# The Selective Activation of C-F Bond with the Auxiliary of Strong 

 Lewis Acid --- A Method to Change the Activation Preference of C-F and C-H bondLin Wang, ${ }^{a}$ Hongjian Sun, ${ }^{a}$ Xiaoyan Li, ${ }^{a, *}$ Olaf Fuhr, ${ }^{b}$ Dieter Fenske ${ }^{b}$

SI. Selected X-ray crystallographic data

|  | 10* | 11 | 12 |
| :---: | :---: | :---: | :---: |
| formula | $\mathrm{C}_{25} \mathrm{H}_{44} \mathrm{BF}_{6} \mathrm{FeNP}_{4}$ | $\mathrm{C}_{23} \mathrm{H}_{38} \mathrm{BrFFeNP}_{3}$ | $\mathrm{C}_{26} \mathrm{H}_{44} \mathrm{FFeINP}_{3}$ |
| $M_{z}$ | 663.15 | 576.21 | 665.28 |
| crystal system | Monoclinic | Monoclinic | Monoclinic |
| space group | P2(1)/n | P2(1)/c | P2(1)/c |
| $\mathrm{a}[\AA]$ | 11.7534(1) | 8.9625(3) | 15.6776(12) |
| b [ $\AA$ ] | 19.4894(2) | 33.2859(13) | 13.1670(11) |
| c [ $\AA$ ] | 14.8134(1) | 9.0190(3) | 16.6228(17) |
| $\left.\alpha{ }^{[ }\right]$ | 90.00 | 90.00 | 90.00 |
| $\beta{ }^{\left[{ }^{\circ}\right]}$ | 93.7660 (1) | 95.341(3) | 118.742(5) |
| $\gamma\left[{ }^{\circ}\right]$ | 90.00 | 90.00 | 90.00 |
| $\mathrm{V}\left[\AA^{3}\right]$ | 3385.9(5) | 2678.91(2) | 3008.6(5) |
| T [K] | 296(2) | 153(2) | 296(2) |
| Z | 4 | 4 | 4 |
| $\mu\left[\mathrm{mm}^{-1}\right]$ | 0.683 | 2.252 | 1.708 |
| total reflns | 25229 | 7826 | 42857 |
| unique reflns | 7774 | 4112 | 7226 |
| $\mathrm{R}_{\text {int }}$ | 0.0306 | 0.0487 | 0.0858 |
| $\mathrm{R}_{1}[\mathrm{I}>2 \sigma(\mathrm{I})]$ | 0.0352 | 0.0533 | 0.0624 |
| wR( $\mathrm{F}^{2}$ ) | 0.1014 | 0.1477 | 0.1558 |
| [ $\mathrm{I}>2 \sigma(\mathrm{I})$ ] |  |  |  |
| $\mathrm{R}_{1}$ (all data) | 0.0456 | 0.0594 | 0.0957 |
| $w \mathrm{R}\left(\mathrm{F}^{2}\right)$ (all data) | 0.1135 | 0.1544 | 0.1778 |
| GOF on $\mathrm{F}^{2}$ | 0.541 | 1.079 | 1.084 |

[^0]
## SII. The detailed crystal data of complex

## Complex 10


$\mathrm{Fe}(1)-\mathrm{C}(13): 2.0164(2) \AA, \mathrm{Fe}(1)-\mathrm{N}(1): 1.9858(2) \AA, \mathrm{F}(3)-\mathrm{B}(1): 1.389(3) \AA, \mathrm{F}(4)-\mathrm{B}(1): 1.370(3) \AA$, $\mathrm{F}(5)-\mathrm{B}(1): 1.380(3) \AA, \mathrm{F}(6)-\mathrm{B}(1): 1.367(3) \AA . \mathrm{C}(13)-\mathrm{Fe}(1)-\mathrm{P}(3): 171.72(6)^{\circ}, \mathrm{P}(1)-\mathrm{Fe}(1)-\mathrm{P}(3):$ 91.19(2) ${ }^{\circ}$

## Complex 11


$\mathrm{Fe}(1)-\mathrm{Br}(1): 2.5653(7) \AA, \mathrm{Fe}(1)-\mathrm{N}(1): 1.945(4) \AA, \mathrm{P}(2)-\mathrm{Fe}(1)-\mathrm{N}(1): 175.45(1)^{\circ}, \mathrm{P}(1)-\mathrm{Fe}(1)-\mathrm{C}(1):$ 93.38(15) ${ }^{\circ}$

