

Supporting Information

For

N-Heterocyclic Olefins as Ancillary Ligands in Catalysis: A Study of their Behaviour in Transfer Hydrogenation Reactions

*A. Iturmendi,^a N. García,^b E. A. Jaseer,^b J. Munárriz,^c P. J. Sanz Miguel,^a V. Polo,^c M. Iglesias,^{*a} and L. A. Oro^{*a,b}*

^aDepartamento de Química Inorgánica – Instituto de Síntesis Química y Catálisis Homogénea-ISQCH, Universidad de Zaragoza – CSIC. C/ Pedro Cerbuna 12, 50009 Zaragoza, Spain.

^c King Fahd University of Petroleum & Minerals (KFUPM)
Dhahran 31261 (Saudi Arabia)

^c Departamento Química Física - Instituto de Biocomputación y Física de Sistemas Complejos (BIFI)
Universidad de Zaragoza (Spain), C/ Pedro Cerbuna 12, 50009 Zaragoza

E-mail: miglesia@unizar.es and oro@unizar.es

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Plots of conversion vs. time

Table 1 Entry 1

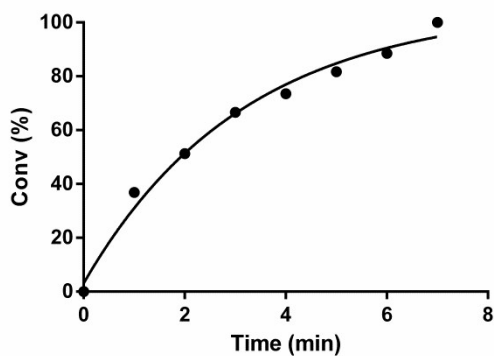


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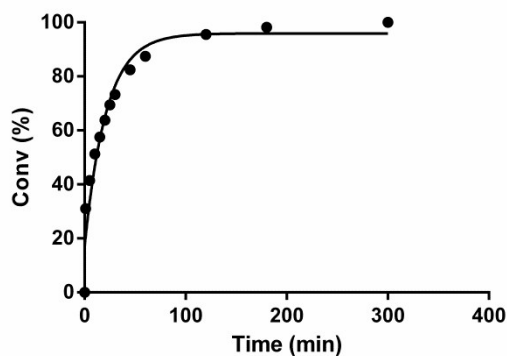


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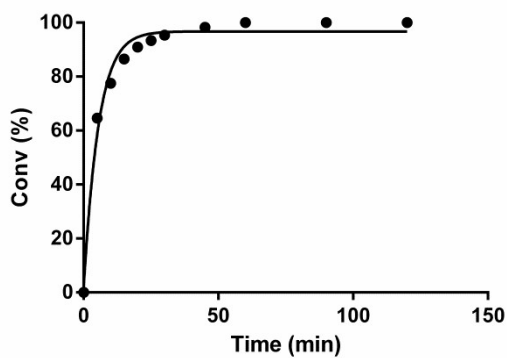


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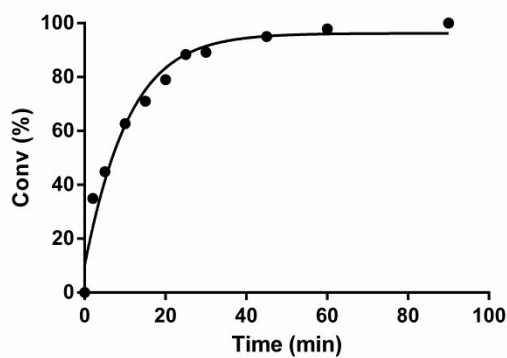


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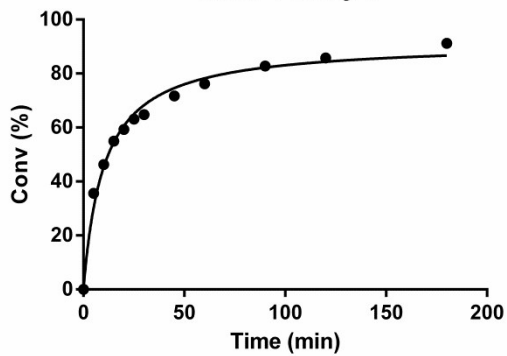


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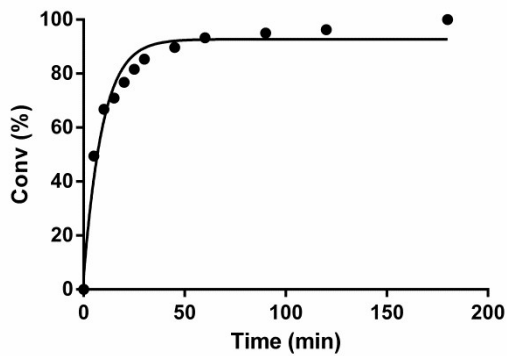


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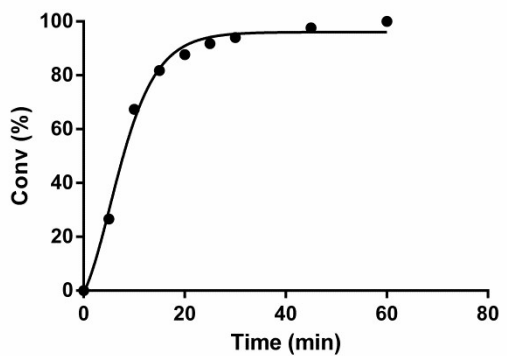
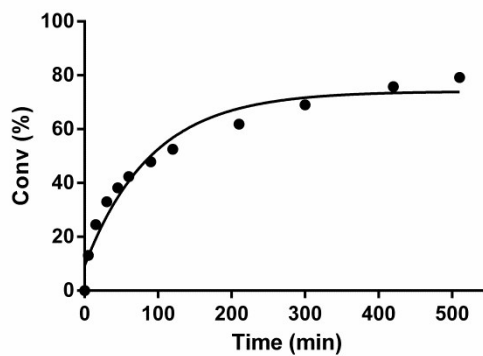


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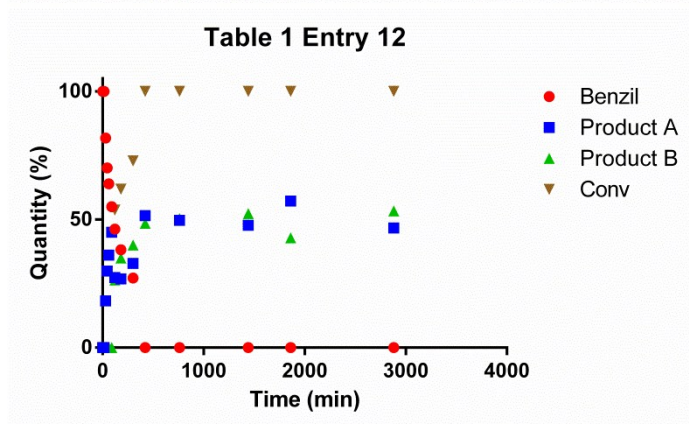
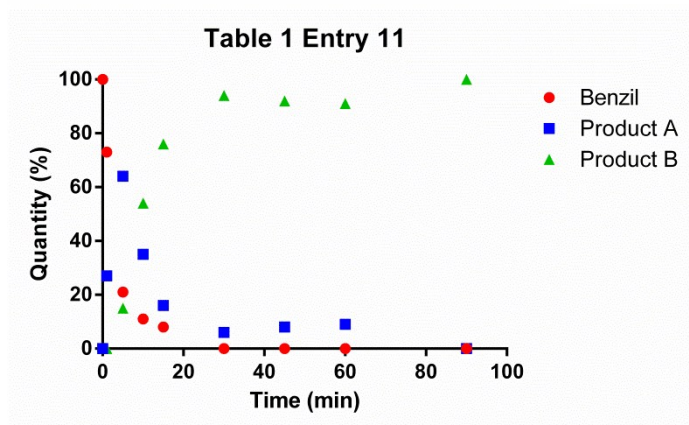
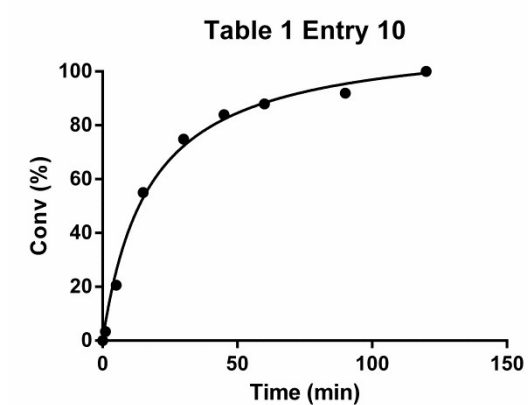
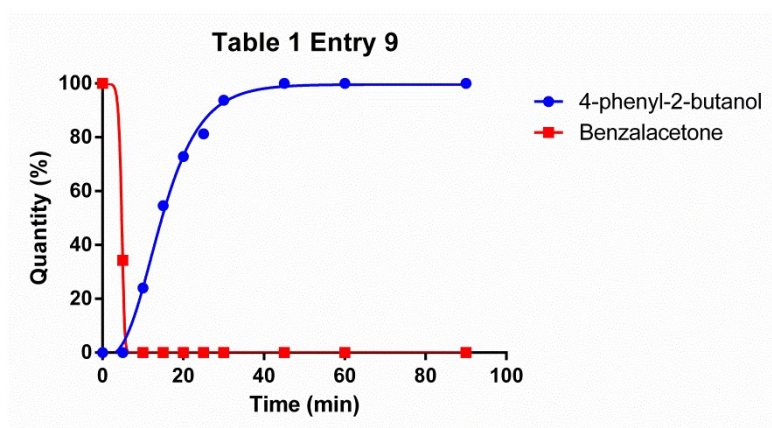


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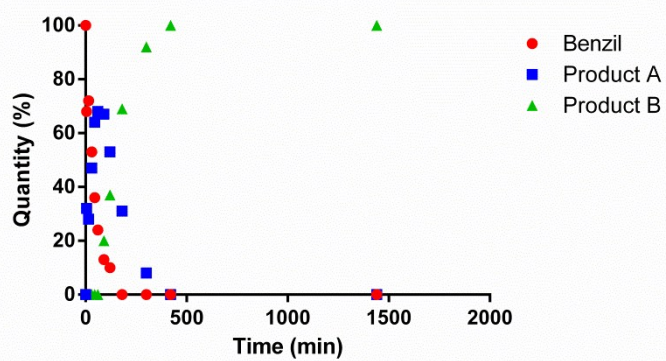


