

Supporting Information for:

Core-Shell $\text{Cd}_{0.2}\text{Zn}_{0.8}\text{S}@\text{BiOX}$ (X = Cl, Br, I) Microspheres: A Family of Hetero-Structured Catalysts with Adjustable Bandgaps, Enhanced Stability and Photocatalytic Performance under Visible Light Irradiation

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Figures caption:

Fig.S1 SEM images of pristine BiOX (a:BiOCl, b: BiOBr, and c: BiOI).

Fig.S2 EDX results of CZS and CZS@BiOX (a: CZS, b: CZS@BiOCl, c: CZS@BiOBr, and d: CZS@BiOI) samples.

Fig.S3 TEM images of BiOX (a: BiOCl, b: BiOBr, and c: BiOI). Scale bar: 500 nm.

Fig.S4 TEM images of 1:1 CZS@BiOX: CZS@BiOCl (a), CZS@BiOBr (b), CZS@BiOI (c); 1:5 CZS@BiOX: CZS@BiOCl (d), CZS@BiOBr (e), CZS@BiOI (f).

Scale bar: 500 nm.

Fig.S5 PL spectra of CZS, BiOX and CZS@BiOX ($X = \text{Cl, Br, I}$).

Fig.S6 Decomposition pathway for RhB under light irradiation.

Fig.S7 Photocatalytic degradation of RhB on blank, CZS and BiOX ($X = \text{Cl, Br, I}$) under visible-light irradiation.

Table S1 Comparison of the morphologies and photocatalytic activities of BiOX ($X = \text{Cl, Br, I}$) between current synthesis method and previous reports.

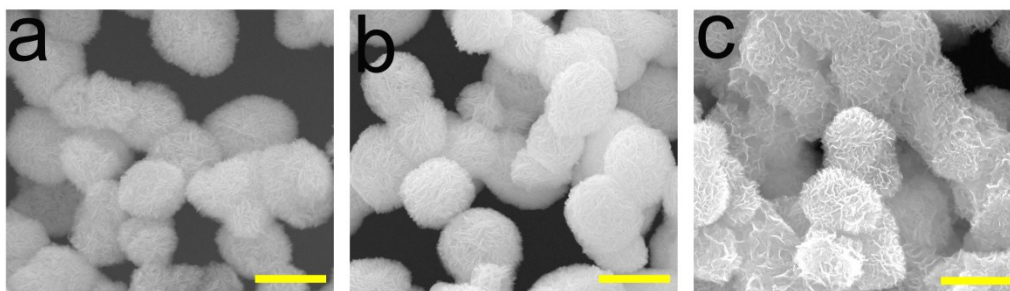


Fig.S1 SEM images of pristine BiOX (a: BiOCl, b: BiOBr, and c: BiOI). Scale bar: 1 μ m.

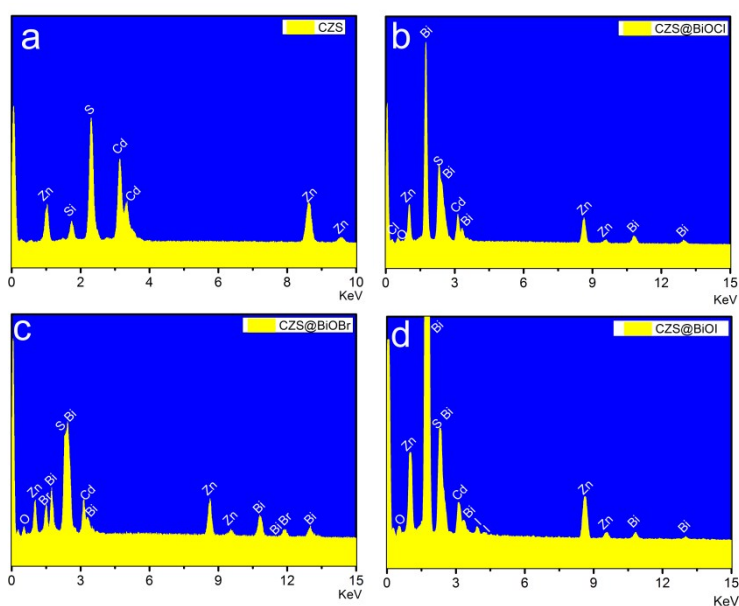


Fig.S2 EDX results of CZS and CZS@BiOX (a: CZS, b: CZS@BiOCl, c: CZS@BiOBr, and d: CZS@BiOI) samples.

