

Electronic Supplementary Information

for

X-Ray Crystallographic and Computational Study on Uranyl-Salophen Complexes Bearing Nitro Groups

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Additional details on the crystallographic characterization of complexes 2-6.

COLLECT¹ software was used for the data collection (θ and ω scans) and DENZO-SMN² for the processing. The absorption effects were corrected with multi-scan method by SADABS³ (**3**·MeCN, **4**·DMF and **5**·H₂O) or empirical method by DENZO-SMN² (**2**·EtOH and **2**·MeOH). The structures were solved by direct methods with SIR2002⁴ and refined by full-matrix least-squares methods with WinGX-software,⁵ which utilizes the SHELXL-2013⁶ module. All C–H hydrogen positions (except the formyl hydrogens of DMF molecules in **4**·DMF and **6**·DMF) were calculated and refined as a riding atom model with 1.2 or 1.5 times the thermal parameter of the C atoms. The formyl hydrogens in **4**·DMF and **6**·DMF were found from the electron density map and restrained ($s = 0.02$) to a distance of 0.95 Å from parent C atom with thermal parameter of 1.2 times the C atom parameter. H-atoms bonded to O atoms were also found from the electron density maps, restrained ($s = 0.02$) to a distance of 0.84 Å from O atoms and their thermal parameters set to values of 1.5 times the O atom parameters. In the structure of **5**·H₂O, PLATON/SQUEEZE⁷ was used for a region of electron densities, because it was impossible to refine the present solvent molecule(s) due to high disorder. Anisotropic displacement parameters (ADP, for **4**·DMF) and the geometry (for **2**·EtOH) of the disordered nitroaryl ring atoms were constrained to equal. Also, O atom ADPs of the nitro group in a minor component of **4**·DMF were constrained to equal. For some other disordered atoms, the ADPs were restrained to be more similar (for **2**·EtOH and **4**·DMF, $s_1 = 0.01$ and $s_2 = 0.02$) and also more isotropic **2**·EtOH, $s = 0.01$ or 0.005). Furthermore, some of the atom distances in disordered moieties of the complexes were restrained to be rational (for **2**·EtOH, $s = 0.01$ or 0.02; **4**·DMF, $s = 0.02$). In **4**·DMF, the atom ADPs of coordinated dimethylformamide molecule, in **5**·H₂O the ADPs of co-crystallized acetone molecule and in **6**·DMF the ADPs of cocrystallized DMF molecule were restrained to be more similar ($s_1 = 0.01$ and $s_2 = 0.02$). Formyl O atoms of cocrystallized DMF molecule in **6**·DMF were also restrained to be more isotropic. Furthermore, the atom distances in acetone and the geometry of uncoordinated water molecule in **5**·H₂O, as well as, the atom distances and geometry of cocrystallized DMF molecule in **6**·DMF were restrained ($s = 0.02$) to be chemically reasonable.

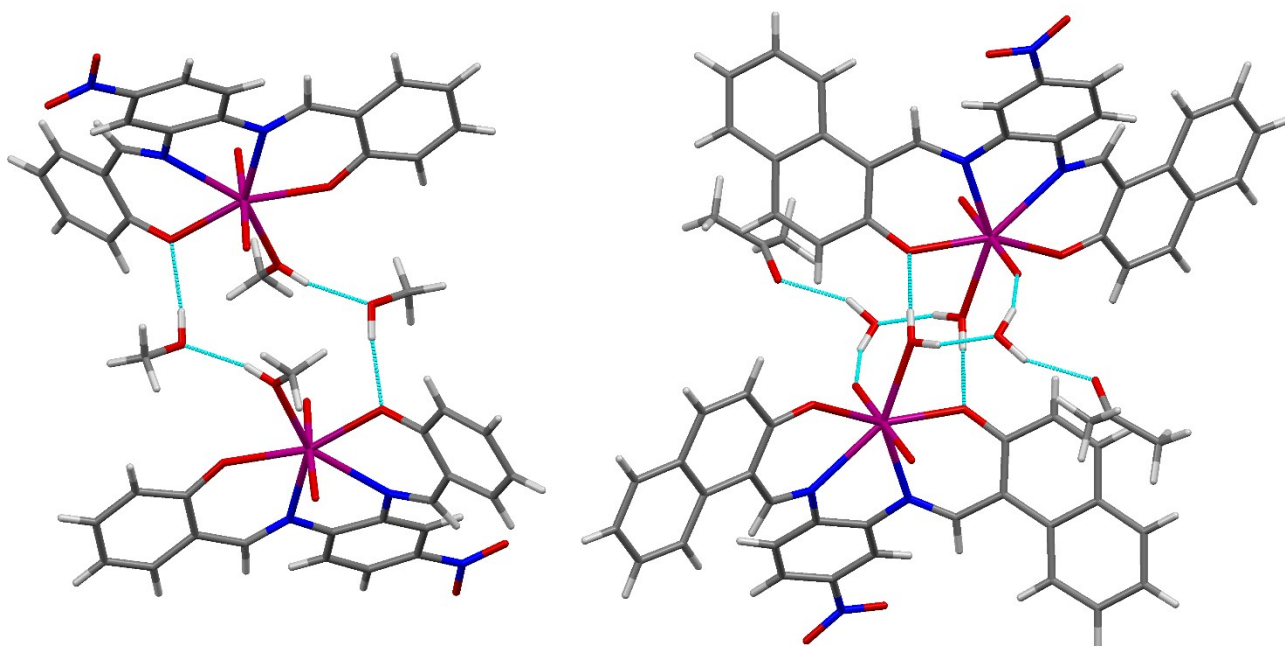
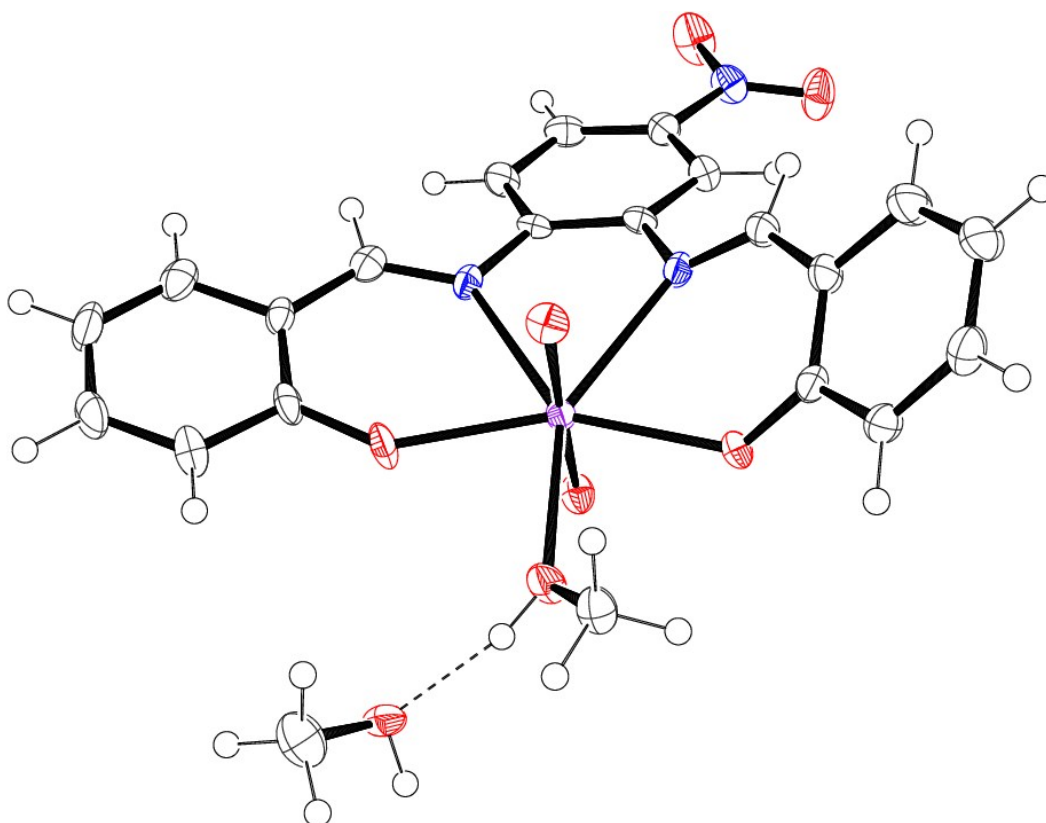


