

Supplementary Information

A Large Area (70 cm²) Monolithic Perovskite Solar Module with High Efficiency and Stability

Anish Priyadarshi,^a Lew Jia Haur,^a Paul Murray,^b Dongchuan Fu,^b Sneha Kulkarni,^a Guichuan Xing,^c Tze Chien Sum,^d Nripan Mathews^{*ae} and Subodh G. Mhaisalkar^{*ae}

^aEnergy Research Institute @ NTU (ERI@N), Research Techno Plaza, X-Frontier Block, Level 5, 50 Nanyang Drive, Singapore 637553. E-mail: Subodh@ntu.edu.sg; Nripan@ntu.edu.sg

^bDyesol Limited Company, Queanbeyan, NSW 2620, Australia

^cInstitute of Applied Physics and Materials Engineering, Faculty of Science and Technology, University of Macau, Macao SAR, China

^dDivision of Physics and Applied Physics, Nanyang Technological University, 21 Nanyang Link, Singapore 637371

^eSchool of Materials Science and Engineering, Nanyang Technological University, Nanyang Avenue, Singapore 639798

Table S1: Reported Data for Perovskite large scale solar device as of 15-July-2016

Title	Size of active area	Efficiency	Architecture
Current Manuscript work ¹	70cm ²	10.74%	FTO/TiO ₂ /ZrO ₂ /Carbon/Perovskite (Drop Casted perovskite through Carbon layer)
Perovskite solar cells and large area modules (100 cm ²) based on an air flow-assisted PbI ₂ blade coating deposition process ²	10.1cm ²	10.4%	FTO/TiO ₂ /Perovskite/Spiro-OMeTAD/Au
	100cm ²	4.3%	
Interfacial modification of hole transport layers for efficient large-area perovskite solar cells achieved via blade-coating ³	6cm ²	10.15%	ITO/PEDOT:PSS(PSSH)/Perovskite/PCBM/Al
Large-area, high-quality organic–inorganic hybrid perovskite thin films via a controlled vapor–solid reaction ⁴	1.5cm ²	6.06%	ITO/PEDOT:PSS/Perovskite /PCBM/C60/BCP/Al
Upscaling of Perovskite Solar Cells: Fully Ambient Roll Processing of Flexible Perovskite Solar Cells with Printed Back Electrodes ⁵	Roll to roll	4.90%	ITO/PEDOT:PSS/Perovskite /PCBM/ZnO/Ag
Pinhole-free perovskite films for efficient solar modules ⁶	4cm ²	13.6%	ITO/TiO ₂ /Perovskite/Spiro-OMeTAD/Au
Efficient and stable large-area perovskite solar cells with inorganic charge extraction layers ⁷	1cm ²	16.20%	FTO/NiMgLiO/Perovskite /PCBM/Ti(Nb)Ox/Ag
Toward Large Scale Roll-to-Roll Production of Fully Printed Perovskite Solar Cells ⁸	Small device	11.96%	ITO/ZnO/Perovskite /P3HT/Ag
Solvent Engineering Boosts the Efficiency of Paintable Carbon-Based Perovskite Solar Cells to Beyond 14% ⁹	1cm ²	9.72%	FTO/TiO ₂ /Perovskite/Carbon
A vacuum flash–assisted solution process for high-efficiency large-area perovskite solar cells ¹⁰	1cm ²	19.60%	FTO/TiO ₂ /Perovskite/Spiro-OMeTAD/Au
Large-area hysteresis-free perovskite solar cells via temperature controlled doctor blading under ambient environment ¹¹	1cm ²	7.32%	ITO/PEDOT:PSS/Perovskite /PCBM/C60/BCP/Al

