Supporting Information: Rotsides and Woerpel

Diastereoselective Silylene Transfer Reactions to Chiral Enantiopure Alkenes: Effects of Ligand Size and Substrate Bias

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Supporting Information

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¹H NMR, and ¹³C NMR spectra were recorded at ambient temperature using Bruker AV–400 (400 and 100 MHz, respectively), AV-500 (500 and 125 MHz, respectively) or AVIII-600 (600 and 150 MHz, respectively) spectrometers, as indicated. ²⁹Si NMR spectra were recorded at ambient temperature using Bruker AVIII-400 (79 MHz) or AV-500 (99 MHz) spectrometers, as indicated. The data are reported as follows: chemical shift in ppm from internal tetramethylsilane or referenced to residual solvent (¹H NMR: $C_6D_6 \delta$ 7.16; CDCl₃ δ ¹³C NMR: C₆D₆ δ 128.1; CDCl₃ δ 77.2. ²⁹Si NMR: referenced to external 7.26. tetramethylsilane C₆D₆ δ 0; CDCl₃ δ 0.) on the δ scale, multiplicity (appar = apparent, br = broad, s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants (Hz), and integration. Due to difficulties with purification for certain products, only distinctive peaks are listed in tabulated ¹H NMR and ¹³C NMR spectroscopic data as indicated. ¹H NMR yields were determined relative to a known concentration of internal standard, mesitylene, using a single scan. Diastereomer ratios were determined using a combination of ¹H (single scan) and ¹³C NMR spectroscopy.¹ Infrared (IR) spectra were obtained using a Thermo Nicolet AVATAR 360 FT-IR 5000 spectrometer using attenuated total reflectance (ATR). High-resolution mass spectra (HRMS) were acquired on an Agilent 6224 Accurate-Mass time-of-flight spectrometer and were obtained by peak matching. Microanalyses were performed by Atlantic Microlab Inc., Norcross, GA. Optical rotations were obtained using a digital polarimeter. Analytical thin layer chromatography was performed on Silicycle silica gel 60 Å F₂₅₄ plates. Liquid chromatography was performed using forced flow (flash chromatography) of the indicated solvent system on Silicycle silica gel (SiO₂) 60 (230-400 mesh). Tetrahydrofuran, methylene chloride, 1,2-dichloroethane, chloroform, toluene, diethyl ether and triethylamine were dried by

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filtration through alumina according to the method of Grubbs.² C_6D_6 and trifluorotoluene were dried over 3 Å molecular sieves for 48 h and degassed prior to use. Dimethyl sulfoxide was distilled over CaH₂. Ethanol was degassed prior to use. All reactions were run under an atmosphere of nitrogen in glassware that was flame-dried under a stream of nitrogen unless otherwise stated. Silacyclopropanes were stored and manipulated in a Vacuum Atmospheres nitrogen-atmosphere drybox. Cyclohexene silacyclopropane **2** was synthesized by known methods.³ Methyltriphenylphosphonium iodide was prepared by known methods.⁴

II. Preparation of Starting Materials



 $\begin{array}{c} \underset{M \in \mathcal{O}}{\overset{\circ}{}} \underset{S2}{\overset{\circ}{}} \\ (S)-Ethyl 2-(Triisopropylsiloxy) propanoate S2. To a cooled (0 °C) solution of ethyl L-lactate S1 (8.6 mL, 75 mmol) in CH₂Cl₂ (190 mL) were added triisopropylsilyl chloride (17.5 mL, 82.0 mmol) and imidazole (7.70 g, 113 mmol). The reaction mixture was allowed to warm to room temperature (22 °C). After 16 h, hexanes (50 mL) and H₂O (50 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 20 mL). The combined organic layers were washed with H₂O (40 mL) and brine (60 mL), dried over Na₂SO₄, and concentrated$ *in vacuo* $to provide protected lactate S2 as a colorless oil and was used without further purification (20.4 g, 100%). The spectroscopic data are consistent with the data reported:⁵ <math>[a]^{22}_{D}$ –16.99 (*c* 0.90, CHCl₃); ¹H

NMR (600 MHz, CDCl₃) δ 4.40 (q, J = 6.7, 1H), 4.22–4.14 (m, 2H), 1.42 (d, J = 6.7, 3H), 1.28 (t, J = 7.1, 3H), 1.07–1.05 (m, 21H); HRMS (TOF MS ES+) m / z calcd for C₁₄H₃₁O₃Si (M+H)⁺275.2037, found 275.2038.

Me Aldehyde S3. To a cooled (-78 °C) solution of ester S2 (2.1 g, 7.7 mmol) in Et₂O S3 (75 mL) was added DIBAL-H (11 mL, 1.0 M in hexanes, 11 mmol) dropwise over 5 min After 1 h. 2.0 M acueous Bochelle Salt (30 mL) and H₂O (30 mL) were added. The

min. After 1 h, 2.0 M aqueous Rochelle Salt (30 mL) and H₂O (30 mL) were added. The aqueous layer was extracted with Et₂O (3 x 40 mL) and the combined organic layers were washed with brine (50 mL), dried over Mg₂SO₄ and concentrated *in vacuo* to provide aldehyde **S3** as a pale yellow oil that was used without further purification (1.7 g, 97%). The spectral data are consistent with the data reported.⁶ [a]²⁰_D-8.10 (*c* 1.10, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 9.66 (d, *J* = 1.7, 1H), 4.18 (qd, 6.8, 1.6, 1H), 1.31 (d, *J* = 6.8, 3H), 1.08–1.07 (m, 21H); ¹³C NMR (125 MHz, CDCl₃) δ 204.8, 74.0, 19.1, 18.1, 12.3; HRMS (TOF MS ES+) *m* / z calcd for C₁₂H₂₇O₂Si (M+H)⁺ 231.1775, found 231.1777.