Figure S1. Dimeric structures of the complexes **2**·MeOH and **5**·H₂O showing the connecting hydrogen bonds (color codes: C=gray, H=white, N =blue, O=red, U =purple, H-bond=turquoise).

Table S1. Classical hydrogen bond geometries found from **5·H₂O**, **2·MeOH** and **2·EtOH**.

D-H···A	<i>d</i> (D-H)	<i>d</i> (H···A)	<i>d</i> (D···A)	<(DHA)
2·MeOH				
(U)-O(Me)-H···O(H)(Me)	0.84(2)	1.80(3)	2.618(8)	165(10)
O(Me)-H···O ⁱ	0.84(2)	1.90(3)	2.725(8)	170(9)
2·EtOH				
(U)-O(Et)-H···O ⁱⁱ	0.72(6)	1.91(6)	2.624(4)	170(7)
5·H₂O				
(U)-O(H)-H···O ⁱ	0.84(2)	1.84(2)	2.677(5)	173(7)
(U)-O(H)-H···O(H ₂)	0.83(2)	1.80(2)	2.611(7)	166(6)
O(H)-H···O(=C(CH ₃) ₂)	0.89(2)	1.93(2)	2.757(11)	154(5)
O(H)-H···O(=U) ⁱ	0.89(2)	2.04(4)	2.832(7)	148(6)

Symmetry operations: ⁱ -x+1,-y+1,-z+1; ⁱⁱ -x+1,-y,-z+2

**Figure S2.** Ortep of **2·MeOH** with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple, H-bond=dashed line).

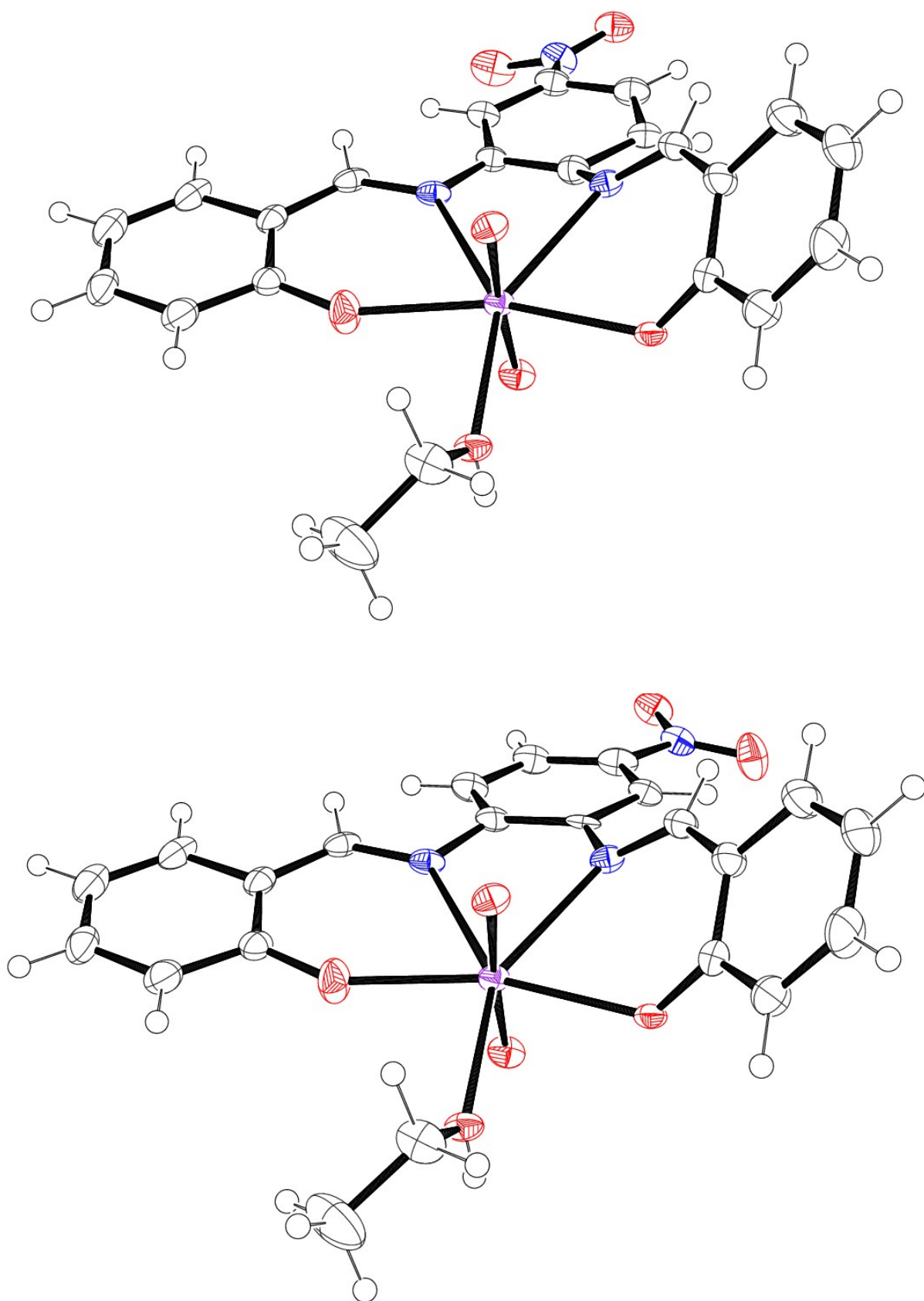


Figure S3. Ortep of 2-EtOH and its major (top) and minor (below) components with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple).