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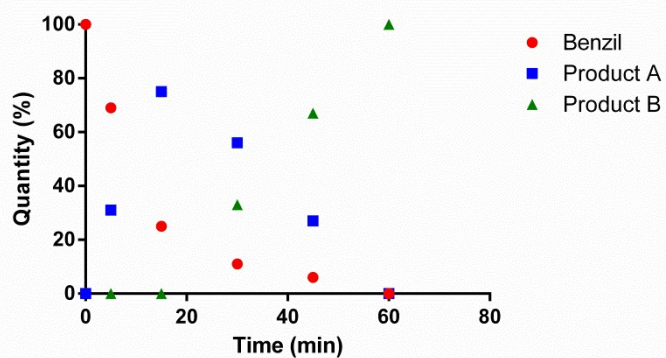


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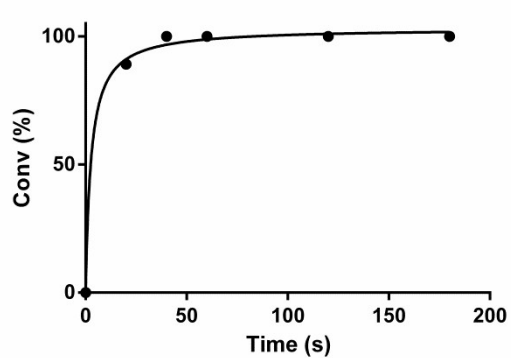


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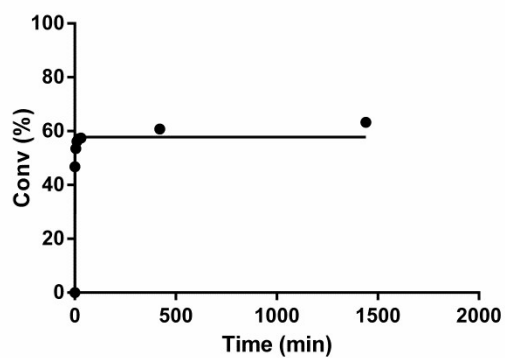


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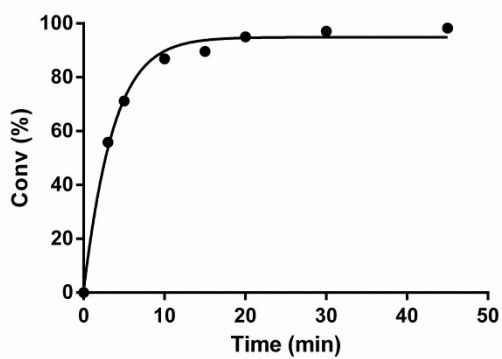


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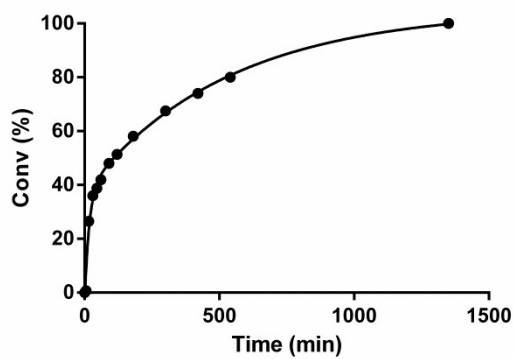


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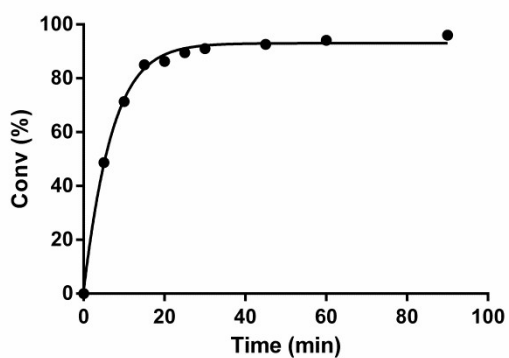


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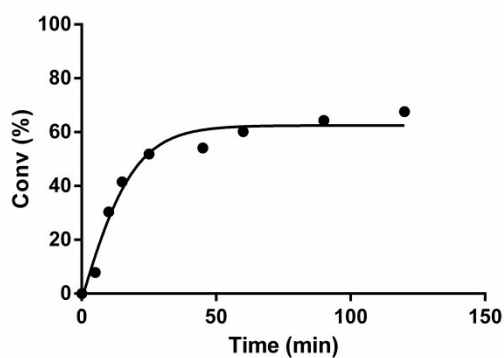


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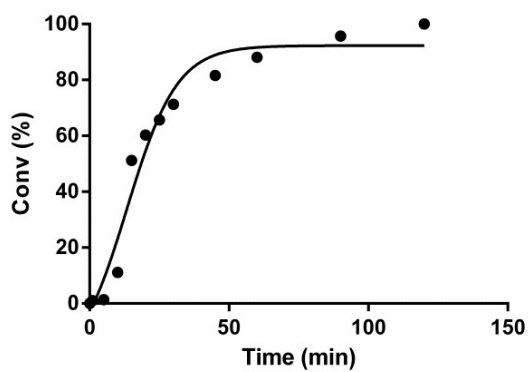


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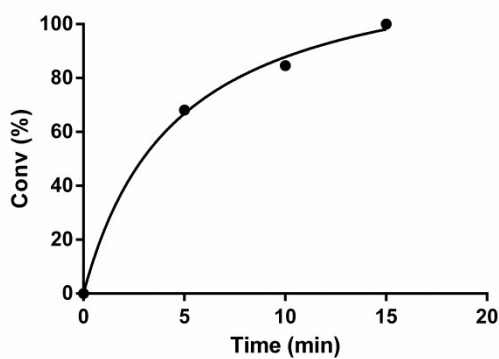


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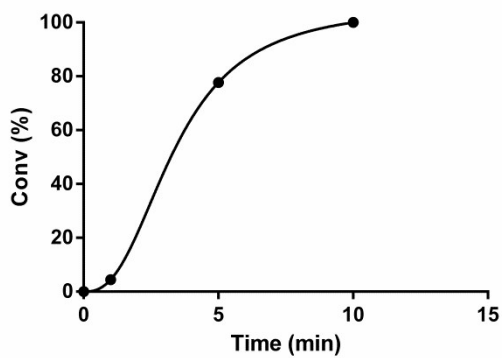


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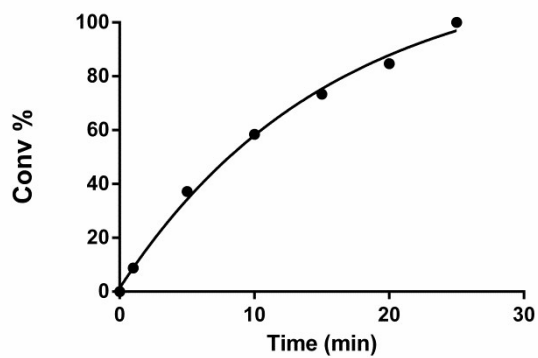


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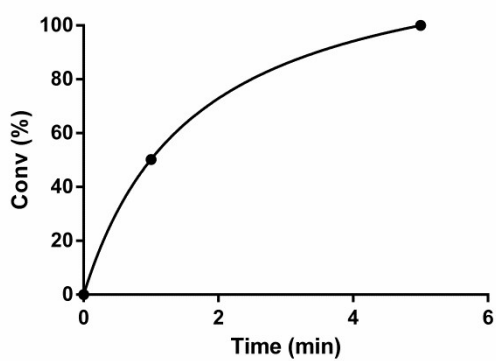


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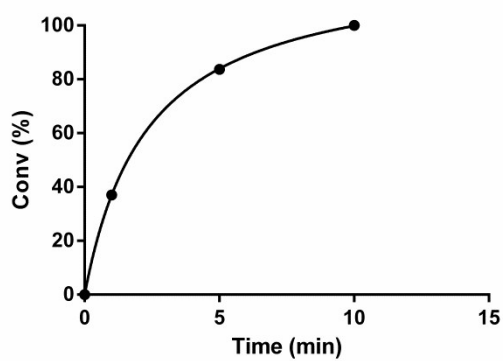


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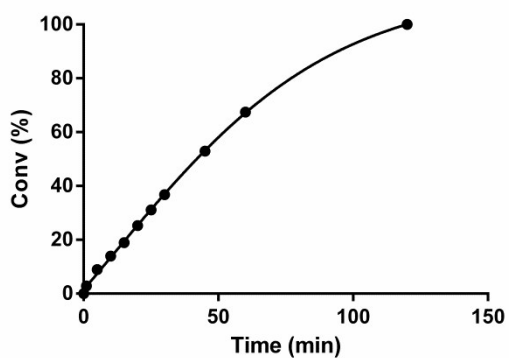


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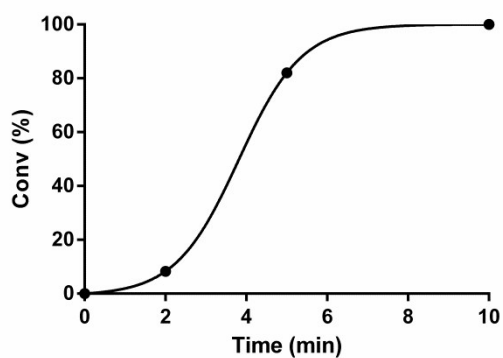


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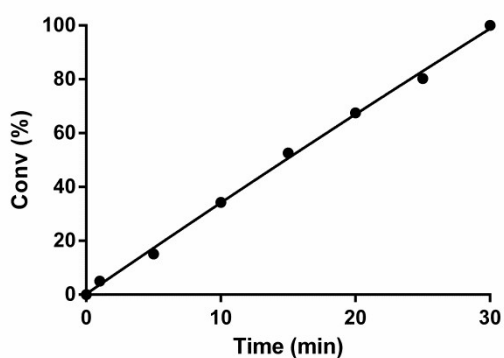


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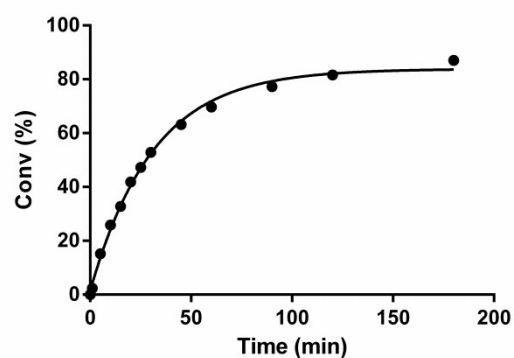


