

# Supporting information

<b>h</b>	<b>k</b>	<b>l</b>	<b>d</b>
0	0	3	<b>4.683</b>
1	0	1	2.401
0	0	6	2.341
0	1	2	2.303

Lithium cobalt oxide:

Hexagonal structure,  
 $a = 2.815\text{\AA}$ ,  $c = 14.05\text{\AA}$

<b>h</b>	<b>k</b>	<b>l</b>	<b>d</b>
1	1	1	<b>2.460</b>
2	0	0	2.13
3	1	1	1.284
2	2	2	1.230

Cobalt (II) oxide:

Rock-salt structure,  
 $a = 4.2615\text{\AA}$

Table 1: Crystal structure and d-spacing parameters for LCO and Co(II)O. The values shown in red are denoted in the TEM images

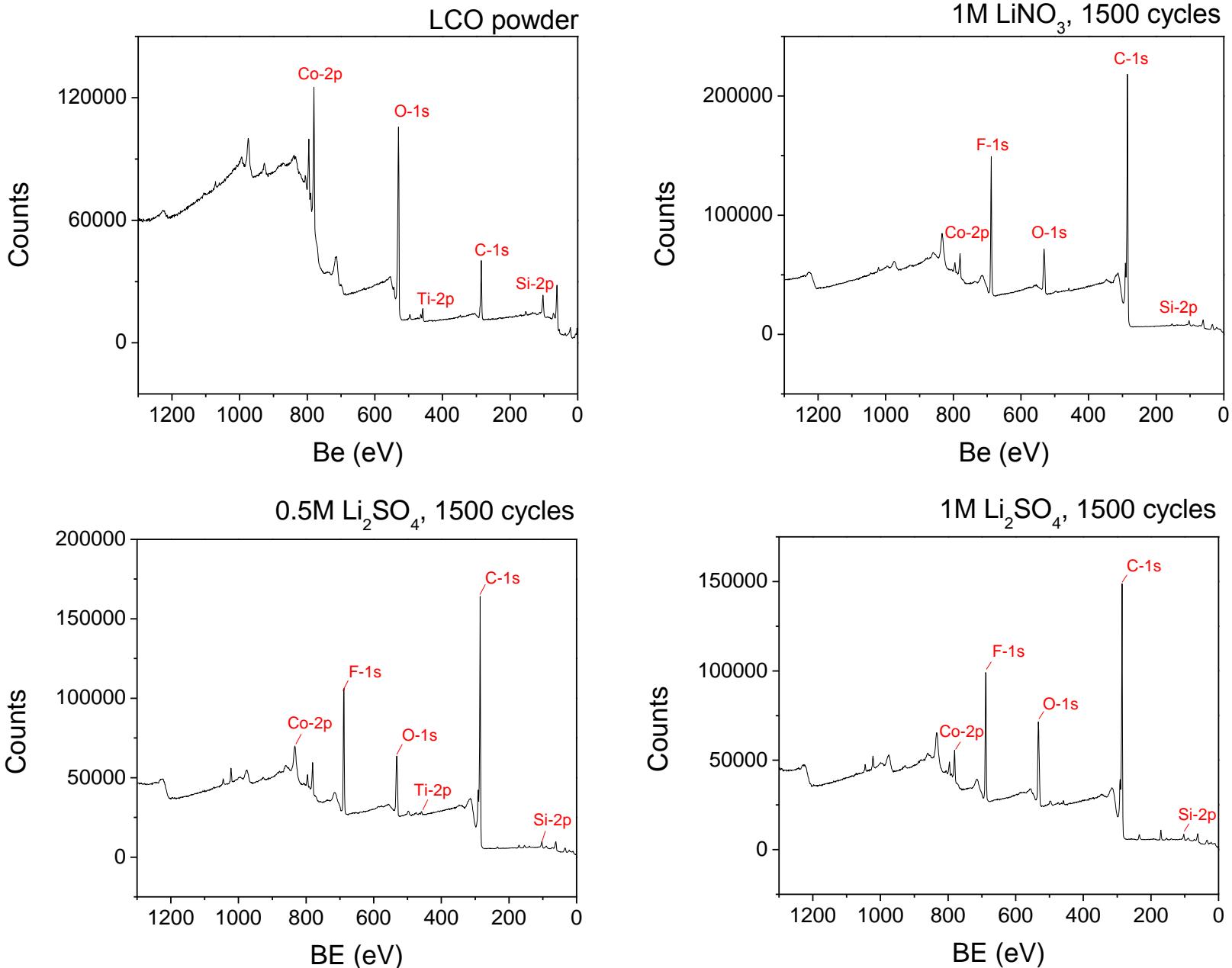


Figure S1. Detailed XPS scans for (a) Uncycled LCO powder, (b), (c), (d) LCO cycled in 1M LiNO<sub>3</sub>, 0.5M Li<sub>2</sub>SO<sub>4</sub> and 1M Li<sub>2</sub>SO<sub>4</sub> respectively

