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Supporting information for:

# PNPCB Heterocycles via Thermal and Lewis Acid Catalyzed trans-Hydroborations

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#### 1. NMR Spectra

Figure 1 <sup>1</sup>H NMR spectrum of 8 (400 MHz,  $d_6$ -benzene, 298K).



Figure 2 <sup>13</sup>C NMR spectrum of 8 (100 MHz, *d*<sub>6</sub>-benzene, 298K)





**Figure 3** <sup>31</sup>P NMR spectrum of **8** (162 MHz, *d*<sub>6</sub>-benzene, 298K).







210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)











Figure 10 <sup>13</sup>C NMR spectrum of 10 (100 MHz, *d*<sub>6</sub>-benzene, 298K)



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)

**Figure 11** <sup>31</sup>P NMR spectrum of **10** (162 MHz, *d*<sub>6</sub>-benzene, 298K)







**Figure 15** <sup>31</sup>P NMR spectrum of **11** (162 MHz, *d*<sub>6</sub>-benzene, 298K)





210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 10 f1 (ppm)



Figure 21 <sup>1</sup>H NMR spectrum of 13 (400MHz, *d*<sub>6</sub>-benzene, 298K)



Figure 22 <sup>13</sup>C NMR spectrum of 13(100 MHz, *d*<sub>6</sub>-benzene, 298K)







Figure 23 <sup>31</sup>P NMR spectrum of 13 (162 MHz, *d*<sub>6</sub>-benzene, 298K)

Figure 24 <sup>11</sup>B NMR spectrum of 13 (128MHz, *d*<sub>6</sub>-benzene, 298K)









210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 10 f1 (ppm)

**Figure 27** <sup>31</sup>P NMR spectrum of **14** (100 MHz, *d*<sub>6</sub>-benzene, 298K)







**Figure 31** <sup>31</sup>P NMR spectrum of **15** (100 MHz, *d*<sub>6</sub>-benzene, 298K)





Figure 32 <sup>11</sup>B NMR spectrum of 15 (100 MHz, *d*<sub>6</sub>-benzene, 298K)





### 2. Crystallographic Details:

#### Table 1 Geometric Parameters for Compounds 8-10, 11, 13-15

Coumpound	8	9	10	11
Emperical Formula	C <sub>22</sub> H <sub>40</sub> BNP <sub>2</sub>	C <sub>20</sub> H <sub>44</sub> BNP <sub>2</sub>	C <sub>22</sub> H <sub>46</sub> BNP <sub>2</sub>	C <sub>26</sub> H <sub>32</sub> BNP <sub>2</sub>
Crystal System	Monoclinic	Triclinic	Triclinic	Monoclinic
Space Group	C2/c	P-1	P-1	P2(1)/n
a/ Å	16.6512(13)	8.8873(4)	8.3977(8)	17.1910(12)
b/ Å	8.7599(5)	9.6687(5)	9.8974(8)	9.0756(8)
c/ Å	32.962(2)	15.5632(8)	17.1444(14)	17.2932(13)
α/ °	90	85.608(2)	86.080(5)	90
β/ °	97.279(6)	88.203(2)	78.918(4)	111.777(4)
γ/ °	90	65.678(2)	65.594(4)	90
V/ Å <sup>3</sup>	4769.2(6)	1215.05(10)	1273.34(19)	2505.3
Z	8	2	2	4
т/ к	150	150	150	150
D <sub>c</sub> / g.cm <sup>-3</sup>	1.090	1.015	1.036	1.143
Total data	5453	5573	5791	5713
Unique data	3775	4675	4840	3823
R <sub>int</sub>	0.0699	0.0297	0.0348	0.0915
R <sub>1</sub> [F <sup>2</sup> >2 σ(F <sup>2</sup> )]	0.0457(3775)	0.0348(4675)	0.0363(4840)	0.0463(3823)
wR2 (all data)	0.1148(5453)	0.1319(5573)	0.1316(5791)	0.1061(5713)
GoF	1.013	1.009	0.985	1.009
min/ max/ eÅ-3	0.396/-0.340	0.354/-0.295	0.0329/-0.267	0.396/ -0.340

	13	14	15
Emperical Formula	$C_{26}H_{38}BNP_2$	$C_{26}H_{32}BNP_2$	$C_{22}H_{40}BNP_2$
Crystal System	Monoclinic	Triclinic	Monoclinic
Space Group	P2(1)/n	P-1	P21/c
a/ Å	8.9183(5)	10.3495(7)	9.7047(7)
b/ Å	16.6266(8)	15.2609(11)	16.3976(9)
c/ Å	17.4581(10)	16.9887(12)	15.2998(10)
α/ °	90	108.455(3)	90
β/ °	91.698(2)	104.696(3)	105.659(4)
γ/ °	90	92.445(4)	90
V/ Å <sup>3</sup>	2587.6(2)	2439.8(3)	2344.4(3)
Z	4	2	4
т/ к	150	150	150
D <sub>c</sub> / g.cm <sup>-3</sup>	1.123	1.174	1.109
Total data	5962	11157	5738
Unique data	4443	9548	3064
R <sub>int</sub>	0.0494	0.0380	
R <sub>1</sub> [F <sup>2</sup> >2 σ(F <sup>2</sup> )]	0.0418(4443)	0.0395(9548)	0.0846(3064)
wR2 (all data)	0.1102(5962)	0.1455(11157)	0.1711(5738)
GoF	1.010	1.148	1.034
min/max/eÅ-3	0.422/-0.267	0.468/-0.359	0.750/-0.579