Supplementary Information.

Enantiopure and racemic radical-cation salts of bis(2'hydroxylpropylthio)(ethylenedithio)TTF with polyiodide anions

Lee Martin,*^a John D.Wallis,^a Milena Guziak,^a Peter Maksymiw,^a Florence Konalian-Kempf,^a Anthony Christian,^a Shin'ichi Nakatsuji,^b Jun'ichi Yamada^b and Hiroki Akutsu^{b,c}

^a School of Science and Technology, Nottingham Trent University, Clifton Lane, Nottingham, NG11 8NS, United Kingdom; E-mail: lee.martin@ntu.ac.uk

^b Graduate School of Material Science, University of Hyogo, Kamigori-cho, Ako-gun, Hyogo, 678-1297, Japan

^c Department of Chemistry, Graduate School of Science, Osaka University, 1-1 Machikaneyama-cho, Toyonaka, Osaka 560-0043, Japan

Table S1. Torsion angles for the donor cation side chains in RR-1/°, e.s.d.s in range 0.8-1.2°.



		Torsion α	Torsion β	Torsion γ
DONOR 1,	01	-45.0	-78.2	-173.8
	02	63.3	63.3	57.3
DONOR 2	03	39.5	55.6	55.4
	04	-72.3	-72.7	-165.8
DONOR 3	05	-33.9	-90.5	81.6
	06	119.9	-155.7	-51.8
DONOR 4	07	57.6	74.0	70.2
	08	-23.8	-178.8	66.0
DONOR 5	09	-15.1	177.0	76.6
	010	55.2	75.4	66.4
DONOR 6	011	34.9	69.0	70.9
	012	-98.3	133.8	173.5

Table S2. Torsion angles for the donor cation side chains in S,S-2 / °, e.s.d.s 1-2°.

	Torsion α	Torsion β	Torsion γ
DONOR 1, 01	40	64	172
02	-67	-81	-171
DONOR 2 O3	-57	-113	54
04	61	153	-168

	Torsion α	Torsion β	Torsion γ
DONOR 1, 01	58	57	-173
02	-9	169	174
DONOR 2 O3	-97	69	175
04	7	168	174

Table S3. Torsion angles for the donor cation side chains in SS-3 donor. / °, e.s.d.s 2-3°.

Table S4. Torsion angles for the donor cation side chains in *rac*-4. / °, e.s.d.s 0.5-0.8°, main conformation only.

	Torsion α	Torsion β	Torsion γ
DONOR 1, 01	109.3	-82.4	-68.0
02	12.6	-103.0	178.5

Table S5. Relative geometry of O atoms involved in hydrogen bonding for *R*,*R*-1, *S*,S-2, *S*,*S*-3, *rac*-4.

R,R- 1	O(8)-H(8)···· O(3)-H(3)(x-1, y, z)··· O(6)- H(6)(x-1, y-1, z)··· O(9)-H(9)(x, y-1, z)	O(8)···O(3): 2.594(16) O(3)···O(6): 2.781(10)
		O(6)…O(9): 2.742(18)
		O(8)…O(3)…O(6): 98.6(5)°
		O(3)…O(6)…O(9): 125.4(3)°
	O(11)-H(11)···· O(2)-H(2)(x-1, y, z+1) ···	O(11)…O(2): 2.940(13)
	O(5)-H(5)(x, y, z+1)… O(4)-H(4)(x-1, y+1,	O(2)···O(5): 2.832(16)
	z+1)	O(5)…O(4): 2.914(13)
		O(11)…O(2)…O(5): 104.1(4)°
		O(2)…O(5)…O(4): 106.9(4)°
	O(10)-H(10)···· O(7)-H(7)(x, y + 1, z)····	O(10)…O(7): 2.777(13)
	O(12)-H(12)(x + 1, y + 1, z)	O(7)…O(12): 2.703(15)
		O(10)…O(7)…O(12): 117.1(4)°
<i>S,S</i> - 2	O(3)-H(3)···O(4)-H(4)(x+1, y, z)···	O(3)…O(4): 2.890(14)
	O(1)-H(1)(x+1, y+1, z-1)…O(2)-H(2)(x+1,	O(4)…O(1): 2.652(14)
	y+1, z-1)	O(1)…O(2): 2.619(14)
		O(3)…O(4)…O(1): 94.2(5)°
		O(4)…O(1)…O(2): 114.1(5)°
S,S- 3	O(3)-H(3)···O(2)(x, y, z + 1)	O(3)…O(2): 3.01(3)
	O(4)-H(4)···O(3) (x, y-1, z)	O(4)…O(3): 2.85(3)
		O(2)···O(3)···O(4): 121.3(9)
rac- 4	O(1)-H(1)···O(2) (-x-1, -y, -z)	O(1)…O(2): 2.751(10)

Fig S1a Hydrogen bonding in *R***,** *R***-1**





Fig S1c Hydrogen bonding in *R***,** *R***-1**



Fig S1d Hydrogen bonding in *R***,** *R***-1**







Fig S3a. Asymmetric unit of *S***,** *S***-1**.



Fig S3b. Asymmetric unit of *S,S-2*.



Fig S3c. Asymmetric unit of *S*, *S*-**3**; two parts of two different octaiodide dianions are shown in the asymmetric unit.



Fig S3d. Asymmetric unit of *rac-4*; the octaiodide dianion sits on a centre of symmetry so only half is crystallographically unique.