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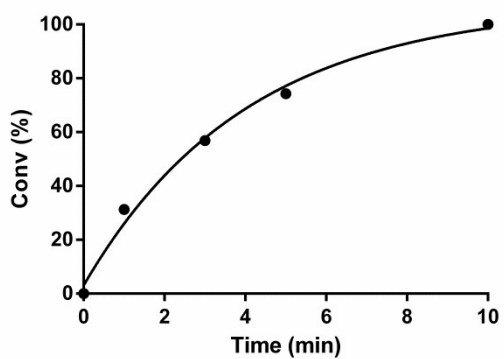
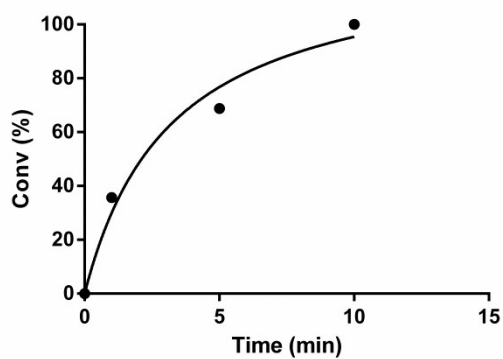
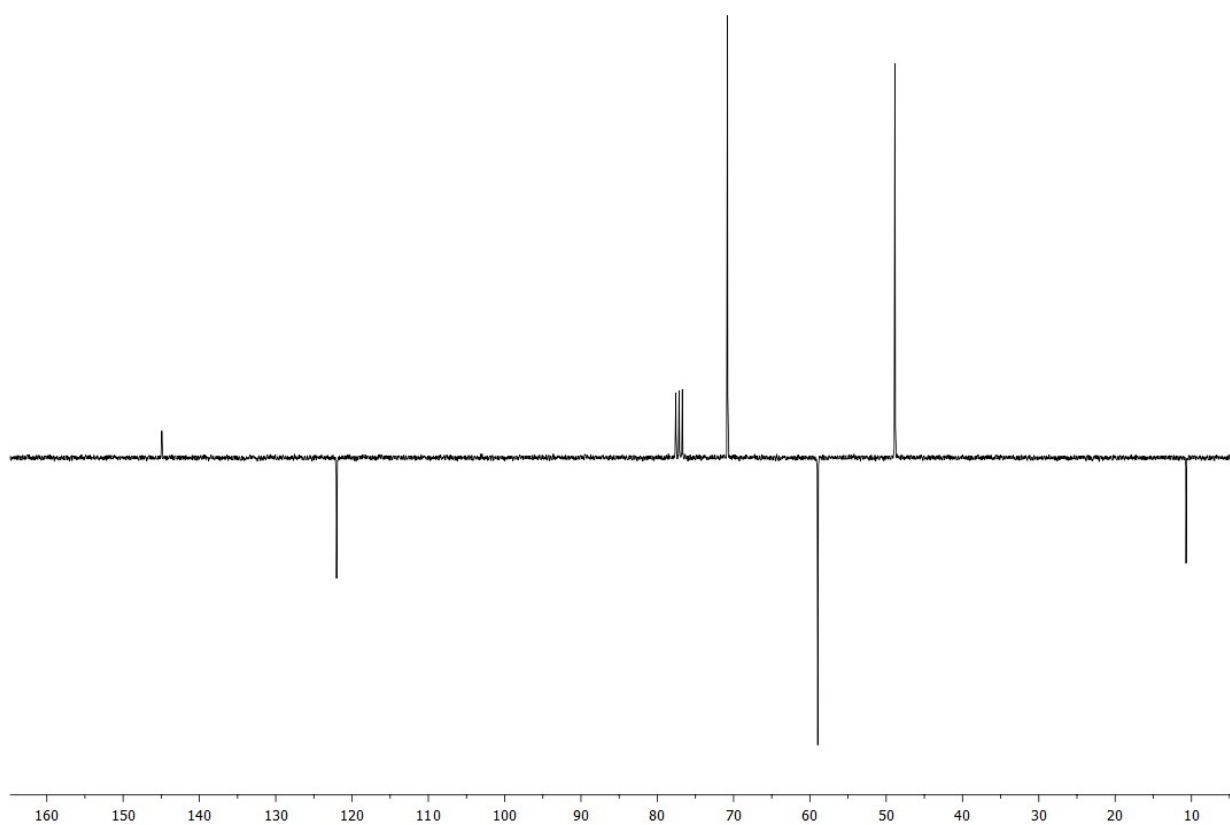
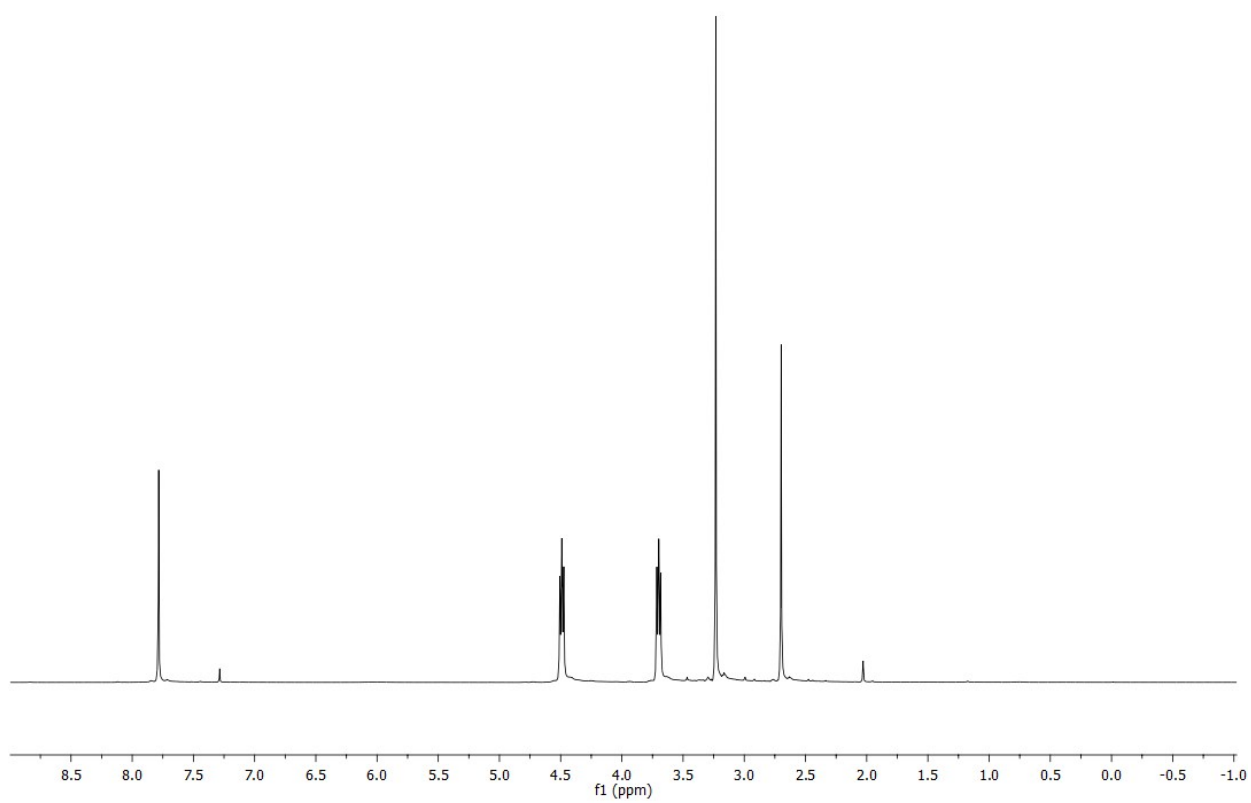


