

Supporting Information for

**Room-Temperature Switching of Magnetic Hysteresis
by Reversible Single-Crystal-to-Single-Crystal
Solvent Exchange in Imidazole-Inspired Fe(II)
Complex**

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Table S1. Summary of the Solvent- and Anion-Ligand Hydrogen Bond Parameters Found in **1-Et₂O** and **1-3H₂O**.

complex	D-H	A	d(D-H)/Å	d(H...A)/Å	d(D...A)/Å	∠(DHA)/deg
1-Et₂O @298K	N2-H2B	F3	0.86	2.339	2.984	132.1
	N2-H2B	F1	0.86	2.254	2.821	123.41
	N2-H2B	F4	0.86	2.541	3.072	120.82
	N5-H5A	O2	0.86	2.236	3.085	169.22
	N6-H6A	F2	0.86	2.106	2.931	160.49
	N6-H6A	F4	0.86	2.342	3.053	140.22
1-Et₂O @120K	N2-H2B	F1	0.88	2.072	2.903	157.12
	N3-H3A	O3	0.88	2.16	3.024	167.19
	N7-H7B	F2	0.88	2.584	3.062	115.02
	N7-H7B	F3	0.88	2.321	3.014	135.67
	N7-H7B	F4	0.88	2.292	2.845	120.82
	N8-H8B	F5	0.88	2.113	2.741	127.7
	N8-H8B	F6	0.88	2.444	3.085	130.06
	N13-H13A	F7	0.88	1.982	2.835	162.94
complex	D-H	A	d(D-H)/Å	d(H...A)/Å	d(D...A)/Å	∠(DHA)/deg
1-3H₂O @298K	C29-H29A	F7A_b	0.95	2.456	3.026	118.43
	N2-H2B	F2_a	0.88	2.108	2.958	162.11
	N2-H2B	F4A_b	0.88	2.191	2.887	135.7
	N6-H6B	F1	0.88	2.357	2.913	121.3
	N6-H6B	F3	0.88	2.356	3.043	134.97
	N6-H6B	F4_a	0.88	2.576	3.092	118.33
	N8-H8B	F5	0.88	2.053	2.91	164.4
	N10-H10A	O3W	0.88	2.219	3.084	167.55
	N14-H14B	F8_a	0.88	1.988	2.71	138.47
1-3H₂O @120K	N1-H1B	F7	0.86	2.006	2.84	162.96
	N7-H7B	F5	0.86	2.359	2.991	130.69
	N7-H7B	F6	0.86	2.141	2.758	128.31
	N8-H8B	F4	0.86	2.062	2.88	158.77
	N14-H14B	F1	0.86	2.276	2.947	134.94
	N14-H14B	F2	0.86	2.332	2.884	122.25
	O1W-H1WA	F3	0.85	2.436	2.899	115.03
	O1W-H1WB	O3W	0.85	1.974	2.789	160.38
	O2W-H2WB	O1W	0.85	1.855	2.676	161.98
	O3W-H3WA	O4W	0.85	2.079	2.578	117.02

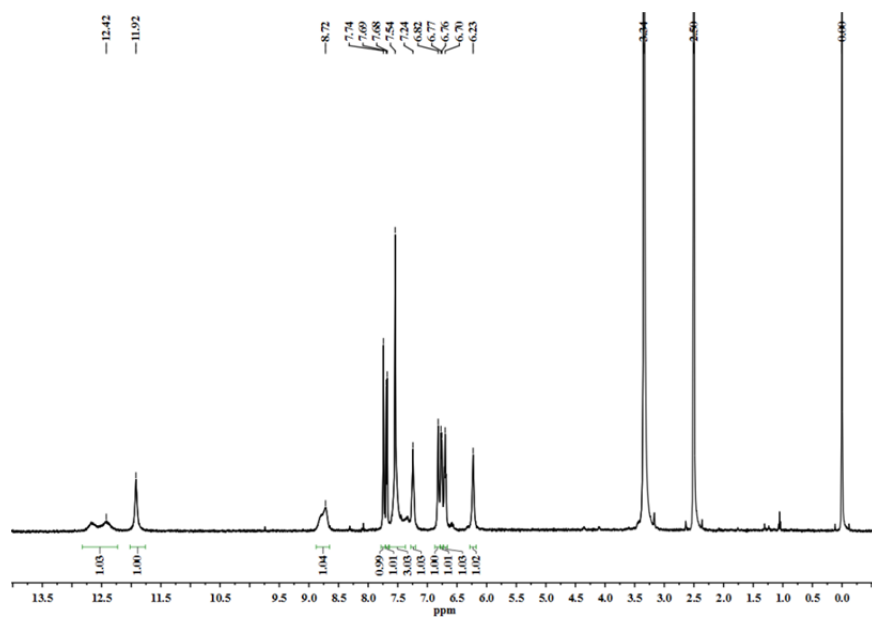


Figure S1. ^1H NMR spectrum of **Liq** in DMSO at room temperature.

