

## Electronic supplementary information (ESI)

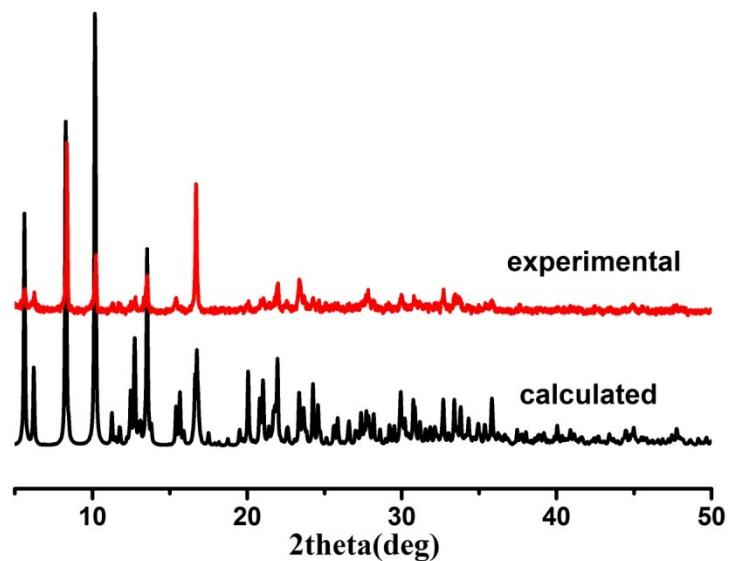
### Novel coordination polymer based on Co(II) hexanuclear clusters with azide and carboxylate bridges: structure, magnetism and its application as Li-ion battery anode

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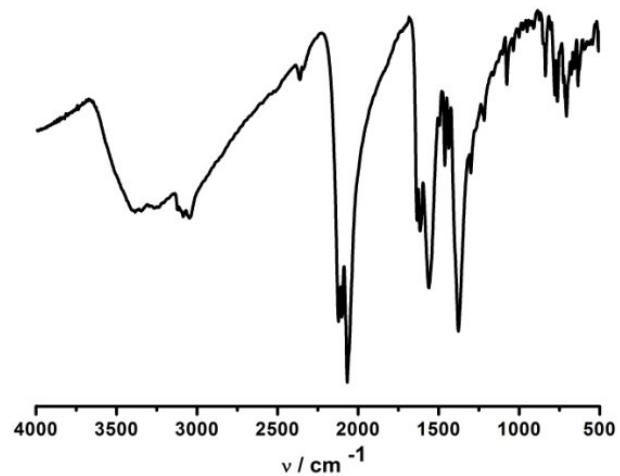
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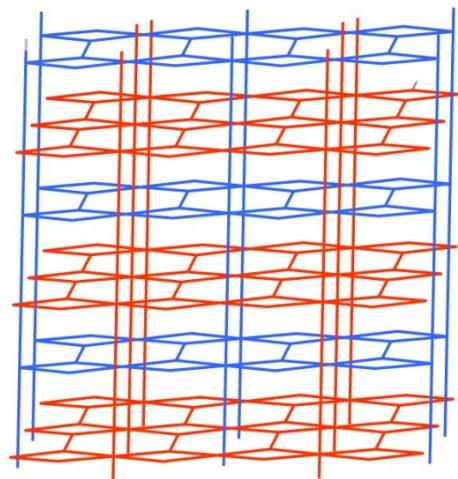
‡ These authors contributed equally to this work.



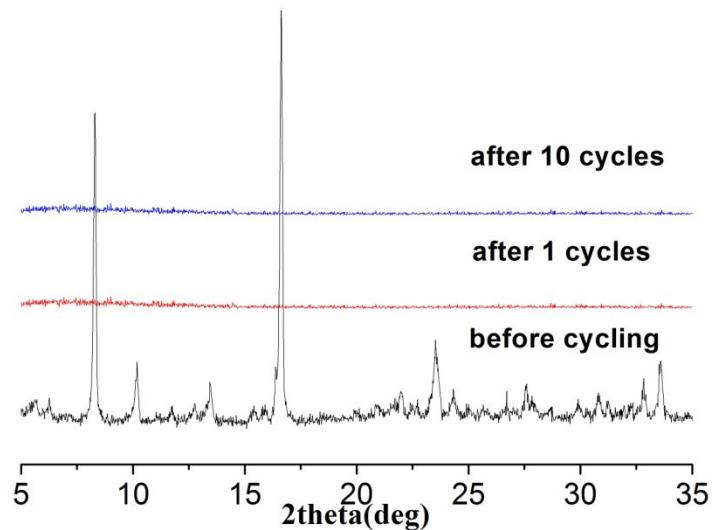
**Fig. S1** PXRD patterns of **1**.



