

Supporting Information

Visible-light-activated Copper(I) catalyzed Oxidative C_{sp}-C_{sp} Cross-Coupling Reaction: Efficient Synthesis of Unsymmetrical Conjugated Diynes without Ligands and Base

Arunachalam Sagadevan,¹ Ping-Chiang Lyu,² and Kuo Chu Hwang^{1*}

¹Department of Chemistry, ² Institute of Bioinformatics and Structural Biology and College of Life Sciences, National Tsing Hua University, Hsinchu, Taiwan, R. O. C. E-mail: kchwang@mx.nthu.edu.tw

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Experimental section

General: All reactions were conducted under 1 atm. of an oxygen atmosphere and oven-dried glass wares were used. All reactions were conducted using a blue light-emitting diode (LED) as the visible-light source (30 lamps, power density: 40 mW/cm² at 460 nm). All solvents were dried according to the known methods, and distilled prior to use. Starting materials (including starting materials for synthesis of epoxide hydrolase inhibitors) were commercially available (Sigma-Aldrich, Alfa-Aesar or TCI-chemicals) and used as received. NMR spectra were recorded ¹H NMR at 400 MHz and ¹³C NMR at 100 MHz using deuterated CDCl₃ as a solvent. Chemical shifts (δ) were reported as parts per million (ppm). The following abbreviations were used to identify the multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, b = broad and all combinations thereof can be explained by their integral parts. Unless otherwise specified, the proton/carbon signal of 2 residual solvent (at δ 7.24 and δ 77.00 ppm, respectively) was used as the internal reference.

General procedure:

A dry test tube (20 mL) with rubber septum and magnetic stirrer bar was charged with 1.2 -1.4 mmol of **1a** (0.6-0.7 M) terminal alkynes, 0.7 mmol of **2a** (0.35 M), and 5 mol% (relative to **1a**) CuCl is combined in CH₃CN (2 mL). The mixture was irradiated with blue LEDs under an oxygen (balloon) atmosphere for hours until completion of heterocoupling reaction (it was determined by thin layer chromatography). The reaction mixture was diluted with 40 % ethyl acetate in hexane and stirred in for 10 min. The mixture was filtered through celite and silica gel pads, and washed with ethyl acetate. The filtrate was concentrated and the residue was purified by flash column chromatography on silica gel to afford the desired heterocoupling products.

Evaluation of Green metrics of the process;

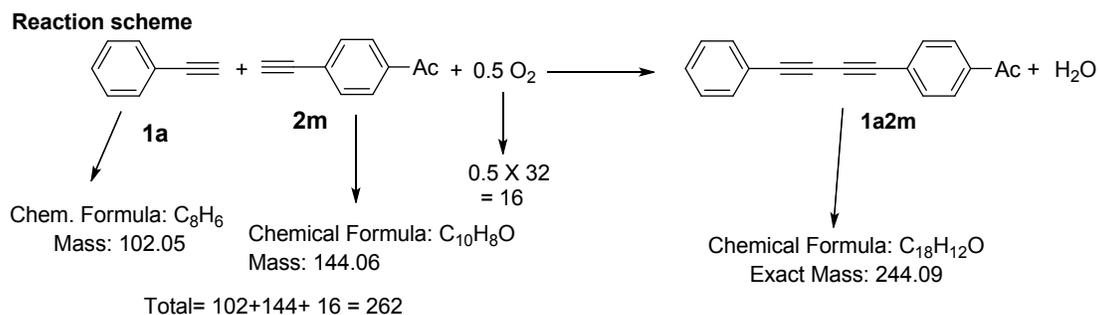
Atom economy defined as “how much of the reactants remain in the final desired product”

$$\text{Atom economy (AE)} = \frac{\text{Molecular mass of desired product}}{\text{Molecular mass of all reactants}} \times 100$$

