

## Supporting Information

### Bonding trends across the series of tricarbonato-actinyl anions

#### $[(\text{AnO}_2)(\text{CO}_3)_3]^{4-}$ (An = U–Cm): the plutonium turn

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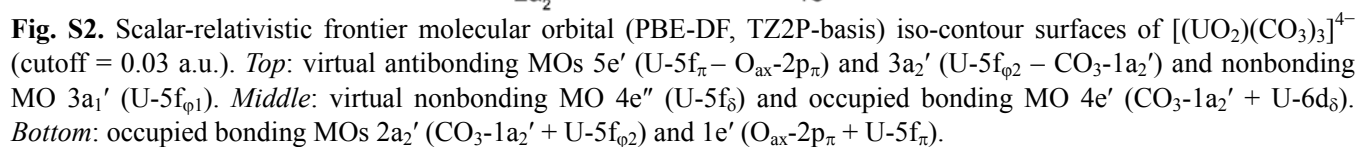
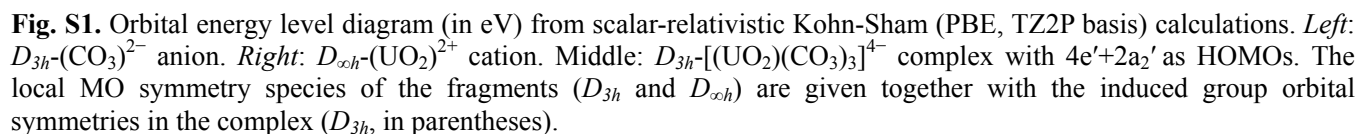
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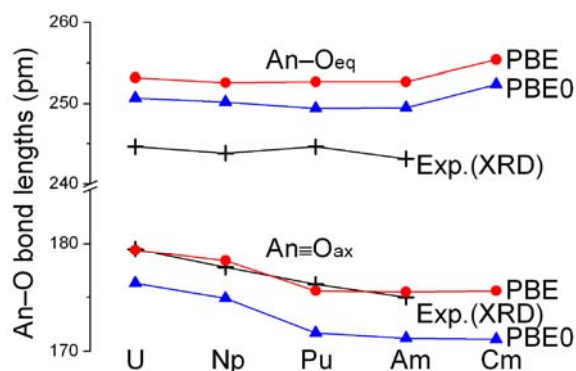
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**Fig. S3.** Averaged  $\text{An}=\text{O}_{\text{ax}}$  and  $\text{An}-\text{O}_{\text{eq}}$  bond lengths in  $[(\text{AnO}_2)(\text{CO}_3)_3][\text{C}(\text{NH}_2)_3]_9^{5+}$ .

**Table S1.** Calculated  $\text{An}-\text{O}$  bond lengths (in pm) in  $[(\text{UO}_2)(\text{CO}_3)_3]^{4-}$  and  $[(\text{NpO}_2)(\text{CO}_3)_3]^{4-}$  complexes at the PBE/TZ2P level by using SR-ZORA and SO-ZORA respectively

|           | $\text{U}=\text{O}_{\text{ax}}$ | $\text{U}-\text{O}_{\text{eq}}$ | $\text{Np}=\text{O}_{\text{ax}}$ | $\text{Np}-\text{O}_{\text{eq}}$ |
|-----------|---------------------------------|---------------------------------|----------------------------------|----------------------------------|
| SR-ZORA   | 184.7                           | 254.2                           | 183.7                            | 253.8                            |
| SO-ZORA   | 184.7                           | 253.9                           | 182.5                            | 255.2                            |
| SO change | 0.0                             | -0.3                            | -1.2                             | +1.4                             |

**Table S2.** Relative energies (in kcal/mol) of different configurations of tricarbonato-actinyl(VI) ions for different DFs

| An | Config.                         | SVWN  | PBE   | PBE0  | BLYP  | B3LYP | M06-L | M06   | M06-2X | TPSS | TPSSH |
|----|---------------------------------|-------|-------|-------|-------|-------|-------|-------|--------|------|-------|
| Np | $(f_{\phi 1u})^1$               | 0     | 0     | 0     | 0     | 0     | 0     | 0     | 0      | 0    | 0     |
|    | $(f_{\delta})^1$                | -2.27 | -0.79 | 23.4  | -0.64 | 18.9  | 4.42  | 28.98 | 42.65  | 0.99 | 10.62 |
| Pu | $(f_{\delta})^2$                | 0     | 0     | 0     | 0     | 0     | 0     | 0     | 0      | 0    | 0     |
|    | $(f_{\delta})^1(f_{\phi 1u})^1$ | 2.99  | 5.06  | 27.16 | 5.68  | 22.79 | 10.09 | 29.76 | 52.65  | —    | —     |

**Table S3.** Correlation of symmetry species of point groups  $D_{3h}$ ,  $C_{3v}$  and  $C_{2v}$ .<sup>a</sup>

| $D_{3h}$ | $C_{3v}$ | $C_{2v}$  |
|----------|----------|-----------|
| $A_1'$   | $A_1$    | $A_1$     |
| $A_2'$   | $A_2$    | $B_2$     |
| $E'$     | $E$      | $A_1+B_2$ |
| $A_1''$  | $A_2$    | $A_2$     |
| $A_2''$  | $A_1$    | $B_1$     |
| $E''$    | $E$      | $A_2+B_1$ |

<sup>a</sup> Since the CASSCF module of MOLPRO can only handle Abelian point-groups, the  $C_{2v}$  symmetry was used.

**Table S4.** An≡O<sub>ax</sub> Bond Indices in [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>][C(NH<sub>2</sub>)<sub>3</sub>]<sub>9</sub><sup>5+</sup> from DFT(PBE-DZP): *n* = number of unpaired electrons on An; *R*<sub>An≡O<sub>ax</sub></sub> = bond length (in pm), *k*<sub>An≡O<sub>ax</sub></sub> = relaxed force constant (in N/cm); BO<sub>NM2</sub> = Nalewajski- Mrozek's valence index type 2; BO<sub>FM</sub>, BO<sub>M</sub> = Mayer's fuzzy and conventional bond orders; BO<sub>L</sub> = Laplacian bond order.

| An | <i>n</i> | <i>R</i> <sub>An≡O<sub>ax</sub></sub> | <i>k</i> <sub>An≡O<sub>ax</sub></sub> | BO <sub>NM2</sub> | BO <sub>FM</sub> | BO <sub>M</sub> | BO <sub>L</sub> |
|----|----------|---------------------------------------|---------------------------------------|-------------------|------------------|-----------------|-----------------|
| U  | 0        | 179.4                                 | 6.31                                  | 2.80              | 2.89             | 1.94            | 1.27            |
| Np | 1        | 178.4                                 | 6.30                                  | 2.90              | 2.89             | 1.91            | 1.26            |
| Pu | 2        | 175.6                                 | 6.33                                  | 2.78              | 2.90             | 1.92            | 1.26            |
| Am | 3        | 175.5                                 | 5.95                                  | 2.48              | 2.88             | 1.89            | 1.19            |
| Cm | 4        | 175.6                                 | 5.32                                  | 2.43              | 2.80             | 1.81            | 0.96            |

**Table S5.** An≡O<sub>ax</sub> bond lengths (in pm) in isolated [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> ions obtained with different DFs, and experimental values for condensed phases

| An               | Density functional calculations of the isolated tricarbonato-uranyl anion                      |       |        |       |       |       |       |       |       |       |       | crystal     | solution |
|------------------|------------------------------------------------------------------------------------------------|-------|--------|-------|-------|-------|-------|-------|-------|-------|-------|-------------|----------|
|                  | El. Config.                                                                                    | M06   | M06-2X | PBE0  | M06-L | B3LYP | TPSSH | SVWN  | PBE   | TPSS  | BLYP  | XRD         | EXAFS    |
| <sup>92</sup> U  | (f) <sup>0</sup>                                                                               | 179.4 | 179.5  | 180.8 | 182.3 | 183.0 | 182.9 | 183.4 | 184.7 | 184.9 | 186.8 | 179.5(±1.0) | 181      |
| <sup>93</sup> Np | (f <sub>φ1</sub> ) <sup>1</sup>                                                                | 177.6 | 177.9  | 179.1 | 180.8 | 181.5 | 181.5 | 182.3 | 183.7 | 183.7 | 185.8 | 177.8(±1.5) | 179      |
| <sup>94</sup> Pu | (f <sub>δ</sub> ) <sup>2</sup>                                                                 | 175.9 | 174.4  | 176.1 | 179.1 | 179.0 | 179.0 | 178.7 | 180.7 | 181.0 | 183.2 | 176.2(±1.2) | 175      |
| <sup>95</sup> Am | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup>                                 | 174.5 | 174.6  | 175.5 | 179.0 | 178.6 | 178.8 | 178.6 | 180.9 | 182.8 | 183.6 | 175.0±(1.2) |          |
| <sup>96</sup> Cm | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup> (f <sub>φ2</sub> ) <sup>1</sup> | 175.5 | 185.6  | 176.4 | 186.6 | 183.1 | 183.3 | 178.8 | 183.2 | 184.7 | 186.8 |             |          |

**Table S6.** An–O<sub>eq</sub> bond lengths (in pm) in isolated [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> ions obtained with different DFs, and experimental values for condensed phases

| An               | Density Functional calculations of the isolated tricarbonato-uranyl anion                      |       |        |       |       |       |       |       |       |       |       | crystal     | solution |
|------------------|------------------------------------------------------------------------------------------------|-------|--------|-------|-------|-------|-------|-------|-------|-------|-------|-------------|----------|
|                  | El. Config.                                                                                    | SVWN  | M06-2X | PBE0  | TPSSH | M06   | TPSS  | B3LYP | PBE   | M06L  | BLYP  | XRD         | EXAFS    |
| <sup>92</sup> U  | (f) <sup>0</sup>                                                                               | 246.6 | 250.6  | 251.4 | 251.4 | 253.1 | 252.6 | 254.4 | 254.2 | 254.4 | 257.8 | 244.6(±0.6) | 244      |
| <sup>93</sup> Np | (f <sub>φ1</sub> ) <sup>1</sup>                                                                | 245.7 | 250.3  | 251.0 | 250.8 | 252.9 | 251.9 | 254.1 | 253.8 | 254.1 | 257.5 | 243.8(±0.9) | 245      |
| <sup>94</sup> Pu | (f <sub>δ</sub> ) <sup>2</sup>                                                                 | 247.4 | 250.4  | 252.2 | 254.4 | 254.8 | 256.2 | 256.7 | 257.6 | 260.7 | 262.3 | 244.6(±1.0) | 246      |
| <sup>95</sup> Am | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup>                                 | 247.2 | 250.6  | 252.2 | 253.1 | 252.8 | 255.4 | 256.3 | 258.0 | 257.1 | 262.5 | 243.1(±1.2) |          |
| <sup>96</sup> Cm | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup> (f <sub>φ2</sub> ) <sup>1</sup> | 252.0 |        | 257.8 | 264.9 | 257.2 | 267.5 | 267.9 | 270.2 | 269.8 | 277.5 |             |          |