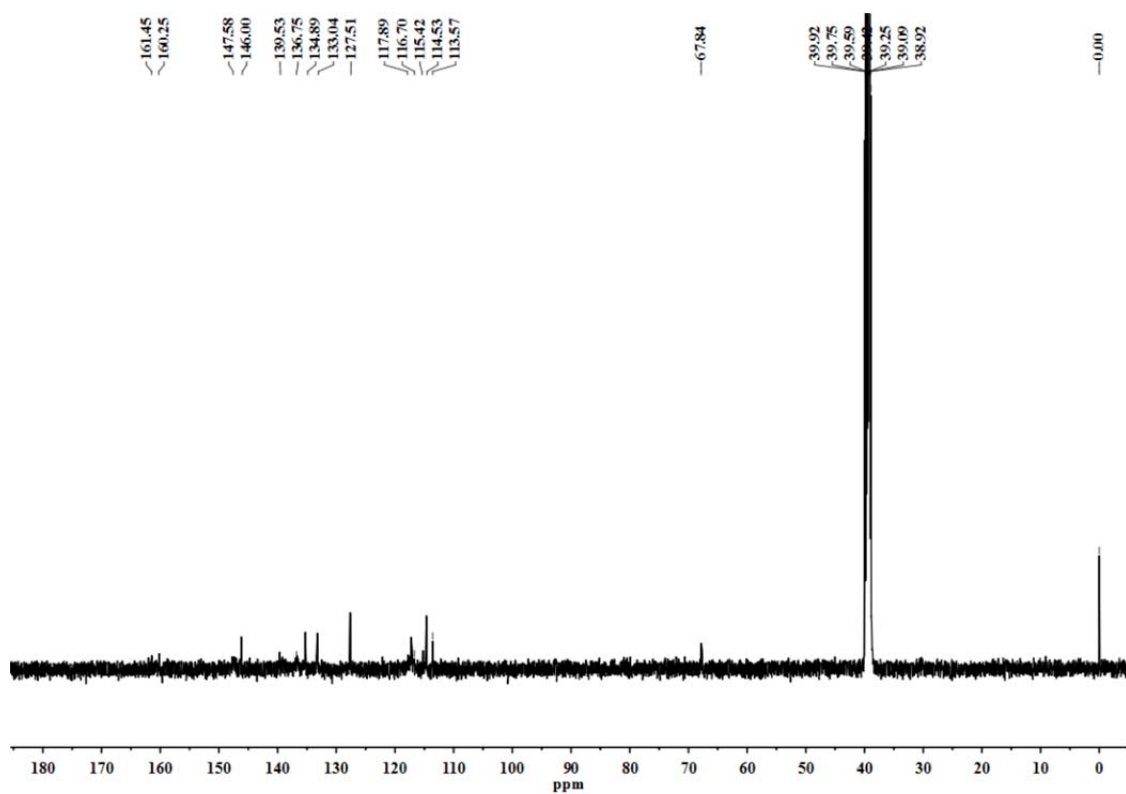


Figure S2. ^{13}C NMR spectrum of **Liq** in DMSO at room temperature.

