

Supporting Information

Redox State Manipulation of a Tris(*p*-tetrazolylphenyl)amine Ligand and its Mn²⁺ Coordination Frameworks

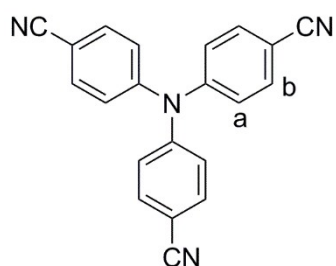
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Tris(*p*-cyanophenyl)amine.



Di(*p*-cyanophenyl) amine (0.500 g, 2.28 mmol) and *p*-fluorobenzonitrile (0.304 g, 2.51 mmol) were dissolved in dry and distilled DMSO (5.0 mL). Potassium carbonate (0.946 g, 6.84 mmol) was added to yield a yellow-red solution that was heated at 140 °C under nitrogen for 4 days. The reaction mixture was cooled to room temperature and water added to induce precipitation. The solid was filtered and washed with HCl (1 M) and water several times. The precipitate was dried under vacuum to yield the product as a pale yellow solid (0.600 g, 82%). ¹H NMR (CDCl₃, 200 MHz): 7.61 (d, ³*J*_{Ha-Hb} = 8.4 Hz, 6H, **H_a**), 7.16 (d, ³*J*_{Ha-Hb} = 8.4 Hz, 6H, **H_b**) ppm.

Crystallographic Tables

Table S1. Crystal data for **1**.

Parameter	
Formula	C ₄₈ H ₄₈ Mn ₃ N ₂₆ O ₆
M/g mol ⁻¹	1177.71
Temperature (K)	150(2)
Crystal system	trigonal
Crystal size (mm ³)	0.120 × 0.085 × 0.050
Crystal colour	Colourless
Crystal Habit	Block
a (Å)	16.5089(10)
b (Å)	16.5089(10)
c (Å)	45.518(6)
γ (°)	120
V (Å ³)	10743.6(18)
Z	6
ρ _{calc} (mg mm ⁻³)	1.092
λ(MoKα)	0.71073 Å
μ(MoKα)	0.574 mm ⁻¹
2θ _{max} (°)	65.28
T(CRYSALISPRO) _{min, max}	0.94508, 1.00000
hkl range	-24 24, -24 24, -67 68
Reflections collected	83300/4331 [R(int) = 0.0680]
Data/restraints/parameters	2918/126
Final R indexes [all data]	R ₁ = 0.0623, wR ₂ = 0.1965
Goodness-of-fit on F ²	1.055
Residual Extrema	-0.663, 3.345 e ⁻ Å ⁻³

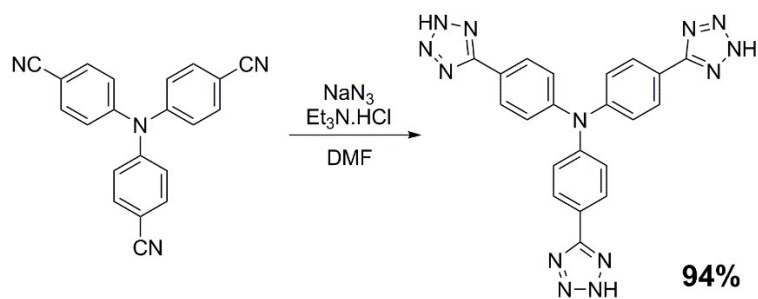
$$RI = \Sigma(|F_o| - |F_c|)/\Sigma(|F_o|); wR_2 = [\Sigma\{w(F_o^2 - F_c^2)^2\}/\Sigma\{w(F_o^2)^2\}]^{1/2}, wR2 = (\Sigma w(F_o^2 - F_c^2)^2/\Sigma(wF_c^2)^2)^{1/2} \quad \text{all reflections} \quad w=1/[\sigma^2(F_o^2)+(0.1308P)^2] \quad \text{where} \\ P=(F_o^2+2F_c^2)/3$$

Table S2. Crystal data for **2**.

Parameter	
Formula	C ₆₀ H ₆₆ Mn ₃ N ₃₂ O ₆
M/g mol ⁻¹	1496.26
Temperature (K)	150(2)
Crystal system	Trigonal
Crystal size (mm ³)	0.160 × 0.105 × 0.050
Crystal colour	Colourless
Crystal Habit	Block
a (Å)	16.4650(14)
b (Å)	16.4650(14)
c (Å)	47.040(4)
γ (°)	120
V (Å ³)	11044(2)
Z	6
ρ _{calc} (mg mm ⁻³)	1.350
λ(MoKα)	0.71073 Å
μ(MoKα)	0.577 mm ⁻¹
T(CRYSALISPRO) _{min, max}	0.93508, 1.00000
2θ _{max}	62.38
hkl range	-22 22, -23 23, -67 68
Reflections collected	62386/3890 [R(int) = 0.1143]
Data/restraints/parameters	2571/157
Final R indexes [all data]	R ₁ = 0.0597, wR ₂ = 0.2088
Goodness-of-fit on F ²	1.051
Residual Extrema	-0.779, 0.849 e ⁻ Å ⁻³

$$RI = \Sigma(|F_o| - |F_c|)/\Sigma(|F_o|); wR_2 = [\Sigma\{w(F_o^2 - F_c^2)^2\}/\Sigma\{w(F_o^2)^2\}]^{1/2}, wR2 = (\Sigma w(F_o^2 - F_c^2)^2/\Sigma(wF_c^2)^2)^{1/2} \quad \text{all reflections} \quad w=1/[\sigma^2(F_o^2)+(0.1388P)^2] \quad \text{where} \\ P=(F_o^2+2F_c^2)/3$$

Figures



Scheme S1. Synthesis of the tris(*p*-tetrazolylphenyl)amine (H₃TPA) ligand.

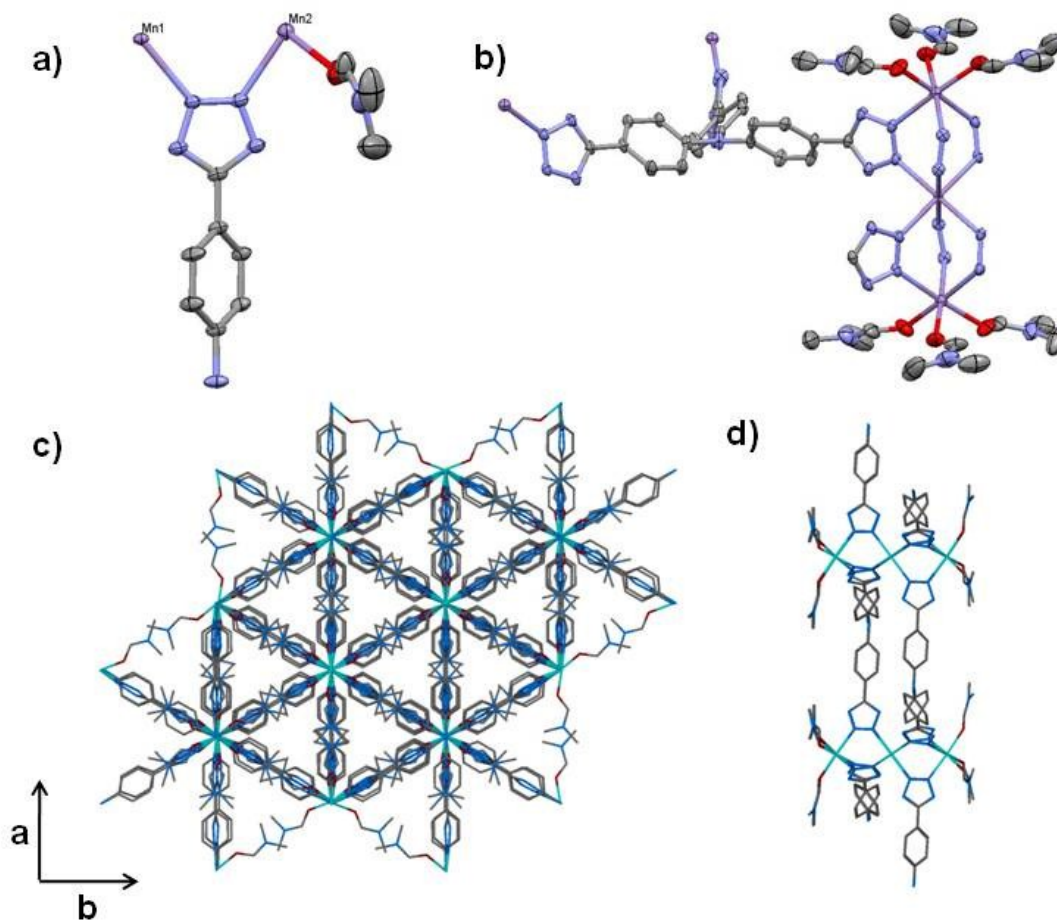


Figure S1. Solid state structure of **2** showing (a) the asymmetric unit, (b) the metal cluster formed upon coordination of three tetrazole rings to two Mn centres, (c) view down the *c* axis of the framework and (d) side on view of one of the networks.

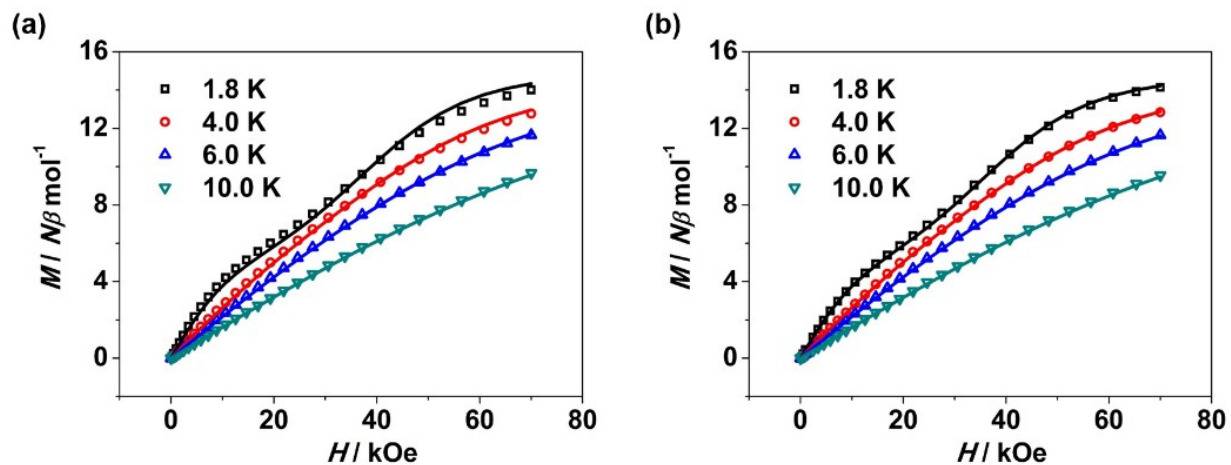


Figure S2. The field-dependent magnetizations of (a) **1** and (b) **2** from 0 to 70 kOe at indicated temperatures. The solid lines are fitted by the PHI program (together with $\chi_M T - T$).

