

Electronic Supplementary information

An electron poor iridium pincer complex in catalytic alkane dehydrogenation

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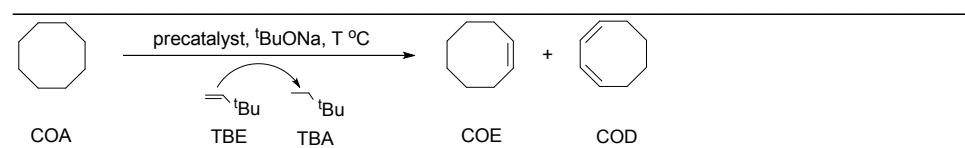
Table S1.Crystal data and refinement details for compounds **4**, **5**, **6** and **8**.

	4	5	6	8
Formula	C ₂₄ H ₃₈ F ₆ O ₂ P ₂	C ₂₄ H ₃₇ ClF ₆ IrO ₂ P ₂	C ₁₆ H ₃₁ ClIrOP	C ₃₈ H ₅₇ F ₆ IrO ₂ P ₂
Fw	534.48	761.12	498.03	913.97
Space group	<i>P</i> 2 ₁ /c	<i>P</i> 2 ₁ /c	<i>P</i> 2 ₁ /n	<i>P</i> 2 ₁
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic
<i>T</i> /K	293(2)	293(2)	293(2)	293(2)
<i>a</i> /Å	12.132(5)	8.318(5)	7.96730(10)	8.709(5)
<i>b</i> /Å	18.353(5)	31.333(5)	15.3786(2)	22.841(5)
<i>c</i> /Å	14.104(5)	12.075(5)	14.8875(3)	10.007(5)
<i>β</i> /deg	109.774(5)	109.444(5)	98.120(2)	92.394(5)
<i>V</i> /Å ³	2955.2(18)	2968(2)	1805.82(5)	1988.9(16)
<i>Z</i>	4	4	4	2
<i>D</i> _{calcd} /g cm ⁻³	1.201	1.704	1.832	1.526
<i>μ</i> /mm ⁻¹	0.202	4.754	7.625	3.496
<i>θ</i> -range /deg	2.935-28.838	2.904-28.719	2.902-28.869	2.943-28.803
No. reflns collected	33067	18421	19901	22511
No. of unique reflns	7037	6622	4335	9107
<i>R</i> (<i>F</i>) (<i>I</i> > 2σ(<i>I</i>)) ^a	0.0649	0.0374	0.0334	0.0550
<i>wR</i> ² (<i>F</i> ²) (all data) ^b	0.1841	0.0867	0.0666	0.1082
<i>S</i> ^c	1.018	1.033	1.054	1.018
<i>R</i> _{int}	0.0429	0.0279	0.0515	0.0783
CCDC	1478500	1478501	1478503	1478502
Flack param.				0.420(9)

^a $R = \Sigma(|F_o| - |F_c|) / \Sigma |F_o|$. ^b $wR2 = [\Sigma w(F_o^2 - F_c^2)^2 / \Sigma (F_o^2)^2]^{1/2}$.

^c $S = [\Sigma w(F_o^2 - F_c^2)^2 / (n-p)]^{1/2}$.

Table S2. Catalytic activity of complex II in the transfer dehydrogenation of COA with TBE.



Entry ^a	t/h	TONs ^b II	Entry ^a	t/h	TONs ^b II
1	0,17	1136	7	8	1621
2	0,5	1346	8	15	1701
3	1	1435	9	24	1730
4	3	1499	10	40	1730
5	4	1530			
6	6	1581			

^a Average of three runs, using a 3030:3030:1 ratio of COA/TBE/precatalyst and 1.5 equiv of ^tBuONa at 200 °C. All reactions were performed under an argon atmosphere.

^b Determined by ¹H NMR, the sum of COE and COD double bonds equals TON of TBE within 2% difference.

Figure S1-1. ^1H NMR spectrum of **1,3-Dibromo-4,6-bis(trifluoromethyl)benzene (1)**
Chloroform- d ; 400 MHz

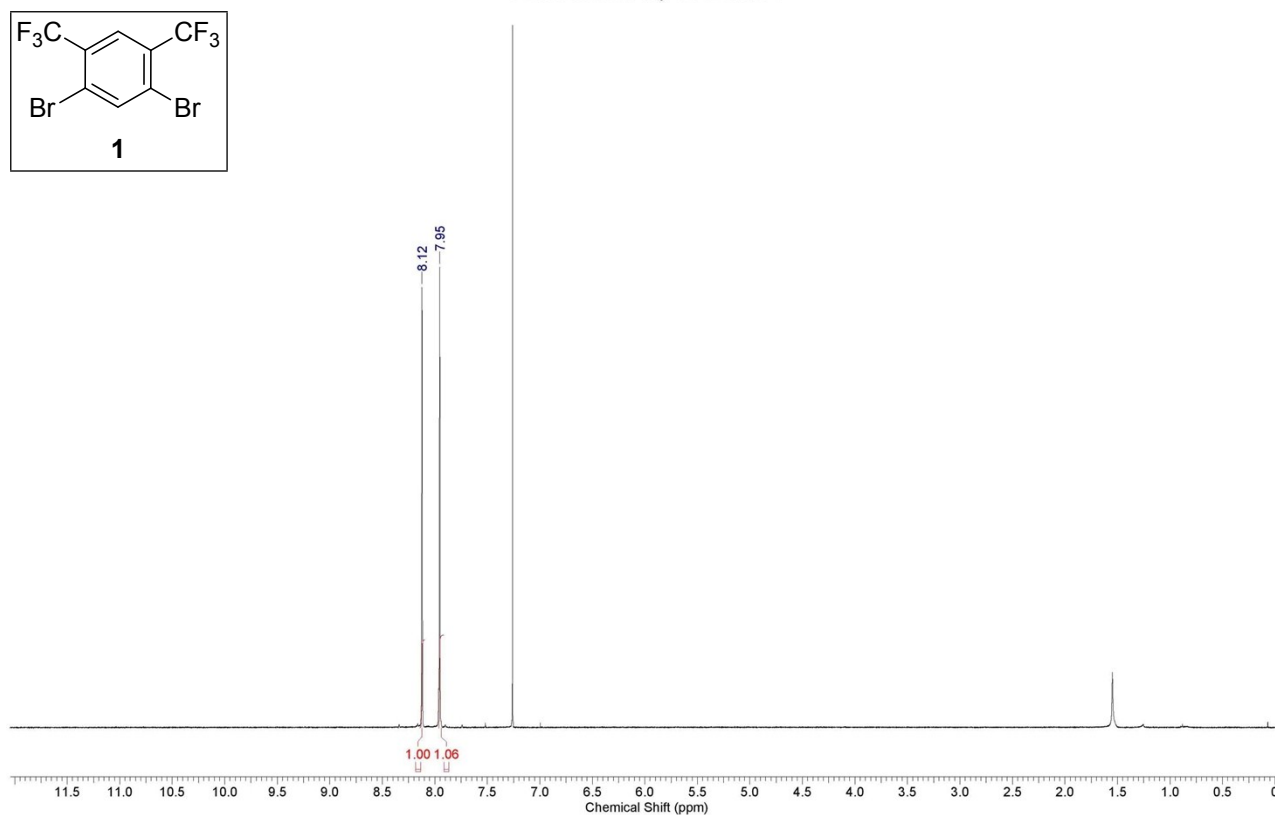


Figure S1-2. ^{19}F NMR spectrum of **1,3-Dibromo-4,6-bis(trifluoromethyl)benzene (1)**
Chloroform- d ; 376 MHz

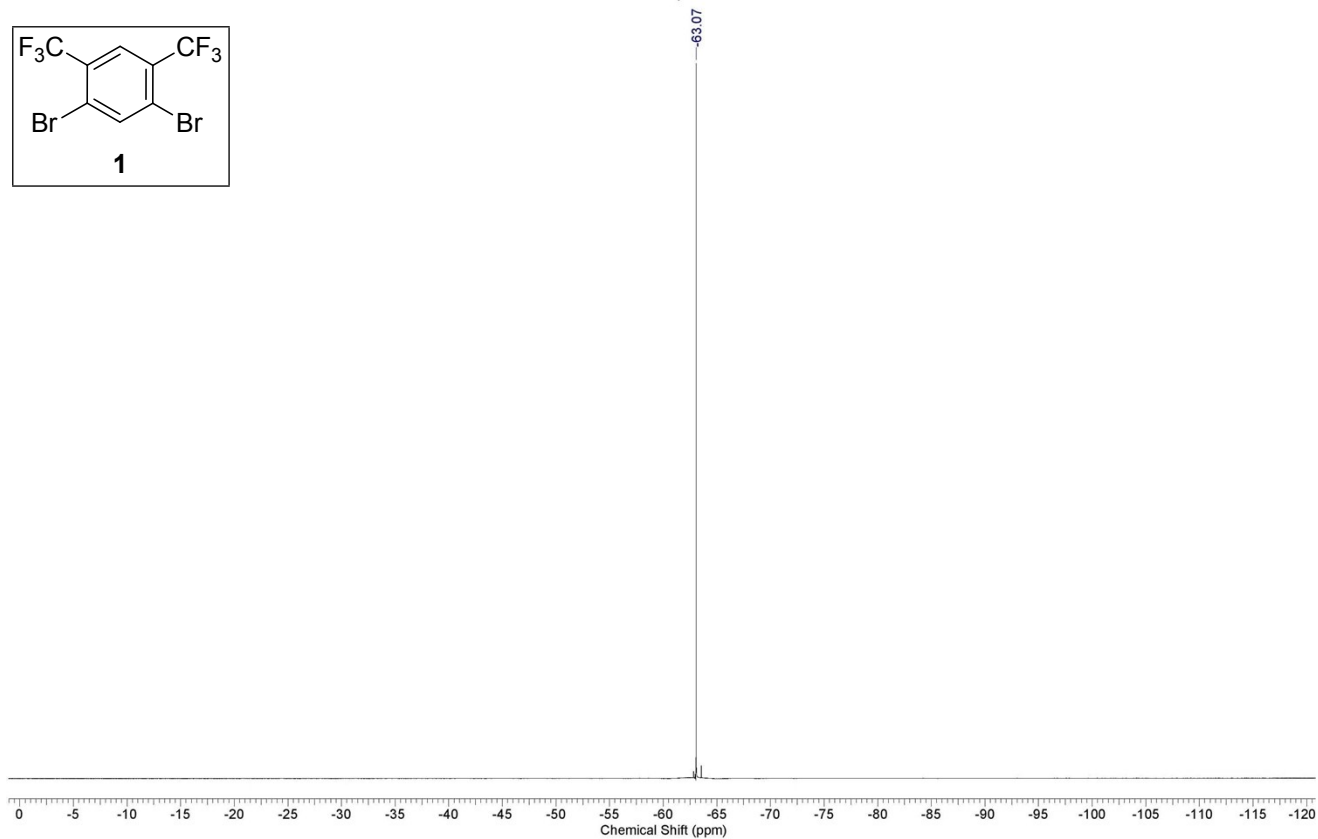


Figure S1-3. ^{13}C NMR spectrum of **1,3-Dibromo-4,6-bis(trifluoromethyl)benzene (1)**
Chloroform- d ; 100 MHz

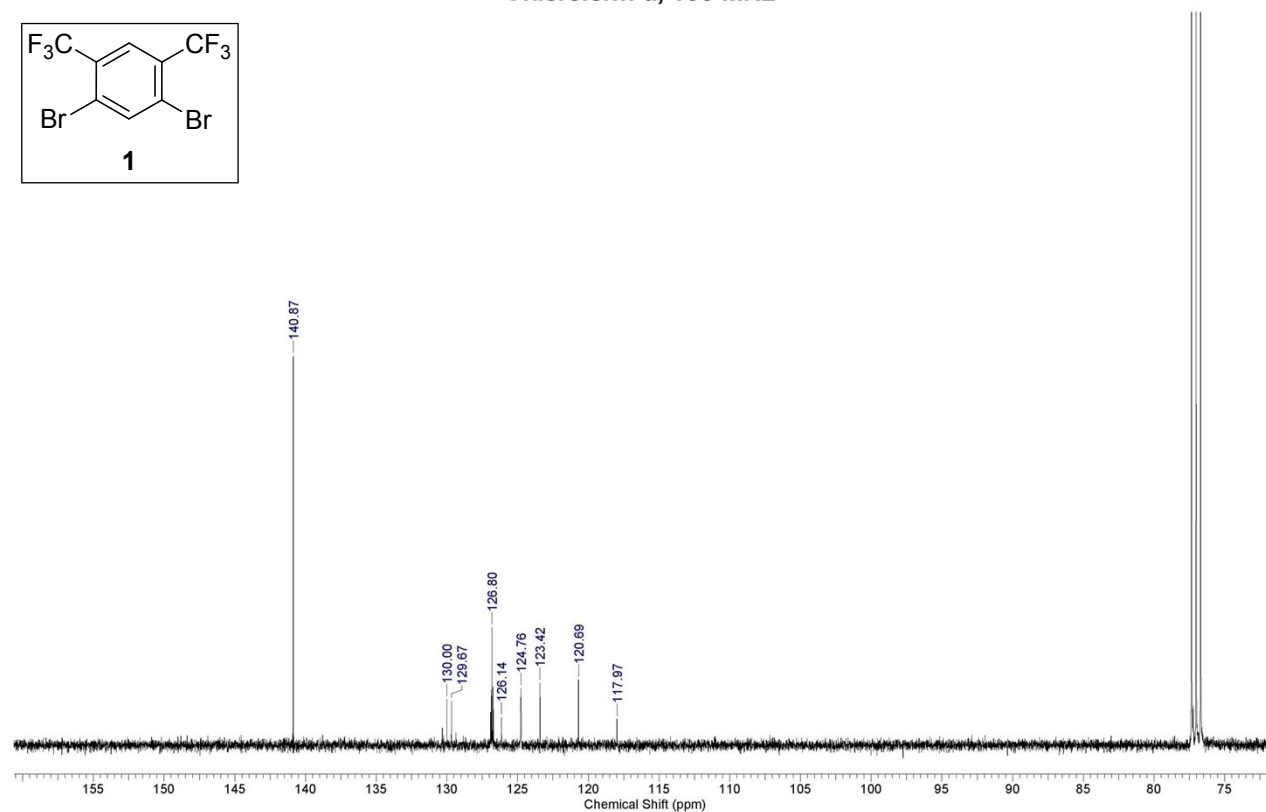


Figure S2-1. ^1H NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol dibenzyl ether (2)**
Chloroform- d ; 400 MHz