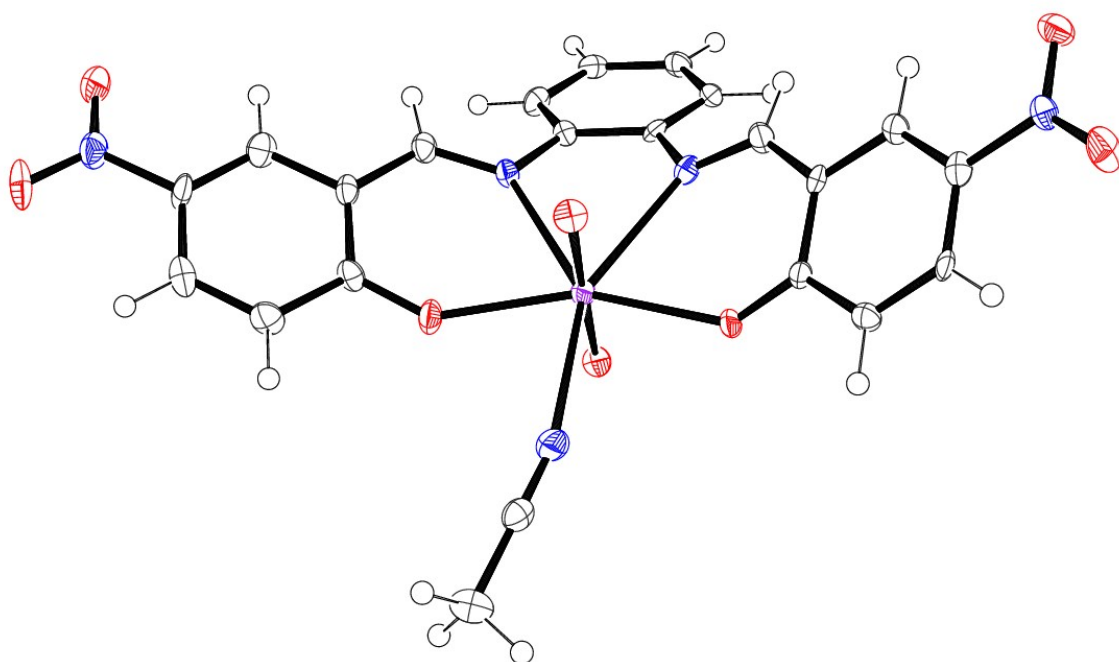


Figure S4. Ortep of **3·MeCN** with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple).

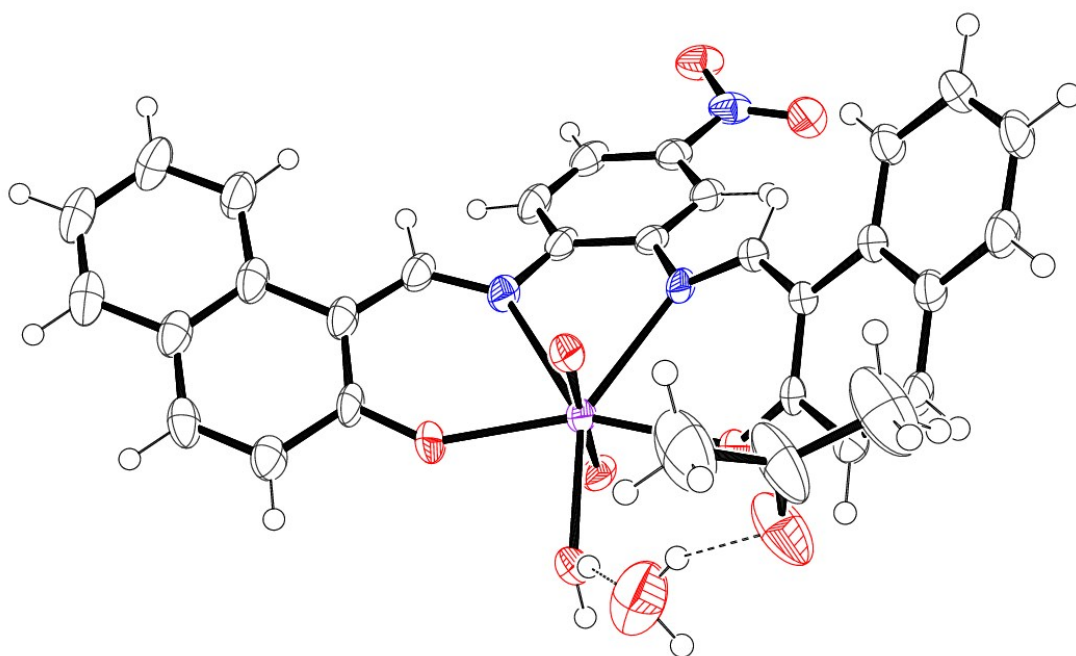


Figure S5. Ortep of **5·H₂O** with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple, H-bond=dashed line).

