

A Series of Anderson-type Polyoxometalate-based Metal–organic Complexes: pH-dependent electrochemical behaviours and as Electrocatalysts and Photocatalysts

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Table. S1. Selected bond distances (Å) and angles (°) for the title complexes.

Complex 1			
O(1)-Cu(1)	1.990(4)	N(1)-Cu(2)	2.002(5)
O(1W)-Cu(1)	2.173(5)	Cu(1)-O(3)	1.953(4)
Cu(1)-O(2W)	1.963(4)	Cu(1)-N(3)	1.981(5)
N(2)-Cu(2)	1.975(4)	Cu(2)-O(4)	1.920(4)
Cu(2)-O(7)#1	1.936(4)	O(7)-Cu(2)#1	1.936(4)
Mo(1)-O(1)-Cu(1)	140.6(2)	C(10)-N(1)-Cu(2)	129.4(4)
C(6)-N(1)-Cu(2)	111.5(4)	O(3)-Cu(1)-O(2W)	172.16(19)
O(3)-Cu(1)-N(3)	88.03(18)	O(2W)-Cu(1)-N(3)	97.67(19)
O(3)-Cu(1)-O(1)	87.43(17)	O(2W)-Cu(1)-O(1)	85.57(18)
N(3)-Cu(1)-O(1)	162.84(18)	O(3)-Cu(1)-O(1W)	93.48(19)
O(2W)-Cu(1)-O(1W)	91.0(2)	N(3)-Cu(1)-O(1W)	98.3(2)
O(1)-Cu(1)-O(1W)	98.5(2)	O(4)-Cu(2)-O(7)#1	87.94(17)
O(4)-Cu(2)-N(2)	175.08(19)	O(7)#1-Cu(2)-N(2)	95.36(18)
O(4)-Cu(2)-N(1)	95.51(18)	O(7)#1-Cu(2)-N(1)	164.42(18)
N(2)-Cu(2)-N(1)	82.28(18)		
Symmetry codes for 1 : #1 -x,-y-1,-z+1			
Complex 2			
Cu(1)-O(1W)	1.985(6)	Cu(1)-O(4W)	1.970(6)
Cu(1)-N(1)	2.017(7)	Cu(1)-O(3W)	2.026(6)
Cu(1)-O(2W)	2.185(6)	O(1W)-Cu(1)-O(4W)	172.6(3)
O(1W)-Cu(1)-N(1)	90.9(3)	O(4W)-Cu(1)-N(1)	89.2(3)
O(1W)-Cu(1)-O(3W)	89.6(3)	O(4W)-Cu(1)-O(3W)	89.7(3)
N(1)-Cu(1)-O(3W)	174.6(3)	O(1W)-Cu(1)-O(2W)	87.8(3)
O(4W)-Cu(1)-O(2W)	99.6(2)	N(1)-Cu(1)-O(2W)	94.2(3)
O(3W)-Cu(1)-O(2W)	91.2(3)		
Complex 3			
Cu(1)-N(1)	1.990(3)	Cu(1)-O(1W)	1.997(3)
Cu(1)-N(1)#1	1.990(3)	Cu(1)-O(1W)#1	1.997(3)
Cu(1)-O(1)	2.393(2)	Cu(1)-O(1)#1	2.393(2)
N(1)-Cu(1)-N(1)#1	180.0(15)	N(1)-Cu(1)-O(1W)	89.17(11)
N(1)#1-Cu(1)-O(1W)	90.83(11)	N(1)-Cu(1)-O(1W)#1	90.83(11)

N(1)#1-Cu(1)-O(1W)#1	89.17(11)	O(1W)-Cu(1)-O(1W)#1	180.000(2)
N(1)-Cu(1)-O(1)	87.72(9)	N(1)#1-Cu(1)-O(1)	92.28(9)
O(1W)-Cu(1)-O(1)	93.73(9)	O(1W)#1-Cu(1)-O(1)	86.27(9)
N(1)-Cu(1)-O(1)#1	92.28(9)	N(1)#1-Cu(1)-O(1)#1	87.72(9)
O(1W)-Cu(1)-O(1)#1	86.27(9)	O(1W)#1-Cu(1)-O(1)#1	93.73(9)
O(1)-Cu(1)-O(1)#1	180.000(1)		