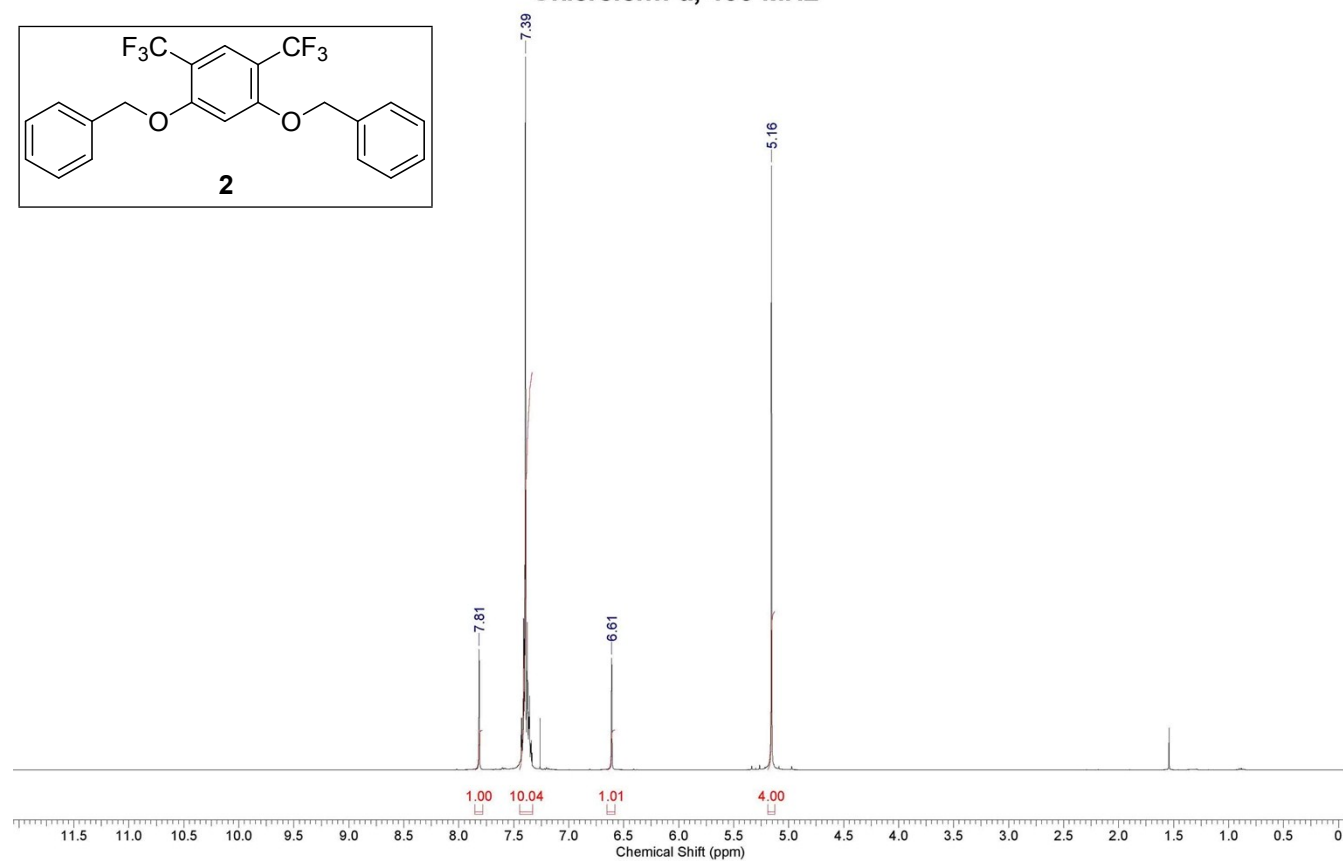


Figure S2-2. ^{19}F NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol dibenzyl ether (2)**
Chloroform- d ; 376 MHz

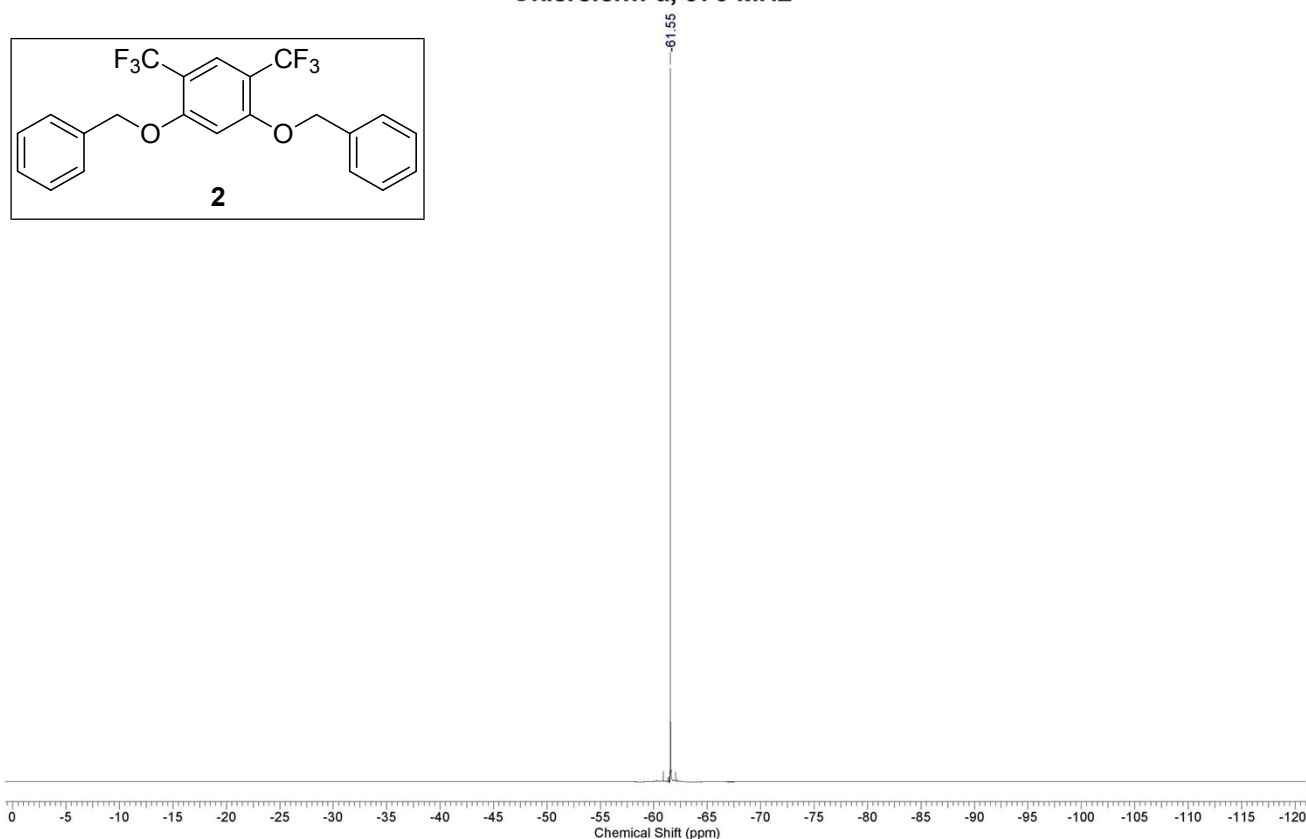


Figure S2-3. ^{13}C NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol dibenzyl ether (2)**
Chloroform- d ; 100 MHz

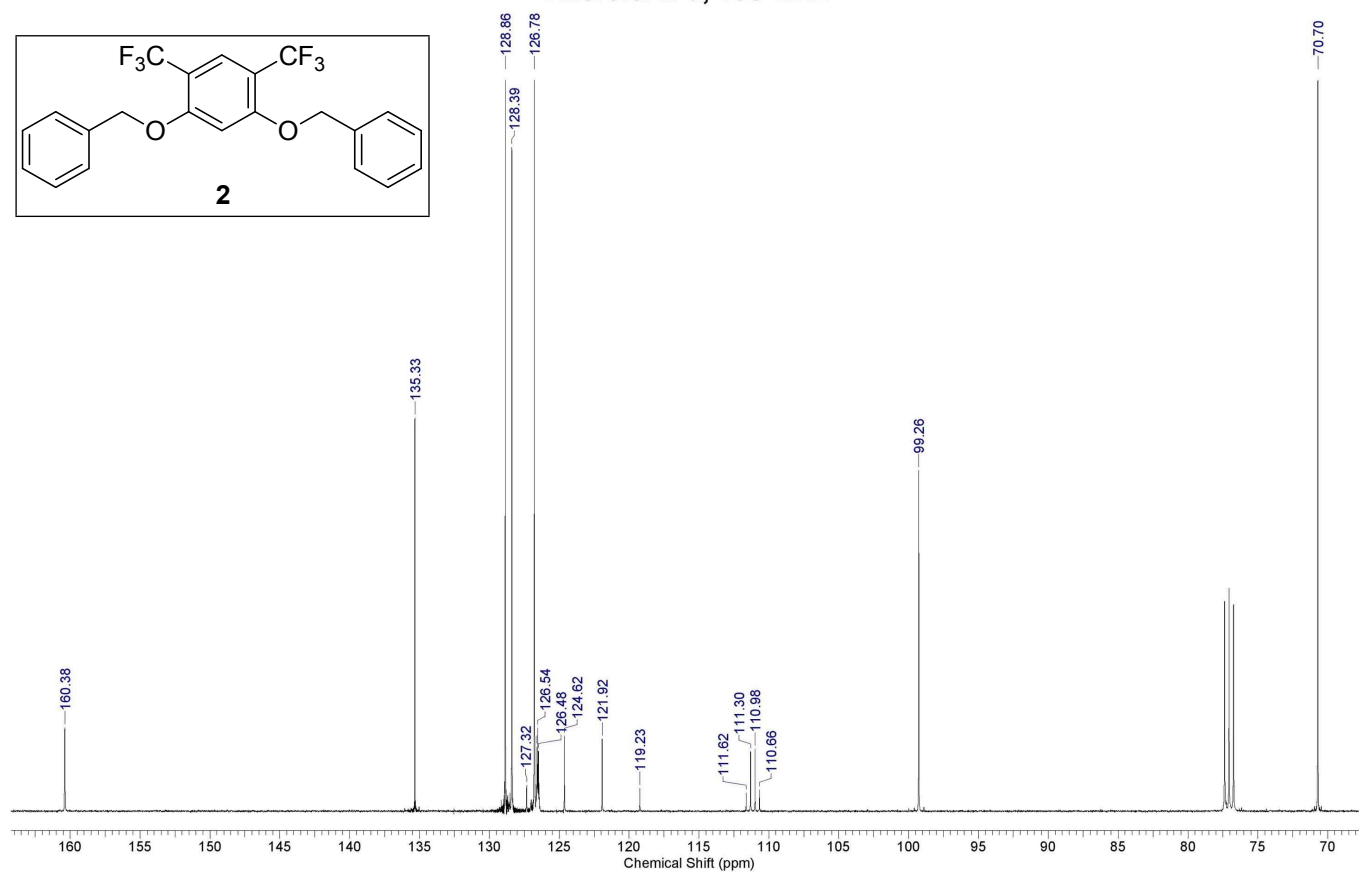


Figure S3-1. ^1H NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol (3)**
Chloroform- d ; 400 MHz

