

An Aminopyrimidine-Functionalized Cage-Based Metal-Organic Framework Exhibiting Highly Selective Adsorption of C₂H₂ and CO₂ over CH₄

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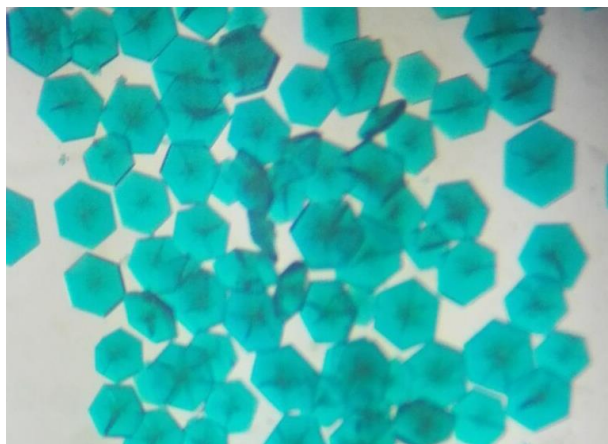


Fig. S1 Photograph of the as-synthesized crystals of **ZJNU-54**.

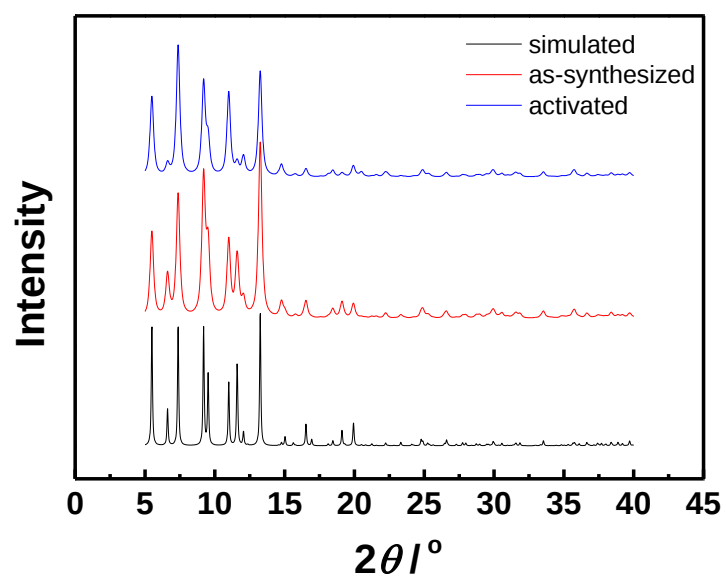


Fig. S2 PXRD patterns.

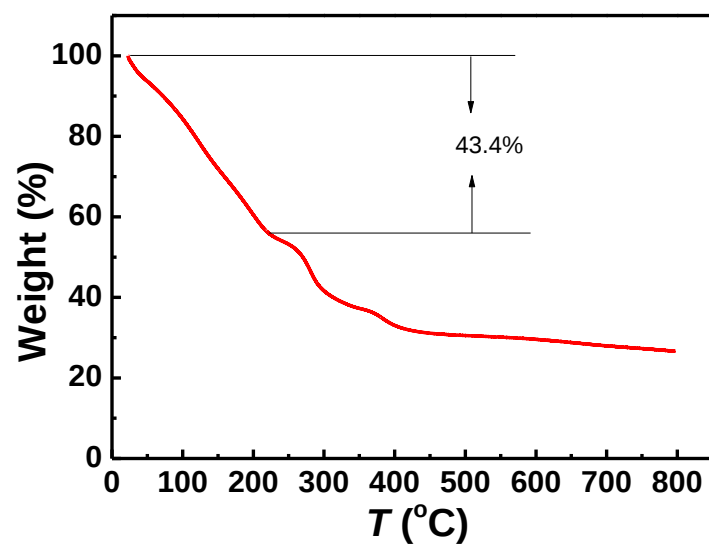


Fig. S3 TGA curve of as-synthesized **ZJNU-54** under a nitrogen atmosphere with a heating rate of 5 K min⁻¹.

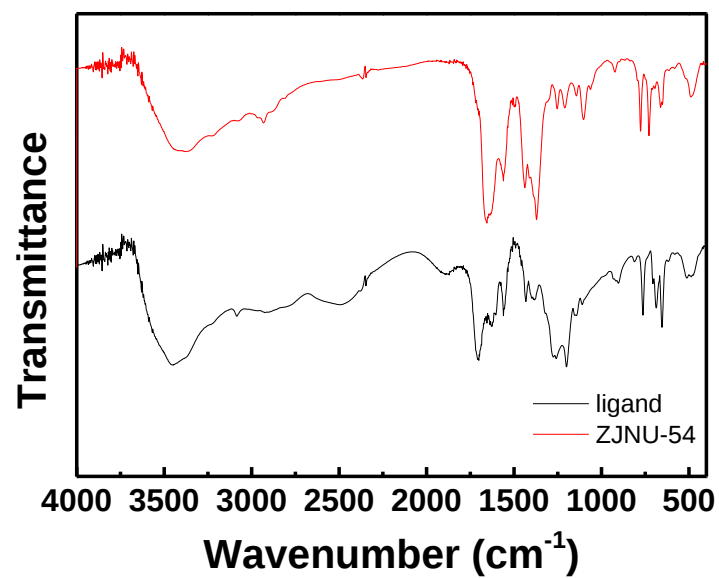


Fig. S4 FTIR spectra of the organic ligand (black) and as-synthesized **ZJNU-54** (red).