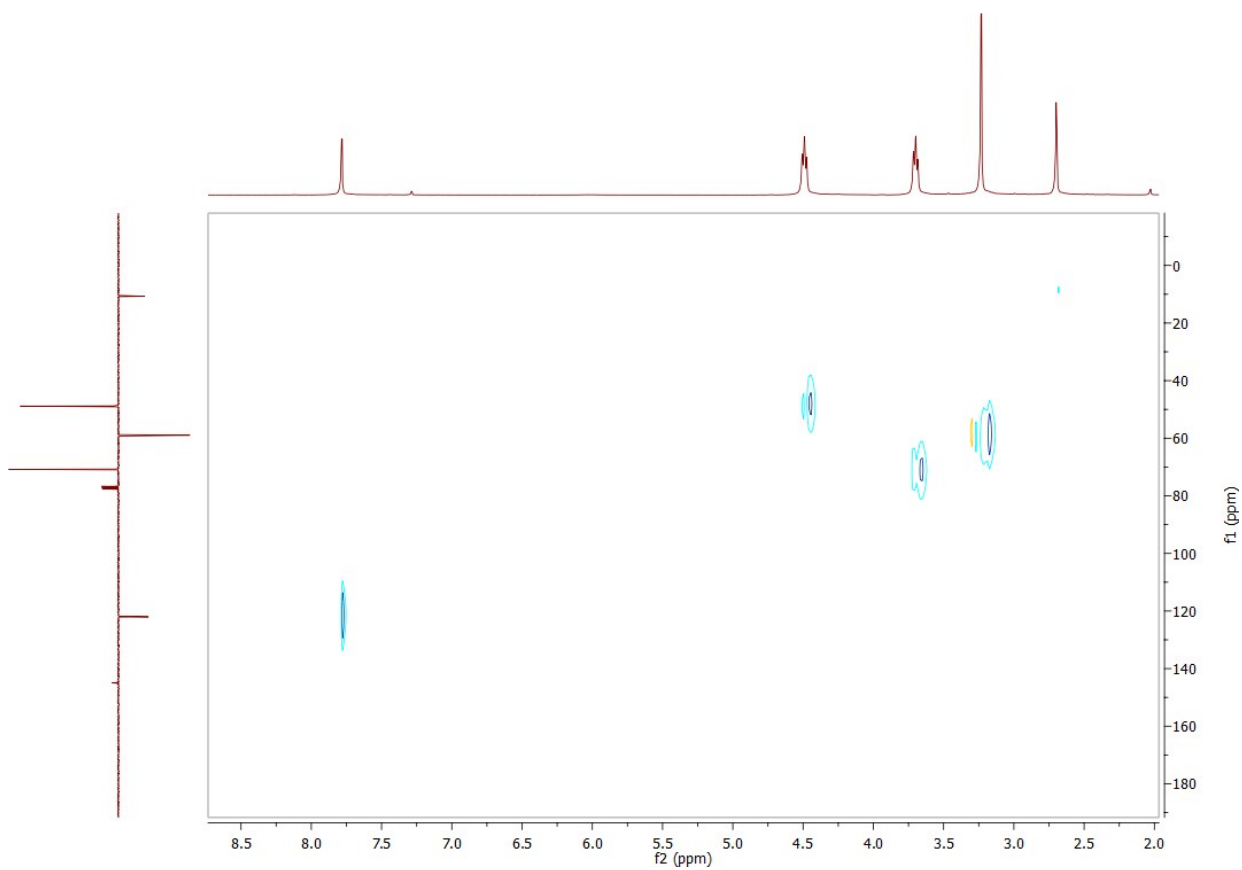
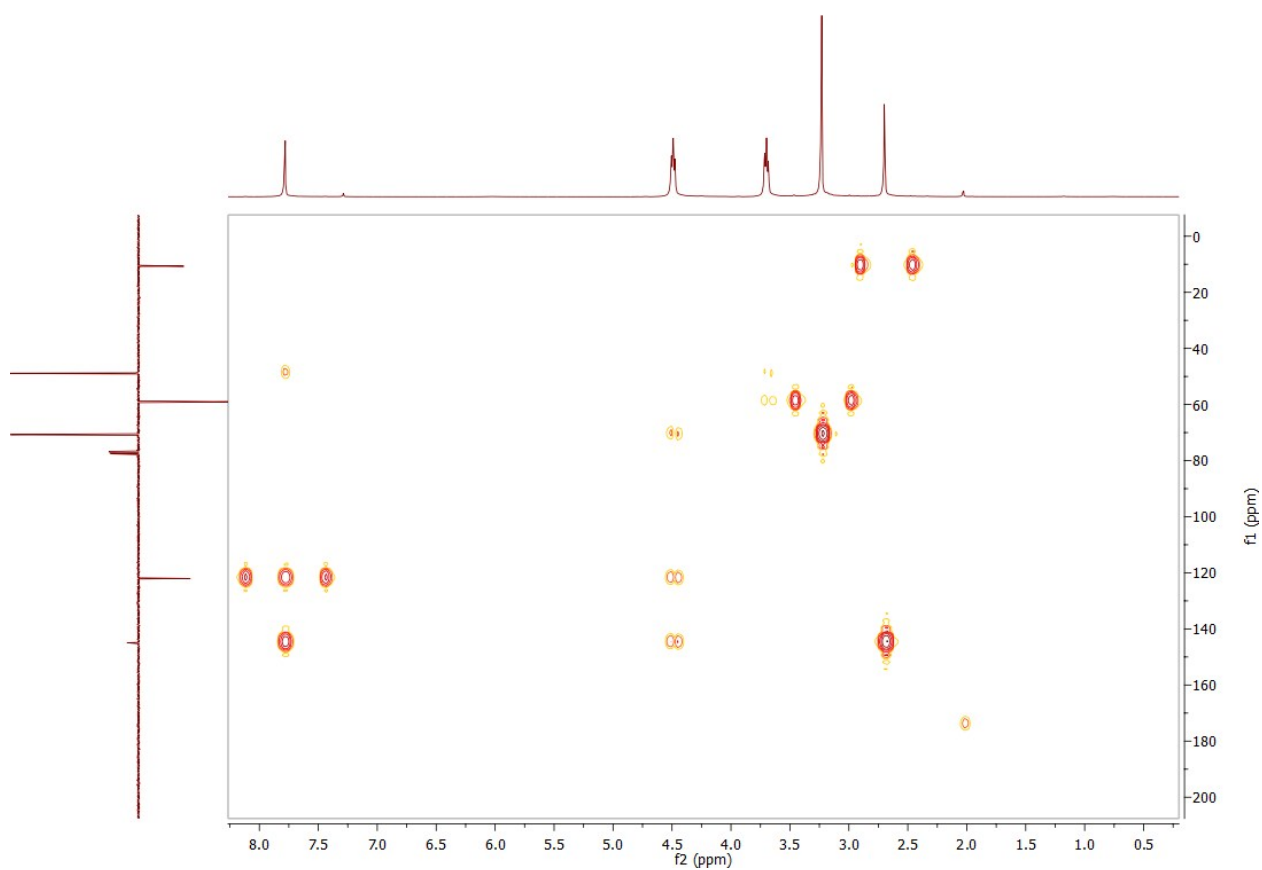
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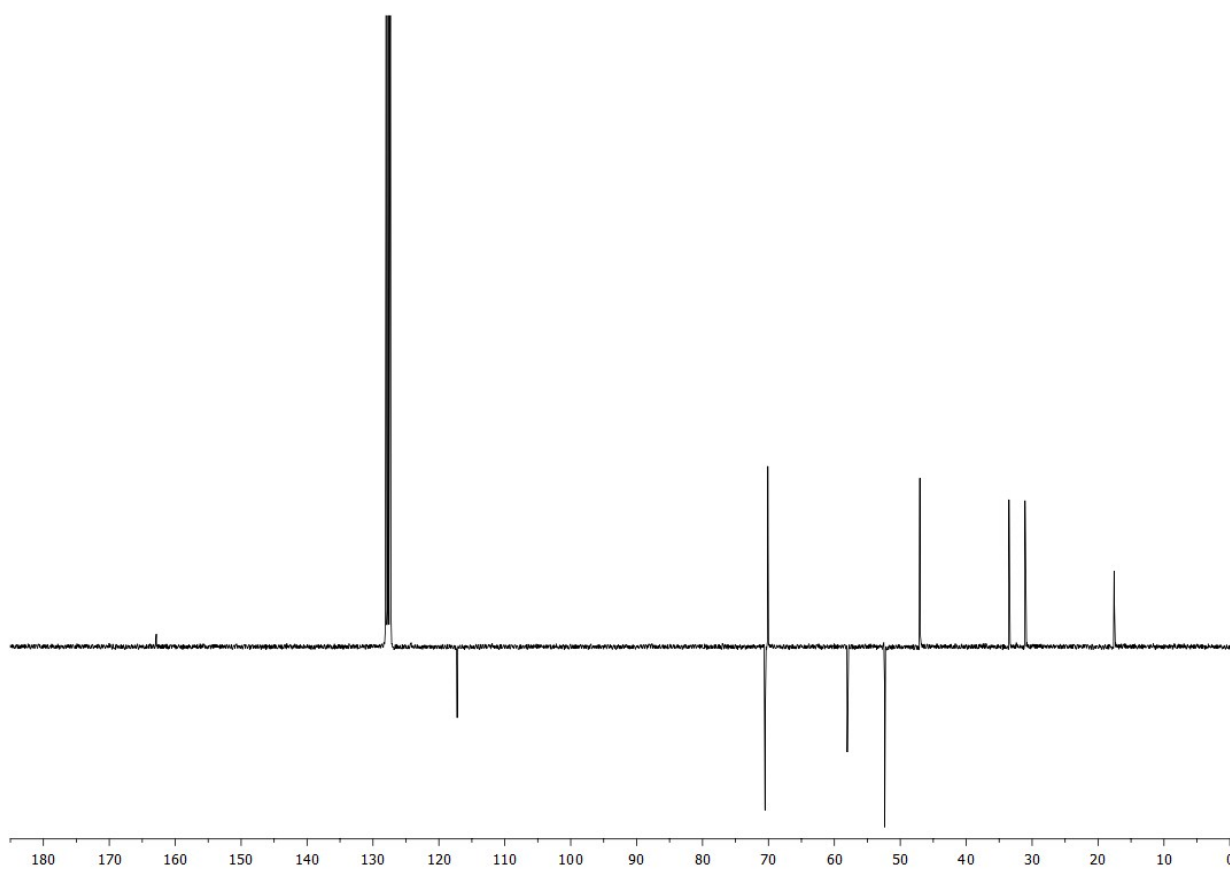
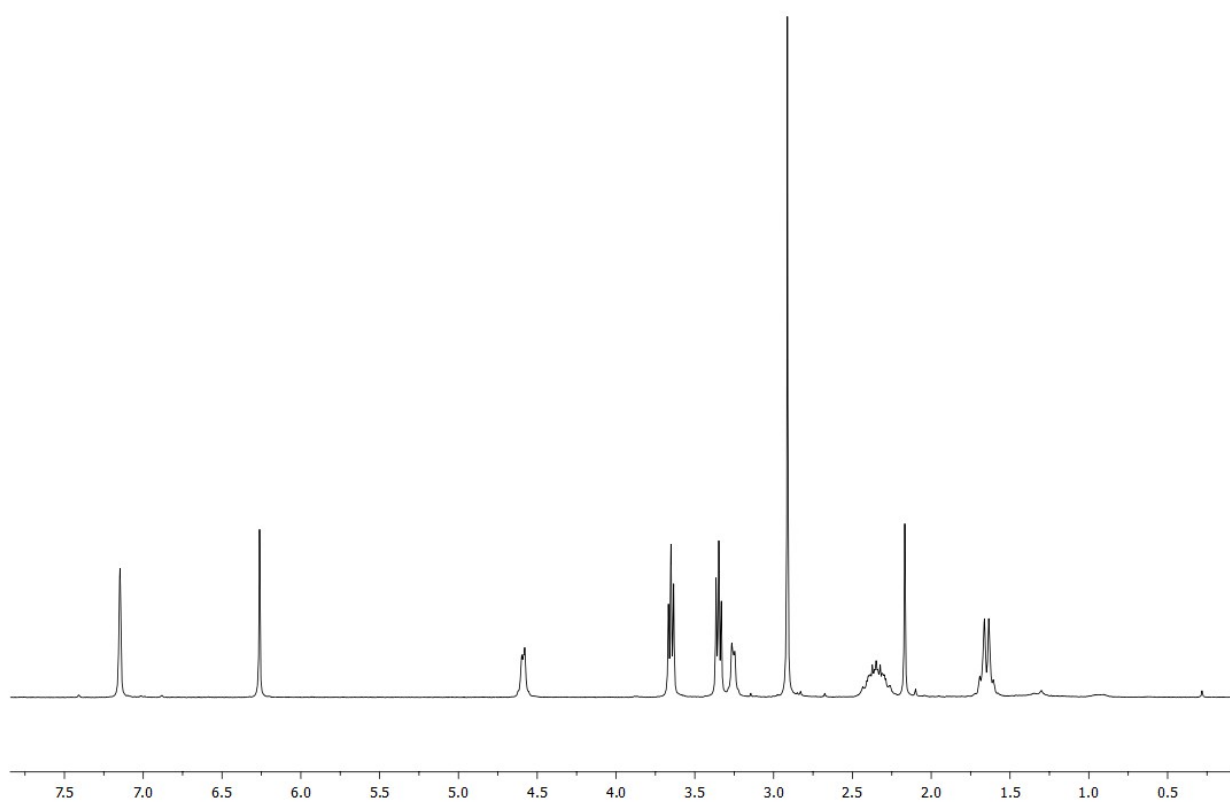
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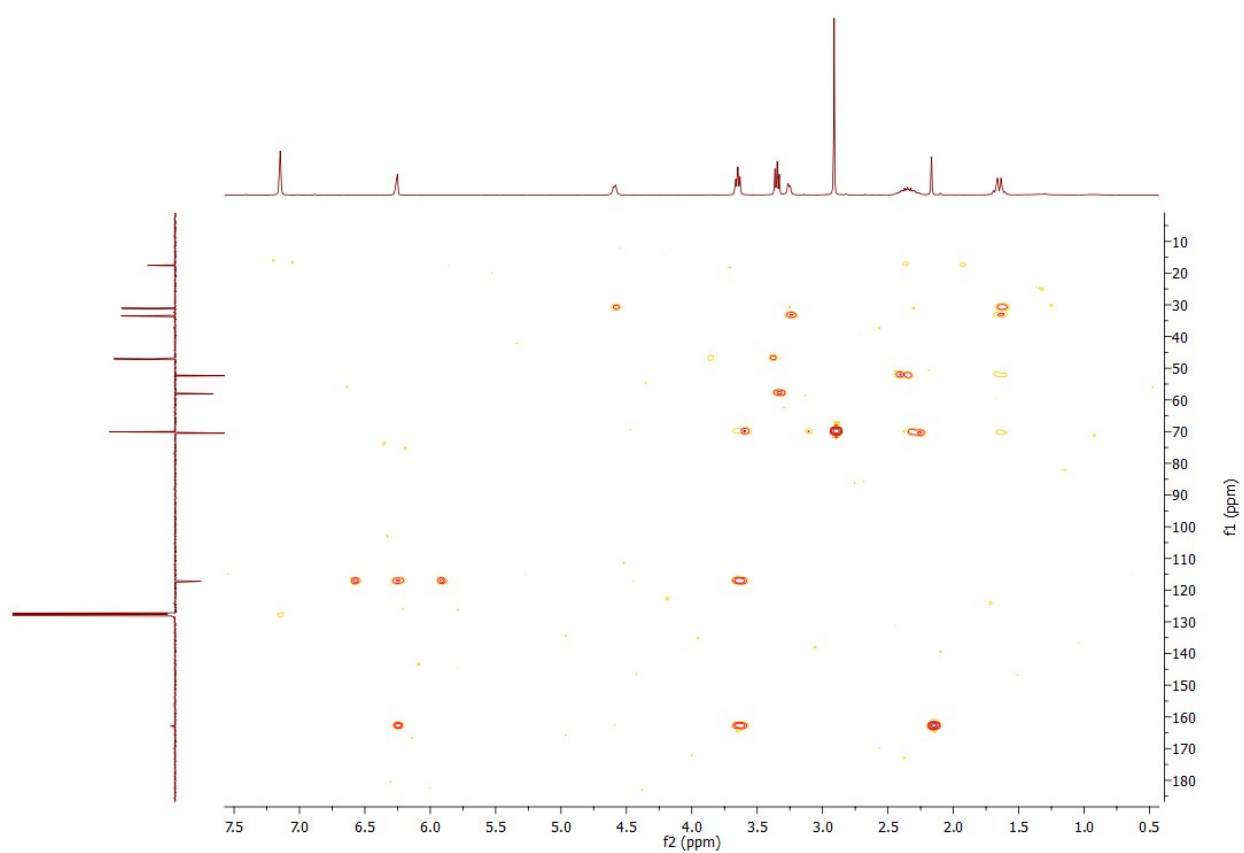
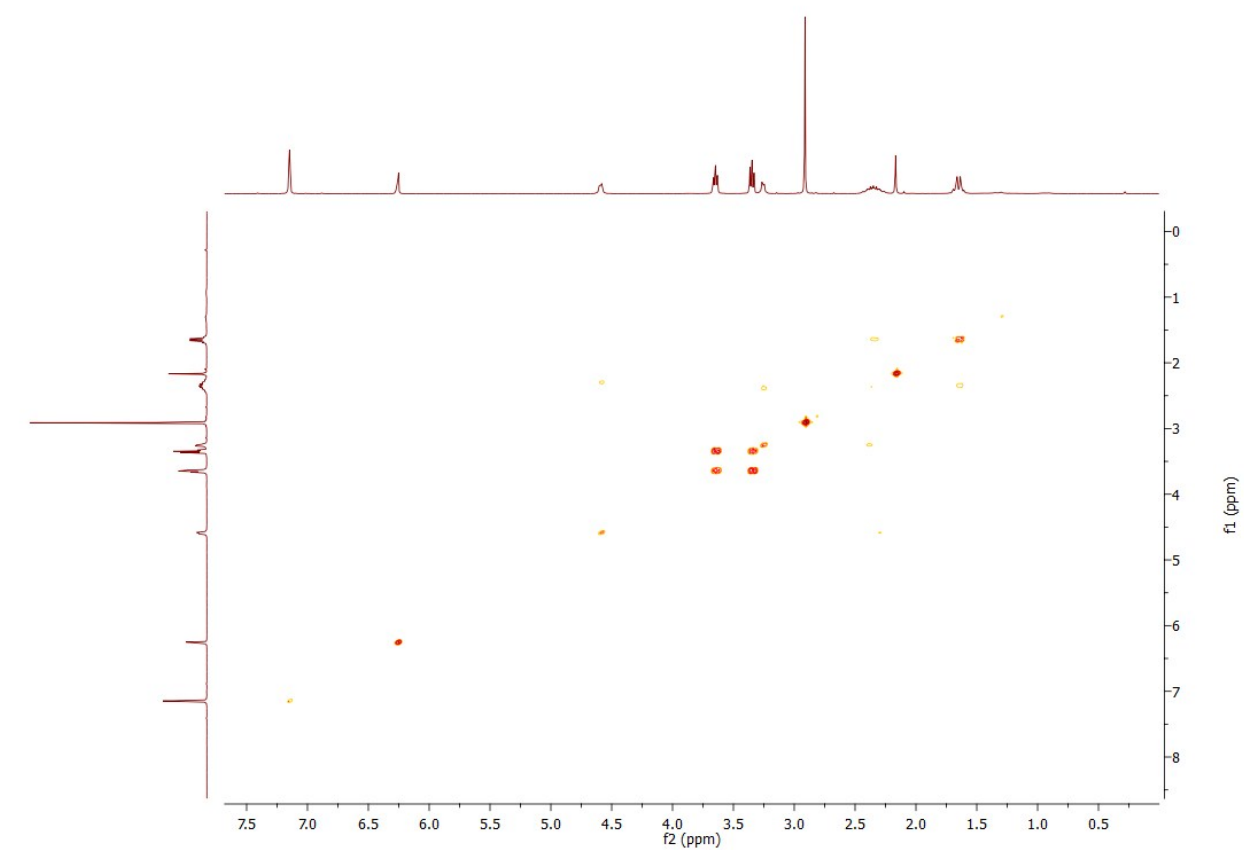
1,3-bis(2-(methoxyethyl)-2-methylimidazolium chloride (4)