Reaction mass efficiency (RME) defined as “the percentage of the mass of the reactants that remain in the product”

$$\text{Reaction mass efficiency (RME)} = \frac{\text{mass of desired product}}{\text{mass of all reactants}} \times 100$$

a) Evaluation of Green metrics of the current photochemical process



Evaluation of Green metrics of the current photochemical process

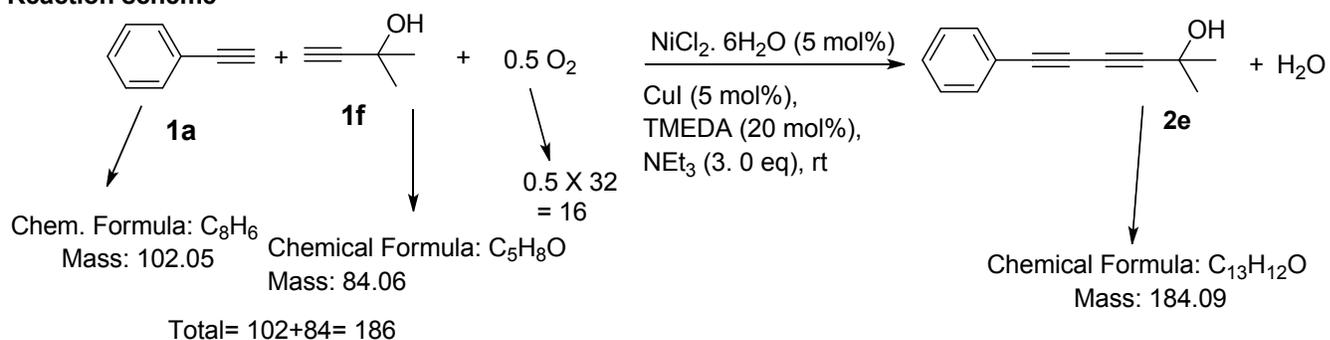
Reactant A	phenylacetylene (1a)	0.95g	0.0093 mol	FW 102.05
Reactant B	1-(4-ethynylphenyl)ethanone (2m)	0.75g	0.0052 mol	FW 144.06
Solvent	ACN	11.79g	---	---
Auxiliary	---	---	---	---
Product	1,3-diynes (1a2m)	1.06g	0.0043 mol	FW 244.09

Product yield= 84%

$$\begin{aligned}
 \text{E-factor} &= \frac{0.95\text{g} + 0.75\text{g} + 11.79\text{g} - 1.06\text{g}}{1.06\text{g}} = 11.7 \text{ kg waste/1 kg product} \\
 \text{Atom economy} &= \frac{244}{262} \times 100 = 93\% \\
 \text{Atom efficiency} &= 84\% \times 93\% / 100 = 78\% \\
 \text{Carbon efficiency} &= \frac{18}{8 + 10} \times 100 = 100\% \\
 \text{Reaction mass efficiency} &= \frac{1.06\text{g}}{0.95\text{g} + 0.75\text{g}} \times 100 = 63\%
 \end{aligned}$$

b) Evaluation of Green metrics of the reported process ^{s1}

Reaction scheme^{s1}



Evaluation of Green metrics of the reported process

Reactant A	phenylacetylene (1a)	0.5g	0.005m ol	FW 102.05
Reactant B	2-methylbut-3-yn- 2-ol (2m)	0.084g	0.001m ol	FW 84.06
Solvent	THF	3.55g	---	---
Auxiliary	TMEDA	0.023g	0.0002	116.21
Base	NEt ₃	0.303g	0.003	101.19
Product	1,3-diynes (2e)	0.158g	0.00086 mol	FW 184.09

Product yield= 86%

TMEDA: *Tetramethylethylenediamine*

$$\begin{aligned}
 \text{E-factor} &= \frac{0.5\text{g} + 0.084\text{g} + 3.55\text{g} + 0.023\text{g} + 0.303\text{g} - 0.158\text{g}}{0.158\text{g}} = 27\text{kg waste/1 kg product} \\
 \text{Atom economy} &= \frac{184}{186 + 101 + 23.2 + 16} \times 100 = 56\% \\
 \text{Atom efficiency} &= 86\% \times 56\% / 100 = 48\% \\
 \text{Carbon efficiency} &= \frac{13}{8 + 5} \times 100 = 100\% \\
 \text{Reaction mass efficiency} &= \frac{0.158\text{g}}{0.5\text{g} + 0.084\text{g}} \times 100 = 27\%
 \end{aligned}$$