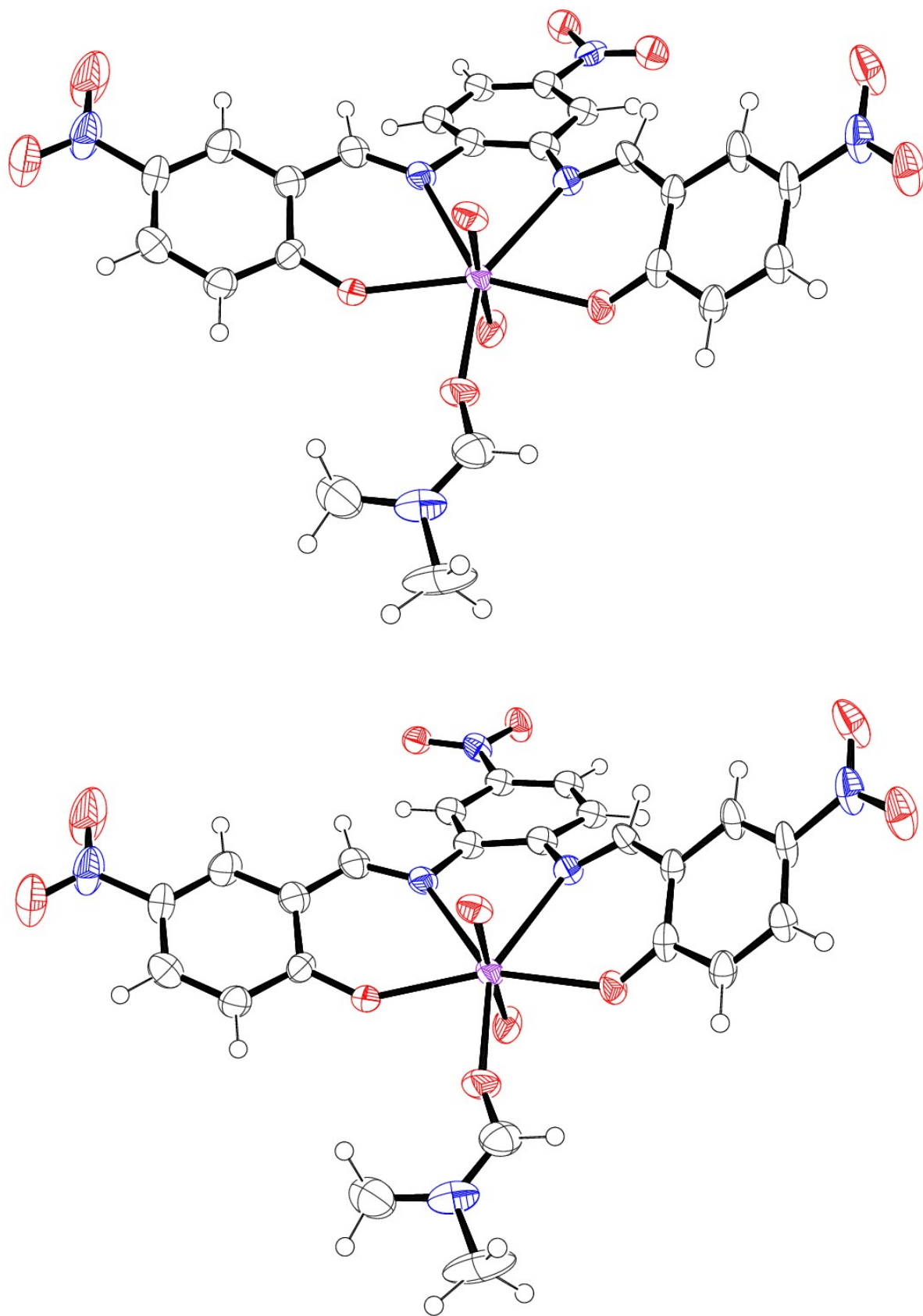


Figure S6. Ortep of 4·DMF and its major (top) and minor (below) components with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple).

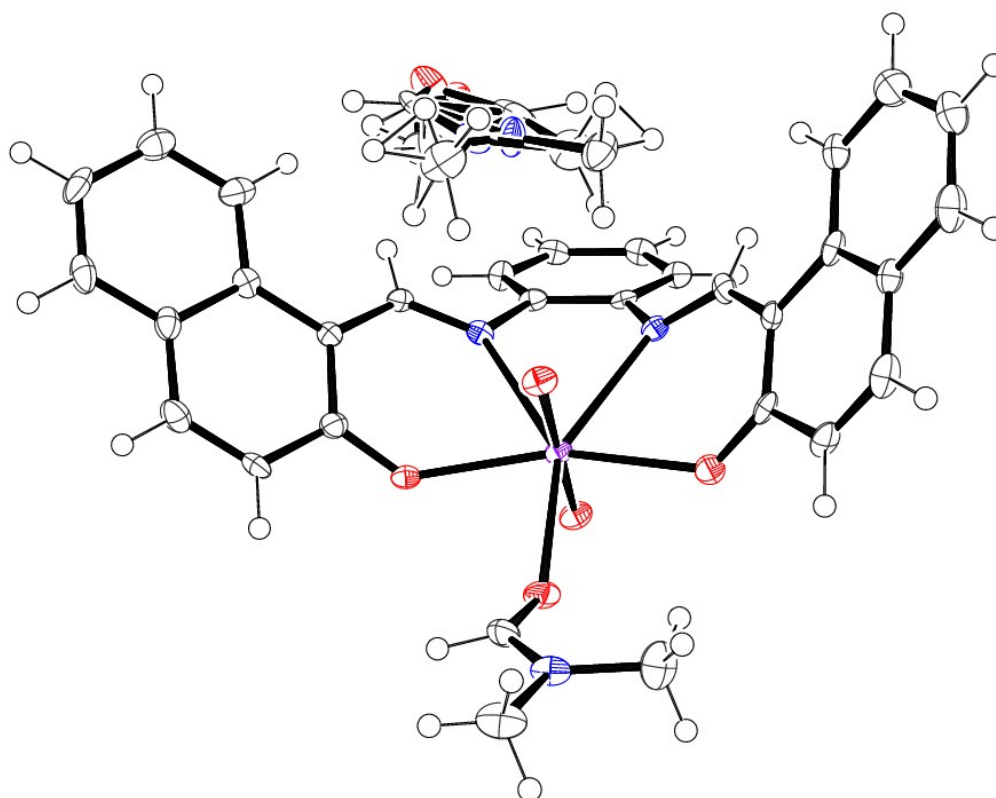


Figure S7. Ortep of **6·DMF** with 50 % ellipsoid probability (color codes: C=gray, H=black circles, N=blue, O=red, U=purple).