Enones S5. To a cooled (-78 °C) of triethyl phosphonoacetate **S4** (3.6 mL, 24 mmol) in THF (75 mL) was added *n*-BuLi (9.0 mL, 2.5 M in hexanes, 23 mmol) dropwise over 3 min. After 30 min, aldehyde **S3** (3.5 g, 15 mmol) was added as a solution in THF (30 mL), by cannula. The reaction mixture was warmed to room temperature (22 °C). After 16 h, the reaction mixture was concentrated *in vacuo* and CH₂Cl₂ (30 mL) was added to the residue. The organic layer was washed with 1 M HCl (10 mL), H₂O (15 mL), and brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) provided enones **S5** as a clear oil (2.7 g, 59%, 74:26 d.r.): ¹H NMR (400 MHz, CDCl₃) δ 6.95 (dd, J = 15.5, 4.4, 1H), 6.26 (dd, J = 11.7, 7.6,

0.26H), 6.02 (dd, J = 15.5, 1.6, 1H), 5.64 (dd, J = 11.7, 1.4, 0.26H), 5.55–5.48 (m, 0.26H), 4.60–4.54 (m, 1H), 4.25–4.14 (m, 2.6H), 1.31–1.26 (m, 7.8H), 1.08–1.04 (m, 27H); ¹³C NMR (100 MHz, CDCl₃) δ 167.1, 166.1, 155.4, 152.4, 119.1, 116.8, 68.0, 65.8, 60.5, 60.3, 24.1, 19.6, 18.23, 18.21, 18.14, 18.11, 18.07, 14.4, 12.5, 12.3; IR (ATR) 1720, 1090, 825 cm⁻¹; HRMS (TOF MS ES+) m / z calcd for C₁₆H₃₃O₃Si (M+H)⁺ 301.2193, found 301.2200. Anal. Calcd for C₁₆H₃₂O₃Si: C, 63.95; H, 10.73. Found: C, 64.03; H, 10.80.

Ester S6. To a solution of enones **S5** (1.5 g, 5.0 mmol) in EtOAc (25 mL) was added Pd/C (0.53 g, 0.5 mmol Pd). The solution was sparged with N₂ and then stirred vigorously under an H₂ atmosphere (balloon). After 3 days, the reaction mixture was filtered through a short pad of Celite and concentrated *in vacuo*. Purification by flash chromatography (1:99 EtOAc:hexanes) provided ester **S6** as a colorless oil (1.26 g, 83%): [a]¹⁷_D +8.03 (*c* 2.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 4.12 (q, *J* = 7.1, 2H), 4.07–3.98 (m, 1H), 2.46–2.33 (m, 2H), 1.85–1.72 (m, 2H), 1.25 (t, *J* = 7.2, 3H), 1.17 (d, *J* = 6.1, 3H), 1.06 (s, 21H); ¹³C NMR (100 MHz, CDCl₃) δ 174.2, 67.6, 60.6, 34.7, 30.1, 23.5, 18.33, 18.30, 14.4, 12.7; IR (ATR) 2943, 1464, 1091, 848 cm⁻¹; HRMS (TOF MS ES+) *m* / *z* calcd for C₁₆H₃₅O₃Si (M+H)⁺ 303.2350, found 303.2364. Anal. Calcd for C₁₆H₃₄O₃Si: C, 63.52; H, 11.33. Found: C, 63.79; H, 11.31.

Me OTIPS OH OTIPS S7
 Alcohol S7. To a cooled (0 °C) solution of LiAlH₄ (0.31 g, 7.9 mmol) in Et₂O (24 mL) was added ester S6 (1.2 g, 4.0 mmol) as a solution in Et₂O (40 mL) by cannula. After 1h, H₂O (20 mL) was added followed by 10% aqueous NaOH (100 mL) and H₂O (60 mL). The layers were separated and the aqueous layer was extracted with diethyl ether (5 x 15 mL). The combined organic layers were washed with brine (30 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography provided alcohol S7

as a colorless oil (0.95 g, 91%): $[a]^{20}_{D}$ +6.44 (*c* 1.10, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 4.08–4.02 (m, 1H), 3.69–3.59 (m, 2H), 2.05 (br s, 1H), 1.74–1.55 (m, 4H), 1.20 (d, *J* = 6.2, 3H), 1.07 (s, 21H); ¹³C NMR (125 MHz, CDCl₃) δ 68.4, 63.4, 36.4, 28.3, 23.1, 18.32, 18.28, 12.6; IR (ATR) 3328, 2942, 1463 cm⁻¹; HRMS (TOF MS ES+) *m* / z calcd for C₁₄H₃₃O₂Si (M+H)⁺ 261.2244, found 261.2248. Anal. Calcd for C₁₄H₃₂O₂Si: C, 64.55; H, 12.38. Found: C, 64.48; H, 12.56.

^{Me} Aldehyde S8. To a cooled (-78 °C) solution of DMSO (0.11 mL, 1.5 mmol) in CH₂Cl₂ (10 mL) was added oxalyl chloride (0.10 mL, 1.2 mmol) dropwise over 2 min. After 30 min, alcohol S7 (0.20 g, 0.76 mmol) was added as a solution in CH₂Cl₂ (1.5 mL) by cannula. After 30 min, Et₃N (0.55 mL, 4.0 mmol) was added. After 2 h, H₂O (5 mL) was added, the layers were separated, and the aqueous layer was extracted with CH₂Cl₂ (3 x 5 mL). The combined organic layers were washed with H₂O (3 x 5 mL) and brine (10 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (5:95 EtOAc:hexnes) provided aldehyde S8 as a yellow oil (0.18 g, 90%). The aldehyde was used immediately in subsequent steps: [a]¹⁸_D+12.85 (*c* 1.25, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 9.80 (t, *J* = 1.6, 1H), 4.09–4.04 (m, 1H), 2.59–2.49 (m, 2H), 1.88–1.82 (m, 1H), 1.79–1.73 (m, 1H), 1.18 (d, *J* = 6.1, 3H), 1.06 (s, 21H); ¹³C NMR (150 MHz, CDCl₃) δ 202.9, 67.5, 39.7, 31.8, 23.5, 18.33, 18.29, 12.7; HRMS (TOF MS ES+) *m* / *z* calcd for C₁₄H₃₀NaO₂Si (M+Na)⁺ 281.1907, found 281.1925.