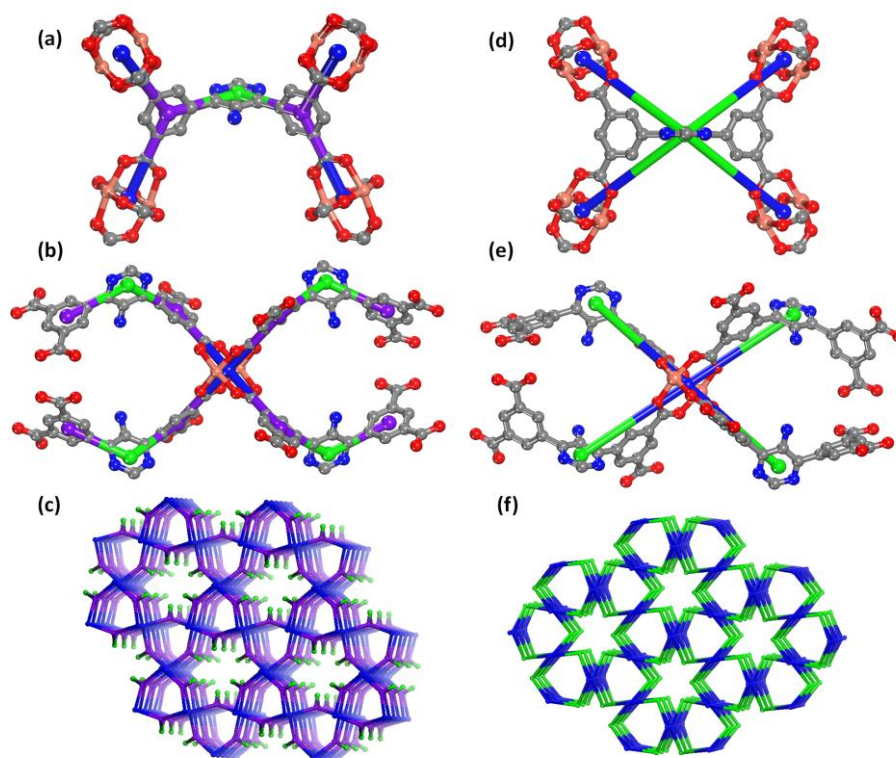
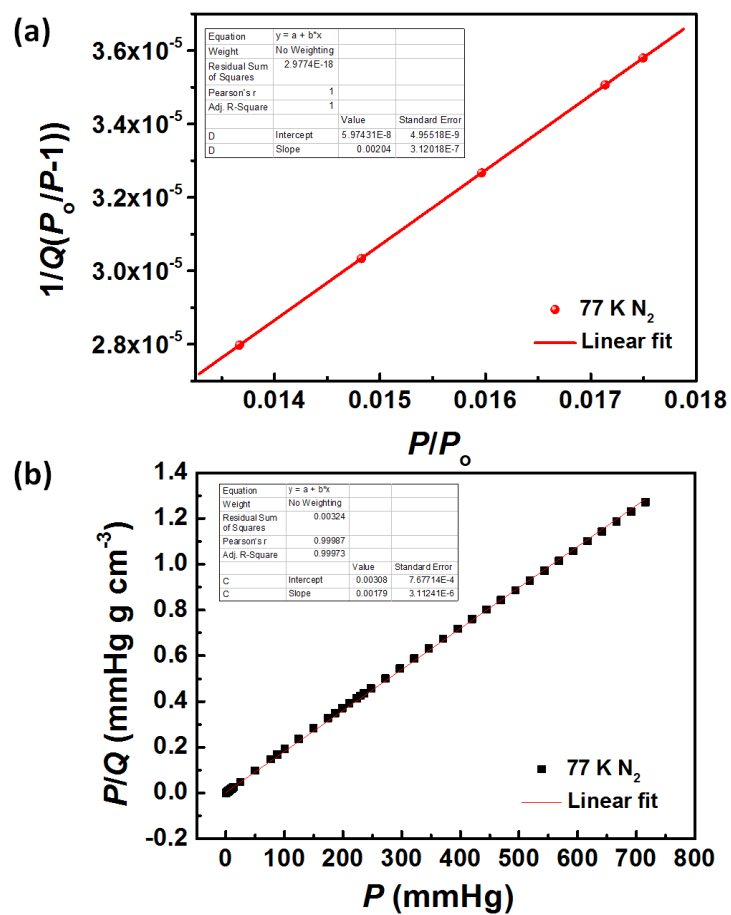


Fig. S5 Topological analysis based on two different simplifications of the organic ligands. If the organic ligand is considered as having a pair of 3-coordinated branch points (a), and the dicopper paddlewheel is regarded as 4-connected node (b), the overall network is a (3,4)-c binodal network with Schläfli symbol of $(6^2 \cdot 10^4)(6^3)_2$ (c). If the organic ligand and the dicopper paddlewheel are taken as 4-connected nodes (d, e), the overall network is a 4-connected 2-binodal network with the Schläfli symbol of $(4^2 \cdot 6^4)(4^2 \cdot 8^4)$ (f).



$$S_{\text{BET}} = 1/(5.97431 \times 10^{-8} + 0.00204)/22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 2134 \text{ m}^2 \text{ g}^{-1}$$

$$S_{\text{Langmuir}} = (1/0.00179)/22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 2432 \text{ m}^2 \text{ g}^{-1}$$

Fig. S6 BET (a) and Langmuir (b) plots for **ZJNU-54a**.

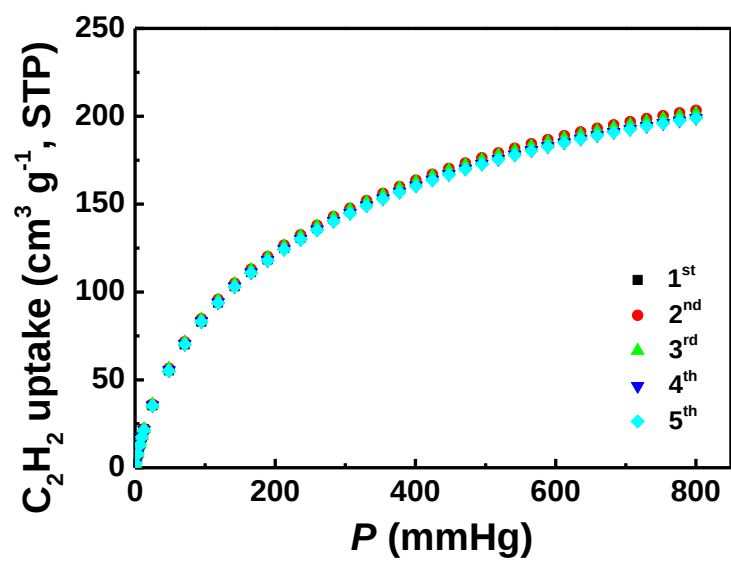


Fig. S7 Five cycles of C_2H_2 adsorption at 298 K in **ZJNU-54a**. Note that the reactivation process was applied between each cycle, which was achieved by heating the sample at 333 K until the degassed rate reached $3 \mu\text{mHg min}^{-1}$.