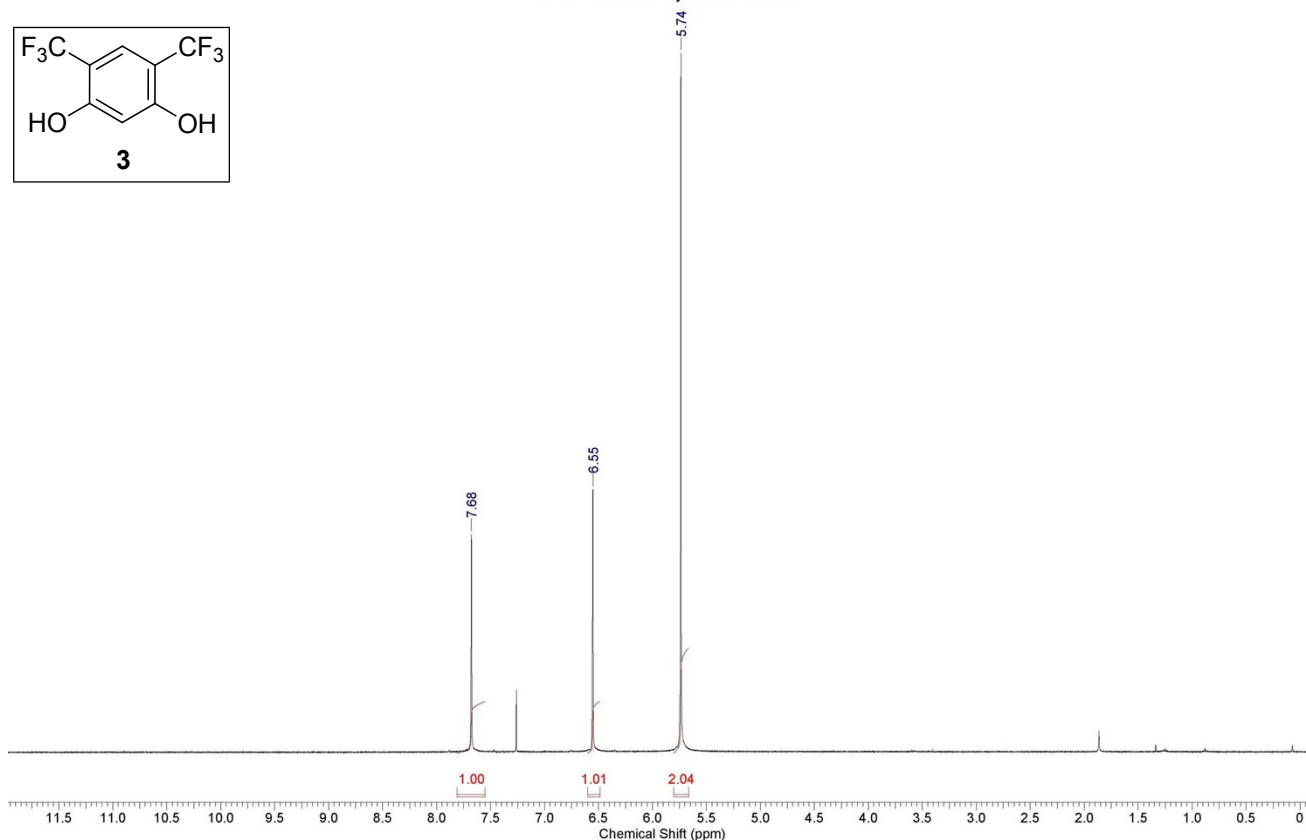


Figure S3-2. ^{19}F NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol (3)**
Chloroform- d ; 376 MHz

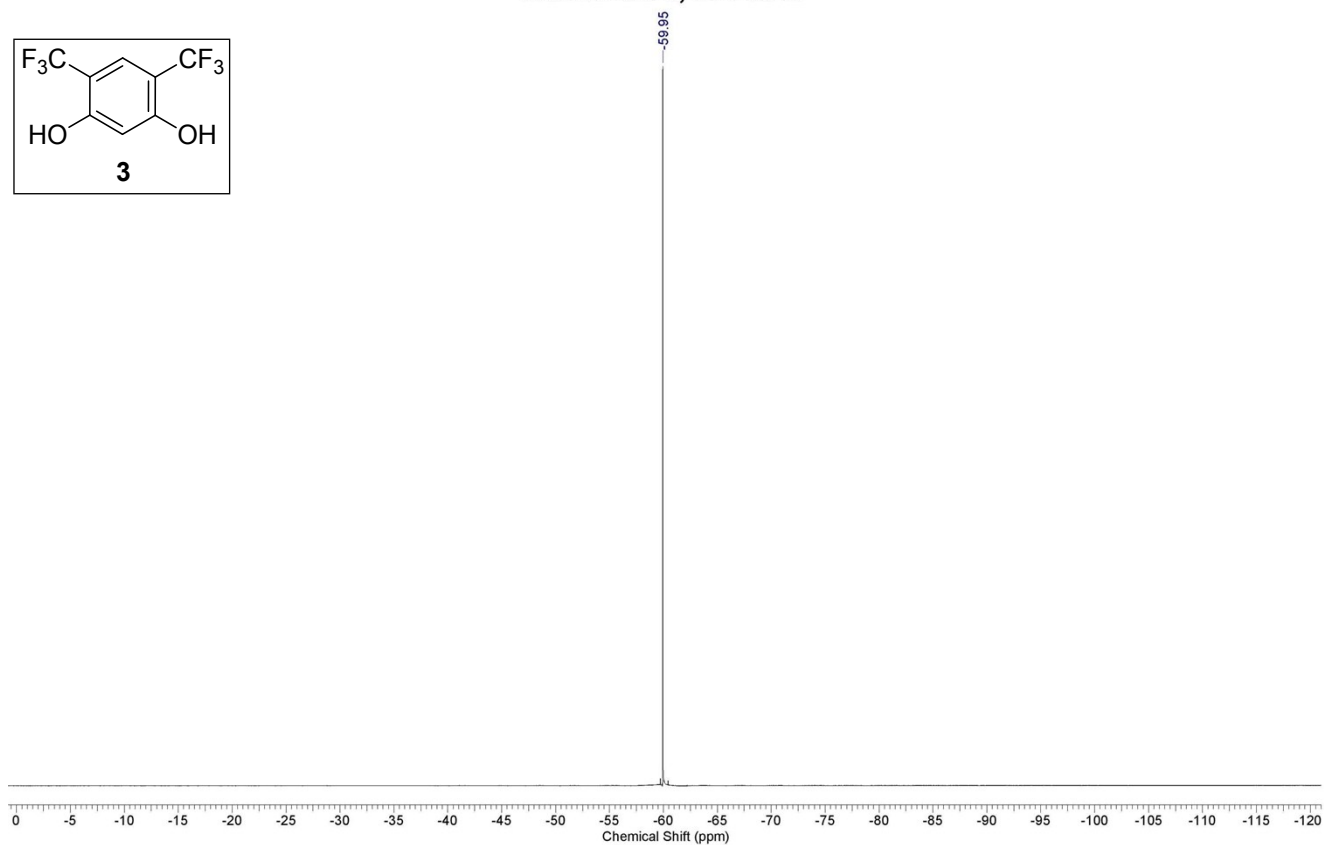


Figure S3-3. ^{13}C NMR spectrum of **4,6-bis(trifluoromethyl)resorcinol (3)**
Chloroform- d ; 100 MHz

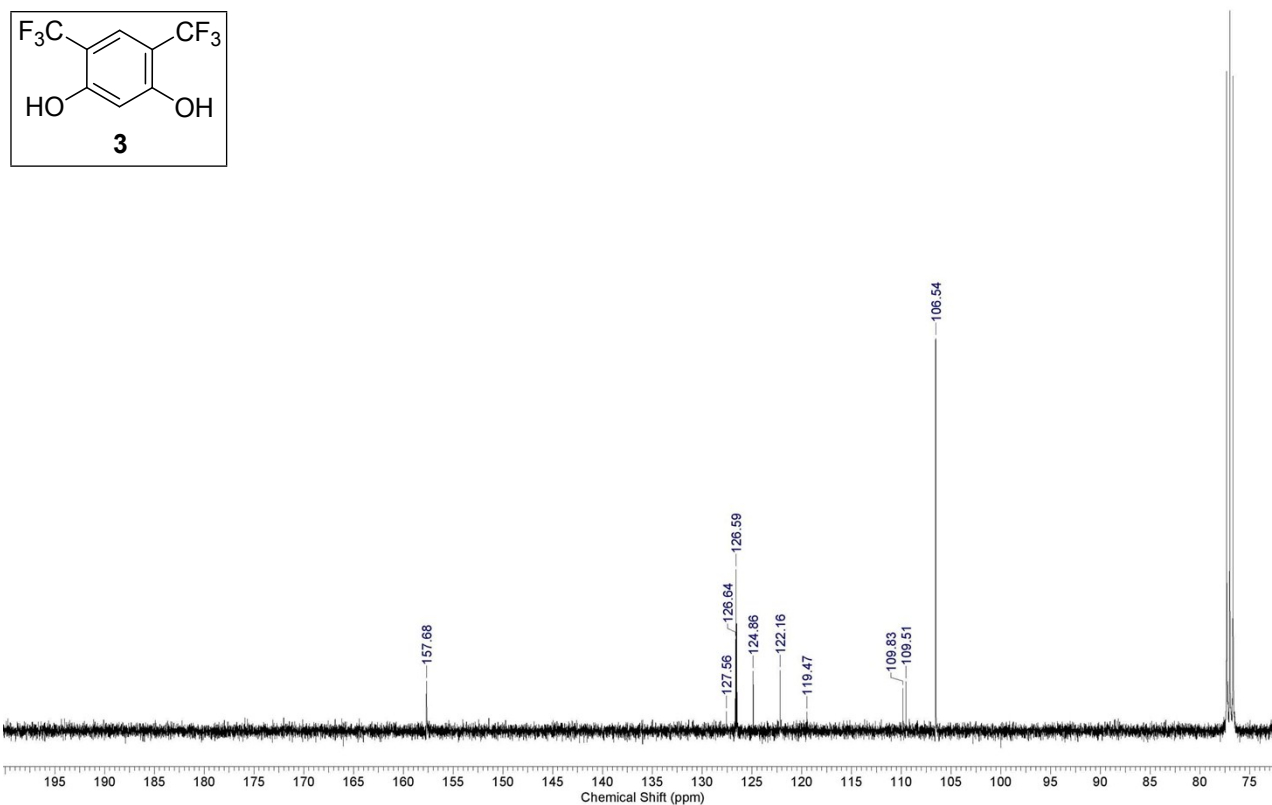


Figure S4-1. ^1H NMR spectrum of **((4,6-bis(trifluoromethyl)-1,3-phenylene)bis(oxy))bis(di-*tert*-butylphosphine) (4)**
Benzene- d_6 ; 400 MHz

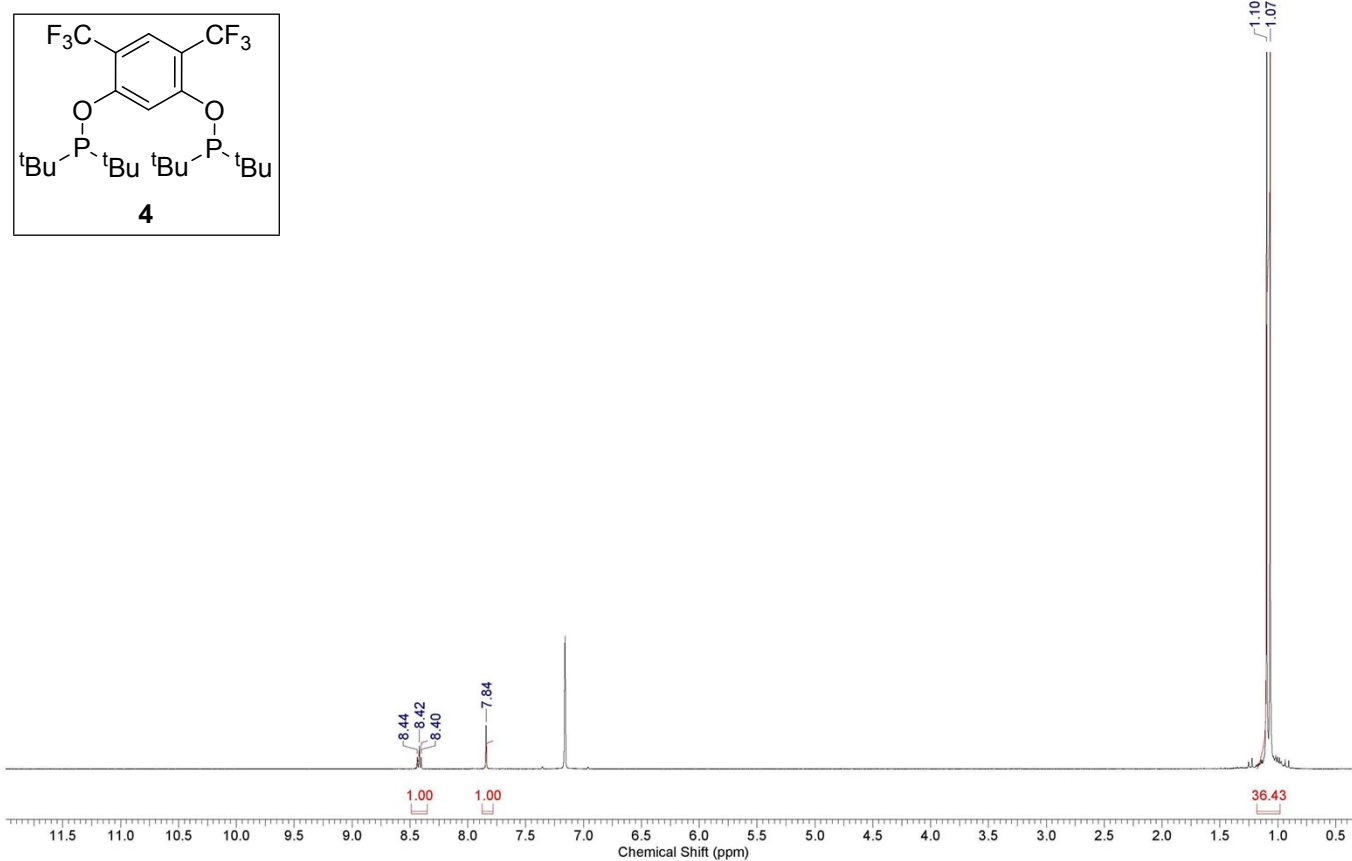


Figure S4-2. ^{19}F NMR spectrum of ((4,6-bis(trifluoromethyl)-1,3-phenylene)bis(oxy))bis(di-*tert*-butylphosphine) (**4**)
Benzene- d_6 ; 376 MHz

