## **Electronic Supplementary Information**

## for

## N-H···O and N-H···Cl Supported 1D Chains of Heterobimetallic Cu<sup>II</sup>/Ni<sup>II</sup>- Sn<sup>IV</sup> Cocrystals

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Scheme S1 Preparation of cocrystals salts, cocrystals and salts by the metalloligand and the selfassembly approaches.



Scheme S2. Chemical diagram of a family of related H<sub>2</sub>L Schiff bases.

Compounds	D–H···A	H···A	D···A	D–H…A	Symmetry	
	N3–H3C····Cl2 $^i$	2.32	3.201	172.0		
	N3-H3A…O1	2.09	2.827	140.0	<i>i</i> ) x, y, 1+z	
1	N3-H3B…O2	2.18	2.834	131.0		
	N3-H3A…O3	2.15	2.891	141.0		
	N3−H3B…O4	2.12	2.920	151.0		
	N3–H3C····Cl3 $^i$	2.44	3.259	157.0		
2	N3-H3A…O1	2.01	2.760	143.0	<i>i</i> ) x, 1+y, z; <i>ii</i> ) 1-x, 1-y, 1-z	
	N3−H3B···O2	2.13	2.803	135.0		
	N3−H3A…O3	2.13	2.856	139.0		
	N3−H3B…O4	2.12	2.880	147.0		
	O5−H5M···Cl3 <sup><i>ii</i></sup>	2.474	3.316	162.0		
3	N3−H3C···Cl2	2.35	3.227	166.0	_	
	N3−H3A…O1	2.16	2.886	137.0		
	N3−H3B…O2	2.15	2.867	137.0		
	N3−H3A…O3	2.05	2.824	143.0		
	N3−H3B…O4	2.06	2.822	142.0		
	N3–H3A····Cl2 <sup><math>i</math></sup>	2.52	3.301	148.0		
	N3−H3A…O1	2.27	2.840	121.0		
4	N3-H3B…O2	1.97	2.788	149.0	<i>i</i> ) 1–x,1–y,–z	
	N3−H3A…O3	1.95	2.815	160.0		
	N3-H3B…O4	2.12	2.795	131.0		
	O5−H5M···Cl2	2.39	3.261	171.0		

Table S1 Hydrogen bonds distances (Å) and angles (°) in 1–4



Fig. S1 Schematic presentation of the encapsulation of a chain of  $(H_2ED)^{2+} \cdot [SnCl_6]^{2-}$  species in the metallo-Schiff base pockets of [ML].



Fig. S2 Supramolecular 1D chain in 2–4 supported by non-covalent interactions.



**Fig. S3** Entrapping of 1D chains of  $(H_2ED)^{2+} \cdot [SnCl_6]^{2-}$  adducts by mononuclear complexes in 1–4. Methanol molecules in 2 and 4 are omitted for clarity.



**Fig. S4** Entrapping of 1D chain of  $(H_2ED)^{2+} \cdot [(R1)_2SnCl_4]^{2-}$  adducts by the mononuclear [CuL] complexes in  $(H_2ED)^{2+} \cdot 2[CuL] \cdot [(R1)_2SnCl_4]^{2-}$  (L = L<sup>1</sup>, R = Me, R1 = Me, Et, *n*-Bu and Ph; L = L<sup>2</sup>, R = Et, R1 = Ph; ED = 1,2-ethylenediamine; excluding the solvent of crystallization). (Reference 9c of the main text).

	1	2	3	4
Formula	$C_{38}H_{46}Cl_6Cu_2N_6O_8Sn$	$C_{44}H_{62}Cl_6Cu_2N_6O_{10}Sn$	C <sub>38</sub> H <sub>46</sub> Cl <sub>6</sub> Ni <sub>2</sub> N <sub>6</sub> O <sub>8</sub> Sn	C <sub>44</sub> H <sub>62</sub> Cl <sub>6</sub> Ni <sub>2</sub> N <sub>6</sub> O <sub>10</sub> Sn
FW	1173.28	1293.46	1163.62	1283.80
Crystal system	Triclinic	Triclinic	Triclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
a/Å	10.466(4)	10.5572(4)	10.4892(12)	10.3172(4)
b/Å	11.006(5)	11.5990(4)	10.9238(12)	11.7507(5)
c/Å	11.089(4)	12.9106(5)	11.1384(12)	12.8219(6)
α/°	80.040(15)	112.4350(10)	90.216(4)	114.355(2)
β/°	62.536(18)	111.101(2)	117.436(4)	108.037(2)
γ/°	79.168(15)	96.452(2)	99.734(4)	96.323(2)
$V/Å^3$	1107.7(8)	1304.86(9)	1111.5(2)	1294.78(10)
Ζ	1	1	1	1
<i>T</i> /K	298(2)	150(2)	298(2)	150(2)
<i>θ/</i> °	2.266 - 26.478	2.475 - 25.744	2.229 - 25.692	2.246 - 25.723
$\mu$ (Mo K $\alpha$ )/mm <sup>-1</sup>	1.930	1.650	1.814	1.568
$\rho_{\rm calcd}/{\rm g~cm^{-3}}$	1.759	1.646	1.738	1.646
F(000)	590	658	588	656
Index ranges	-13 <h<13< td=""><td>-12<h<12< td=""><td>-12<h<12< td=""><td>-12<h<12< td=""></h<12<></td></h<12<></td></h<12<></td></h<13<>	-12 <h<12< td=""><td>-12<h<12< td=""><td>-12<h<12< td=""></h<12<></td></h<12<></td></h<12<>	-12 <h<12< td=""><td>-12<h<12< td=""></h<12<></td></h<12<>	-12 <h<12< td=""></h<12<>
	-13 <k<13< td=""><td>-14&lt;<i>k</i>&lt;14</td><td>-13<k<13< td=""><td>-14&lt;<i>k</i>&lt;14</td></k<13<></td></k<13<>	-14< <i>k</i> <14	-13 <k<13< td=""><td>-14&lt;<i>k</i>&lt;14</td></k<13<>	-14< <i>k</i> <14
	-13< <i>l</i> <13	-15<1<15	-13 13</td <td>-15<l<15< td=""></l<15<></td>	-15 <l<15< td=""></l<15<>
Rfs. collected	17669	22907	18140	23394
Rfs. unique/observed	4531/3256	4949/3918	4215/2855	4879/4091
R <sub>int</sub>	0.0732	0.0488	0.0842	0.0312
$R_1^{a}/wR_2^{b} [I > 2\sigma(I)]$	0.0507/0.1106	0.0292/0.0637	0.0509/0.1094	0.0267/0.0577
$R_1^{a}/wR_2^{b}$ [for all $F_0^2$ ]	0.0815/0.1237	0.0466/0.0696	0.0930/0.1265	0.0388/0.0629
GOF on $F^2$	1.026	1.019	1.033	1.058

## Table S2 Crystallographic data for 1–4