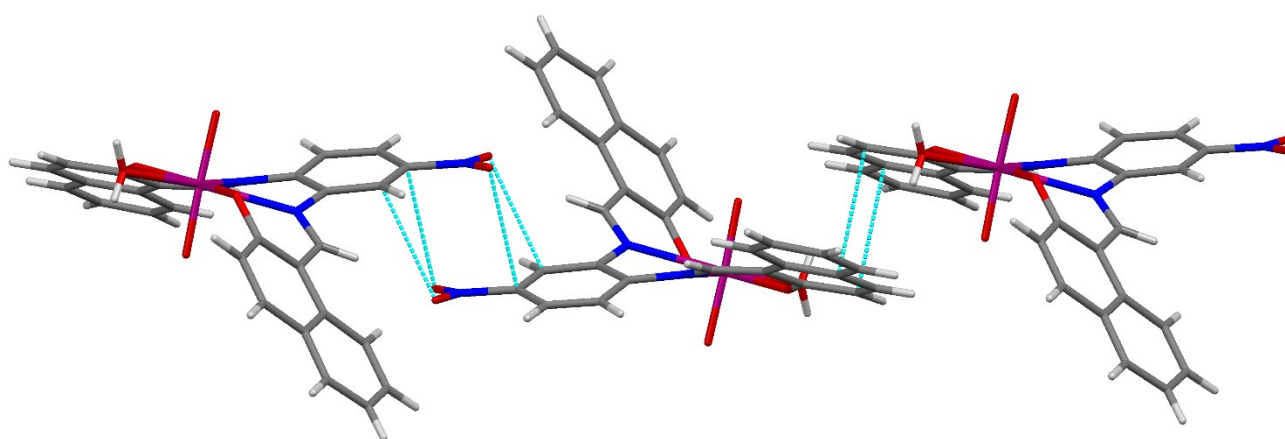


Figure S8. Two stacking interactions found in **5·H₂O** showing the closest intermolecular contacts between atoms. The first on the left between nitroaryl moieties (symmetry operation $-x, -y, 1-z$) has a close C $\cdots$ O contact of 3.16 Å, while the second on the right between naphthalene moieties (symmetry operation $1-x, -y, 2-z$) shows close contacts of C $\cdots$ C 3.31 Å and C_g $\cdots$ C_g 3.61 Å. Color codes: C=gray, H=white, N =blue, O=red, U =purple, contact=turquoise.

Table S2. Crystallographic data and parameters.

	2·MeOH	2·EtOH	3·MeCN	4·DMF	5·H₂O	6·DMF
CCDC	1519432	1519433	1519434	1519435	1519436	1519437
Formula	C ₂₂ H ₂₁ N ₃ O ₈ U	C ₂₂ H ₁₉ N ₃ O ₇ U	C ₂₂ H ₁₅ N ₅ O ₈ U	C ₂₃ H ₁₈ N ₆ O ₁₁ U	C ₅₉ H ₄₈ N ₆ O ₁₇ U ^b	C ₃₄ H ₃₂ N ₄ O ₆ U
<i>F</i> _w	693.45	675.43	715.42	792.46	1589.09 ^b	830.66
Crystal system	Triclinic	Triclinic	Triclinic	Triclinic	Triclinic	Orthorhombic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>Pbca</i>
Unit cell						
<i>a</i> [Å]	7.7986(2)	8.10340(10)	7.80830(10)	8.6139(3)	11.6878(3)	9.7279(2)
<i>b</i> [Å]	11.3229(2)	11.5084(2)	12.8964(2)	12.4803(5)	12.1737(3)	23.6223(3)
<i>c</i> [Å]	13.4185(3)	12.0611(2)	13.0519(2)	13.4843(5)	12.4532(4)	27.0822(3)
α [°]	74.707(2)	94.8897(10)	117.7150(10)	115.801(2)	73.370(2)	90
β [°]	81.9930(10)	90.5348(11)	96.8180(10)	102.929(2)	66.535(2)	90
γ [°]	84.047(2)	91.4276(12)	100.0270(10)	96.343(2)	83.915(2)	90
<i>V</i> [Å ³]	1129.01(4)	1120.27(3)	1115.29(3)	1235.81(8)	1557.31(8)	6223.37(17)
<i>Z</i>	2	2	2	2	1	8
ρ [Mg/m ³]	2.040	2.002	2.130	2.130	1.694 ^b	1.773
μ [mm ⁻¹]	7.243	7.293	7.338	6.644	5.264 ^b	5.268
<i>F</i> (000)	660	640	676	756	764 ^b	3232
Crystal size	0.14×0.08×0.04	0.20×0.13×0.06	0.20×0.15×0.12	0.10×0.06×0.04	0.40×0.08×0.03	0.22×0.10×0.07
θ range	2.12–25.24	2.35–25.25	2.72–25.25	2.50–25.25	2.57–25.25	1.72–25.00
Index ranges	-8≤ <i>h</i> ≤9 -13≤ <i>k</i> ≤13 -16≤ <i>l</i> ≤16	-9≤ <i>h</i> ≤9 -13≤ <i>k</i> ≤13 -14≤ <i>l</i> ≤14	-8≤ <i>h</i> ≤9 -15≤ <i>k</i> ≤14 -15≤ <i>l</i> ≤15	-10≤ <i>h</i> ≤9 -14≤ <i>k</i> ≤14 -16≤ <i>l</i> ≤16	-14≤ <i>h</i> ≤14 -14≤ <i>k</i> ≤14 -13≤ <i>l</i> ≤14	-11≤ <i>h</i> ≤11 -28≤ <i>k</i> ≤28 -32≤ <i>l</i> ≤32
Reflections	6897	7084	15413	17012	20447	52552
Data	4081	4045	4040	4476	5620	5450
<i>R</i> _{int}	0.0404	0.0325	0.0667	0.0728	0.0402	0.0827
Restraints / parameters	2 / 317	321 / 359	0 / 326	38 / 377	27 / 409	51 / 467
Completeness to θ [%]	99.6	99.5	99.9	99.9	99.6	99.4
Reflections [<i>I</i> > 2 σ (<i>I</i>)]	3750	3877	3521	3676	5050	4291
Goodness-of-fit on <i>F</i> ²	1.065	1.041	1.022	1.032	1.045	1.071
Final <i>R</i> [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0372 <i>wR</i> ₂ = 0.0956	<i>R</i> ₁ = 0.0235 <i>wR</i> ₂ = 0.0585	<i>R</i> ₁ = 0.0345 <i>wR</i> ₂ = 0.0609	<i>R</i> ₁ = 0.0430 <i>wR</i> ₂ = 0.0715	<i>R</i> ₁ = 0.0300 <i>wR</i> ₂ = 0.0683	<i>R</i> ₁ = 0.0368 <i>wR</i> ₂ = 0.0649
Final <i>R</i> (all data)	<i>R</i> ₁ = 0.0423 <i>wR</i> ₂ = 0.0988	<i>R</i> ₁ = 0.0250 <i>wR</i> ₂ = 0.0595	<i>R</i> ₁ = 0.0448 <i>wR</i> ₂ = 0.0636	<i>R</i> ₁ = 0.0596 <i>wR</i> ₂ = 0.0754	<i>R</i> ₁ = 0.0366 <i>wR</i> ₂ = 0.0710	<i>R</i> ₁ = 0.0543 <i>wR</i> ₂ = 0.0699
Largest peak / hole [e./Å ³]	2.524 / -1.334 ^a	1.294 / -1.046	0.805 / -0.753	1.089 / -0.966	1.060 / -0.752	0.643 / -0.807

^a Highest peak at 0.2988 0.2851 0.4817 (1.09 Å from U1), deepest hole at 0.4269 0.3276 0.3785 (0.88 Å from U1)); ^b Formula and values calculated without the squeezed⁷ out solvent.

