

## Electronic supplementary information

### Hydrothermally grown CdS nanograin-sensitized 1D Zr:Fe<sub>2</sub>O<sub>3</sub>/FTO photoanode for efficient solar-light-driven photoelectrochemical performance

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**Table S1.** Recent reports on photoelectrochemical performance of CdS/Fe<sub>2</sub>O<sub>3</sub> photoanodes.

Year	Photoelectrode	Method	Electrolyte	Performance	Ref
2016	<b>CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> Nanorod arrays</b>	<b>Hydrothermal + Immersion</b>	<b>0.50 M Na<sub>2</sub> S and 0.50 M Na<sub>2</sub>SO<sub>3</sub></b>	<b>3.3 mA/cm<sup>2</sup> 0.2 V versus Ag/AgCl</b>	<b>[Our work]</b>
2016	CdS /Ti-Fe <sub>2</sub> O <sub>3</sub> 2D nanosheets	Electrodeposition + Dipping	0.50 M Na <sub>2</sub> S and 0.50 M Na <sub>2</sub> SO <sub>3</sub>	2.7 mA/cm <sup>2</sup> 0.2 V versus Ag/AgCl	[1]
2016	Fe <sub>2</sub> O <sub>3</sub> /CdS co- sensitized TiO <sub>2</sub> nanotube arrays (TNA)	ultrasonic- assisted chemical bath deposition (CBD) method	0.35 M Na <sub>2</sub> SO <sub>3</sub> + 0.25 M Na <sub>2</sub> S.	0.904 mA/cm <sup>2</sup> (0.0 V <sub>Ag/AgCl</sub> )	[2]
2015	mesoporous Fe <sub>2</sub> O <sub>3</sub> -CdS heterostructures	interfacial thermal decomposition +chemical bath deposition method	phosphate buffer saline (pH 7.4)	0.9 mA/cm <sup>2</sup> 0.0 V (vs Ag/AgCl)	[3]
2013	3D hierarchical CdS/ $\alpha$ -Fe <sub>2</sub> O <sub>3</sub> heterojunction nanocomposites	facile chemical bath method	0.1 M Na <sub>2</sub> SO <sub>4</sub>	1.4 $\mu$ A/cm <sup>2</sup> Vs saturated calomel electrode (SCE)	[13]

**Table S2:** Variation of crystallite size and micro strain according to the annealing at different temperatures.

CdS/Zr:Fe <sub>2</sub> O <sub>3</sub>	D (nm)	$\epsilon_{\mu}$ ( $\times 10^{-4}$ )
As-grown	74.7	8.7
250 °C	69.7	9.3
350 °C	109.1	16.2
400 °C	77.4	12.0

**Table S3:** Comparison of photoelectrochemical performance of annealed CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> heterostructured array at different experimental conditions.

Electrode	Applied potential	Electrolyte	Performance
CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub> annealed at 350 °C	(-0.3 V <sub>Ag/AgCl</sub> )	0.1 M Na <sub>2</sub> S + 0.02 M Na <sub>2</sub> SO <sub>3</sub> .	2.2 mA/cm <sup>2</sup>
	(0.2 V <sub>Ag/AgCl</sub> )	0.1 M Na <sub>2</sub> S and 0.02 M Na <sub>2</sub> SO <sub>3</sub>	3.1 mA/cm <sup>2</sup>
	(-0.3 V <sub>Ag/AgCl</sub> )	0.5 M Na <sub>2</sub> S and 0.5 M Na <sub>2</sub> SO <sub>3</sub>	3.0 mA/cm <sup>2</sup>
	<b>(0.2 V<sub>Ag/AgCl</sub>)</b>	<b>0.5 M Na<sub>2</sub>S and 0.5 M Na<sub>2</sub>SO<sub>3</sub></b>	<b>3.3 mA/cm<sup>2</sup></b>

**Table S4:** EIS fitting parameters of the bare 1D Zr:α-Fe<sub>2</sub>O<sub>3</sub>, CdS, and CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> heterostructured array deposited at various time intervals.