EPR measurements: EPR spectra were recorded at room temperature on a Bruker ESP-300E(X band, 9.8 GHz) with parameters setting as shown below: receiver gain= 30 n; receiver phase= 0 deg; receiver harmonic= 1; field modulation frequency= 100000 Hz; microwave frequency [Hz]= 9.660469 e+09; field modulation amplitude [T]= 0.00016; receiver time constant [S] = 0.32768; microwave power= 0.015 W; receiver offset [%FS]= 0; DMPO (5-,5-dimethyl-1-pyrroline N-oxide) was employed as a radical trap for trapping of the superoxide radical anion.

The reaction under an standard condition (**1a**, CuCl, 1 atm. O₂) in CH₃CN was irradiated with blue LEDs for 30 min in the presence of DMPO in an EPR chamber while recording the EPR spectra. The EPR signals shown in Figure S1 is corresponding to DMPO-OO(H). This result

indicates that superoxide anion radical was formed in the reaction solution. No superoxide EPR signal was observed from the reaction solution under standard condition in the absence of CuCl (Fig. S2). These results indicate that copper(I) phenylacetylide undergoes single electron transfer to O₂, and generate superoxide free radical upon blue LEDs irradiation.

EPR spectra of the reaction mixture after blue LEDs irradiation

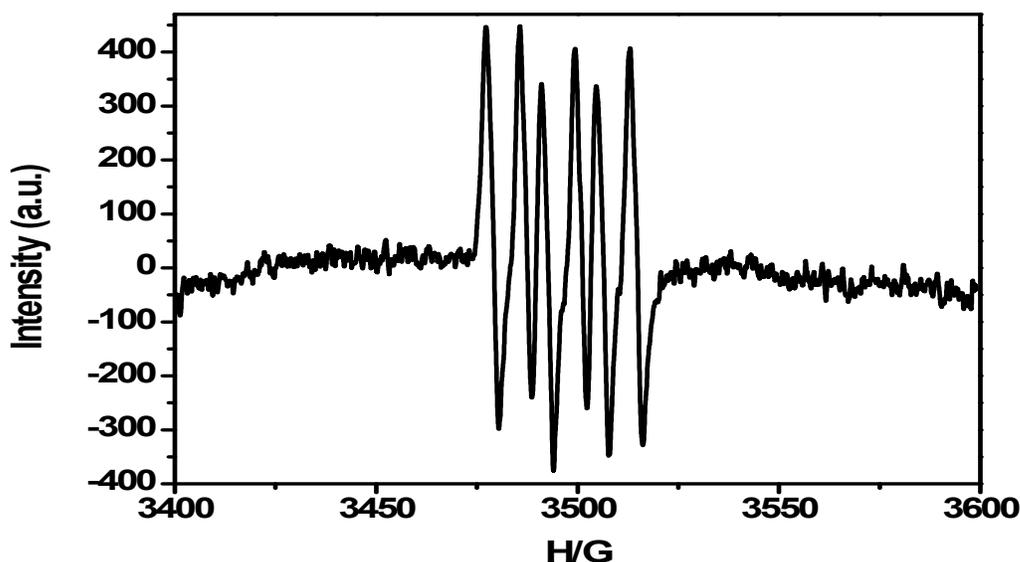


Figure S1: EPR spectra of the reaction mixture: phenylacetylene(**1a**) (0.6 M), and 5 mol% of CuCl in CH₃CN, 0.5 ml of this reaction solution was taken out into a small vial, followed by the addition of 0.01 ml of DMPO (5×10^{-2} M). The mixture was irradiated with blue LEDs at room temperature under an oxygen atmosphere (1 atm) for 30 minutes. The reaction mixture was then analysed by EPR spectra. There are classical 6 peaks, the signals corresponding to (DMPO-OO(H)).