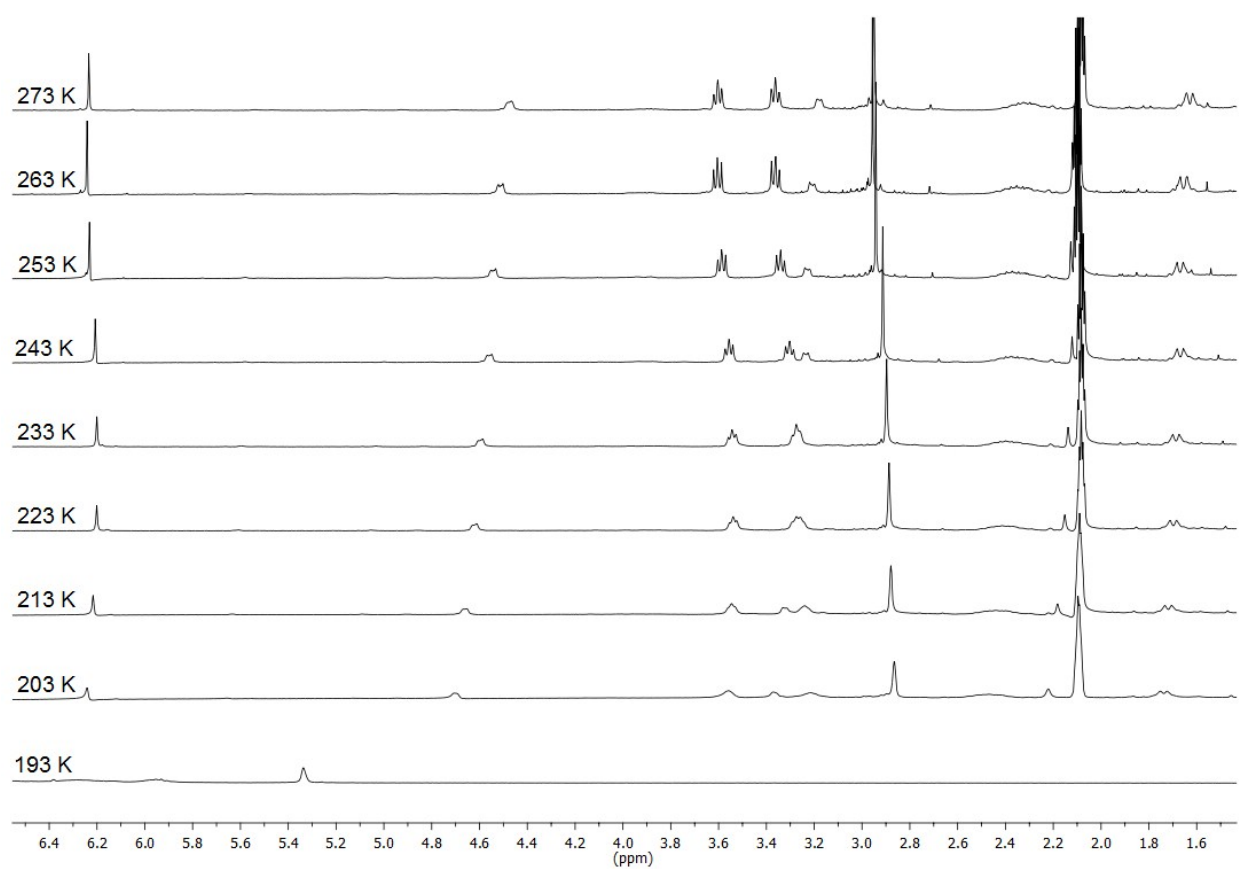
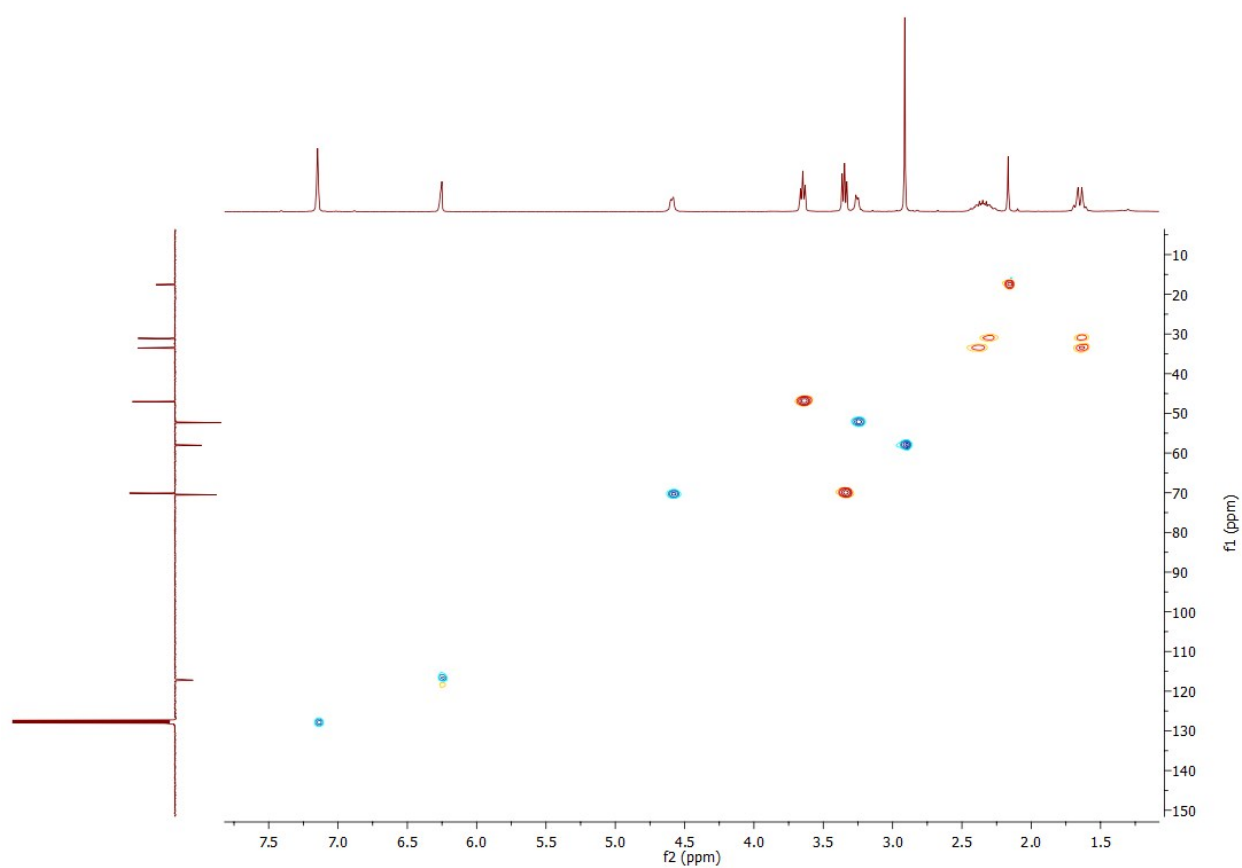




