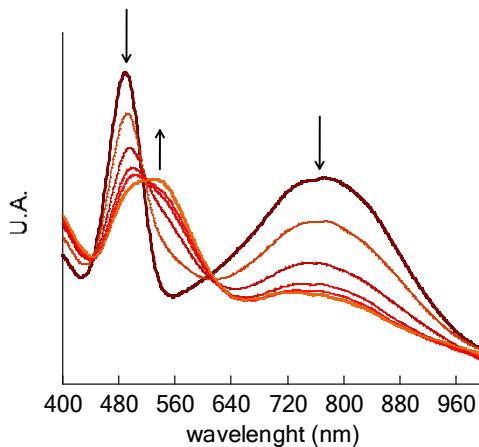


## **A nonlinear optical active polymer film based on Pd(II) dithione/dithiolate second-order NLO chromophore**

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Angela Serpe, Francesca Tessore and Paola Deplano

### **Supplementary Material**

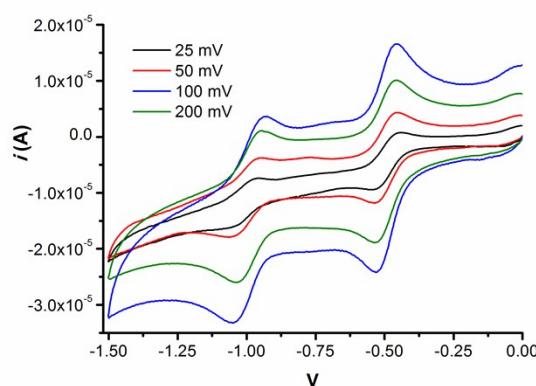


**Figure S1.** Spectroelectrochemical reduction of  $[\text{Pd}(\text{Bz}_2\text{pipdt})(\text{dmit})]$ . Study carried out in 0.1 M TBABF<sub>4</sub>/DMF, at  $-40^\circ\text{C}$ , with an applied potential of  $-0.6\text{ V}$  (vs. Ag/AgCl).

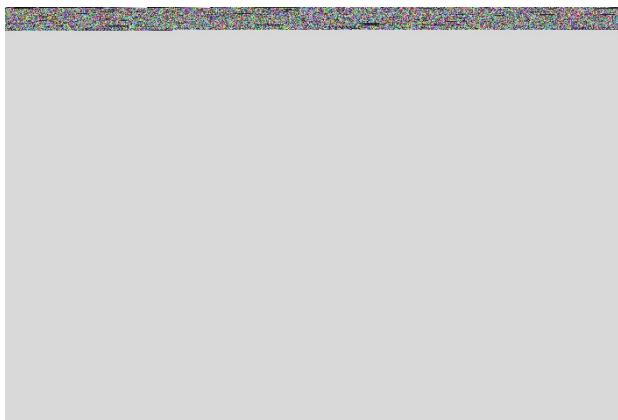
**Table S1.** Cyclic voltammetric data. Measured at the Pt electrode in DMF, 0.2 M  $n\text{-Bu}_4\text{NClO}_4$  (reference electrode Ag/AgCl) for **1**. Corresponding data for R= Pr<sup>i</sup> and Bz are reported for comparison.

Complex	$E_a$ (V) <sup>a</sup> $0 \rightarrow +1$	$E_{1/2}^1$ (V) $0 \rightarrow -1$	$E_{1/2}^2$ (V) $-1 \rightarrow -2$	Ref.
$[\text{Pd}(\text{Pr}_2\text{pipdt})(\text{dmit})]$	+0.84	-0.38	-0.93	21
$[\text{Pd}(\text{Bz}_2\text{pipdt})(\text{dmit})]$	+0.84	-0.36	-0.81	5c
$[\text{Pd}(\text{DOD}_2\text{pipdt})(\text{dmit})]$	+0.76	-0.49	-0.99	This work

<sup>a</sup> Irreversible



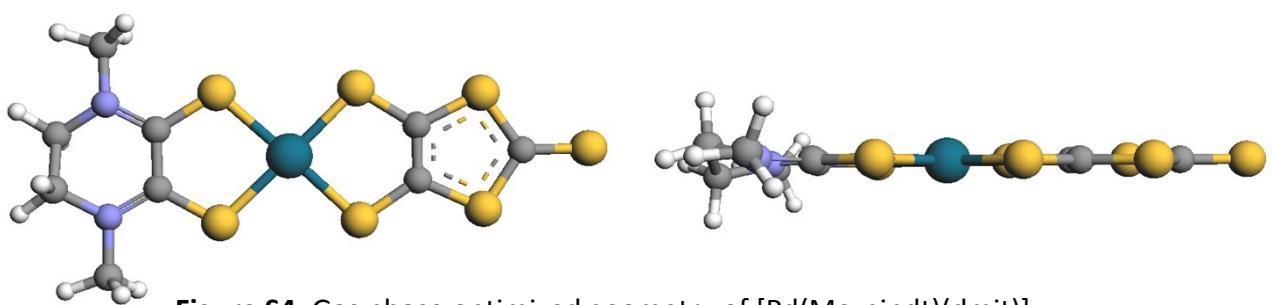
**Figure S2.** Reduction processes of **1**. Cyclic voltammograms recorded at different scan rates in a DMF solution, containing 0.2 M  $n\text{-Bu}_4\text{NClO}_4$  as supporting electrolyte (V vs Ag/AgCl in KCl 3 M).