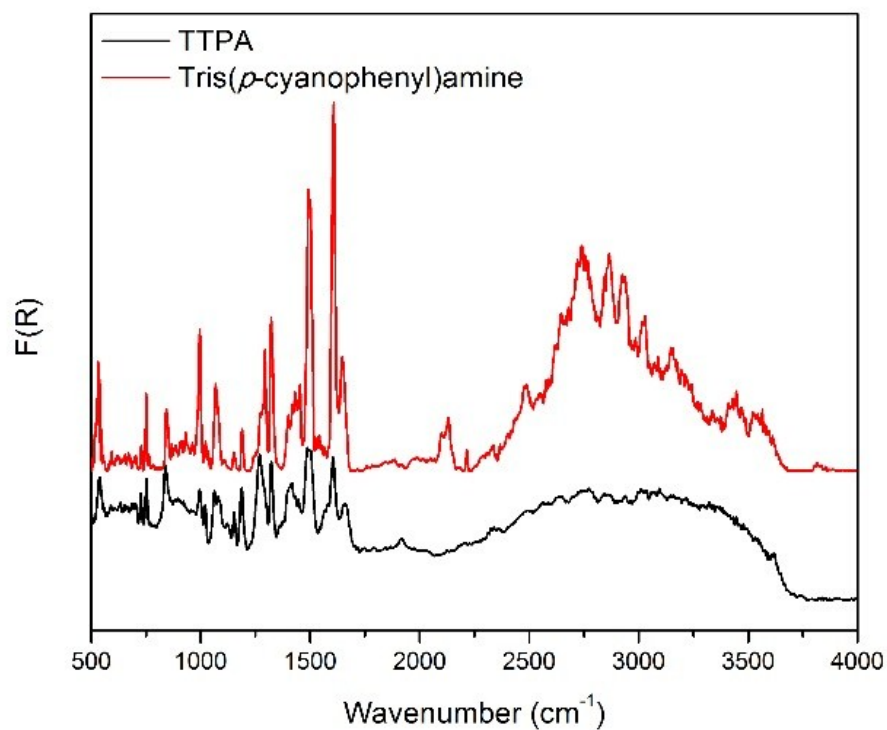


Figure S3. Infrared spectra of tris(*p*-tetrazolylphenyl)amine (H₃TTPA) vs. tris(*p*-cyanophenyl)amine over the range 4000-500 cm^{-1} in KBr matrix.

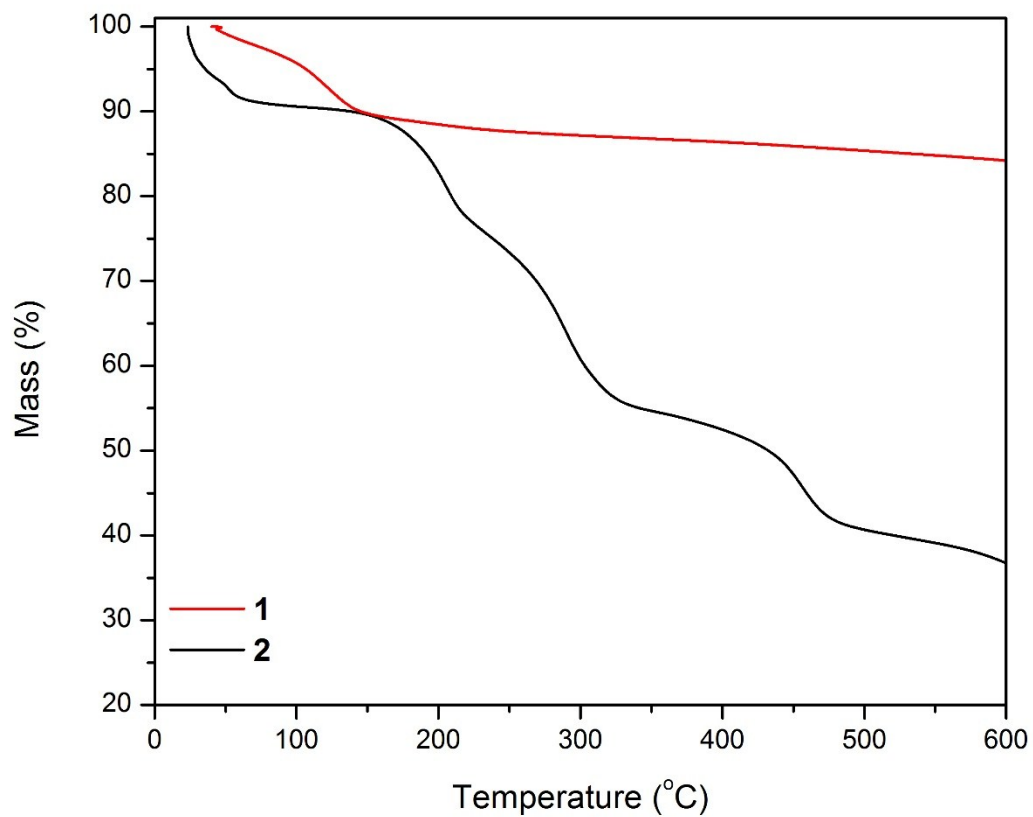


Figure S4. Thermal Gravimetric Analysis (TGA) of **1** and **2** over the range 50-600 °C.

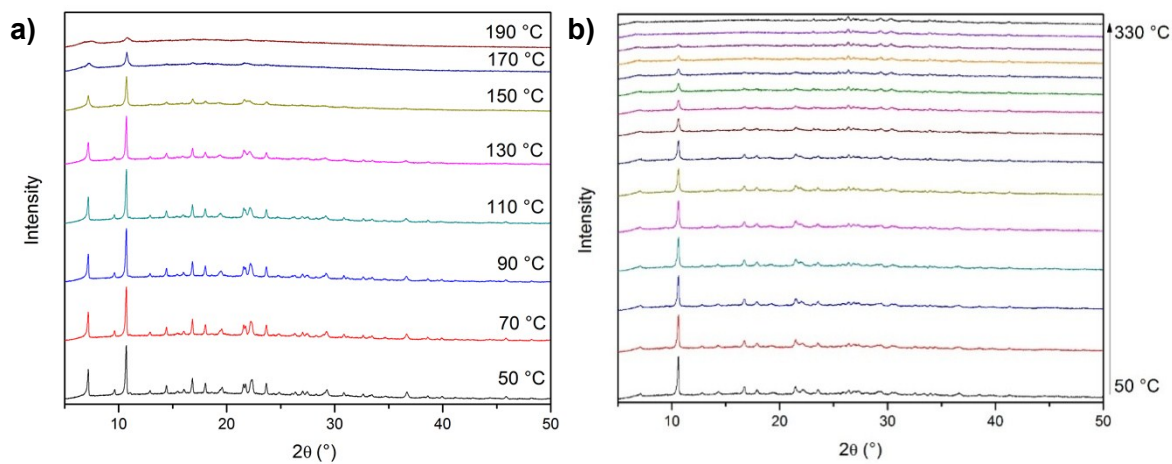


Figure S5. Variable temperature powder X-ray diffraction (VT-PXRD) for (a) **1** from 50-190 °C and (b) **2** from 50-330 °C in 20 °C increments.

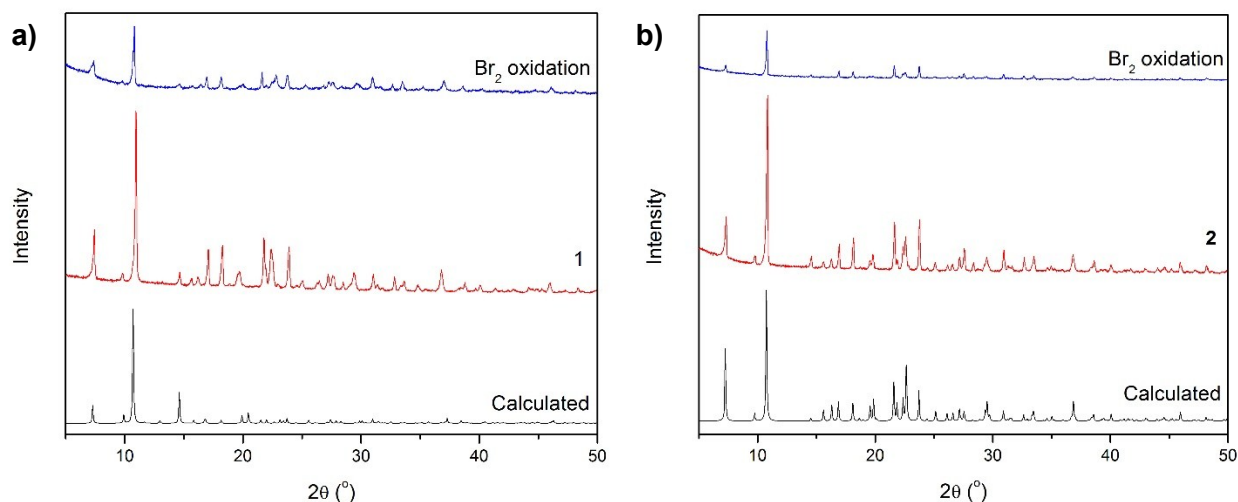


Figure S6. PXRD pattern for (a) **1**; predicted pattern, as synthesised framework and oxidised with bromine and (b) **2**; predicted pattern, as synthesised framework, oxidised with bromine over 5-50° 2 θ .