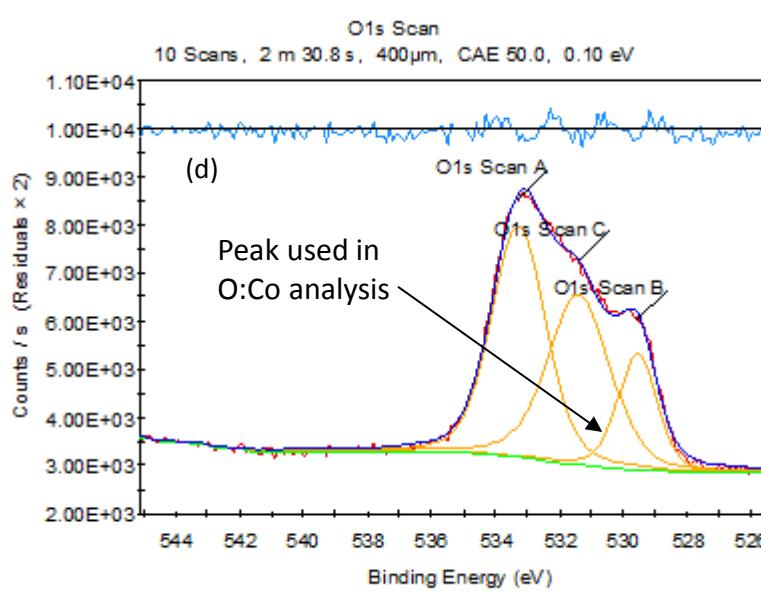
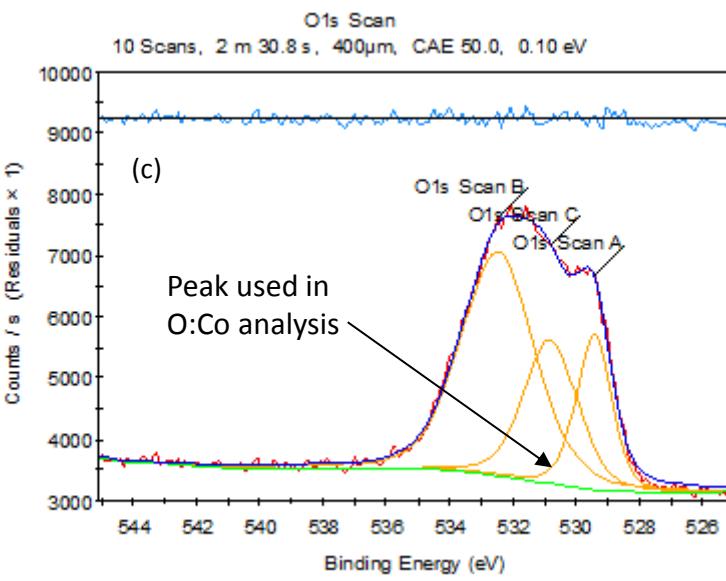