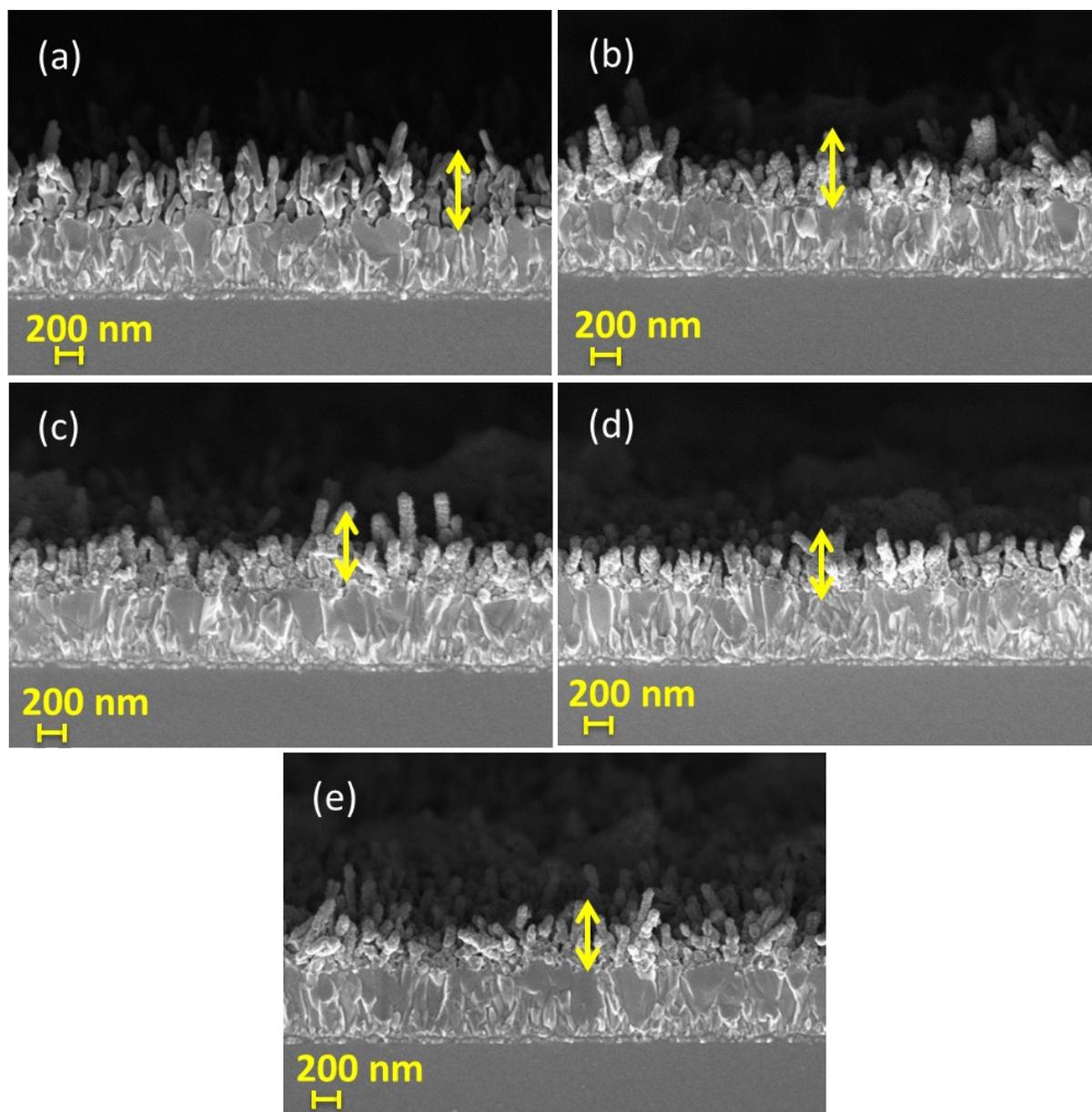
Samples/ Parameters	$R_s$ Ω	$R_2$ Ω	$R_3$ Ω	$C_{[CPE1]}$ μF	$C_{[CPE2]}$ μF
1D Zr:Fe <sub>2</sub> O <sub>3</sub>	51	123	929	0.33	196
CdS	47	722	3681	3.2	5.3
10_CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	52	93	870	0.41	157
20_CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	42	87	618	0.6	229
30_CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	57	95	641	0.46	135
40_CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	89	144	643	0.31	150

**Table S5:** EIS fitting parameters of the bare CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> heterostructured array annealed at various temperatures.

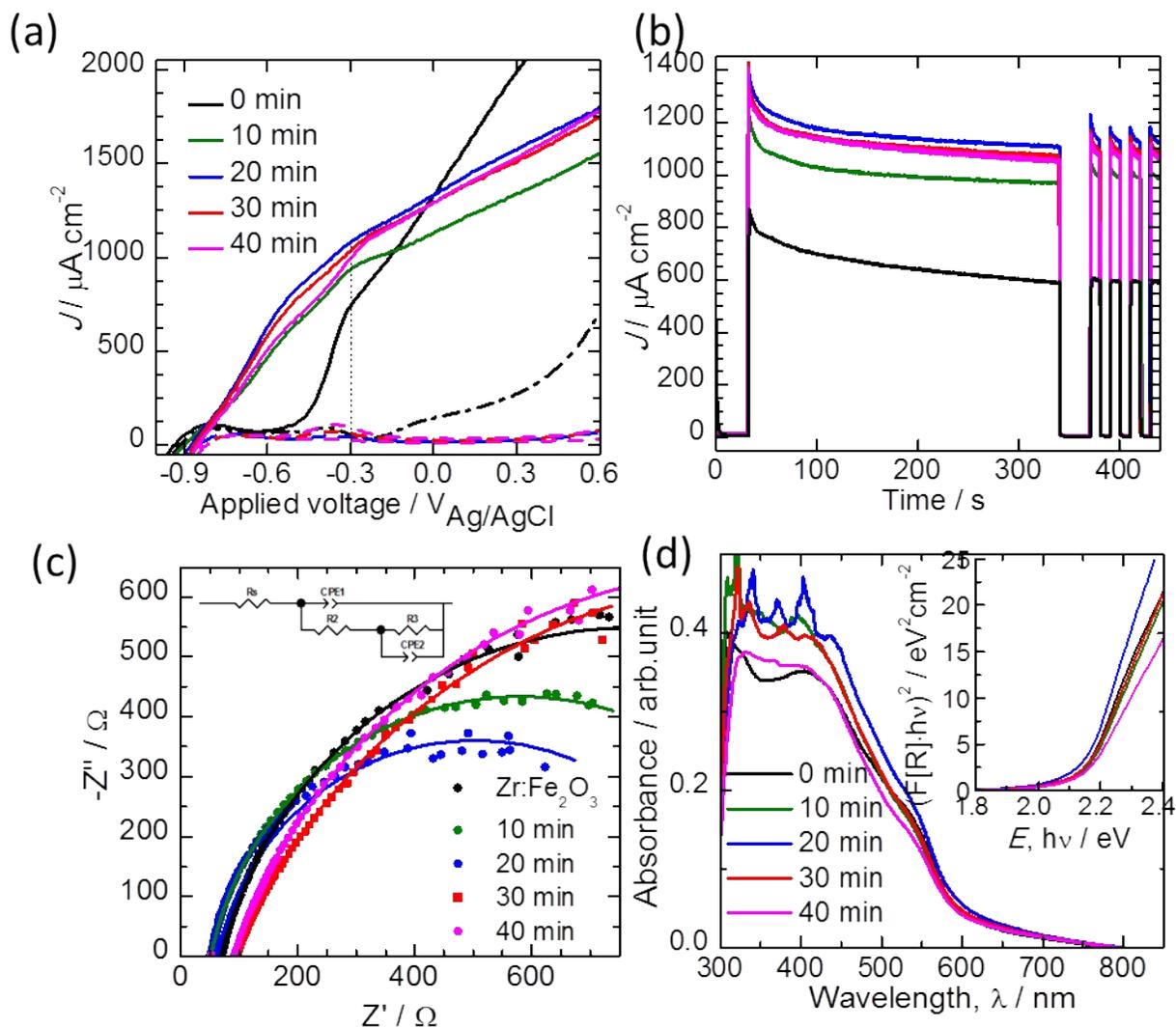
Annealing temperature/ EIS parameters	$R_s$ $\Omega$	$R_2$ $\Omega$	$R_3$ $\Omega$	$C_{[CPE1]}$ $\mu F$	$C_{[CPE2]}$ $\mu F$
250 °C	96	24	840	0.28	1.9
300 °C	62	17	734	0.20	3.0
350 °C	63	14	664	0.20	3.1
400 °C	62	19	861	0.29	2.1