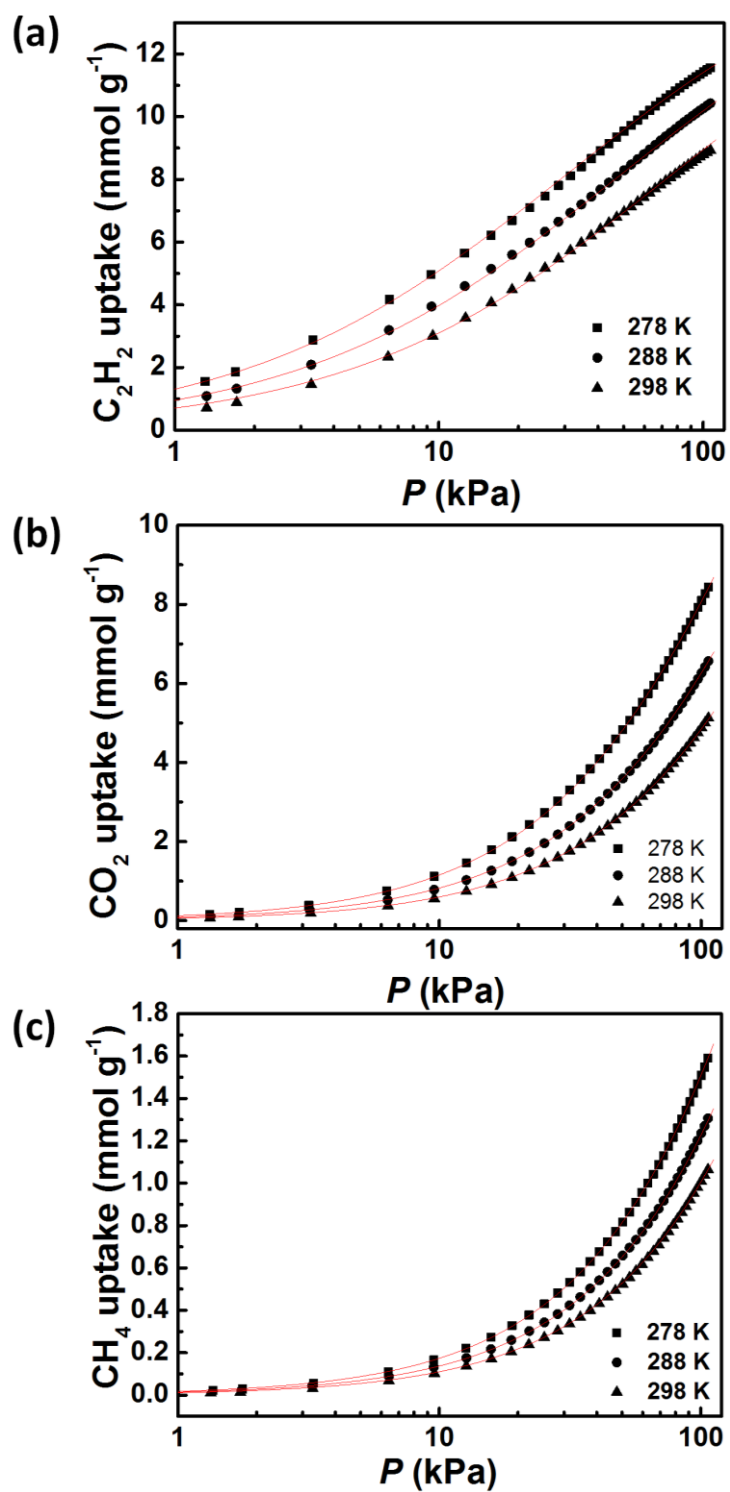


Fig. S8 Comparison of the pure-component isotherm data for (a) C_2H_2 , (b) CO_2 , and (c) CH_4 in ZJNU-54a with the fitted isotherms (shown by continuous solid lines) at 278 K, 288 K and 298 K.

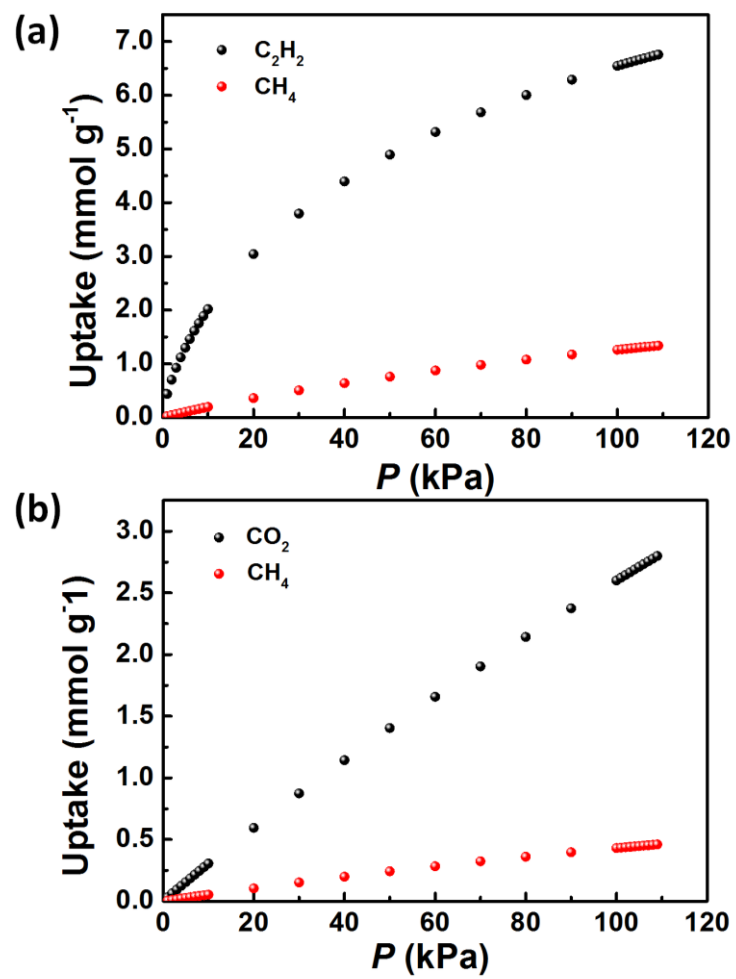


Fig. S9 IAST calculations of mixture adsorption isotherms of **ZJNU-54a** for (a) 50/50 C₂H₂/CH₄ and (b) 50/50 CO₂/CH₄ gas mixtures at 298 K.