**Table S7.** Am≡O<sub>ax</sub> and Am–O<sub>eq</sub> bond lengths (in pm) in [(AmO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> depending on the percentage of Hartree-Fock exchange (HF<sub>x</sub>) in the PBE0 density functional

| <i>D</i>           | XRD   | HF <sub>x</sub> = 0 | 5     | 10    | 15    | 20    | 25    | 30    | 35    | 40    | 45    | 50    | Δ <sub>50-0</sub> |
|--------------------|-------|---------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------------------|
| Am≡O <sub>ax</sub> | 175.0 | 180.9               | 179.7 | 178.6 | 177.5 | 176.5 | 175.5 | 174.5 | 173.7 | 172.9 | 172.2 | 171.5 | -9.4pm            |
| Am–O <sub>eq</sub> | 243.1 | 258.0               | 256.8 | 255.6 | 254.4 | 253.3 | 252.2 | 251.3 | 250.6 | 250.1 | 249.6 | 249.2 | -8.8pm            |

**Table S8.** An≡O<sub>ax</sub> bond lengths (in pm) in isolated [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> ions obtained from a dispersion corrected functional (DFT-D3-BJ) of Grimme

| An | El. Config.                     | XRD/EXAFS   | PBE   | PBE-D3 | BLYP  | BLYP-D3 | B3LYP | B3LYP-D3 |
|----|---------------------------------|-------------|-------|--------|-------|---------|-------|----------|
| U  | (f) <sup>0</sup>                | 179.5/179.0 | 184.7 | 184.7  | 186.8 | 186.7   | 183.0 | 183.0    |
| Np | (f <sub>φ1</sub> ) <sup>1</sup> | 177.8/178.0 | 183.7 | 183.6  | 185.8 | 185.7   | 181.5 | 181.5    |
| Pu | (f <sub>δ</sub> ) <sup>2</sup>  | 176.2/175.0 | 180.7 | 180.6  | 183.2 | 183.1   | 179.0 | 178.9    |

**Table S9.** An–O<sub>eq</sub> bond lengths (in pm) in isolated [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> ions obtained from a dispersion corrected functional (DFT-D3-BJ) of Grimme

| An | El. Config.                     | XRD/EXAFS   | PBE   | PBE-D3 | BLYP  | BLYP-D3 | B3LYP | B3LYP-D3 |
|----|---------------------------------|-------------|-------|--------|-------|---------|-------|----------|
| U  | (f) <sup>0</sup>                | 244.6/242.0 | 254.2 | 253.9  | 257.8 | 257.2   | 254.4 | 253.9    |
| Np | (f <sub>φ1</sub> ) <sup>1</sup> | 243.8/242.0 | 253.8 | 253.5  | 257.5 | 256.9   | 254.1 | 253.6    |
| Pu | (f <sub>δ</sub> ) <sup>2</sup>  | 244.6/246.0 | 257.6 | 257.2  | 262.3 | 261.4   | 256.7 | 256.0    |

**Table S10.** Np–O bond lengths (in pm) in isolated [(NpO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>]<sup>4-</sup> ions obtained for different DFs and two different electronic configurations

|                    | Config.                        | XRD/EXAFS       | M06   | M06-2X | PBE0  | M06-L | B3LYP | TPSSH | SVWN  | PBE   | TPSS  | BLYP  |
|--------------------|--------------------------------|-----------------|-------|--------|-------|-------|-------|-------|-------|-------|-------|-------|
| Np=O <sub>ax</sub> | (f <sub>δ</sub> ) <sup>1</sup> | 177.8(±1.5)/178 | 177.0 | 176.5  | 178.0 | 180.0 | 180.5 | 180.4 | 180.8 | 182.4 | 182.5 | 184.7 |
|                    | (f <sub>φ</sub> ) <sup>1</sup> | 177.8(±1.5)/178 | 177.6 | 177.9  | 179.1 | 180.8 | 181.5 | 181.5 | 182.3 | 183.7 | 183.7 | 185.8 |

|                    | Config.                        | XRD/EXAFS       | SVWN  | M06-2X | PBE0  | TPSSH | M06   | TPSS  | B3LYP | PBE   | M06-L | BLYP  |
|--------------------|--------------------------------|-----------------|-------|--------|-------|-------|-------|-------|-------|-------|-------|-------|
| Np-O <sub>eq</sub> | (f <sub>δ</sub> ) <sup>1</sup> | 243.8(±0.9)/242 | 246.8 | 250.4  | 251.3 | 251.7 | 252.8 | 253.3 | 254.6 | 255.3 | 255.5 | 259.3 |
|                    | (f <sub>φ</sub> ) <sup>1</sup> | 243.8(±0.9)/242 | 245.7 | 250.3  | 251.0 | 250.8 | 252.9 | 251.9 | 254.1 | 253.8 | 254.1 | 257.5 |

**Table S11.** Averaged An≡O<sub>ax</sub> and An–O<sub>eq</sub> bond lengths (in pm) in [(AnO<sub>2</sub>)(CO<sub>3</sub>)<sub>3</sub>][C(NH<sub>2</sub>)<sub>3</sub>]<sub>9</sub><sup>5+</sup> (bond lengths of the complex in vacuum in parentheses)

| An | Config.                                                                                        | An≡O <sub>ax</sub> |       |              |              | An–O <sub>eq</sub> |       |              |              |
|----|------------------------------------------------------------------------------------------------|--------------------|-------|--------------|--------------|--------------------|-------|--------------|--------------|
|    |                                                                                                | XRD                | EXAFS | PBE          | PBE0         | XRD                | EXAFS | PBE          | PBE0         |
| U  | (f) <sup>0</sup>                                                                               | 179.5(±1.0)        | 179   | 179.4(184.7) | 176.3(180.8) | 244.6(±0.6)        | 242   | 253.2(254.2) | 250.7(251.4) |
| Np | (f <sub>φ1</sub> ) <sup>1</sup>                                                                | 177.8(±1.5)        | 178   | 178.4(183.7) | 174.9(179.1) | 243.8(±0.9)        | 242   | 252.6(253.8) | 250.2(251.0) |
| Pu | (f <sub>δ</sub> ) <sup>2</sup>                                                                 | 176.2(±1.2)        | 175   | 175.6(180.7) | 171.7(176.1) | 244.6(±1.0)        | 246   | 252.7(257.6) | 249.4(252.2) |
| Am | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup>                                 | 175.0±(1.2)        | –     | 175.5(180.9) | 171.2(175.5) | 243.1(±1.2)        | –     | 252.7(258.0) | 249.5(252.2) |
| Cm | (f <sub>δ</sub> ) <sup>2</sup> (f <sub>φ1</sub> ) <sup>1</sup> (f <sub>φ2</sub> ) <sup>1</sup> | –                  | –     | 175.6(183.2) | 171.1(176.4) | –                  | –     | 255.4(270.2) | 252.4(257.8) |

**Table S12.** Full reference No. 17

Gaussian 09, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009

**Table S13.** Electron configurations of the (non-specified) S-L-J ground states of chemically unbound free atomic cations of charge  $q+$  in physical vacuum from the first half-group of the lanthanide and actinide elements. <sup>a</sup>

| $q+$ | 1+                                                | 2+                                  | 3+              | 4+              | 5+                                   | 6+                                   | 7+                                   | 8+                                   |
|------|---------------------------------------------------|-------------------------------------|-----------------|-----------------|--------------------------------------|--------------------------------------|--------------------------------------|--------------------------------------|
| La   | <u>5d<sup>2</sup></u>                             | <u>5d<sup>1</sup></u>               | 4f <sup>0</sup> |                 |                                      |                                      |                                      |                                      |
| Ce   | <u>4f<sup>1</sup>5d<sup>2</sup></u>               | 4f <sup>2</sup>                     | 4f <sup>1</sup> | 4f <sup>0</sup> |                                      |                                      |                                      |                                      |
| Pr   | <u>4f<sup>3</sup>6s<sup>1</sup></u>               | 4f <sup>3</sup>                     | 4f <sup>2</sup> | 4f <sup>1</sup> | 4f <sup>0</sup>                      | <u>4f<sup>1</sup>5p<sup>-2</sup></u> | <u>4f<sup>1</sup>5p<sup>-3</sup></u> | <u>4f<sup>1</sup>5p<sup>-4</sup></u> |
| Nd   | <u>4f<sup>4</sup>6s<sup>1</sup></u>               | 4f <sup>4</sup>                     | 4f <sup>3</sup> | 4f <sup>2</sup> | <u>4f<sup>2</sup>5p<sup>-1</sup></u> | <u>4f<sup>2</sup>5p<sup>-2</sup></u> | <u>4f<sup>2</sup>5p<sup>-3</sup></u> | <u>4f<sup>2</sup>5p<sup>-4</sup></u> |
| Pm   | <u>4f<sup>5</sup>6s<sup>1</sup></u>               | 4f <sup>5</sup>                     | 4f <sup>4</sup> | 4f <sup>3</sup> | <u>4f<sup>3</sup>5p<sup>-1</sup></u> | <u>4f<sup>3</sup>5p<sup>-2</sup></u> | <u>4f<sup>3</sup>5p<sup>-3</sup></u> | <u>4f<sup>3</sup>5p<sup>-4</sup></u> |
| Sm   | <u>4f<sup>6</sup>6s<sup>1</sup></u>               | 4f <sup>6</sup>                     | 4f <sup>5</sup> | 4f <sup>4</sup> | 4f <sup>3</sup>                      | <u>4f<sup>4</sup>5p<sup>-2</sup></u> | <u>4f<sup>4</sup>5p<sup>-3</sup></u> | <u>4f<sup>4</sup>5p<sup>-4</sup></u> |
| Eu   | <u>4f<sup>7</sup>6s<sup>1</sup></u>               | 4f <sup>7</sup>                     | 4f <sup>6</sup> | 4f <sup>5</sup> | <u>4f<sup>5</sup>5p<sup>-1</sup></u> | <u>4f<sup>5</sup>5p<sup>-2</sup></u> | <u>4f<sup>3</sup>5p<sup>-1</sup></u> | <u>4f<sup>3</sup>5p<sup>-4</sup></u> |
| Gd   | <u>4f<sup>7</sup>5d<sup>1</sup>6s<sup>1</sup></u> | <u>4f<sup>7</sup>5d<sup>1</sup></u> | 4f <sup>7</sup> | 4f <sup>6</sup> | <u>4f<sup>6</sup>5p<sup>-1</sup></u> | <u>4f<sup>6</sup>5p<sup>-2</sup></u> | <u>4f<sup>6</sup>5p<sup>-3</sup></u> | <u>4f<sup>6</sup>5p<sup>-4</sup></u> |