**Table S6:** EIS fitting parameters of the bare CdS coated Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>, and 5, 10 mM Ni(OH)<sub>2</sub> loaded CdS coated Zr:Fe<sub>2</sub>O<sub>3</sub> heterostructured array.

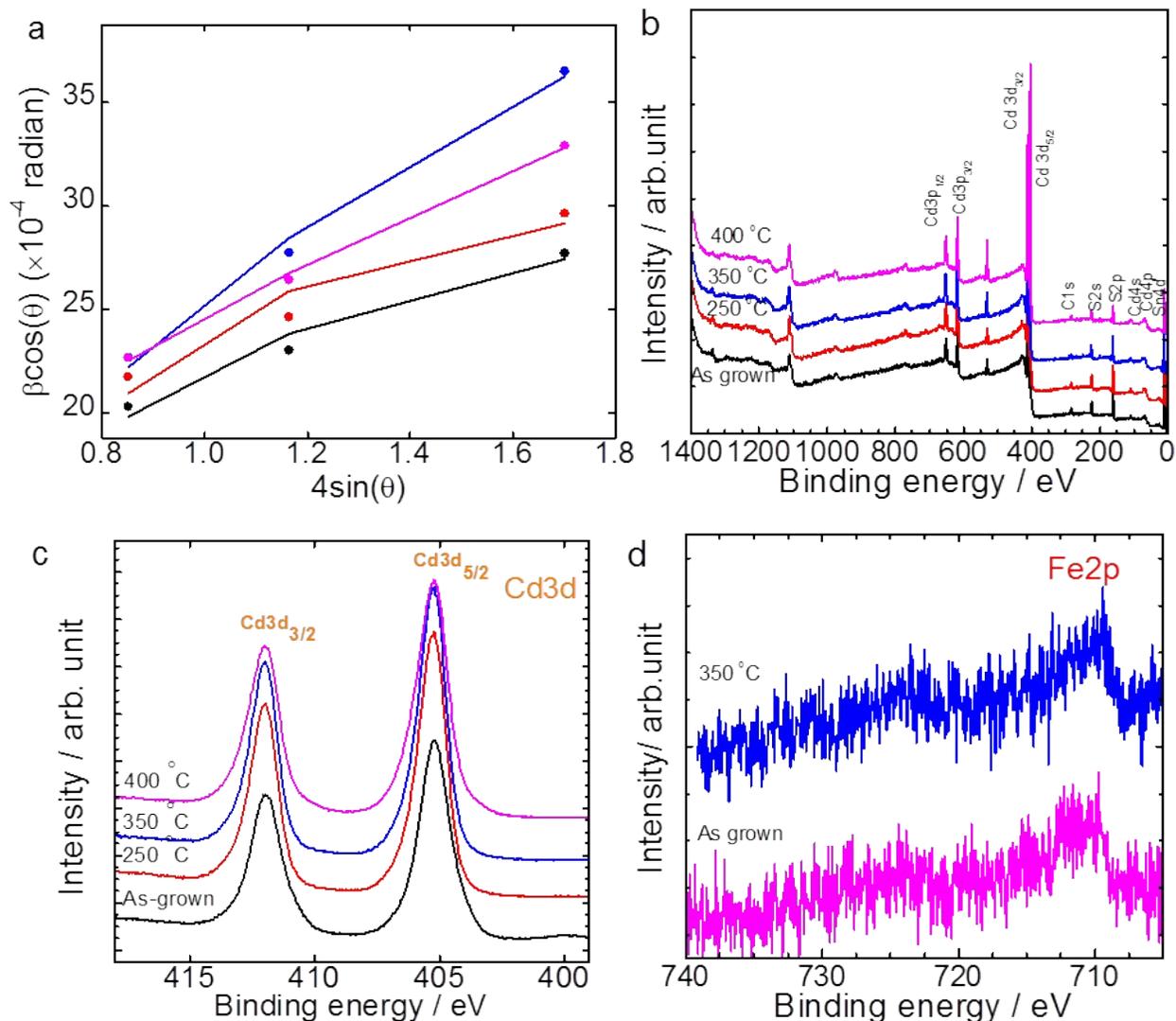
Sample/ parameters	$R_s$ $\Omega$	$R_2$ $\Omega$	$R_3$ $\Omega$	$C_{[CPE1]}$ $\mu F$	$C_{[CPE2]}$ $\mu F$
CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	63	14	664	0.20	3.1
5 mM Ni(OH) <sub>2</sub> loaded CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	124	7.5	700	17	34
10 mM Ni(OH) <sub>2</sub> loaded CdS/1D Zr:Fe <sub>2</sub> O <sub>3</sub>	99	7.4	883	0.24	33