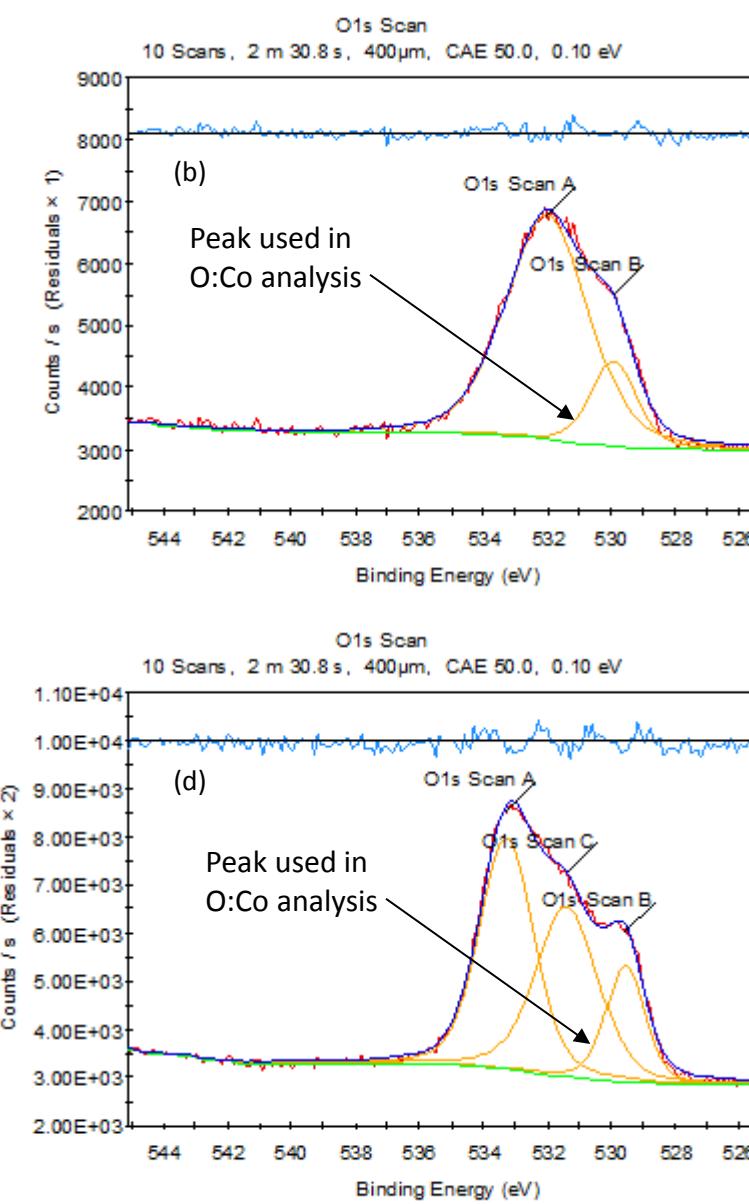
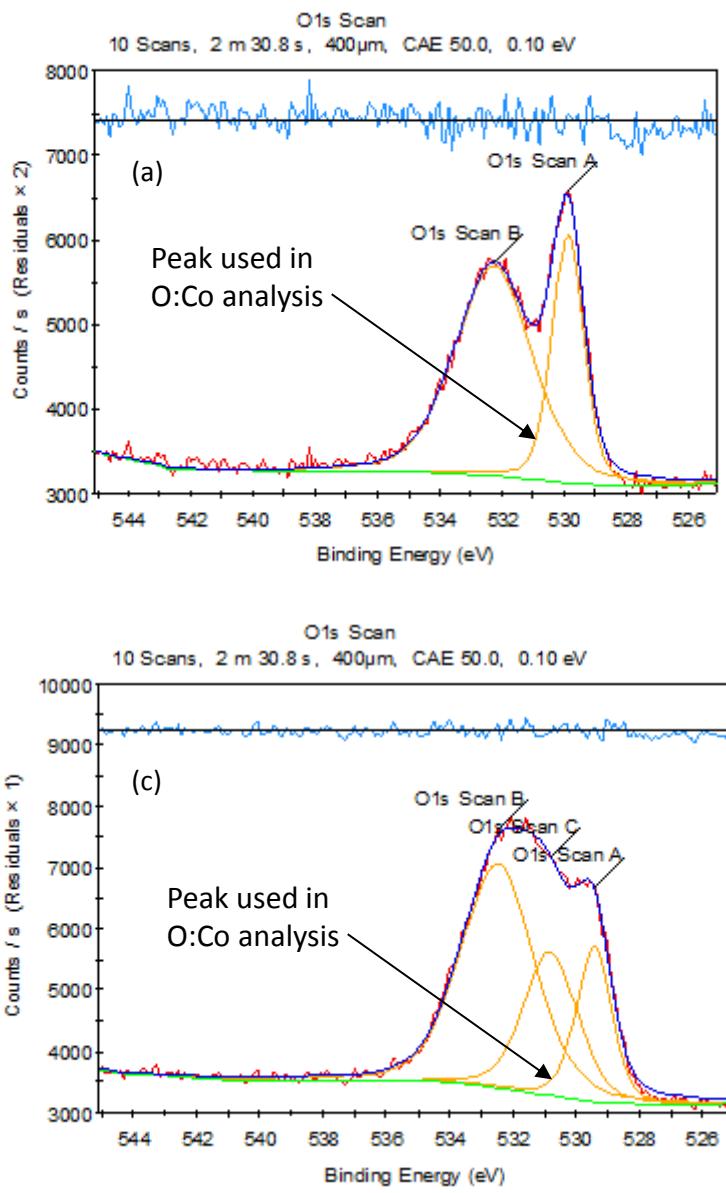