EPR spectra of the reaction mixture in the absence of CuCl

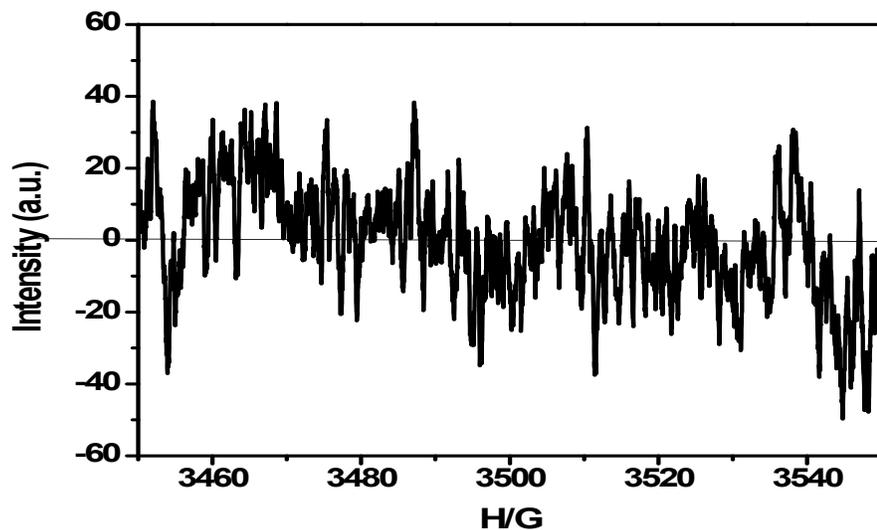


Figure S2: EPR spectra of the reaction mixture: phenylacetylene(**1a**) (0.6 M), 0.5 mL of this reaction solution was taken out into a small vial, followed by the addition of 0.01 mL of DMPO (5×10^{-2} M). The mixture was irradiated with blue LEDs at room temperature under an oxygen atmosphere (1 atm) for 30 minutes (in the absence of CuCl). The reaction mixtures was analysed by EPR spectra. No signals were detected.

Estimation of association constant using isothermal titration calorimetry (ITC):

Experiment 1: Isothermal titration calorimetry experiments were carried out at 25 °C on a high precision ITC-200 (MicroCal, LLC, and Northampton, MA). The solution of copper(I) phenylacetylide (Cu(I)-**1a**, 1 mM) and (1-(4-ethynylphenyl) ethanone, **2m**) (2 mM) were prepared by using CH₃CN as a solvent. Before the measurements, samples were degassed for at least 7 minutes. The calorimeter was initially calibrated using water-water titration, in which the reference power of 5 µcal/s was applied. As a control experiment, solvent-to-solvent titration was also performed. Then, the Cu(I)-**1a** solution was loaded into the cell and **2m**-acetonitrile solution was taken in the syringe. 20 injections were performed with an each titration volume of 2 µL. The reference power of 5 µcal/s was applied while the sample contents were stirred at 1000 rpm (rotations per minute).

The binding curve is obtained from a plot of the heat change from each injection against the molar ratio of **2m** (in syringe) and binding partner Cu(I)-phenylacetylide (Cu(I)-**1a**) in the cell (Fig. S1). The binding curve is analyzed with an appropriate binding model to determine the value of K (binding affinity). The isothermal titration reveals the association constant, $K_a \sim 199 \mu\text{M}^{-1}$ and this affinity value suggests a moderate extent of interaction.

Likewise, a **1a**-acetonitrile solution was titrated into a Cu(I)-**1a** acetonitrile solution. No clear complex formation was observed from the ITC measurements in this case (see, results shown in supplementary Figure S4).