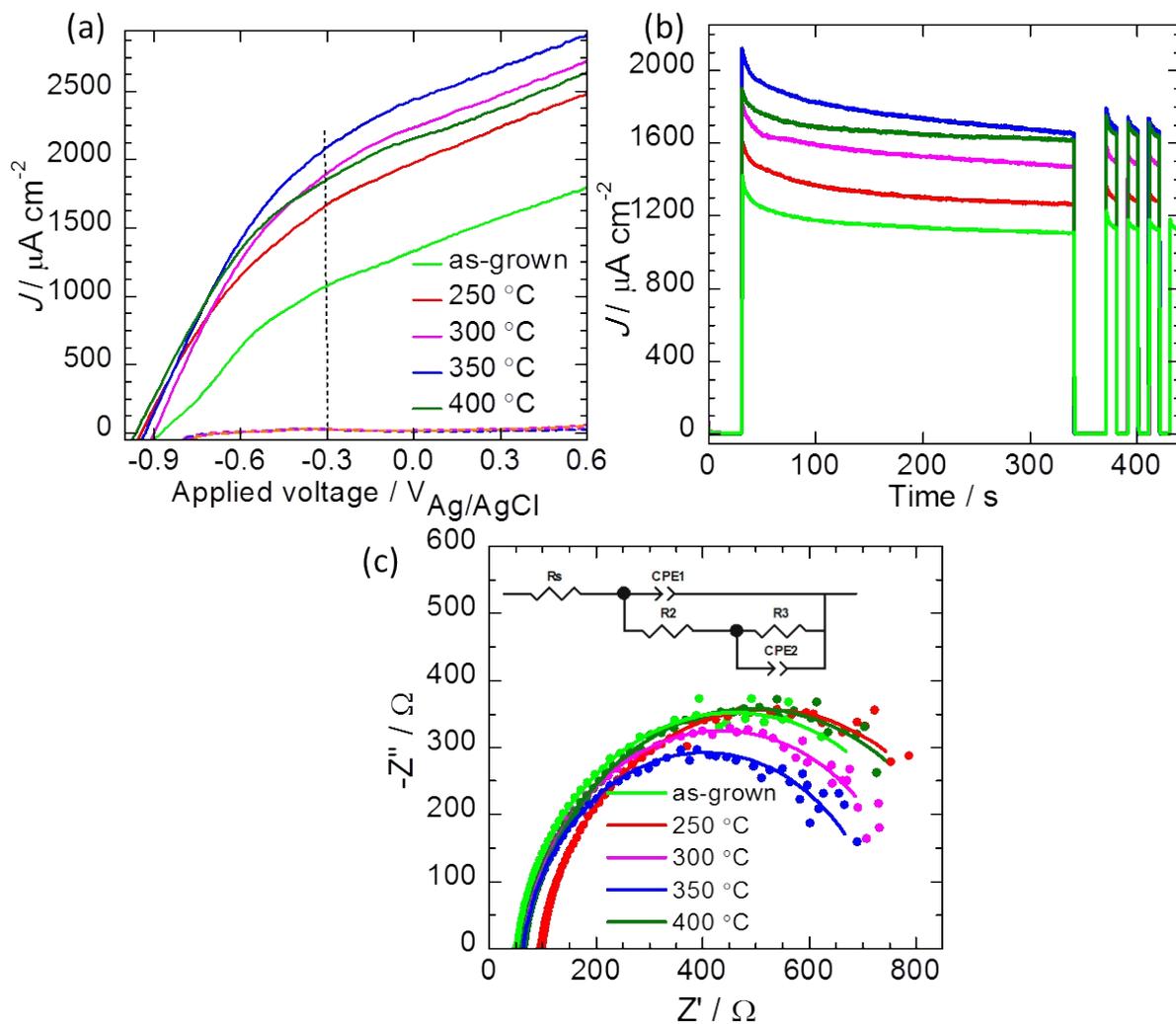
**Fig. S1:** FE-SEM cross-sectional images of (a) pristine 1D Zr:Fe<sub>2</sub>O<sub>3</sub>, and CdS deposited on 1D Zr:Fe<sub>2</sub>O<sub>3</sub> for (b) 10 min, (c) 20 min, (d) 30 min, (e) 40 min, respectively.