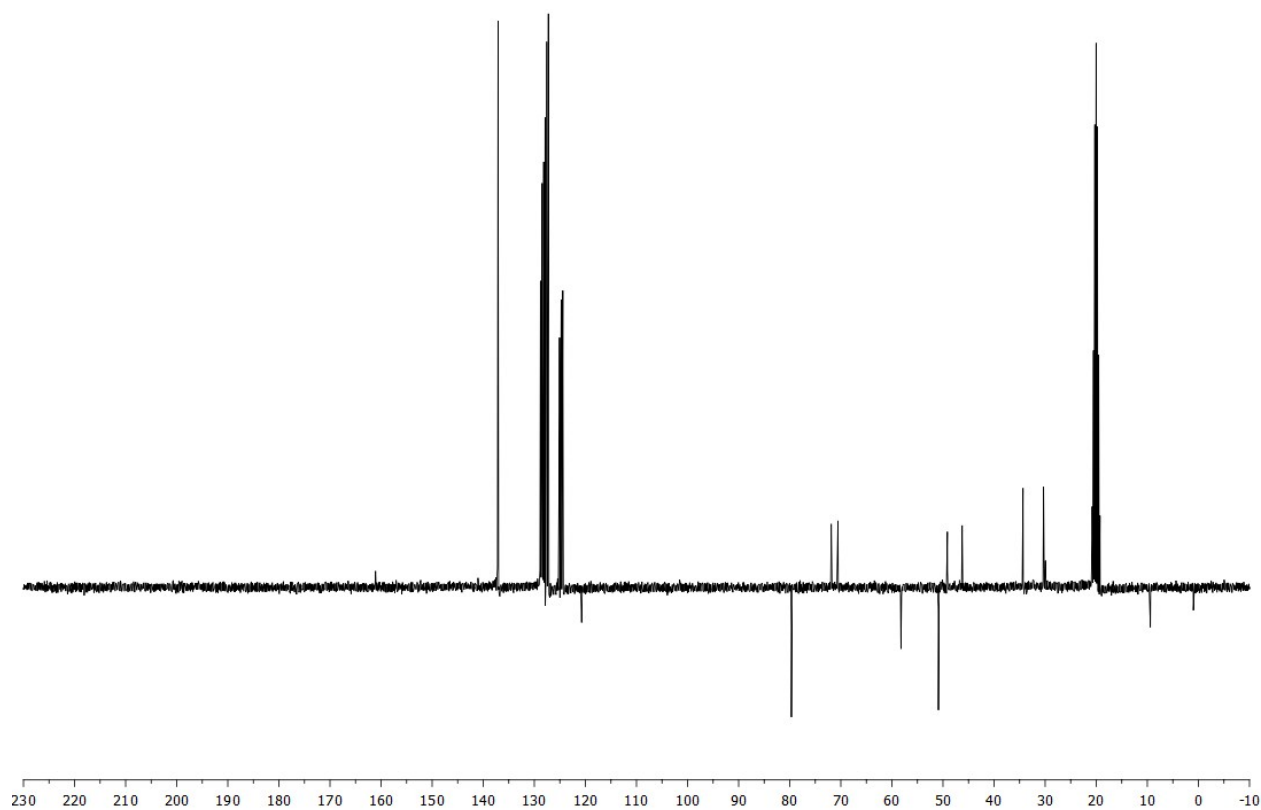
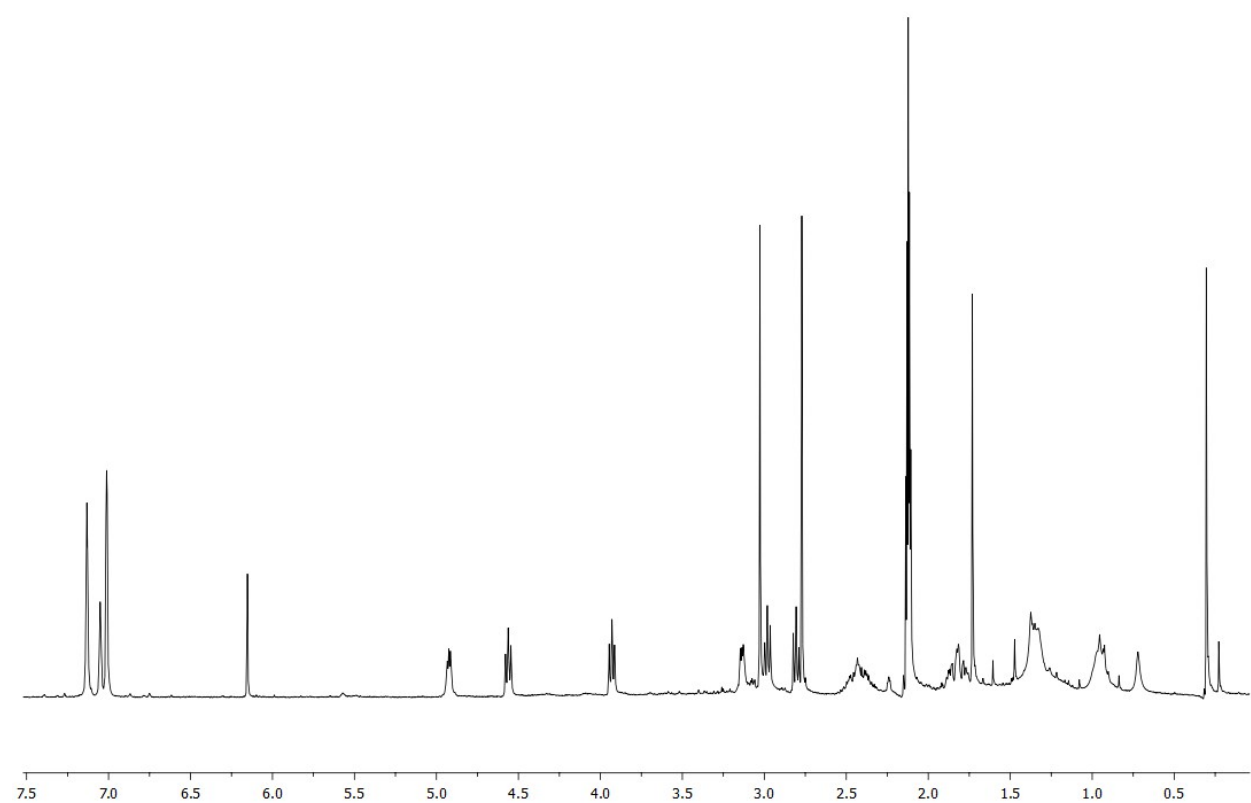
Complex 5