¹ Current Paper² Journal of Power Sources, 2015, Volume 277, Pages 286–291³ Solar Energy Materials and Solar Cells, 2016, Volume 144, 309–315⁴ J. Mater. Chem. A, 2016,4, 9124-9132⁵ Advanced Energy Materials, 2015, 201500569⁶ Energy Environ. Sci.,2016, 9, 484⁷ Science, 2015, 350, 944⁸ Advanced Material, 2015, 201404598⁹ Advanced Energy Materials, 2016,201502087¹⁰ Science, 2016 (DOI: 10.1126/science.aaf8060)¹¹ Applied Materials Today, 2016, Volume 3, 96–102

Table S2: Statistical device analysis for eighteen modules (Fig. 5a)

	Voc (V)	Jsc (mA/cm²strip)	Pmax (mW)	Fill Factor (%)	Efficiency (%)
Average	3.49	19.09	270.36	53.99	8.98
STD Dev	0.08	1.28	19.70	2.23	0.65
Max	3.72	21.39	315.00	57.96	10.46
Min	3.35	17.31	250.17	50.17	8.31

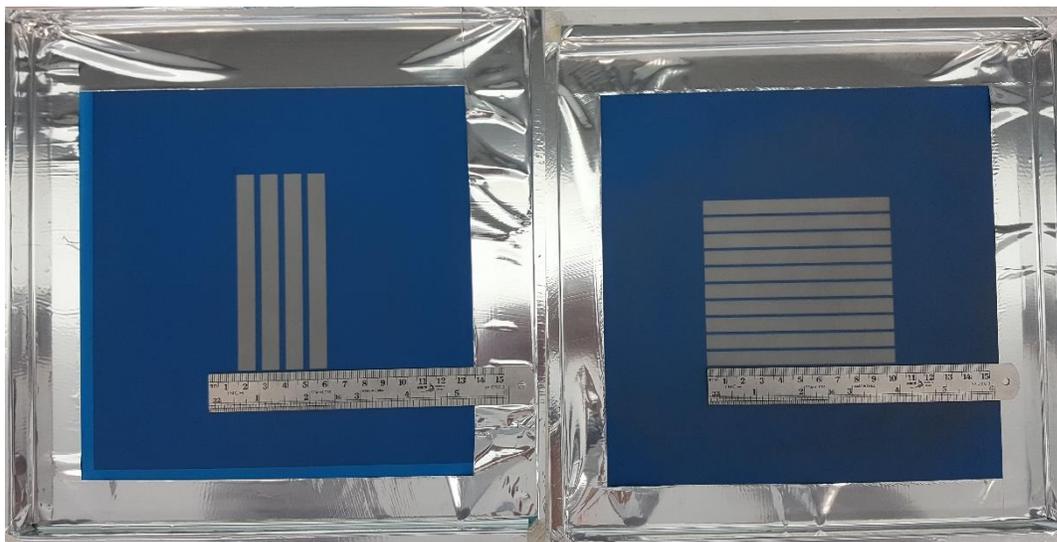
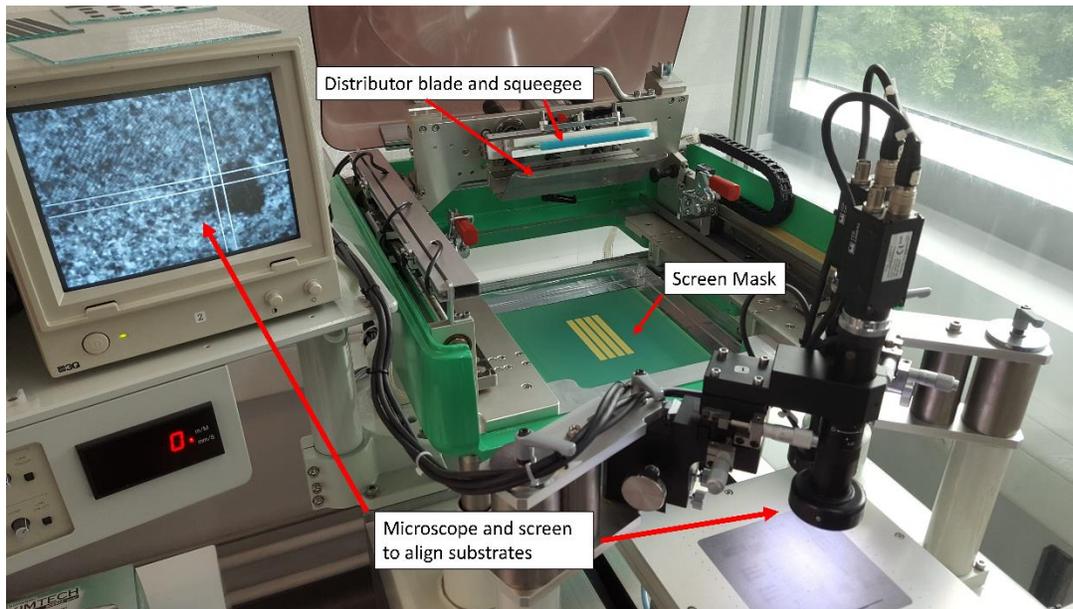


