Supplementary Material for

Revisiting oxo-centered carbonyl-triruthenium clusters: investigating CO photorelease and some spectroscopic and electrochemical correlations

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Compound	1 ·H₂O	$7 \cdot C_2 H_4 Cl_2$	8
Empirical formula	$C_{25}H_{36}N_4O_{15}Ru_3$	$C_{29}H_{42}Cl_2N_4O_{14}Ru_3$	$C_{23}H_{30}N_4O_{14}Ru_3$
Formula weight	935.79	1044.77	889.72
Temperature (K)	296(2)	296(2)	296(2)
Wavelength (Å)	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic	Orthorhombic
Space group	C2/c	$P2_1/n$	Pnma
Unit cell dimensions	a (Å) = 20.3927(15)	a (Å) = 13.3193(4)	a(Å) = 19.2404(4)
	b (Å) = 11.1794(8)	b (Å) = 22.2592(8)	b (Å) = 19.8237(4)
	c (Å) = 17.9721(13)	c (Å) = 14.4161(5)	c (Å) = 8.2389(2)
	α (°) = 90°	α (°) = 90°	α (°) = 90°
	β (°) = 119.214(2)°	β (°) = 111.062(1)°	$\beta (^{\circ}) = 90^{\circ}$
	γ (°) = 90°	γ (°) = 90°	γ (°) = 90°
Z	4	4	4
Density (mg/m ³)	1.738	1.740	1.881
Absorption coefficient (mm ⁻¹)	1.319	1.321	1.494
F(000)	1864	2088	1760
Crystal size (mm ³)	0.300 x 0.100 x 0.070	0.610 x 0.040 x 0.010	0.140 x 0.060 x 0.030
Theta range for data collection (°)	2.151 to 25.249°	1.769 to 26.404°	2.055 to 25.089°
Index ranges	$\begin{array}{c} -23 \rightarrow h \rightarrow 24, -13 \rightarrow k \rightarrow 13, \\ -21 \rightarrow l \rightarrow 21 \end{array}$	$-16 \rightarrow h \rightarrow 16, -27 \rightarrow k \rightarrow 27,$ $-18 \rightarrow l \rightarrow 17$	$-16 \rightarrow h \rightarrow 22, -20 \rightarrow k \rightarrow 23,$ $-9 \rightarrow 1 \rightarrow 5$
Reflections collected	18964	72486	17509
Independent reflections	3238 [R(int) = 0.0281]	8157 [R(int) = 0.0879]	2884 [R(int) = 0.0393]
Absorption correction	Semi-empirical from	Semi-empirical from	Semi-empirical from
	equivalentes	equivalents	equivalents
T_{max}/T_{min}	0.7454/0.6173	0.7454/0.6950	0.7452/0.6797
Data / restraints / parameters	3238 / 0 / 223	8157 / 0 / 479	2221 / 4 / 227
Goodness-of-fit on F ²	1.201	1.020	1.009
Final R indices [I>2 σ (I)]	$R_1 = 0.0446, wR_2 = 0.0957$	$R_1 = 0.0428, wR_2 = 0.0952$	$R_1 = 0.0275, wR_2 = 0.0553$
Deposition number*	CCDC 1482393	CCDC 1482394	CCDC 1482395

Table S1. Crystallographic and refinement data for $[Ru_3O(CH_3COO)_6(dmpz)_2(CO)] \cdot H_2O$ (1·H₂O), $[Ru_3O(CH_3COO)_6(dmap)_2(CO)] \cdot C_2H_4Cl_2$ (7· $C_2H_4Cl_2$), and $[Ru_3O(CH_3COO)_6(4-ampy)_2(CO)]$ (8).

*Supplementary crystallographic data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.



Figure S1. ORTEP plot of complexes $[Ru_3O(CH_3COO)_6CO(L)_2]$ where (top) L = 2,6-dimethylpyrazine (dmpz) (complex 1), (middle) L = 4-(dimethyl)aminopyridine (dmap) (complex 7), and (bottom) L = 4-aminopyridine (ampy) (complex 8)



Figure S2.¹H NMR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(dmap)_2]$ in CD₃CN at 298 K.