Me Alkene 4. To a solution methyltriphenylphosphonium iodide (0.55 g, 1.4 mmol) in THF (3.6 mL) was added *t*-BuOK (0.19 g, 1.7 mmol). After 2 h, aldehyde S8 (0.28 g, 1.1 mmol) was added as a solution in THF (1.8 mL). After 20 min hexanes (10 mL) and saturated aqueous NH₄Cl (10 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 5 mL). The combined organic layers were washed with H₂O (10 mL) and brine (15 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided alkene **4** as a colorless oil (0.22 g, 79%): $[a]^{20}_{D}$ +5.75 (*c* 0.94, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 5.86–5.79 (m, 1H), 5.01 (ddd, *J* = 17.1, 3.4, 1.6, 1H), 4.95–4.92 (m, 1H), 3.98–3.93 (m, 1H), 2.13–2.08 (m, 2H), 1.64–1.58 (m, 1H), 1.55–1.49 (m, 1H), 1.17 (d, *J* = 6.1, 3H), 1.06 (s, 21H); ¹³C NMR (150 MHz, CDCl₃) δ 139.2, 114.4, 68.2, 39.2, 29.8, 23.6, 18.4, 18.3, 12.7; IR (ATR) 2866, 1642 cm⁻¹; HRMS (TOF MS ES+) *m* / z calcd for C₁₅H₃₂NaOSi (M+Na)⁺ 279.2115, found 279.2109. Anal. Calcd for C₁₅H₃₂OSi: C, 70.24; H, 12.57. Found: C, 70.50; H, 12.73.



Diene 6. To a solution of allyltrimethylphosphonium bromide **S9** (0.087 g, 0.44 mmol) in THF (1 mL) was added *t*-BuOK (0.052 g, 0.46 mmol). After 30 min, aldehyde **S3** (0.106 g, 0.460 mmol) was added as a solution in THF (0.5 mL). After 16 h Et₂O (3 mL) and H₂O (5 mL) were added, the layers were separated, and the aqueous layer was extracted with Et₂O (5 x 3 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided diene **6** as a colorless oil (0.041 g, 35%): $[a]^{19}_{D}$ +9.71 (*c* 0.58, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 6.32 (dt, *J* = 16.9, 10.3, 1H), 6.17 (dd, *J* = 15.2, 10.5, 1H), 5.73 (dd, *J* = 15.2, 5.9, 1H), 5.16 (d, *J* = 14.0, 1H), 5.04 (d, *J* = 10.1, 1H), 4.46–4.41 (m, 1H), 1.26 (d, *J* = 6.3, 3H), 1.06 (s, 21H); ¹³C NMR (125 MHz, CDCl₃) δ 139.3, 136.9, 128.9, 116.6, 69.0, 25.0, 18.2, 12.5; IR (ATR) 2943,

1463 cm⁻¹; HRMS (TOF MS ES+) m / z calcd for C₁₅H₃₁OSi (M+H)⁺ 255.2139, found 255.2141.⁶



Alcohol S11. To a cooled (-78 °C) solution of (*R*)-camphor S10 (3.1 g, 20 mmol) in Et₂O (100 mL) was added allylmagnesium chloride (25 mL, 1.0 M in Et₂O, 25 mmol) dropwise over 5 min. After 30 min, MeOH (5 mL) was added and the reaction mixture was warmed to 22 °C. H₂O (30 mL) was added, the layers were separated, and the aqueous layer was extracted with Et₂O (3 x 20 mL). The combined organic layers were washed with brine (25 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (5:95 EtOAc:hexanes) provided alcohol S11 as a yellow oil (3.8 g, 97%). The spectroscopic data are consistent with the data reported:⁷ [a]²⁰_D+8.02 (*c* 4.00, toluene); ¹H NMR (500 MHz, CDCl₃) δ 5.98–5.90 (m, 1H), 5.18–5.14 (m, 2H), 2.34–2.25 (m, 2H), 1.97 (dt, *J* = 13.1, 3.7, 1H), 1.76–1.69 (m, 3H), 1.44–1.42 (m, 2H), 1.10 (s, 3H), 1.06–1.02 (m, 1H), 0.86 (s, 3H), 0.85 (s, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 135.2, 119.1, 79.9, 52.4, 49.6, 46.2, 45.2, 44.7, 30.7, 27.2, 21.6, 21.1, 11.1.



Homoallylic Ether 18. To a cooled (0 °C) solution of NaH (0.31 g, 60% in mineral oil, 7.9 mmol) in THF (10 mL) was added alcohol **S11** (0.75 g, 3.9 mmol) as a solution in THF (5 mL) by cannula. The reaction mixture was allowed to warm to room temperature (22 °C) over 3 h,

and iodomethane (0.60 mL, 9.7 mmol) was added. After 16 h, saturated aqueous NaHCO₃ (10 mL) and H₂O (10mL) were added and the aqueous layer was extracted with Et₂O (3 x 15 mL). The combined organic layers were washed with brine (20 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided allyl ether **18** as a colorless oil (0.70 g, 87%): $[a]^{20}_{D}$ +9.91 (*c* 0.40, CHCl₃); ¹H NMR (500 MHz, CDCl₃) δ 5.94 (dddd, *J* = 17.2, 10.1, 8.9, 5.4, 1H), 5.08–5.00 (m, 2H), 3.13 (s, 3H), 2.71–2.66 (m, 1H), 2.20 (dt, *J* = 12.9, 3.7, 1H), 2.13 (dd, *J* = 15.7, 8.7, 1H), 1.76–1.68 (m, 1H), 1.66–1.65 (m, 1H), 1.57–1.52 (m, 1H), 1.38 (ddd, *J* = 13.3, 11.8, 5.6, 1H), 1.03 (ddd, *J* = 12.6, 9.2, 5.7, 1H), 0.96 (s, 3H), 0.88 (s, 3H and m, 1H), 0.82 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 136.4, 115.8, 85.6, 53.1, 50.3, 47.8, 45.1, 41.1, 40.0, 30.8, 27.4, 21.2, 21.1, 12.3; IR (ATR) 3074, 2963, 1639 cm⁻¹; Anal. Calcd for C₁₄H₂₄O: C, 80.71; H, 11.61. Found: C 80.95; H, 11.75.