| $q+$ | 1+                                                | 2+                                  | 3+              | 4+              | 5+              | 6+                                   | 7+                                   | 8+                                   |
|------|---------------------------------------------------|-------------------------------------|-----------------|-----------------|-----------------|--------------------------------------|--------------------------------------|--------------------------------------|
| Ac   | <u>7s<sup>2</sup></u>                             | <u>7s<sup>1</sup></u>               | 5f <sup>0</sup> |                 |                 |                                      |                                      |                                      |
| Th   | <u>6d<sup>1</sup>7s<sup>2</sup></u>               | <u>5f<sup>1</sup>6d<sup>1</sup></u> | 5f <sup>1</sup> | 5f <sup>0</sup> |                 |                                      |                                      |                                      |
| Pa   | <u>5f<sup>2</sup>7s<sup>2</sup></u>               | <u>5f<sup>2</sup>6d<sup>1</sup></u> | 5f <sup>2</sup> | 5f <sup>1</sup> | 5f <sup>0</sup> |                                      |                                      |                                      |
| U    | <u>5f<sup>3</sup>7s<sup>2</sup></u>               | 5f <sup>4</sup>                     | 5f <sup>3</sup> | 5f <sup>2</sup> | 5f <sup>1</sup> | 5f <sup>0</sup>                      |                                      |                                      |
| Np   | <u>5f<sup>4</sup>6d<sup>1</sup>7s<sup>1</sup></u> | 5f <sup>5</sup>                     | 5f <sup>4</sup> | 5f <sup>3</sup> | 5f <sup>2</sup> | 5f <sup>1</sup>                      | <u>5f<sup>1</sup>6p<sup>-1</sup></u> | <u>5f<sup>1</sup>6p<sup>-2</sup></u> |
| Pu   | <u>5f<sup>6</sup>7s<sup>1</sup></u>               | 5f <sup>6</sup>                     | 5f <sup>5</sup> | 5f <sup>4</sup> | 5f <sup>3</sup> | <u>5f<sup>3</sup>6p<sup>-1</sup></u> | <u>5f<sup>3</sup>6p<sup>-2</sup></u> | <u>5f<sup>3</sup>6p<sup>-3</sup></u> |
| Am   | <u>5f<sup>7</sup>7s<sup>1</sup></u>               | 5f <sup>7</sup>                     | 5f <sup>6</sup> | 5f <sup>5</sup> | 5f <sup>4</sup> | 5f <sup>3</sup>                      | <u>5f<sup>3</sup>6p<sup>-1</sup></u> | <u>5f<sup>3</sup>6p<sup>-2</sup></u> |
| Cm   | <u>5f<sup>7</sup>7s<sup>2</sup></u>               | 5f <sup>8</sup>                     | 5f <sup>7</sup> | 5f <sup>6</sup> | 5f <sup>5</sup> | 5f <sup>4</sup>                      | <u>5f<sup>5</sup>6p<sup>-2</sup></u> | <u>5f<sup>5</sup>6p<sup>-3</sup></u> |

<sup>a</sup> **Bold face:** ‘Nonsystematic’ configurations of the ground states of the 4f- and 5f-elemental free atomic cations with, respectively, 5d6s- and 6d7s-valence occupations, or with 5p- and 6p-core holes. Empty entries have empty f<sup>0</sup>d<sup>0</sup>s<sup>0</sup> valence shells and the expected number of p core holes. The (n-1)p<sup>-x</sup> symbol means that x electron(s) are ionized from the (n-1)p<sup>6</sup> outer core shell.

**Table S14.** Atomic coordinates of the optimized  $[(\text{AnO}_2)(\text{CO}_3)_3]^{4-}$  and  $[(\text{AnO}_2)(\text{CO}_3)_3][\text{C}(\text{NH}_2)_3]_9^{5+}$  species using the PBE0 density functional.

**$\text{UO}_2(\text{CO}_3)_3^{4-}$**

|      |           |           |           |
|------|-----------|-----------|-----------|
| 1.U  | 0.000000  | 0.000000  | 0.000000  |
| 2.O  | 0.000000  | 0.000000  | 1.808061  |
| 3.O  | 0.000000  | 0.000000  | -1.808061 |
| 4.O  | -2.124960 | -3.680538 | 0.000000  |
| 5.O  | -2.078811 | -1.413937 | 0.000000  |
| 6.O  | -0.185100 | -2.507271 | 0.000000  |
| 7.O  | 2.263911  | -1.093335 | 0.000000  |
| 8.O  | 4.249919  | 0.000000  | 0.000000  |
| 9.O  | -2.078811 | 1.413937  | 0.000000  |
| 10.O | -2.124960 | 3.680538  | 0.000000  |
| 11.O | -0.185100 | 2.507271  | 0.000000  |
| 12.O | 2.263911  | 1.093335  | 0.000000  |
| 13.C | -1.485113 | -2.572292 | 0.000000  |
| 14.C | 2.970227  | 0.000000  | 0.000000  |
| 15.C | -1.485113 | 2.572292  | 0.000000  |

**$\text{NpO}_2(\text{CO}_3)_3^{4-}$**

|      |           |           |           |
|------|-----------|-----------|-----------|
| 1.Np | 0.000000  | 0.000000  | 0.000000  |
| 2.O  | 0.000000  | 0.000000  | 1.791489  |
| 3.O  | 0.000000  | 0.000000  | -1.791489 |
| 4.O  | -2.123145 | -3.677394 | 0.000000  |
| 5.O  | -2.077064 | -1.410062 | 0.000000  |
| 6.O  | -0.182617 | -2.503821 | 0.000000  |
| 7.O  | 2.259681  | -1.093759 | 0.000000  |
| 8.O  | 4.246289  | 0.000000  | 0.000000  |
| 9.O  | -2.077064 | 1.410062  | 0.000000  |
| 10.O | -2.123145 | 3.677394  | 0.000000  |
| 11.O | -0.182617 | 2.503821  | 0.000000  |
| 12.O | 2.259681  | 1.093759  | 0.000000  |
| 13.C | -1.482404 | -2.567599 | 0.000000  |
| 14.C | 2.964808  | 0.000000  | 0.000000  |
| 15.C | -1.482404 | 2.567599  | 0.000000  |

**$\text{PuO}_2(\text{CO}_3)_3^{4-}$**

|      |           |           |           |
|------|-----------|-----------|-----------|
| 1.Pu | 0.000000  | 0.000000  | 0.000000  |
| 2.O  | 0.000000  | 0.000000  | 1.760761  |
| 3.O  | 0.000000  | 0.000000  | -1.760761 |
| 4.O  | -2.128791 | -3.687175 | 0.000000  |
| 5.O  | -2.082981 | -1.422530 | 0.000000  |
| 6.O  | -0.190457 | -2.515179 | 0.000000  |
| 7.O  | 2.273438  | -1.092649 | 0.000000  |
| 8.O  | 4.257583  | 0.000000  | 0.000000  |
| 9.O  | -2.082981 | 1.422530  | 0.000000  |
| 10.O | -2.128791 | 3.687175  | 0.000000  |
| 11.O | -0.190457 | 2.515179  | 0.000000  |
| 12.O | 2.273438  | 1.092649  | 0.000000  |
| 13.C | -1.487704 | -2.576779 | 0.000000  |
| 14.C | 2.975408  | 0.000000  | 0.000000  |
| 15.C | -1.487704 | 2.576779  | 0.000000  |

**AmO<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub><sup>4-</sup>**

|      |           |           |           |
|------|-----------|-----------|-----------|
| 1.Am | 0.000000  | 0.000000  | 0.000000  |
| 2.O  | 0.000000  | 0.000000  | 1.754944  |
| 3.O  | 0.000000  | 0.000000  | -1.754944 |
| 4.O  | -2.128604 | -3.686851 | 0.000000  |
| 5.O  | -2.083524 | -1.421578 | 0.000000  |
| 6.O  | -0.189361 | -2.515174 | 0.000000  |
| 7.O  | 2.272884  | -1.093596 | 0.000000  |
| 8.O  | 4.257209  | 0.000000  | 0.000000  |
| 9.O  | -2.083524 | 1.421578  | 0.000000  |
| 10.O | -2.128604 | 3.686851  | 0.000000  |
| 11.O | -0.189361 | 2.515174  | 0.000000  |
| 12.O | 2.272884  | 1.093596  | 0.000000  |
| 13.C | -1.486775 | -2.575169 | 0.000000  |
| 14.C | 2.973549  | 0.000000  | 0.000000  |
| 15.C | -1.486775 | 2.575169  | 0.000000  |