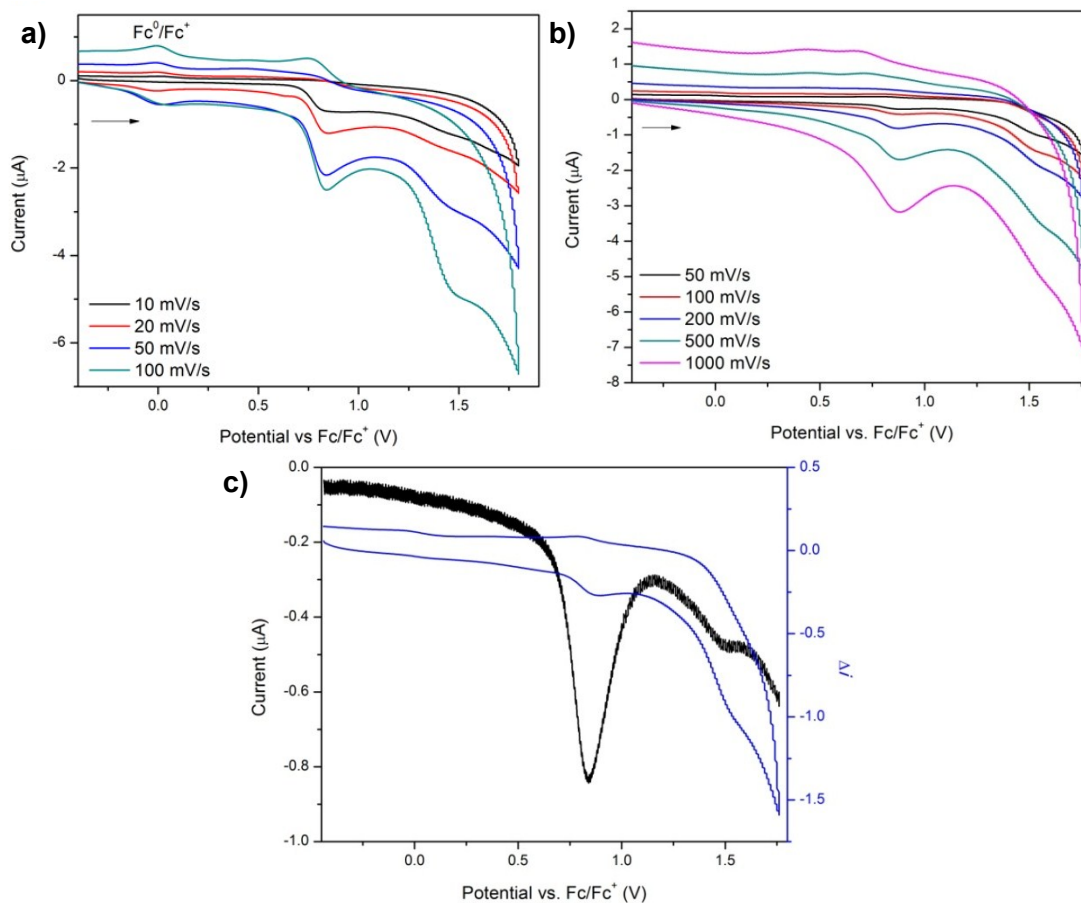


Figure S7. Solution state electrochemistry of the H₃TPPA ligand where cyclic voltammogram over the scan rates of (a) 10-100 mV s⁻¹ and (b) 50-1000 mV s⁻¹ in [(*n*-C₄H₉)₄N]PF₆/CH₃CN electrolyte and (c) square wave in [(*n*-C₄H₉)₄N]PF₆/CH₂Cl₂ electrolyte referenced against the Fc⁺/Fc couple. The arrow indicates the direction of the forward scan.

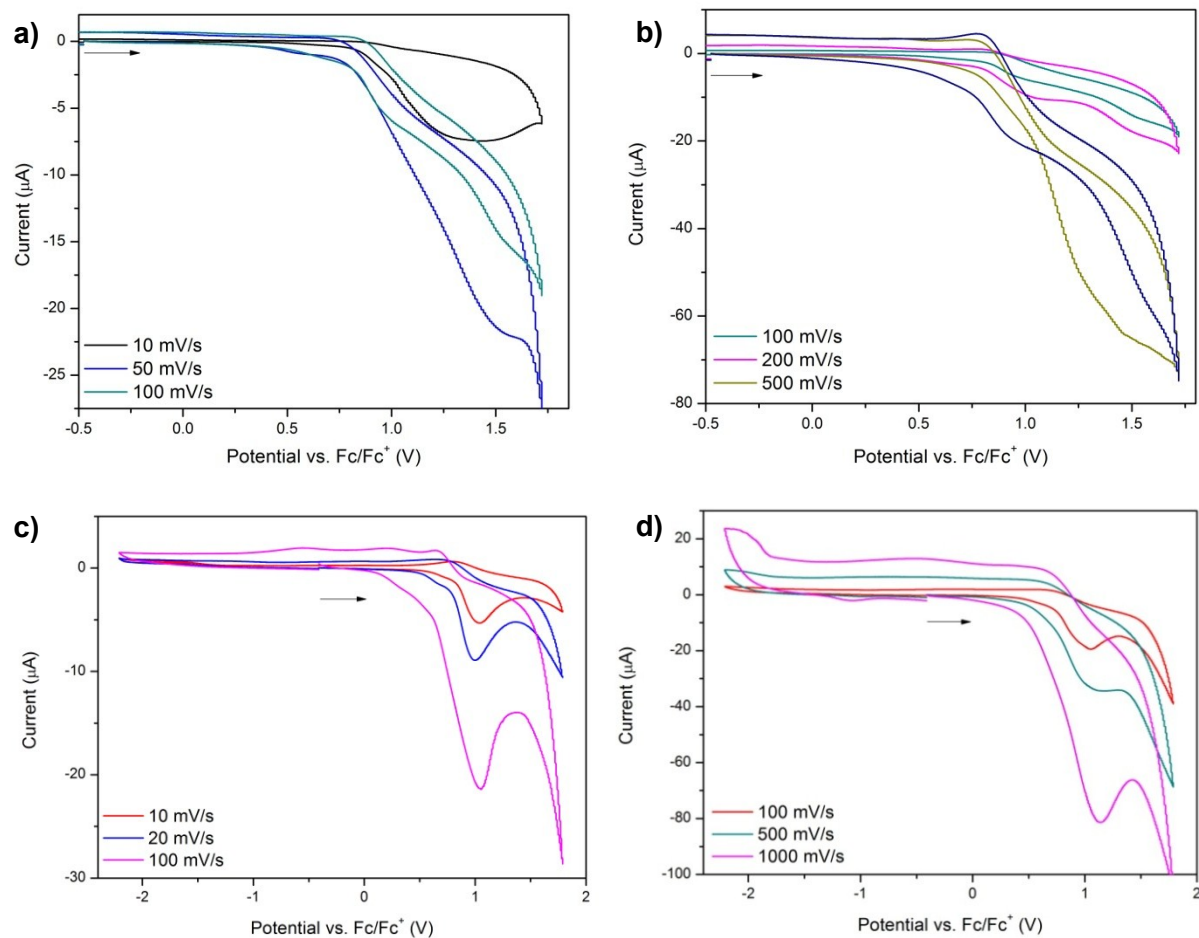


Figure S8. Solid state electrochemistry of **1** in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_3\text{CN}$ electrolyte over the scan rates of (a) 10-100 mV s^{-1} , (b) 100-500 mV/s and in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_2\text{Cl}_2$ electrolyte over the scan rates of (c) 10-100 mV s^{-1} and (d) 100-1000 mV s^{-1} .

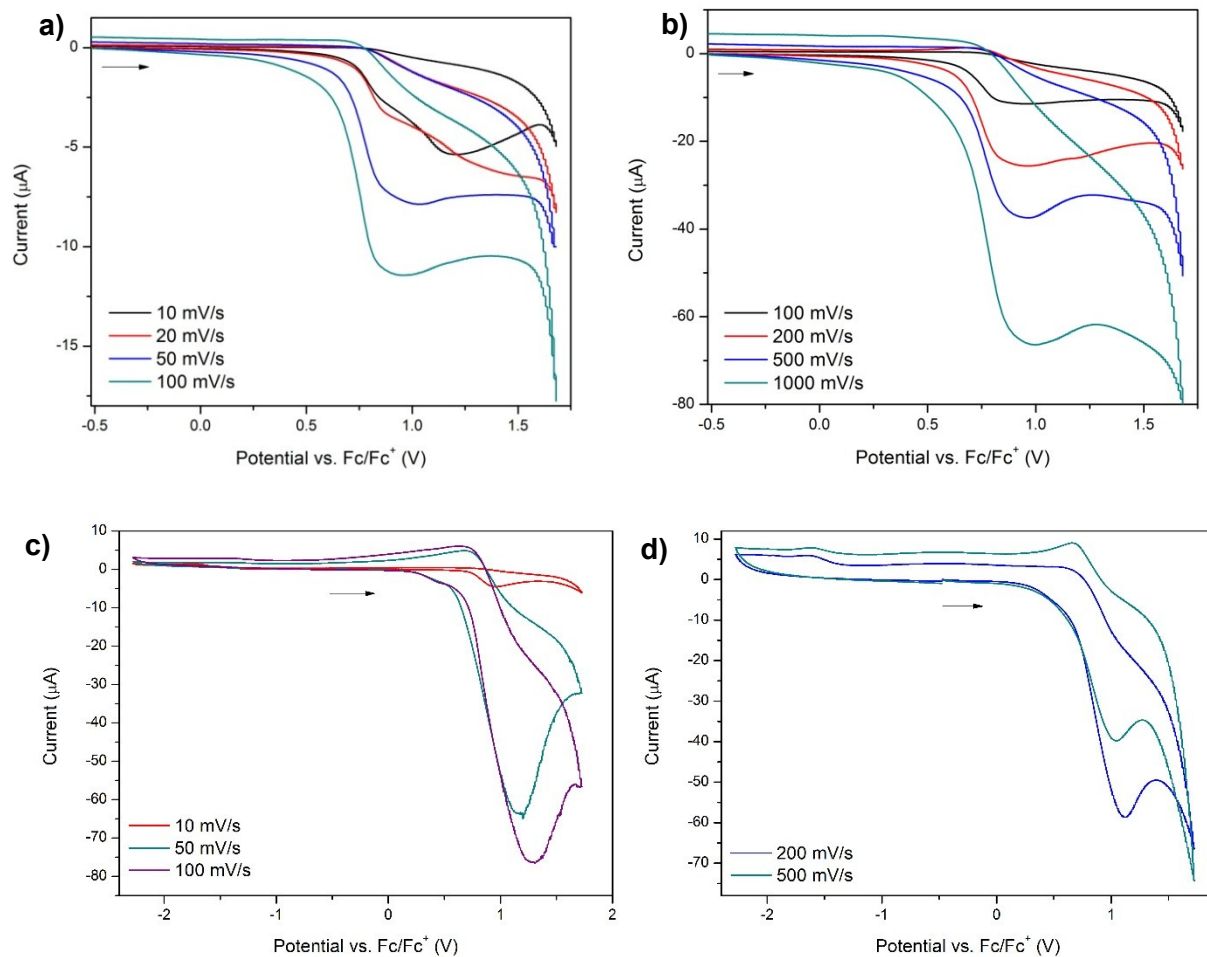


Figure S9. Solid state electrochemistry of **2** in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_3\text{CN}$ electrolyte over the scan rates of (a) 10-100 mV/s, (b) 100-1000 mV s^{-1} and in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_2\text{Cl}_2$ electrolyte over the scan rates of (c) 10-100 mV/s and (d) 100-1000 mV s^{-1} .