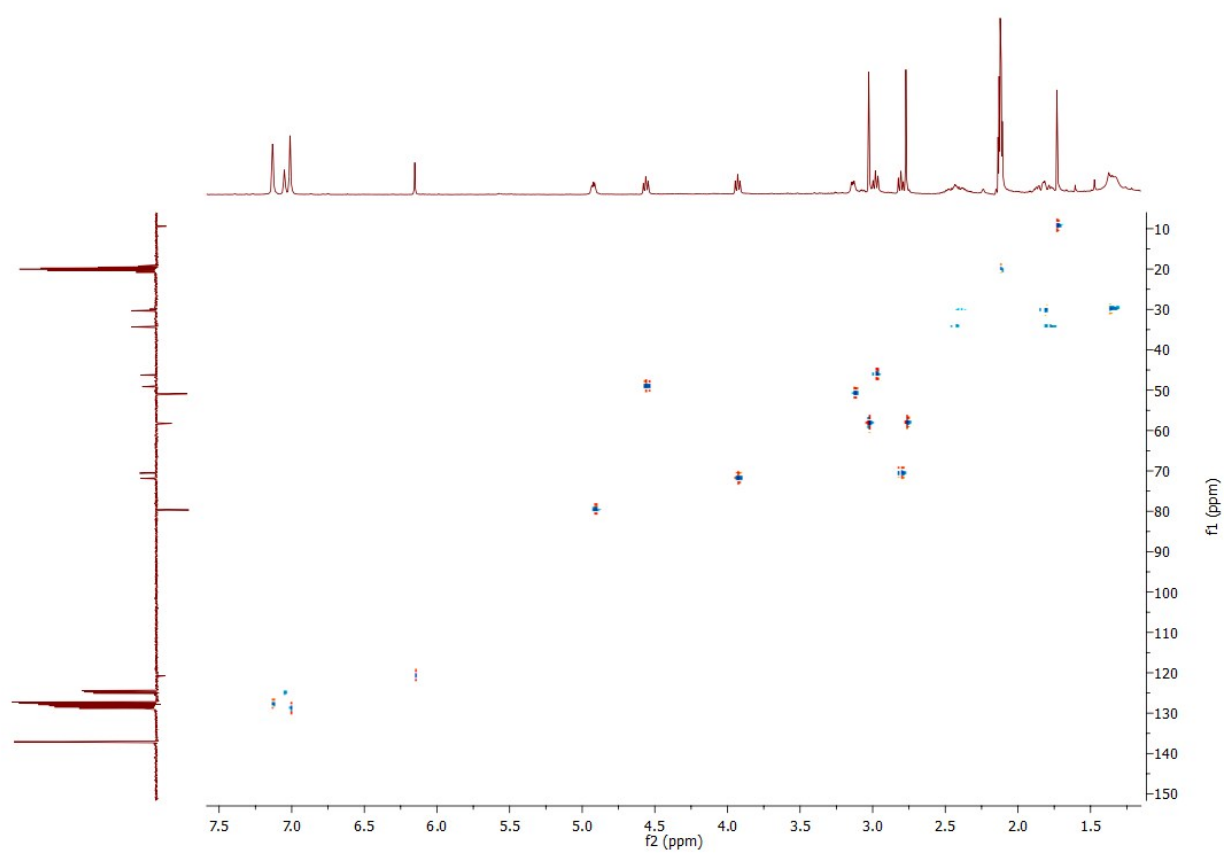
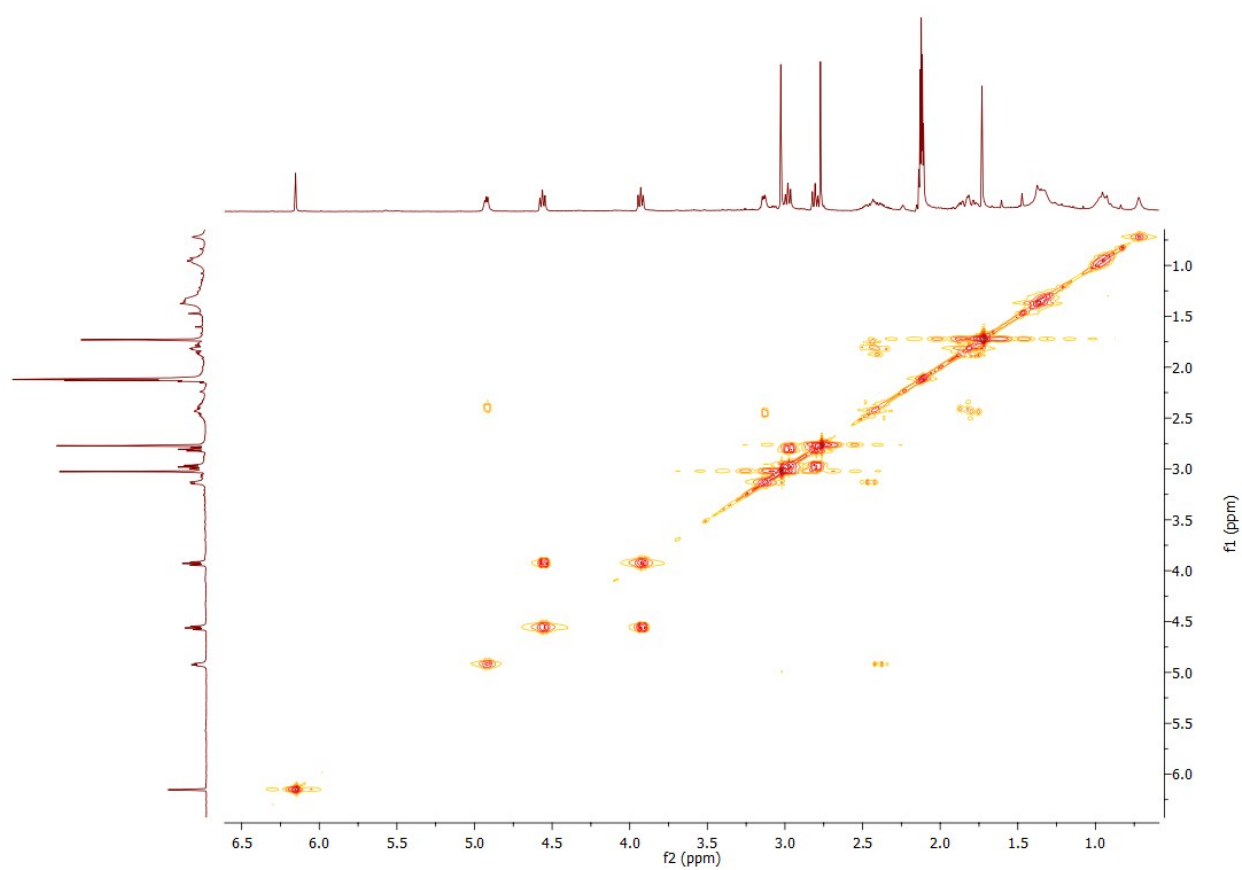


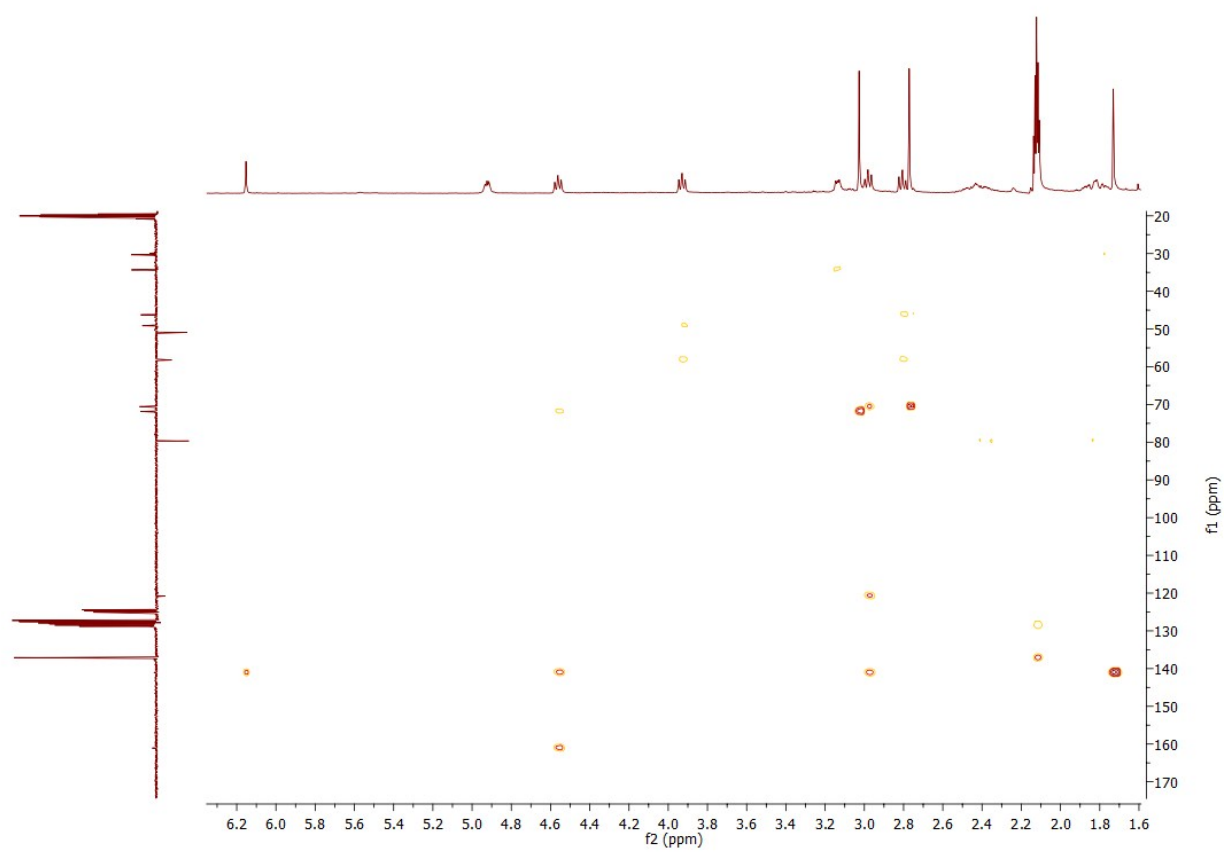




Complex 6







Crystal structure analysis: X-ray diffraction data for compound **3** was recorded on an APEX-II diffractometer equipped with an area detector and graphite monochromated Mo K α radiation (λ = 0.71073 Å). Data reduction and cell refinement were carried out using the APEX2 software.^[S1] Salient structures were found by direct methods and refined by full-matrix least-squares methods based on F^2 using the SHELXL-97 and WinGX software.^[S2] All non-hydrogen atoms were refined anisotropically, and all the hydrogen atoms were included in geometrically calculated positions and refined with isotropic displacement parameters according to the riding model. All geometrical calculations were performed using the SHELXL-97 and WinGX programs.^[S2]

Crystal data for compound **3**: [C₈₄H₁₀₀Cl₁₀Ir₂N₄O₂P₄], triclinic, $P\bar{1}$, a = 13.9013(7) Å, b = 17.4906(9) Å, c = 20.0339(11) Å, α = 93.6950(10)°, β = 105.6320(10)°, γ = 110.4170(10)°, Z = 2, M_r = 2060.46, V = 4327.9(4) Å³, D_{calcd} = 1.581 g cm⁻³, $\lambda(\text{Mo K}\alpha)$ = 0.71073 Å, T = 100 K, μ = 3.502 mm⁻¹, 52738 reflections collected, 20117 unique (R_{int} = 0.0355), 16485 observed, $R1(F_o)$ = 0.0576 [$I > 2\sigma(I)$], $wR2(F_o^2)$ = 0.1628 (all data), GOF = 1.043.

[S1] APEX2 Bruker AXS Inc., Madison, Wisconsin, USA, 2011.

[S2] a) G. M. Sheldrick, SHELXS-97 and SHELXL-97; University of Göttingen, Germany, 1997; b) L. J. Farrugia, WinGX; University of Glasgow, Great Britain, 1998.

Computational Details

Table S1. Absolute (a.u.) and relative (kcal/mol, to isolated molecules and **A** or **B** for Figure 3 or Figure 4, respectively) energies of all B3LYP-D3 calculated structures using def2-SVP (DZ) basis set and solvent PCM (2-propanol) for geometry optimization and Gibbs free energy corrections at standard conditions. Basis set corrections using the def2-TZVP (TZ) have been calculated by single point (SP) energy calculations.