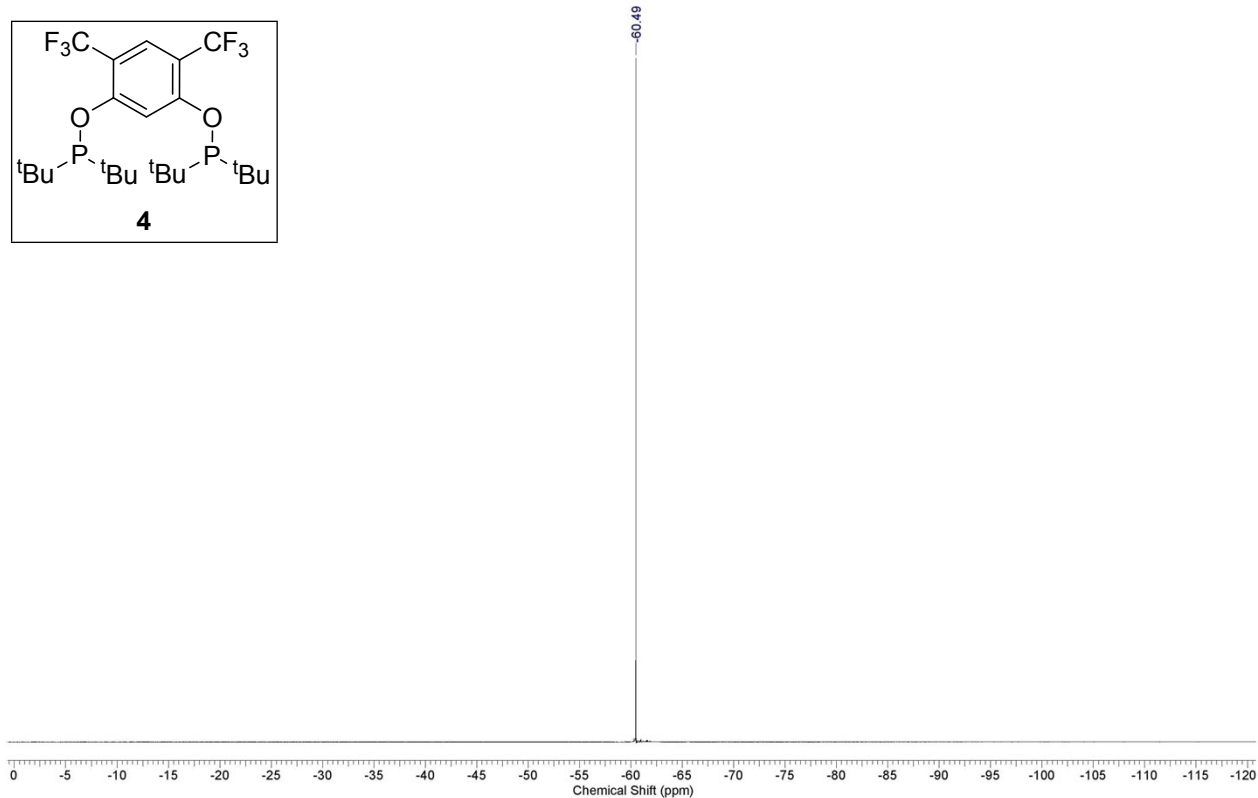


Figure S4-3. ^{31}P NMR spectrum of ((4,6-bis(trifluoromethyl)-1,3-phenylene)bis(oxy))bis(di-*tert*-butylphosphine) (**4**)
Benzene- d_6 ; 162 MHz

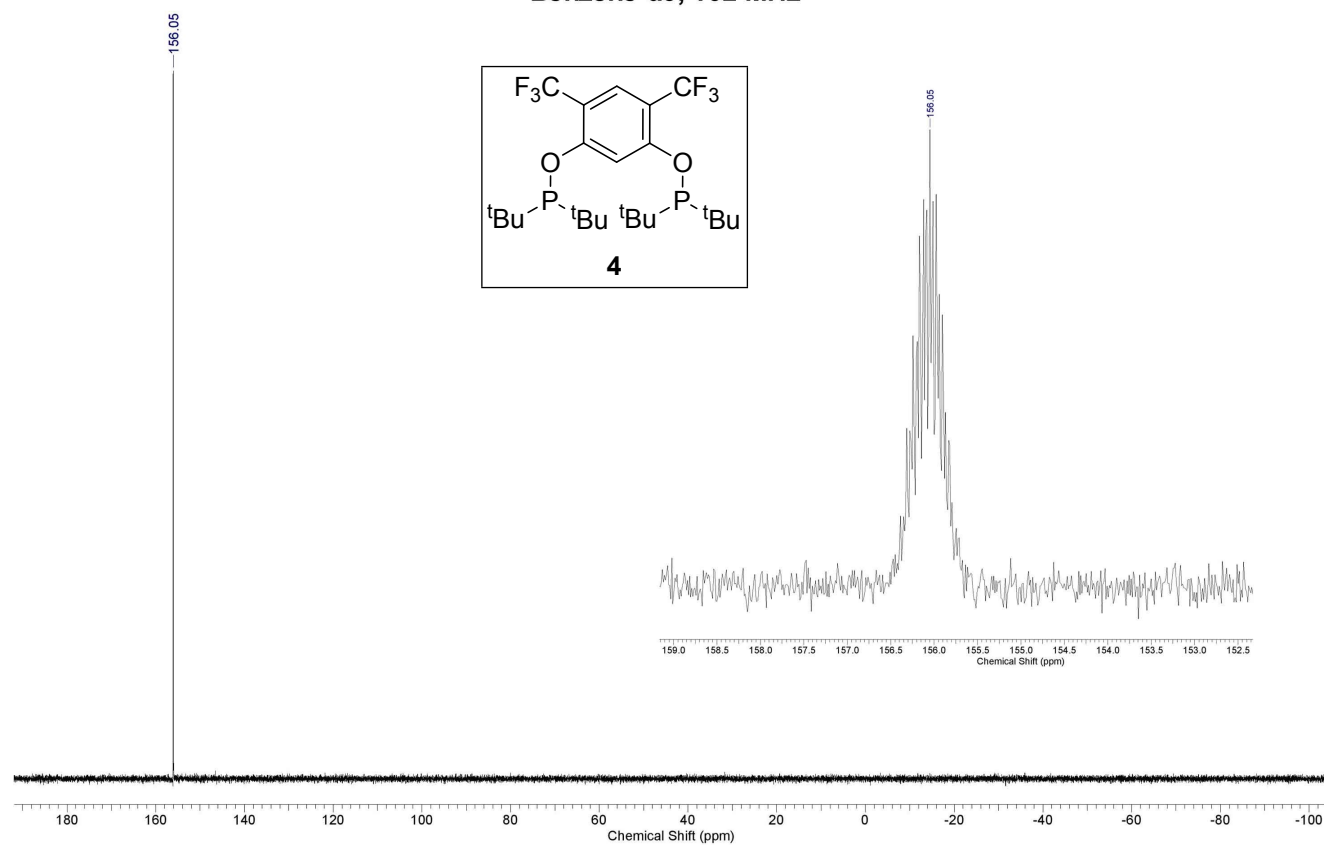


Figure S4-4. ^{13}C NMR spectrum of ((4,6-bis(trifluoromethyl)-1,3-phenylene)bis(oxy))bis(di-*tert*-butylphosphine) (**4**)
Benzene- d_6 ; 100 MHz

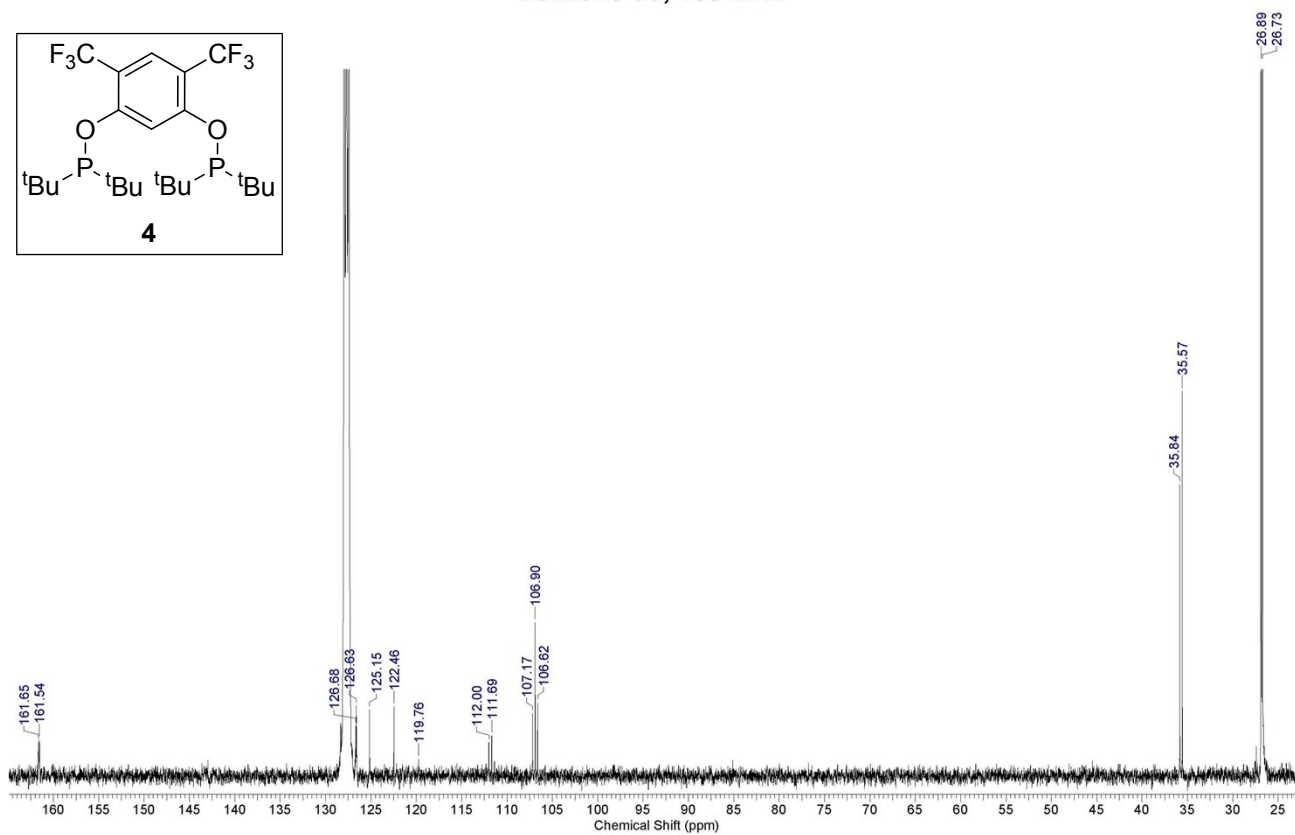


Figure S5-1. ^1H NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrHCl}$ (**5**)
Benzene- d_6 ; 400 MHz