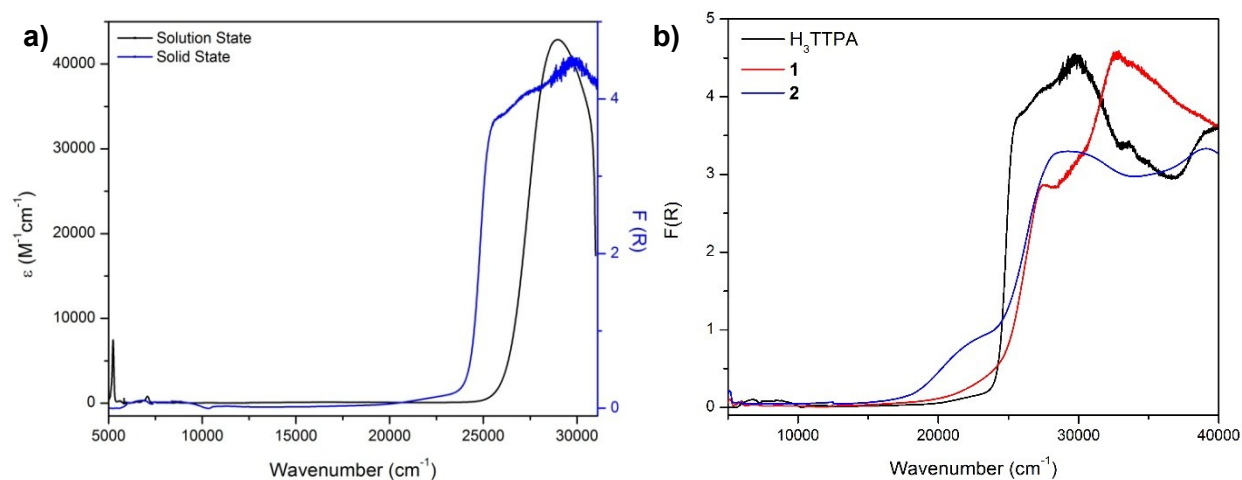


Figure S10. (a) Solution and solid state UV/Vis/NIR spectrum of the H_3TPPA ligand over the range 5000-35000 cm^{-1} and (b) UV/Vis/NIR spectra in the solid state of the H_3TPPA ligand, **1** and **2** from 5000-40000 cm^{-1} .

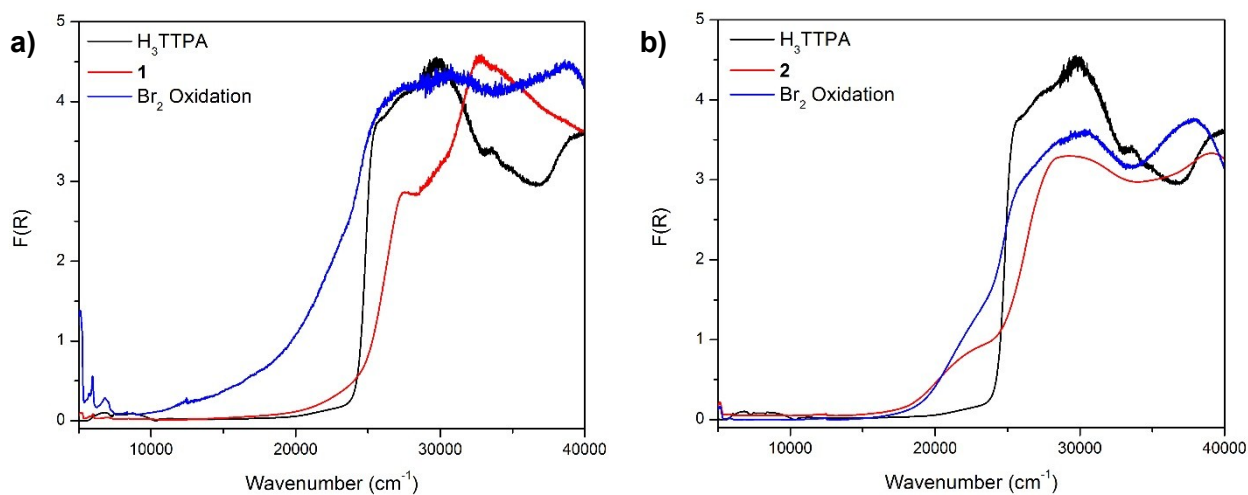


Figure S11. Solid state UV/Vis/NIR spectra of (a) **1** and (b) **2** and the spectra obtained by oxidation using bromine.

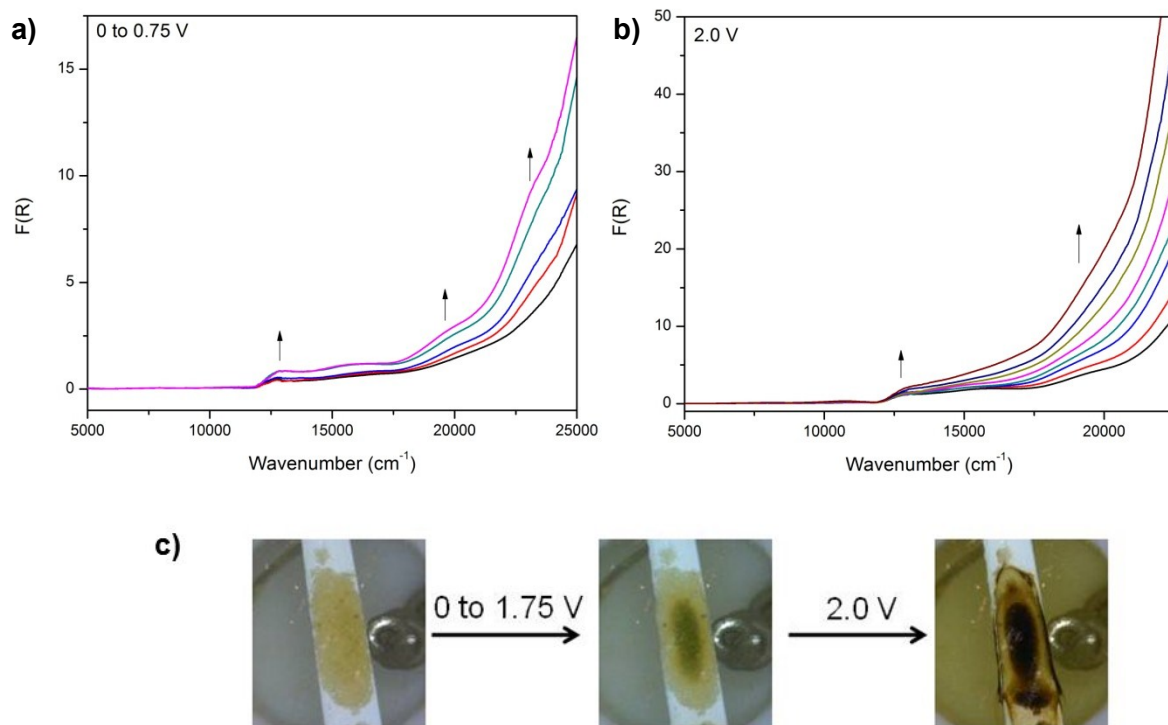


Figure S12. *In situ* solid state spectroelectrochemical experiment on **2** in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_3\text{CN}$ over the scan range of 5000-25000 cm^{-1} where a) the potential was increased from 0-1.75 V, b) the potential was held at 2.0 V and c) photos of the framework during the experiment.

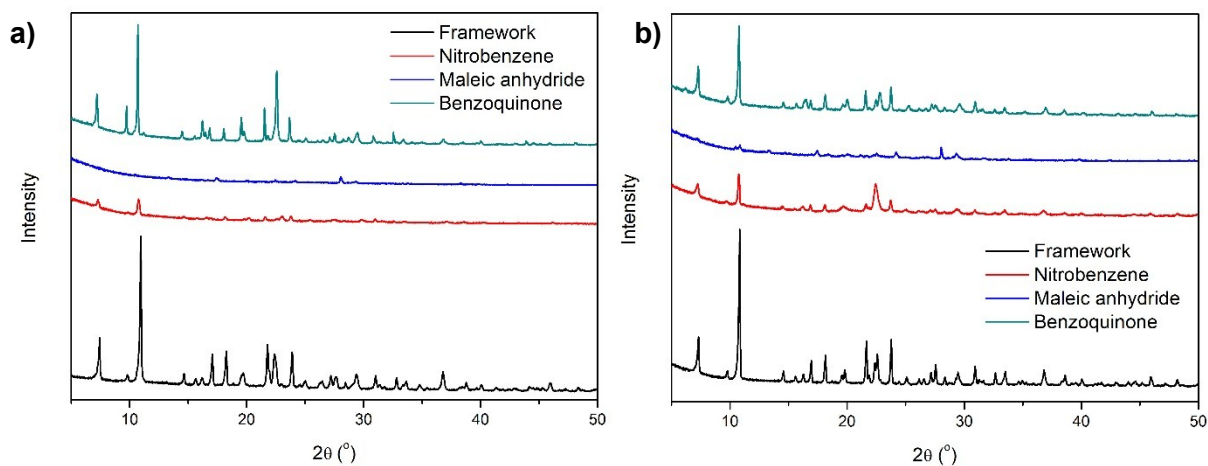


Figure S13. PXRD of (a) **1** and (b) **2** upon infiltration with nitrobenzene, maleic anhydride and benzoquinone electron acceptor guests.

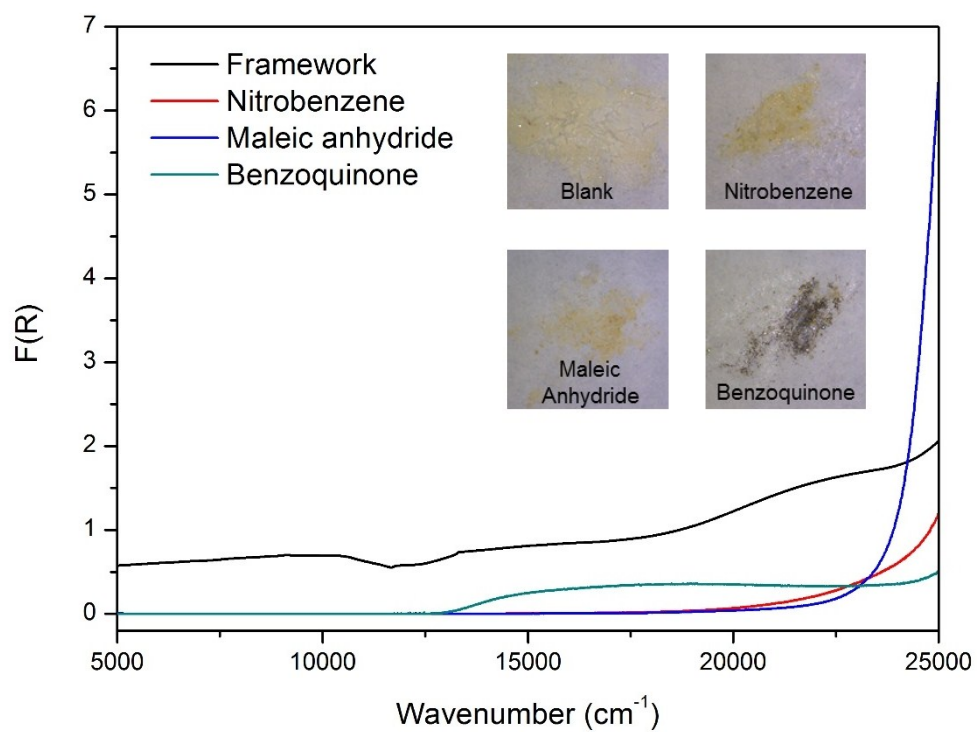


Figure S14. Vis/NIR spectra of **1** and upon infiltration with nitrobenzene, maleic anhydride and benzoquinone electron acceptor guests.