Table S3: Dimensions (<i>a</i> , <i>b</i> , <i>c</i> in Å, α, β, γ in °) of the model's super-cells submitted to MD simulations						
	2·MeOH	2·EtOH	3·MeCN	4·DMF	5·H₂O	YALSOU
<i>a</i>	24.3102	23.3958	23.3756	25.8417	23.4249	12.5727
<i>b</i>	23.0168	22.6458	24.3474	24.9606	25.7928	16.8834
<i>c</i>	24.1222	26.837	24.9064	26.9686	26.1038	18.9578
α	94.8897	74.707	73.37	115.801	117.715	90
β	90.5348	81.993	66.535	102.929	96.818	90
γ	91.4276	84.047	83.915	96.343	100.027	90

Table S4. Relevant Φ angle parameters for CSD entries and complexes 2-6.						
CSD code	averaged ϕ		ϕ	ϕ'	ϕ''	$ \phi' - \phi'' $
DEYGUK	100.66	DEYGUK	108.91	35.36	45.41	7.0775
		DEYGUK	100.88	47.05	33.67	
		DEYGUK	91.58	46.69	51.17	
		DEYGUK	101.29	41.36	41.76	
DOZPAK	104.87			47.77	35.11	12.66
DOZPEO	141.66	DOZPEO	139.06	34.77	23.49	12.385
		DOZPEO	144.27	16.03	29.52	
DOZPOY	108.27			47.71	34.73	12.98
DOZPUE	100.25			35.23	48.85	13.62
ELUFUM	110.79			37.63	35.73	1.90
ELUGAT	112.28	ELUGAT	110.41	34.47	46.90	14.91
		ELUGAT	114.15	48.76	31.37	
EWOLAE	111.65			39.76	32.68	7.08
EWOLEI	109.82			29.66	42.45	12.79
EWOLOS	113.50			28.96	39.80	10.84
EWOLUI	113.24			45.75	25.18	20.57
EWOLIM	103.40			41.47	41.47	0.00
EWOMAF	125.35	EWOMAF	123.00	29.28	25.78	6.57
		EWOMAF	127.71	35.18	25.54	
FITKIC	121.82			41.01	26.94	14.07
FITKIC10	121.82			40.96	26.81	14.15
GOKVOT	104.55			43.51	36.94	6.57
HAJNOV	113.19			37.34	37.85	0.51
HENWIH	110.42			41.72	41.38	0.34
HEYXEP	120.45			24.04	44.25	20.21
HEYXIT	99.135	HEYXIT	104.56	51.94	43.52	4.55
		HEYXIT	93.71	44.67	45.35	
HEYXOZ	119.97	HEYXOZ	122.50	30.27	27.97	2.22
		HEYXOZ	122.17	30.44	27.49	
		HEYXOZ	122.84	28.85	29.82	
		HEYXOZ	122.39	27.71	30.37	
HEYXUF	93.13			47.86	54.51	6.65
HEYYAM	111.53			32.93	44.52	11.59
ISIWIR	111.87			36.70	34.06	2.64
KIRHEY	122.64			20.02	44.75	24.73
LADCUO	102.06			44.62	44.51	0.11
NICQAS	100.02	NICQAS	103.64	38.33	54.54	9.68
		NICQAS	96.14	39.81	42.96	
NICQEW	99.96	NICQEW	100.89	50.75	47.45	7.0175
		NICQEW	109.49	43.71	42.26	
		NICQEW	98.30	45.22	37.57	
		NICQEW	91.18	32.50	48.17	
NICQIA	103.48			42.92	41.93	0.99
NOGSOS	92.81			55.73	38.97	16.76
NOGSUY	107.29			36.71	36.78	0.07
OGOKIF	101.01			40.63	39.00	1.63
PIYCEF	119.05			37.87	36.99	0.88
PIYCOP	120.19			47.30	13.95	33.35
QUBMIL	136.71			23.14	25.89	2.75
QUJGEI	114.36	QUJGEI	112.59	29.26	39.48	9.405
		QUJGEI	116.13	29.83	38.42	
REPPEJ	106.20					
SAJMOG	108.87			48.84	33.28	15.56
SONTEU	110.49			43.91	32.32	11.59
WIBTIK	111.835	WIBTIK	107.15	34.82	30.81	4.18

		WIBTIK	116.52	36.77	41.12	
YALRAF	120.71			40.52	26.29	14.23
YALREY	105.15			37.74	37.22	0.52
YALRIN	101.16			37.97	38.88	0.91
YALROT	110.025	YALROT	115.41	37.91	32.31	15.81
		YALROT	104.64	52.47	26.45	
YALRUZ	104.41		105.23	32.48	42.41	6.695
			103.59	37.24	40.70	
YALSAG	112.05			38.34	38.34	0.00
YALSEK	112.74			35.95	35.95	0.00
YALSIO	118.695	YALSIO	130.88	19.50	29.68	11.41
		YALSIO	106.51	45.21	32.57	
YUYKAG	109.82			28.02	54.98	26.96
1-MeCN YALSOU	100.13	YALSOU		41.43	41.43	0.00
compound name	averaged ϕ		ϕ	ϕ'	ϕ''	$ \phi' - \phi'' $
2-EtOH	127.1300			2.9200	53.5100	50.59
2-MeOH	129.2700			8.9700	42.8200	33.85
3-MeCN	133.4000			16.6300	34.6900	18.06
4-DMF	130.6400			20.0300	33.4700	13.44
5-H2O	127.8200			9.9700	42.8000	32.83
6-DMF	105.0500			37.5000	45.7100	8.21

