Electronic supplementary information

Structural diversity and magnetic properties of six cobalt coordination polymers based on 2,2'-phosphinico-dibenzoate ligand

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Table S1. Selected Bond Distances ((Å) and Angles ((deg) for $1-6$

			1				
Co(1)-O(5A)	2.037(3)	Co(1)-O(42)	2.053(3)	Co(1)-O(38)	2.235(3)		
$C_0(1)-O(41)$	2.069(3)	$C_0(1) - O(17)$	2.096(3)	$C_0(1)$ - $O(40)$	2.119(3)		
$C_0(2) - O(9)$	2.020(3)	$C_0(2) - O(6A)$	2.027(3)	$C_0(2) - O(8)$	2 157(3)		
$C_0(2) - O(17)$	2.096(3)	$C_0(2) = O(16)$	2.027(3) 2.110(3)	$C_0(2) = O(38)$	2 188(3)		
$C_0(3)$ - $O(3)$	1.969(3)	$C_0(3) - O(12)$	2.018(3)	$C_0(3)$ - $O(1)$	2.082(3)		
$C_0(3) - O(31)$	2.034(3)	$C_0(3) - O(10)$	2.010(3) 2.048(3)	00(0) 0(1)	2.002(5)		
$C_0(4) - O(37)$	2.051(3)	$C_0(4) - O(2)$	2.095(3)	$C_{0}(4) - O(15)$	1 991(3)		
$C_0(4) - O(37)$	2.003(3) 2.122(3)	$C_0(4) - O(2)$	2.000(3)	$C_0(4) - O(13)$	2.251(3)		
$C_0(4) - O(30)$	2.122(3) 2.075(3)	$C_0(5) O(10)$	2.105(3) 2.116(3)	$C_0(5) O(27)$	2.251(5)		
$C_0(5) - O(16)$	2.073(3) 2.122(3)	$C_0(5) - O(10)$	2.110(3) 2.142(3)	$C_0(5) O(37)$	2.000(3)		
$C_0(5) - O(10)$	2.123(3) 1.079(3)	$C_0(5) - O(8)$	2.143(3) 2.002(3)	$C_0(5) - O(32)$	1.021(3)		
$C_0(6) - O(39)$	1.978(3)	$C_0(0) - O(7)$	2.002(3)	00(0)-0(33)	1.951(5)		
$C_0(0) - O(19)$	2.027(3) 2.020(3)	$C_0(0) - O(32)$	2.555(5) 2.046(2)	$C_{2}(7) O(14)$	2,217(2)		
$C_0(7) - O(39)$	2.020(3)	$C_0(7) - O(37)$	2.040(3)	$C_0(7) - O(14)$	2.51/(5) 2.184(2)		
$C_0(7) - O(29)$	2.047(3)	$C_0(7) - O(27)$	2.110(3)	$C_0(7) - O(34)$	2.184(3)		
$C_0(8) - O(22)$	1.998(3)	Co(8) - O(26)	2.008(3)	Co(8)-O(20)	2.062(3)		
Co(8) - O(13)	2.04/(3)	Co(8) - O(27)	2.052(3)		2 (22)		
Co(9)-O(24B)	2.038(3)	$C_0(9) - O(36)$	2.062(3)	Co(9)-O(28)	2.023(3)		
Co(9)-O(29)	2.083(3)	Co(9)-O(34)	2.136(3)	Co(9)-O(44)	2.195(3)		
Co(10)-O(45)	2.077(3)	Co(10)-O(46)	2.088(3)	Co(10)-O(23B)	2.052(3)		
Co(10)-O(47)	2.094(3)	Co(10)-O(36)	2.128(3)	Co(10)-O(44)	2.181(3)		
Co(1)-O(17)-Co(2)	99.60(11)	Co(2)-O(38)-Co(1)	92.77(10)	Co(2)-O(16)-Co(5)	100.84(11)		
Co(3)-O(10)-Co(5)	113.62(12)	Co(4)-O(14)-Co(7)	89.65(10)	Co(5)-O(37)-Co(4)	123.90(13)		
Co(5)-O(8)-Co(2)	98.67(11)	Co(5)-O(32)-Co(6)	87.18(10)	Co(6)-O(39)-Co(5)	100.89(11)		
Co(6)-O(39)-Co(7)	120.89(13)	Co(7)-O(37)-Co(5)	96.12(11)	Co(7)-O(37)-Co(4)	103.21(11)		
Co(7)-O(29)-Co(9)	104.39(11)	Co(7)-O(39)-Co(5)	96.49(11)	Co(8)-O(27)-Co(7)	115.18(12)		
Co(9)-O(34)-Co(7)	98.11(10)	Co(9)-O(36)-Co(10)	98.66(11)	Co(10)-O(44)-Co(9)	93.18(10)		
		2					
$C_0(1)-O(3A)$	2.006(2)	$C_{0}(1)-O(3)$	2.006(2)	$C_{0}(1)-O(2)$	2,119(3)		
$C_0(1)$ - $O(2A)$	2.119(3)	$C_0(1)$ -N(6B)	2.169(3)	$C_0(1)$ -N(6C)	2.169(3)		
$C_0(2)$ - $O(4)$	1.964(3)	$C_0(2) - O(1D)$	2.021(3)	$C_0(2)-O(5E)$	2.109(3)		
$C_0(2) - N(1)$	2 129(3)	$C_0(2) - O(6E)$	2.258(3)		2.10)(3)		