$$\begin{array}{c} \mathsf{OH} \\ \mathsf{Ph} & \underbrace{\mathsf{TBDPSCl, imidazole}}_{\mathsf{S12}} & \underbrace{\mathsf{OTBDPS}}_{\mathsf{0}-22\ ^\circ\mathsf{C},\ \mathsf{CH}_2\mathsf{Cl}_2} & \underbrace{\mathsf{OTBDPS}}_{\mathsf{Ph}} & \underbrace{\mathsf{OTBDPS}}_{\mathsf{24}} \end{array}$$

Silyl Ether 24. To a cooled (0 °C) solution of alcohol S12 (0.51 g, 3.4 mmol) in CH₂Cl₂ (7 mL) were added imidazole (0.20 g, 4.1 mol) and *tert*-butyl(chloro)diphenylsilane (0.85 mL, 3.3 mmol). After 24 h, an additional portion of *tert*-butyl(chloro)diphenylsilane (0.5 mL, 1.9 mmol) was added. After 24 h, hexanes (15 mL) and H₂O (15 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 10 mL). The combined organic layers were washed with H₂O (15 mL) and brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided silyl ether **24** as a colorless oil (0.467 g, 37%): ¹H NMR (400 MHz, CDCl₃) δ 7.71–7.69 (m, 2H), 7.46–7.31 (m, 6H), 7.26–7.18 (m, 7H), 5.58–5.48 (m, 1H), 4.87–4.78 (m, 2H), 4.72–4.69 (m, 1H), 2.48–2.35 (m, 2H), 1.04 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 144.2, 136.13, 136.10, 134.5, 134.4,

133.8, 129.8, 129.6, 128.0, 127.7, 127.5, 127.1, 126.5, 117.3, 75.7, 44.9, 27.2, 19.5; IR (ATR) 2931, 1454 cm⁻¹; HRMS (TOF MS ES+) *m* / z calcd for C₂₆H₃₀NaOSi (M+Na)⁺ 409.1958, found 409.1971. Anal. Calcd for C₂₆H₃₀OSi: C, 80.78; H, 7.82. Found: C, 80.74; H, 7.66.

(*S*)-Alcohol S12. To a solution of (*R*)-BINOL (0.29 g, 1.0 mmol) in CH₂Cl₂ (20 mL) with powdered molecular sieves (4 Å, 4.3 g) was added Ti(O*i*-Pr)₄ (0.30 mL, 1.0 mmol) and the reaction mixture was heated to reflux. After 1 h, the reaction mixture was cooled to 22 °C and benzaldehyde S13 (1.0 mL, 10 mmol) was added. After 15 min, the reaction mixture was cooled to -78 °C and allyltributyltin S14 (4.0 mL, 13 mmol) was added. After 10 min, the reaction mixture was brought to -20 °C. After 3 d, the reaction mixture was warmed to 22 °C, diluted with CH₂Cl₂ (20 mL), and filtered. The organic layer was washed with saturated aqueous NaHCO₃ (20 mL) and brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (10:90 EtOAc:hexanes) provided alcohol (*S*)-S12 as a yellow oil (0.52 g, 35%). The spectroscopic data are consistent with the data reported:⁸ [a]²⁰_D +50.48 (*c* 0.53, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 7.38–7.34 (m, 4H), 7.29–7.26 (m, 1H), 5.85–5.78 (m, 1H), 5.19–5.14 (m, 2H), 4.75 (dd, *J* = 7.9, 5.0, 1H), 2.57–2.47 (m, 2H).

$$\begin{array}{c} OH \\ Ph \\ \hline (S)-S12 \end{array} \xrightarrow{TBSCl, imidazole} \\ 0-22 \ ^{\circ}C, \ CH_2Cl_2 \end{array} \xrightarrow{OTBS} \\ Ph \\ \hline (S)-S12 \end{array} \xrightarrow{OTBS} \\ (S)-23 \end{array}$$

(S)-Homoallylic Ether 23. To a cooled (0 °C) solution of alcohol (S)-S12 (0.50 g, 3.4 mmol) in CH_2Cl_2 (7 mL) were added imidazole (0.28 g, 4.1 mmol) and *tert*-butyl(chloro)dimethylsilane (0.51 g, 3.4 mmol) and the reaction mixture was allowed to warm

to room temperature (22 °C). After 16 h, hexanes (5 mL) and H₂O (5 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 5 mL). The combined organic layers were washed with H₂O (5 mL) and brine (10 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided protected alcohol (*S*)-23 as a colorless oil (0.50 g, 57%): $[a]^{18}_{D}$ +47.22 (*c* 1.00, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 7.32–7.29 (m, 4H), 7.24–7.21 (m, 1H), 5.81–5.74 (m, 1H), 5.03–4.99 (m, 2H), 4.69–4.67 (m, 1H), 2.48–2.36 (m, 2H), 0.88 (s, 9H), 0.02 (s, 3H), -0.13 (s, 3H).

