## Supplementary Information

## Insight into the roles of structures and energy levels of mono- and bis-β-diketones on sensitizing Nd(III) NIR-luminescence

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Figure S1 400 MHz <sup>1</sup>H NMR spectrum of BDT in CDCl<sub>3</sub>



Figure S2 Thermogravimetric curves of 1, 2 and 3.



Figure S3 The bending patterns of the ligands in 2 and 3.



**Figure S4** Fluorescence spectra of BDT and BTT in their Nd (red line) and Gd(III) complexes (black line).



Figure S5 FT–IR spectra of 1, 2 and 3.



**Figure S6** UV/Vis absorption spectra of a  $3 \times 10^{-6}$  M solution of [BDT]<sup>2–</sup> in DMSO upon addition of increasing amounts of NdCl<sub>3</sub>, from 0 to 2.5 equiv. (left); Calculated UV/Vis absorption spectra of the species (right). Inset: Species distribution diagram drawn with the conditional constants reported in Table 1 for Nd<sub>2</sub>(BDT)<sub>3</sub>.



**Figure S7** UV/Vis absorption spectra of a  $6 \times 10^{-6}$  M solution of [TTA]<sup>-</sup> in DMSO upon addition of increasing amounts of NdCl<sub>3</sub>, from 0 to 1.25 equiv. (left); Calculated UV/Vis absorption spectra of the species (right). Inset: Species distribution diagram drawn with the conditional constants reported in Table 1 for Nd(TTA)<sub>3</sub>.



**Figure S8** Emission spectra of a  $3 \times 10^{-6}$  M solution of  $[BTT]^{2-}$  in DMSO upon addition of increasing amounts of NdCl<sub>3</sub>, from 0 to 2.5 equiv. (left); Calculated emission spectra of the species (right).



**Figure S9** Emission spectra of a  $3 \times 10^{-6}$  M solution of [BDT]<sup>2–</sup> in DMSO upon addition of increasing amounts of NdCl<sub>3</sub>, from 0 to 2.5 equiv. (left); Calculated emission spectra of the species (right).



**Figure S10** Emission spectra of a  $6 \times 10^{-6}$  M solution of [TTA]<sup>-</sup> in DMSO upon addition of increasing amounts of NdCl<sub>3</sub>, from 0 to 1.25 equiv. (left); Calculated emission spectra of the species (right).

Complexe s		H <sub>3</sub> C CH <sub>3</sub>	$C_{5} \xrightarrow{S}_{C_{4}} \xrightarrow{O}_{C_{3}} \xrightarrow{O}_{C_{4}} \xrightarrow{O}_{C_{3}} \xrightarrow{O}_{C_{4}} \xrightarrow{O}_{C_{4$	$ \begin{array}{c}         S \\         C_3 \\         BDT \\         O \\         CF_3 \\         CF_3         S         S         $	
	Nd <sup></sup> C <sub>mothyno</sub>	Nd <sup></sup> Cpmso	Nd <sup></sup> C <sub>3</sub>	NdC4	Nd…C5
	3 8032	4 1854	5 9504	6 9177	6 6266
	3 8798	4 3456	6 1007	7 0315	6 7011
1	3 8853	4 6102	6 1395	7 0346	6 744
	5.0005	5.0333			
Average	3.8516	4.5436	6.0635	6.9946	6.6905
menage	3.9443	4.9360	6.0270	7.0299	
	3.9672	4.8609	6.0609	7.0567	
2	3.9443	4.8395	6.0944	7.0816	
		4.8069			
		4.7058			
		4.5446			
Average	3.9646	4.7822	6.0607	7.0560	
0	3.7926	4.3003	5.7048	6.6877	
	3.8485	4.3412	5.8653	6.7991	
3	3.8488	4.9135	6.0046	7.0816	
		4.9196			
Average	3.8299	4.6186	5.8582	6.8561	

 Table S1 Distances of metal centres to C–H bonds.

	1	2	3
Formula	$C_{28}H_{24}F_{9}NdO_{8}$	$C_{50}H_{54}F_{18}Nd_2O_{19}$	$C_{60}H_{54}F_{18}Nd_2O_{18}$
Formula	$S_5$	$\mathbf{S}_{10}$	S <sub>12</sub>
Mr	964.01	1910.01	2078.35
Color	colorless	yellow	yellow
Cryst. syst.	monoclinic	monoclinic	monoclinic
Space group	$P2_1/n$	C2/c	C2/c
a ( Å)	12.2314(2)	19.7938(5)	41.621(4)
<i>b</i> ( Å)	17.1983(4)	28.2072(5)	11.9064(10)
c ( Å)	17.3127(3)	15.0963(4)	16.7402(15)
a (deg)	90	90	90
$\beta$ (deg)	91.1317(16)	118.891(4)	98.961(2)
γ (deg)	90	90	90
V(Å3)	3641.15(12)	7379.7(3)	8194.5(13)
Ζ	4	4	4
ρ (g cm <sup>3</sup> )	1.759	1.719	1.685
μ (mm <sup>-1</sup> )	1.804	1.782	1.661
F(000)	1908.0	3792.0	4128.0
$R_1, [I > 2\sigma(I)]$	0.0519	0.0321	0.0682
$wR_2, [I \ge 2\sigma(I)]$	0.1025	0.0720	0.1516
R <sub>1</sub> ,(all data)	0.0635	0.0466	0.1428
wR <sub>2</sub> ,(all data)	0.1091	0.0813	0.1916
GOF on $F^2$	1.036	1.064	0.992

 Table S2 Crystal data and structure refinements for 1, 2 and 3.

Table S3	Binding	constants	and	species	distribution	obtained	by	fitting
the change	es in emis	ssion spect	tra.					

Spacias		10 00	% Species for	Concentration of		
	Species	logp <sub>M/L</sub>	M/L	species (10 <sup>-6</sup> M)		
1	$\log \beta_{13}$	28.213	61.3	1.411		
	$\log \beta_{12}$	22.168	23.1	0.531		
	$\log \beta_{11}$	12.544	0.2	0.004		
	[TTA] <sup>-</sup>	—	15.4	0.354		
2	$\log \beta_{23}$	28.200	68.1	0.754		
	$\log \beta_{22}$	21.532	15.8	0.175		
	$\log \beta_{21}$	14.160	1.1	0.012		
	[BDT] <sup>2–</sup>	_	14.1	0.156		
3	$log\beta_{23}$	28.164	65.3	0.725		
	$\log \beta_{22}$	21.741	17.4	0.193		
	$\log \beta_{21}$	14.189	0.8	0.009		
	[BTT] <sup>2–</sup>	_	16.5	0.183		