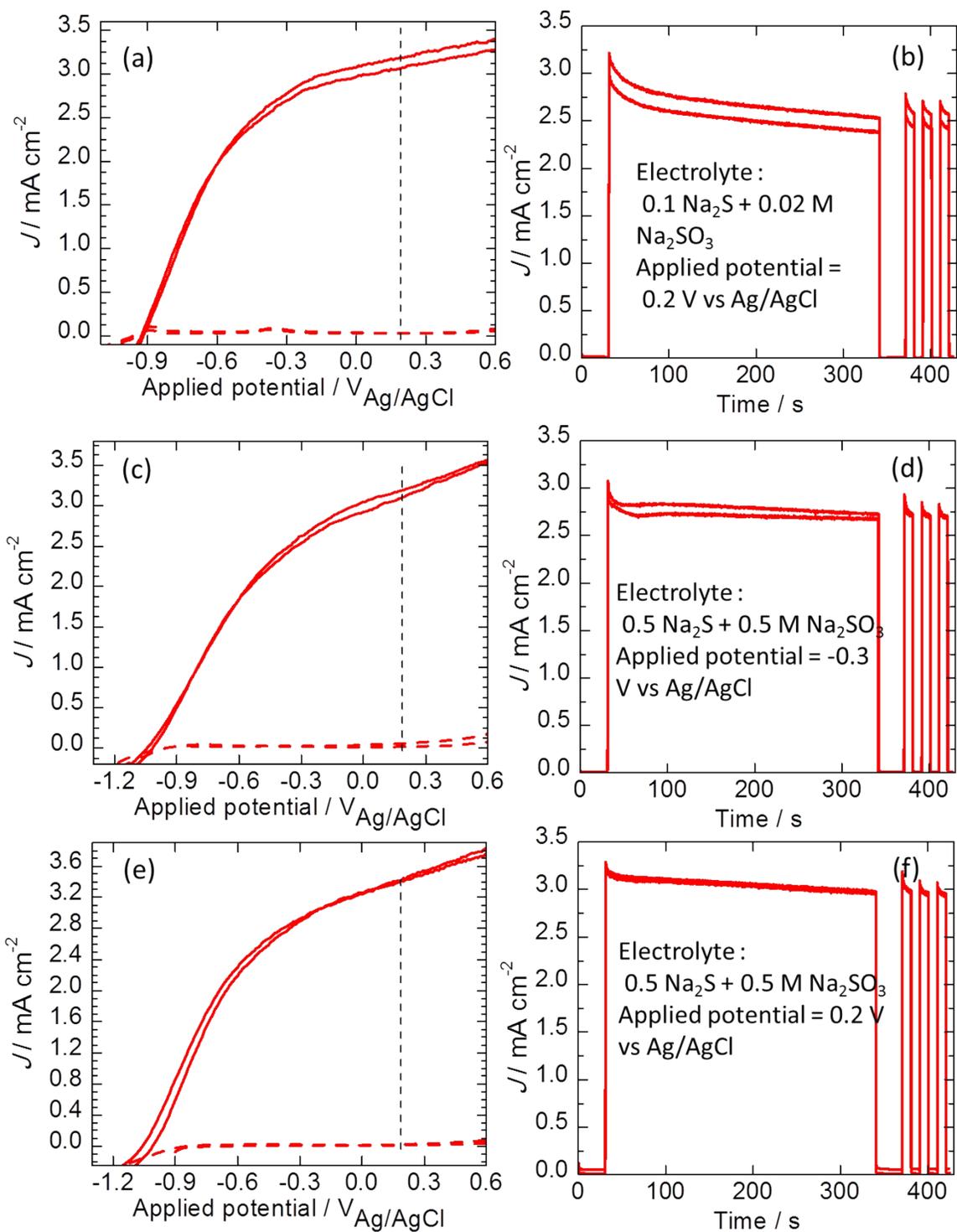
**Fig. S2:** (a) Current density-voltage characteristics (solid lines) and in the dark (dash lines) at a scan rate of  $50 \text{ mVs}^{-1}$ , (b) Potentiostatic photocurrent density-time characteristic, under simulated AM 1.5G illumination, (c) EIS spectrum of CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> (d) UV-Vis absorption spectrum for CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> deposited at (a), 10 min, 20 min, 30 min, and 40 min. Inset shows the corresponding band gaps.



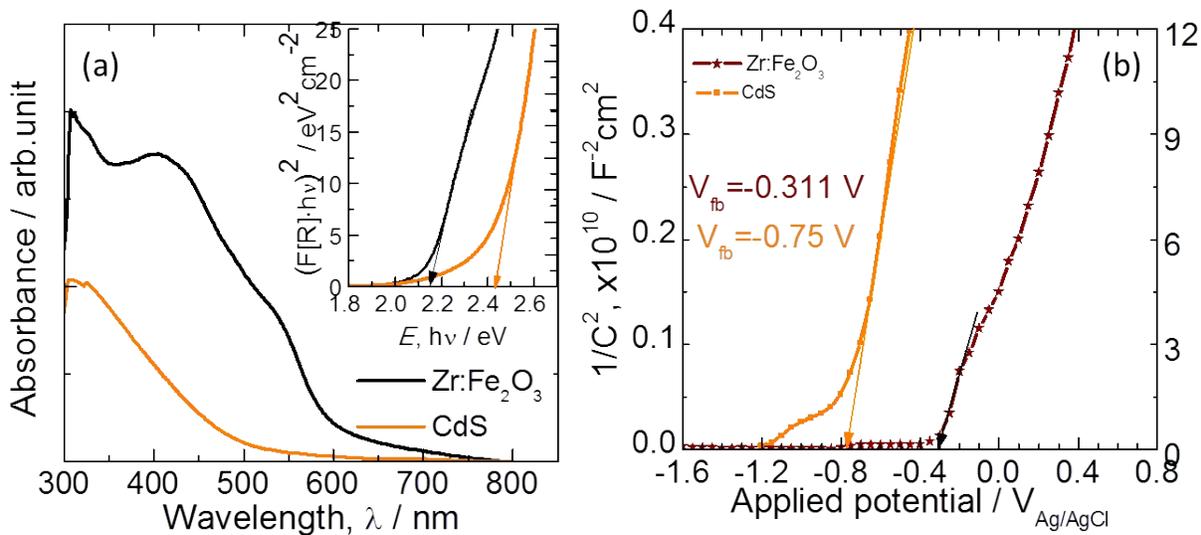
**Fig. S3:** (a) Williamson-Hall plot for the as-grown and the annealed CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> photoanodes, (b) XPS survey scan and (c,d) narrow scan spectra of Cd3d and Fe2p for CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> heterostructure photoanodes. The presence of Cd peaks (Cd 3d<sub>5/2</sub> at 405.3 eV and Cd 3d<sub>3/2</sub> at 411.9 eV) corresponds to the presence of the oxidation state +2 of Cd 3d in CdS<sup>4-6</sup>.