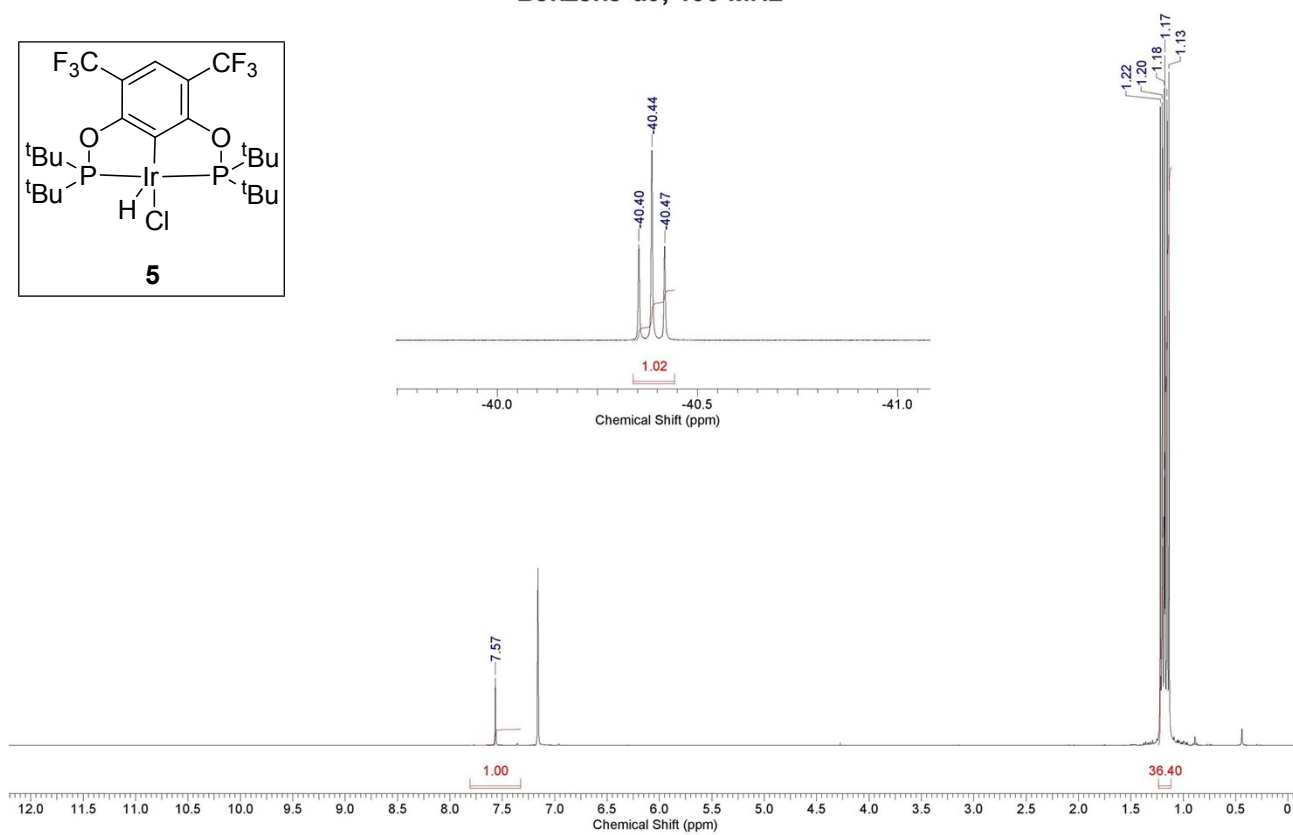


Figure S5-2. ^{19}F NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrHCl}$ (**5**)
Benzene- d_6 ; 376 MHz

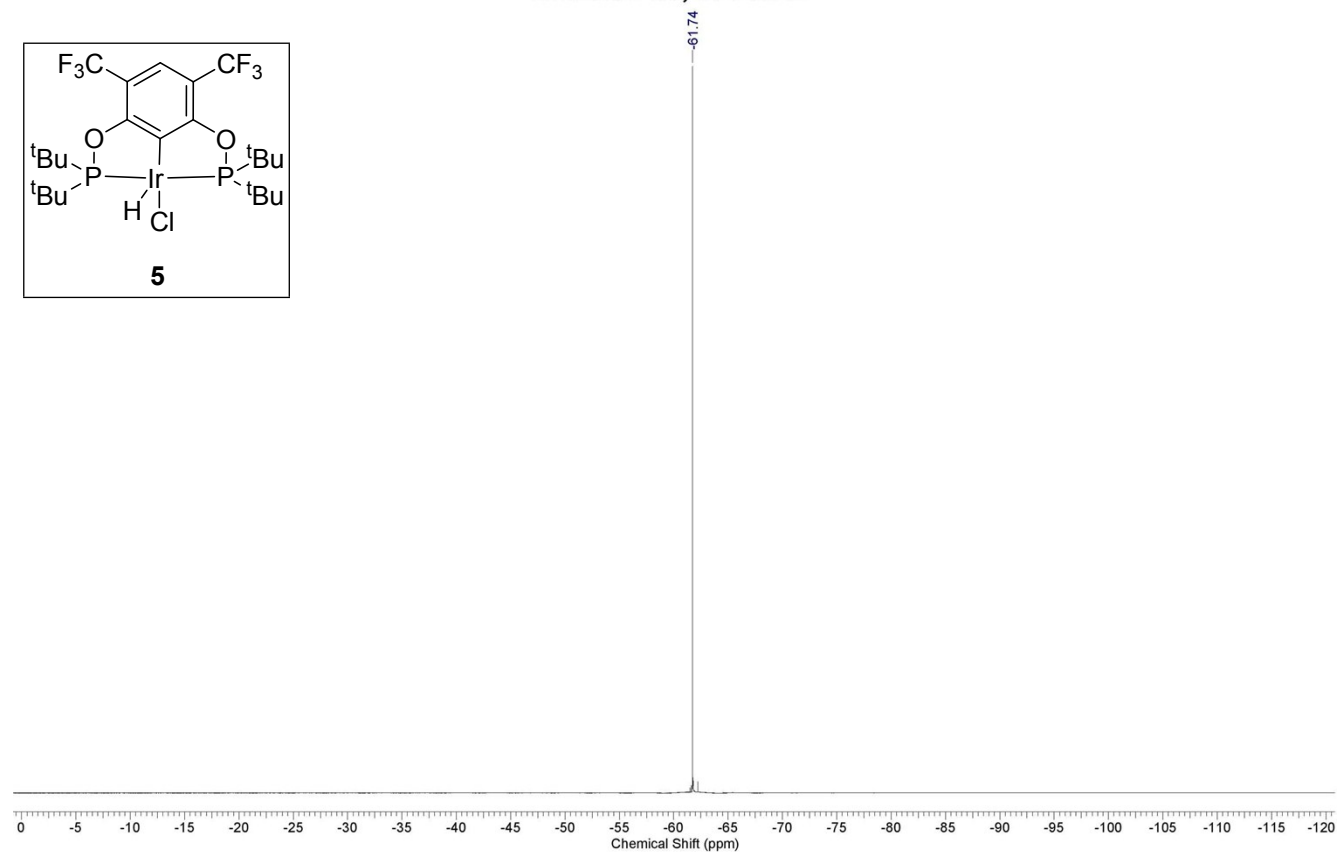


Figure S5-3. ^{31}P NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrHCl}$ (**5**)
Benzene- d_6 ; 162 MHz

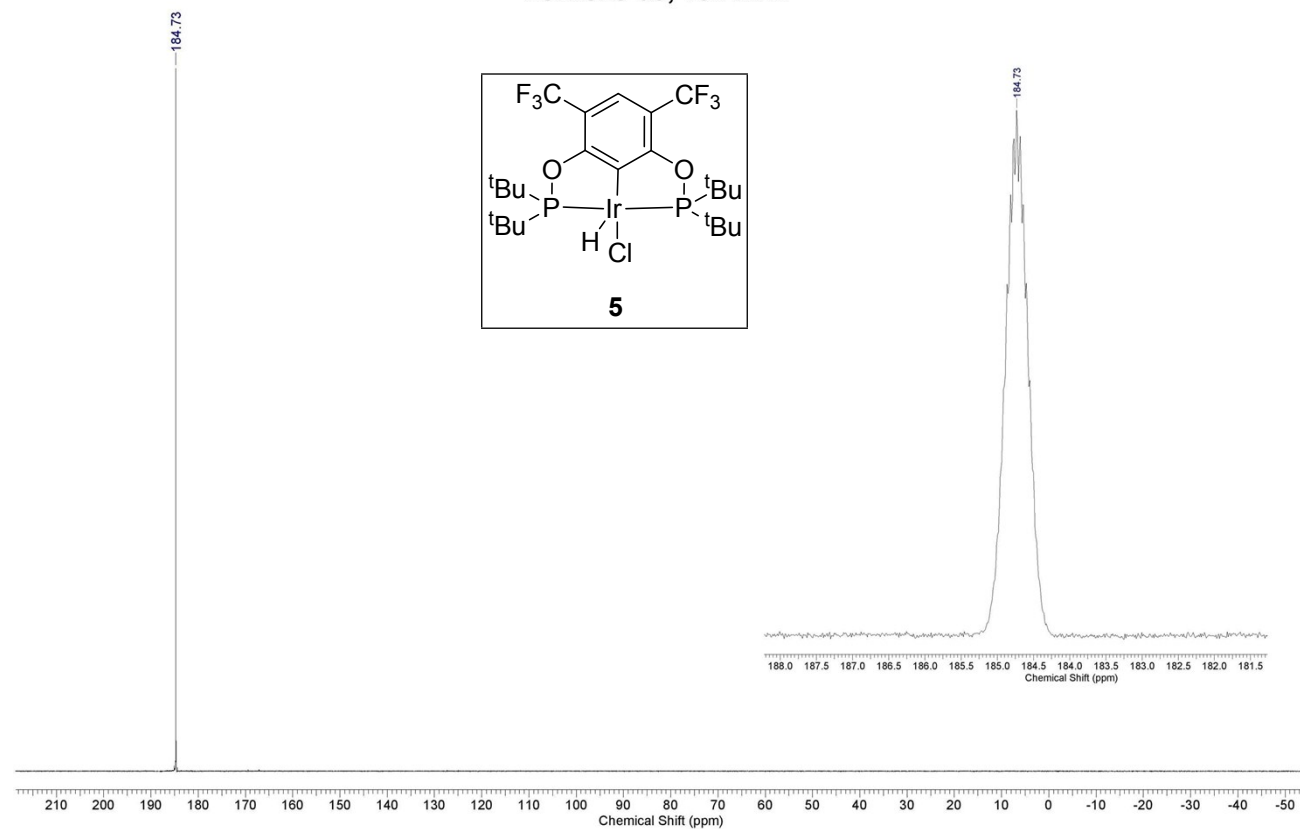


Figure S5-4. ^{13}C NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrHCl}$ (**5**)
Benzene- d_6 ; 100 MHz

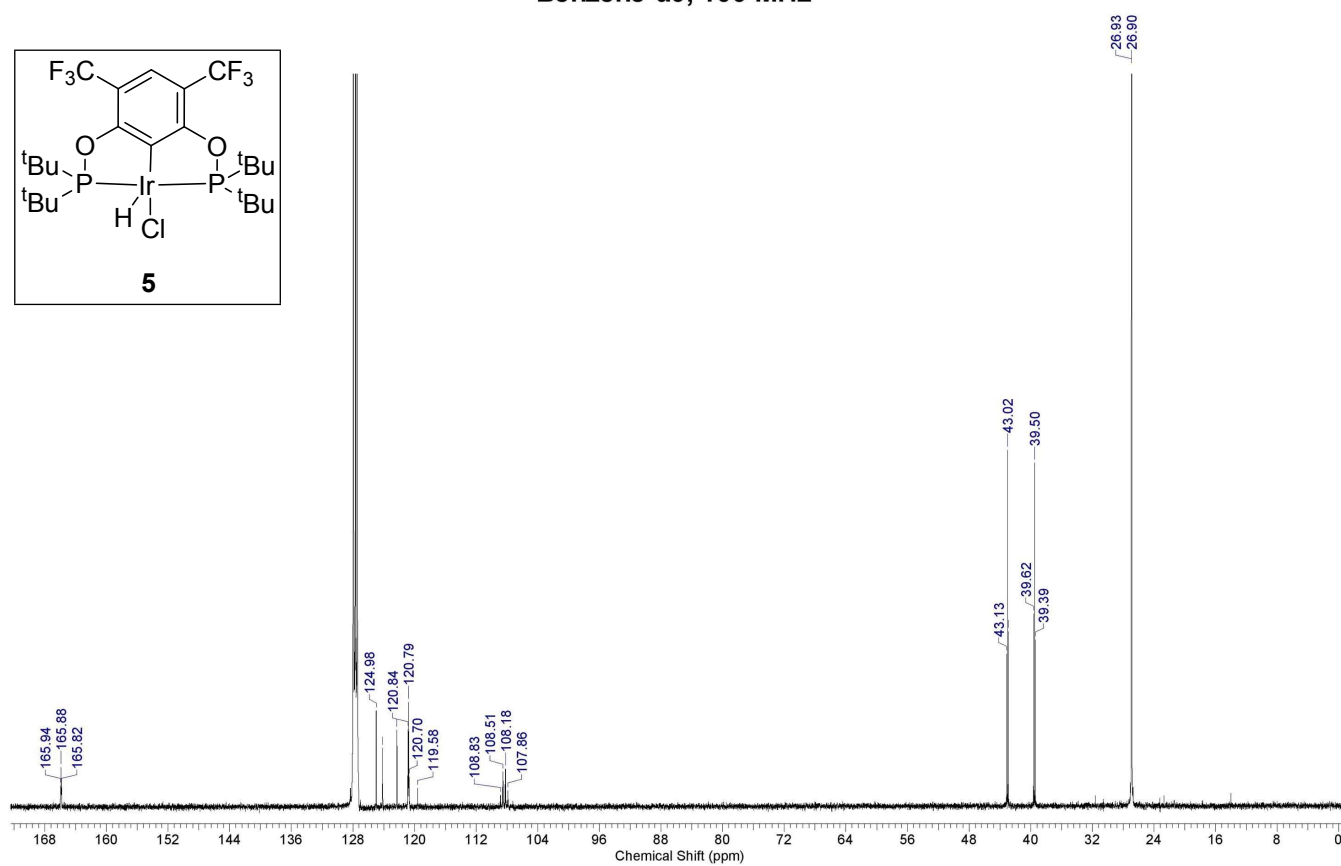


Figure S6-1. ^1H NMR spectrum of $\text{CF}_3(\text{POCOP})\text{Ir}$ (**7**)
Benzene- d_6 ; 400 MHz

