

## Electronic Supplementary Information

# Half-Sandwich Complexes of Iridium and Ruthenium Containing Cysteine-Derived Ligands

María Carmona,<sup>a</sup> Ricardo Rodríguez,<sup>a,\*</sup> Fernando J. Lahoz,<sup>a</sup> Pilar García-Orduña,<sup>a</sup> Carlos Cativiela,<sup>b</sup> José A. López<sup>a</sup> and Daniel Carmona<sup>a,\*</sup>

<sup>a</sup> Instituto de Síntesis Química y Catálisis Homogénea (ISQCH), CSIC - Universidad de Zaragoza, Departamento de Química Inorgánica, Pedro Cerbuna 12, 50009 Zaragoza, Spain

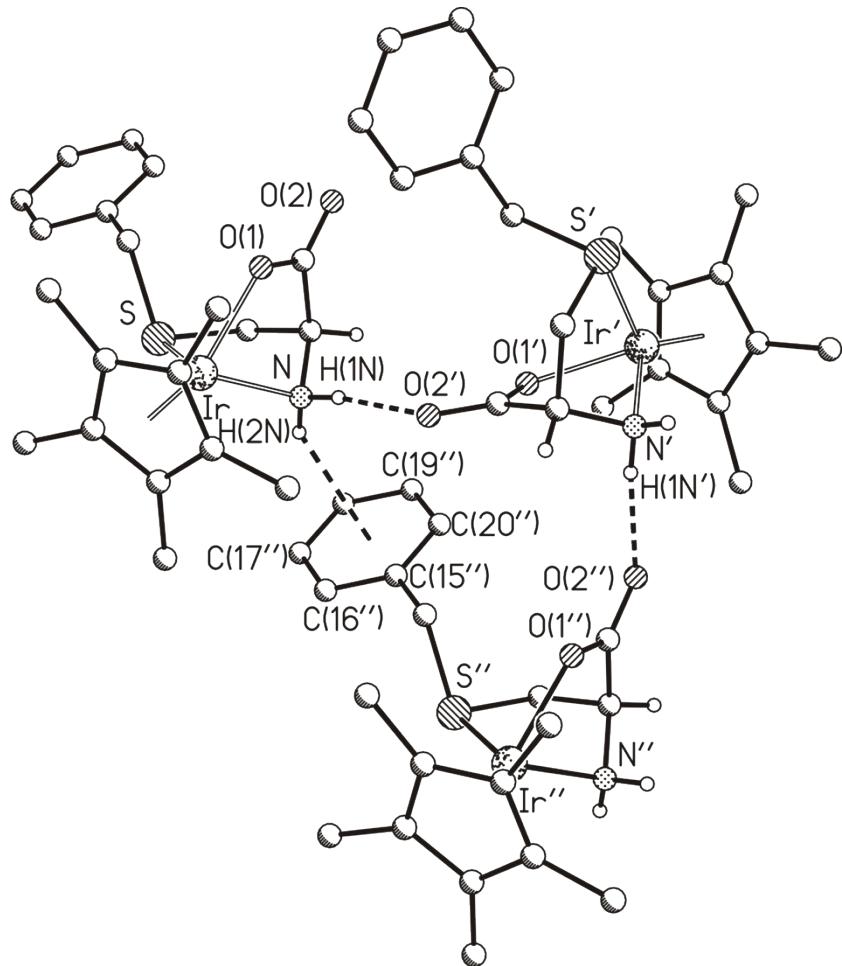
<sup>b</sup> Instituto de Síntesis Química y Catálisis Homogénea (ISQCH), CSIC - Universidad de Zaragoza, Departamento de Química Orgánica, Pedro Cerbuna 12, 50009 Zaragoza, Spain

E-mails: dcarmona@unizar.es (D. C.), riromar@unizar.es (R. R.).

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## 1- Hydrogen bond interactions in **6aSb**



**Figure S1.** Intermolecular interactions in **6aSb**. Symmetry codes:  
)  $x - 1/2, -y + 1/2, -z + 1$ ; ")  $x - 1, y, z$ .