	2.12)(0)	3	2.200(0)				
$C_{\alpha}(1) \mathbf{N}(1)$	2.197(2)	G ₂ (1) O (2)	2 125(2)	$C_{2}(1) O(2\Lambda)$	2 2008(10)		
$C_0(1) - N(1)$	2.18/(2)	$C_0(1) - O(2)$	2.133(2)	$C_0(1) - O(2A)$	2.2908(19)		
$C_0(1) = O(3)$	2.018/(19)	$C_0(1) - O(7)$	2.051(2)	$C_0(1) - O(8)$	2.117(3)		
$C_0(2) - O(4)$	2.054(2)	$C_0(2) - O(4B)$	2.054(2)	$C_0(2) - O(5)$	2.0589(19)		
$C_0(2) - O(5B)$	2.059(2)	Co(2)-O(9B)	2.155(2)	Co(2)-O(9)	2.153(2)		
Co(1)-O(2)-Co(1A)	103.68(8)						
		4					
Co(1)-N(1)	2.2281(18)	Co(1)-O(2)	2.1222(16)	Co(1)-O(2C)	2.2625(15)		
Co(1)-O(3)	2.0079(15)	Co(1)-O(7)	2.0621(15)	Co(1)-O(8)	2.1337(16)		
Co(2)-N(2)	2.160(2)	Co(2)-N(2B)	2.160(2)	Co(2)-O(4)	2.0767(15)		
Co(2)-O(4B)	2.0767(15)	Co(2)-O(5)	2.0404(15)	Co(2)-O(5B)	2.0404(15)		
Co(1)-O(2)-Co(1C)	104.90(6)						
		5					
Co(1)-N(2)	2.167(3)	Co(1)-N(3B)	2.137(3)	Co(1)-O(4)	1.9956(15)		
$C_0(1)-O(4A)$	1.9956(15)	$C_0(1)-O(5)$	2,1580(15)	$C_0(1)-O(5A)$	2.1580(15)		
$C_0(2)-N(1)$	2,1014(19)	$C_0(2) - O(1C)$	2.2242(16)	$C_0(2)$ - $O(2C)$	2.0938(16)		
$C_0(2) - O(3)$	1 9783(15)	$C_0(2) - O(6D)$	2.0282(16)	00(1) 0(10)			
	1.5 / 05(10)	6	2.0202(10)				
$C_{2}(1) O(6)$	2 0755(17)	$C_{\alpha}(1) O(fA)$	2.0755(17)	$C_{2}(1) O(0)$	2 1042(10)		
$C_0(1) - O(0)$	2.0733(17) 2.1042(10)	$C_0(1) - O(0A)$	2.0733(17)	$C_0(1) - O(9)$	2.1045(19)		
$C_0(1) = O(3A)$	2.1045(19)	$C_0(1) = O(10A)$	2.0023(17) 1.0220(19)	$C_0(2) - N(1)$	2.023(2)		
$C_{0}(2) - O(2B)$	1.9030(18)	$C_{0}(2) - O(4)$	1.9220(18)	$C_0(2) - O(10)$	2.01/3(18) 1.0491(10)		
$C_{0}(3) - O(3B)$	1.9384(17)	C0(5)-O(5A)	2.0304(18)	$CO(3) - O(\delta C)$	1.9401(19)		
$C_0(3) - O(10)$	2.002/(1/)	$O_{2}(2) O(10) O_{2}(1)$	117.00(0)	$G_{1}(2) = O(10) = G_{1}(1)$	10((2(0))		
Co(3)-O(10)-Co(2)	104./9(8)	Co(3)-O(10)-Co(1)	117.00(8)	Co(2)-O(10)-Co(1)	100.02(8)		
oyinneuy cours. A. A-1, y, Z, B. X ⁺ 1, y, Z (1), AX ⁺ 1, -y, -Z ⁺ 1, B. X ⁺ 1/2, -Y ⁺ 1/2, Z ⁻ 1/2, U: -X ⁺ 1/2, Y ⁻ 1/2, -Z ⁺ 1/2, U: -X ⁺ 1, -Y ⁺ 1, Z ⁺ 1; E: X, Y ⁺ 1, Z (2); AX ⁺ 1, -Y, -Z ⁺ 2, U: -X ⁺ 1/2, -Y ⁺ 1/2, -Z ⁺ 1/2, U: -X ⁺ 1, -Y ⁺ 1, -Z ⁺ 1; E: X, Y ⁺ 1, Z (2); AX ⁺ 1, -Y, -Z ⁺ 2, U: -X ⁺ 1/2, -X ⁺							
$D_{1} = -x_{1} + y_{1} + z_{2} + (z_{2} + x_{3} + y_{3} + z_{1} + (z_{3} + y_{3} + z_{1} + z_{3} + z_{2} + z_{3} + z_{1} + (z_{3} + z_{3} + y_{3} + z_{1} + z_{2} + z_{3} + $							
y+1, -2+1, b. x+1, y, z, U. x, -y+1/2, Z-1/2 (0).							







Fig. S3. 3D supramolecular structure of 1.



Fig. S4. 3D supramolecular structure of 2.



Fig. S5. 3D supramolecular structure of 3.



Fig. S6. Temperature dependence of the imaginary parts of 1 measured under various oscillating frequencies with zero field.



T/KFig. S7. Temperature dependence of the in-phase of 1 measured under various oscillating frequencies with zero field.



Fig. S8. Hysteresis loop at 1.8 K and 5 K for 1.