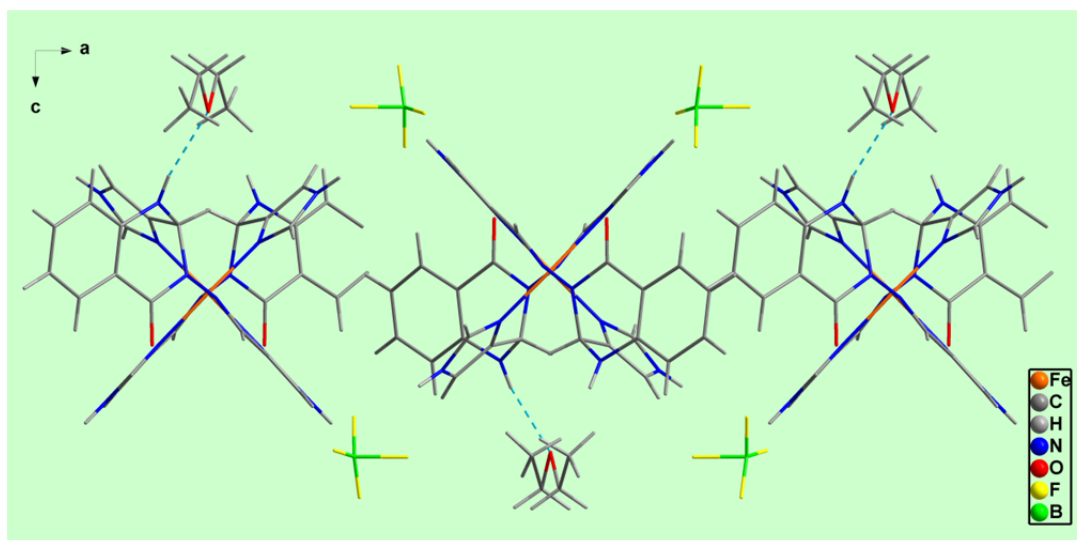


Figure S3. View of the hydrogen bonding interactions between O atoms of Et₂O and N on the ligand through N–H···O interactions in **1**-Et₂O.

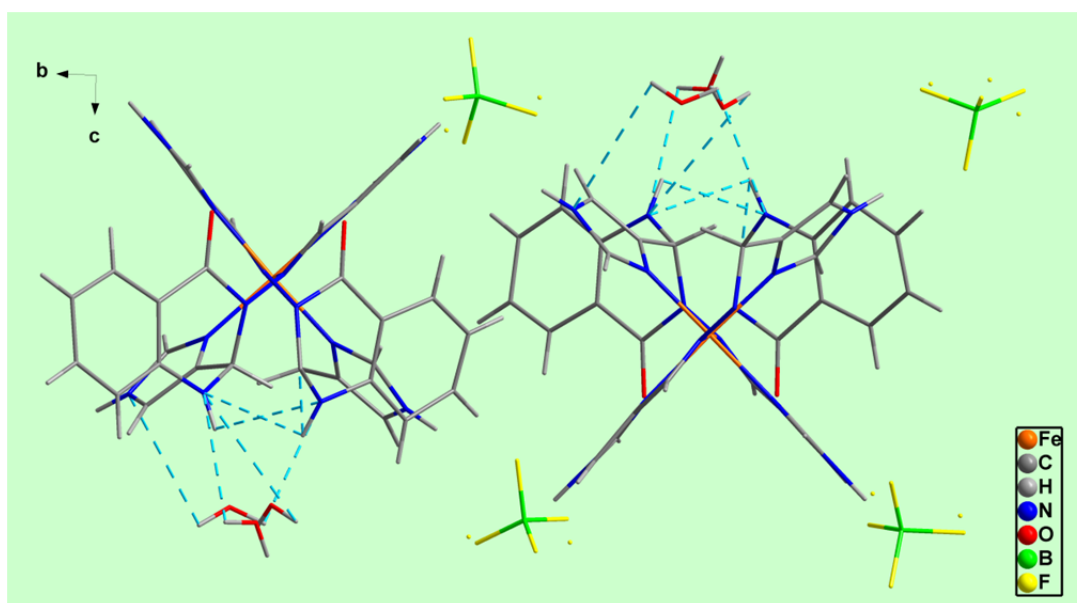


Figure S4. View of the hydrogen bonding interactions between O atoms of H₂O and N on the ligand through N···H–O interactions in **1**-3H₂O.

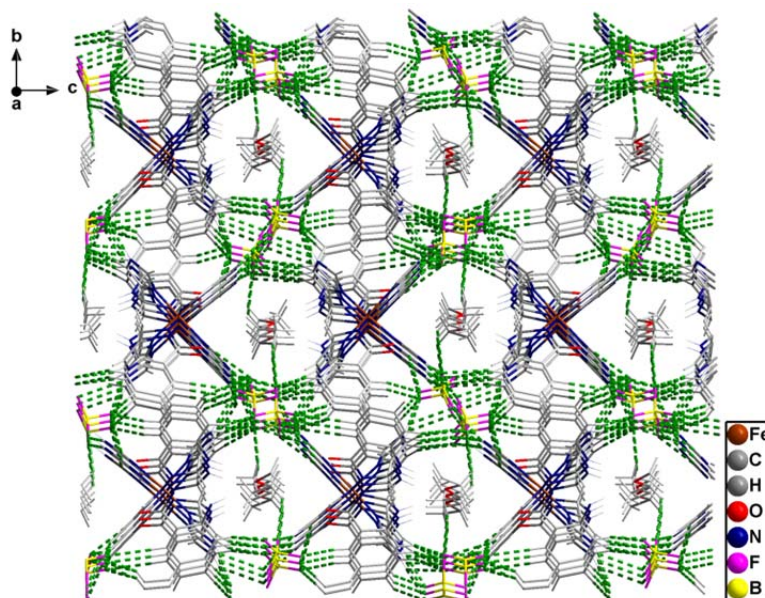


Figure S5. View of the supramolecular three-dimensional arrangement through N-H $\cdots$ F interactions in **1**-Et₂O at 298K.