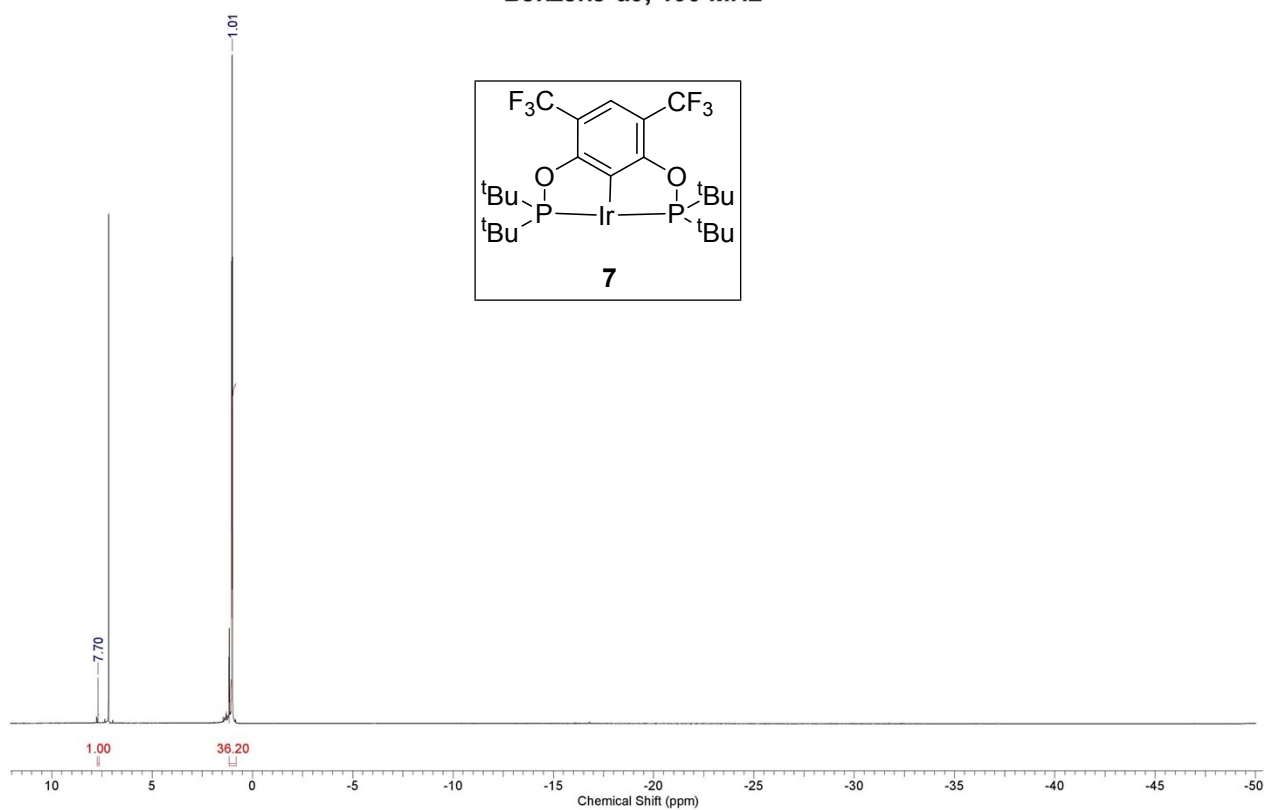


Figure S6-2. ^{19}F NMR spectrum of $\text{CF}_3(\text{POCOP})\text{Ir}$ (**7**)
Benzene- d_6 ; 376 MHz

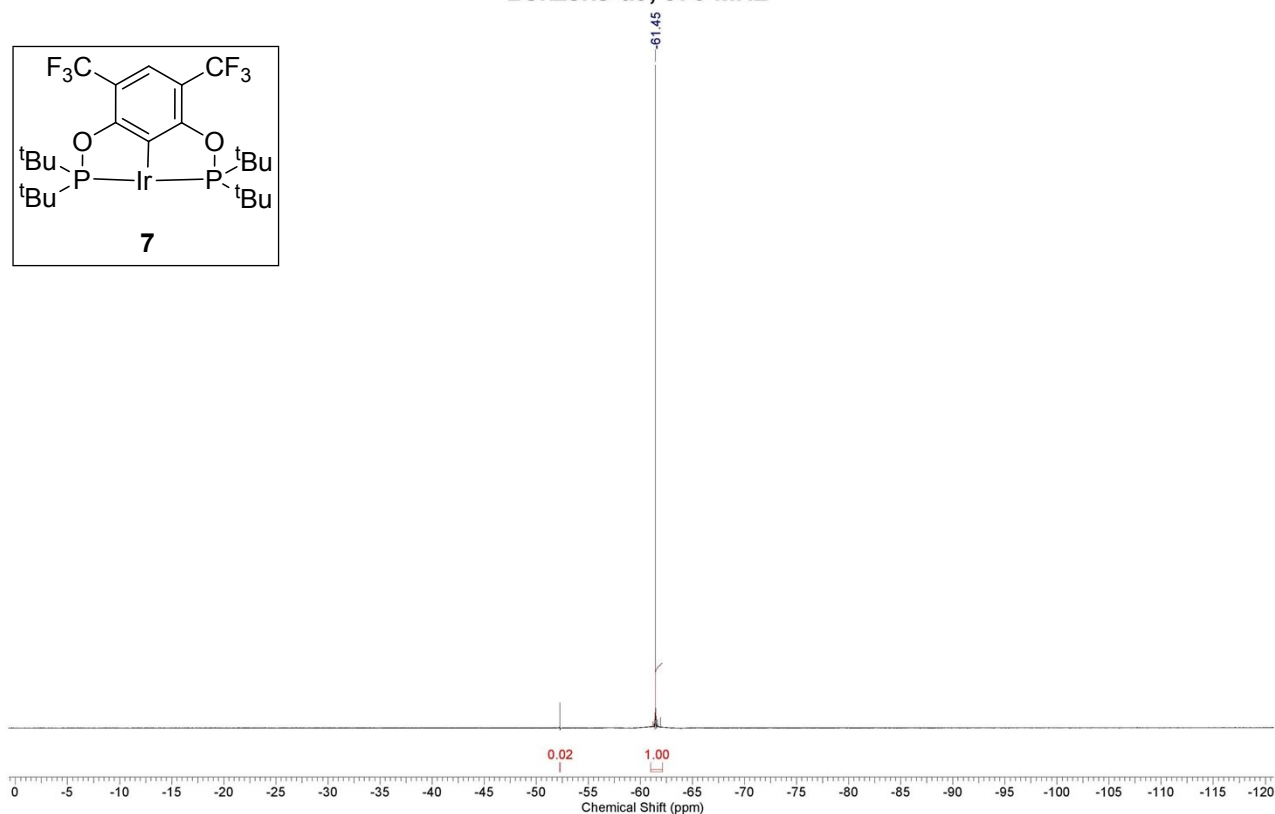


Figure S6-3. ^{31}P NMR spectrum of $\text{CF}_3(\text{POCOP})\text{Ir}$ (**7**)
Benzene- d_6 ; 162 MHz

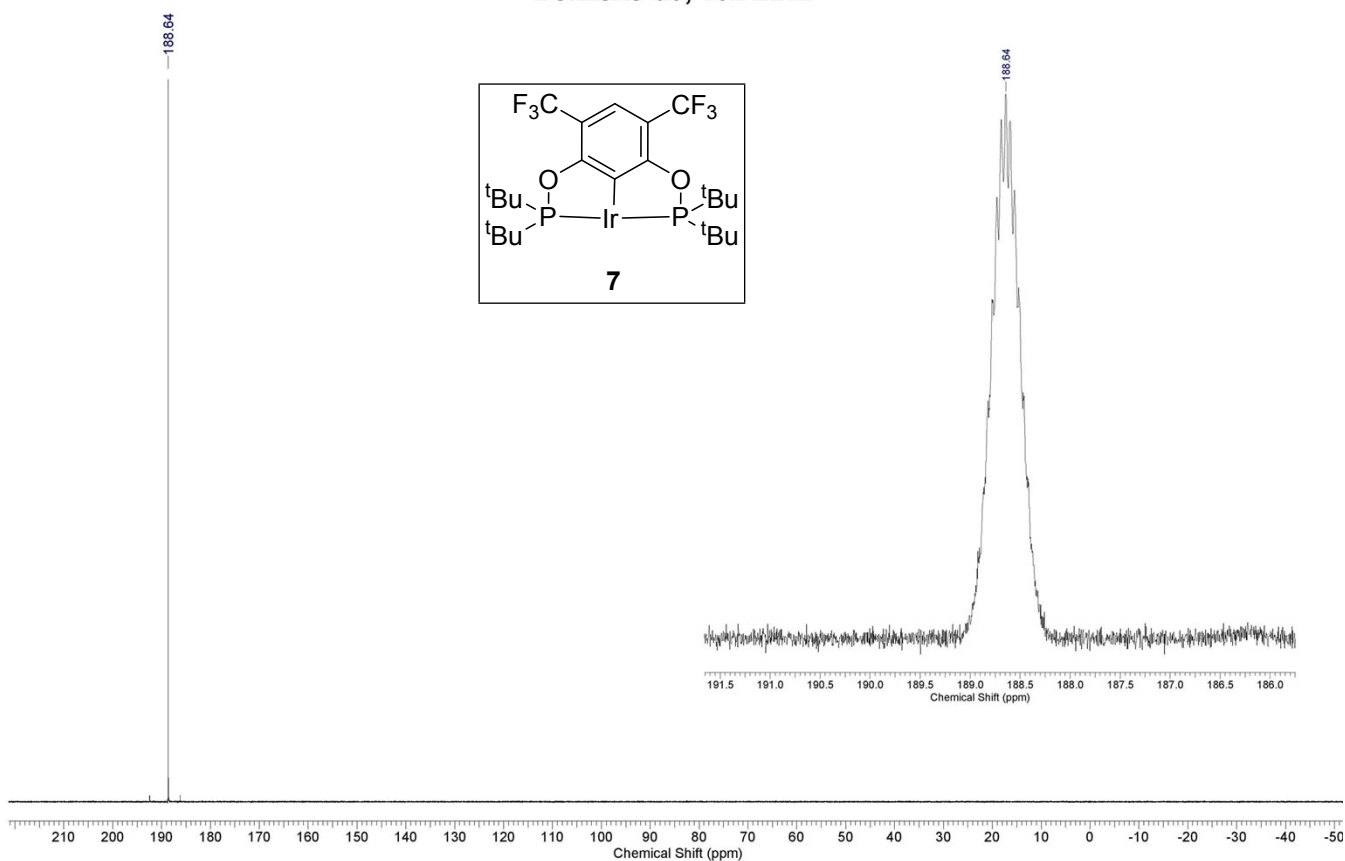


Figure S6-4. ^{13}C NMR spectrum of $\text{CF}_3(\text{POCOP})\text{Ir}$ (**7**)
Benzene- d_6 ; 100 MHz

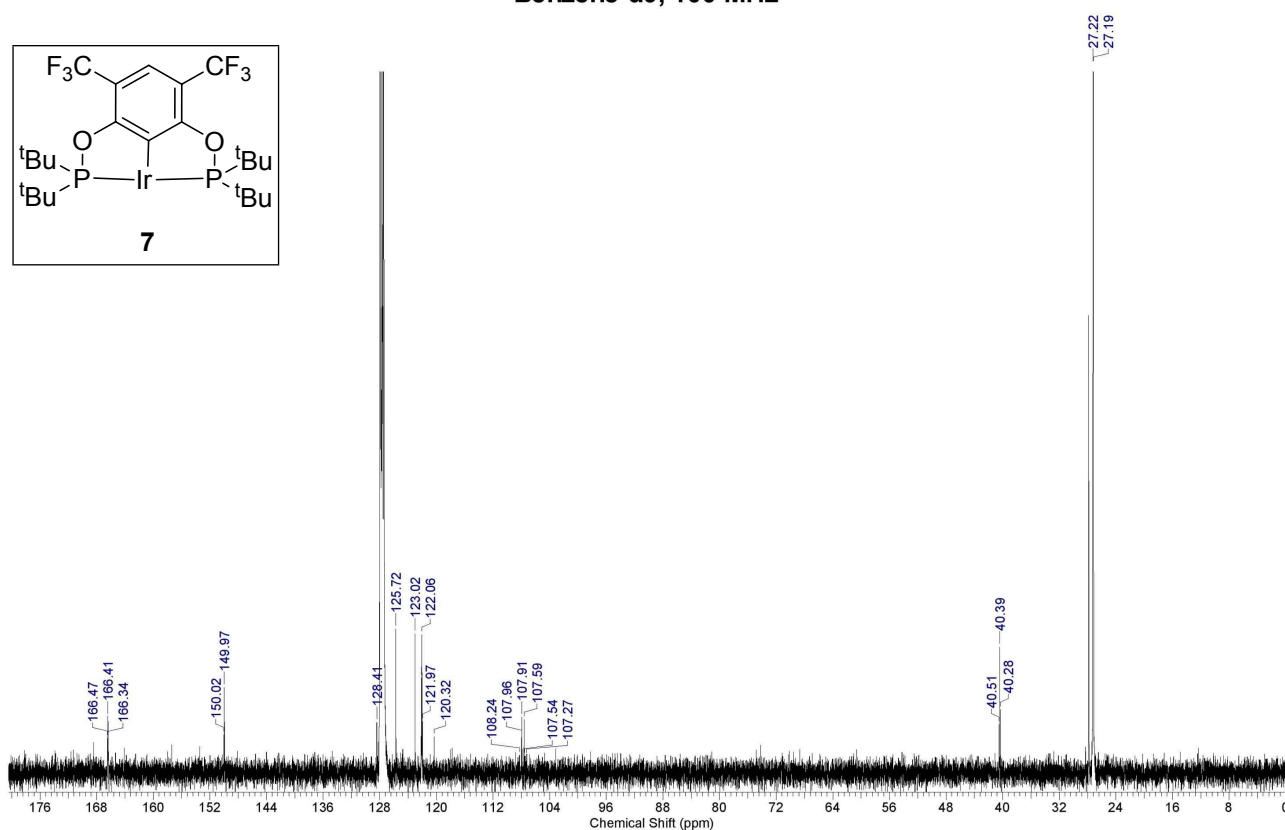


Figure S7-1. ^1H NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) after drying in vacuo
Benzene- d_6 ; 400 MHz