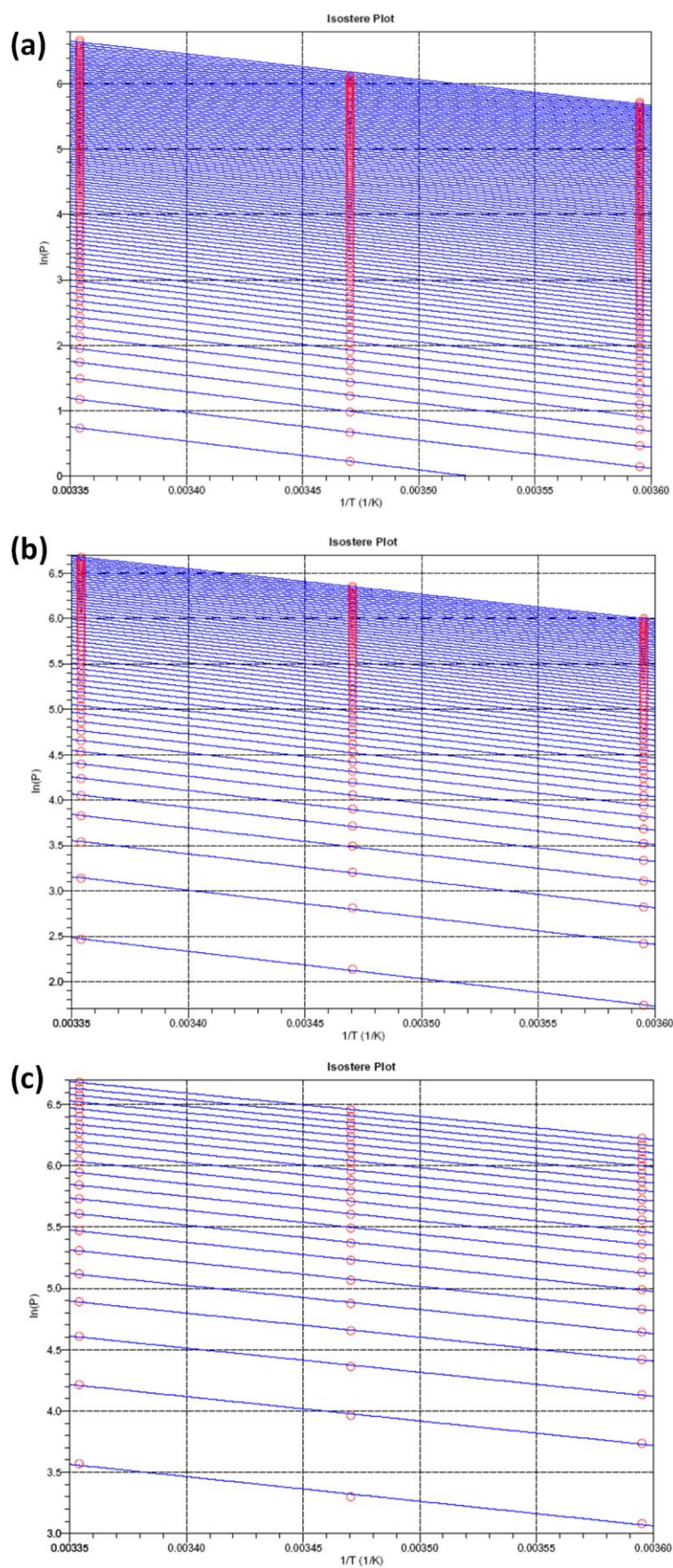
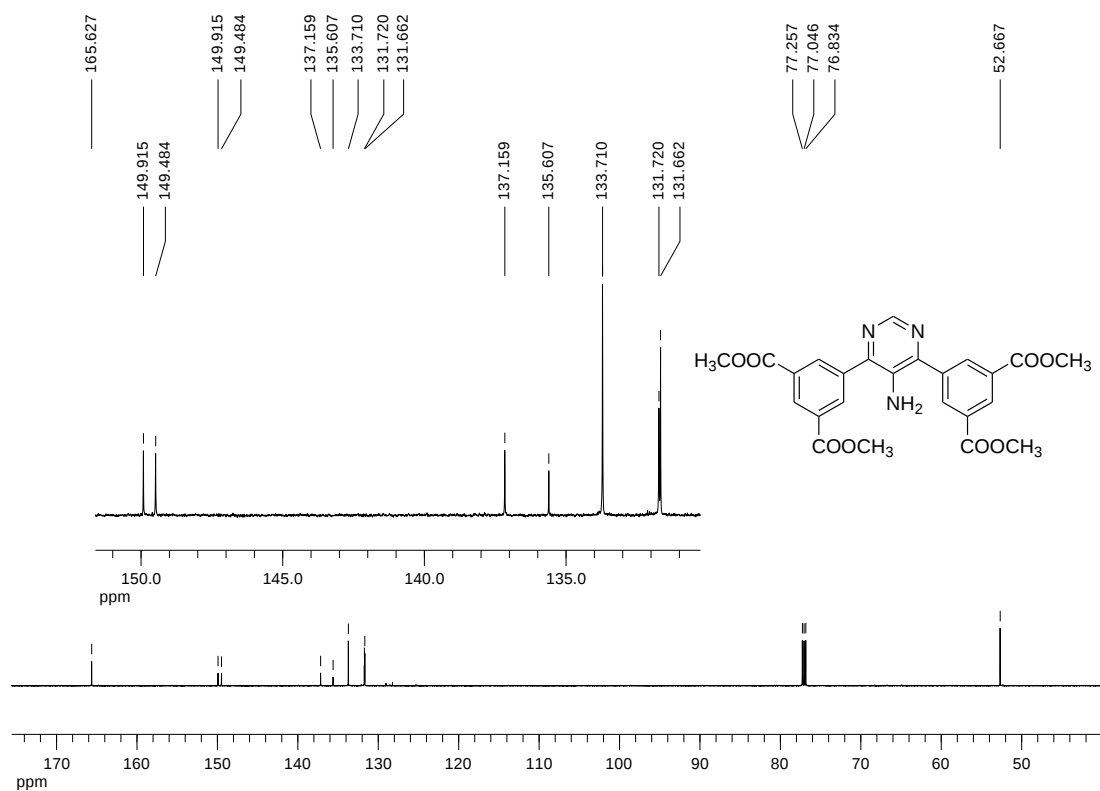
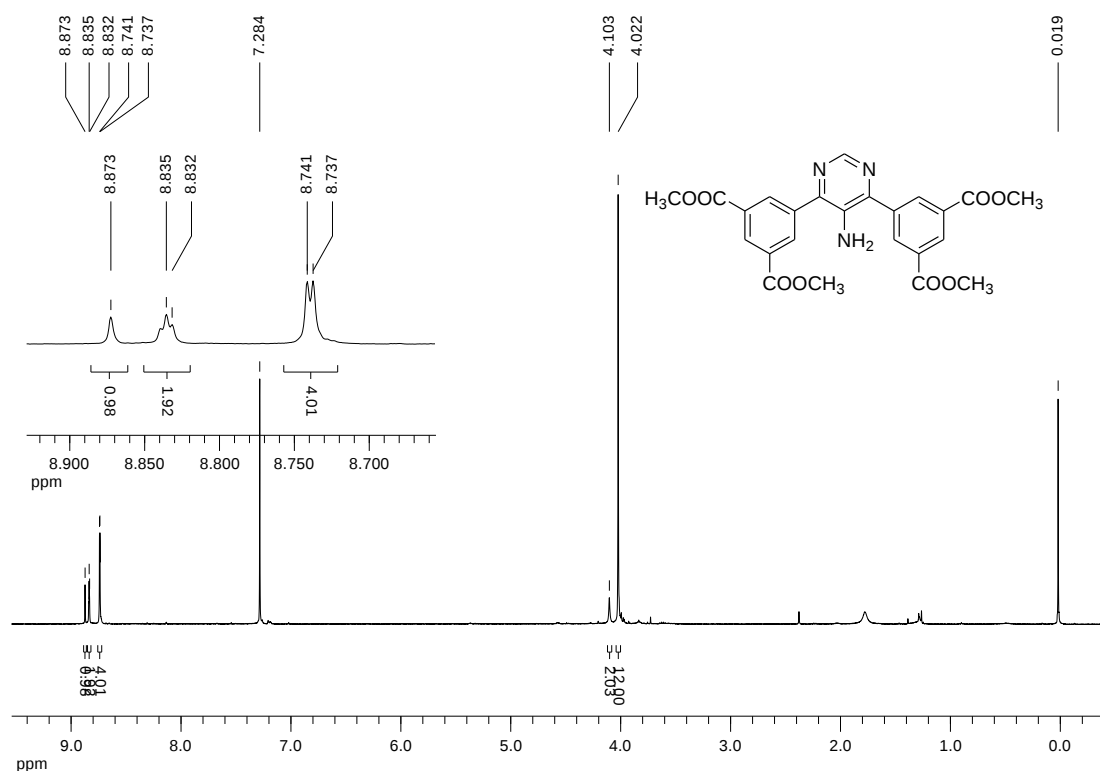