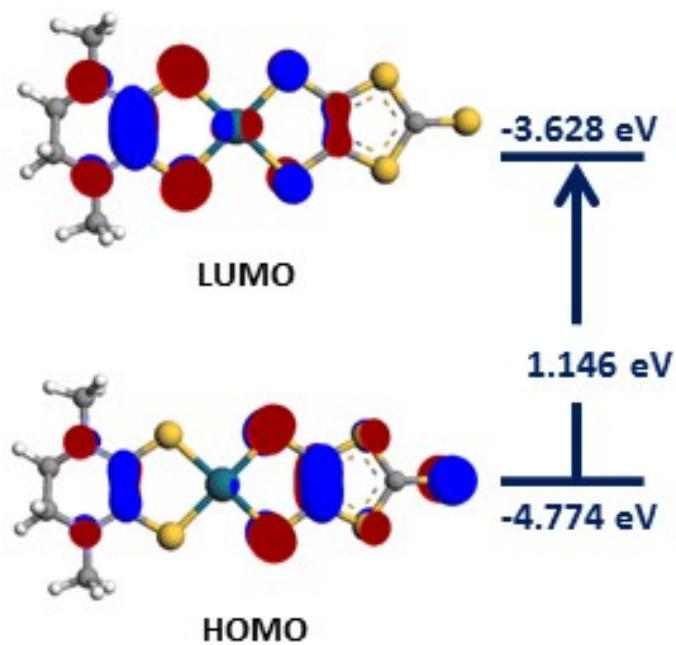
**Figure S3.** First reduction process of **1**. Cyclic voltammograms recorded at different scan rates in a DMF solution, containing 0.2 M  $n\text{-Bu}_4\text{NClO}_4$  as supporting electrolyte (V vs Ag/AgCl in KCl 3 M).

*Computational studies.*

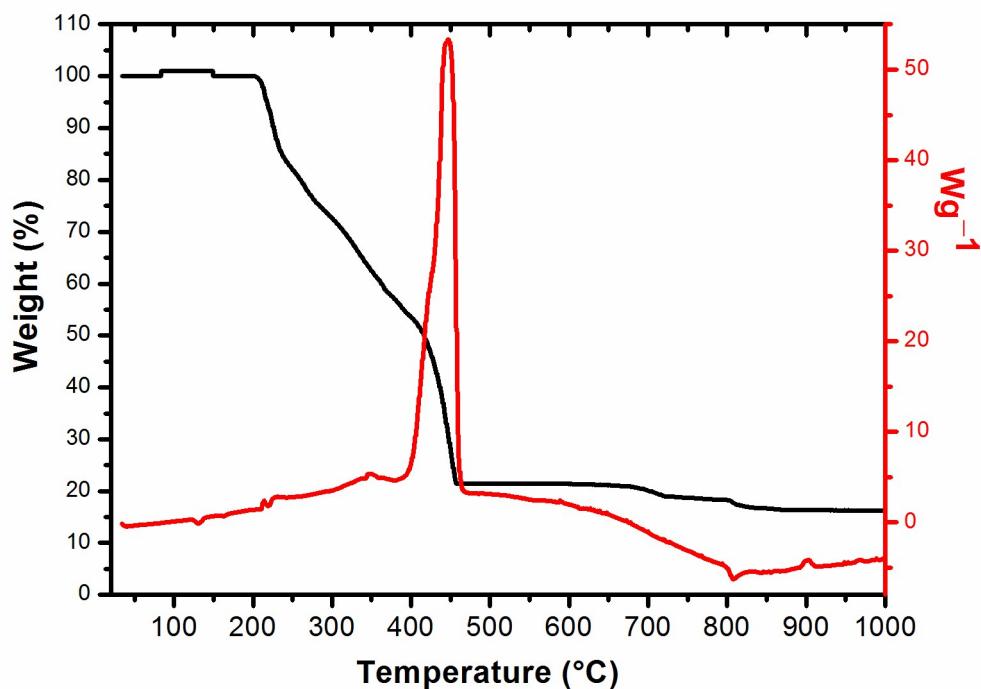
Density Functional Theory (DFT)<sup>1</sup> calculations were carried out, using the GAUSSIAN 09<sup>2</sup> software package, with the aim to investigate the electronic structure of the complex  $[\text{Pd}(\text{Me}_2\text{pipdt})(\text{dmit})]$ . In order to reduce the calculation's time, the methyl group was used as substituent instead of the dodecyl. Previous studies on complexes bearing the  $\text{R}_2\text{pipdt}$  ligand, have shown that the substituent at the nitrogen atoms plays a negligible role in determining the frontier orbitals.<sup>3</sup> B3LYP<sup>4</sup> was employed as functional, whereas the basis set 6-31G(d,p)<sup>5</sup> was used for C, H, N, and S atoms and the pseudopotential LanL2DZ for Pd.<sup>6</sup> ArgusLab 4.0 program<sup>7</sup> was employed to input the structure and to visualize the optimized molecular structure and the orbital isosurfaces.