Fig S1: (a) Semi-automatic Screen Printer (MicroTec MT320TV); (b) Mask for 5cm x 10cm and 10cm x 10cm

(a)



(b)

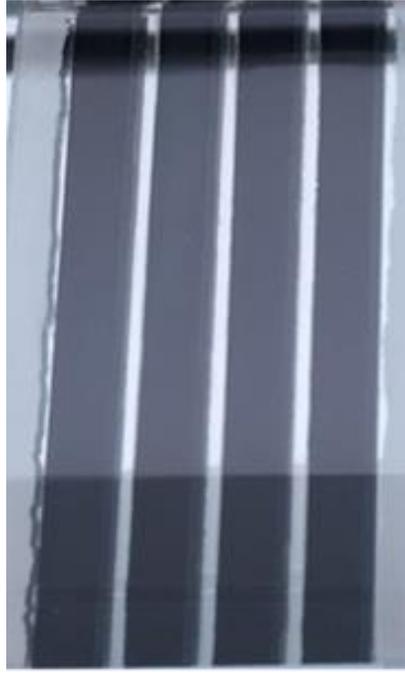


Fig. S2: Back side image of Perovskite solar module (a) unfilled or non-homogeneous mesoporous layer with perovskite crystals; (b) Uniformly filled mesoporous layers with perovskite crystals