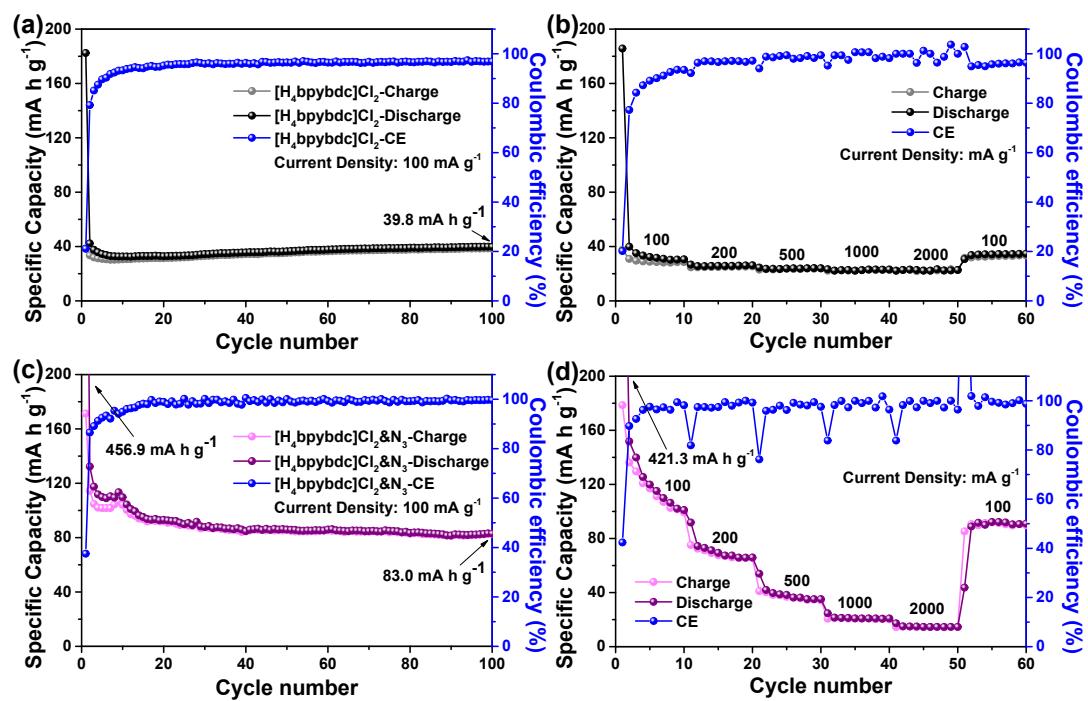
**Fig. S2** IR spectrum of **1**.



**Fig. S3** Topological presentation of the two-fold interpenetration in **1**.



**Fig. S4** PXRD patterns of the electrode before and after cycling.



**Fig. S5** Cyclic stability and Coulombic efficiency of  $[H_4bpybdc]Cl_2$  (a) and  $[H_4bpybdc]Cl_2 \& Na_3^+$  (c) at 100 mA g<sup>-1</sup>. Rate performance of  $[H_4bpybdc]Cl_2$  (b) and  $[H_4bpybdc]Cl_2 \& Na_3^+$  (d) at different current densities.

**Table S1** Hydrogen bond distances ( $\text{\AA}$ ) and angles ( $^\circ$ ) for compounds **1**.

D-H...A	dD-H)/ $\text{\AA}$	d(H...A)/ $\text{\AA}$	d(D...A)/ $\text{\AA}$	$\angle$ D-H...A/ $^\circ$
O9-H9B...O4A	0.86	1.91	2.72(6)	159(8)
O9-H9C...O5B	0.88	2.13	2.97(1)	158(5)
O10-H10B...O16	0.87	2.03	2.88(3)	168(7)
O10-H10B...O16'	0.87	1.92	2.76(3)	162(7)
O10-H10C...O6B	0.86	2.02	2.87(6)	169(7)
O11-H11B...O6C	0.87	1.78	2.63 (9)	167(10)
O11-H11B...O6'C	0.87	1.96	2.74(9)	151(9)
O11-H11C...O3D	0.87	1.95	2.76(0)	153(9)
O12-H12B...O4E	0.89	1.92	2.71(7)	148(5)
O12-H12C...O5C	0.87	1.95	2.80(4)	166(8)
O13-H13B...O3E	0.86	2.00	2.77(2)	148(4)
O13-H13C...O10F	0.87	2.11	2.90(8)	152(5)

