

Modulating the Hydration Behaviour of Calcium Chloride by Lactam Complexation

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SUPPLEMENTARY INFORMATION

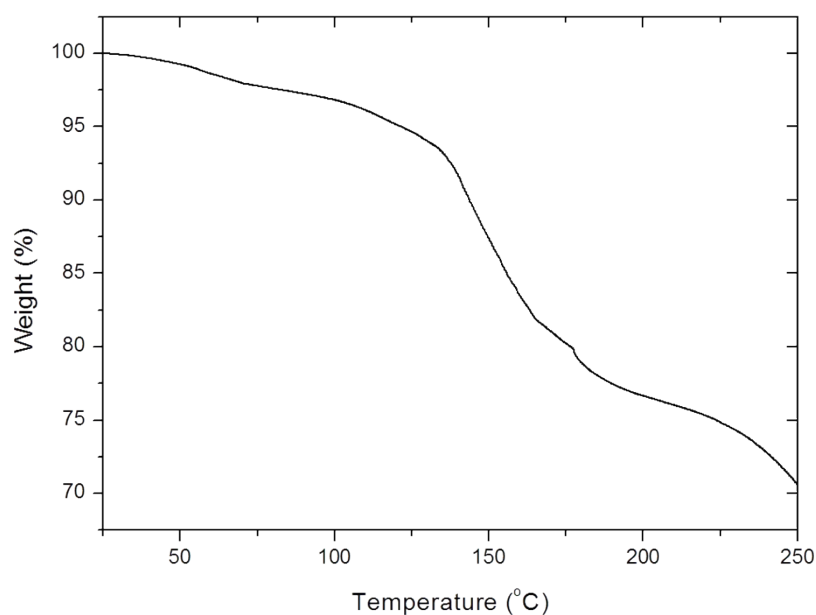


Figure S1. Thermogravimetric trace for $\{[\text{Ca}(\mu\text{-L1})(\text{H}_2\text{O})_5]\text{Cl}_2\cdot\text{H}_2\text{O}\}_n$ (**2**).

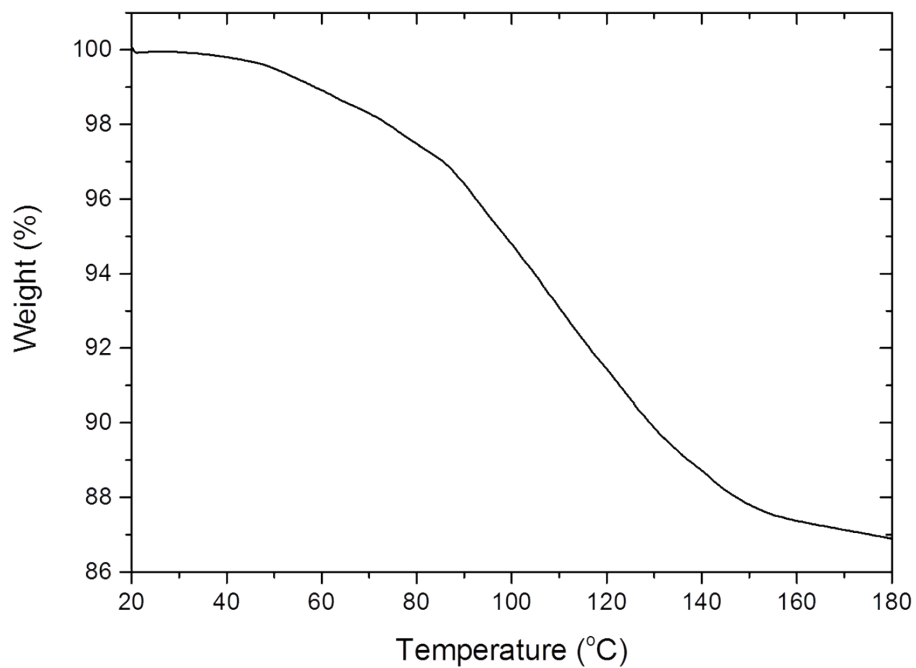


Figure S2. Thermogravimetric trace for the mechanochemically prepared product of reaction of CaCl₂ with ligand L1. Product was previously dried at 70 °C for 3 hours.

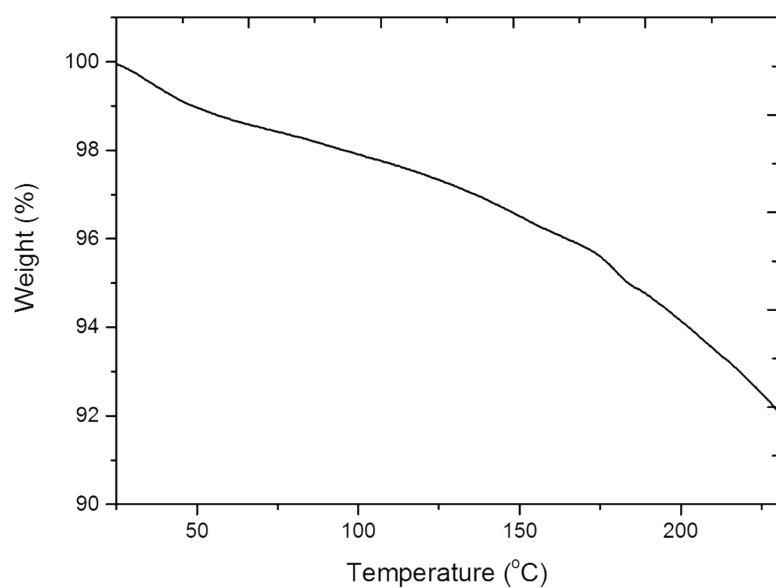


Figure S3. Thermogravimetric trace for the ethanol solvate {[CaCl₂(L2)(EtOH)]_n (4).