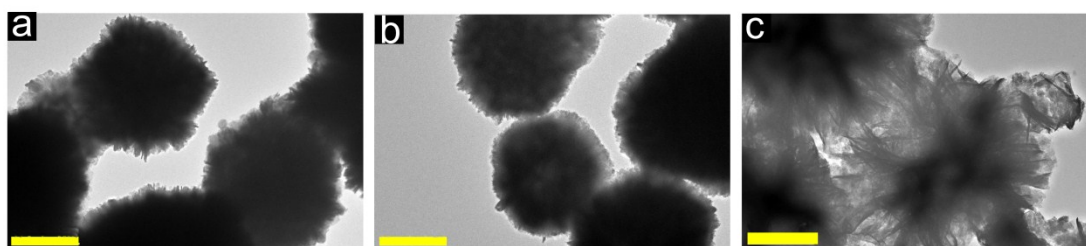


Fig.S3 TEM images of BiOX (a: BiOCl, b: BiOBr, and c: BiOI). Scale bar: 500 nm.

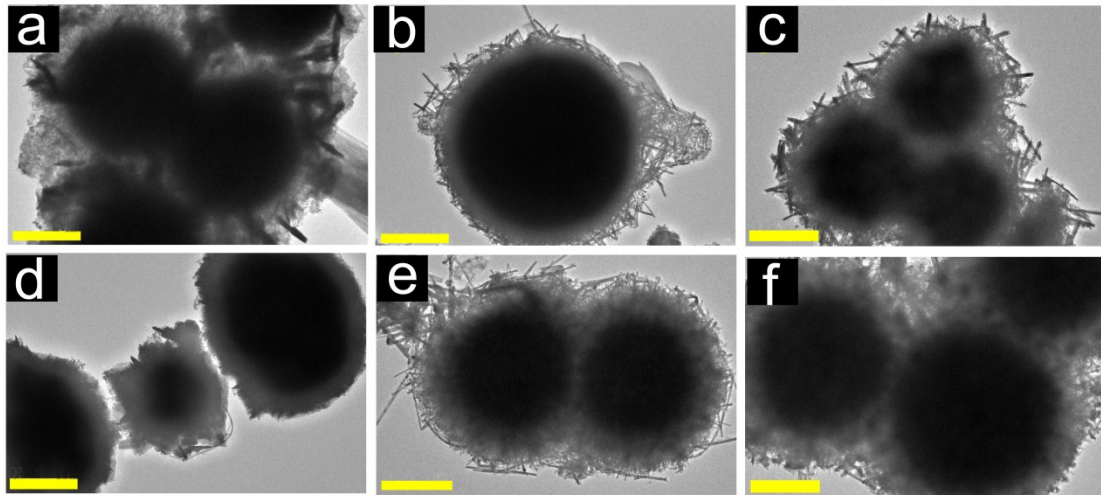


Fig.S4 TEM images of 1:1 CZS@BiOX: CZS@BiOCl (a), CZS@BiOBr (b), CZS@BiOI (c); 1:5 CZS@BiOX: CZS@BiOCl (d), CZS@BiOBr (e), CZS@BiOI (f). Scale bar: 500 nm.