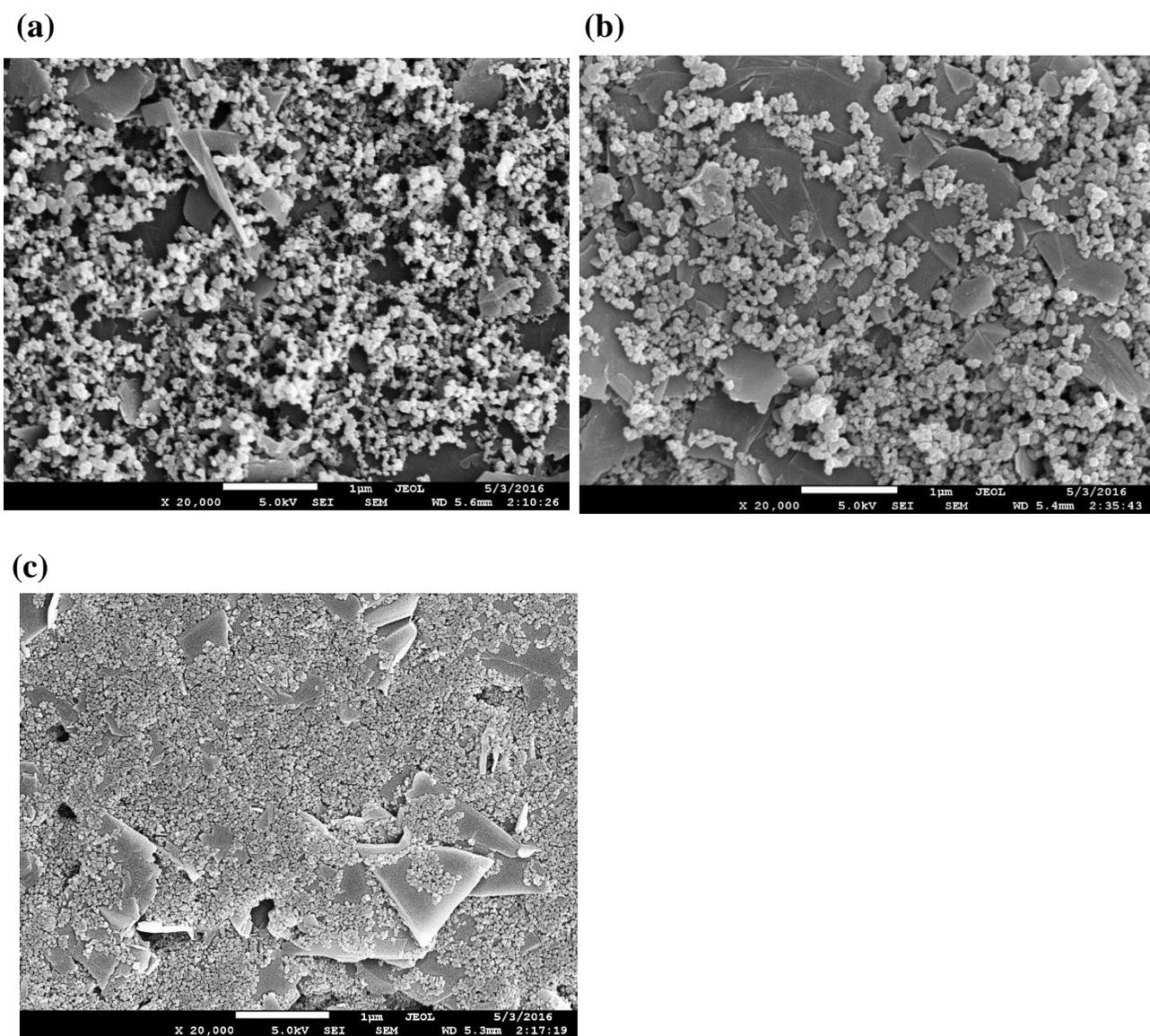
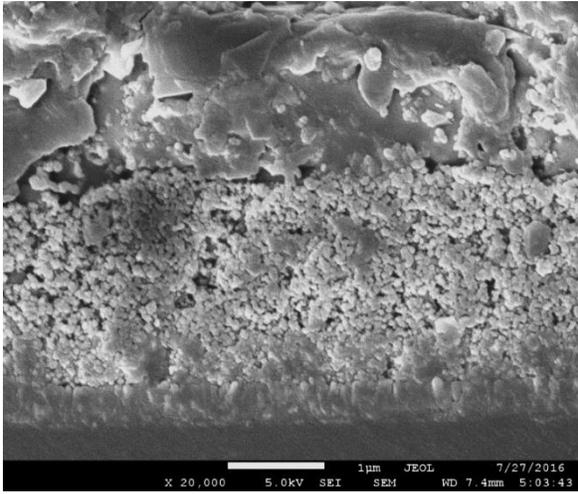


Fig. S3: High Resolution SEM image of carbon Film (a) Carbon 1 (b) Carbon 2 and (c) Carbon 3

(a)



(b)



(c)

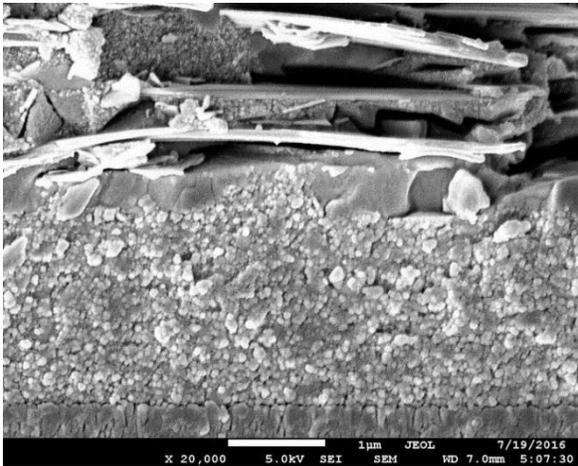


Fig. S4: Cross Section image of full device with infiltration of perovskite crystals (a) Carbon 1 (b) Carbon 2 and (c) Carbon3 as counter electrode