	E_PCM_DZ	E_PCM_TZ(SP)	G_PCM_TZ	DE_PCM_DZ	DE_PCM_TZ	DG_PCM_TZ
A	-1873.41739	-1874.86562	-1874.30334	0.0	0.0	0.0
A1	-1873.37465	-1874.81989	-1874.25573	26.8	28.7	29.9
A2	-1873.39238	-1874.83779	-1874.26976	15.7	17.5	21.1
A3	-1873.40803	-1874.85874	-1874.30041	5.9	4.3	1.8
B	-1561.55510	-1562.66835	-1562.27941	0.0	0.0	0.0
TSB/C	-1561.50791	-1562.62694	-1562.24390	29.6	26.0	22.3
C	-1561.55529	-1562.67228	-1562.28932	-0.1	-2.5	-6.2
D	-1753.18277	-1754.50216	-1754.06960	-1.3	-4.9	-9.0
TSD/E	-1753.13375	-1754.45376	-1754.02339	29.4	25.4	20.0
TSB/E	-1946.17956	-1947.71309	-1947.19878	18.6	22.2	34.9
E	-1753.18151	-1754.49581	-1754.05841	-0.5	-1.0	-2.0
TSE/F	-1947.41159	-1948.94689	-1948.40994	0.0	3.9	15.2
F+B	-385.85201	-386.28134	-386.15383	2.8	2.2	0.6
cod	-311.83973	-312.17955	-312.03237			
Acetone	-193.02861	-193.25413	-193.19909			
Phenylacetone	-384.65418	-385.08007	-384.97498			
isopropanol	-194.23093	-194.45884	-194.37891			

Table S2. Absolute (a.u.) and relative (kcal/mol, to **A** and isolated molecules) energies of all B3LYP-D3 calculated structures using def2-SVP (DZ) basis set for geometry optimization and Gibbs free energy corrections and def2-TZVP (TZ) combined with PCM solvent corrections (2-propanol) using single point energy calculations.

	E_DZ	E_TZ_PCM	G_TZ_PCM	DE_DZ	DE_TZ_PCM	DG_TZ_PCM
A	-1873.40216	-1874.86467	-1874.29945	0.0	0.0	0.0
TSA/B'	-1873.34366	-1874.80681	-1874.25018	36.7	36.3	30.9
B'	-1873.40620	-1874.87047	-1874.31506	-2.5	-3.6	-9.8
C'	-2065.03321	-2066.69969	-2066.09561	-3.7	-5.6	-13.5
TSC/D'	-2064.96850	-2066.63468	-2066.02931	36.9	35.2	28.1
D'	-2065.02996	-2066.69515	-2066.08108	-1.6	-2.8	-4.4
TSD/E'	-1947.39768	-1948.94649	-1948.41051	9.7	13.5	5.1
E'+B	-385.84642	-386.28129	-386.15358	1.9	1.0	-1.4
cod	-311.83838	-312.17959	-312.03210			
Acetone	-193.02324	-193.25397	-193.20008			
Phenylacetone	-384.64842	-385.07999	-384.97471			
isopropanol	-194.22685	-194.45875	-194.37852			

Figure S1. DFT calculated Gibbs free energy profile (relative to **A** and isolated molecules, in kcal/mol) for the stepwise and concerted transfer hydrogenation of ketones by 2-propanol catalyzed by **2** without decooordination of cod from the metallic complex.

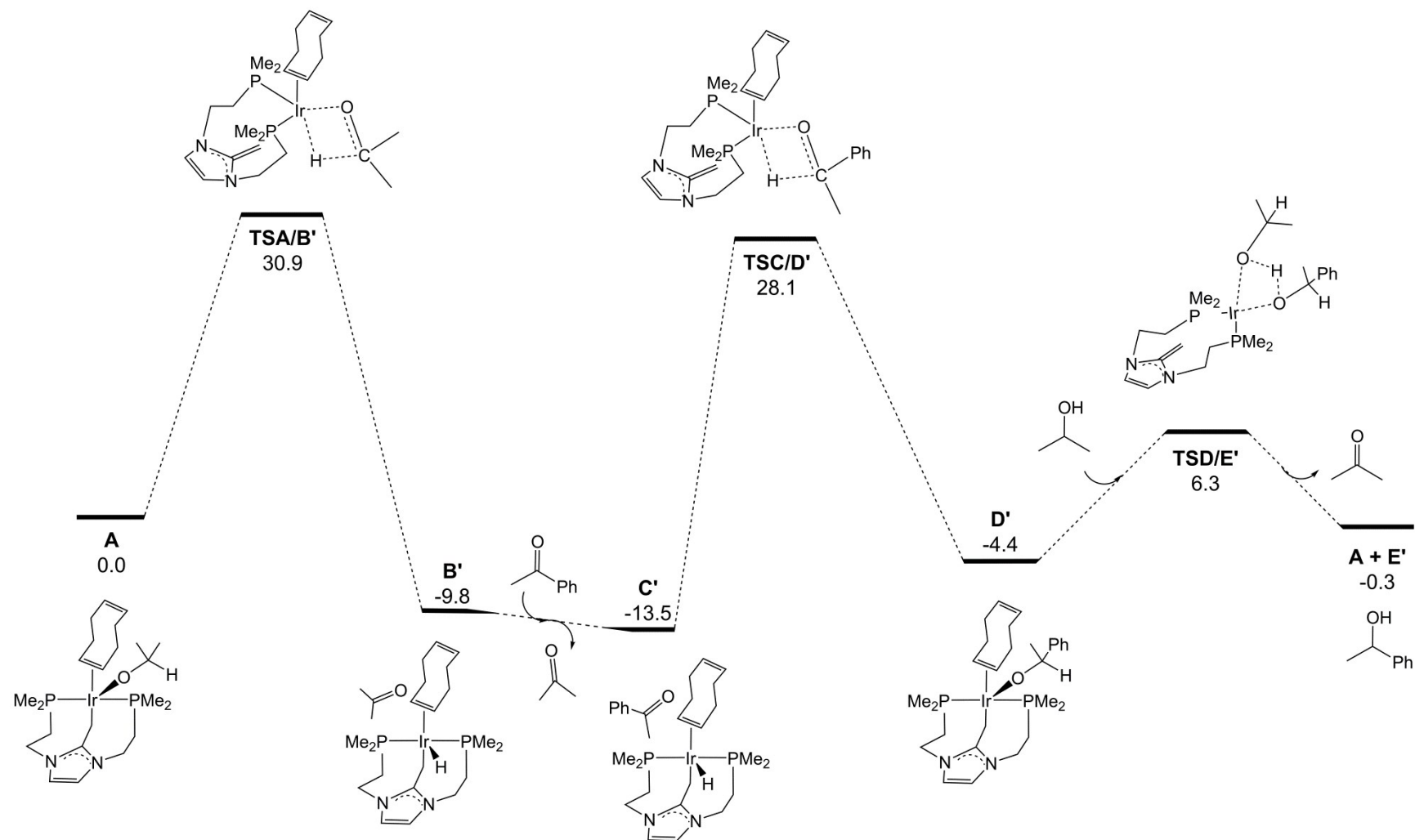


Table S3. Cartesian coordinates (in Å) for all DFT optimized structures.

A				6	2.681407	-1.122314	-1.470824
77	0.139852	-0.035290	0.165910	1	2.852008	-0.049584	-1.590248
15	-1.086056	-1.861706	0.707908	1	2.189719	-1.463028	-2.399313
7	-2.911899	-1.203440	-1.782328	6	4.059258	-1.841291	-1.332923
15	-1.084613	1.562773	1.443569	1	3.943931	-2.687426	-0.631979
6	-2.393286	0.046180	-1.555335	1	4.349840	-2.279720	-2.301450
7	-3.421100	0.790107	-1.038922	1	5.857112	-0.571210	-1.640543
6	-4.574743	0.005088	-0.960204	1	6.123251	0.284209	0.537065
1	-5.513959	0.409960	-0.595710	1	1.693423	-1.623718	1.829817
6	-4.258447	-1.230460	-1.420105	1	1.417864	-2.439424	-0.355700
1	-4.862483	-2.125628	-1.531479	1	-2.626735	-3.095835	-2.601223
6	-2.062454	-2.364409	-2.005750	1	-1.211574	-2.039014	-2.615128
6	-1.573180	-3.005680	-0.700183	8	1.386135	1.644598	-0.326306
1	-0.683964	-3.614631	-0.923404	6	1.657767	2.176830	-1.572871
1	-2.339862	-3.689843	-0.303255	1	1.450018	1.448801	-2.392044
6	-1.050580	0.441640	-1.751578	6	0.835986	3.445731	-1.886873
6	-3.269166	2.087358	-0.392512	1	1.082807	4.243909	-1.167419
1	-4.230488	2.611616	-0.490139	1	1.047036	3.824260	-2.902048
1	-2.530730	2.673280	-0.957157	1	-0.245914	3.262997	-1.815385
6	-2.885513	1.975446	1.090421	6	3.151690	2.545852	-1.668273
1	-3.525419	1.221046	1.572815	1	3.789391	1.668883	-1.486658
1	-3.106540	2.935576	1.584950	1	3.411668	2.973449	-2.652372
6	-2.698271	-1.652273	1.581844	1	3.391758	3.294712	-0.894069
1	-3.350576	-0.950438	1.053321	1	-0.540860	-0.117069	-2.541099
1	-3.218490	-2.617197	1.674868	1	-0.938998	1.516849	-1.872821
1	-2.501125	-1.260788	2.588441	A1			
6	-0.342056	-3.136418	1.818861	6	4.271615	-0.739943	-1.285416
1	-1.026603	-3.991498	1.922987	6	3.801341	0.397413	-1.812856
1	0.616753	-3.489111	1.417369	6	3.717396	1.768685	-1.200607
1	-0.168221	-2.689643	2.807876	1	4.577415	2.367804	-1.561816
6	-1.190283	1.309692	3.274085	1	2.813817	2.237906	-1.609691
1	-1.784420	2.096875	3.766124	6	3.628683	1.951771	0.328488
1	-1.627746	0.329683	3.506002	1	4.625728	1.831533	0.787618
1	-0.164809	1.327735	3.672552	1	3.376120	3.013154	0.474875
6	-0.389685	3.269015	1.405917	6	2.628952	1.152787	1.166986
1	-0.824783	3.882116	2.210248	6	2.691255	-0.213355	1.498045
1	0.701026	3.205329	1.493141	6	3.762729	-1.242520	1.160842
1	-0.607387	3.728893	0.434466	1	3.283530	-2.192727	0.872974
6	1.735338	-1.384398	-0.316908	1	4.295995	-1.460644	2.106131
6	1.904892	-0.880652	1.050879	6	4.822853	-0.902942	0.101970
6	5.159187	-0.920790	-0.869620	1	5.376963	-0.005348	0.408157
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