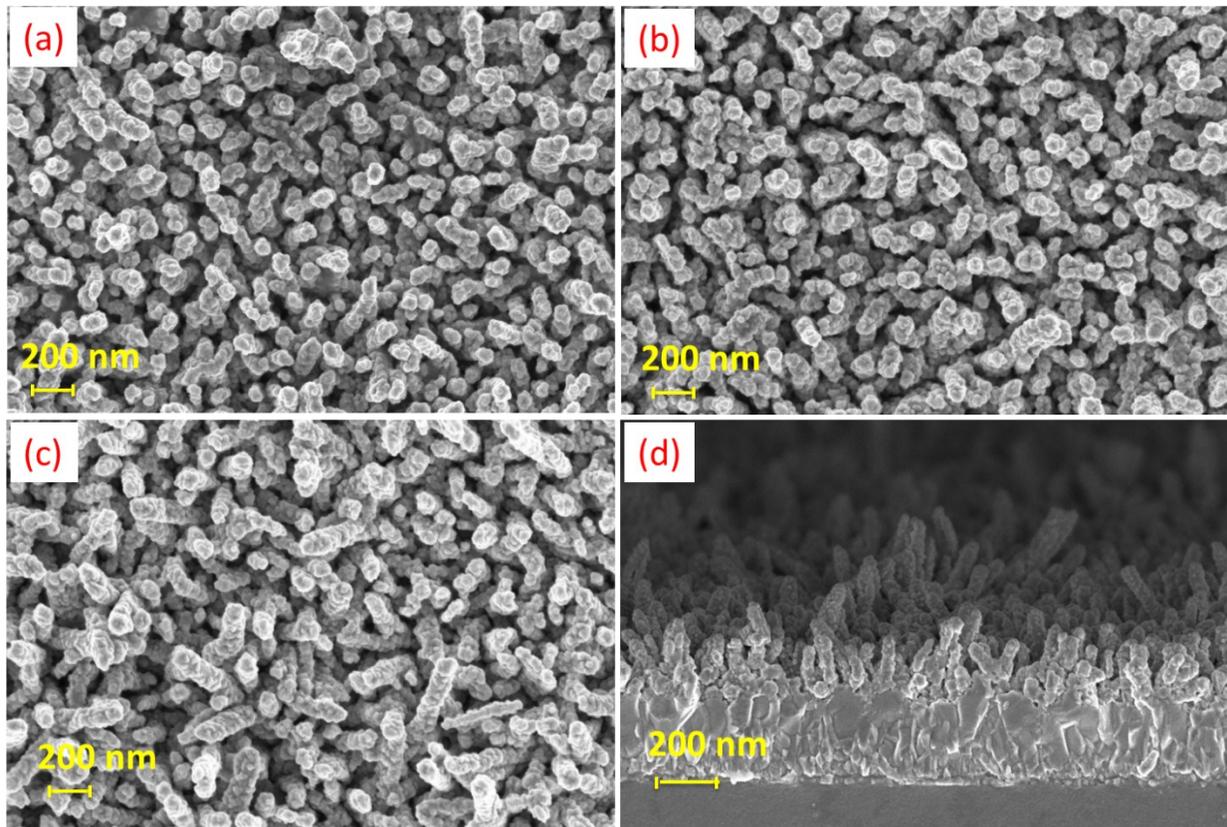
**Fig. S4:** (a) Current density-voltage characteristics (solid lines) and in the dark (dash lines), (b) Potentiostatic photocurrent density-time characteristic, (c) EIS spectra under simulated AM 1.5G illumination for CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> annealed at different temperatures, Inset shows the simple equivalent circuit used to fit the EIS spectra.



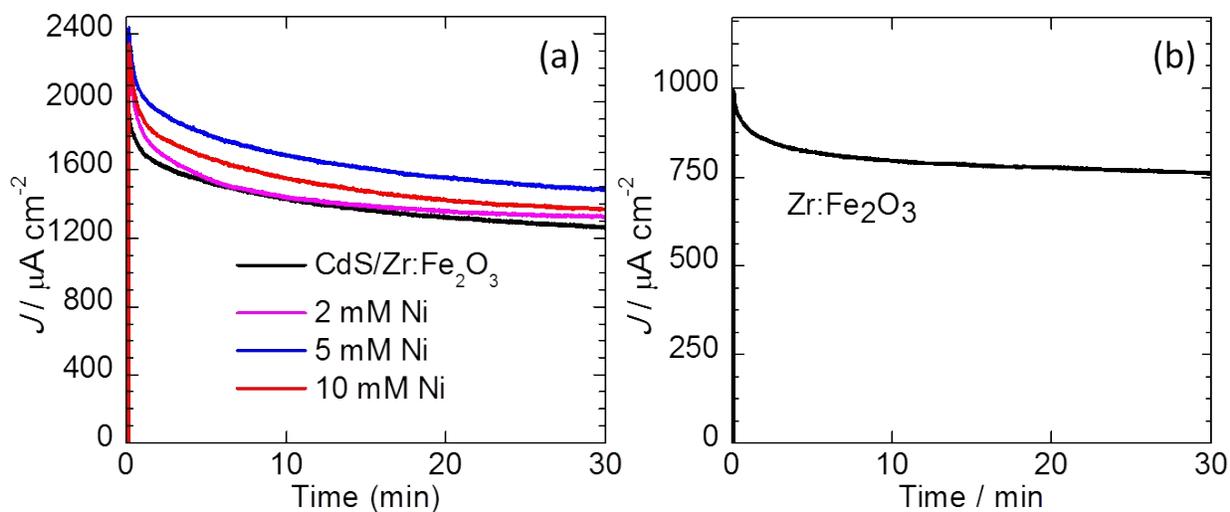
**Fig. S5:** Current density-voltage characteristics (a,c, e) and potentiostatic photocurrent density-time (b,d,f) characteristics of annealed CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> at different experimental conditions.