**[UO<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub>][C(NH<sub>2</sub>)<sub>3</sub>]<sub>9</sub><sup>5+</sup>**

|           |           |           |           |
|-----------|-----------|-----------|-----------|
| 1.U       | 0.000000  | 0.000000  | 0.155164  |
| 2.O       | 0.000000  | 0.000000  | 1.918051  |
| 3.O       | 0.000000  | 0.000000  | -1.608101 |
| 4.O       | -2.077302 | 1.401140  | 0.067488  |
| 5.O       | -0.174772 | 2.499567  | 0.067488  |
| 6.O       | -2.077298 | 3.597985  | -0.377781 |
| 7.C.tz2p  | -1.464353 | 2.536333  | -0.087666 |
| 8.O       | -0.174772 | -2.499567 | 0.067488  |
| 9.O       | -2.077302 | -1.401140 | 0.067488  |
| 10.O      | -2.077298 | -3.597985 | -0.377781 |
| 11.C.tz2p | -1.464353 | -2.536333 | -0.087666 |
| 12.O      | 2.252074  | 1.098426  | 0.067488  |
| 13.O      | 2.252074  | -1.098426 | 0.067488  |
| 14.O      | 4.154596  | 0.000000  | -0.377781 |
| 15.C.tz2p | 2.928706  | 0.000000  | -0.087666 |
| 16.C.dzp  | -0.231366 | 5.991089  | 1.373710  |
| 17.N      | -0.912143 | 6.005892  | 0.237888  |
| 18.H      | -1.158746 | 6.889887  | -0.192795 |
| 19.H      | -1.355869 | 5.121709  | -0.084892 |
| 20.N      | 0.231856  | 4.826542  | 1.830840  |
| 21.H      | 0.141451  | 3.982334  | 1.255830  |
| 22.H      | 0.439409  | 4.726513  | 2.816261  |
| 23.N      | 0.010787  | 7.111308  | 2.046500  |
| 24.H      | -0.467335 | 7.978216  | 1.822894  |
| 25.C.dzp  | -0.231366 | -5.991089 | 1.373710  |
| 26.N      | -0.912143 | -6.005892 | 0.237888  |
| 27.H      | -1.355869 | -5.121709 | -0.084892 |
| 28.H      | -1.158746 | -6.889887 | -0.192795 |
| 29.N      | 0.010787  | -7.111308 | 2.046500  |
| 30.H      | -0.467335 | -7.978216 | 1.822894  |
| 31.N      | 0.231856  | -4.826542 | 1.830840  |
| 32.H      | 0.141451  | -3.982334 | 1.255830  |
| 33.H      | 0.439409  | -4.726513 | 2.816261  |
| 34.C.dzp  | 5.304118  | 2.795176  | 1.373710  |
| 35.N      | 5.657327  | 2.213007  | 0.237888  |
| 36.H      | 5.113464  | 1.386637  | -0.084892 |
| 37.H      | 6.546190  | 2.441440  | -0.192795 |
| 38.N      | 6.153180  | 3.564996  | 2.046500  |

|          |           |           |           |
|----------|-----------|-----------|-----------|
| 39.H     | 7.143005  | 3.584384  | 1.822894  |
| 40.H     | 5.835934  | 4.187932  | 2.780900  |
| 41.N     | 4.063980  | 2.614065  | 1.830840  |
| 42.H     | 3.378077  | 2.113667  | 1.255830  |
| 43.H     | 3.873576  | 2.743795  | 2.816261  |
| 44.C.dzp | -5.072752 | -3.195913 | 1.373710  |
| 45.N     | -4.745183 | -3.792885 | 0.237888  |
| 46.H     | -5.387444 | -4.448447 | -0.192795 |
| 47.H     | -3.757595 | -3.735072 | -0.084892 |
| 48.N     | -4.295837 | -2.212478 | 1.830840  |
| 49.H     | -3.519528 | -1.868667 | 1.255830  |
| 50.H     | -4.312985 | -1.982717 | 2.816261  |
| 51.N     | -6.163967 | -3.546312 | 2.046500  |
| 52.H     | -6.544823 | -2.960101 | 2.780900  |
| 53.C.dzp | 5.304118  | -2.795176 | 1.373710  |
| 54.N     | 5.657327  | -2.213007 | 0.237888  |
| 55.H     | 6.546190  | -2.441440 | -0.192795 |
| 56.H     | 5.113464  | -1.386637 | -0.084892 |
| 57.N     | 4.063980  | -2.614065 | 1.830840  |
| 58.H     | 3.378077  | -2.113667 | 1.255830  |
| 59.H     | 3.873576  | -2.743795 | 2.816261  |
| 60.N     | 6.153180  | -3.564996 | 2.046500  |
| 61.H     | 7.143005  | -3.584384 | 1.822894  |
| 62.C.dzp | -5.072752 | 3.195913  | 1.373710  |
| 63.N     | -4.745183 | 3.792885  | 0.237888  |
| 64.H     | -3.757595 | 3.735072  | -0.084892 |
| 65.H     | -5.387444 | 4.448447  | -0.192795 |
| 66.N     | -6.163967 | 3.546312  | 2.046500  |
| 67.H     | -6.675670 | 4.393832  | 1.822894  |
| 68.H     | -6.544823 | 2.960101  | 2.780900  |
| 69.N     | -4.295837 | 2.212478  | 1.830840  |
| 70.H     | -3.519528 | 1.868667  | 1.255830  |
| 71.H     | -4.312985 | 1.982717  | 2.816261  |
| 72.C.dzp | -4.104555 | 0.000000  | -2.763235 |
| 73.N     | -4.826268 | 0.000000  | -3.881948 |
| 74.H     | -5.148604 | 0.861544  | -4.307619 |
| 75.H     | -5.148604 | -0.861544 | -4.307619 |
| 76.N     | -3.745379 | -1.145416 | -2.199583 |
| 77.H     | -3.989932 | -2.023168 | -2.637552 |
| 78.H     | -3.151764 | -1.171294 | -1.358340 |
| 79.N     | -3.745379 | 1.145416  | -2.199583 |
| 80.H     | -3.989932 | 2.023168  | -2.637552 |
| 81.H     | -3.151764 | 1.171294  | -1.358340 |
| 82.C.dzp | 2.052278  | -3.554649 | -2.763235 |
| 83.N     | 2.413134  | -4.179671 | -3.881948 |
| 84.H     | 1.828183  | -4.889594 | -4.307619 |
| 85.H     | 3.320421  | -4.028050 | -4.307619 |
| 86.N     | 2.864649  | -2.670886 | -2.199583 |
| 87.H     | 3.747081  | -2.443798 | -2.637552 |
| 88.H     | 2.590252  | -2.143860 | -1.358340 |
| 89.N     | 0.880731  | -3.816302 | -2.199583 |
| 90.H     | 0.242851  | -4.466966 | -2.637552 |
| 91.H     | 0.561512  | -3.315154 | -1.358340 |
| 92.C.dzp | 2.052278  | 3.554649  | -2.763235 |
| 93.N     | 2.413134  | 4.179671  | -3.881948 |
| 94.H     | 3.320421  | 4.028050  | -4.307619 |

|       |           |           |           |
|-------|-----------|-----------|-----------|
| 95.H  | 1.828183  | 4.889594  | -4.307619 |
| 96.N  | 0.880731  | 3.816302  | -2.199583 |
| 97.H  | 0.242851  | 4.466966  | -2.637552 |
| 98.H  | 0.561512  | 3.315154  | -1.358340 |
| 99.N  | 2.864649  | 2.670886  | -2.199583 |
| 100.H | 3.747081  | 2.443798  | -2.637552 |
| 101.H | 2.590252  | 2.143860  | -1.358340 |
| 102.H | 5.835934  | -4.187932 | 2.780900  |
| 103.H | 0.708889  | -7.148033 | 2.780900  |
| 104.H | -6.675670 | -4.393832 | 1.822894  |
| 105.H | 0.708889  | 7.148033  | 2.780900  |

**[NpO<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub>][C(NH<sub>2</sub>)<sub>3</sub>]<sub>9</sub><sup>5+</sup>:**

|           |           |           |           |
|-----------|-----------|-----------|-----------|
| 1.Np      | 0.000000  | 0.000000  | 0.170049  |
| 2.O       | 0.000000  | 0.000000  | 1.919513  |
| 3.O       | 0.000000  | 0.000000  | -1.579298 |
| 4.O       | -2.074612 | 1.396252  | 0.082940  |
| 5.O       | -0.171884 | 2.494792  | 0.082940  |
| 6.O       | -2.072383 | 3.589473  | -0.380003 |
| 7.C.tz2p  | -1.460481 | 2.529627  | -0.077194 |
| 8.O       | -0.171884 | -2.494792 | 0.082940  |
| 9.O       | -2.074612 | -1.396252 | 0.082940  |
| 10.O      | -2.072383 | -3.589473 | -0.380003 |
| 11.C.tz2p | -1.460481 | -2.529627 | -0.077194 |
| 12.O      | 2.246495  | 1.098540  | 0.082940  |
| 13.O      | 2.246495  | -1.098540 | 0.082940  |
| 14.O      | 4.144766  | 0.000000  | -0.380003 |
| 15.C.tz2p | 2.920962  | 0.000000  | -0.077194 |
| 16.C.dzp  | -0.212620 | 5.997253  | 1.337919  |
| 17.N      | -0.902532 | 5.997828  | 0.207366  |
| 18.H      | -1.159790 | 6.876937  | -0.226840 |
| 19.H      | -1.354971 | 5.110251  | -0.093970 |
| 20.N      | 0.253438  | 4.838238  | 1.805644  |
| 21.H      | 0.150856  | 3.984805  | 1.245023  |
| 22.H      | 0.472476  | 4.752613  | 2.790014  |
| 23.N      | 0.035342  | 7.126022  | 1.994326  |
| 24.H      | -0.442374 | 7.990483  | 1.762284  |
| 25.C.dzp  | -0.212620 | -5.997253 | 1.337919  |
| 26.N      | -0.902532 | -5.997828 | 0.207366  |
| 27.H      | -1.354971 | -5.110251 | -0.093970 |
| 28.H      | -1.159790 | -6.876937 | -0.226840 |
| 29.N      | 0.035342  | -7.126022 | 1.994326  |
| 30.H      | -0.442374 | -7.990483 | 1.762284  |
| 31.N      | 0.253438  | -4.838238 | 1.805644  |
| 32.H      | 0.150856  | -3.984805 | 1.245023  |
| 33.H      | 0.472476  | -4.752613 | 2.790014  |
| 34.C.dzp  | 5.300083  | 2.814493  | 1.337919  |
| 35.N      | 5.645537  | 2.217299  | 0.207366  |
| 36.H      | 5.103093  | 1.381687  | -0.093970 |
| 37.H      | 6.535497  | 2.434061  | -0.226840 |
| 38.N      | 6.153645  | 3.593618  | 1.994326  |
| 39.H      | 7.141148  | 3.612134  | 1.762284  |
| 40.H      | 5.842298  | 4.222261  | 2.725915  |
| 41.N      | 4.063318  | 2.638603  | 1.805644  |
| 42.H      | 3.375514  | 2.123048  | 1.245023  |
| 43.H      | 3.879646  | 2.785483  | 2.790014  |

