Electronic Supplementary Information for:

Linker design for the modular assembly of multifunctional and targeted platinum(II)-containing anticancer agents

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S1. NMR SPECTROSCOPY



Figure S1. ¹H NMR spectrum of **P1** in MeOH- d_4 .



Figure S2. ¹³C NMR spectrum of P1 in MeOH- d_4 .



Figure S3. ¹H NMR spectrum of **P2** in MeOH- d_4 .

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Figure S4. ¹³C NMR spectrum of P2 in MeOH- d_4 .



Figure S5. ¹H NMR spectrum of **P1-N7** in DMF- d_7 .



Figure S6. ¹³C NMR spectrum of P1-N7 in DMF- d_7 .



Figure S7. 1 H- 1 H COSY spectrum of compound **P1-N7** in DMF- d_{7} .



Figure S8. ${}^{1}\text{H}-{}^{13}\text{C}$ HMBC spectrum of compound **P1-N7** in DMF- d_7 .



S2. LC-MS ANALYSIS OF PURIFIED COMPOUNDS





Figure S10. LC-MS analysis of purified P2.



Figure S11. LC-MS analysis of purified P1-N7.



Figure S12. Synthesis of **P1** and **P2**. i) AgNO₃, DMF, rt, ii) HOOC(CH₂)₂C(O)O(CH₂)₃CN (**3**), DMF, 60 °C, 4 h, iii) N¹-(acridin-9-yl)-N²-methylethane-1,2-diamine (**A1**), DMF, 4 °C, iv) N¹-(benzo[c]acridin-7-yl)-N²-methylethane-1,2-diamine (**B1**), DMF, 4 °C.

S3. LC-MS ANALYSIS OF COUPLING REACTIONS



Figure S13. LC-MS analysis of the reaction mixture for the preparation of P1-N1.



Figure S14. LC-MS analysis of the reaction mixture for the preparation of P1-N2.





Figure S16. LC-MS analysis of the reaction mixture for the preparation of P1-N5.





Figure S18. (A) Reverse-phase HPLC trace for the reaction of **P1** and **N7**. (B) ESMS spectrum recorded in positive-ion mode. Characteristic molecular and fragment ions for **P1-N7** are m/z $[M]^+$ 1113.6, $[M+H]^{2+}$ 557.3, $[M+2H]^{3+}$ 371.6.





Figure S20. LC-MS analysis of the reaction mixture for the preparation of P2-N1.



Figure S21. LC-MS analysis of the reaction mixture for the preparation of P2-N2.



Figure S22. LC-MS analysis of the reaction mixture for the preparation of P2-N4.



Figure S23. LC-MS analysis of the reaction mixture for the preparation of P2-N5.



Figure S24. LC-MS analysis of the reaction mixture for the preparation of P2-N6.



Figure S25. LC-MS analysis of the reaction mixture for the preparation of P2-N7.



Figure S26. LC-MS analysis of the reaction mixture for the preparation of P2-N9.



Figure S27. (A) "Two-step" synthesis of platinum-acridine conjugates under aqueous conditions. (B) Reverse-phase HPLC trace of the reaction mixture for the preparation of **P2-N11**. (C) ESMS spectrum of **P2** and target conjugate **P2-N11** recorded in positive-ion mode. (D) Structures and characteristic molecular ions and fragment ions (m/z) for **P2** and **P2-N11**.





Figure S28. LC-ESMS analysis of the mixture of compound **P1-N2** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S29. LC-ESMS analysis of the mixture of compound **P1-N5** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S30. LC-ESMS analysis of the mixture of compound **P1-N7** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S31. LC-ESMS analysis of the mixture of compound **P1-N9** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S32. LC-ESMS analysis of the mixture of compound **P2-N1** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S33. LC-ESMS analysis of the mixture of compound **P2-N6** in phosphate buffer (PB, pH 7.4) incubated at 37 °C.



Figure S34. LC-ESMS analysis of the mixture of compound **P1-N2** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.



Figure S35. LC-ESMS analysis of the mixture of compound **P1-N5** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.



Figure S36. LC-ESMS analysis of the mixture of compound **P1-N7** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.



Figure S37. LC-ESMS analysis of the mixture of compound **P1-N9** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.



Figure S38. LC-ESMS analysis of the mixture of compound **P2-N1** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.



Figure S39. LC-ESMS analysis of the mixture of compound **P2-N6** in phosphate buffered saline (PBS, pH 7.4) incubated at 37 °C.

Compd	Conjugate structures	Conv. (%) ± S.D. ^a	Calculated $[M]^+$	Observed [M] ⁺
P1-N1		98.9 ± 1.4	808.2778	808.2786
P2-N1		98.1 ± 1.7	858.2934	858.2940
P1-N2		96.6 ± 1.2	963.4955	963.4954
P2-N2		96.9 ± 2.3	1013.5111	1013.5125
P1-N4		99.5 ± 1.9	910.3458	910.3466
P2-N4		99.4 ± 1.1	960.3615	960.3610
P1-N5	$H_{2N} \rightarrow H_{2}$	96.4 ± 2.5	1045.3619	1045.3645
P2-N5		98.9 ± 1.4	1095.3776	1095.3784
P1-N6		97.1 ± 2.3	1095.4227	1095.4242
P2-N6		96.7 ± 0.9	1145.4383	1145.4391
P1-N7		97.8 ± 1.5	1111.3604	1111.3615
P2-N7		96.4 ± 2.5	1162.3760	1162.3744
P1-N9		99.8 ± 0.2	612.7688	612.7696
P2-N9		99.7 ± 0.5	637.7766 (2+) [»]	637.7769 (2+) ^ø

Table S1. Conversion Yields and HR-ESMS Results for Conjugates

^{*a*} Each reaction was performed in triplicate. The conversion yields were determined from HPLC traces recorded at an acridine-specific wavelength. Reactions with conversion yields lower than 90% are not reported. ^{*b*} Singly charged $[M]^+$ not observed in positive-ion mode. The additional fused ring in the benz[c]acridine derivatives (**P2**) is shown as dashed bonds.

Acronym, Common Name	System. Nomenclature	Structure
EDC	<i>N</i> -(3-Dimethylaminopropyl)- <i>N'</i> - ethylcarbodiimide hydrochloride	$H_{\text{CI}} \sim N^{2} C^{2} N^{2}$
DCC	<i>N,N'</i> -Dicyclohexylcarbodiimide	
CDI	1,1'-Carbonyldiimidazole	
РуВОР	Benzotriazol-1-yl- oxytripyrrolidinophosphonium hexafluorophosphate	PF ₆ N-O-P-N
СОМИ	1-[(1-(Cyano-2-ethoxy-2- oxoethylideneaminooxy)-dimethylamino- morpholinomethylene)] methanaminium hexafluorophosphate	$ \begin{array}{c} $
TSTU	O-(N-Succimidinyl)- N,N,N',N'- tetramethyluronium tetrafluoroborate	$ \begin{array}{c} $
HBTU	<i>O</i> -(Benzotriazol-1-yl)- <i>N,N,N',N'</i> - tetramethyluronium hexafluorophosphate	$PF_{6}^{-} N$

Table S2.	Summary	of Cour	ling Rea	ogents Used
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