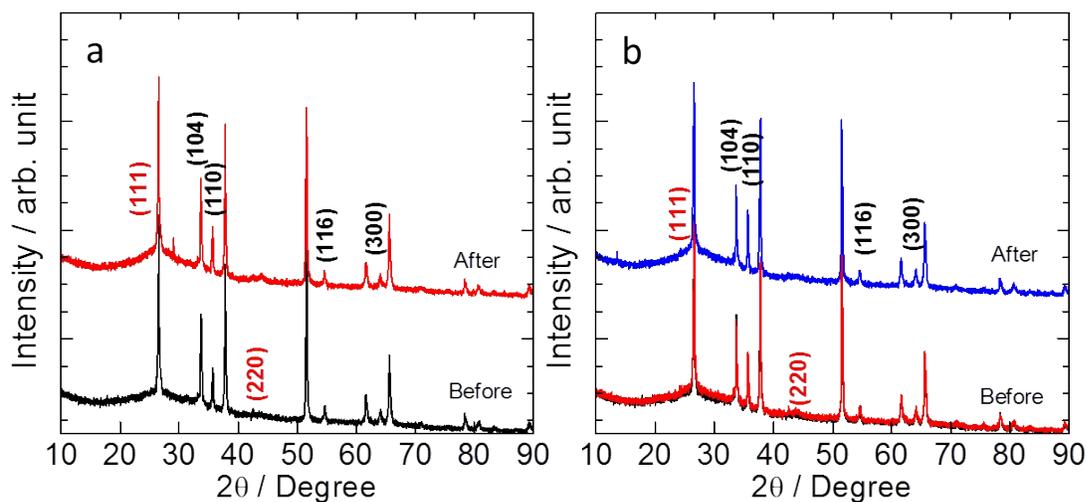
**Fig. S6:** (a) UV-Visible absorption spectrum. Inset shows the corresponding band gaps, (b) Mott-Schottky plots measured in a 0.1 M Na<sub>2</sub>S + 0.02 M Na<sub>2</sub>SO<sub>3</sub> electrolyte for bare CdS and 1D Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> on FTO.



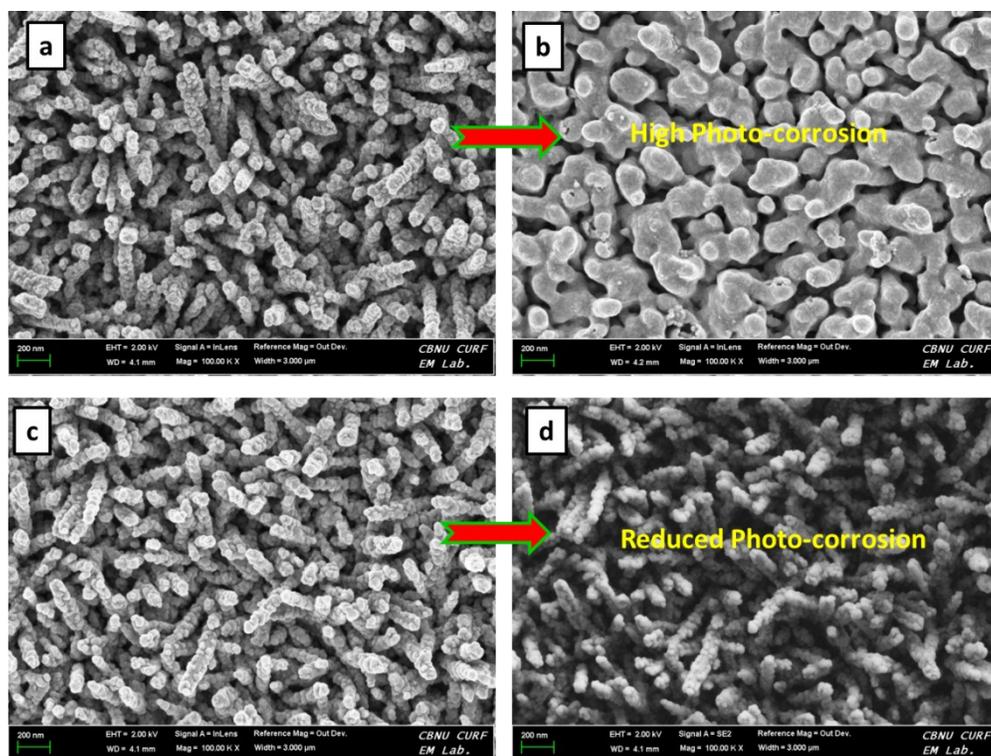
**Fig. S7** Top-view FE-SEM images of (a) 2 mM, (b) 5 mM, (c) 10 mM, and (d) cross-sectional FE-SEM image of the 5 mM nickel precursor loaded on annealed CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub>.



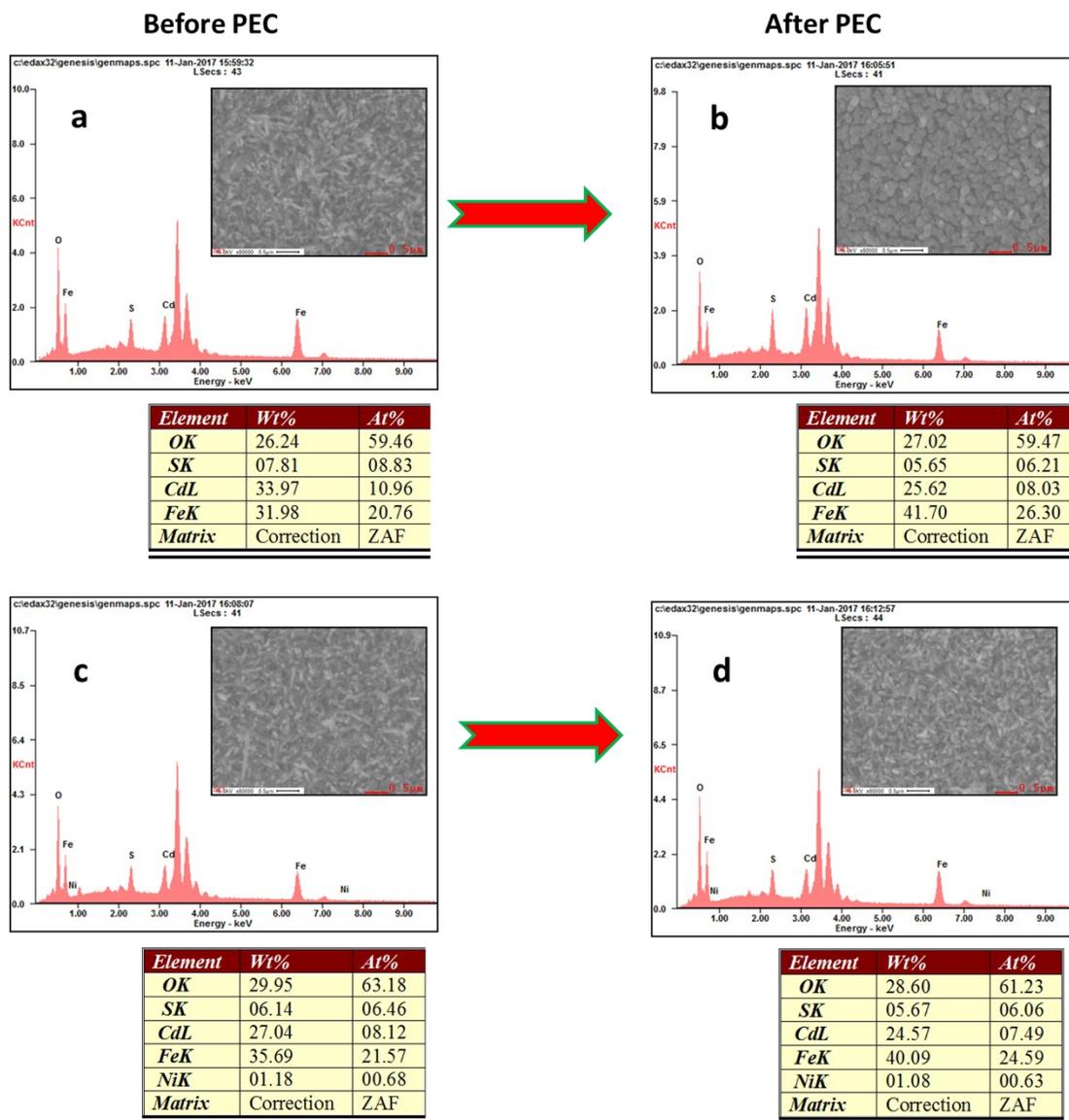
**Fig. S8** Stability test of Ni(OH)<sub>2</sub>/CdS/1D Zr:Fe<sub>2</sub>O<sub>3</sub> and pristine 1D Zr:α-Fe<sub>2</sub>O<sub>3</sub> deposited on FTO substrates.



