

1 **Synthesis of three crystalline forms of the Al<sub>2</sub>O<sub>3</sub> featuring rod-like fibers and their effect**

2 **to gas degradation of 1-chloronaphthalene**

3 Huijie Lu,<sup>abc</sup> Qianqian Li,<sup>ab</sup> Guijin Su,\*<sup>ab</sup> Minghui Zheng,<sup>ab</sup> Yuyang Zhao,<sup>ab</sup> Xue Miao,<sup>ab</sup> Yalu

4 Liu,<sup>ab</sup> Xinchen Huang<sup>ab</sup> and Yanhui Zhao<sup>ab</sup>

5 <sup>a</sup>State Key Laboratory of Environmental Chemistry and Ecotoxicology, Research Center for

6 Eco-Environmental Sciences, Chinese Academy of Sciences, P.O. Box 2871, Beijing 100085,

7 China;

8 <sup>b</sup>University of Chinese Academy of Sciences, Beijing 100049, China;

9 <sup>c</sup>Beijing Environmental Sanitation Engineering Research Institute, Jia no. 48 Shangjialou,

10 Chaoyang District, Beijing 100028, China

11 \*Corresponding author: Dr. Guijin Su; Tel: +86 10 62849356; Fax: + 86 10 62923563; E-mail

12 address: [gjsu@rcees.ac.cn](mailto:gjsu@rcees.ac.cn) (G. Su).

13 **List of Supporting Information Contents:**

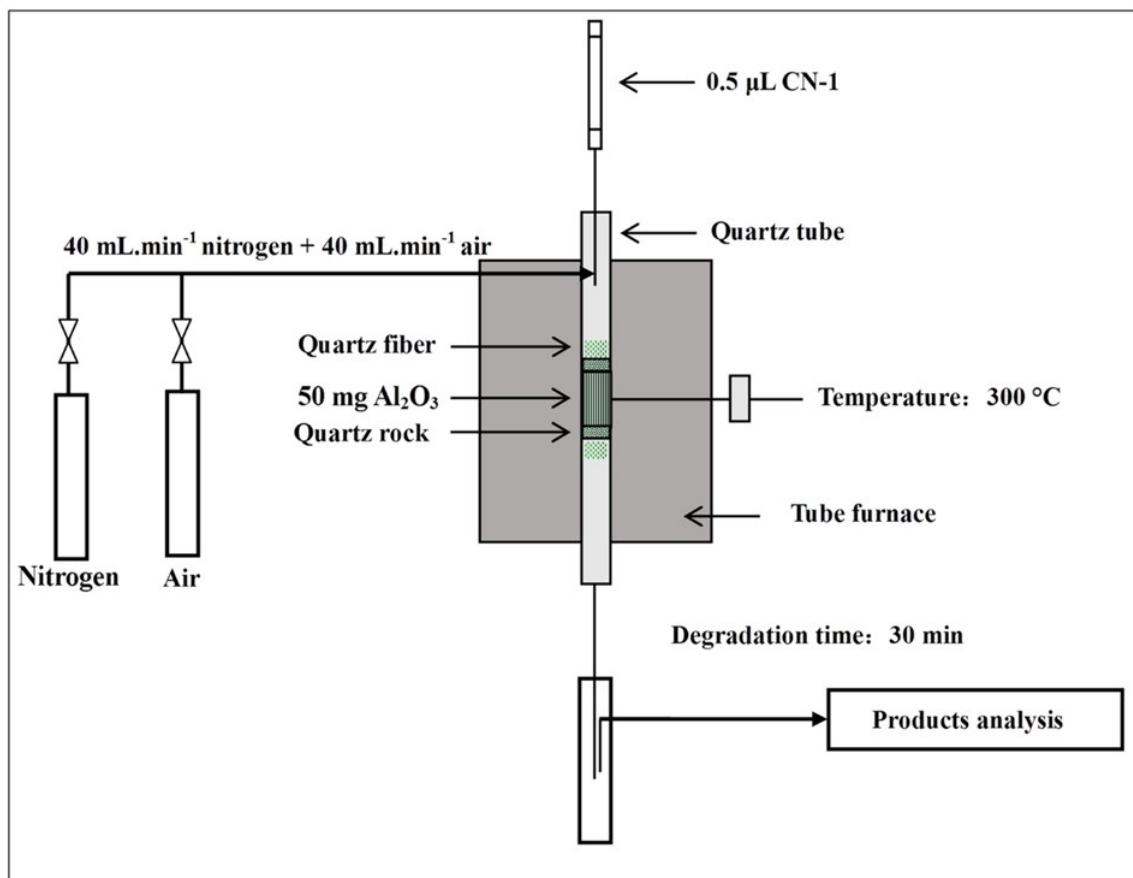
14 ➤ **Fig. S1** Schematic of the pulsed flow microreactor used in the present study.

15 ➤ **Table S1** The mass-to-charge ratios (*m/z*) of PCNs.

16 ➤ **Table S2** Physical properties of the prepared Al<sub>2</sub>O<sub>3</sub> materials.

17 ➤ **Table S3** Acid site assignments of the IR bands (cm<sup>-1</sup>) observed upon adsorption of NH<sub>3</sub>

18 on the three crystalline Al<sub>2</sub>O<sub>3</sub> samples.



19

20

21 **Fig. S1** Schematic of the pulsed flow microreactor used in the present study.

22

PCN Congeners	m/z	Type of m/z
Naphthalene	128	M
	102	M-C <sub>2</sub> H <sub>2</sub>
MoCN	162	M
	164	M+2
DiCN	196	M
	198	M+2

23 **Table S1** The mass-to-charge ratios (m/z) of PCNs.

24

Materials	S <sub>BET</sub> (m <sup>2</sup> /g)	Total pore volume (cm <sup>3</sup> /g)	Average pore diameter (nm)	O <sub>a</sub> (%)
$\alpha$ -Al <sub>2</sub> O <sub>3</sub>	5.2	0.025	3.39	14.0
$\eta$ -Al <sub>2</sub> O <sub>3</sub>	34.9	0.063	3.84	10.5
$\gamma$ -Al <sub>2</sub> O <sub>3</sub>	135.0	0.199	3.83	31.3

25 **Table S2** Physical properties of the prepared Al<sub>2</sub>O<sub>3</sub> materials.

26

Al <sub>2</sub> O <sub>3</sub>	Vibrational feature (cm <sup>-1</sup> )						
samples	Lewis acid sites				Brønsted acid sites		
$\alpha$ -Al <sub>2</sub> O <sub>3</sub>		1587					
$\eta$ -Al <sub>2</sub> O <sub>3</sub>	1625	1587	1255	1172		1480	1453
$\gamma$ -Al <sub>2</sub> O <sub>3</sub>	1625	1587	1255	1172	1682	1480	1453

27 **Table S3** Acid site assignments of the IR bands (cm<sup>-1</sup>) observed upon adsorption of NH<sub>3</sub> on

28 the three crystalline Al<sub>2</sub>O<sub>3</sub> samples.