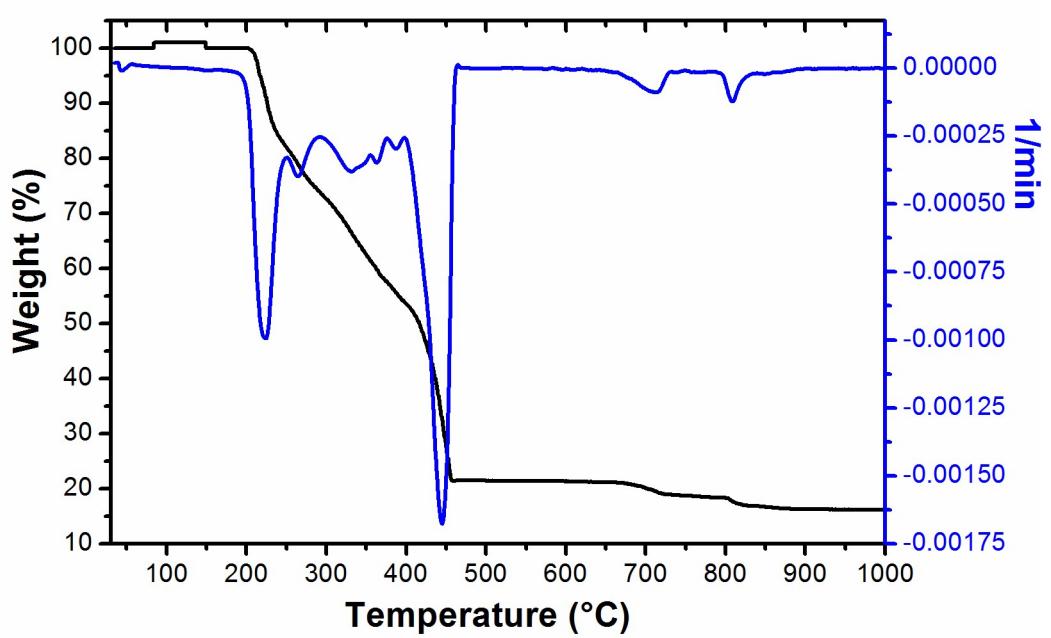
**Figure S4.** Gas phase optimized geometry of  $[\text{Pd}(\text{Me}_2\text{pipdt})(\text{dmit})]$ .



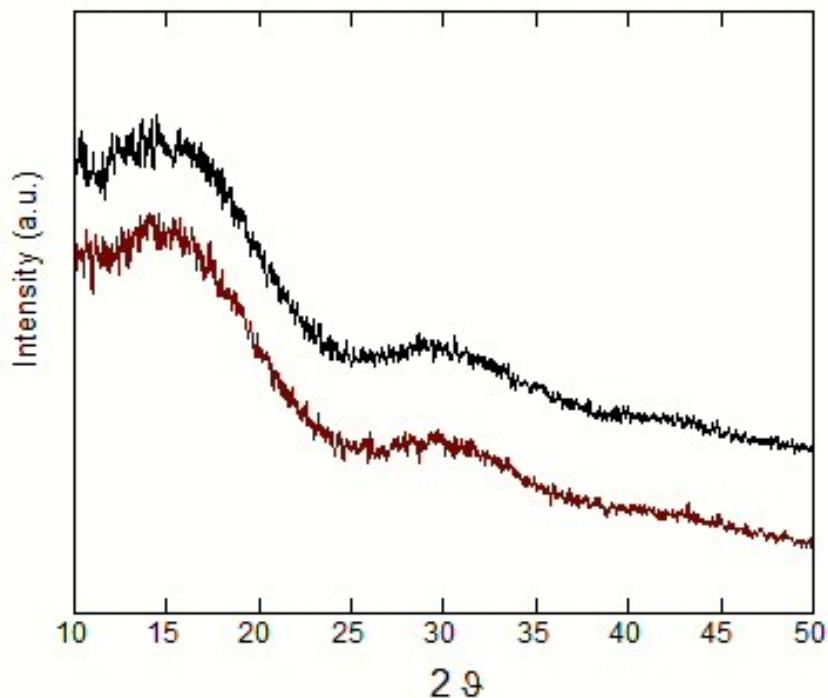
**Figure S5.** FOs of  $[\text{Pd}(\text{Me}_2\text{pipdt})(\text{dmit})]$  calculated by DFT methods; the contour plot value is 0.040.



**Figure S6.** Comparison between TGA (black) and DCS (red) curves of **1**.



**Figure S7.** Comparison between TGA curve of **1** (black) and its first derivative.



**Figure S8.** Comparison between the XRD patterns of films of pure PMMA (red-brown) and **1**-PMMA (black).

Wide-angle XRD patterns were recorded with a Panalytical Empyrean diffractometer equipped with a graphite monochromator and a X'Celerator linear detector. The scans were collected within the range 10–50° ( $2\theta$ ) using Cu K $\alpha$  radiation.

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