Fig S10 Isostere plots for C₂H₂ (a), CO₂ (b) and CH₄ (c) adsorption in ZJNU-54a.



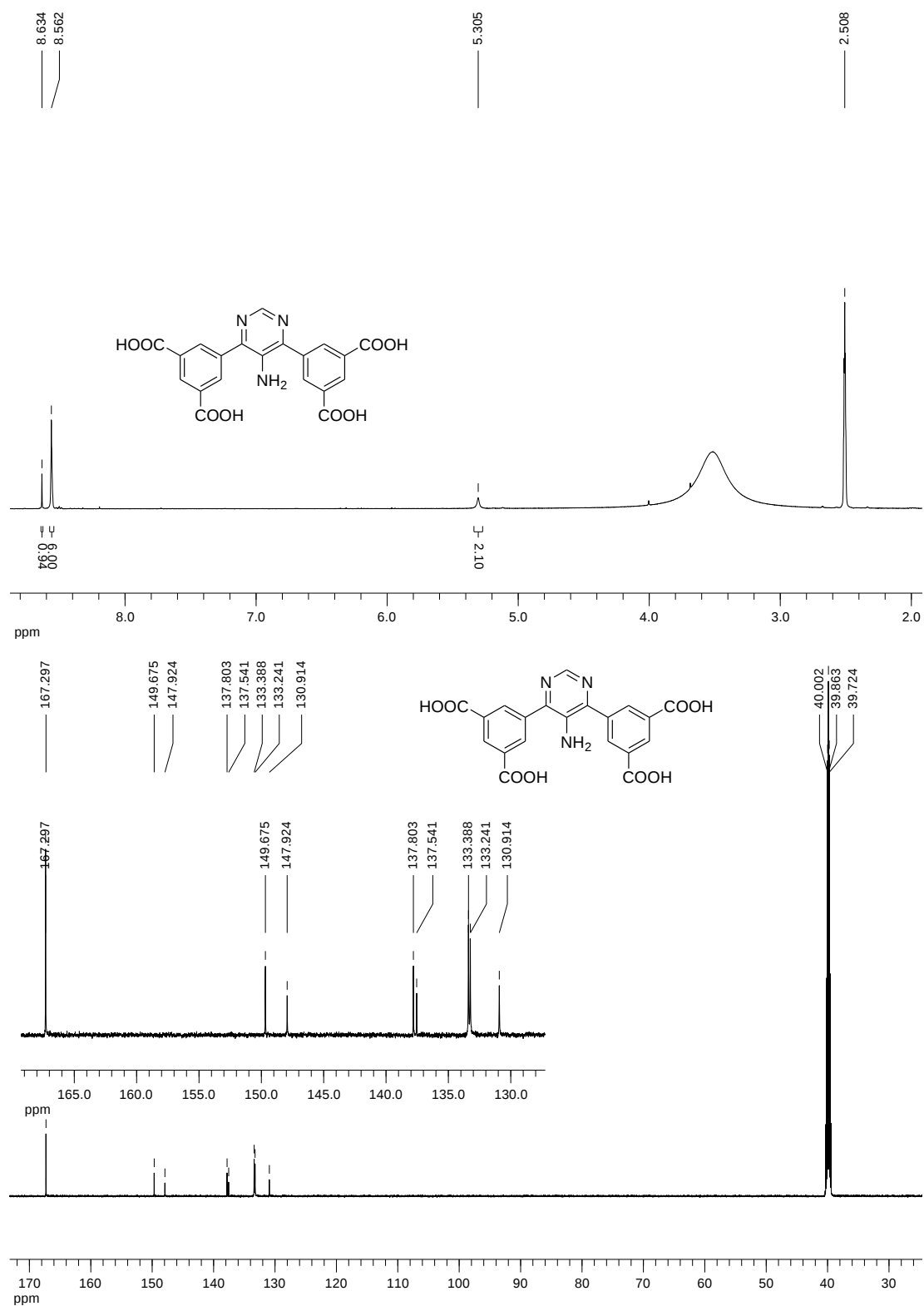


Fig. S11 ^1H NMR and ^{13}C NMR spectra.

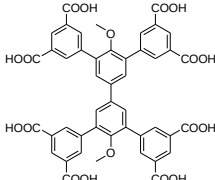
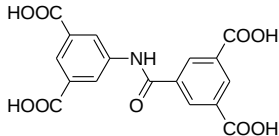
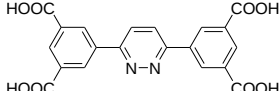
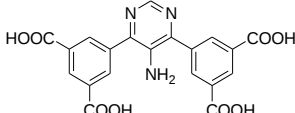
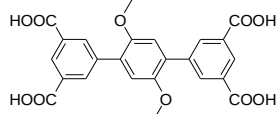
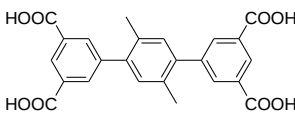
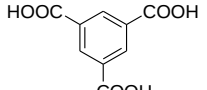
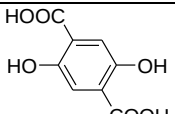
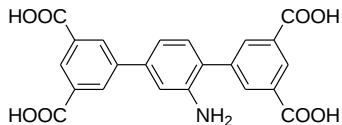
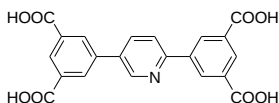
Table S1 Crystal data and structure refinement for **ZJNU-54**.

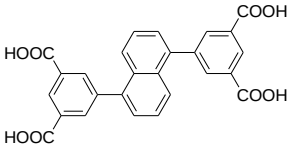
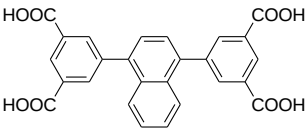
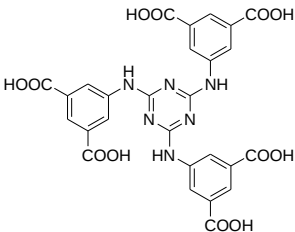
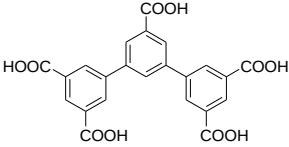
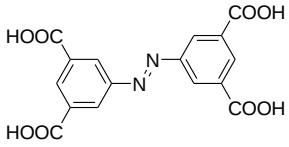
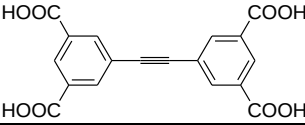
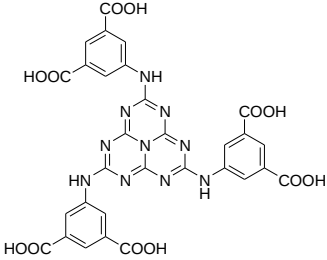
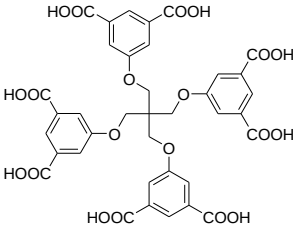
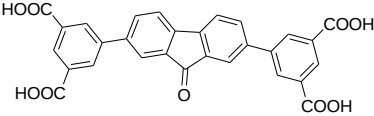
Empirical formula	C ₂₀ H ₁₄ N ₃ O ₁₀ Cu ₂
Formula weight	583.44
Temperature (K)	150
Wavelength (Å)	0.71073
Crystal system	Hexagonal
Space group	<i>P</i> 6 ₃ /mmc
Unit cell dimensions	<i>a</i> = 18.5653(6) Å <i>b</i> = 18.5653(6) Å <i>c</i> = 23.9636(11) Å <i>α</i> = 90° <i>β</i> = 90° <i>γ</i> = 120°
Volume (Å ³)	7153.0(5)
<i>Z</i>	6
Calculated density (g cm ⁻³)	0.813
Absorption coefficient (mm ⁻¹)	0.921
<i>F</i> (000)	1758
Crystal size (mm)	0.24 × 0.21 × 0.09
<i>θ</i> range for data collection (°)	1.27 to 27.60
Limiting indices	-24 ≤ <i>h</i> ≤ 24, -21 ≤ <i>k</i> ≤ 24, -31 ≤ <i>l</i> ≤ 31
Reflections collected / unique	115179 / 3099
<i>R</i> _{int}	0.0866
Completeness to <i>θ</i> = 27.60	99.9 %
Max. and min. transmission	0.9217 and 0.8092
Refinement method	Full-matrix least-squares on <i>F</i> ²
Data / restraints / parameters	3099 / 0 / 101
Goodness-of-fit on <i>F</i> ²	1.169
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0777, <i>wR</i> ₂ = 0.1629
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.1024, <i>wR</i> ₂ = 0.1774
Largest diff. peak and hole (e.Å ⁻³)	0.632 and -0.717
CCDC	1480522