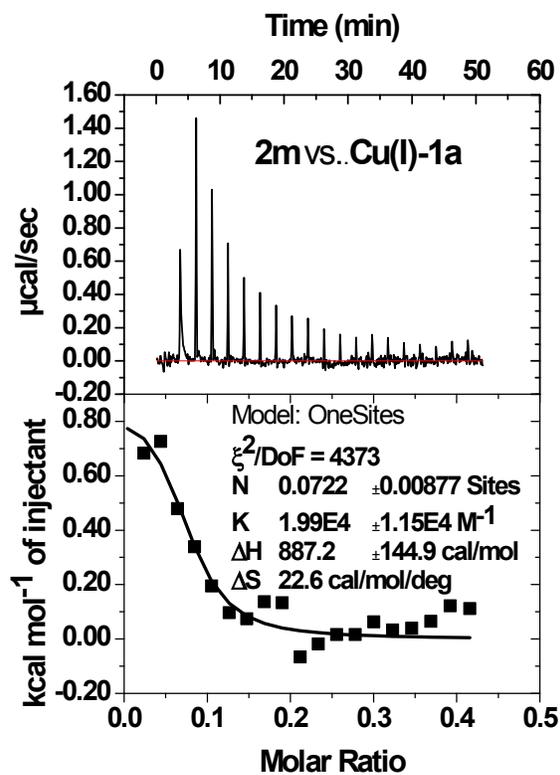


Figure S3. Isothermal titration calorimetry (ITC) data for the determination of the association constant values for combination Cu(I)-phenylacetylide (Cu(I)-1a) and **2m** (1-(4-ethynylphenyl) ethanone). The inset of the bottom panel indicates the peak fitting results of one set of binding sites obtained from the inbuilt Origin Pro software of the Microcal ITC-200.

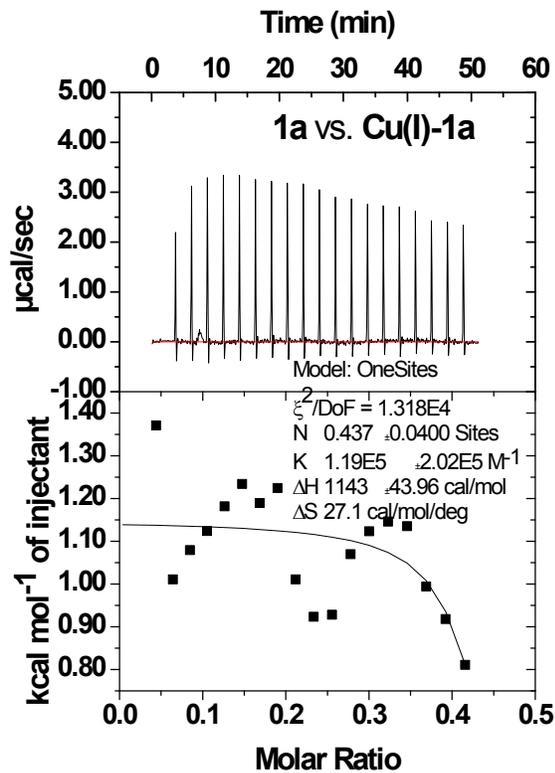
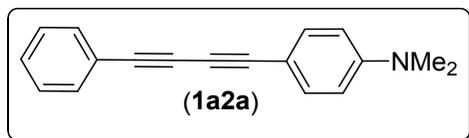


Figure S4. Isothermal titration calorimetry (ITC) data for the determination of the association constant values for combination copper(I) phenylacetylde (Cu(I)-**1a**) and phenylacetylene (**1a**). The inset of the bottom panel indicates the peak fitting results of one set of binding sites obtained from the inbuilt Origin Pro software of the Microcal ITC-200

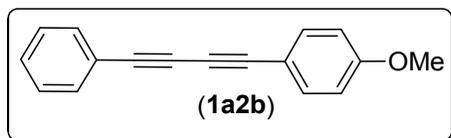
Spectroscopic data

N, N-dimethyl-4-(phenylbuta-1, 3-diyne-1-yl) aniline (1a2a)^{s2}



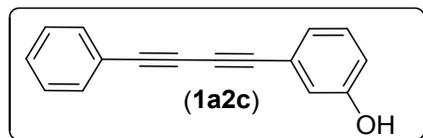
Pale yellow solid; m.p.= 110-114°C; ¹H NMR (400 MHz, CDCl₃): δ 7.50-7.48 (m, 2 H), 7.40-7.38 (m, 2 H), 7.32-7.30 (m, 3 H), 6.60 (d, *J*= 4.0 Hz, 2 H), 2.98 (s, 6 H); ¹³C NMR (100 MHz, CDCl₃): δ 150.5, 133.8, 132.2, 128.6, 128.3, 122.3, 111.6, 107.8, 83.4, 80.7, 74.7, 72.0, 40.0; HRMS calcd for C₁₈H₁₅N: 245.1204, found: 245.1207.