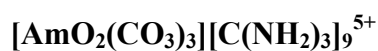
|          |           |           |           |
|----------|-----------|-----------|-----------|
| 44.C.dzp | -5.087464 | -3.182761 | 1.337919  |
| 45.N     | -4.743006 | -3.780529 | 0.207366  |
| 46.H     | -5.375707 | -4.442876 | -0.226840 |
| 47.H     | -3.748122 | -3.728565 | -0.093970 |
| 48.N     | -4.316756 | -2.199635 | 1.805644  |
| 49.H     | -3.526371 | -1.861757 | 1.245023  |
| 50.H     | -4.352122 | -1.967131 | 2.790014  |
| 51.N     | -6.188987 | -3.532403 | 1.994326  |
| 52.H     | -6.577734 | -2.948448 | 2.725915  |
| 53.C.dzp | 5.300083  | -2.814493 | 1.337919  |
| 54.N     | 5.645537  | -2.217299 | 0.207366  |
| 55.H     | 6.535497  | -2.434061 | -0.226840 |
| 56.H     | 5.103093  | -1.381687 | -0.093970 |
| 57.N     | 4.063318  | -2.638603 | 1.805644  |
| 58.H     | 3.375514  | -2.123048 | 1.245023  |
| 59.H     | 3.879646  | -2.785483 | 2.790014  |
| 60.N     | 6.153645  | -3.593618 | 1.994326  |
| 61.H     | 7.141148  | -3.612134 | 1.762284  |
| 62.C.dzp | -5.087464 | 3.182761  | 1.337919  |
| 63.N     | -4.743006 | 3.780529  | 0.207366  |
| 64.H     | -3.748122 | 3.728565  | -0.093970 |
| 65.H     | -5.375707 | 4.442876  | -0.226840 |
| 66.N     | -6.188987 | 3.532403  | 1.994326  |
| 67.H     | -6.698774 | 4.378349  | 1.762284  |
| 68.H     | -6.577734 | 2.948448  | 2.725915  |
| 69.N     | -4.316756 | 2.199635  | 1.805644  |
| 70.H     | -3.526371 | 1.861757  | 1.245023  |
| 71.H     | -4.352122 | 1.967131  | 2.790014  |
| 72.C.dzp | -4.152076 | 0.000000  | -2.701934 |
| 73.N     | -4.892548 | 0.000000  | -3.808618 |
| 74.H     | -5.232440 | 0.861367  | -4.220885 |
| 75.H     | -5.232440 | -0.861367 | -4.220885 |
| 76.N     | -3.781973 | -1.145142 | -2.144809 |
| 77.H     | -4.033082 | -2.022893 | -2.579016 |
| 78.H     | -3.173870 | -1.170347 | -1.313350 |
| 79.N     | -3.781973 | 1.145142  | -2.144809 |
| 80.H     | -4.033082 | 2.022893  | -2.579016 |
| 81.H     | -3.173870 | 1.170347  | -1.313350 |
| 82.C.dzp | 2.076038  | -3.595804 | -2.701934 |
| 83.N     | 2.446274  | -4.237070 | -3.808618 |
| 84.H     | 1.870255  | -4.962110 | -4.220885 |
| 85.H     | 3.362186  | -4.100743 | -4.220885 |
| 86.N     | 2.882709  | -2.702714 | -2.144809 |
| 87.H     | 3.768418  | -2.481305 | -2.579016 |
| 88.H     | 2.600485  | -2.163478 | -1.313350 |
| 89.N     | 0.899265  | -3.847856 | -2.144809 |
| 90.H     | 0.264664  | -4.504198 | -2.579016 |
| 91.H     | 0.573384  | -3.333825 | -1.313350 |
| 92.C.dzp | 2.076038  | 3.595804  | -2.701934 |
| 93.N     | 2.446274  | 4.237070  | -3.808618 |
| 94.H     | 3.362186  | 4.100743  | -4.220885 |
| 95.H     | 1.870255  | 4.962110  | -4.220885 |
| 96.N     | 0.899265  | 3.847856  | -2.144809 |
| 97.H     | 0.264664  | 4.504198  | -2.579016 |
| 98.H     | 0.573384  | 3.333825  | -1.313350 |
| 99.N     | 2.882709  | 2.702714  | -2.144809 |

|       |           |           |           |
|-------|-----------|-----------|-----------|
| 100.H | 3.768418  | 2.481305  | -2.579016 |
| 101.H | 2.600485  | 2.163478  | -1.313350 |
| 102.H | 5.842298  | -4.222261 | 2.725915  |
| 103.H | 0.735436  | -7.170709 | 2.725915  |
| 104.H | -6.698774 | -4.378349 | 1.762284  |
| 105.H | 0.735436  | 7.170709  | 2.725915  |

**[PuO<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub>][C(NH<sub>2</sub>)<sub>3</sub>]<sub>9</sub><sup>5+</sup>**

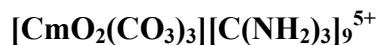
|           |           |           |           |
|-----------|-----------|-----------|-----------|
| 1.Pu      | 0.000000  | 0.000000  | 0.166609  |
| 2.O       | 0.000000  | 0.000000  | 1.883476  |
| 3.O       | 0.000000  | 0.000000  | -1.551077 |
| 4.O       | -2.069489 | 1.390075  | 0.084963  |
| 5.O       | -0.169096 | 2.487267  | 0.084963  |
| 6.O       | -2.068594 | 3.582909  | -0.375026 |
| 7.C.tz2p  | -1.456196 | 2.522205  | -0.074502 |
| 8.O       | -0.169096 | -2.487267 | 0.084963  |
| 9.O       | -2.069489 | -1.390075 | 0.084963  |
| 10.O      | -2.068594 | -3.582909 | -0.375026 |
| 11.C.tz2p | -1.456196 | -2.522205 | -0.074502 |
| 12.O      | 2.238585  | 1.097192  | 0.084963  |
| 13.O      | 2.238585  | -1.097192 | 0.084963  |
| 14.O      | 4.137187  | 0.000000  | -0.375026 |
| 15.C.tz2p | 2.912391  | 0.000000  | -0.074502 |
| 16.C.dzp  | -0.209059 | 5.987693  | 1.353337  |
| 17.N      | -0.897820 | 5.993044  | 0.222104  |
| 18.H      | -1.148073 | 6.873692  | -0.213194 |
| 19.H      | -1.345525 | 5.106843  | -0.089947 |
| 20.N      | 0.255923  | 4.826859  | 1.819199  |
| 21.H      | 0.156512  | 3.975278  | 1.255400  |
| 22.H      | 0.468575  | 4.738486  | 2.804939  |
| 23.N      | 0.040239  | 7.114329  | 2.013359  |
| 24.H      | -0.436240 | 7.980676  | 1.783453  |
| 25.C.dzp  | -0.209059 | -5.987693 | 1.353337  |
| 26.N      | -0.897820 | -5.993044 | 0.222104  |
| 27.H      | -1.345525 | -5.106843 | -0.089947 |
| 28.H      | -1.148073 | -6.873692 | -0.213194 |
| 29.N      | 0.040239  | -7.114329 | 2.013359  |
| 30.H      | -0.436240 | -7.980676 | 1.783453  |
| 31.N      | 0.255923  | -4.826859 | 1.819199  |
| 32.H      | 0.156512  | -3.975278 | 1.255400  |
| 33.H      | 0.468575  | -4.738486 | 2.804939  |
| 34.C.dzp  | 5.290023  | 2.812796  | 1.353337  |
| 35.N      | 5.639039  | 2.218987  | 0.222104  |
| 36.H      | 5.095418  | 1.388163  | -0.089947 |
| 37.H      | 6.526828  | 2.442586  | -0.213194 |
| 38.N      | 6.141070  | 3.592013  | 2.013359  |
| 39.H      | 7.129588  | 3.612543  | 1.783453  |
| 40.H      | 5.825919  | 4.222875  | 2.742000  |
| 41.N      | 4.052221  | 2.635065  | 1.819199  |
| 42.H      | 3.364435  | 2.123183  | 1.255400  |
| 43.H      | 3.869362  | 2.775041  | 2.804939  |
| 44.C.dzp  | -5.080965 | -3.174896 | 1.353337  |
| 45.N      | -4.741218 | -3.774057 | 0.222104  |
| 46.H      | -5.378755 | -4.431106 | -0.213194 |
| 47.H      | -3.749893 | -3.718680 | -0.089947 |
| 48.N      | -4.308144 | -2.191793 | 1.819199  |

