Supporting Information

Unusual $\kappa 1$ Coordination of a β -Diketiminate Ligand in Niobium Complexes

Jessica A. Ziegler, Robert G. Bergman*, and John Arnold*

Department of Chemistry, University of California, Berkeley, California 94720, United States

A.	NMR Spectroscopic Analysis	S2
	A.1 NMR spectroscopy of 2	S2
	A.2 NMR spectroscopy of 3	S3
	A.3 NMR spectroscopy of 4	S4
	A.4 ¹ H NMR spectrum of 5	S5
	A.5 NMR characterization of 6	S5
B.	Infrared Spectroscopic Analysis 6 and 6-D	S7
C.	X-ray Crystallography of 5	S7

A. NMR Spectroscopic Analysis



Figure S1: ¹H NMR spectrum of 2 in C_6D_6 at 298 K.



Figure S2: ${}^{13}C{}^{1}H$ NMR spectrum of 2 in C₆D₆ at 298 K.



Figure S4: Representative ¹H NMR spectrum of **3** in CDCl₃ before decomposition occurs.



Figure S5: ¹H NMR spectrum of 4 in CD₂Cl₂ at 298 K.



Figure S6: ${}^{13}C{}^{1}H$ NMR spectrum of 4 in CD₂Cl₂ at 298 K.



Figure S7: ¹H NMR spectrum of paramagnetic complex **5** in C₆D₆ at room temperature.



Figure S8: ¹H NMR spectrum of 6 in C₆D₆ at 298 K.



Figure S9: Stacked spectra of **6** (blue, top) and **6-D** (red, bottom) with the aromatic region enlarged, showing the chemical shift for the Nb-*H* that appears at 7.36 ppm for **6** and is absent in the spectrum of **6-D**.



Figure S10: ${}^{13}C{}^{1}H$ NMR spectrum of 6 in C₆D₆ at 298 K.

B. Infrared Spectroscopic Analysis



Figure S11: Overlaid IR spectra of 6 (blue) and 6-D (red).

C. X-ray Crystallography of 5



Figure S12: Molecular structure of **5** as determined by X-ray diffraction. Hydrogen atoms, a second equivalent molecule of **5**, and aryl isopropyl groups have been omitted for clarity; thermal ellipsoids are set at the 50% probability level. Selected bond lengths (Å): Nb(1)-N(1) 1.794(2), Nb(1)-N(2) 2.210(2), Nb(1)-N(3) 2.207(2). Selected bond angles (°): C(6)-N(1)-Nb(1) 168.07(2), N(2)-Nb(1)-N(3) 83.68(8).