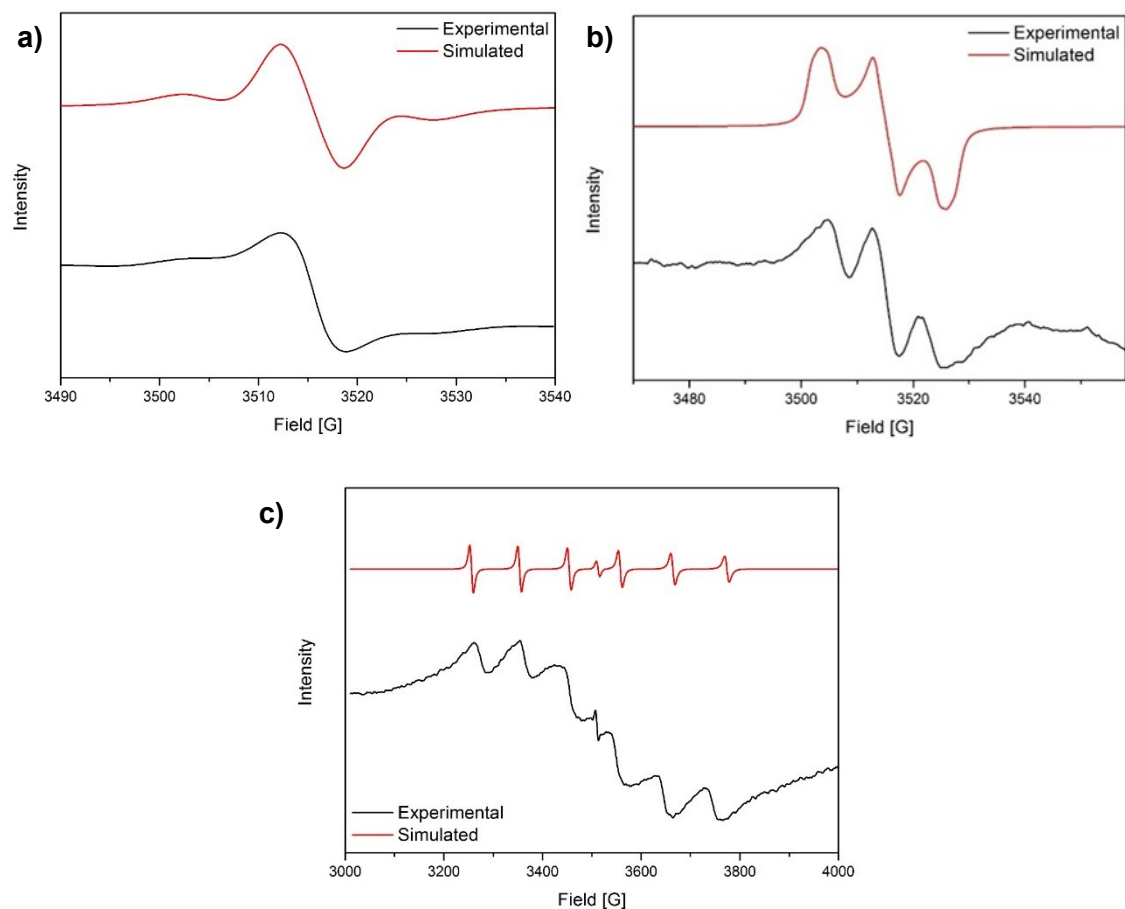


Figure S15. EPR spectra (X-band) of (a) **1** and (b) **2** during the spectroelectrochemical experiment in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_2\text{Cl}_2$ showing the EPR signal of the radical cation and c) of the Mn^{2+} signal of **1** against the experimentally obtained spectrum at 1.8 V. Parameters for the simulated spectrum are shown in Table S3.

Table S3. EPR parameters used for simulated spectra.

Compound	Solvent	Nucleus	<i>g</i> -factor	<i>A</i> _{iso} (MHz)	Reference
H ₃ TTPA	CH ₂ Cl ₂	Nitrogen	2.0068	11.5	Figure 5
[{Mn ₃ (TTPA)(MeOH) ₃ }Cl ₃] _n (1)	CH ₂ Cl ₂	Nitrogen	2.005	2.8	Figure S15
[{Mn ₃ (TTPA)(MeOH) ₃ }Cl ₃] _n (1)	CH ₂ Cl ₂	Manganese	2.001	291	Figure S15
[{Mn ₃ (TTPA)(DMF) ₃ }(NO ₃) ₃] _n (2)	CH ₂ Cl ₂	Nitrogen	2.001	4.2	Figure S18
[{Mn ₃ (TTPA)(DMF) ₃ }(NO ₃) ₃] _n (2)	CH ₂ Cl ₂	Manganese	2.001	280	Figure S18

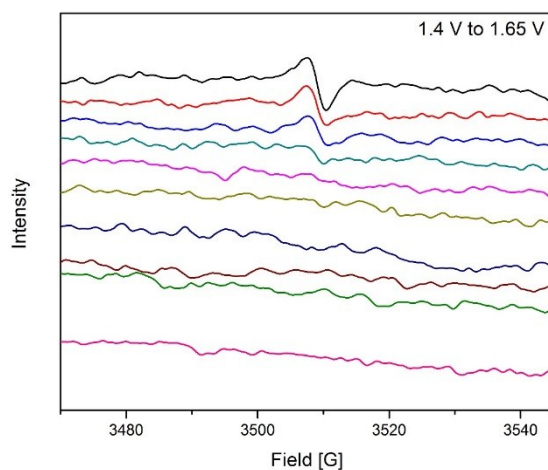


Figure S16. EPR spectra of **1** during the spectroelectrochemical experiment in [(*n*-C₄H₉)₄N]PF₆/CH₂Cl₂ where the potential was increased from 1.4 to 1.65 V.

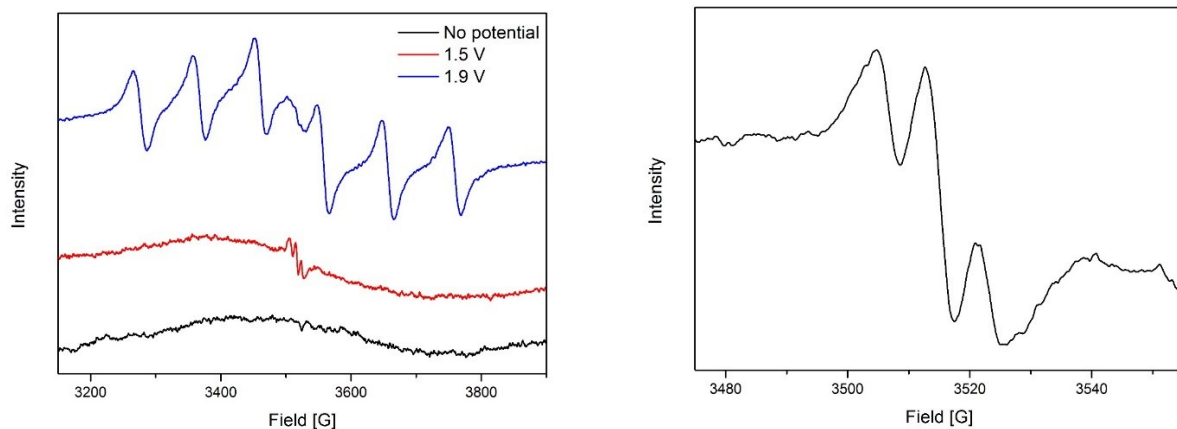


Figure S17. EPR spectra of **2** during the spectroelectrochemical experiment in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_2\text{Cl}_2$ where a) over the full spectral range the potential was increased from 0 to 1.9 V and b) radical cation.

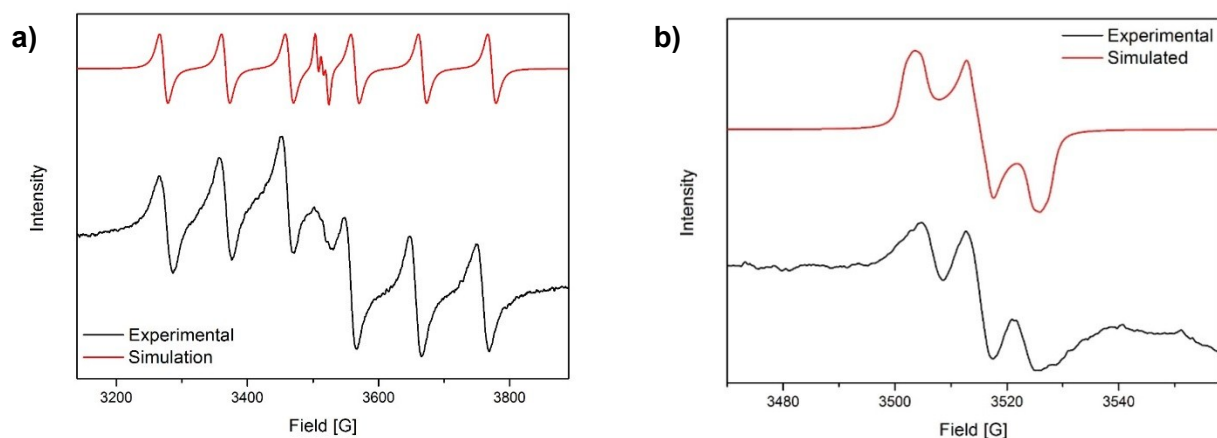


Figure S18. Simulated vs. experimental EPR spectra of **2** during the spectroelectrochemical experiment in $[(n\text{-C}_4\text{H}_9)_4\text{N}]\text{PF}_6/\text{CH}_2\text{Cl}_2$ b) at 1.9 V over the full spectral range and d) close up spectrum of the radical. Parameters for the simulated spectrum are shown in Table S3.

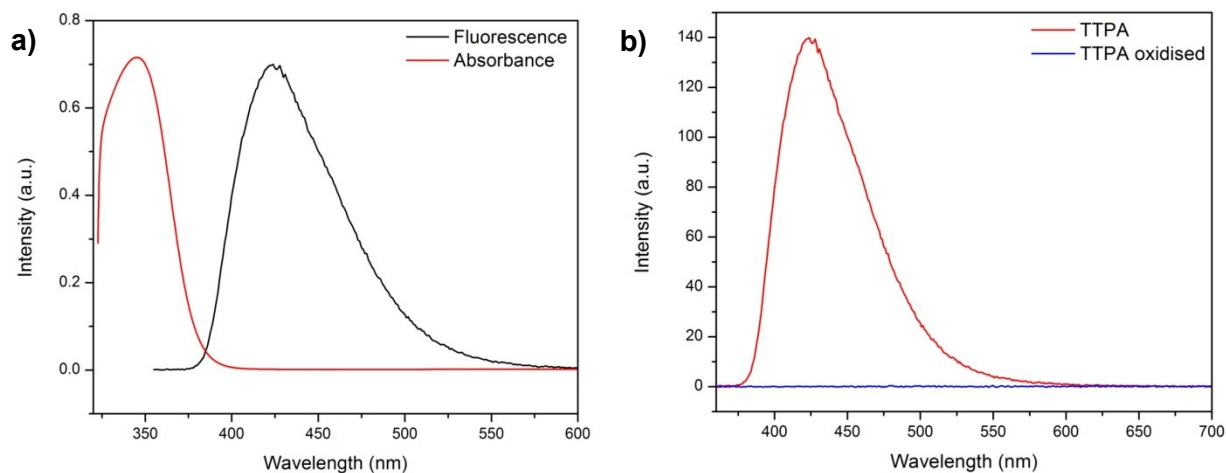


Figure S19. Fluorescence spectrum of (a) the H_3TTPA ligand as a solution in acetonitrile where $\lambda_{ex} = 345$ nm (28986 cm^{-1}) and (b) a comparison of the H_3TTPA ligand against the oxidised TTPA ligand with $NOPF_6$ in acetonitrile.