|          |           |           |           |
|----------|-----------|-----------|-----------|
| 49.H     | -3.520948 | -1.852095 | 1.255400  |
| 50.H     | -4.337937 | -1.963445 | 2.804939  |
| 51.N     | -6.181310 | -3.522316 | 2.013359  |
| 52.H     | -6.570077 | -2.933956 | 2.742000  |
| 53.C.dzp | 5.290023  | -2.812796 | 1.353337  |
| 54.N     | 5.639039  | -2.218987 | 0.222104  |
| 55.H     | 6.526828  | -2.442586 | -0.213194 |
| 56.H     | 5.095418  | -1.388163 | -0.089947 |
| 57.N     | 4.052221  | -2.635065 | 1.819199  |
| 58.H     | 3.364435  | -2.123183 | 1.255400  |
| 59.H     | 3.869362  | -2.775041 | 2.804939  |
| 60.N     | 6.141070  | -3.592013 | 2.013359  |
| 61.H     | 7.129588  | -3.612543 | 1.783453  |
| 62.C.dzp | -5.080965 | 3.174896  | 1.353337  |
| 63.N     | -4.741218 | 3.774057  | 0.222104  |
| 64.H     | -3.749893 | 3.718680  | -0.089947 |
| 65.H     | -5.378755 | 4.431106  | -0.213194 |
| 66.N     | -6.181310 | 3.522316  | 2.013359  |
| 67.H     | -6.693348 | 4.368133  | 1.783453  |
| 68.H     | -6.570077 | 2.933956  | 2.742000  |
| 69.N     | -4.308144 | 2.191793  | 1.819199  |
| 70.H     | -3.520948 | 1.852095  | 1.255400  |
| 71.H     | -4.337937 | 1.963445  | 2.804939  |
| 72.C.dzp | -4.107261 | 0.000000  | -2.732755 |
| 73.N     | -4.845398 | 0.000000  | -3.840811 |
| 74.H     | -5.174793 | 0.861518  | -4.261098 |
| 75.H     | -5.174793 | -0.861518 | -4.261098 |
| 76.N     | -3.736953 | -1.144825 | -2.175474 |
| 77.H     | -3.999265 | -2.023424 | -2.601345 |
| 78.H     | -3.139548 | -1.169024 | -1.335582 |
| 79.N     | -3.736953 | 1.144825  | -2.175474 |
| 80.H     | -3.999265 | 2.023424  | -2.601345 |
| 81.H     | -3.139548 | 1.169024  | -1.335582 |
| 82.C.dzp | 2.053630  | -3.556992 | -2.732755 |
| 83.N     | 2.422699  | -4.196238 | -3.840811 |
| 84.H     | 1.841300  | -4.912262 | -4.261098 |
| 85.H     | 3.333494  | -4.050743 | -4.261098 |
| 86.N     | 2.859924  | -2.663883 | -2.175474 |
| 87.H     | 3.751969  | -2.451753 | -2.601345 |
| 88.H     | 2.582179  | -2.134417 | -1.335582 |
| 89.N     | 0.877029  | -3.808709 | -2.175474 |
| 90.H     | 0.247296  | -4.475177 | -2.601345 |
| 91.H     | 0.557370  | -3.303441 | -1.335582 |
| 92.C.dzp | 2.053630  | 3.556992  | -2.732755 |
| 93.N     | 2.422699  | 4.196238  | -3.840811 |
| 94.H     | 3.333494  | 4.050743  | -4.261098 |
| 95.H     | 1.841300  | 4.912262  | -4.261098 |
| 96.N     | 0.877029  | 3.808709  | -2.175474 |
| 97.H     | 0.247296  | 4.475177  | -2.601345 |
| 98.H     | 0.557370  | 3.303441  | -1.335582 |
| 99.N     | 2.859924  | 2.663883  | -2.175474 |
| 100.H    | 3.751969  | 2.451753  | -2.601345 |
| 101.H    | 2.582179  | 2.134417  | -1.335582 |
| 102.H    | 5.825919  | -4.222875 | 2.742000  |
| 103.H    | 0.744158  | -7.156832 | 2.742000  |
| 104.H    | -6.693348 | -4.368133 | 1.783453  |
| 105.H    | 0.744158  | 7.156832  | 2.742000  |



|           |           |           |           |
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| 1.Am      | 0.000000  | 0.000000  | 0.174552  |
| 2.O       | 0.000000  | 0.000000  | 1.886374  |
| 3.O       | 0.000000  | 0.000000  | -1.537654 |
| 4.O       | -2.070232 | 1.390033  | 0.093118  |
| 5.O       | -0.168688 | 2.487891  | 0.093118  |
| 6.O       | -2.067124 | 3.580364  | -0.380815 |
| 7.C.tz2p  | -1.455203 | 2.520486  | -0.070580 |
| 8.O       | -0.168688 | -2.487891 | 0.093118  |
| 9.O       | -2.070232 | -1.390033 | 0.093118  |
| 10.O      | -2.067124 | -3.580364 | -0.380815 |
| 11.C.tz2p | -1.455203 | -2.520486 | -0.070580 |
| 12.O      | 2.238920  | 1.097857  | 0.093118  |
| 13.O      | 2.238920  | -1.097857 | 0.093118  |
| 14.O      | 4.134248  | 0.000000  | -0.380815 |
| 15.C.tz2p | 2.910406  | 0.000000  | -0.070580 |
| 16.C.dzp  | -0.218220 | 5.999438  | 1.328343  |
| 17.N      | -0.913407 | 5.995729  | 0.201215  |
| 18.H      | -1.175042 | 6.873190  | -0.233357 |
| 19.H      | -1.358491 | 5.104608  | -0.102357 |
| 20.N      | 0.253157  | 4.842973  | 1.797017  |
| 21.H      | 0.154138  | 3.987405  | 1.237558  |
| 22.H      | 0.477938  | 4.762026  | 2.780379  |
| 23.N      | 0.030256  | 7.130460  | 1.981134  |
| 24.H      | -0.449886 | 7.993880  | 1.748637  |
| 25.C.dzp  | -0.218220 | -5.999438 | 1.328343  |
| 26.N      | -0.913407 | -5.995729 | 0.201215  |
| 27.H      | -1.358491 | -5.104608 | -0.102357 |
| 28.H      | -1.175042 | -6.873190 | -0.233357 |
| 29.N      | 0.030256  | -7.130460 | 1.981134  |
| 30.H      | -0.449886 | -7.993880 | 1.748637  |
| 31.N      | 0.253157  | -4.842973 | 1.797017  |
| 32.H      | 0.154138  | -3.987405 | 1.237558  |
| 33.H      | 0.477938  | -4.762026 | 2.780379  |
| 34.C.dzp  | 5.304776  | 2.810735  | 1.328343  |
| 35.N      | 5.649157  | 2.206831  | 0.201215  |
| 36.H      | 5.099966  | 1.375816  | -0.102357 |
| 37.H      | 6.539878  | 2.418979  | -0.233357 |
| 38.N      | 6.160031  | 3.591432  | 1.981134  |
| 39.H      | 7.147846  | 3.607327  | 1.748637  |
| 40.H      | 5.849160  | 4.226696  | 2.707610  |
| 41.N      | 4.067559  | 2.640727  | 1.797017  |
| 42.H      | 3.376125  | 2.127190  | 1.237558  |
| 43.H      | 3.885067  | 2.794920  | 2.780379  |
| 44.C.dzp  | -5.086556 | -3.188703 | 1.328343  |
| 45.N      | -4.735750 | -3.788898 | 0.201215  |
| 46.H      | -5.364836 | -4.454211 | -0.233357 |
| 47.H      | -3.741475 | -3.728792 | -0.102357 |
| 48.N      | -4.320716 | -2.202246 | 1.797017  |
| 49.H      | -3.530263 | -1.860216 | 1.237558  |
| 50.H      | -4.363005 | -1.967107 | 2.780379  |
| 51.N      | -6.190287 | -3.539028 | 1.981134  |
| 52.H      | -6.585006 | -2.952174 | 2.707610  |
| 53.C.dzp  | 5.304776  | -2.810735 | 1.328343  |
| 54.N      | 5.649157  | -2.206831 | 0.201215  |
| 55.H      | 6.539878  | -2.418979 | -0.233357 |

|          |           |           |           |
|----------|-----------|-----------|-----------|
| 56.H     | 5.099966  | -1.375816 | -0.102357 |
| 57.N     | 4.067559  | -2.640727 | 1.797017  |
| 58.H     | 3.376125  | -2.127190 | 1.237558  |
| 59.H     | 3.885067  | -2.794920 | 2.780379  |
| 60.N     | 6.160031  | -3.591432 | 1.981134  |
| 61.H     | 7.147846  | -3.607327 | 1.748637  |
| 62.C.dzp | -5.086556 | 3.188703  | 1.328343  |
| 63.N     | -4.735750 | 3.788898  | 0.201215  |
| 64.H     | -3.741475 | 3.728792  | -0.102357 |
| 65.H     | -5.364836 | 4.454211  | -0.233357 |
| 66.N     | -6.190287 | 3.539028  | 1.981134  |
| 67.H     | -6.697960 | 4.386552  | 1.748637  |
| 68.H     | -6.585006 | 2.952174  | 2.707610  |
| 69.N     | -4.320716 | 2.202246  | 1.797017  |
| 70.H     | -3.530263 | 1.860216  | 1.237558  |
| 71.H     | -4.363005 | 1.967107  | 2.780379  |
| 72.C.dzp | -4.143827 | 0.000000  | -2.686094 |
| 73.N     | -4.900696 | 0.000000  | -3.781630 |
| 74.H     | -5.233111 | 0.861655  | -4.199263 |
| 75.H     | -5.233111 | -0.861655 | -4.199263 |
| 76.N     | -3.763650 | -1.144856 | -2.135440 |
| 77.H     | -4.033334 | -2.023248 | -2.556926 |
| 78.H     | -3.156735 | -1.169806 | -1.301586 |
| 79.N     | -3.763650 | 1.144856  | -2.135440 |
| 80.H     | -4.033334 | 2.023248  | -2.556926 |
| 81.H     | -3.156735 | 1.169806  | -1.301586 |
| 82.C.dzp | 2.071914  | -3.588660 | -2.686094 |
| 83.N     | 2.450348  | -4.244127 | -3.781630 |
| 84.H     | 1.870340  | -4.962834 | -4.199263 |
| 85.H     | 3.362771  | -4.101179 | -4.199263 |
| 86.N     | 2.873300  | -2.686988 | -2.135440 |
| 87.H     | 3.768852  | -2.481346 | -2.556926 |
| 88.H     | 2.591450  | -2.148910 | -1.301586 |
| 89.N     | 0.890350  | -3.831845 | -2.135440 |
| 90.H     | 0.264483  | -4.504594 | -2.556926 |
| 91.H     | 0.565286  | -3.318716 | -1.301586 |
| 92.C.dzp | 2.071914  | 3.588660  | -2.686094 |
| 93.N     | 2.450348  | 4.244127  | -3.781630 |
| 94.H     | 3.362771  | 4.101179  | -4.199263 |
| 95.H     | 1.870340  | 4.962834  | -4.199263 |
| 96.N     | 0.890350  | 3.831845  | -2.135440 |
| 97.H     | 0.264483  | 4.504594  | -2.556926 |
| 98.H     | 0.565286  | 3.318716  | -1.301586 |
| 99.N     | 2.873300  | 2.686988  | -2.135440 |
| 100.H    | 3.768852  | 2.481346  | -2.556926 |
| 101.H    | 2.591450  | 2.148910  | -1.301586 |
| 102.H    | 5.849160  | -4.226696 | 2.707610  |
| 103.H    | 0.735846  | -7.178869 | 2.707610  |
| 104.H    | -6.697960 | -4.386552 | 1.748637  |
| 105.H    | 0.735846  | 7.178869  | 2.707610  |