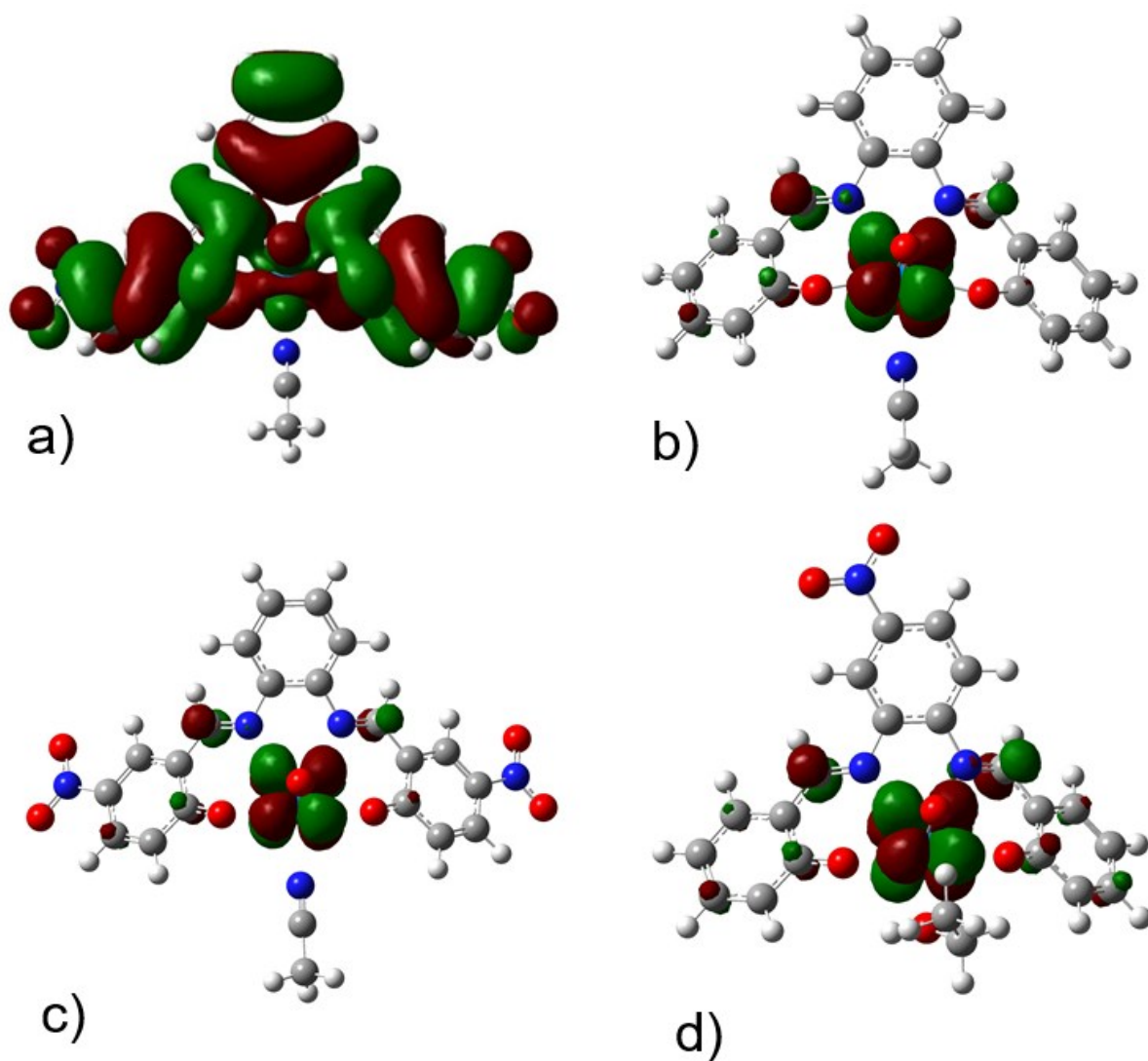
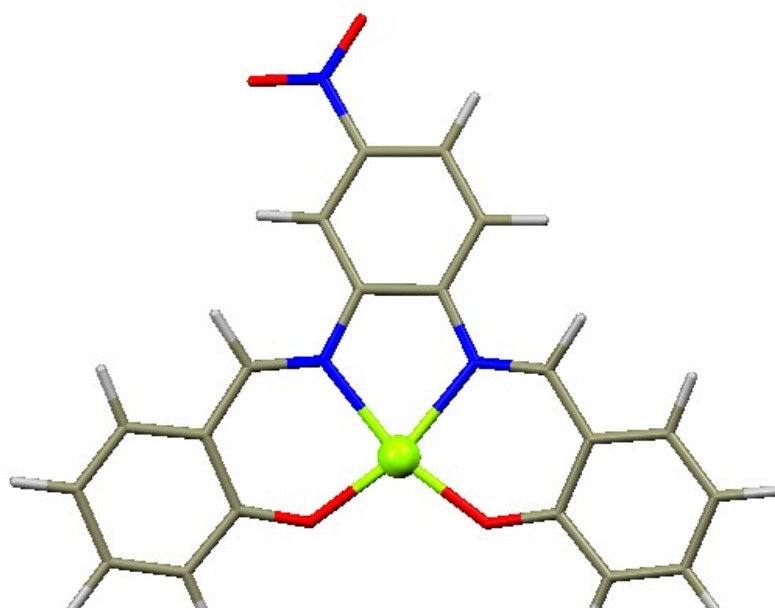


Figure S9. a) 3-MeCN HOMO isosurface, and LUMO isosurfaces for 1-MeCN (b), 3-MeCN (c) and 2-EtOH (d), showing positive (red) and negative (green) values. Calculated HOMO-LUMO energy differences are 34.2, 34.0 and 35.4 Kcal/mol for 1-MeCN, 2-EtOH and 3-MeCN, respectively.

a)



b)

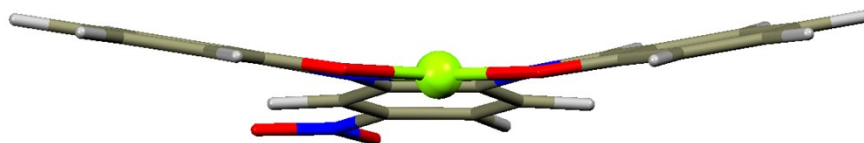


Figure S10. Top a) and side b) views of the calculated structure for the Mg-salophen complex. $\Phi = 160.70^\circ$. Colour code: C = grey, H = light grey, N = blue, O = red, Mg = yellow.

