Electronic Supplementary Material (ESI) for Dalton Transactions. This journal is © The Royal Society of Chemistry 2016

Electronic Supplementary Information for Dalton Transactions

This journal is © The Royal Society of Chemistry XXXX

Supporting Information for Manuscript:

Iron(III) Bis(pyrazol-1-yl)acetate Based Decanuclear Metallacycle: Synthesis, Structure, Magnetic Properties and DFT Calculations

Małgorzata J. Gajewska,^a Alina Bieńko,^b Radovan Herchel,^{c*} Matti Haukka,^d Maria Jerzykiewicz,^b Andrzej Ożarowski,^e Krzysztof Drabent,^b Chen-Hsiung Hung,^{a*}

^a Institute of Chemistry, Academia Sinica, Nankang, Taipei 105, Taiwan; E-mail: <u>chhung@gate.sinica.edu.tw</u>
^b Faculty of Chemistry, Wrocław University, F. Joliot-Curie 14; 50-383 Wrocław Poland
^c Regional Centre of Advanced Technologies and Materials & Department of Inorganic Chemistry, Faculty of Science, Palacký University, 17. listopadu 12, 771 46 Olomouc, Czech Republic; E-mail: <u>radovan.herchel@upol.cz</u>
^c Department of Chemistry, FI-40014 University of Jyväskylä, Finland
^e Nafional High Magnefic Field Laboratory, Florida State University, Tallahassee, FL 32310, USA

*To whom correspondence should be addressed:

Phone: 886-2-27898570; Fax: 886-2-27831237; E-mail: <u>chhung@chem.sinica.edu.tw</u>

Electronic Supplementary Information for Dalton Transactions

This journal is © The Royal Society of Chemistry XXXX

Figure S1 - The IR spectra of $[Fe_{10}(bdtbpza)_{10}(MeO)_{20}]$ (bottom) and bis(3,5-di-t-butylpyrazol-1-yl)acetate ligand (top).



Figure S2 – UV-vis spectra of $[Fe_{10}(bdtbpza)_{10}(MeO)_{20}]$ (1) at room temperature in DCM solvent with concentration 6.35 x 10⁻⁵ M







Figure S4 The space-filling representations of {Fe₁₀} wheel showing the interior cavity diameters; (a) whole molecule, (b) without H atoms; (C-grey, H-cream, O-red, N-blue, Fe green)





b)

| Atom no. | Valence state | Most consistent | BVS | % Deviation from |
|----------|---------------|-----------------|-------|-----------------------|
| | assumed | valence state | | assumed valence state |
| Fe1 | Fe1(2) | | 2.939 | 47 |
| Fe1 | Fe1(3) | * | 3.144 | 5 |
| Fe1 | Fe1(6) | | 3.039 | 49 |
| Fe1 | Fe1(9) | | 3.053 | 66 |
| Fe2 | Fe2(2) | | 3.022 | 51 |
| Fe2 | Fe2(3) | * | 3.233 | 8 |
| Fe2 | Fe2(6) | | 3.13 | 48 |
| Fe2 | Fe2(9) | | 3.162 | 65 |
| Fe3 | Fe3(2) | | 3.001 | 50 |
| Fe3 | Fe3(3) | * | 3.211 | 7 |
| Fe3 | Fe3(6) | | 3.108 | 48 |
| Fe3 | Fe3(9) | | 3.136 | 65 |
| Fe4 | Fe4(2) | | 2.956 | 48 |
| Fe4 | Fe4(3) | * | 3.162 | 5 |
| Fe4 | Fe4(6) | | 3.058 | 49 |
| Fe4 | Fe4(9) | | 3.077 | 66 |
| Fe5 | Fe5(2) | | 3.016 | 51 |
| Fe5 | Fe5(3) | * | 3.227 | 8 |
| Fe5 | Fe5(6) | | 3.123 | 48 |
| Fe5 | Fe5(9) | | 3.154 | 65 |

Table S1 Selected Bond valence sums for Fe atoms in (1)

(Only bonds that have all of the valence states defined in a given model have their bond valence parameters reported)