|      |           |          |           |
|------|-----------|----------|-----------|
| 1.Cm | 0.000000  | 0.000000 | 0.209598  |
| 2.O  | 0.000000  | 0.000000 | 1.920497  |
| 3.O  | 0.000000  | 0.000000 | -1.501050 |
| 4.O  | -2.087987 | 1.415158 | 0.122268  |

|           |           |           |           |
|-----------|-----------|-----------|-----------|
| 5.O       | -0.181569 | 2.515829  | 0.122268  |
| 6.O       | -2.079616 | 3.602000  | -0.365971 |
| 7.C.tz2p  | -1.467852 | 2.542394  | -0.045765 |
| 8.O       | -0.181569 | -2.515829 | 0.122268  |
| 9.O       | -2.087987 | -1.415158 | 0.122268  |
| 10.O      | -2.079616 | -3.602000 | -0.365971 |
| 11.C.tz2p | -1.467852 | -2.542394 | -0.045765 |
| 12.O      | 2.269556  | 1.100671  | 0.122268  |
| 13.O      | 2.269556  | -1.100671 | 0.122268  |
| 14.O      | 4.159232  | 0.000000  | -0.365971 |
| 15.C.tz2p | 2.935703  | 0.000000  | -0.045765 |
| 16.C.dzp  | -0.227853 | 6.051865  | 1.280746  |
| 17.N      | -0.940683 | 6.021814  | 0.164899  |
| 18.H      | -1.219432 | 6.889653  | -0.278657 |
| 19.H      | -1.381922 | 5.119946  | -0.116247 |
| 20.N      | 0.256856  | 4.907364  | 1.765100  |
| 21.H      | 0.144895  | 4.038501  | 1.228861  |
| 22.H      | 0.502596  | 4.849304  | 2.745385  |
| 23.N      | 0.025467  | 7.197047  | 1.906526  |
| 24.H      | -0.460985 | 8.053747  | 1.662609  |
| 25.C.dzp  | -0.227853 | -6.051865 | 1.280746  |
| 26.N      | -0.940683 | -6.021814 | 0.164899  |
| 27.H      | -1.381922 | -5.119946 | -0.116247 |
| 28.H      | -1.219432 | -6.889653 | -0.278657 |
| 29.N      | 0.025467  | -7.197047 | 1.906526  |
| 30.H      | -0.460985 | -8.053747 | 1.662609  |
| 31.N      | 0.256856  | -4.907364 | 1.765100  |
| 32.H      | 0.144895  | -4.038501 | 1.228861  |
| 33.H      | 0.502596  | -4.849304 | 2.745385  |
| 34.C.dzp  | 5.354995  | 2.828606  | 1.280746  |
| 35.N      | 5.685385  | 2.196252  | 0.164899  |
| 36.H      | 5.124964  | 1.363193  | -0.116247 |
| 37.H      | 6.576331  | 2.388767  | -0.278657 |
| 38.N      | 6.220092  | 3.620578  | 1.906526  |
| 39.H      | 7.205241  | 3.627649  | 1.662609  |
| 40.H      | 5.920139  | 4.272336  | 2.623038  |
| 41.N      | 4.121474  | 2.676125  | 1.765100  |
| 42.H      | 3.424997  | 2.144733  | 1.228861  |
| 43.H      | 3.948323  | 2.859913  | 2.745385  |
| 44.C.dzp  | -5.127143 | -3.223259 | 1.280746  |
| 45.N      | -4.744702 | -3.825562 | 0.164899  |
| 46.H      | -5.356899 | -4.500886 | -0.278657 |
| 47.H      | -3.743042 | -3.756753 | -0.116247 |
| 48.N      | -4.378330 | -2.231238 | 1.765100  |
| 49.H      | -3.569892 | -1.893768 | 1.228861  |
| 50.H      | -4.450919 | -1.989392 | 2.745385  |
| 51.N      | -6.245559 | -3.576468 | 1.906526  |
| 52.H      | -6.660021 | -2.990823 | 2.623038  |
| 53.C.dzp  | 5.354995  | -2.828606 | 1.280746  |
| 54.N      | 5.685385  | -2.196252 | 0.164899  |
| 55.H      | 6.576331  | -2.388767 | -0.278657 |
| 56.H      | 5.124964  | -1.363193 | -0.116247 |
| 57.N      | 4.121474  | -2.676125 | 1.765100  |
| 58.H      | 3.424997  | -2.144733 | 1.228861  |
| 59.H      | 3.948323  | -2.859913 | 2.745385  |
| 60.N      | 6.220092  | -3.620578 | 1.906526  |

|          |           |           |           |
|----------|-----------|-----------|-----------|
| 61.H     | 7.205241  | -3.627649 | 1.662609  |
| 62.C.dzp | -5.127143 | 3.223259  | 1.280746  |
| 63.N     | -4.744702 | 3.825562  | 0.164899  |
| 64.H     | -3.743042 | 3.756753  | -0.116247 |
| 65.H     | -5.356899 | 4.500886  | -0.278657 |
| 66.N     | -6.245559 | 3.576468  | 1.906526  |
| 67.H     | -6.744257 | 4.426098  | 1.662609  |
| 68.H     | -6.660021 | 2.990823  | 2.623038  |
| 69.N     | -4.378330 | 2.231238  | 1.765100  |
| 70.H     | -3.569892 | 1.893768  | 1.228861  |
| 71.H     | -4.450919 | 1.989392  | 2.745385  |
| 72.C.dzp | -4.204684 | 0.000000  | -2.605719 |
| 73.N     | -4.982126 | 0.000000  | -3.687354 |
| 74.H     | -5.321403 | 0.861624  | -4.099411 |
| 75.H     | -5.321403 | -0.861624 | -4.099411 |
| 76.N     | -3.813916 | -1.145541 | -2.064006 |
| 77.H     | -4.096833 | -2.022665 | -2.479603 |
| 78.H     | -3.190663 | -1.176677 | -1.240211 |
| 79.N     | -3.813916 | 1.145541  | -2.064006 |
| 80.H     | -4.096833 | 2.022665  | -2.479603 |
| 81.H     | -3.190663 | 1.176677  | -1.240211 |
| 82.C.dzp | 2.102342  | -3.641363 | -2.605719 |
| 83.N     | 2.491063  | -4.314648 | -3.687354 |
| 84.H     | 1.914513  | -5.039283 | -4.099411 |
| 85.H     | 3.406890  | -4.177658 | -4.099411 |
| 86.N     | 2.899025  | -2.730177 | -2.064006 |
| 87.H     | 3.800095  | -2.536629 | -2.479603 |
| 88.H     | 2.614363  | -2.174856 | -1.240211 |
| 89.N     | 0.914890  | -3.875718 | -2.064006 |
| 90.H     | 0.296737  | -4.559293 | -2.479603 |
| 91.H     | 0.576299  | -3.351533 | -1.240211 |
| 92.C.dzp | 2.102342  | 3.641363  | -2.605719 |
| 93.N     | 2.491063  | 4.314648  | -3.687354 |
| 94.H     | 3.406890  | 4.177658  | -4.099411 |
| 95.H     | 1.914513  | 5.039283  | -4.099411 |
| 96.N     | 0.914890  | 3.875718  | -2.064006 |
| 97.H     | 0.296737  | 4.559293  | -2.479603 |
| 98.H     | 0.576299  | 3.351533  | -1.240211 |
| 99.N     | 2.899025  | 2.730177  | -2.064006 |
| 100.H    | 3.800095  | 2.536629  | -2.479603 |
| 101.H    | 2.614363  | 2.174856  | -1.240211 |
| 102.H    | 5.920139  | -4.272336 | 2.623038  |
| 103.H    | 0.739882  | -7.263159 | 2.623038  |
| 104.H    | -6.744257 | -4.426098 | 1.662609  |
| 105.H    | 0.739882  | 7.263159  | 2.623038  |