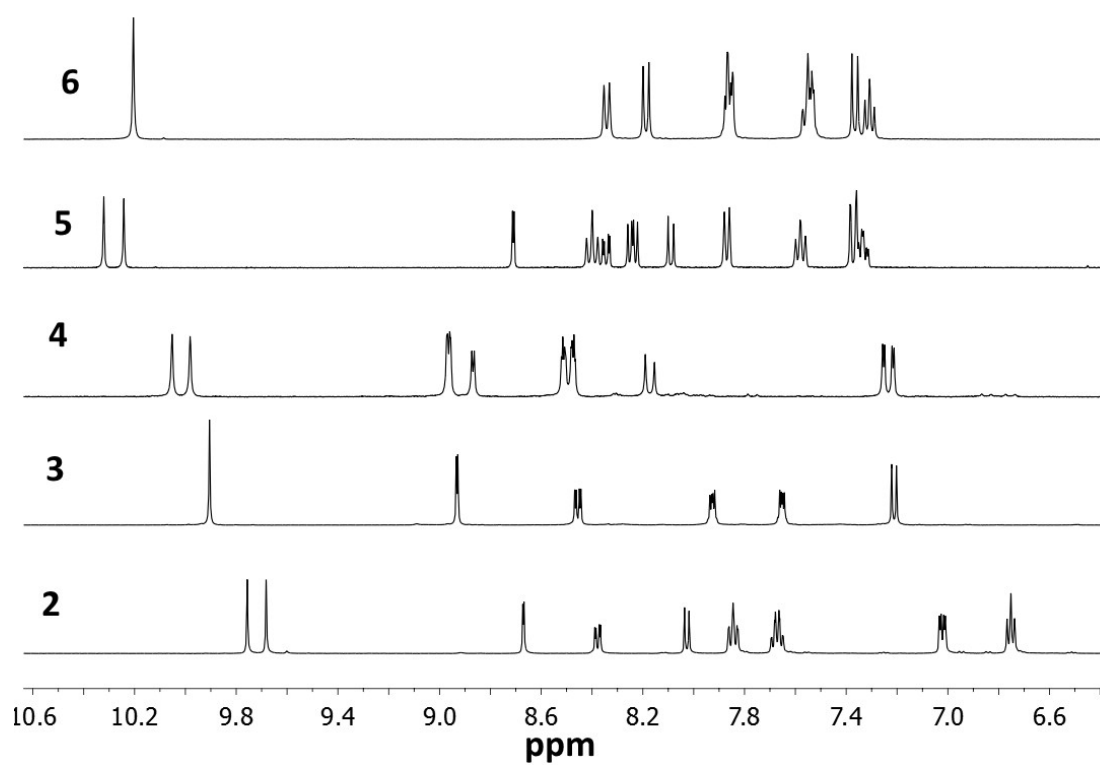


Fig. S11. Portion of the ^1H -NMR spectra of complexes **2-6** in $\text{DMSO-}d_6$.

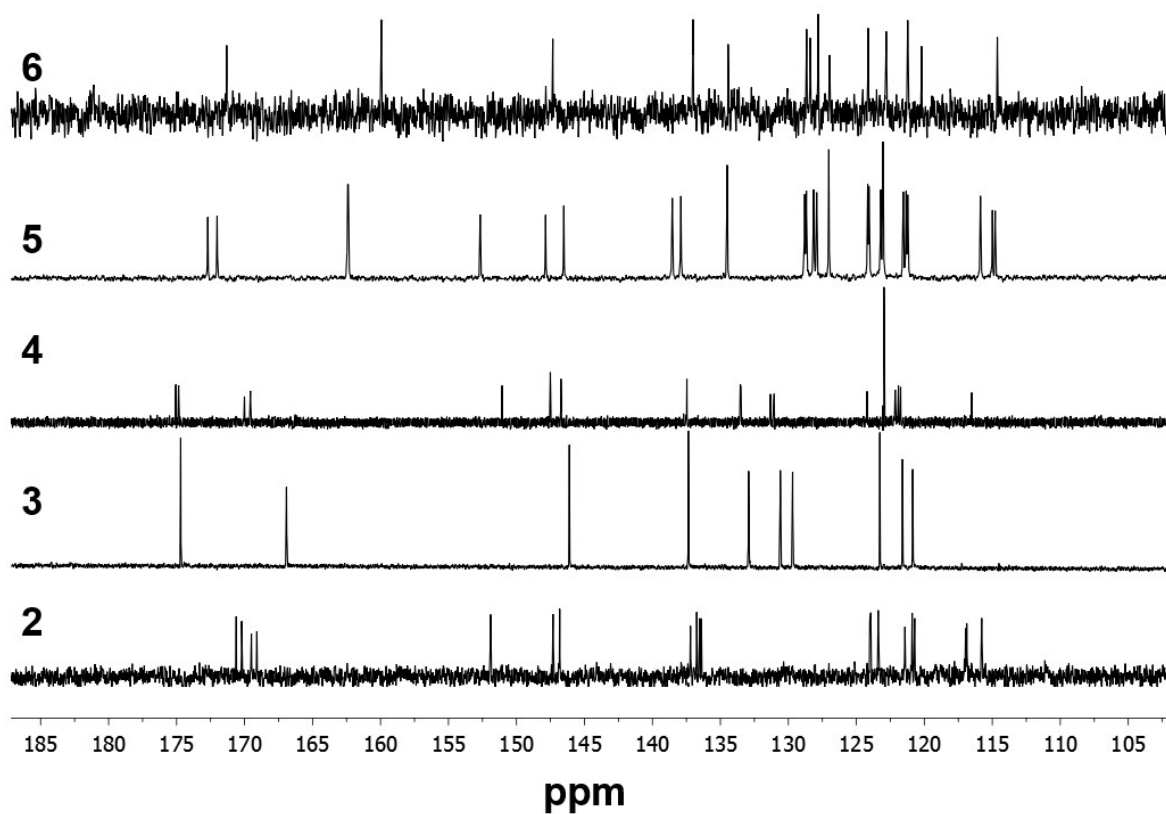


Fig. S12. Portion of the ^{13}C -NMR spectra of complexes **2-6** in $\text{DMSO-}d_6$.

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- ¹ COLLECT, Bruker AXS, Inc., Madison, Wisconsin, USA, 2008.
- ² Z. Otwinowski and W. Minor, *Methods Enzymol.*, vol. 276, *Macromolecular Crystallography*, Part A, eds. C. W. Carter Jr and R. M. Sweet, Academic Press, New York, 1997, pp. 307-326.
- ³ G. M. Sheldrick, SADABS, University of Göttingen, Germany, 1996.
- ⁴ M. C. Burla, M. Camalli, B. Carrozzini, G. L. Cascarano, C. Giacovazzo, G. Polidori and R. Spagna, *J. Appl. Crystallogr.*, 2003, **36**, 1103.
- ⁵ L. J. Farrugia, *J. Appl. Crystallogr.*, 2012, **45**, 849.
- ⁶ G. M. Sheldrick, *Acta Crystallogr., Sect. A: Found. Crystallogr.*, 2008, **64**, 112.
- ⁷ A. L. Spek, *Acta Crystallogr., Sect. D: Biological Crystallography*, 2009, **65**, 148.