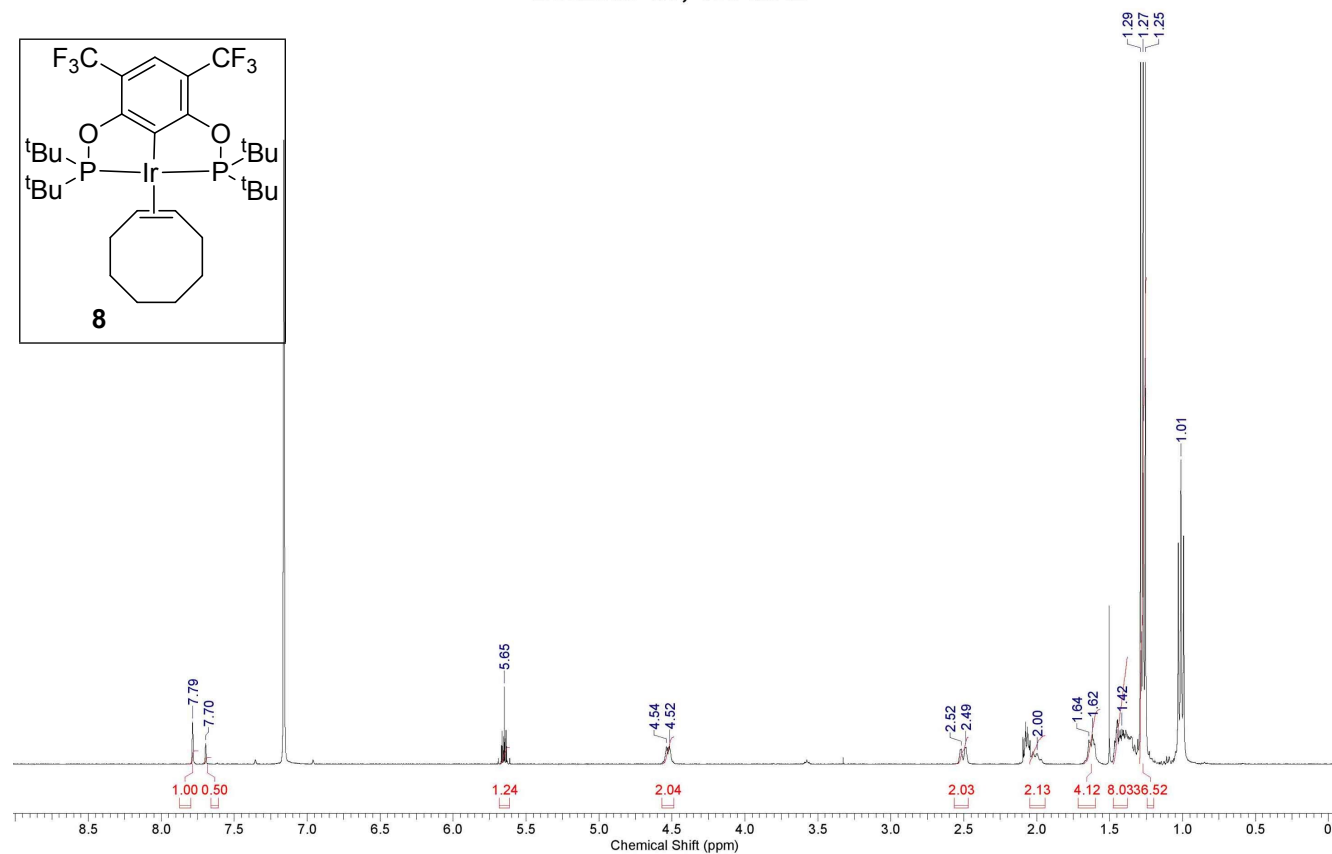


Figure S7-2. ^{19}F NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) with 10-fold excess of COE
Benzene- d_6 ; 376 MHz

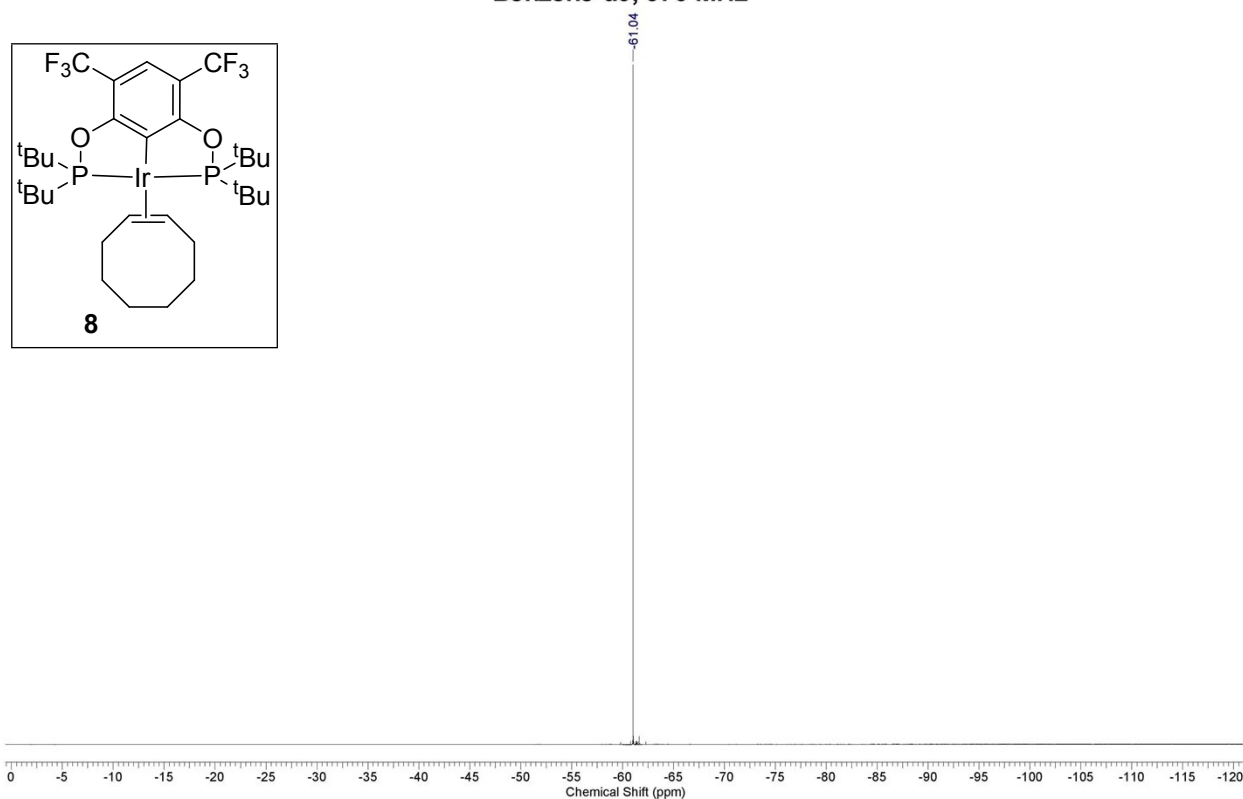


Figure S7-3. ^{19}F NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) after drying in vacuo
Benzene- d_6 ; 376 MHz

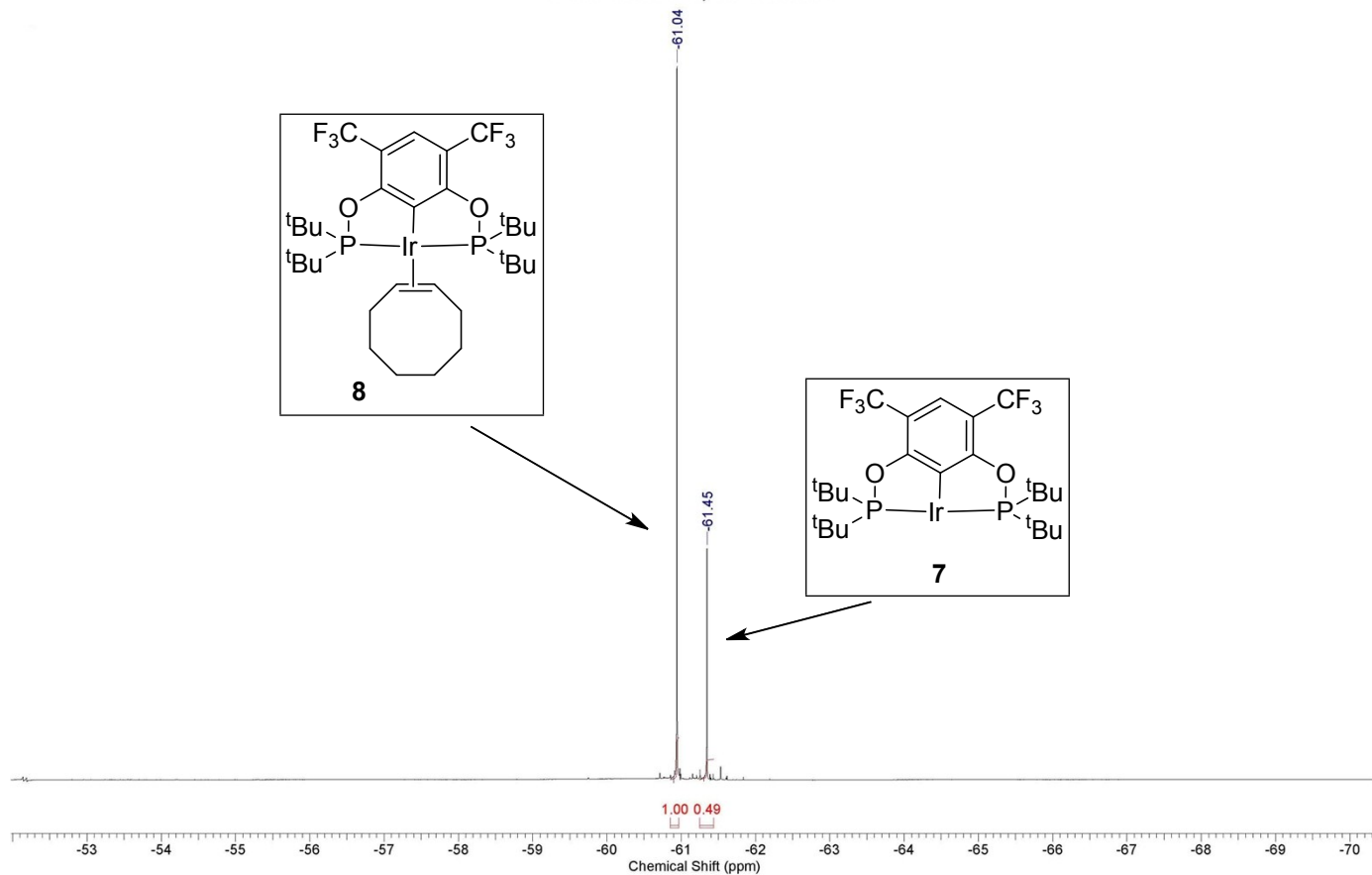


Figure S7-4. ^{31}P NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) with 10-fold excess of COE
Benzene- d_6 ; 162 MHz