Symmetry codes for **3**: #1 -x+2,-y+1,-z+2

Complex 4			
Cu(1)-O(4)	1.953(19)	Cu(1)-N(1)	1.987(3)
Cu(1)-O(7)#2	2.018(2)	Cu(1)-O(2)	2.128(2)
Cu(1)-O(1)	2.161(2)	Cu(1)-O(3)	2.307(3)
Cu(2)-O(5)	1.926(2)	Cu(2)-O(6)	1.934(2)
Cu(2)-O(4)#4	1.967(2)	Cu(2)-O(4)	1.975(2)
Cu(2)-O(3)#4	2.358(2)	Cu(2)-Cu(2)#4	2.9880(7)
O(3)-Cu(2)#4	2.358(2)	O(4)-Cu(2)#4	1.967(2)
O(7)-Cu(1)#2	2.018(2)	O(4)-Cu(1)-N(1)	176.80(10)
O(4)-Cu(1)-O(7)#2	89.94(8)	N(1)-Cu(1)-O(7)#2	89.98(10)
O(4)-Cu(1)-O(2)	86.24(9)	N(1)-Cu(1)-O(2)	93.35(10)
O(7)#2-Cu(1)-O(2)	170.28(9)	O(4)-Cu(1)-O(1)	92.42(8)
N(1)-Cu(1)-O(1)	90.78(10)	O(7)#2-Cu(1)-O(1)	93.28(9)
O(2)-Cu(1)-O(1)	95.80(9)	O(4)-Cu(1)-O(3)	83.58(8)
N(1)-Cu(1)-O(3)	93.22(10)	O(7)#2-Cu(1)-O(3)	89.42(9)
O(2)-Cu(1)-O(3)	81.29(9)	O(1)-Cu(1)-O(3)	175.19(9)
O(5)-Cu(2)-O(6)	90.31(9)	O(5)-Cu(2)-O(4)#4	168.51(9)
O(6)-Cu(2)-O(4)#4	97.05(9)	O(5)-Cu(2)-O(4)	94.33(9)
O(6)-Cu(2)-O(4)	161.62(9)	O(4)#4-Cu(2)-O(4)	81.43(8)
O(5)-Cu(2)-O(3)#4	88.56(9)	O(6)-Cu(2)-O(3)#4	95.17(9)
O(4)#4-Cu(2)-O(3)#4	81.95(8)	O(4)-Cu(2)-O(3)#4	102.13(9)
O(5)-Cu(2)-Cu(2)#4	134.02(7)	O(6)-Cu(2)-Cu(2)#4	135.09(7)
O(4)#4-Cu(2)-Cu(2)#4	40.82(6)	O(4)-Cu(2)-Cu(2)#4	40.60(6)
O(3)#4-Cu(2)-Cu(2)#4	92.68(7)		

Symmetry codes for **4**: #2 -x+1,-y+2,-z+1 #4 -x,-y+2,-z+1

Complex 5			
Cu(1)-O(3W)	1.953(2)	Cu(1)-O(2W)	1.988(3)
Cu(1)-N(1)	2.003(3)	Cu(1)-O(1)	2.004(3)
Cu(1)-O(1W)	2.274(3)	Cu(1)-O(3)#1	2.355(3)
O(3W)-Cu(1)-O(2W)	90.59(11)	O(3W)-Cu(1)-N(1)	174.67(12)
O(2W)-Cu(1)-N(1)	93.10(11)	O(3W)-Cu(1)-O(1)	87.07(11)
O(2W)-Cu(1)-O(1)	177.66(10)	N(1)-Cu(1)-O(1)	89.23(11)
O(3W)-Cu(1)-O(1W)	92.97(11)	O(2W)-Cu(1)-O(1W)	87.28(11)
N(1)-Cu(1)-O(1W)	91.04(11)	O(1)-Cu(1)-O(1W)	92.89(10)
O(3W)-Cu(1)-O(3)#1	91.22(11)	O(2W)-Cu(1)-O(3)#1	90.85(10)
N(1)-Cu(1)-O(3)#1	84.90(11)	O(1)-Cu(1)-O(3)#1	89.15(10)
O(1W)-Cu(1)-O(3)#1	175.43(10)		