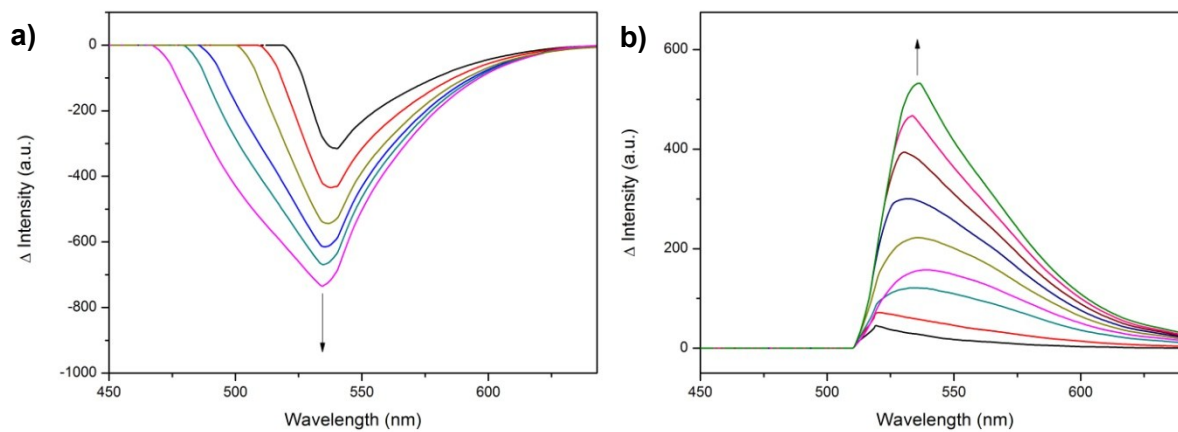


Figure S20. Spectroelectrochemical fluorescence spectrum of H_3TTPA as the potential was (a) increased from 0 to 1.5 V, (b) decreased from 1.5 to 0 V in $[(n-C_4H_9)_4N]PF_6/CH_3CN$ electrolyte.

Hückel Calculations Output: N(Ph-tetrazole)₃

 ψ_{tot}

Structure type : delocalized structure

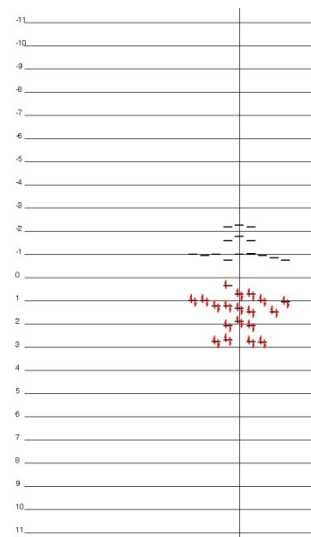
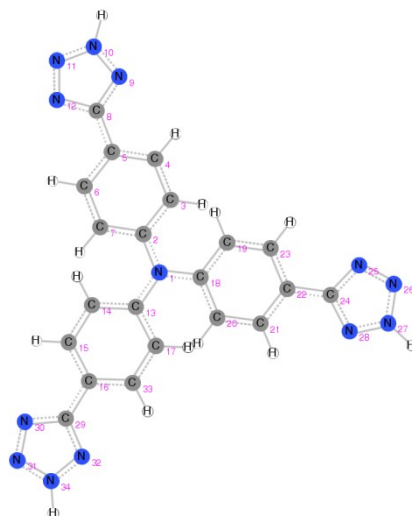
Number of atoms : 49

Number of Hückel atoms : 34

Number of π electrons : 37

Total charge : 1

1- Hamiltonian :

[illegible]

19	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	1.00β	$\alpha + 0.00\beta$	0.00β	
20	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	1.00β	0.00β	$\alpha + 0.00\beta$	
21	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	1.00β	$\alpha +$
22	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
23	C	0.00β	$\alpha + 0.00\beta$	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
24	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
25	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
26	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
27	N2	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
28	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
29	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
30	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
31	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
32	N1	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
33	C	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	
34	N2	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	0.00β	

2- Orbitals :																					
ei :	$\alpha + 2.79\beta$	$\alpha + 2.79\beta$	$\alpha + 2.79\beta$	$\alpha + 2.69\beta$	$\alpha + 2.09\beta$	$\alpha + 2.09\beta$	$\alpha + 1.91\beta$	$\alpha + 1.48\beta$	$\alpha + 1.48\beta$	$\alpha + 1.34\beta$	$\alpha + 1.25\beta$	$\alpha + 1.25\beta$	$\alpha + 1.06\beta$	$\alpha + 1.00\beta$	$\alpha + 1.00\beta$	$\alpha + 1.00\beta$	$\alpha + 0.73\beta$	$\alpha + 0.73\beta$	$\alpha + 0.37\beta$	$\alpha - 0.75\beta$	$\alpha -$
0.75β	$\alpha - 0.84\beta$	$\alpha - 0.94\beta$	$\alpha - 0.94\beta$	$\alpha - 1.00\beta$	$\alpha - 1.00\beta$	$\alpha - 1.00\beta$	$\alpha - 1.03\beta$	$\alpha - 1.57\beta$	$\alpha - 1.57\beta$	$\alpha - 1.76\beta$	$\alpha - 2.19\beta$	$\alpha - 2.19\beta$	$\alpha - 2.26\beta$								
ni :	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	2	1	0	0
0	0	0	0	0	0	0	0	0	0	0	0	0	0								
1	N2	-0.21	-0.00	-0.00	-0.65	-0.00	0.00	-0.30	-0.00	0.16	-0.00	-0.00	0.29	0.00	0.00	-0.00	0.00	0.00	0.44	0.00	
2	C	0.00	0.13	0.00	-0.00	0.00	0.00	0.20	0.00	0.00	-0.25	0.00	-0.00	-0.18							
0.16		-0.11	0.18	-0.10	-0.00	-0.00	-0.18	-0.17	0.33	0.29	0.11	-0.19	0.24								
3	C	-0.06	0.03	-0.02	-0.14	0.25	-0.11	0.08	-0.24	-0.04	-0.07	0.12	-0.07	-0.14	0.33	-0.19	0.32	-0.11	-0.05	-0.23	-0.10
		0.06	-0.01	-0.08	0.05	0.15	0.39	-0.27	0.00	0.13	-0.26	-0.15	-0.12	0.21	-0.20						
4	C	-0.06	0.06	-0.03	-0.06	0.29	-0.13	0.21	-0.03	-0.00	-0.09	-0.04	0.02	-0.12	0.33	-0.19	0.32	0.23	0.09	0.08	-0.20
		0.12	0.12	-0.10	0.06	-0.15	-0.39	0.27	0.18	-0.04	0.08	-0.03	0.15	-0.26	0.20						
5	C	-0.11	0.13	-0.07	-0.03	0.35	-0.16	0.32	0.19	0.03	-0.06	-0.17	0.10	0.02	-0.00	-0.00	0.00	0.28	0.12	0.26	0.25
0.15		-0.09	0.18	-0.10	-0.00	-0.00	-0.18	-0.07	0.14	0.21	-0.21	0.37	-0.25								-
6	C	-0.06	0.06	-0.03	-0.06	0.29	-0.13	0.21	-0.03	-0.00	-0.09	-0.04	0.02	-0.12	-0.33	0.19	-0.32	0.23	0.09	0.08	-0.20
		0.12	0.12	-0.10	0.06	0.15	0.39	-0.27	0.18	-0.04	0.08	-0.03	0.15	-0.26	0.20						
7	C	-0.06	0.03	-0.02	-0.14	0.25	-0.11	0.08	-0.24	-0.04	-0.07	0.12	-0.07	-0.14	-0.33	0.19	-0.32	-0.11	-0.05	-0.23	-0.10
		0.06	-0.01	-0.08	0.05	-0.15	-0.39	0.27	0.00	0.13	-0.26	-0.15	-0.12	0.21	-0.20						
8	C	-0.19	0.25	-0.14	0.05	0.16	-0.07	0.19	0.34	0.05	0.12	-0.13	0.07	0.26	0.00	0.00	0.00	-0.25	-0.10	-0.07	0.21
0.12		-0.16	0.03	-0.02	-0.00	-0.00	-0.16	0.19	-0.37	-0.30	0.16	-0.28	0.18								-
9	N1	-0.22	0.29	-0.17	0.09	-0.03	0.02	-0.01	-0.04	-0.01	-0.18	-0.34	0.20	0.29	0.00	0.00	-0.00	-0.33	-0.14	-0.18	-0.36
		0.21	0.32	0.15	-0.09	-0.00	0.00	0.00	0.05	-0.10	0.20	0.14	-0.06	0.11	-0.07						