Figure S3. COSY spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(dmap)_2]$ in CD₃CN at 298 K.



Figure S4.¹H NMR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(adpy)₂] in CD₃CN at 298 K.



Figure S5. COSY spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(adpy)₂] in CD₃CN at 298 K.



Figure S6.¹H NMR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(ampy)₂] in CD₃CN at 298 K.



Figure S7. COSY spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(ampy)_2]$ in CD₃CN at 298 K.



Figure S8.¹H NMR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(dmpz)₂] in CD₃CN at 298 K.



Figure S9. COSY spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(dmpz)_2]$ in CD₃CN at 298 K.



Figure S10.¹H NMR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(acpy)_2]$ in CD₃CN at 298 K.



Figure S11. COSY spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(acpy)_2]$ in CD₃CN at 298 K.



Figure S12.¹H NMR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(tbpy)₂] in CD₃CN at 298 K.



Figure S13. COSY spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(tbpy)₂] in CD₃CN at 298 K.



Figure S14. Electronic spectra of compounds of the complexes 1-3 and 6-8 in acetonitrile solutions.



Figure S15. FT-IR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(dmap)_2]$, collected from KBr pellets.



Figure S16. FT-IR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(adpy)_2]$, collected from KBr pellets.



Figure S17. FT-IR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(ampy)_2]$, collected from KBr pellets.



Figure S18. FT-IR spectrum of the complex $[Ru_3O(CH_3COO)_6(CO)(dmpz)_2]$, collected from KBr pellets.



Figure S19. FT-IR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(4-acpy)₂], collected from KBr pellets.



Figure S20. FT-IR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(tbpy)₂], collected from KBr pellets.



Figure S21. FT-IR spectrum of the complex [Ru₃O(CH₃COO)₆(CO)(ampy)₂], collected in dichloromethane solution, in KBr window (spacer = 0.25 mm).

compound	peak	assignment
	(cm ⁻¹)	
[Ru ₃ O(CH ₃ COO) ₆ (CO)(adpy) ₂]	1947s	v(CO)
	1684s	v(C=O) adpy
	1603s	V _{as} (COO) Ac
	1566s	$v_{as}(COO)$ Ac
	1421s	v _s (COO) Ac
	1347w	$\delta(CH_3)$ Ac
	1218w	δ(CH) py
	1064w	v(ring)py
	1029w	δ(CH) Ac
	769w	π (NH) adpy
	687m	δ(OCO)
	624m	π(COO) Ac
$[Ru_3O(CH_3COO)_6(CO)(ampy)_2]$	3377m	v(NH ₂) ampy
	1920s	v(CO)
	1639s	$\delta(NH_2)$ ampy
	1617s	v _{as} (COO) Ac
	1575s	v _{as} (COO) Ac
	1420s	$v_{s}(COO) Ac$
	1347	$\delta(CH_3)$ Ac
	1215m	δ(CH) py
	1060w	V(ring) py
	1024S	<u> 0(CH) AC</u> δ(OCO)
	622w	======================================
$[Ru_{2}O(CH_{2}COO)_{2}(CO)(dman)_{2}]$	1927s	x(CO)
[Ku ₃ 0(CH ₃ COO) ₆ (CO)(uniap) ₂]	16118	v(CO0) Ac
	1570	
	1579s	$V_{as}(COO)$ Ac
	14238	$V_{s}(COU)$ Ac
	1340Sh 1227a	o(CH ₃) Ac
	12278 1067m	$\delta(CH)$ py
	1007m	$\delta(CH) \Delta c$
	688w	$\delta(OCO)$ Ac
	624w	$\pi(COO)$ Ac
$[Ru_3O(CH_3COO)_6(dmpz)_2(CO)]$	1945s	v(CO)
	1609s	v _{as} (COO) Ac
	1569s	V _{as} (COO) Ac
	1420s	v _s (COO) Ac
	1348w	δ(CH ₃)Ac
	1253w	δ(ring in plane) dmpz
	1032w	δ(CH) Ac
	772w	δ(CH) dmpz
	732w	δ(ring) dmpz
	688m	δ(OCO)
	625m	π(COO)Ac
$\left[Ru_{3}O(CH_{3}COO)_{6}(CO)(4-acpy)_{2}(CO) \right]$	1940s	v(CO)
	1608s	v _{as} (COO) Ac
	1569s	v _{as} (COO) Ac

Table S2: Tentative assignment of the more relevant peaks observed in the infrared spectra, obtained from KBr pellets.