Symmetry codes for **5**: #1 -x,-y+4,-z-1

Complex 6			
Cu(1)-N(1)#2	1.994(4)	Cu(1)-N(1)	1.994(4)
Cu(1)-O(1W)#2	2.007(3)	Cu(1)-O(1W)	2.007(3)
Cu(1)-O(1)#2	2.293(3)	Cu(1)-O(1)	2.293(3)
Cu(2)-O(4W)	1.949(4)	Cu(2)-O(2W)	2.002(4)
Cu(2)-N(2)	2.011(4)	Cu(2)-O(3W)	2.024(4)
Cu(2)-O(2)	2.317(3)	Cu(2)-O(3)	2.350(3)
N(1)#2-Cu(1)-N(1)	180.0(3)	N(1)#2-Cu(1)-O(1W)#2	89.23(15)
N(1)-Cu(1)-O(1W)#2	90.77(15)	N(1)#2-Cu(1)-O(1W)	90.77(15)
N(1)-Cu(1)-O(1W)	89.23(15)	O(1W)#2-Cu(1)-O(1W)	180.0(1)
N(1)#2-Cu(1)-O(1)#2	91.56(14)	N(1)-Cu(1)-O(1)#2	88.44(14)
O(1W)#2-Cu(1)-O(1)#2	84.01(13)	O(1W)-Cu(1)-O(1)#2	95.98(13)
N(1)#2-Cu(1)-O(1)	88.44(14)	N(1)-Cu(1)-O(1)	91.56(14)
O(1W)#2-Cu(1)-O(1)	95.98(13)	O(1W)-Cu(1)-O(1)	84.01(13)
O(1)#2-Cu(1)-O(1)	180.0(13)	O(4W)-Cu(2)-O(2W)	86.84(18)
O(4W)-Cu(2)-N(2)	175.94(18)	O(2W)-Cu(2)-N(2)	91.44(17)
O(4W)-Cu(2)-O(3W)	88.51(18)	O(2W)-Cu(2)-O(3W)	172.11(14)
N(2)-Cu(2)-O(3W)	93.60(17)	O(4W)-Cu(2)-O(2)	90.60(14)
O(2W)-Cu(2)-O(2)	102.47(13)	N(2)-Cu(2)-O(2)	86.17(14)
O(3W)-Cu(2)-O(2)	83.94(13)	O(4W)-Cu(2)-O(3)	93.50(14)
O(2W)-Cu(2)-O(3)	86.07(13)	N(2)-Cu(2)-O(3)	90.05(14)
O(3W)-Cu(2)-O(3)	87.86(13)	O(2)-Cu(2)-O(3)	170.73(12)

Symmetry codes for **6**: #2 -x,-y+1,-z

Complex 7			
Cu(1)-O(3W)	1.963(4)	Cu(1)-O(2)	1.984(4)
Cu(1)-O(2W)	1.985(4)	Cu(1)-N(1)	2.012(4)
Cu(1)-O(1W)	2.278(4)	Cu(1)-O(1)	2.370(5)
O(3W)-Cu(1)-O(2)	176.99(17)	O(3W)-Cu(1)-O(2W)	90.85(19)
O(2)-Cu(1)-O(2W)	86.13(19)	O(3W)-Cu(1)-N(1)	93.86(18)
O(2)-Cu(1)-N(1)	89.13(18)	O(2W)-Cu(1)-N(1)	172.0(2)
O(3W)-Cu(1)-O(1W)	90.65(18)	O(2)-Cu(1)-O(1W)	89.55(16)
O(2W)-Cu(1)-O(1W)	93.4(2)	N(1)-Cu(1)-O(1W)	93.01(18)
O(3W)-Cu(1)-O(1)	89.69(16)	O(2)-Cu(1)-O(1)	90.15(16)
O(2W)-Cu(1)-O(1)	87.4(2)	N(1)-Cu(1)-O(1)	86.21(17)
O(1W)-Cu(1)-O(1)	179.17(16)		