Table S2 Langmuir-Freundlich parameters for adsorption of C₂H₂, CO₂, and CH₄ in ZJNU-54a

	q_{sat} (mmol g ⁻¹)	b_0 (kPa) ^{-ν}	E (kJ mol ⁻¹)	ν
C ₂ H ₂	16.21986	4.98323 × 10 ⁻⁶	22.60	0.71494
CO ₂	25.96059	2.07758 × 10 ⁻⁷	23.24	0.98438
CH ₄	10.13841	1.78756 × 10 ⁻⁶	15.91	1.0

Table S3 Comparison of C₂H₂ adsorption in the existing reports.

MOFs	Metal ions	Ligand structure	C ₂ H ₂ uptake	ref
FJI-H8	Cu ²⁺		224 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	1
NJU-Bai17	Cu ²⁺		222.4 cm ³ (STP) g ⁻¹ (296 K and 1 bar)	2
ZJNU-47	Cu ²⁺		213 cm ³ (STP) g ⁻¹ (295 K and 1 bar)	3
ZJNU-54	Cu ²⁺		211 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	This work
Cu₂TPTC-OMe	Cu ²⁺		204 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	4
Cu₂TPTC-Me	Cu ²⁺		203 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	4
HKUST-1	Cu ²⁺		201 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	5
CoMOF-74	Co ²⁺		197 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	6
ZJU-8	Cu ²⁺		195 cm ³ (STP) g ⁻¹ (296 K and 1 bar)	7
ZJU-5	Cu ²⁺		193 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	8

ZJU-9	Cu^{2+}		$193 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	9
ZJU-7	Cu^{2+}		$180 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 bar)	10
Cu-TDPAT	Cu^{2+}		$177.7 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	11
ZJU-72	Cu^{2+}		$167.7 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	12
CPM-200-Fe/Mg	$\text{Fe}^{3+},$ Mg^{2+}		$160.8 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 bar)	13
Cu-EBTC	Cu^{2+}		$160 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (296 K and 1 bar)	14
Cu-TDPAH	Cu^{2+}		$155.7 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	15
Cu-L	Cu^{2+}		$154 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (296 K and 1 atm)	16
ZJU-61	Cu^{2+}		$139.23 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	17