Symmetry codes: (A) -x+1, -y, -z+2; (B) x+1, y, z+1; (C) x+1, y, z; (D) x, y+1, z-1; (E) 1-x, -y, 1-z; (F) x, y, z-1.

**Table S2** Carboxylate-coordinated MOFs or CPs for LIBs anode for LIBs.

Organic ligands	MOFs or CPs	Voltage (V vs Li <sup>+</sup> /Li)	Current Density (mA g <sup>-1</sup> )/C rate	Cycle Number	Reversible Capacity (mA h g <sup>-1</sup> )	Refs
<b>H<sub>3</sub>BTB</b>	MOF-177	0.1–1.6	50	2	105	1
<b>H<sub>3</sub>TTPCA</b>	Cd-MOF	0.1–3	100	100	302	2
<b>Asp</b>	Asp-Cu	0.01–3	50	200	233	3
<b>HCOOH</b>	Zn <sub>3</sub> (HCOO) <sub>6</sub>	0.005–3.0 V	60	60	560	
	Co <sub>3</sub> (HCOO) <sub>6</sub>	0.005–3.0 V	60	60	410	4
	Zn <sub>1.5</sub> Co <sub>1.5</sub> (HCOO) <sub>6</sub>	0.005–3.0 V	60	60	510	
	Ni-NTC	0.01–3	100	80	248	
<b>NTCDA</b>	Li-NTC	0.01–3	100	80	468	5
	Li/Ni-NTC	0.01–3	100	80	482	
	Co <sub>2</sub> (OH) <sub>2</sub> BDC	0.02–3	50	100	650	6
<b>H<sub>2</sub>BDC</b>	Ni-MOF	0.01–3	100	100	620	7
	[Cu <sub>2</sub> (C <sub>8</sub> H <sub>4</sub> O <sub>4</sub> ) <sub>4</sub> ] <sub>n</sub>	0.01–2.5	48	50	161	8
	Co-Zn-MOF	0.01–3	100	100	1211	9
<b>H<sub>3</sub>BTC</b>	Mn-BTC	0.01–2	100	100	694	10
	CoBTC-EtOH	0.01–3	100	100	856	11
	Cu <sub>3</sub> (BTC) <sub>2</sub>	0.05–3	383	50	474	12
<b>H<sub>4</sub>BTC</b>	MnCo-BTC	0.01–3	100	150	901	13
<b>H<sub>2</sub>TFBDC &amp; 4,4'-bpy</b>	Mn-LCP	0.1–3	50	50	390	14
<b>terephthalate</b>	Li <sub>2</sub> C <sub>8</sub> H <sub>4</sub> O <sub>4</sub>	0.7–3.0	1C	80	125	15
<b>trans-trans-muconate</b>	Li <sub>2</sub> C <sub>6</sub> H <sub>4</sub> O <sub>4</sub>	0.7–3.0	1C	50	234	
<b>pda</b>	[Li <sub>6</sub> (pda) <sub>3</sub> ]·2EtO <sub>H</sub>	0.2–2.0	30	50	160	16

H<sub>3</sub>BTB: 4,4',4''-benzene-1,3,5-triyl-tri-benzoic acid; H<sub>3</sub>TTPCA: 1,1',1''-(1,3,5-triazine-2,4,6-triyl)tripiperidine-4-carboxylic acid; Asp: aspartic acid; HCOOH: methanoic acid; NTCDA: 1,4,5,8-Naphthalenetetracarboxylic dianhydride; H<sub>2</sub>BDC: terephthalic acid; H<sub>3</sub>BTC: trimesic acid; H<sub>4</sub>BTC: 1,2,4,5-Benzenetetracarboxylic acid; H<sub>2</sub>TFBDC: 2,3,5,6-tetrafluoroterephthalic acid; 4,4'-bpy: 4,4'-bipyridine; pda: 2,6-pyridinedicarboxylic acid.

## Notes and references

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