Table S2a. Selected hydrogen–bonding geometry (Å, °) for complex **1**

D–H···A	D–H	H···A	D···A	D–H···A
C(9)–H(9A)···O(26)	0.93	2.56	3.262	133

Table S2b. Selected hydrogen-bonding geometry (Å, °) for complex **2**

D–H⋯A	D–H	H⋯A	D⋯A	D–H⋯A
C(7)–H(7A) ⋯O(2)	0.93	2.55	3.3355	142

Table S2c. Selected hydrogen-bonding geometry (Å, °) for complex **3**

D–H⋯A	D–H	H⋯A	D⋯A	D–H⋯A
C(12)–H(12A) ⋯O(8),	0.93	2.37	3.167(5)	143

Table S2d. Selected hydrogen-bonding geometry (Å, °) for complex **5**

D–H⋯A	D–H	H⋯A	D⋯A	D–H⋯A
N(4)–H(4C) ⋯O(12),	0.86	1.78	2.639(4)	177
O3W–H3WB ⋯O(7)	0.85	2.48	2.922(4)	113

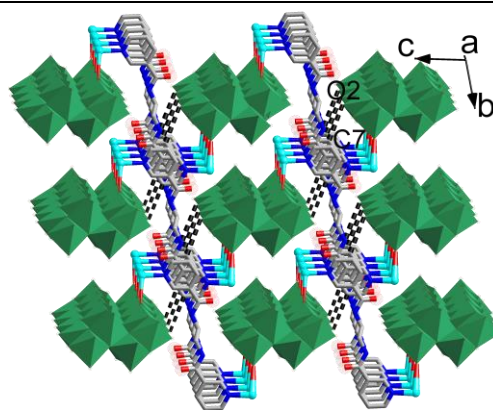


Fig. S1. The 3D supramolecular framework in **2**

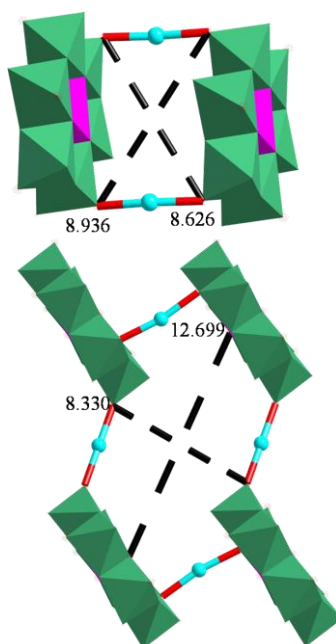


Fig. S2. Two kinds of the $\text{Cu}_2(\text{TeMo}_6)_2$ and $\text{Cu}_4(\text{TeMo}_6)_4$ loops in complex **6**

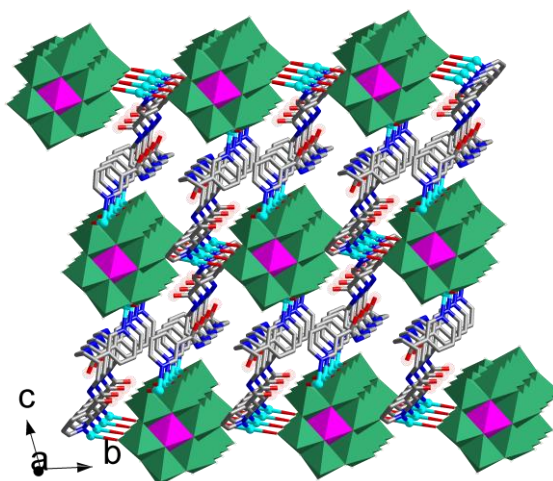


Fig. S3. The 3D framework of complex 6.