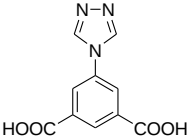
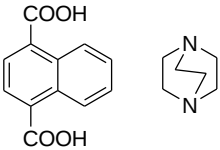
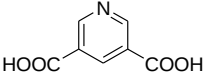
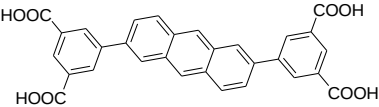
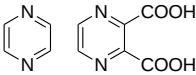
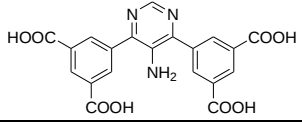
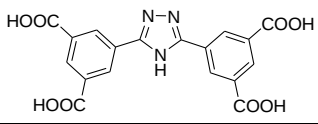
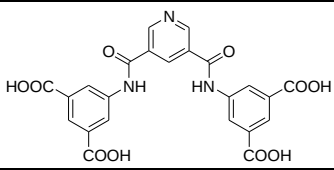
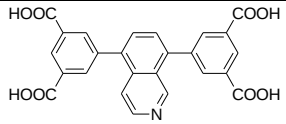
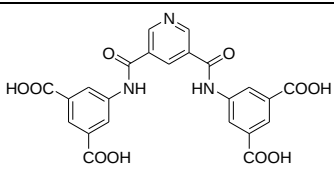
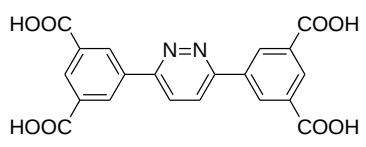
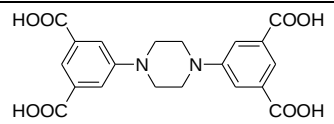
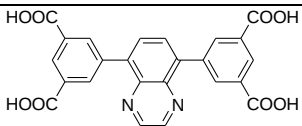
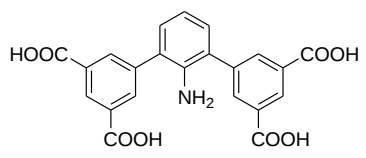
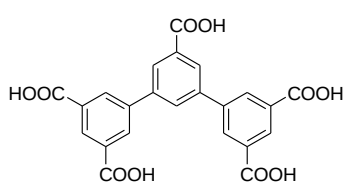
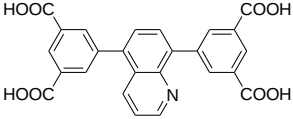
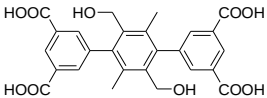
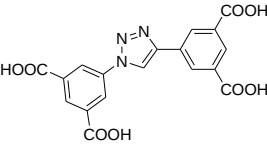
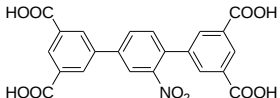
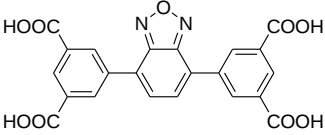
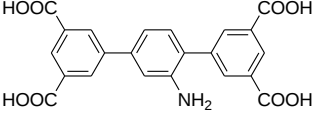
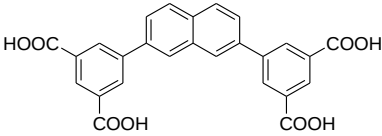
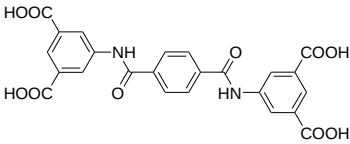
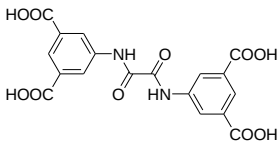
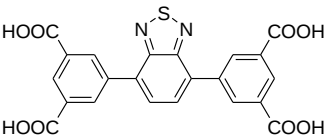
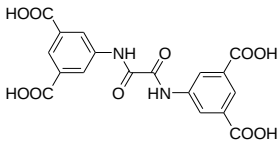
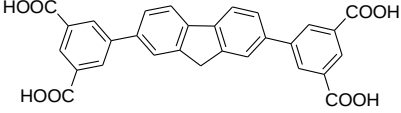
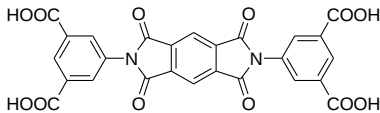
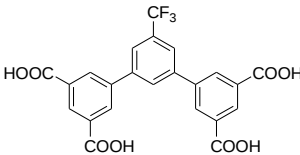
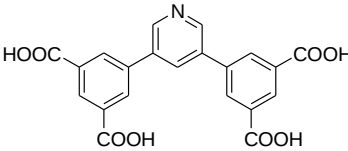
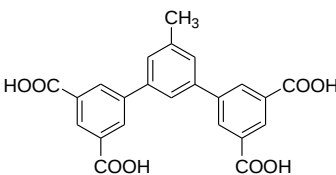
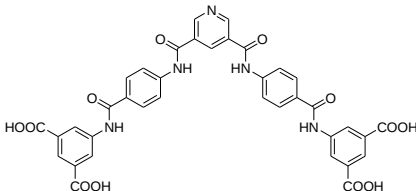
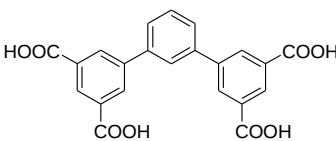
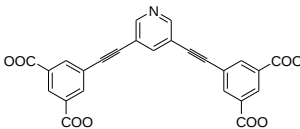
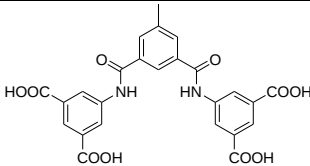
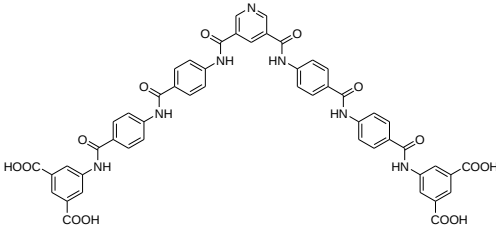
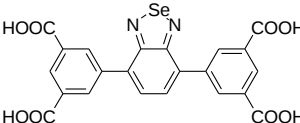
FJU-22a	Cu^{2+}		$114.8 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (296 K and 1 bar)	18
$\text{Zn}_2\text{L}_2(\text{dabco})$	Zn^{2+}		$106 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	19
UTSA-50	Cu^{2+}		$91 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (296 K and 1 atm)	20
ZJU-26	Cu^{2+}		$84 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	21
$\text{Mg}(\text{HCOO})_2$	Mg^{2+}		$65.7 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$ (298 K and 1 atm)	22
$\text{Cu}_2(\text{pzdc})_2(\text{pyz})$	Cu^{2+}		$42 \text{ cm}^3 (\text{STP}) \text{ g}^{-1}$	23