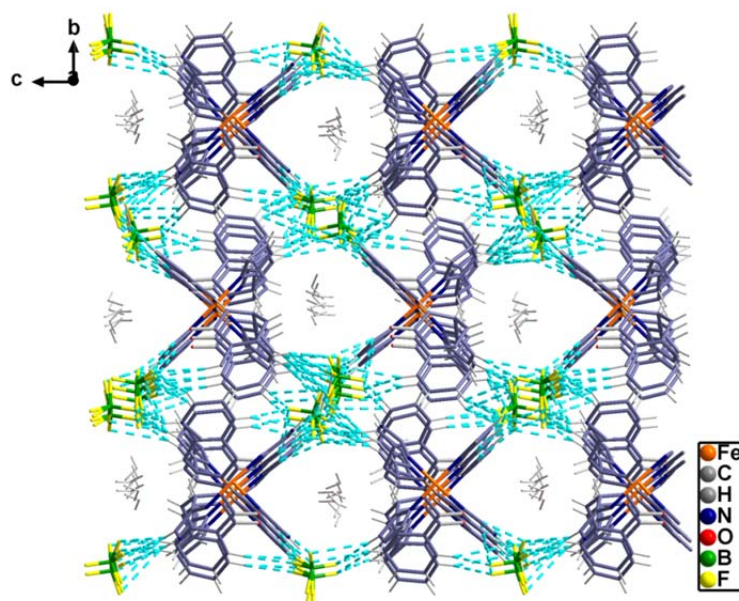


Figure S6. View of the supramolecular three-dimensional arrangement through N-H $\cdots$ F interactions in **1**-3H₂O at 298K.

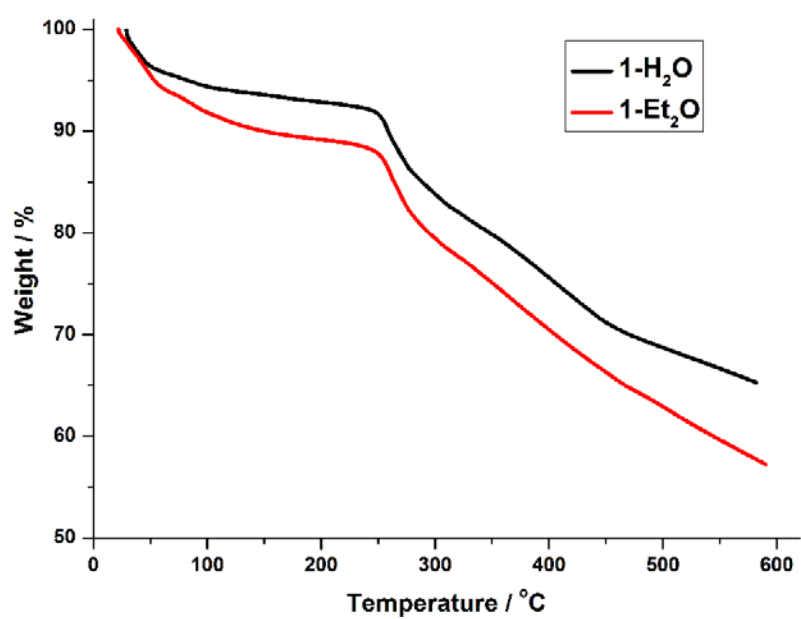


Figure S7. TG analysis of $[\text{Fe}(\text{Liq})_2](\text{BF}_4)_2 \cdot \text{C}_4\text{H}_{10}\text{O}$ (**1-Et₂O**) and $[\text{Fe}(\text{Liq})_2](\text{BF}_4)_2 \cdot \text{H}_2\text{O}$ (**1-3H₂O**).