**Table S15.** Standard oxidation energies of elements (in eV).<sup>a</sup>

| Group    | Element   | $M^{0-}$<br>$M^+$ | $M^{0-}$<br>$M^{2+}$ | $M^{+-}$<br>$M^{2+}$     | $M^{2+-}$<br>$M^{3+}$  | $M^{3+-}$<br>$M^{4+}$ | $M^{4+-}$<br>$M^{5+}$ | $M^{5+-}$<br>$M^{6+}$          | $M^{6+-}$<br>$M^{7+}$        | $M^{7+-}$<br>$M^{8+}$ |
|----------|-----------|-------------------|----------------------|--------------------------|------------------------|-----------------------|-----------------------|--------------------------------|------------------------------|-----------------------|
| <b>1</b> | <b>Li</b> | 3.04              |                      |                          |                        |                       |                       |                                |                              |                       |
|          | Na        | 2.71              |                      |                          |                        |                       |                       |                                |                              |                       |
|          | K         | 2.93              |                      |                          |                        |                       |                       |                                |                              |                       |
|          | Rb        | 2.98              |                      |                          |                        |                       |                       |                                |                              |                       |
|          | Cs        | 3.03              |                      |                          |                        |                       |                       |                                |                              |                       |
|          | Fr        | 2.90              |                      |                          |                        |                       |                       |                                |                              |                       |
| <b>2</b> | <b>Be</b> | 2+ $\delta$       | 3.69                 | 1.69- $\delta$           |                        |                       |                       |                                |                              |                       |
|          | Mg        | 2.70              | 4.74                 | 2.04                     |                        |                       |                       |                                |                              |                       |
|          | Ca        | 3.80              | 5.74                 | 1.94                     |                        |                       |                       |                                |                              |                       |
|          | Sr        | 4.10              | 5.80                 | 1.70                     |                        |                       |                       |                                |                              |                       |
|          | Ba        | 4+ $\delta$       | 5.82                 | 1.82- $\delta$           |                        |                       |                       |                                |                              |                       |
|          | Ra        | 4+ $\delta$       | 5.6                  | 1.6- $\delta$            |                        |                       |                       |                                |                              |                       |
| <b>3</b> | <b>Sc</b> | 3+ $\delta$       |                      | 2- $\delta$ + $\epsilon$ | 1.23- $\epsilon$       |                       |                       |                                |                              |                       |
|          | Y         | 3+ $\delta$       |                      | 1.3- $\delta$            | 2.8                    |                       |                       |                                |                              |                       |
|          | La        | 3+ $\delta$       | 4.04                 | 1.04- $\delta$           | 3.1                    |                       |                       |                                |                              |                       |
|          | Ac        | 2+ $\delta$       | 1.7                  | -0.3- $\delta$           | 4.9                    |                       |                       |                                |                              |                       |
| <b>4</b> | <b>Ti</b> | 2+ $\delta$       | 3.25                 | 1.25- $\delta$           | 0.85                   | [-0.2]                |                       |                                |                              |                       |
|          | Zr        | 2+ $\delta$       |                      | 2- $\delta$ + $\epsilon$ | 1- $\epsilon$ + $\eta$ | 0.80- $\eta$          |                       |                                |                              |                       |
|          | Hf        | 2+ $\delta$       |                      | 2- $\delta$ + $\epsilon$ | 1- $\epsilon$ + $\eta$ | [1.2- $\eta$ ]        |                       |                                |                              |                       |
|          | Ce        | 2+ $\delta$       | 3.8                  | 1.8- $\delta$            | 3.2                    | -1.7                  |                       |                                |                              |                       |
|          | Th        | 2+ $\delta$       | -0.9                 | -2.9- $\delta$           | 4.8                    | 3.6                   |                       |                                |                              |                       |
| <b>5</b> | <b>V</b>  | 2+ $\delta$       | 2.26                 | 0.26- $\delta$           | 0.26                   | -0.34                 | -0.99                 |                                |                              |                       |
|          | Nb        | 2+ $\delta$       | 1.5                  | -0.5- $\delta$           | [1.8]                  | [-0.5]                | [0.5]                 |                                |                              |                       |
|          | Ta        | 2+ $\delta$       |                      | 0- $\delta$ + $\epsilon$ | -0.2- $\epsilon$       | 1+ $\eta$             | [0.95- $\eta$ ]       |                                |                              |                       |
|          | Pr        | 2+ $\delta$       | 4.0                  | 2- $\delta$              | 2.8                    | -3.0                  |                       |                                |                              |                       |
|          | Pa        | 2+ $\delta$       |                      | 0- $\delta$ + $\epsilon$ | 2.0                    | 1.94                  |                       |                                |                              |                       |
| <b>6</b> | <b>Cr</b> | 1+ $\delta$       | 1.82                 | 0.82- $\delta$           | 0.42                   | $\epsilon$            | -1.34                 | [-2.35- $\epsilon$ ]           |                              |                       |
|          | Mo        | 0+ $\delta$       |                      | 0- $\delta$ + $\epsilon$ | 0.6- $\epsilon$        | [0]                   | [ $\eta$ ]            | [-0.2- $\eta$ ]                |                              |                       |
|          | W         | 0+ $\delta$       |                      | 0- $\delta$ + $\epsilon$ | -0.3- $\epsilon$       | [0.8]                 | [0.0]                 | [0.0]                          |                              |                       |
|          | Nd        | 3+ $\delta$       | 4.25                 | 1.25- $\delta$           | 2.7                    | -4.5                  |                       |                                |                              |                       |
|          | U         | 1+ $\delta$       | 0.7                  | -0.3- $\delta$           | 4.7                    | 0.5                   | [-0.15]               | [-0.4]                         |                              |                       |
| <b>7</b> | <b>Mn</b> | 2+ $\delta$       | 2.37                 | 0.37- $\delta$           | -1.54                  | [-0.91]               | [-2+ $\eta$ ]         | [-2.6- $\eta$ ]                | [-0.56]                      |                       |
|          | Tc        | -1+ $\delta$      |                      | 0- $\delta$ + $\epsilon$ | -0.1- $\epsilon$       | [0.15]                | [-1+ $\eta$ ]         | [-1- $\eta$ + $\beta$ ]        | [-0.35- $\beta$ ]            |                       |
|          | Re        | -1+ $\delta$      |                      | 0- $\delta$ + $\epsilon$ | 0.1- $\epsilon$        | [-0.1]                | [ $\eta$ ]            | [-0.8- $\eta$ ]                | [-0.77]                      |                       |
|          | Pm        | 2+ $\delta$       | 4.4                  | 2.4- $\delta$            | 2.6                    | -4.7                  |                       |                                |                              |                       |
|          | Np        | 1+ $\delta$       | 0.9                  | -0.1- $\delta$           | 4.7                    | -0.15                 |                       |                                |                              |                       |
| <b>8</b> | <b>Fe</b> | 1+ $\delta$       | 0.9                  | -0.1- $\delta$           | -0.78                  |                       |                       |                                |                              |                       |
|          | Ru        | 0+ $\delta$       | -0.9                 | -0.9- $\delta$           | -0.25                  | [-2.0]                | [-2+ $\eta$ ]         | [-1.55- $\eta$ ]               | [-0.6]                       | [-1.0]                |
|          | Os        | 0+ $\delta$       |                      | 0- $\delta$ + $\epsilon$ | 0- $\epsilon$ + $\eta$ | [0.73- $\eta$ ]       | [-2+ $\delta'$ ]      | [-1- $\delta'$ + $\epsilon'$ ] | [-1- $\epsilon'$ + $\eta'$ ] | [-0.1- $\eta'$ ]      |
|          | Sm        | 3+ $\delta$       | 5.36                 | 2.36- $\delta$           | 1.55                   | -5.1                  |                       |                                |                              |                       |
|          | Pu        | 1+ $\delta$       | 2.6                  | 1.6- $\delta$            | 3.5                    | -1.0                  | -1.1                  |                                |                              |                       |

| Group     | Element   | $M^{0 \rightarrow M^+}$ | $M^{0 \rightarrow M^{2+}}$ | $M^{1 \rightarrow M^{2+}}$ | $M^{2+ \rightarrow M^{3+}}$ | $M^{3+ \rightarrow M^{4+}}$ | $M^{4+ \rightarrow M^{5+}}$ | $M^{5+ \rightarrow M^{6+}}$ | $M^{6+ \rightarrow M^{7+}}$ | $M^{7+ \rightarrow M^{8+}}$ |
|-----------|-----------|-------------------------|----------------------------|----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|
| <b>9</b>  | <b>Co</b> | $\delta$                | <i>0.56</i>                | $0.56-\delta$              | -1.92                       |                             |                             |                             |                             |                             |
|           | Rh        | -0.6                    |                            | $-1+\delta$                | $-1.67-\delta$              |                             |                             |                             |                             |                             |
|           | Ir        | $-1+\delta$             |                            | $-1-\delta+\epsilon$       | $-1.47+\delta-\epsilon$     |                             |                             |                             |                             |                             |
|           | Eu        | $3+\delta$              | <i>5.6</i>                 | $2.6-\delta$               | 0.36                        | -6.3                        |                             |                             |                             |                             |
|           | Am        | $2+\delta$              | <i>3.8</i>                 | $1.8-\delta$               | 2.3                         | -2.6                        |                             |                             |                             |                             |
| <b>10</b> | <b>Ni</b> | $\delta$                | <i>0.5</i>                 | $0.5-\delta$               | $[-2+\epsilon]$             | $[-1.36-\epsilon]$          |                             |                             |                             |                             |
|           | Pd        | $-1+\delta$             | <i>-1.9</i>                | $-0.9-\delta$              |                             |                             |                             |                             |                             |                             |
|           | Pt        | $-1+\delta$             | <i>-2.36</i>               | $-1.36-\delta$             |                             |                             |                             |                             |                             |                             |
|           | Gd        | $2+\delta$              | <i>2.9</i>                 | $0.9-\delta$               | 3.9                         | -7.9                        |                             |                             |                             |                             |
|           | Cm        | $1+\delta$              | <i>1.7</i>                 | $0.7-\delta$               | 4.4                         | -3.5                        |                             |                             |                             |                             |
| <b>11</b> | <b>Cu</b> | -0.50                   | <i>-0.65</i>               | -0.15                      | -2.40                       |                             |                             |                             |                             |                             |
|           | Ag        | -0.80                   | <i>-1.98</i>               | -1.18                      | -1.80                       |                             |                             |                             |                             |                             |
|           | Au        | -1.69                   | <i>-3.49</i>               | -1.80                      | -1.00                       |                             |                             |                             |                             |                             |
|           | Tb        | $2+\delta$              | <i>3.15</i>                | $1.15-\delta$              | 3.7                         | -3.1                        |                             |                             |                             |                             |
|           | Bk        | $-1+\delta$             | <i>-0.8</i>                | $0.2-\delta$               | 2.8                         | -1.67                       |                             |                             |                             |                             |
| <b>12</b> | <b>Zn</b> | $1+\delta$              | <i>1.52</i>                | $0.52-\delta$              |                             |                             |                             |                             |                             |                             |
|           | Cd        | $1+\delta$              | <i>0.8</i>                 | $-0.2-\delta$              |                             |                             |                             |                             |                             |                             |
|           | Hg        | -0.8                    | <i>-1.7</i>                | -0.9                       |                             |                             |                             |                             |                             |                             |
|           | Dy        | $2+\delta$              | <i>4.35</i>                | $2.35-\delta$              | 2.6                         | -5.0                        |                             |                             |                             |                             |
|           | Cf        | $2+\delta$              | <i>4.24</i>                | $2.24-\delta$              | 1.6                         | -3.3                        |                             |                             |                             |                             |

<sup>a</sup> Values after Lide<sup>1</sup> and Nugent et al.<sup>2</sup> Positive values mean strong tendency toward higher oxidation. Values in brackets [] refer to oxo-ions such as titanyl  $[\text{Ti}^{4+}\text{O}^{2-}]^{2+}$ . If only double or triple oxidation energies are available, unknown increments such as  $+\delta$  and  $-\delta$  are introduced for the individual oxidation steps, the  $\delta$ ,  $\epsilon$  etc. being specific for each element row).

<sup>1</sup> D. R. Lide, Ed., CRC Handbook of Chemistry and Physics, CRC Press, Boca Raton FL, 2005. Internet Version <http://www.hbcpnetbase.com>.

<sup>2</sup> L. J. Nugent, R. D. Baybarz and J. L. Burnett, J. Inorg. Nucl. Chem., 1971, 33, 2503-2530. L. J. Nugent, R. D. Baybarz, J. L. Burnett and J. L. Ryan, J. Phys. Chem., 1973, 77, 1528-1539.