Table S4 Comparison of CO₂ adsorption in copper-based MOFs constructed from diisophthalate ligands.

MOFs	Ligand structure	CO ₂ uptake (conditions)	Ref
ZJNU-54		120 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	This work
JLU-Liu21		118 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	24
NJU-Bai21		115.1 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	25
ZJNU-44		116 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	26
PCN-124		114 cm ³ (STP) g ⁻¹ (295 K and 1 atm)	27
ZJNU-47		108 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	28
NPC-6		108 cm ³ (STP) g ⁻¹ (293 K and 1 bar)	29
ZJNU-45		107 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	26
Cu-L1		104.4 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	30
ZJU-72		103.9 cm ³ (STP) g ⁻¹ (298 K and 1 atm)	12

ZJNU-43		103 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	26
QI-Cu		102 cm ³ (STP) g ⁻¹ (293 K and 1 bar)	31
NTU-101-Cu		101 cm ³ (STP) g ⁻¹ (273 K and 1 atm)	32
NJU-Bai14		100 cm ³ (STP) g ⁻¹ (296 K and 1bar)	33
ZJNU-41		97.4 cm ³ (STP) g ⁻¹ (298 K and 1 atm)	34
ZJU-8		95 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	7
PCN-88		94 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	35
HNUST-1		93 cm ³ (STP) g ⁻¹ (296 K and 1 bar)	36
NOTT-125		92.6 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	37
ZJNU-40		85.7 cm ³ (STP) g ⁻¹ (298 K and 1 atm)	34
HNUST-3		84.5 cm ³ (STP) g ⁻¹ (298 K and 1bar)	38
ZJU-25		82 cm ³ (STP) g ⁻¹ (296 K and 1 atm)	39

SNU-50		80 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	40
PCN-308		78 cm ³ (STP) g ⁻¹ (297 K and 1 bar)	41
PCN-305		74 cm ³ (STP) g ⁻¹ (297 K and 1 bar)	41
PCN-307		73 cm ³ (STP) g ⁻¹ (297 K and 1 bar)	41
NJU-Bai22		72.9 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	25
PCN-306		70 cm ³ (STP) g ⁻¹ (297 K and 1 bar)	41
Cu-L		64 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	42
HNUST-4		62.9 cm ³ (STP) g ⁻¹ (296 K and 1 bar)	43
NJU-Bai23		49.0 cm ³ (STP) g ⁻¹ (298 K and 1 bar)	25
ZJNU-42		37.6 cm ³ (STP) g ⁻¹ (298 K and 1 atm)	34

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