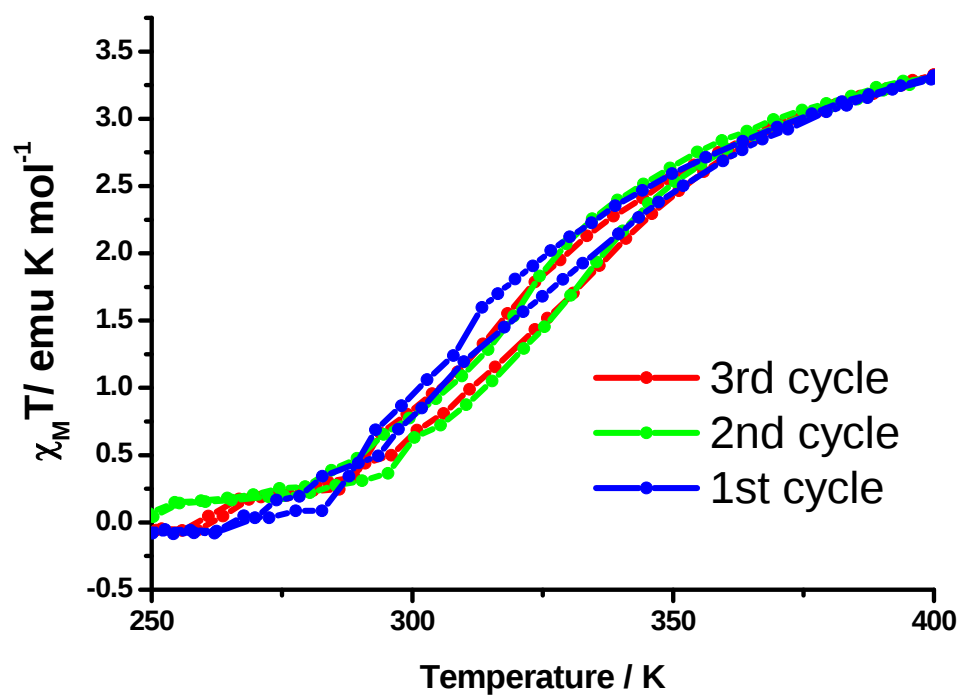


Figure S8. The magnetic data of sample **1**-Et₂O in the temperature range of 5-400 K.

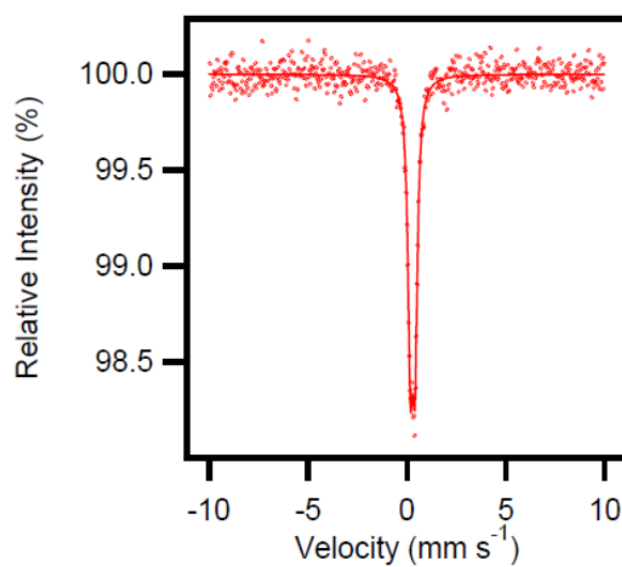


Figure S9. Mössbauer spectrum of complex **1**-3H₂O at 300 K.

Area: 100%

Isomer Shift: 0.300 (Standard Deviation 0.0026)

Quadrupole Splitting: 0.235 (Standard Deviation 0.0047)

Line Width: 0.299 (Standard Deviation 0.0081)

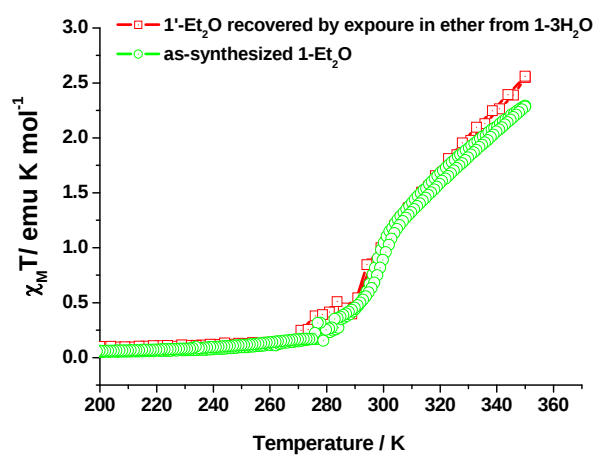


Figure S10. Variable-temperature magnetic susceptibility data of recovered sample **1'**-Et₂O which is exposed in Et₂O vapour by **1**-3H₂O after 2 days and the comparison with the data of as-synthesized **1**-Et₂O.