$$\begin{array}{c} O \\ Ph \end{array} + SnBu_3 \xrightarrow{\text{Ti}(Oi-Pr)_4, (S)-BINOL, 4 \text{ Å ms}} \\ S13 \\ S14 \end{array} \xrightarrow{\text{OH}} Ph \xrightarrow{OH} \\ \hline (R)-S12 \\ \hline$$

(*R*)-Alcohol S12. To a solution of (*S*)-BINOL (0.29 g, 1.0 mmol) in CH₂Cl₂ (20 mL) with powdered molecular sieves (4 Å, 4.3 g) was added Ti(O*i*-Pr)₄ (0.30 mL, 1.0 mmol) and the reaction mixture was heated to reflux. After 1 h, the reaction mixture was cooled to 22 °C and benzaldehyde S13 (1.0 mL, 10 mmol) was added. After 15 min, the reaction mixture was cooled to -78 °C and allyltributyltin S14 (4.0 mL, 13 mmol) was added. After 10 min, the reaction mixture was brought to -20 °C. After 3 d, the reaction mixture was warmed to 22 °C, diluted with CH₂Cl₂ (20 mL), and filtered. The organic layer was washed with saturated aqueous NaHCO₃ (20 mL) and brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (10:90 EtOAc:hexanes) provided alcohol (*R*)-S12 as a yellow oil (0.68 g, 47%): [a]²⁰_D-44.36 (*c* 1.55, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.38–7.34 (m, 4H), 7.30–7.27 (m, 1H), 5.87–5.77 (m, 1H), 5.20–5.13 (m, 2H), 4.77–4.72 (m, 1H), 2.58–2.46 (m, 2H).



(*R*)-Homoallylic Ether 23 To a cooled (0 °C) solution of alcohol (*R*)-S12 (0.68 g, 4.6 mmol) mL) were added imidazole (0.38 g, 5.5 mmol) and tertin CH_2Cl_2 (9) butyl(chloro)dimethylsilane (0.56 g, 3.7 mmol). The reaction mixture was allowed to warm to room temperature (22 °C). After 16 h, hexanes (5 mL) and H₂O (5 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 5 mL). The combined organic layers were washed with H₂O (5 mL) and brine (10 mL), dried over Na₂SO₄, and concentrated in vacuo. Purification by flash chromatography (hexanes) provided protected alcohol (*R*)-23 as a colorless oil (0.72 g, 60%): $[a]_{D}^{18}$ -48.89 (*c* 1.02, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 7.30–7.29 (m, 4H), 7.23–7.21 (m, 1H), 5.81–5.74 (m, 1H), 5.02–4.99 (m, 2H), 4.67 (dd, J = 7.3, 5.2, 1H), 2.48–2.43 (m, 1H), 2.40–2.36 (m, 1H), 0.88 (s, 9H), 0.02 (s, 3H), – 0.13 (s. 3H): ¹³C NMR (150 MHz, CDCl₃) & 145.3, 135.5, 128.2, 127.1, 126.1, 117.0, 75.2, 45.7, 26.0, 18.4, -4.5, -4.7; IR (ATR) 1641, 1085, 832 cm⁻¹; HRMS (TOF MS ES+) *m* / z calcd for C₁₆H₂₇OSi (M+H)⁺ 263.1826, found 236.1814. Anal. Calcd for C₁₆H₂₆OSi: C, 73.22; H, 9.98. Found: C, 73.38; H, 9.97.

Alkene 28. To a cooled (-15 °C) solution of methyltriphentylphosphonium bromide (5.1 g, 14 mmol) in THF (150 mL) was added *n*-BuLi (6.0 mL, 2.5 M in hexanes, 15 mmol) dropwise over 5 min. After 15 min, the reaction mixture was cooled to -78 °C and aldehyde S3 (1.7 g, 7.4 mmol) was added as a solution in THF (50 mL) by cannula. After 15 min, the reaction

mixture was brought to room temperature (22 °C). After 1 h, saturated aqueous NH₄Cl (75 mL) was added and the aqueous layer was extracted with Et₂O (3 x 45 mL). The combined organic layers were washed with brine (50 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) provided alkene **28** as a colorless oil (0.83 g, 48%): [a]²⁰_D+14.12 (*c* 0.46, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ 5.87 (ddd, *J* = 17.0, 10.4, 5.5, 1H), 5.17 (dt, *J* = 17.2, 1.4, 1H), 4.98 (dt, *J* = 10.4, 1.3, 1H), 4.40–4.36 (m, 1H), 1.24 (d, *J* = 6.4, 3H), 1.07 (s, 10H), 1.06 (s, 11H); ¹³C NMR (150 MHz, CDCl₃) δ 143.5, 112.5, 69.8, 24.9, 18.3, 12.5; IR (ATR) 2943, 1464, 1091, 848 cm⁻¹; HRMS (TOF MS ES+) *m* / z calcd for C₁₃H₂₉OSi (M+H)⁺ 229.1988, found 229.1961. Anal. Calcd for C₁₃H₂₈OSi: C, 80.71; H, 11.61. Found: C, 80.95; H, 11.75.

Me ÖH	$OMe \xrightarrow{i-Pr_3SiCl, imidazole} Me \xrightarrow{O}_{\underline{i}} OMe \xrightarrow{O}_{\underline{i}} OMe$	DIBAL-H −78 °C, Et ₂ O OTIPS	n-BuLi, PPh₃Mel Me 0 to −78 °C, THF OTIPS
S15	S16	ent-S3	ent-28

(*R*)-Methyl 2-(Triisopropylsiloxy)propanoate S16. To a cooled (0 °C) solution of methyl-D-lactate S15 (0.50 mL, 5.2 mmol) in CH₂Cl₂ (13 mL) was added imidazole (0.54 g, 7.9 mmol). Triisopropylchlorosilane (1.3 mL, 6.1 mmol) was added dropwise over 5 min. After 16 h, hexanes (15 mL) was added, and the reaction mixture was washed with H₂O (3 x 15 mL) and brine (20 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Ester S16 was obtained as a colorless oil without any further purification (1.2 g, 86%). The spectroscopic data are consistent with the data reported:^{5a 1}H NMR (500 MHz, CDCl₃) δ 4.44 (q, *J* = 6.7, 1H), 3.72 (s, 3H), 1.43 (d, *J* = 6.7, 3H), 1.07–1.05 (m, 21H); ¹³C NMR (125 MHz, CDCl₃) δ 174.9, 68.7, 51.9, 22.0, 18.0, 12.3.