10 N2	-0.32	0.42	-0.24	0.14	-0.22	0.10	-0.21	-0.39	-0.06	-0.27	-0.12	0.07	-0.10	-0.00	-0.00	-0.00	0.19	0.08	0.10	0.24	-
0.14	-0.26	-0.25	0.15	0.00	0.00	-0.00	0.09	0.01	-0.03	-0.02	0.01	-0.02	0.01								
11 N1	-0.23	0.30	-0.17	0.10	-0.12	0.06	-0.11	-0.00	-0.00	-0.00	0.18	0.36	-0.21	-0.26	-0.00	0.00	0.21	0.09	0.08	-0.15	
	0.08	0.27	0.44	-0.25	-0.00	-0.00	-0.00	-0.27	0.06	-0.11	-0.07	0.02	-0.04	0.03							
12 N1	-0.20	0.26	-0.15	0.07	0.02	-0.01	0.06	0.36	0.06	0.38	0.35	-0.21	-0.04	0.00	0.00	0.00	-0.13	-0.05	-0.10	-0.05	
	0.03	-0.10	-0.35	0.20	0.00	0.00	0.00	0.29	-0.13	0.24	0.17	-0.07	0.12	-0.08							
13 C	-0.11	-0.00	0.02	-0.32	-0.03	0.26	-0.06	0.12	0.30	-0.00	0.00	0.22	-0.03	-0.00	-0.00	0.00	0.04	0.33	-0.16	0.00	
	0.32	-0.11	0.00	0.21	-0.00	-0.00	-0.00	-0.18	0.37	-0.02	0.29	-0.22	0.00	0.24							
14 C	-0.06	-0.00	0.03	-0.14	-0.03	0.27	0.08	0.09	0.22	-0.07	0.00	0.14	-0.14	-0.24	-0.44	-0.01	0.02	0.12	-0.23	-0.00	-
0.12	-0.01	-0.00	-0.10	0.09	-0.30	-0.39	0.00	-0.29	0.01	-0.15	0.24	-0.00	-0.20								
15 C	-0.06	-0.00	0.06	-0.06	-0.03	0.31	0.21	0.01	0.03	-0.09	-0.00	-0.05	-0.12	-0.24	-0.44	-0.01	-0.03	-0.24	0.08	-0.00	-
0.23	0.12	-0.00	-0.12	-0.09	0.30	0.39	0.18	0.09	-0.00	-0.03	-0.30	0.00	0.20								
16 C	-0.11	-0.00	0.15	-0.03	-0.04	0.38	0.32	-0.07	-0.18	-0.06	-0.00	-0.20	0.02	0.00	0.00	-0.00	-0.04	-0.30	0.26	0.00	
	0.29	-0.09	0.00	0.21	-0.00	-0.00	-0.00	-0.18	0.16	-0.01	0.21	0.42	-0.01	-0.25							
17 C	-0.06	-0.00	0.03	-0.14	-0.03	0.27	0.08	0.09	0.22	-0.07	0.00	0.14	-0.14	0.24	0.44	0.01	0.02	0.12	-0.23	-0.00	-
0.12	-0.01	-0.00	-0.10	-0.09	0.30	0.39	0.00	-0.29	0.01	-0.15	0.24	-0.00	-0.20								
18 C	-0.11	-0.02	-0.01	-0.32	-0.21	-0.15	-0.06	0.20	-0.25	-0.00	-0.19	-0.11	-0.03	-0.00	-0.00	0.00	0.27	-0.20	-0.16	-0.28	-
0.16	-0.11	-0.18	-0.10	-0.00	-0.00	-0.00	-0.18	-0.20	-0.31	0.29	0.11	0.19	0.24								
19 C	-0.06	-0.03	-0.02	-0.14	-0.22	-0.16	0.08	0.15	-0.19	-0.07	-0.12	-0.07	-0.14	-0.29	0.14	0.38	0.10	-0.07	-0.23	0.10	
	0.06	-0.01	0.08	0.05	-0.47	0.07	-0.16	0.00	0.16	0.25	-0.15	-0.12	-0.20	-0.20							
20 C	-0.06	-0.03	-0.02	-0.14	-0.22	-0.16	0.08	0.15	-0.19	-0.07	-0.12	-0.07	-0.14	0.29	-0.14	-0.38	0.10	-0.07	-0.23	0.10	
	0.06	-0.01	0.08	0.05	0.47	-0.07	0.16	0.00	0.16	0.25	-0.15	-0.12	-0.20	-0.20							
21 C	-0.06	-0.06	-0.03	-0.06	-0.25	-0.18	0.21	0.02	-0.02	-0.09	0.04	0.02	-0.12	0.29	-0.14	-0.38	-0.20	0.15	0.08	0.20	
	0.12	0.12	0.10	0.06	-0.47	0.07	-0.16	0.18	-0.05	-0.07	-0.03	0.16	0.26	0.20							
22 C	-0.11	-0.13	-0.07	-0.03	-0.31	-0.22	0.32	-0.12	0.15	-0.06	0.17	0.10	0.02	0.00	0.00	-0.00	-0.24	0.18	0.26	-0.25	-
0.15	-0.09	-0.18	-0.10	-0.00	-0.00	-0.00	-0.18	-0.08	-0.13	0.21	-0.22	-0.36	-0.25								
23 C	-0.06	-0.06	-0.03	-0.06	-0.25	-0.18	0.21	0.02	-0.02	-0.09	0.04	0.02	-0.12	-0.29	0.14	0.38	-0.20	0.15	0.08	0.20	
	0.12	0.12	0.10	0.06	0.47	-0.07	0.16	0.18	-0.05	-0.07	-0.03	0.16	0.26	0.20							
24 C	-0.19	-0.25	-0.14	0.05	-0.14	-0.10	0.19	-0.22	0.27	0.12	0.13	0.07	0.26	0.00	0.00	0.00	0.22	-0.17	-0.07	-0.21	-
0.12	-0.16	-0.03	-0.02	-0.00	-0.00	-0.00	-0.16	0.23	0.35	-0.30	0.16	0.27	0.18								
25 N1	-0.20	-0.25	-0.15	0.07	-0.01	-0.01	0.06	-0.23	0.28	0.38	-0.35	-0.20	-0.04	-0.00	-0.00	-0.00	0.11	-0.08	-0.10	0.05	
	0.03	-0.10	0.35	0.20	0.00	0.00	0.00	0.29	-0.15	-0.23	0.17	-0.07	-0.12	-0.08							
26 N1	-0.23	-0.30	-0.18	0.10	0.11	0.08	-0.11	0.00	-0.00	0.18	-0.36	-0.21	-0.26	-0.00	-0.00	-0.00	-0.18	0.14	0.08	0.15	
	0.08	0.27	-0.44	-0.25	-0.00	-0.00	-0.00	-0.27	0.07	0.11	-0.07	0.03	0.04	0.03							
27 N2	-0.32	-0.41	-0.24	0.14	0.19	0.14	-0.21	0.25	-0.31	-0.27	0.12	0.07	-0.10	0.00	0.00	0.00	-0.16	0.12	0.10	-0.24	-
0.14	-0.26	0.25	0.14	0.00	0.00	0.00	0.09	0.02	0.03	-0.02	0.01	0.02	0.01								
28 N1	-0.22	-0.29	-0.17	0.09	0.03	0.02	-0.01	0.03	-0.03	-0.18	0.34	0.20	0.29	0.00	0.00	0.00	0.28	-0.22	-0.18	0.36	
	0.21	0.32	-0.15	-0.09	0.00	0.00	0.00	0.05	-0.12	-0.19	0.14	-0.07	-0.11	-0.07							
29 C	-0.19	-0.00	0.28	0.05	-0.02	0.17	0.19	-0.12	-0.32	0.12	-0.00	-0.15	0.26	0.00	0.00	-0.00	0.04	0.27	-0.07	0.00	
	0.25	-0.16	0.00	0.04	-0.00	-0.00	-0.00	-0.16	-0.42	0.02	-0.30	-0.32	0.01	0.18							
30 N1	-0.20	-0.00	0.30	0.07	-0.00	0.02	0.06	-0.13	-0.34	0.38	0.00	0.41	-0.04	-0.00	-0.00	0.00	0.02	0.13	-0.10	-0.00	-
0.05	-0.10	-0.00	-0.41	0.00	0.00	0.00	0.29	0.27	-0.01	0.17	0.14	-0.00	-0.08								
31 N1	-0.23	-0.00	0.35	0.10	0.01	-0.14	-0.11	0.00	0.00	0.18	0.00	0.41	-0.26	-0.00	-0.00	0.00	-0.03	-0.22	0.08	-0.00	-
0.17	0.27	0.00	0.51	-0.00	-0.00	-0.00	-0.27	-0.13	0.01	-0.07	-0.05	0.00	0.03								
32 N1	-0.22	-0.00	0.34	0.09	0.00	-0.04	-0.01	0.02	0.04	-0.18	-0.00	-0.40	0.29	0.00	0.00	-0.00	0.05	0.35	-0.18	-0.00	-
0.41	0.32	0.00	0.17	0.00	0.00	0.00	0.05	0.22	-0.01	0.14	0.13	-0.00	-0.07								
33 C	-0.06	-0.00	0.06	-0.06	-0.03	0.31	0.21	0.01	0.03	-0.09	-0.00	-0.05	-0.12	0.24	0.44	0.01	-0.03	-0.24	0.08	-0.00	-
0.23	0.12	-0.00	-0.12	0.09	-0.30	-0.39	0.18	0.09	-0.00	-0.03	-0.30	0.00	0.20								
34 N2	-0.32	-0.00	0.48	0.14	0.02	-0.24	-0.21	0.14	0.37	-0.27	-0.00	-0.14	-0.10	0.00	0.00	0.00	-0.03	-0.20	0.10	0.00	
	0.27	-0.26	-0.00	-0.29	0.00	-0.00	0.00	0.09	-0.03	0.00	-0.02	-0.02	0.00	0.01							

Total energy : $37\alpha + 59.32\beta$

3- Electronic density matrix : electron densities on and between the atoms (the density non bonded atoms is ill-defined) :

	1 N2	2 C	3 C	4 C	5 C	6 C	7 C	8 C	9 N1	10 N2	11 N1	12 N1	13 C	14 C	15 C	16 C	17 C	18 C	19 C	20 C	21
C	22 C	23 C	24 C	25 N1	26 N1	27 N2	28 N1	29 C	30 N1	31 N1	32 N1	33 C	34 N2								