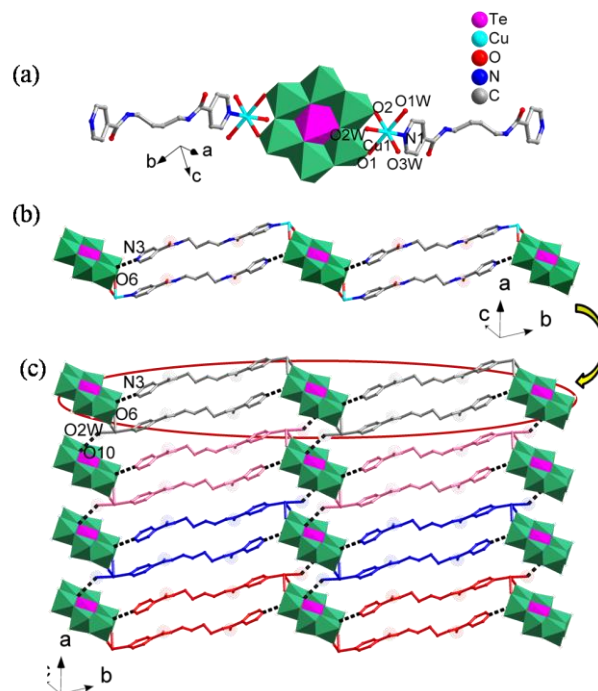


Fig. S4. (a) The coordination environment of the Cu^{II} ion in 7. All H atoms and lattice water molecules are omitted for clarity. (b) The 1D supramolecular chain linked by hydrogen bonding interaction [N(3)–H(3B)⋯O(6)]. (c) The 2D supramolecular network of 7 formed by hydrogen bonding interaction [O2W–H2WB⋯O(10)].

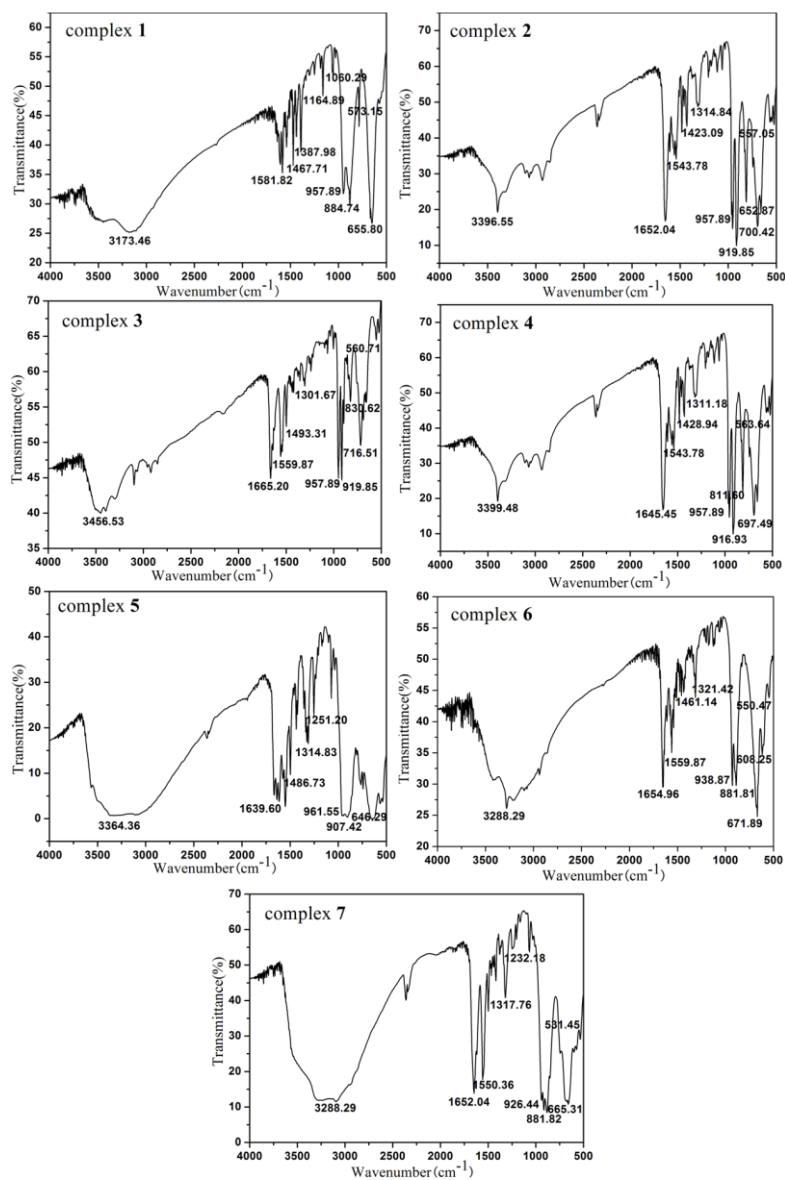


Fig. S5. The IR spectra of the title complexes.