Figure S2: XPS spectra for O-1s peaks in (a) pure LCO and electrodes cycled in different (b) 1M LiNO<sub>3</sub>, (c) 0.5M Li<sub>2</sub>SO<sub>4</sub> and (d) 1M Li<sub>2</sub>SO<sub>4</sub> for 1500 cycles.

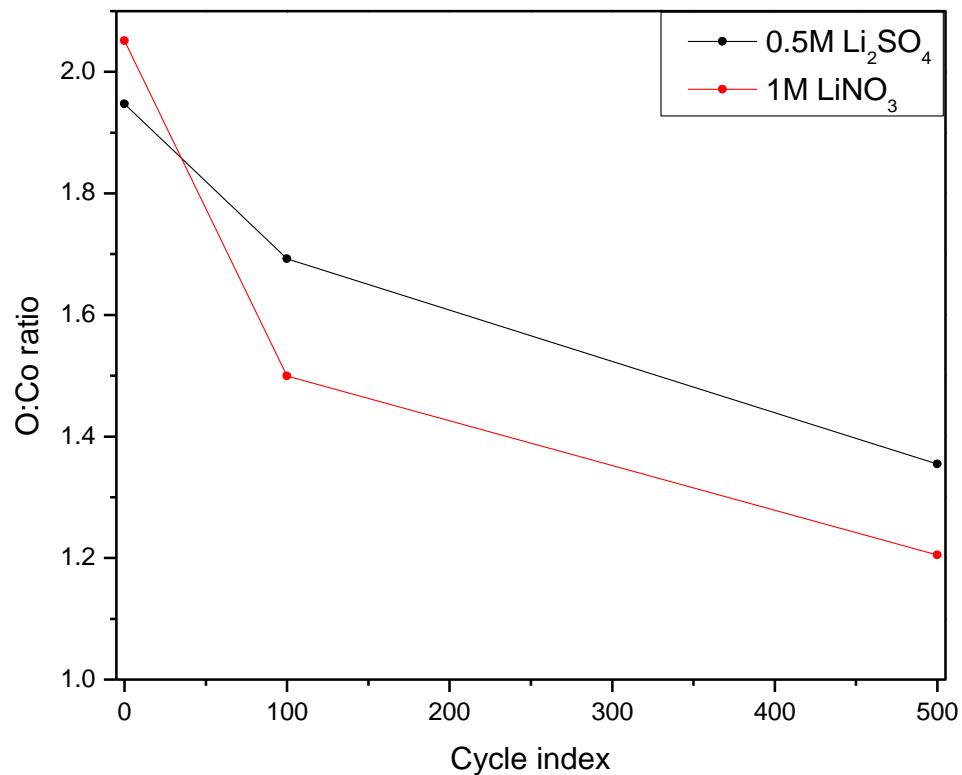


Figure S3: XPS-calculated O:Co ratio of the surface of LCO electrodes cycled in different electrolytes for 1, 100 and 500 cycles.