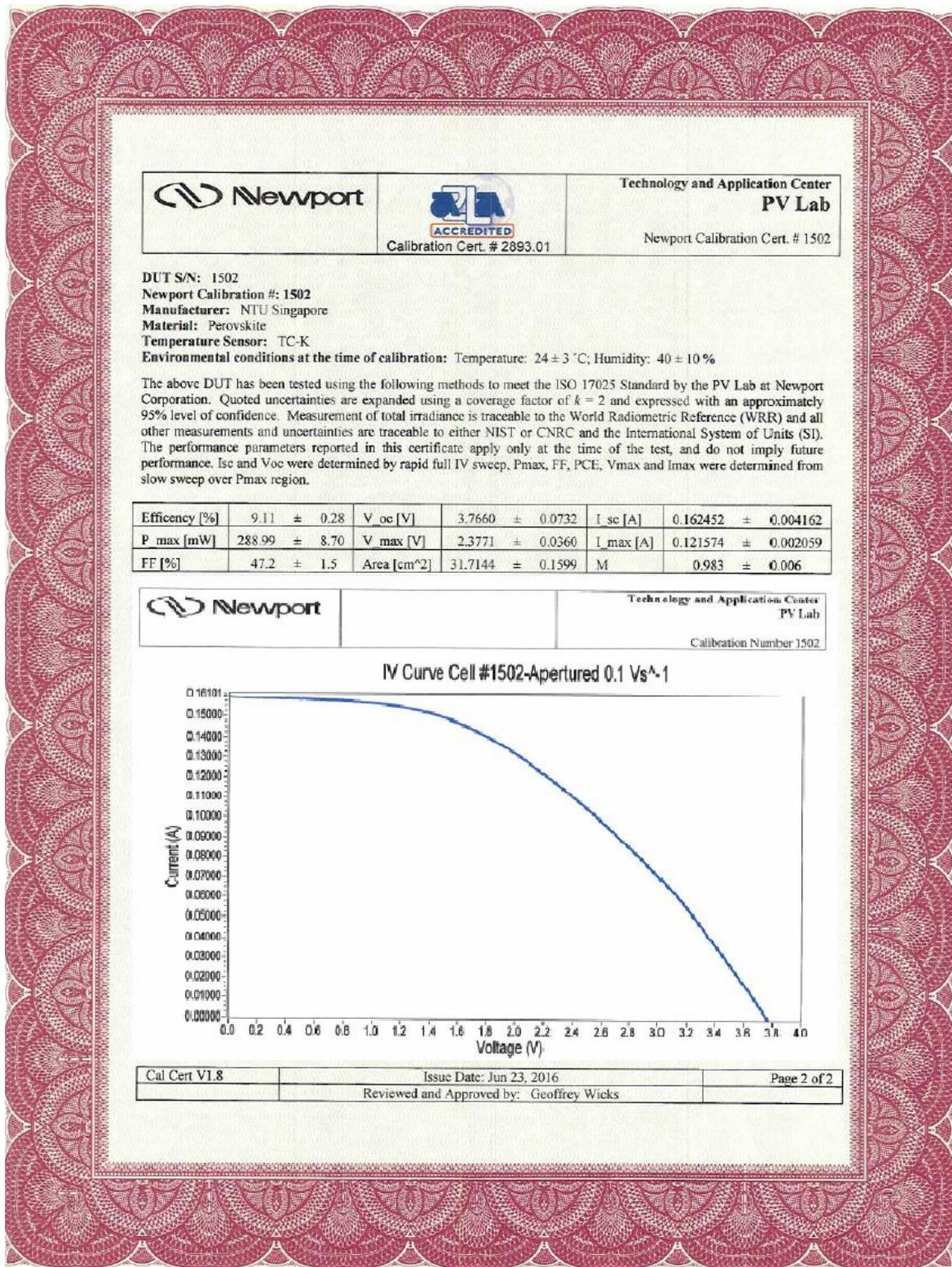


Fig. S5: Independent certification from Newport Corporation, confirming a PCE of 9.11 %.