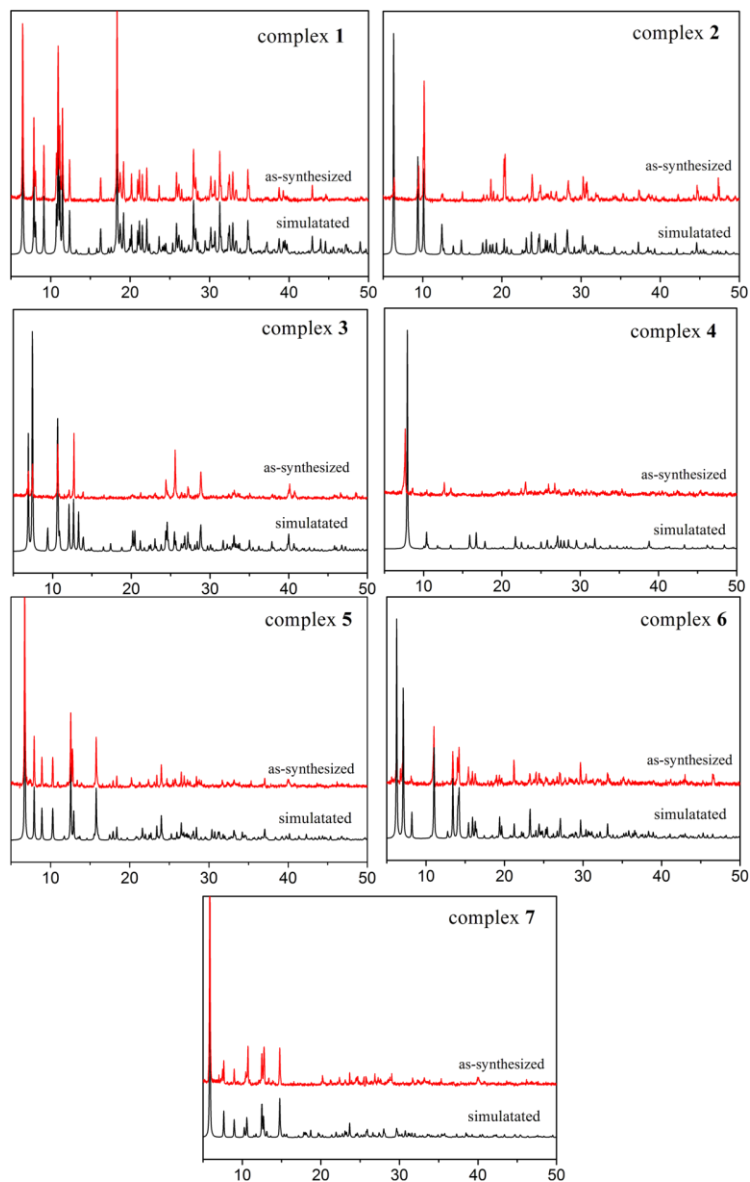


Fig. S6. Powder X-ray diffraction patterns of the title complexes.

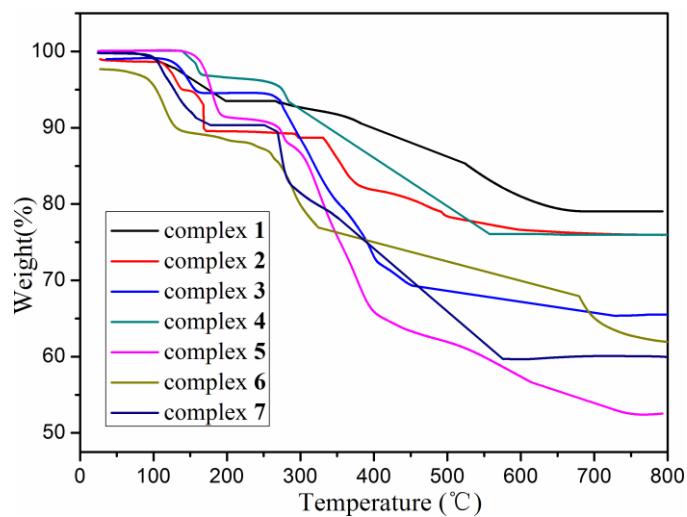


Fig. S7. The TGA curves of the title complexes.