## Complex 12


$\mathrm{Fe}(1)-\mathrm{I}(1): 2.7478(9) \AA, \mathrm{Fe}(1)-\mathrm{N}(1): 1.930(5) \AA, \mathrm{P}(2)-\mathrm{Fe}(1)-\mathrm{P}(3): 94.01(7)^{\circ}, \mathrm{P}(1)-\mathrm{Fe}(1)-\mathrm{P}(3):$ $172.69(7)^{\circ}$

## SIII. The IR and NMR data of complexes

## Complex 8



IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex $\mathbf{8}$ was synthesized in diethyl ether.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex $\mathbf{8}$ was synthesized in ethyl ether.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex $\mathbf{8}$ was synthesized in diethyl ether
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex $\mathbf{8}$ was synthesized in diethyl ether.

## Complex 9



IR spectrum


## ${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$



Note: Complex 9 was synthesized in diethyl ether. The signals at 1.01 and 1.05 ppm are the signals of $\mathrm{PMe}_{3}$ contained by isomers.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 9 was synthesized in diethyl ether. The signals at -111.7 and -105.4 ppm are the signals of isomers.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 9 was synthesized in diethyl ether. The signals at 17.9 ppm and 17.0 ppm are the signals of isomers.

## Complex 10



IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{2} \mathrm{D}_{6} \mathrm{CO}\right)$


Note: Complex 10 was synthesized in THF and crystalized in THF/diethyl ether mix solvent. ${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{2} \mathrm{D}_{6} \mathrm{CO}\right)$


Note: Complex 10 was synthesized in THF and crystalized in THF/diethyl ether mix solvent. The signals at -114.2 and -152.0 ppm are the signal of isomer.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{2} \mathrm{D}_{6} \mathrm{CO}\right)$

${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR ( $\left.121.5 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 298 \mathrm{~K}, \mathrm{ppm}\right): \delta 3.4\left(\mathrm{td},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=36.4 \mathrm{~Hz},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{Pd}(\mathrm{c})\right.$ ), $11.6\left(\mathrm{dd},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=36.4 \mathrm{~Hz},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=51.0 \mathrm{~Hz}, \mathrm{~Pa} \& \mathrm{~Pb}\right), 16.1\left(\mathrm{td},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=51.0 \mathrm{~Hz},{ }^{2} \mathrm{~J}_{\mathrm{P}, \mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{Pc}(\mathrm{d})\right.$ ). The signals at 14 ppm are the signals of isomer produced by dissociation of $\mathrm{PMe}_{3}$ ligand of complex $\mathbf{1 0}$ in solvent and the signal of dissociated $\mathrm{PMe}_{3}$ ligand appears at -62.4 ppm .
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{2} \mathrm{D}_{6} \mathrm{CO}\right)$


## Complex 11



## IR spectrum


${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 11 was synthesized in diethyl ether. In solvent $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$, complex 11 would dissociate a $\mathrm{PMe}_{3}$ ligand in some extent. The signals at $1.52,1.27 \mathrm{ppm}$ are the signal of this product.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 11 was synthesized in diethyl ether.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 11 was synthesized in diethyl ether. In solvent $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$, complex 11 would dissociate a $\mathrm{PMe}_{3}$ ligand in some extent. The doublet at 15 ppm is the signal of this product.
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


## Complex 12



IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$

${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note. In solvent $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$, the arrange of $\mathrm{PMe}_{3}$ will occur. And In solvent $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$, complex 12 would dissociate a $\mathrm{PMe}_{3}$ ligand in some extent. The signals of different isomers: $12.9 \mathrm{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=56.9 \mathrm{~Hz}\right)$ and $19.2 \mathrm{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}\right.$ $=56.9 \mathrm{~Hz}) ; 13.1 \mathrm{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=55.9 \mathrm{~Hz}\right)$ and $20.0 \mathrm{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=55.9 \mathrm{~Hz}\right) ; 12.1 \mathrm{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=58.3 \mathrm{~Hz}\right) ; 11.9$ $\operatorname{ppm}\left(\mathrm{d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=57.1 \mathrm{~Hz}\right)$.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note. The signals of different isomers: -104.9 ppm; -108.8 ppm; -111.6 ppm.
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note. Isomers exist.