^{Me} Aldehyde *ent*-S3. To a cooled (-78 °C) solution of ester S16 (1.2 g, 4.5 mmol) in in Et₂O (45 mL) was added DIBAL-H (5.4 mL, 1.0 M in hexanes, 5.4 mmol) dropwise over 5 min. After 1 h, 2.0 M aqueous Rochelle Salt (10 mL) and H₂O (30 mL) were added and the aqueous layer was extracted with Et₂O (3 x 15 mL). The combined organic layers were washed with brine (20 mL), dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) provided aldehyde *ent*-S3 as a colorless oil (0.55 g, 53%). The spectroscopic data are consistent with the data reported for the enantiomeric compound:⁹ ¹H NMR (400 MHz, CDCl₃) δ 9.66 (d, *J* = 1.6, 1H), 4.18 (qd, *J* = 6.8, 1.6, 1H), 1.31 (d, *J* = 6.8, 3H), 1.08–1.06 (m, 21H); HRMS (TOF MS ES+) *m* / z calcd for C₁₂H₂₇O₂Si (M+H)⁺ 231.1775, found 231.1781.

Me Alkene *ent-28.* To a solution of methyltriphenylphosphonium iodide (0.49 g, 1.3 of the form of the ent-set model in THF (3 mL) was added *t*-BuOK (0.16 g, 1.4 mmol). After 1 h, aldehyde *ent-S3* (0.29 g, 1.3 mmol) was added as a solution in THF (1.4 mL). After 10 min, hexanes (10 mL) and saturated aqueous NH₄Cl (10 mL) were added, the layers were separated, and the aqueous layer was extracted with hexanes (3 x 15 mL). The combined organic layers were washed with H₂O (10 mL) and brine (15 mL), dried over Na₂SO₄, and concentrated *in vacuo*. Purification by flash chromatography (hexanes) provided alkene *ent-28* as a colorless oil (0.212 g, 74%): [a]²⁰_D-14.37 (*c* 0.53, CHCl₃); ¹H NMR (600 MHz, CDCl₃) 5.87 (ddd, *J* = 17.1, 10.4, 5.5, 1H), 5.17 (dt, *J* = 17.2, 1.5, 1H), 4.98 (dt, *J* = 10.4, 1.4, 1H), 4.40–4.36 (m, 1H), 1.24 (d, *J* = 6.3, 3H), 1.07 (s, 10H), 1.06 (s, 11H); ¹³C NMR (125 MHz, CDCl₃) δ 143.5, 112.5, 69.8, 24.9, 18.3, 12.5; HRMS (TOF MS ES+) *m* / *z* calcd for C₁₃H₂₉OSi (M+H)⁺ 229.1982, found 229.1972.

III. Reaction Condition and Ligand Screening





Me OTIPS 4	+ Si	h^{t} -Bu AgO ₂ CCF ₃ (3 mol %) h^{t} -Bu 22 °C, C ₆ D ₆ :solvent	Me Si-t- OTIPS 5a a	Bu + Me C nd 5b	t-Bu Si OTIPS
	Entry	Solvent	Conversion ^a	d.r. ^b	
	1	Trifluorotoluene	95	42:58	
	2	1,2-Dichloroethane	100	48:52	
	3	Chloroform	50	47:53	
	4	Tetrahydrofuran	0	-	
	5	Diethyl Ether	78	50:50	
	6	Toluene	78	47:53	

Table S1. Reaction Condition Screening for Alkene 4

^{*a*}Determined by ¹H NMR spectroscopy. ^{*b*}Determined by ¹H and ¹³C NMR spectroscopy.



Me OTIPS 4	AgOTf (3 (S)-BINA	t-Bu t-Bu mol %) AP OTIP	t-Bu Si - t-Bu + Me S So 5a and 5b	t-Bu ,Si∽t-Bu
Entry	Ag:BINAP	Complex ¹⁰	Conversion (%) ^{<i>a</i>}	d.r. ^b
1	1:2	Α	46	52:48
2	1:1	A:B:C (21:63:16) ^c	68	50:50
3	2:1	С	67	53:47

^{*a*}Determined by ¹H NMR spectroscopy. ^{*b*}Determined by ¹H and ¹³C NMR spectroscopy.

^{*c*}Ratio of complexes in solution.





Me OTIPS 6	AgO ₂ CCF ₃ (3 mo Ligand (6 mol % Temp, tol-d ₈	u t-Bu Si ∽t-Bu Si ∽t-Bu Si ∽t-Bu OTIPS 7a an	+ Me OTIPS	t-Bu ,Si∽t-Bu
Entry	Ligand	Temperature (°C)	Time (h)	d.r. ^{<i>a</i>}
1	(S)-BINAP (8)	22	2	52:48
2	(S)-BINAP (8)	-20	7.5	54:46
3	(S)-BINAP (8)	-20	22	53:47
4	(<i>R</i>)-BINAP (9)	-20	22	51:49
5	(<i>R</i>)-BINAP (9)	-78	7.5	53:47

 Table S3.
 Ligand and Temperature Screen for Alkene 6

^{*a*}Determined by ¹H NMR spectroscopy.