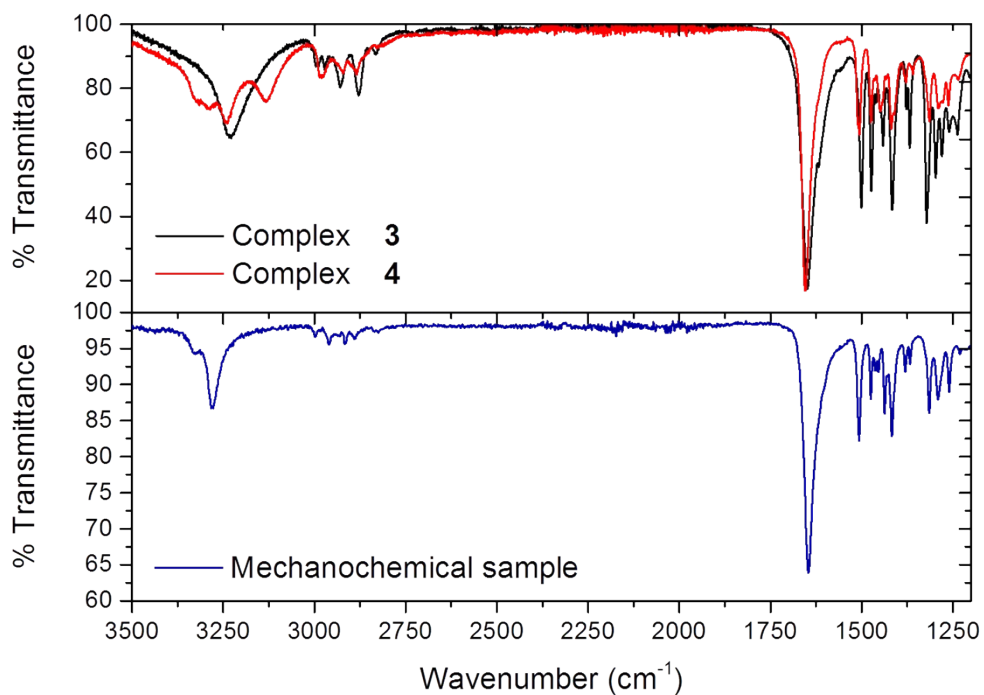


Figure S4. IR spectra for complexes **3** and **4** in comparison to the mechanochemically prepared sample.

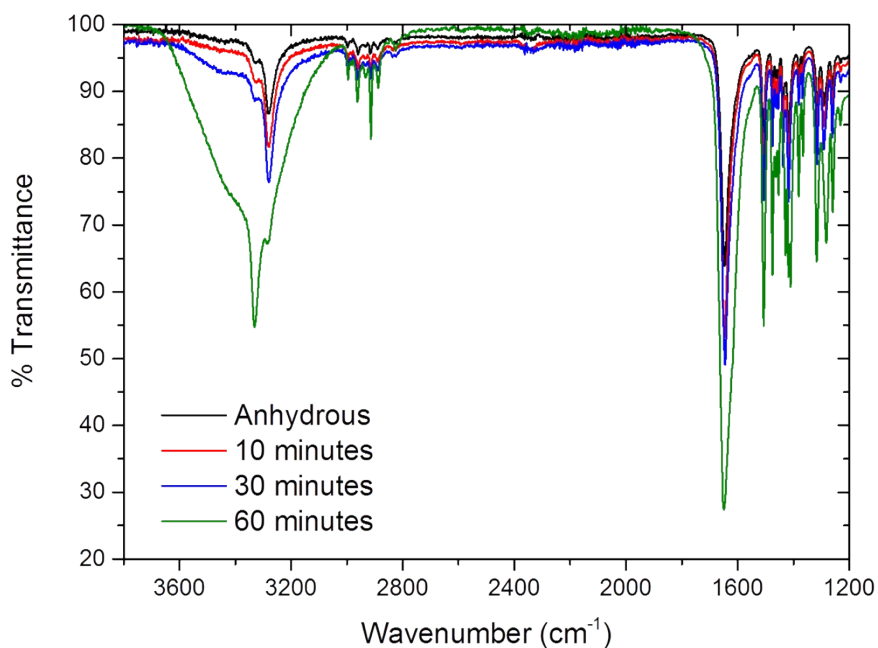


Figure S5. Overlaid IR spectra for the anhydrous mechanochemically prepared complex of CaCl₂ and ligand L2 following removal from the oven (110 °C).