**Hydrogen bonds ( $\text{\AA}$ ,  $^\circ$ ):** Symmetry code: ')  $x - 1/2, -y + 1/2, -z + 1$ .

	D-H	D $\cdots$ A	H $\cdots$ A	D-H $\cdots$ A
N-H(1N) $\cdots$ O(2')	0.90(4)	2.846(4)	1.96(4)	170(4)

**N-H $\cdots$  $\pi$  interactions ( $\text{\AA}$ ,  $^\circ$ ):** Symmetry code: ")  $x - 1, y, z$ .

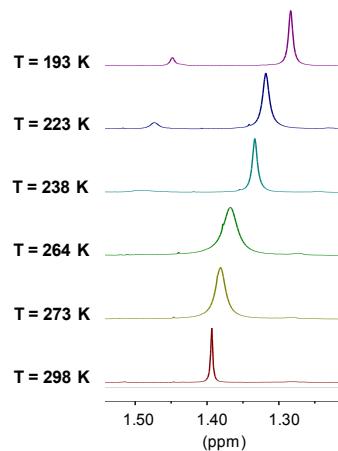
	H $\cdots$ G	H $\cdots$ Ph (plane)	$\gamma$	N-H $\cdots$ C	N-H $\cdots$ C (Ph)
N-H(2N) $\cdots$ C(15'')/C(20'')	2.58(3)	2.54(3)	10.2(6)	C(16''): 2.93(3) C(17''): 2.73(3) C(18''): 2.74(3) C(19''): 2.94(3)	C(15''): 3.10(3) C(20''): 3.11(3)

H $\cdots$ G represents the distance from the H atom to the centroid of the phenyl ring of the reported NH $\cdots$  $\pi$  interaction. H $\cdots$ Ph is the separation from the H atom to the mean plane of the ring.  $\gamma$  angle: angle between the H-G vector and the normal of the phenyl ring. N-H $\cdots$ C: contact distances between H atom and C atoms under the assumed criterion (3.05  $\text{\AA}$ ). N-H $\cdots$ C(Ph): range of separation between H and the rest of the carbon atoms of the ring.

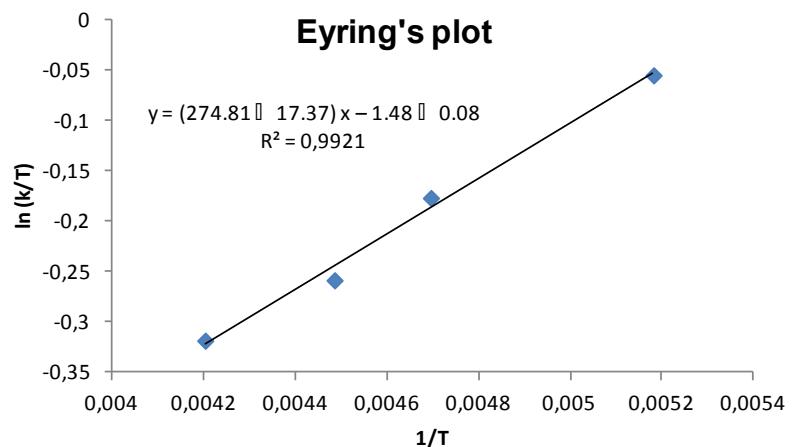
## 2- Kinetic studies for the epimerization of complex 9a

[9a] = 78.61 mM; solvent = CD<sub>2</sub>Cl<sub>2</sub>

T (K)	k (s <sup>-1</sup> )
193	182,7504
213	178,5546
223	172,2720
238	173,1600



ln (k/T)	1/T
0,005181347	-0,054568901
0,004694836	-0,176397729
0,004484305	-0,258097148
0,004201681	-0,318054651



$$\Delta G^\ddagger = 12.43 \pm 2.28 \text{ kcal/mol}$$

$$\Delta H^\ddagger = -0.55 \pm 0.03 \text{ kcal/mol}$$

$$\Delta S^\ddagger = -44.3 \pm 6.0 \text{ cal/K·mol}$$