Table S4. Ligand Screening for Alkene 28 (Silver-Catalyzed)

$\begin{array}{c} \text{Me} \\ & \overbrace{\text{OTIPS}\\ 28} \\ \end{array} \begin{array}{c} \overbrace{\text{AgO}_2\text{CCF}_3(5 \text{ mol }\%)}^{t-\text{Bu}} \\ 2 \\ \hline \text{AgO}_2\text{CCF}_3(5 \text{ mol }\%) \\ 22 \ ^\circ\text{C}, \ \text{C}_6\text{D}_6 \end{array}$	Me OTIPS OTIPS O	t-Bu ,Si∽t-Bu ∕√
Ligand (10 mol %)	Conversion (%) ^{<i>a</i>}	d.r. ^{<i>a,b</i>}
(<i>R</i>)-BINAP (9)	20	53:47
(<i>R</i>)-QUINAP (S17) ^{<i>c</i>}	67	55:45
(<i>S</i> , <i>S</i>)-NORPHOS (30)	0	-
(<i>S</i> , <i>S</i>)-DIOP (31)	0	-
(R,R)-pyrrolidine (S19)	61	54:46
(<i>S</i> , <i>S</i>)-salen (27)	79	54:46
(<i>S</i> , <i>S</i>)- <i>t</i> -Bu-BOX (21)	82	52:48

^{*a*}Determined by ¹H NMR spectroscopy. ^{*b*}Determined using ¹³C NMR spectroscopy. ^{*c*}Heated to 80 °C for 48 h.

Me	TIPS 22 °C, C ₆ D ₆ Me √	t-Bu Si - t-Bu Si - t-Bu H Me OTIPS OTIPS 29a and 29b	_i ∽ <i>t-</i> Bu
Entry	Ligand (6 mol %)	Conversion (%) ^{<i>a</i>}	d.r. ^{<i>a,b</i>}
1	(<i>R</i>)-BINAP (9)	18	50:50
2	(<i>R</i>)-QUINAP (S17)	67	55:45
3	(<i>S</i> , <i>S</i>)- <i>i</i> -Pr-PyBOX (20)	0	-
4	(<i>S</i> , <i>S</i>)- <i>i</i> -Pr-PyOX (22)	84	60:40
5	(<i>S</i> , <i>S</i>)- <i>t</i> -Bu-BOX (21)	30	56:44
6	(<i>S</i> , <i>S</i>)-salen (27)	85	60:40
7	(<i>R</i> , <i>R</i>)-pyrrolidine (S19)	34	59:41

Table S5. Ligand Screening for Alkene 28 (Copper-Catalyzed)

^{*a*}Determined by ¹H NMR spectroscopy. ^{*b*}Determined using ¹³C NMR spectroscopy.

Table So. Ligand and Catalyst Screen 101 ent-2	e S6. Ligand and Catalyst Screen	tor <i>ent-28</i>
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C	Me OTIPS MX, liga	t-Bu t-Bu t-Bu Me Si - t-Bu nd OTIPS +	t-Bu Si∽t-Bu Me U	
	<i>ent-28</i> 22 °C, C	₆ D ₆ ent-29a and e t	nt-29b	
Entry	MX	Ligand	Yield (%) ^{<i>a</i>}	d.r. ^{<i>a</i>}
1	AgO ₂ CCF ₃	None	77	54:46
2	AgO ₂ CCF ₃	(S)-BINAP (8)	61	51:49
3	AgO ₂ CCF ₃	(<i>R</i>)-QUINAP (S17)	53	53:47
4	[Cu(OTf) ₂]•PhH	None	68	60:40
5	[Cu(OTf) ₂]•PhH	(<i>S</i> , <i>S</i>)- <i>t</i> -Bu-BOX (21)	21	52:48
6	[Cu(OTf) ₂]•PhH	(<i>R</i>)-QUINAP (S17)	4	50:50
a _D				

^{*a*}Determined by ¹H NMR spectroscopy.

IV. Details of Computation

Major isomer, 29b and 29b' (anti). A conformational search (Monte Carlo) for the tri(isopropyl)silyl ether **29b** using molecular mechanics (MMFF force field) as implemented in Spartan16¹¹ produced >300 conformers, mostly with similar energies (note that the enantiomeric structure was calculated, which would not affect the results). Minimizing just the lowest energy conformer with DFT (B3LYP/6-31G*) was prohibitively slow, so it was not practical to calculate several hundred conformations at the higher level. The trialkylsilyl ether moiety was changed to a trimethylsilyl group, considering that these two different types of silyloxy groups should be conformationally similar.¹² With the trimethylsilyl ether (compound **29b'**), 20 low-energy structures were identified (MMFF). Those structures were optimized using the PM3 model in Spartan16. The geometries were then optimized with Gaussian09¹³ using density functional theory (DT/6-31G*) to optimize geometry (keywords: opt b3lyp/6-31g(d) geom=connectivity).

Of the 20 structures, the lowest two were nearly identical in structure and energy; the structure of the lowest energy structure (the lower of these two) is listed below (E = -1246.34122902 au). Frequency calculations were performed on the lowest energy conformer (keywords: freq b3lyp/6-31g(d) geom=connectivity), which confirmed that it had no imaginary frequencies. Three more structures were found only 0.2 kcal/mol higher, and another three were 1.6 kcal/mol higher in energy. The only difference among these structures were differences in rotation of the trimethylsilyl group.

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Table S7. Cartesian coordinates for Major isomer, 29b'.

