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Supporting Information

Trinuclear nickel and cobalt complexes containing unsymmetrical tripodal tetradentate ligands: syntheses, structural, magnetic, and catalytic properties

Jong Won Shin, Ah Rim Jeong, Sun Young Lee, Cheal Kim, Shinya Hayami, and Kil Sik Min



Fig. S1 The XRPD patterns of (a) original compound [(Hpmide)₂Co₃(CH₃COO)₄] (**2**) and (b) solid obtained after the reaction of **2**/MCPBA and cyclohexene.



Fig. S2 The XRPD patterns of (a) original compound [(pmidip)₂Co₃(CH₃COO)₄] (**3**) and (b) solid obtained after the reaction of **3**/MCPBA and cyclohexene.

Ni1-N1	2.0651(10)	Ni2-01	2.0244(8)
Ni1-N2	2.1211(10)	Ni2–O4	2.0415(8)
Ni1-O1	2.0036(9)	Ni2-06	2.1535(9)
Ni1-O2	2.1524(9)	Ni1•••Ni2	3.0895(6)
Ni1-O3	2.0236(8)	Ni1•••Ni1 ⁱ	6.1790(12)
Ni1-06	2.1451(9)		
N1-Ni1-N2	80.97(4)	O2-Ni1-O3	94.15(3)
N1-Ni1-O1	102.28(4)	O2-Ni1-O6	87.85(4)
N1-Ni1-O2	88.53(4)	O3-Ni1-O6	88.14(3)
N1-Ni1-O3	94.58(3)	01-Ni2-04	87.16(3)
N1-Ni1-O6	175.62(3)	O1-Ni2-O4 ⁱ	92.84(3)
N2-Ni1-O1	85.07(3)	O1-Ni2-O6	79.94(4)
N2-Ni1-O2	82.06(3)	O1-Ni2-O6 ⁱ	100.06(4)
N2-Ni1-O3	174.20(3)	O4-Ni2-O6	88.19(3)
N2-Ni1-O6	96.07(3)	O4-Ni2-O6 ⁱ	91.81(3)
O1-Ni1-O2	161.70(3)	Ni1-O1-Ni2	100.17(4)
O1-Ni1-O3	99.58(3)	Ni1-06-Ni2	91.90(4)
01-Ni1-06	80.61(4)		

Table S1Selected bond distances (Å) and angles (°) for 1

 $\frac{O1-N11-O6}{Symmetry transformations used to generate equivalent atoms: (i) -x+1, -y, -z.$

Co1-N1	2.150(4)	Co2-O1	2.016(4)
Co1-N2	2.250(4)	Co2–O4	2.110(4)
Co1-O1	1.987(4)	Co2–O6	2.196(4)
Co1–O2	2.196(4)	Co1•••Co2	3.1314(2)
Co1–O3	2.115(5)	Co1•••Co1 ⁱ	6.2627(4)
Co1-O6	2.016(4)		
N1-Co1-N2	78.74(17)	O2-Co1-O3	98.01(15)
N1-Co1-O1	105.05(17)	O2-Co1-O6	83.80(14)
N1-Co1-O2	90.32(16)	O3-Co1-O6	86.71(15)
N1-Co1-O3	96.89(17)	O1-Co2-O4	87.47(16)
N1-Co1-O6	173.49(17)	O1-Co2-O4 ⁱ	92.54(16)
N2-Co1-O1	82.15(16)	O1-Co2-O6	80.14(15)
N2-Co1-O2	79.99(15)	O1-Co2-O6 ⁱ	99.86(15)
N2-Co1-O3	175.14(16)	O4-Co2-O6	86.91(15)
N2-Co1-O6	97.43(16)	O4-Co2-O6 ⁱ	93.09(15)
O1-Co1-O2	153.63(16)	Co1-O1-Co2	102.92(17)
O1-Co1-O3	101.17(15)	Co1-O6-Co2	89.56(14)
O1-Co1-O6	79.44(15)		

Table S2Selected bond distances (Å) and angles (°) for 2

Symmetry transformations used to generate equivalent atoms: (i) -x+2, -y+1, -z.

Co1-N1	1.9441(17)	Co2-O1	2.0719(14)
Co1-N2	1.9380(19)	Co2–O2	2.1621(14)
Co1-O1	1.8999(14)	Co2–O4	2.0980(17)
Co1–O2	1.8904(14)	Co1•••Co2	3.0045(6)
Co1–O3	1.9152(16)	Co1•••Co1 ⁱ	6.0090(11)
Co1-O5	1.9375(15)		
N1-Co1-N2	85.25(8)	O2-Co1-O3	91.58(6)
N1-Co1-O1	171.21(7)	O2-Co1-O5	177.00(7)
N1-Co1-O2	92.47(7)	O3-Co1-O5	91.39(7)
N1-Co1-O3	89.33(7)	O1-Co2-O4	86.99(6)
N1-Co1-O5	87.93(7)	O1-Co2-O4 ⁱ	93.00(6)
N2-Co1-O1	86.09(7)	O1-Co2-O2	75.02(5)
N2-Co1-O2	88.17(7)	O1-Co2-O2 ⁱ	104.98(5)
N2-Co1-O3	174.56(7)	O2-Co2-O4	83.57(6)
N2-Co1-O5	88.90(7)	O2-Co2-O4 ⁱ	96.43(6)
O1-Co1-O2	85.76(6)	Co1-O1-Co2	98.21(6)
O1-Co1-O3	99.31(6)	Co1-O2-Co2	95.47(6)
O1-Co1-O5	93.40(6)		

Table S3Selected bond distances (Å) and angles (°) for 3

 $\frac{O1-Co1-O5}{Symmetry transformations used to generate equivalent atoms: (i) -x, -y+1, -z+1.}$



Scheme S1 Plausible olefin epoxidation mechanism catalyzed by 3 with peracids.