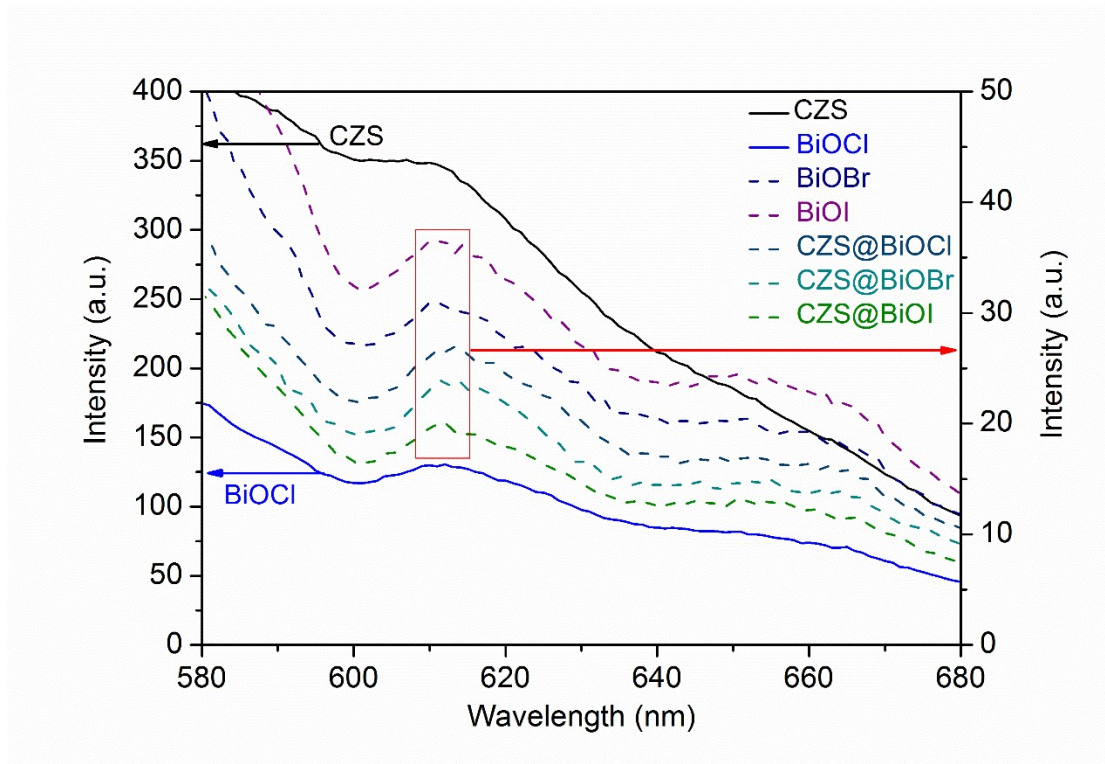


Fig.S5 PL spectra of CZS, BiOX and CZS@BiOX ($X = \text{Cl}, \text{Br}, \text{I}$).

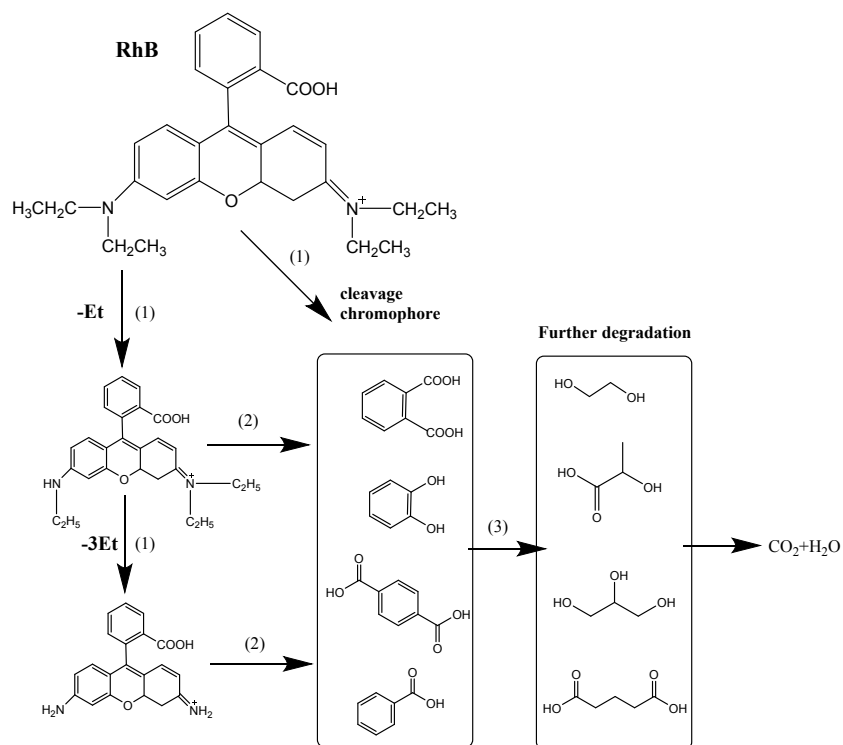


Fig.S6 Decomposition pathway for RhB under light irradiation.