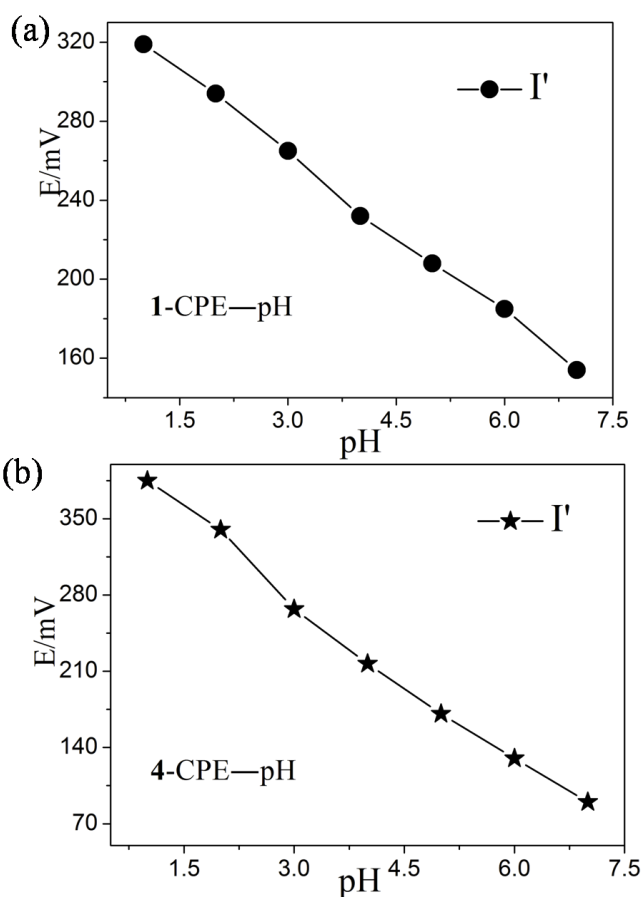


Fig. S8. The dependence of anodic peak potentials on the pH values of medium for 1–(a) and 4–(b)CPE.

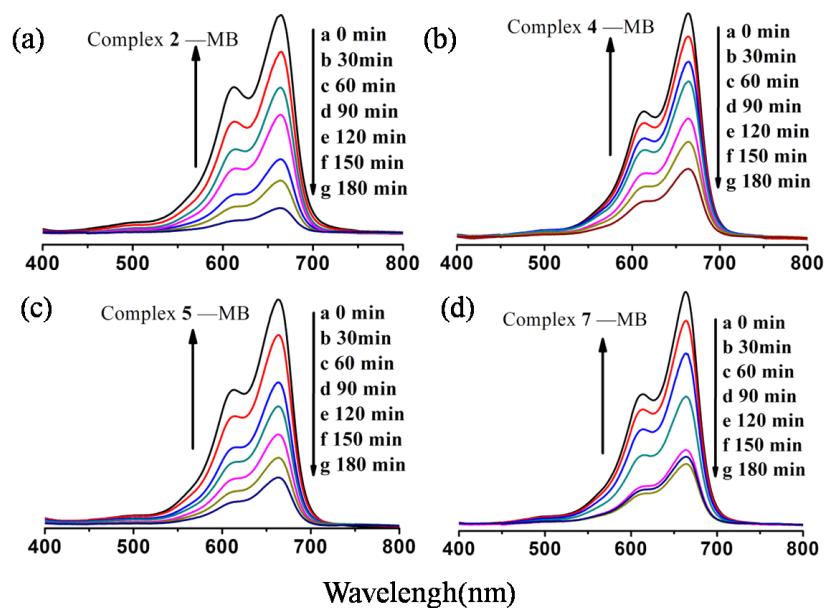


Fig. S9. Absorption spectra of the MB solution during the decomposition reaction under UV irradiation at the presence of complexes 2, 4, 5 and 7.