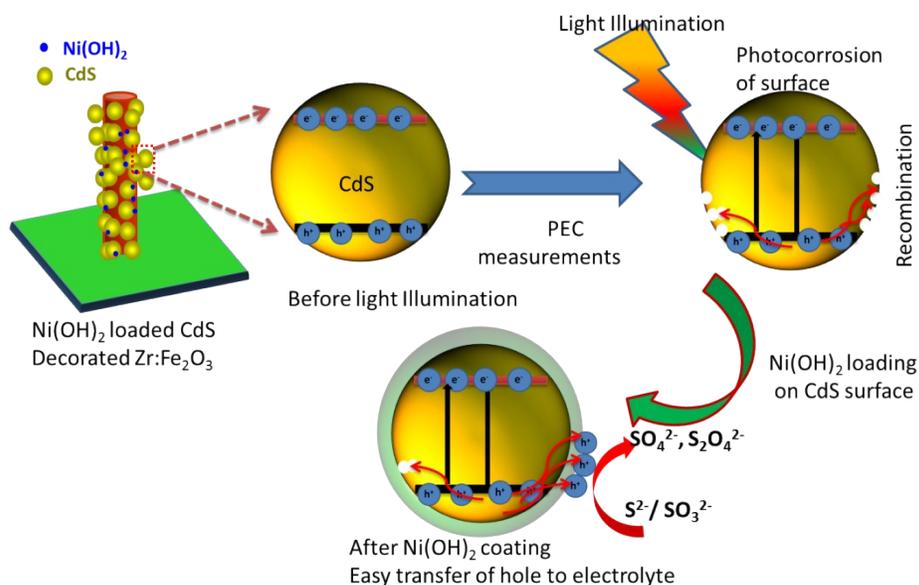
**Fig. S9** XRD patterns of (a) CdS nanograin-sensitized 1D Zr:α-Fe<sub>2</sub>O<sub>3</sub> and (b) Ni(OH)<sub>2</sub> loaded CdS nanograin-sensitized 1D Zr:α-Fe<sub>2</sub>O<sub>3</sub> nanorod arrays before and after PEC measurements.



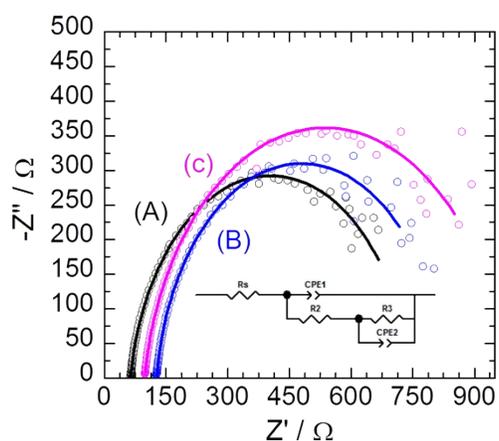
**Fig. S10.** FESEM of (a, c) CdS nanograin-sensitized 1D Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and Ni(OH)<sub>2</sub> loaded CdS nanograin-sensitized 1D Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> nanorod arrays before PEC measurements and (b,d) are of after measurements.



**Fig. S11.** EDS of (a, c) CdS nanograin-sensitized 1D Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and Ni(OH)<sub>2</sub> loaded CdS nanograin-sensitized 1D Zr: $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> nanorod arrays before PEC measurements and (c,d) are the after measurements.



**Fig. S12.** Schematic presentation of prevention of CdS photo-corrosion by  $\text{Ni(OH)}_2$  protecting layer.



**Fig. S13 .** EIS Nyquist plots of (A) CdS nanograin-sensitized 1D  $\text{Zr:}\alpha\text{-Fe}_2\text{O}_3$ , (B) 5 mM and (C) 10 mM  $\text{Ni(OH)}_2$  loaded CdS nanograin-sensitized 1D  $\text{Zr:}\alpha\text{-Fe}_2\text{O}_3$  films at -0.3 V vs Ag/AgCl.

## References

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