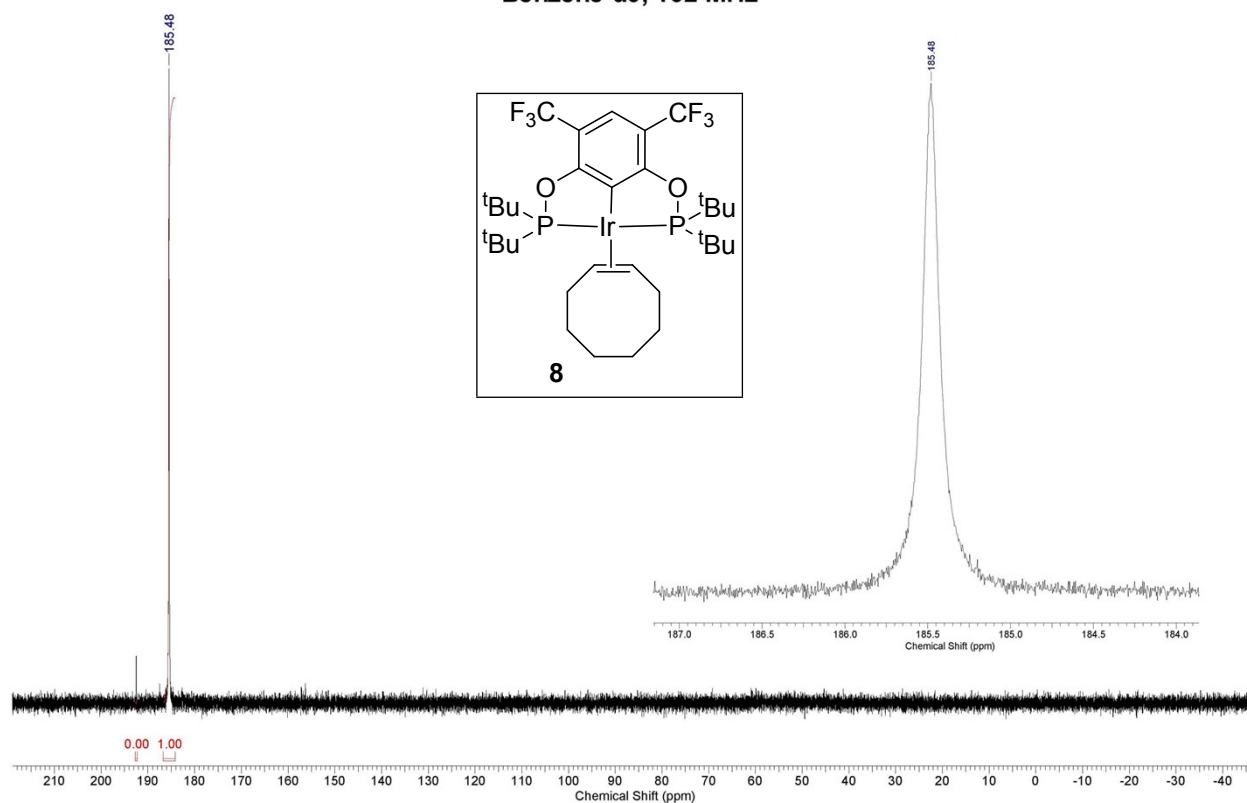


Figure S7-5. ^{31}P NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) after drying in vacuo
Benzene- d_6 ; 162 MHz

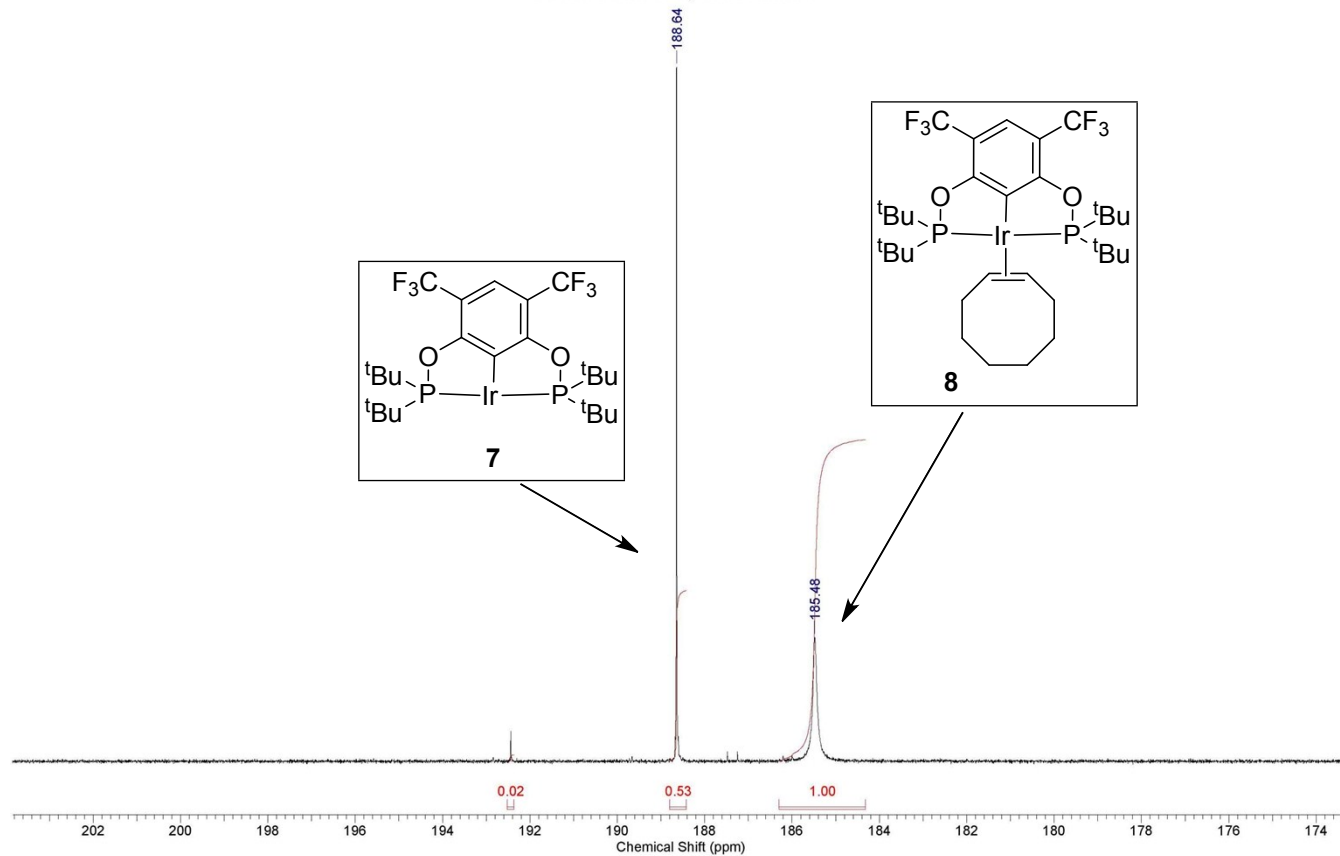


Figure S7-6. ^{13}C NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrCOE}$ (**8**) after drying in vacuo
Benzene- d_6 ; 100 MHz

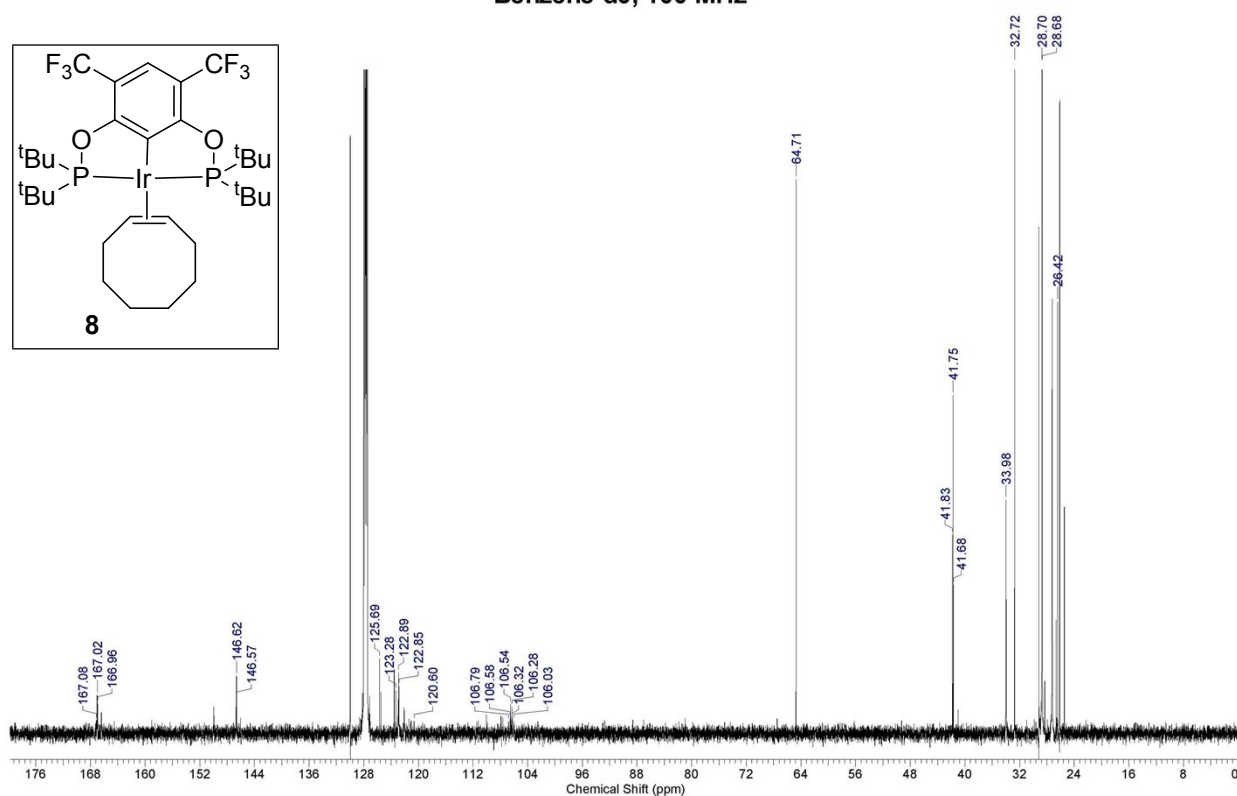


Figure S8-1. ^{19}F NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrTBE}$ (**9**) with 10-fold excess of TBE
Benzene- d_6 ; 376 MHz

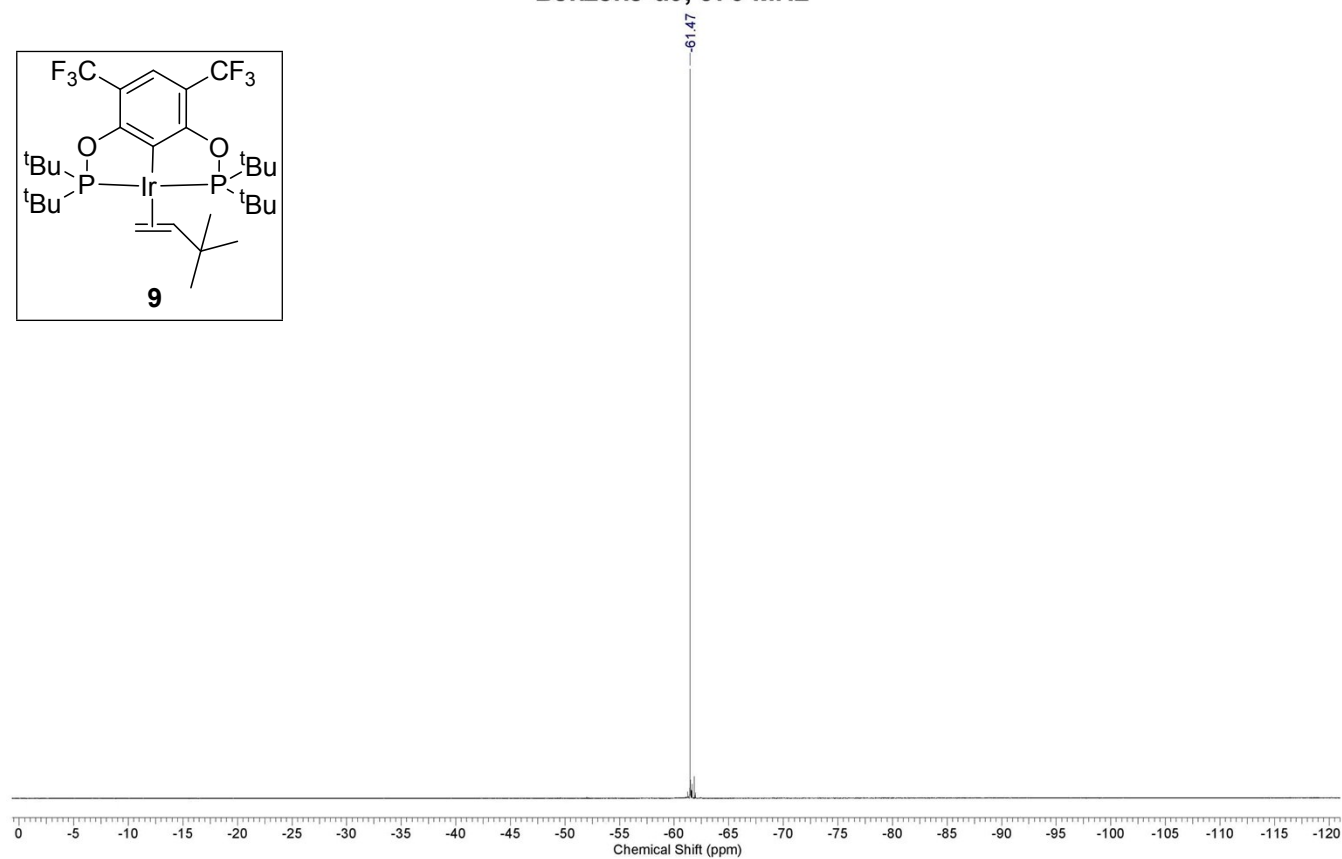


Figure S8-2. ^{31}P NMR spectrum of $\text{CF}_3(\text{POCOP})\text{IrTBE}$ (**9**) with 10-fold excess of TBE
Benzene- d_6 ; 162 MHz