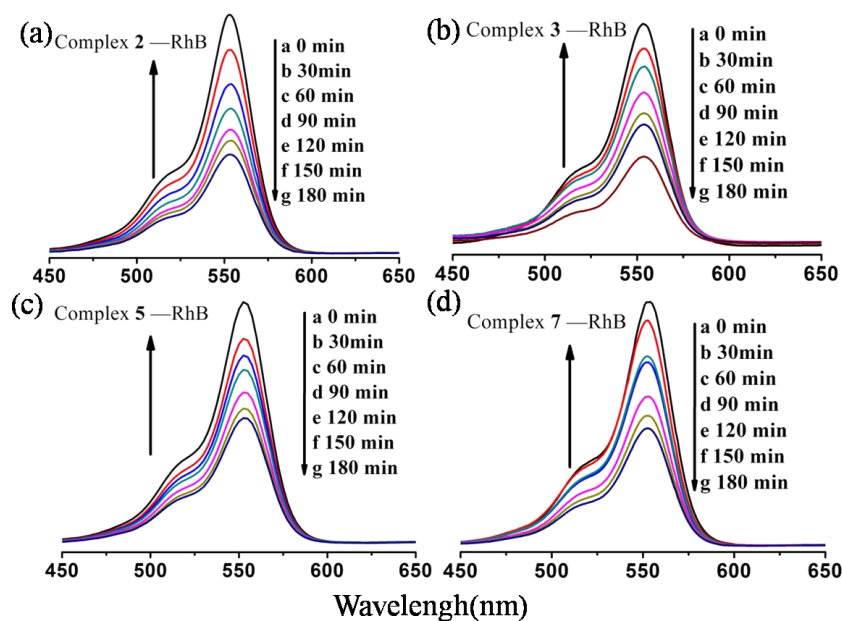


Fig. S10. Absorption spectra of the RhB solution during the decomposition reaction under UV irradiation at the presence of complexes 2, 3, 5 and 7.

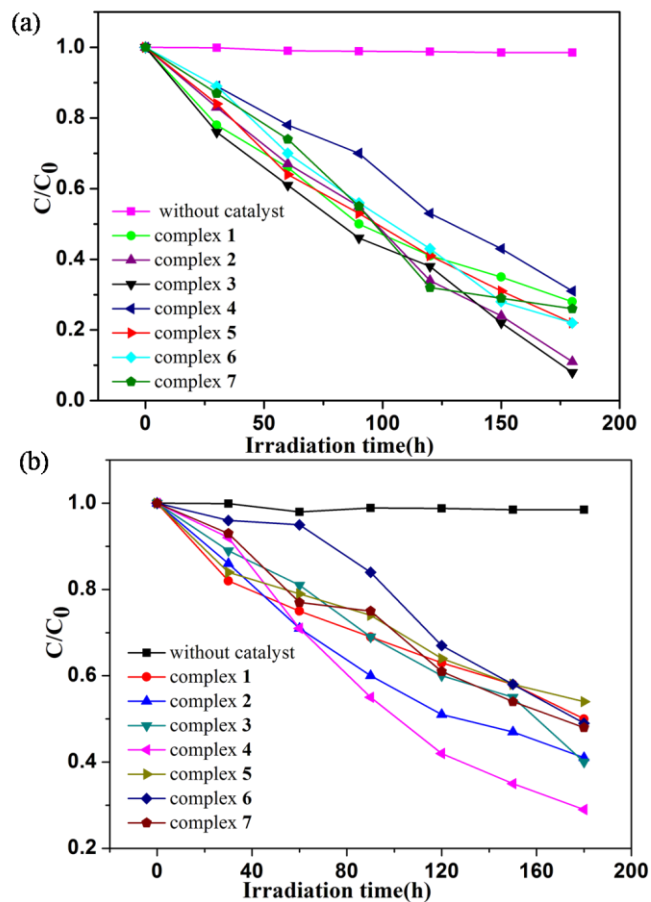


Fig. S11 Photocatalytic decomposition rate of MB solution (a), and RhB (b) under UV irradiation with the use of title complexes.