## Complex 13



13a


13b

IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complexes 13a and 13b were synthesized in diethyl ether and crystalized in n-pentane. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 298 \mathrm{~K}, \mathrm{ppm}$ ): $\delta-15.92$, -15.26 (td, ${ }^{2} J_{\mathrm{P}, \mathrm{H}}=21.0 \mathrm{~Hz},{ }^{2} J_{\mathrm{P}, \mathrm{H}}=84.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Fe}-H$ ), 1.05 (m, $\left.27 \mathrm{H}, \mathrm{P}\left(\mathrm{CH}_{3}\right)_{3}\right), 7.25-8.41(\mathrm{~m}, 8 \mathrm{H}, \mathrm{Ar}-H), 9.53,9.91(\mathrm{~s}, 1 \mathrm{H}, \mathrm{C}=\mathrm{N}-H)$.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complexes 13a and 13b were synthesized in diethyl ether and crystalized in n-pentane. ${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR (282.4 MHz, $\mathrm{C}_{6} \mathrm{D}_{6}, 298 \mathrm{~K}, \mathrm{ppm}$ ): $\delta-108.8,116.4$ (s, F, Ar-F).
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: 13a and 13b: ${ }^{31} \mathrm{P}$ NMR (121.5 MHz, $\left.\mathrm{C}_{6} \mathrm{D}_{6}, 298 \mathrm{~K}, \mathrm{ppm}\right): \delta 24.1,23.8\left(\mathrm{t},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=36.4 \mathrm{~Hz}, 1 \mathrm{P}\right.$, $\left.P\left(\mathrm{CH}_{3}\right)_{3}\right), 19.9,19.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{P}, \mathrm{P}}=36.4 \mathrm{~Hz}, 2 \mathrm{P}, P\left(\mathrm{CH}_{3}\right)_{3}\right)$.

## Complex 14



IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 14 was synthesized in diethyl ether and crystalized in n-pentane.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 14 was synthesized in diethyl ether and crystalized in n-pentane.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 14 was synthesized in diethyl ether and crystalized in $n$-pentane.

## Complex 15



IR spectrum

${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 15 was synthesized in diethyl ether. The signals at 3.4 ppm and 1.2 ppm are the typical signals of solvent.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 15 was synthesized in diethyl ether. The signal at 16 ppm is the arrangement isomer of complex 15.
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex $\mathbf{1 5}$ was synthesized in diethyl ether. The signals at 3.4 ppm and 1.2 ppm are the typical signals of solvent. The signals of $\mathrm{Et}_{2} \mathrm{O}$ appear at 15 and 65 ppm .

## Complex 16



## IR spectrum


${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 16 was synthesized in diethyl ether.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 16 was synthesized in diethyl ether.
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 16 was synthesized in diethyl ether.

## Complex 17



## IR spectrum


${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 17 was synthesized in diethyl ether.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 17 was synthesized in diethyl ether.
${ }^{31} \mathrm{P}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 17 was synthesized in diethyl ether.
${ }^{13} \mathrm{C}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 17 was synthesized in diethyl ether.

## Complex 18



## IR spectrum



## ${ }^{1} \mathrm{H}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$



Note: Complex 18 was synthesized in diethyl ether.
${ }^{19} \mathrm{~F}\{\mathrm{H}\}$ NMR spectrum $\left(\mathrm{C}_{6} \mathrm{D}_{6}\right)$


Note: Complex 18 was synthesized in diethyl ether.


Note: Complex 18 was synthesized in diethyl ether.

## SIV. ${ }^{1}$ H NMR spectra of Alcohol Products

COH2OH
${ }^{1} \mathrm{H}$ NMR ( $\left.300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right): 7.26-7.39(\mathrm{~m}, ~ A r, 5 \mathrm{H}), 4.62\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.70(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H}$ NMR ( $\left.300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right): 6.97-7.28(\mathrm{~m}, ~ A r, 4 \mathrm{H}), 4.55\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.82(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right): 7.33-6.91(\mathrm{~m}, \mathrm{Ar}, 4 \mathrm{H}), 4.61\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.89(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}, \delta\right): 7.23-7.06(\mathrm{~m}, ~ \mathrm{Ar}, 3 \mathrm{H}), 4.53\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.95(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}, \delta\right): 7.41-8.09(\mathrm{~m}, \mathrm{Ar}, 7 \mathrm{H}), 5.08\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.01(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right): 7.23-7.09(\mathrm{~m}, A r, 5 \mathrm{H}), 6.44(\mathrm{~d}, H \mathrm{C}=\mathrm{C}, 1 \mathrm{H}), 6.18(\mathrm{~m}, \mathrm{C}=\mathrm{CH}, 1 \mathrm{H})$, $4.14\left(\mathrm{~d}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 2.45(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.