1 N2	1.51	0.40	-0.04	-0.08	-0.01	-0.08	-0.04	0.05	0.02	-0.02	-0.02	0.01	0.40	-0.04	-0.08	-0.01	-0.04	0.40	-0.04	-0.04	-
0.08	-0.01	-0.08	0.05	0.01	-0.02	-0.02	0.02	0.05	0.01	-0.02	0.02	-0.08	-0.02								
2 C	0.40	0.93	0.58	0.02	-0.24	0.02	0.58	-0.02	0.10	-0.04	-0.02	0.07	-0.07	-0.08	0.02	0.06	-0.08	-0.07	-0.08	-0.08	
	0.02	0.06	0.02	-0.02	-0.02	0.02	0.02	-0.04	-0.02	-0.02	0.02	-0.04	0.02	0.02							
3 C	-0.04	0.58	1.00	0.71	0.02	-0.29	-0.00	-0.06	-0.02	0.01	0.02	-0.01	-0.08	-0.00	0.03	0.02	-0.00	-0.08	-0.00	-0.00	
	0.03	0.02	0.03	-0.02	-0.01	0.01	0.01	-0.02	-0.02	-0.01	0.01	-0.02	0.03	0.01							
4 C	-0.08	0.02	0.71	0.99	0.59	-0.01	-0.29	0.01	-0.15	0.07	0.04	-0.11	0.02	0.03	-0.01	-0.03	0.03	0.02	0.03	0.03	-
0.01	-0.03	-0.01	0.01	0.01	-0.01	-0.01	0.02	0.01	0.01	-0.01	0.02	-0.01	-0.01	-0.01							
5 C	-0.01	-0.24	0.02	0.59	0.97	0.59	0.02	0.38	0.02	-0.03	-0.06	0.00	0.06	0.02	-0.03	-0.03	0.02	0.06	0.02	0.02	-
0.03	-0.03	-0.03	0.02	0.01	-0.02	-0.02	0.03	0.02	0.01	-0.02	0.03	-0.03	-0.02								
6 C	-0.08	0.02	-0.29	-0.01	0.59	0.99	0.71	0.01	-0.15	0.07	0.04	-0.11	0.02	0.03	-0.01	-0.03	0.03	0.02	0.03	0.03	-
0.01	-0.03	-0.01	0.01	0.01	-0.01	-0.01	0.02	0.01	0.01	-0.01	0.02	-0.01	-0.01								
7 C	-0.04	0.58	-0.00	-0.29	0.02	0.71	1.00	-0.06	-0.02	0.01	0.02	-0.01	-0.08	-0.00	0.03	0.02	-0.00	-0.08	-0.00	-0.00	
	0.03	0.02	0.03	-0.02	-0.01	0.01	0.01	-0.02	-0.02	-0.01	0.01	-0.02	0.03	0.01							
8 C	0.05	-0.02	-0.06	0.01	0.38	0.01	-0.06	0.97	0.67	-0.22	-0.14	0.56	-0.02	-0.02	0.01	0.02	-0.02	-0.02	-0.02	-0.02	
	0.01	0.02	0.01	-0.01	-0.01	0.01	0.01	-0.01	-0.01	-0.01	0.01	-0.01	0.01	0.01							
9 N1	0.02	0.10	-0.02	-0.15	0.02	-0.15	-0.02	0.67	1.18	0.53	-0.35	-0.10	-0.04	-0.02	0.02	0.03	-0.02	-0.04	-0.02	-0.02	
	0.02	0.03	0.02	-0.01	-0.01	0.01	0.01	-0.02	-0.01	-0.01	0.01	-0.02	0.02	0.01							
10 N2	-0.02	-0.04	0.01	0.07	-0.03	0.07	0.01	-0.22	0.53	1.51	0.56	-0.27	0.02	0.01	-0.01	-0.02	0.01	0.02	0.01	0.01	-
0.01	-0.02	-0.01	0.01	0.01	-0.01	-0.01	0.01	0.01	0.01	-0.01	0.01	-0.01	-0.01								
11 N1	-0.02	-0.02	0.02	0.04	-0.06	0.04	0.02	-0.14	-0.35	0.56	1.09	0.72	0.02	0.01	-0.01	-0.02	0.01	0.02	0.01	0.01	-
0.01	-0.02	-0.01	0.01	0.01	-0.01	-0.01	0.01	0.01	0.01	-0.01	0.01	-0.01	-0.01								
12 N1	0.01	0.07	-0.01	-0.11	0.00	-0.11	-0.01	0.56	-0.10	-0.27	0.72	1.20	-0.02	-0.01	0.01	0.01	-0.01	-0.02	-0.01	-0.01	
	0.01	0.01	0.01	-0.01	-0.01	0.01	0.01	-0.01	-0.01	-0.01	0.01	-0.01	0.01	0.01							
13 C	0.40	-0.07	-0.08	0.02	0.06	0.02	-0.08	-0.02	-0.04	0.02	0.02	-0.02	0.93	0.58	0.02	-0.24	0.58	-0.07	-0.08	-0.08	
	0.02	0.06	0.02	-0.02	-0.02	0.02	0.02	-0.04	-0.02	0.07	-0.02	0.10	0.02	-0.04							
14 C	-0.04	-0.08	-0.00	0.03	0.02	0.03	-0.00	-0.02	-0.02	0.01	0.01	-0.01	0.58	1.00	0.71	0.02	-0.00	-0.08	-0.00	-0.00	
	0.03	0.02	0.03	-0.02	-0.01	0.01	0.01	-0.02	-0.06	-0.01	0.02	-0.02	-0.29	0.01							
15 C	-0.08	0.02	0.03	-0.01	-0.03	-0.01	0.03	0.01	0.02	-0.01	-0.01	0.01	0.02	0.71	0.99	0.59	-0.29	0.02	0.03	0.03	-
0.01	-0.03	-0.01	0.01	0.01	-0.01	-0.01	0.02	0.01	-0.11	0.04	-0.15	-0.01	0.07								
16 C	-0.01	0.06	0.02	-0.03	-0.03	-0.03	0.02	0.02	0.03	-0.02	-0.02	0.01	-0.24	0.02	0.59	0.97	0.02	0.06	0.02	0.02	-
0.03	-0.03	-0.03	0.02	0.01	-0.02	-0.02	0.03	0.38	0.00	-0.06	0.02	0.59	-0.03								
17 C	-0.04	-0.08	-0.00	0.03	0.02	0.03	-0.00	-0.02	-0.02	0.01	0.01	-0.01	0.58	-0.00	-0.29	0.02	1.00	-0.08	-0.00	-0.00	
	0.03	0.02	0.03	-0.02	-0.01	0.01	0.01	-0.02	-0.06	-0.01	0.02	-0.02	0.71	0.01							
18 C	0.40	-0.07	-0.08	0.02	0.06	0.02	-0.08	-0.02	-0.04	0.02	0.02	-0.02	-0.07	-0.08	0.02	0.06	-0.08	0.93	0.58	0.58	
	0.02	-0.24	0.02	-0.02	0.07	-0.02	-0.04	0.10	-0.02	-0.02	0.02	-0.04	0.02	0.02							
19 C	-0.04	-0.08	-0.00	0.03	0.02	0.03	-0.00	-0.02	-0.02	0.01	0.01	-0.01	-0.08	-0.00	0.03	0.02	-0.00	0.58	1.00	-0.00	-
0.29	0.02	0.71	-0.06	-0.01	0.02	0.01	-0.02	-0.02	-0.01	0.01	-0.02	0.03	0.01								
20 C	-0.04	-0.08	-0.00	0.03	0.02	0.03	-0.00	-0.02	-0.02	0.01	0.01	-0.01	-0.08	-0.00	0.03	0.02	-0.00	0.58	-0.00	1.00	
	0.71	0.02	-0.29	-0.06	-0.01	0.02	0.01	-0.02	-0.02	-0.01	0.01	-0.02	0.03	0.01							
21 C	-0.08	0.02	0.03	-0.01	-0.03	-0.01	0.03	0.01	0.02	-0.01	-0.01	0.01	0.02	0.03	-0.01	-0.03	0.03	0.02	-0.29	0.71	
	0.99	0.59	-0.01	0.01	-0.11	0.04	0.07	-0.15	0.01	0.01	-0.01	0.02	-0.01	-0.01							
22 C	-0.01	0.06	0.02	-0.03	-0.03	-0.03	0.02	0.02	0.03	-0.02	-0.02	0.01	0.06	0.02	-0.03	-0.03	0.02	-0.24	0.02	0.02	
	0.59	0.97	0.59	0.38	0.00	-0.06	-0.03	0.02	0.02	0.01	-0.02	0.03	-0.03	-0.02							
23 C	-0.08	0.02	0.03	-0.01	-0.03	-0.01	0.03	0.01	0.02	-0.01	-0.01	0.01	0.02	0.03	-0.01	-0.03	0.03	0.02	0.71	-0.29	-
0.01	0.59	0.99	0.01	-0.11	0.04	0.07	-0.15	0.01	0.01	-0.01	0.02	-0.01	-0.01								
24 C	0.05	-0.02	-0.02	0.01	0.02	0.01	-0.02	-0.01	-0.01	0.01	0.01	-0.01	-0.02	-0.02	0.01	0.02	-0.02	-0.02	-0.06	-0.06	
	0.01	0.38	0.01	0.97	0.56	-0.14	-0.22	0.67	-0.01	-0.01	0.01	-0.01	0.01	0.01							
25 N1	0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	-0.01	0.01	0.01	-0.01	-0.02	-0.01	0.01	0.01	-0.01	0.07	-0.01	-0.01	-
0.11	0.00	-0.11	0.56	1.20	0.72	-0.27	-0.10	-0.01	-0.01	0.01	-0.01	0.01	0.01								
26 N1	-0.02	0.02	0.01	-0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	0.01	0.02	0.01	-0.01	-0.02	0.01	-0.02	0.02	0.02	
	0.04	-0.06	0.04	-0.14	0.72	1.09	0.56	-0.35	0.01	0.01	-0.01	0.01	-0.01	-0.01							
27 N2	-0.02	0.02	0.01	-0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	0.01	0.02	0.01	-0.01	-0.02	0.01	-0.04	0.01	0.01	
	0.07	-0.03	0.07	-0.22	-0.27	0.56	1.51	0.53	0.01	0.01	-0.01	0.01	-0.01	-0.01							
28 N1	0.02	-0.04	-0.02	0.02	0.03	0.02	-0.02	-0.01	-0.02	0.01	0.01	-0.01	-0.04	-0.02	0.02	0.03	-0.02	0.10	-0.02	-0.02	-
0.15	0.02	-0.15	0.67	-0.10	-0.35	0.53	1.18	-0.01	-0.01	0.01	-0.02	0.02	0.01								

29 C	0.05	-0.02	-0.02	0.01	0.02	0.01	-0.02	-0.01	-0.01	0.01	0.01	-0.01	-0.02	-0.06	0.01	0.38	-0.06	-0.02	-0.02	-0.02	
	0.01	0.02	0.01	-0.01	-0.01	0.01	0.01	-0.01	0.97	0.56	-0.14	0.67	0.01	-0.22							
30 N1	0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	-0.01	0.01	0.01	-0.01	0.07	-0.01	-0.11	0.00	-0.01	-0.02	-0.01	-0.01	
	0.01	0.01	0.01	-0.01	-0.01	0.01	0.01	-0.01	0.56	1.20	0.72	-0.10	-0.11	-0.27							
31 N1	-0.02	0.02	0.01	-0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	0.01	-0.02	0.02	0.04	-0.06	0.02	0.02	0.01	0.01	-
0.01	-0.02	-0.01	0.01	0.01	-0.01	-0.01	0.01	-0.14	0.72	1.09	-0.35	0.04	0.56								
32 N1	0.02	-0.04	-0.02	0.02	0.03	0.02	-0.02	-0.01	-0.02	0.01	0.01	-0.01	0.10	-0.02	-0.15	0.02	-0.02	-0.04	-0.02	-0.02	
	0.02	0.03	0.02	-0.01	-0.01	0.01	0.01	-0.02	0.67	-0.10	-0.35	1.18	-0.15	0.53							
33 C	-0.08	0.02	0.03	-0.01	-0.03	-0.01	0.03	0.01	0.02	-0.01	-0.01	0.01	0.02	-0.29	-0.01	0.59	0.71	0.02	0.03	0.03	-
0.01	-0.03	-0.01	0.01	0.01	-0.01	-0.01	0.02	0.01	-0.11	0.04	-0.15	0.99	0.07								
34 N2	-0.02	0.02	0.01	-0.01	-0.02	-0.01	0.01	0.01	0.01	-0.01	-0.01	0.01	-0.04	0.01	0.07	-0.03	0.01	0.02	0.01	0.01	-
0.01	-0.02	-0.01	0.01	0.01	-0.01	-0.01	0.01	-0.22	-0.27	0.56	0.53	0.07	1.51								

4- Atomic charges :

Atom Charge :

1 N2	0.49
2 C	0.07
3 C	0.00
4 C	0.01
5 C	0.03
6 C	0.01
7 C	0.00
8 C	0.03
9 N1	-0.18
10 N2	0.49
11 N1	-0.09
12 N1	-0.20
13 C	0.07
14 C	0.00
15 C	0.01
16 C	0.03
17 C	0.00
18 C	0.07
19 C	0.00
20 C	0.00
21 C	0.01
22 C	0.03
23 C	0.01
24 C	0.03
25 N1	-0.20
26 N1	-0.09
27 N2	0.49
28 N1	-0.18
29 C	0.03
30 N1	-0.20
31 N1	-0.09
32 N1	-0.18
33 C	0.01
34 N2	0.49

Total charge : 1.00