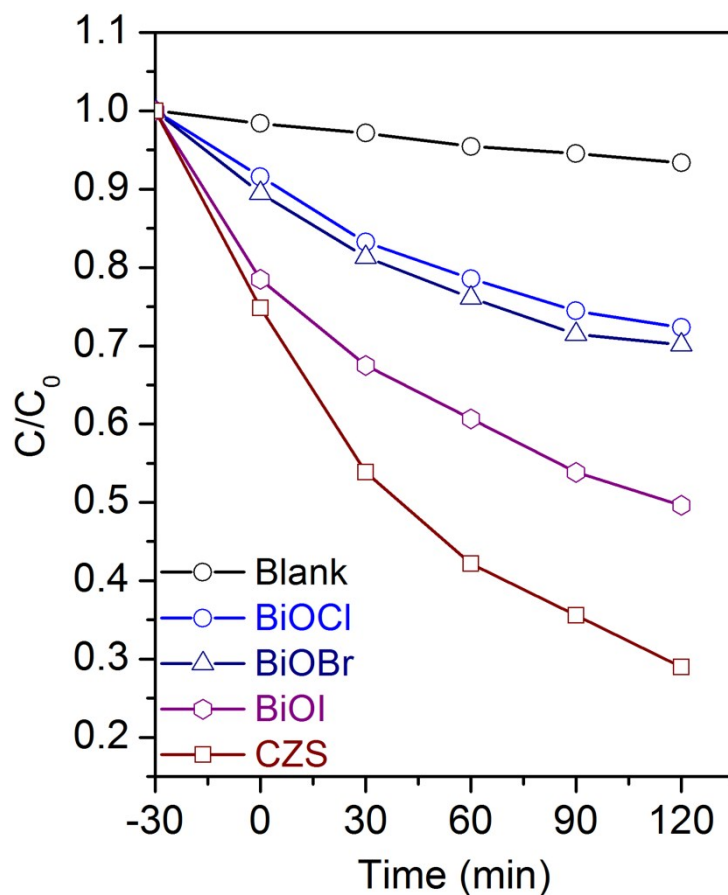


Fig.S7 Photocatalytic degradation of RhB on blank, CZS and BiOX ($X = \text{Cl}, \text{Br}, \text{I}$) under visible-light irradiation.

Table S1 Comparison of the morphologies and photocatalytic activities of BiOX ($X = \text{Cl}, \text{Br}, \text{I}$) between current synthesis method and previous reports.

| Synthesis method | Morphology | S_{BET} (m^2/g) | Catalytic activity | Ref |
|--|-------------------|--|--------------------|-----------|
| Precipitation; Room temperature. | Plates | BiOCl:12.1 BiOBr: 14.3 BiOI: 13.4 | BiOCl<BiOI<BiOBr | 1 |
| Solvothermal; Ethylene glycol; 150 °C. | Plates | BiOI: 6.72 BiOCl:15.99 | BiOI<BiOCl | 2 |
| Solvothermal; Ethylene glycol; 160 °C. | Irregular spheres | BiOBr:12.03 BiOI:56.29 | BiOBr<BiOI | 3 |
| Solvothermal; Ethylene glycol; 160 °C. | Porous spheres | BiOCl:30.35, BiOBr:43.82 BiOI:70.15 | BiOCl<BiOBr<BiOI | This work |

Reference:

1. L. Chen, R. Huang, M. Xiong, Q. Yuan, J. He, J. Jia, M. Yao, S. Luo, C. Au and S. Yin, *Inorg. Chem.*, 2013, 52, 11118–11125.
2. T. Li, G. Chen, C. Zhou, Z. Shen, R. Jin and J. Sun, *Dalton Trans.*, 2011, 40, 6751–6758.
3. Z. Jia, F. Wang, F. Xin and B. Zhang, *Ind. Eng. Chem. Res.*, 2011, 50, 6688–6694.