${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right): 7.34-6.97(\mathrm{~m}, A r, 4 \mathrm{H}), 6.55(\mathrm{~d}, H \mathrm{C}=\mathrm{C}, 1 \mathrm{H}), 6.24(\mathrm{~m}, \mathrm{C}=\mathrm{CH}, 1 \mathrm{H})$, $4.30\left(\mathrm{dd}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 1.72(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H})$.

${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}, \delta\right): 7.30-7.13(\mathrm{~m}, ~ A r, 5 \mathrm{H}), 6.47(\mathrm{~d}, H \mathrm{C}=\mathrm{C}, 1 \mathrm{H}), 4.10\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right)$, $2.71(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H}), 1.83\left(\mathrm{~s}, \mathrm{CH}_{3}, 3 \mathrm{H}\right)$


${ }^{1} \mathrm{H}$ NMR ( $\left.300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta\right)$ : 7.34-7.21 (m, $\left.A r, 5 \mathrm{H}\right), 6.52(\mathrm{~d}, H \mathrm{C}=\mathrm{C}, 1 \mathrm{H}), 4.21\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right)$, $2.28\left(\mathrm{~m}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 1.90(\mathrm{~s}, \mathrm{OH}, 1 \mathrm{H}), 1.51\left(\mathrm{~m}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 1.29\left(\mathrm{~m}, \mathrm{CH}_{2}, 4 \mathrm{H}\right), 0.86\left(\mathrm{t}, \mathrm{CH}_{3}, 3 \mathrm{H}\right)$


${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta$ ): 7.30-7.14 (m, $\left.A r, 5 \mathrm{H}\right), 6.48(\mathrm{~d}, H \mathrm{C}=\mathrm{C}, 1 \mathrm{H}), 4.17\left(\mathrm{~s}, \mathrm{CH}_{2}, 2 \mathrm{H}\right)$, $2.24\left(\mathrm{~m}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 1.43\left(\mathrm{~m}, \mathrm{CH}_{2}, 2 \mathrm{H}\right), 1.24\left(\mathrm{~m}, \mathrm{CH}_{2}, 6 \mathrm{H}\right), 0.83\left(\mathrm{t}, \mathrm{CH}_{3}, 3 \mathrm{H}\right)$


## SV General Procedure of the Operando IR Experiment.

1 mL THF was charged into a two-necked Schlenk tube which had been connected with the Schlenk line. Through an adapter, the test probe was emerged into the solution. At the beginning, the operando IR spectrometer recorded the solvent signal for 10 min . After $5 \mathrm{~min}, 4-\mathrm{FC}_{6} \mathrm{H}_{4} \mathrm{CHO}$ $(0.17 \mathrm{~g}, 1.37 \mathrm{mmol})$ was added into the container. At the $10^{\text {th }}$ minute, 1 mL THF solution of complex $15(0.14 \mathrm{~g}, 0.27 \mathrm{mmol})$ was added into the mixture and the reaction container was heated to $55^{\circ} \mathrm{C}$. The mixture was kept stirring at this temperature for 15 min . During that period, the IR absorption ( $1709 \mathrm{~cm}^{-1}$ ) for $\mathrm{C}=\mathrm{O}$ bond of 4- $\mathrm{FC}_{6} \mathrm{H}_{4} \mathrm{CHO}$ maintained unchanged. At the $28^{\text {th }}$ minute, the $\mathrm{HSi}\left(\mathrm{OCH}_{2} \mathrm{CH}_{3}\right)_{3}(0.26 \mathrm{~g}, 1.6 \mathrm{mmol})$ was added. The operando IR spectra were continually recorded until the reaction was completed. The signal of $\mathrm{C}=\mathrm{O}$ bond disappeared immediately and the one of C-O bond ( $1039 \mathrm{~cm}^{-1}$ ) grew up.

(a)

METTLER TOLEDO

(b)

Figure S1. Operando IR results for the cascade reaction with three components: (a) Kinetic Profiles; (b) Three-Dimensional Fourier Transform IR (3DFT-IR) Profiles.


[^0]:    * The disorder solvent voids were squeezed. Based on the squeezed result, there are about 2.8 ethyl ether molecules in one cell.