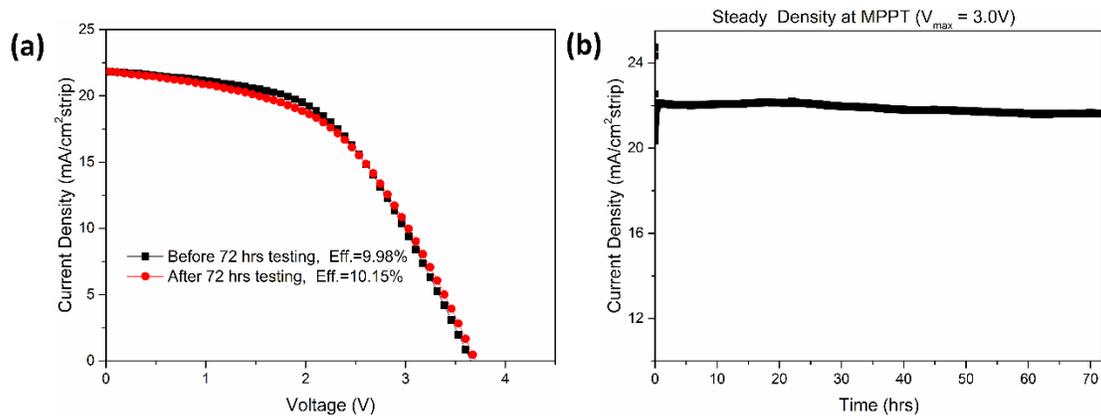


Fig. S6 (a) IV data measured under solar simulator before and after the steady state measurement tests; (b) Steady state current measurement at 2.3V for 3 days using a white light LED light source.

Experimental Section:

Fluorine doped tin oxide (FTO, ~14 ohm per square, 2.2 mm thick) substrates were first etched with CO₂ laser to form the desired pattern, which was cleaned with an ultrasonicator twice with decon soap solution, followed by deionized water (DI water) and finally ethanol. Each sonication process is held at 40°C for 30 mins. These substrates were then immersed in 50 mM of TiCl₄ solution (Wako Pure Chemical Industries, Ltd) for 30 min at 70°C, followed by rinsing with DI water. The substrates are held at 500°C for 30 mins to allow formation of TiO₂ crystals to serve as the seed layer. Using a screen-printer (MicroTec MT320TV, Fig s5a), a layer of compact TiO₂ paste (Dyesol BL-1) is printed onto the substrate using screen mask (Fig s5b). The printed film is allowed to relax for 20 mins before calcined at 500°C for 30 mins. These substrates were finally immersed in 100 mM of TiCl₄ solution (Wako Pure Chemical Industries, Ltd) for 30 min at 70°C to fill up the defective “pin-holes”, followed by rinsing with DI water. A final heat treatment of 500°C completes the formation of the blocking layer.

A 500nm thick layer of mesoporous TiO₂ film is printed onto the substrate with a TiO₂ paste (Dyesol NRD-30, diluted with terpineol (Sigma, FG) in weight ratio 1:1.4). Then, the substrates are sintered for 30 mins at 500°C to improve the crystallinity. Subsequently, a ZrO₂ (Solaronix, Zr-Nanoxide ZT/SP) spacer layer is printed on top of the meso-TiO₂ layer and heated to 500°C for 30 mins to remove the binders. Finally, carbon paste was printed on top of ZrO₂ to complete the device stack. The substrates were heated at 400°C for 30 mins.