	1420s	v _s (COO) Ac
	1347w	δ(CH ₃) Ac
	1361w	δ(CH) py
	1058w	v(ring py)
	1024w	δ(CH) Ac
	683m	δ(OCO)
	623m	π(COO)Ac
	1271s	v(CH ₃) 4-acpy
	1361s	δ _s (CH ₃) 4-acpy
	1694s	v(C=O) 4-acpy
[Ru ₃ O(CH ₃ COO) ₆ (4-tbpy) ₂ (CO)]	1948s	v(CO)
	1613s	v _{as} (COO) Ac
	1574s	v _{as} (COO) Ac
	1421s	v _s COO) Ac
	1346w	δ(CH ₃) Ac
	1273w	δ(CH) py
	1071w	v(ring) py
	1030w	δ(CH) Ac
	686m	δ(OCO)
	623m	π(COO)Ac
	2963s	$\delta_s(CH_3)$ 4-tbpy
	2869s-	v (CH ₃) 4-tbpy

v =stretching; $\delta =$ bending; π : rocking; s =symmetrical; as = assimetrycal; s =strong; m =médium; w =weak; sh = shoulder.



Figure S22. Cyclic voltammogram of compound [Ru₃O(CH₃COO)₆(CO)(dmap)₂] in 0,1 mol dm⁻³ acetonitrile solution TBAPF₆. Scan rate 100 mV s⁻¹.



Figure S23. Cyclic voltammogram of compound [Ru₃O(CH₃COO)₆(CO)(adpy)₂] in 0,1 mol dm⁻³ acetonitrile solution TBAPF₆. Scan rate 100 mV s⁻¹.



Figure S24. Cyclic voltammogram of compound [Ru₃O(CH₃COO)₆(CO)(dmpz)₂] in 0,1 mol dm⁻³ acetonitrile solution TBAPF₆. Scan rate 100 mV s⁻¹.



Figure S25. Cyclic voltammogram of compound [Ru₃O(CH₃COO)₆(CO)(4-acpy)₂] in 0,1 mol dm⁻³ acetonitrile solution TBAPF₆. Scan rate 100 mV s⁻¹.



Figure S26. Cyclic voltammogram of compound [Ru₃O(CH₃COO)₆(CO)(tbpy)₂] in 0,1 mol dm⁻³ acetonitrile solution TBAPF₆. Scan rate 100 mV s⁻¹.



Figure S27. Absorption spectra of complex $[Ru_3O(CH_3COO)_6(CO)(dmap)_2]$ during photolysis at $\lambda_{exc} = 377$ nm, collected every 5 minutes from acetonitrile solutions (total time of irradiation = 80 minutes)



Figure S28. Absorption spectra of complex $[Ru_3O(CH_3COO)_6(CO)(adpy)_2]$ during photolysis at $\lambda_{exc} = 377$ nm, collected every 5 minutes from acetonitrile solutions (total time of irradiation = 80 minutes)



Figure S29. Absorption spectra of complex $[Ru_3O(CH_3COO)_6(CO)(ampy)_2]$ during photolysis at $\lambda_{exc} = 377$ nm, collected every 5 minutes from acetonitrile solutions (total time of irradiation = 80 minutes)



Figure S30. Absorption spectra of complex $[Ru_3O(CH_3COO)_6(CO)(4-acpy)_2]$ during photolysis at $\lambda_{exc} = 377$ nm, collected every 5 minutes from acetonitrile solutions (total time of irradiation = 80 minutes)



Figure S31. Absorption spectra of complex $[Ru_3O(CH_3COO)_6(CO)(tbpy)_2]$ during photolysis at $\lambda_{exc} = 377$ nm, collected every 5 minutes from acetonitrile solutions (total time of irradiation = 80 minutes)