Atom	X (Å)	Y (Å)	Z (Å)
Si	-1.669300	-0.100800	-0.215300
С	-0.239400	-0.640100	-1.292900
С	0.044300	-0.745000	0.236000
С	1.250300	-0.038500	0.826700
0	2.430700	-0.764700	0.439200
Si	3.831100	-0.143800	-0.230400
С	-3.110300	-1.377700	-0.007400
С	-4.026100	-1.018000	1.182800
С	-2.519800	-2.781800	0.245700
С	-3.955000	-1.440600	-1.301500
С	-2.103100	1.785600	-0.121100
С	-2.324800	2.222500	1.344800
С	-3.382800	2.088500	-0.933100
С	-0.957200	2.629500	-0.719000
С	4.656500	1.090800	0.943000
С	3.483200	0.720200	-1.877100
С	4.943900	-1.637200	-0.498800
Н	0.335000	0.132200	-1.806300
Н	-0.241600	-1.573800	-1.855300
Н	0.028700	-1.780600	0.583100
Н	-3.469300	-0.964600	2.126200
Н	-4.538800	-0.060000	1.040900
Н	-4.805000	-1.785000	1.304600
Н	-1.969000	-2.833400	1.191100
Н	-1.843700	-3.098400	-0.556900
Н	-3.333200	-3.519900	0.296600
Н	-3.345700	-1.725100	-2.167000
Н	-4.750700	-2.192700	-1.194800
Н	-4.439700	-0.486500	-1.533700
Н	-1.424100	2.078100	1.951700
Н	-3.141700	1.672700	1.824500
Н	-2.582000	3.291300	1.385100
Н	-3.593500	3.167500	-0.905500

Н	-3.278500	1.801400	-1.986100
Н	-4.263800	1.577900	-0.530200
Н	-0.018700	2.507300	-0.168000
Н	-0.767400	2.379400	-1.768600
Н	-1.222200	3.695800	-0.674400
Н	4.918800	0.617500	1.896600
Н	4.003600	1.944000	1.165000
Н	5.579500	1.492600	0.505900
Н	2.813400	1.581100	-1.761000
Н	3.019600	0.033500	-2.594800
Н	4.414000	1.092300	-2.324100
Н	5.145300	-2.152900	0.447200
Н	5.907700	-1.344700	-0.933200
Н	4.475300	-2.359300	-1.177100
С	1.204300	0.021600	2.354500
Н	2.104000	0.504200	2.750600
Н	1.154400	-0.992700	2.766100
Н	0.326900	0.578500	2.700900
Н	1.307800	0.986000	0.430100

Minor isomer, 29a' (syn). A conformational search of the trimethylsilyl ether was performed as discussed above, which produced 15 low-energy structures. These structures were optimized using the PM3 model in Spartan16. This minimization produced eight unique structures, which were then optimized using Gaussian09 using density functional theory (DFT/6-31G*) as discussed above. Of these structures, the lowest three were essentially of the same energy and structure (<0.0003 kcal/mol). The lowest energy of these conformers is shown below (E = – 1246.34012163 au). Frequency calculations, as discussed above, confirmed that it had no imaginary frequencies. Three other structures were only 0.05 kcal/mol higher in energy; they exhibited a \angle Si–C–C–O dihedral angle of –38°. By comparison, that angle was 78° in the lowest energy conformers. The remaining two structures were considerably higher energy (>3.5 kcal/mol). The assignment of this conformer results from the energy differences between the two isomeric forms of **29a'** compared to **29b'**. All of the conformers of **29a'** are >0.7 kcal/mol higher than for the lowest energy conformers of **29b'**. Although this number is not large, it does suggest which isomer is thermodynamically preferred. The energy difference is likely to be larger in the triisopropyl silyl ether series. In isomer **29a'**, the groups on the two silicon atoms occupy similar regions of space, which is not the case for **29b'**, so making the groups larger on the silyl ether should increase that interaction and destabilize the minor isomer.



Table S8. Cartesian coordinates for Minor isomer, 29a'.

Atom	X (Å)	Y (Å)	Z (Å)
Si	-1.69260	-0.14800	-0.35860
С	-1.24050	-1.14140	-1.88450
С	-0.20850	-1.24170	-0.71830
С	1.23940	-0.86130	-0.98490
0	1.86500	-0.54600	0.26430
Si	3.37310	0.12470	0.54780
С	1.98020	-1.99560	-1.70490
С	-3.03610	-0.93210	0.79180
С	-3.00820	-0.30460	2.20300
С	-2.77150	-2.44660	0.93450
С	-4.44170	-0.74640	0.17520
С	-1.54620	1.78050	-0.43060
С	-0.75310	2.28610	0.79650
С	-2.93810	2.45120	-0.44550
С	-0.80520	2.21270	-1.71380
С	3.23210	0.96180	2.22940
С	3.80420	1.39250	-0.78910
С	4.71930	-1.20000	0.63270
Н	-0.88570	-0.59440	-2.75970
Н	-1.75520	-2.05880	-2.17080
Н	-0.25530	-2.20350	-0.20270

Н	1.26830	0.02750	-1.63210
Н	3.00160	-1.70590	-1.97770
Н	1.45050	-2.26990	-2.62400
Н	2.03440	-2.87950	-1.05920
Н	-2.03130	-0.42870	2.68540
Н	-3.24210	0.76570	2.18970
Н	-3.75700	-0.79050	2.84570
Н	-1.81670	-2.65230	1.43020
Н	-2.76890	-2.96060	-0.03390
Н	-3.56510	-2.90360	1.54320
Н	-4.51170	-1.21180	-0.81480
Н	-5.20010	-1.21830	0.81750
Н	-4.71660	0.30790	0.06860
Н	0.24000	1.82770	0.84740
Н	-1.26780	2.06650	1.73880
Н	-0.62470	3.37760	0.74150
Н	-2.82240	3.54280	-0.51560
Н	-3.54090	2.13050	-1.30360
Н	-3.51190	2.24750	0.46490
Н	-1.32490	1.88010	-2.61910
Н	-0.74210	3.31000	-1.75100
Н	0.21970	1.82960	-1.75270
Н	2.91480	0.24550	2.99610
Н	2.49610	1.77380	2.21090
Н	4.19280	1.38610	2.54660
Н	3.90220	0.93550	-1.78120
Н	4.75960	1.88130	-0.56060
Н	3.04130	2.17730	-0.85760
Н	5.68100	-0.75800	0.92370
Н	4.86750	-1.71120	-0.32510
Н	4.46700	-1.96240	1.37940

V. References and Notes

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VI. Selected Spectra