Three different types of carbon paste are being prepared. For carbon paste (1), 1.5g of graphite (Sigma Aldrich, Part number: 282863) is mixed with 4.4g of ethyl cellulose (Sigma, mixed with ethanol in a 1:10 weight ratio), 0.5g of Super P carbon (TimCal Ltd, part number: H30253) in 10ml of α -terpineol (Sigma, FG). This mixture was then ball milled for 2 hours through 8 cycle repetitions of 10 mins ball milling at 300rpm and resting for 5 mins.

A micropipette (of maximum capacity 10 μ l) is used to control the amount of perovskite solution. For a width of 7.75mm, 7.5 μ l of solution is dispensed from the micropipette. The volume is dispensed as 5 drops, making each drop approximately 1.5 μ l. This amount of solution is sufficient to wet approximately 2.5 to 3cm of an individual strip length. The same process is repeated 16 times to ensure complete coverage of the whole 5cm x 10cm module (4 strips) by perovskite

solution. The amount of perovskite solution and corresponding droplet size is adjusted accordingly to the change in strip width.

For carbon paste (2), 1.5g of graphite (Sigma Aldrich, Part number: 282863) is mixed with 1.3g of ethyl cellulose (Sigma, mixed with ethanol in a 1:10 weight ratio), 0.5g of meso-carbon (Sigma, part number:699624) in 5ml of α -terpineol (Sigma, FG). This mixture was then ball milled for 4 hours through 16 cycle repetitions of 10 mins ball milling at 350rpm and resting for 5 mins.

Rotary Evaporation is used to remove ethanol from both pastes.

Perovskite solution is obtained by mixing equimolar ratio of PbI_2 and $\text{CH}_3\text{HN}_3\text{I}$ (Tokyo Chemical Limited) in γ -Butyrolactone (Sigma, $\geq 99\%$). 5 AVA-I prepared by reacting 5-aminovaleric acid (Sigma) with hydriodic acid (Sigma) in equimolar ratio, followed by filtering and purification. 5 AVA-I was added to the perovskite solutions in molar ratio of 1:20 (AVA-I to MAI). The solutions are kept stirring overnight on a 50°C hotplate for complete dissolution of the precursors. Perovskite solution is dropped on the carbon layer using a micropipette.

The FE-SEM (Jeol JSM-7600F Field Emission Scanning Electron Microscope) is used to obtain the morphology and cross-section imageries of the devices. I-V characteristics of the devices are measured using the solar simulator (San-EI Electric, XEC-301S), which is connected to a Keithley 2612A SourceMeter. The Quantachrome NOVA 3200 BET setup is used to determine the surface area of the printed film. Carbon films were prepared by printing on a glass substrate, annealed, and scrapped off to measure the surface area of the printed film.

A four-probe tester (Advanced Instrument Technologies CMT-SR2000N) was used to carry out experiments on a printed carbon film of dimensions $3\text{cm}\times 5\text{cm}$ for their conductivity measurement. Pin type is an in-line 4 probe electrode with spacing of 1mm between each pin. Current limit is set to be automatic, while current direction is set to flow in both directions. The average film thickness was input into the software for the measurement of sheet resistance of the carbon film, and the reported values are averaged results of 8 data points. FTO measured in a similar manner had a sheet resistance of $14.23 \Omega/\text{sq}$.

Steady state power tracking was done for 3 days on a $5\text{cm}\times 10\text{cm}$ perovskite module. The current was monitored using Autolab machine (PGSTAT302N, Software version- NOVA 1.11) with a

white light LED light source (Model number: ALFL-100, rated @ 100W, CCT of 6000K, Manufactured by Arianetech Pte Ltd, Singapore). The intensity of the light received by the solar module is adjusted by varying the distance of the module and the light source. The distance and the applied voltage (2.3V) were fixed to yield similar currents as the maximum power point current measured under the solar simulator. All measurement was done in ambient (Temperature $\approx 25^{\circ}\text{C}$, and humidity $\approx 65\%\text{RH}$).