1-methoxy-4-(phenylbuta-1, 3-diyne-1-yl) benzene (1a2b)^{s3}



Colorless solid; m.p.= 90-94 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.51-7.49 (m, 2 H), 7.47-7.44 (m, 2 H), 7.33-7.31 (m, 3 H), 6.85-6.83 (m, 2 H), 3.80 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃): δ 160.3, 134.1, 132.4, 129.0, 128.3, 121.9, 114.1, 113.6, 81.8, 81.0, 74.1, 72.7, 55.3; HRMS calcd for C₁₇H₁₂O: 232.0888, found: 232.0890.

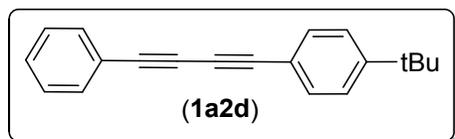
3-(phenylbuta-1,3-diyne-1-yl)phenol (1a2c)^{s4}



Brown solid; m.p.=119-123°C; ¹H NMR (600 MHz, CDCl₃): δ 7.52-7.50 (m, 2 H), 7.32-7.35 (m, 3H), 7.19 (t, *J*= 7.8 Hz 1H), 7.10-7.09 (m, 1H), 6.97-6.96 (m, 1H), 6.85-6.83 (m, 1H), 5.28 (bs, 1 H); ¹³C NMR (150 MHz, CDCl₃): δ 155.3, 132.4, 129.7, 129.2, 128.4, 125.2, 122.9, 121.6,

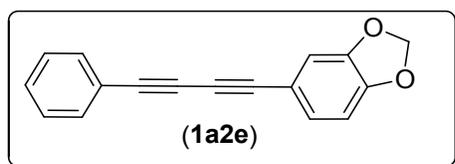
118.9, 116.8, 81.6, 81.1, 73.87 and 73.81; **HRMS** calcd for C₁₆H₁₀O: 218.0732, found: 218.0726.

1-(tert-butyl)-4-(phenylbuta-1, 3-diyn-1-yl) benzene (1a2d)



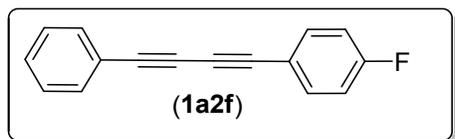
Colorless solid; m.p.=78-82°C, ¹H NMR (400 MHz, CDCl₃): δ 7.54-7.52 (m, 2 H), 7.48-7.46 (m, 1 H), 7.37-7.33 (m, 6 H), 1.31 (s, 9 H); ¹³C NMR (100 MHz, CDCl₃): δ 152.6, 132.4, 132.2, 129.1, 128.4, 125.4, 121.7, 118.6, 81.8, 81.1, 74.1, 73.2, 34.8, 31.0; **HRMS** calcd for C₂₀H₁₈: 258.1409, found: 258.1412.

5-(phenylbuta-1,3-diyn-1-yl)benzo[d][1,3]dioxole (1a2e)^{s3}



Colorless solid; m.p.= 110-114°C, ¹H NMR (600 MHz, CDCl₃): δ 7.50-7.49 (m, 2 H), 7.34-7.30 (m, 3 H), 7.07-7.05 (m, 1 H), 6.94-6.93 (m, 1 H), 6.75 (d, J= 8.1, 1 H); ¹³C NMR (150 MHz, CDCl₃): δ 148.8, 147.4, 132.4, 129.1, 128.4, 127.7, 121.9, 114.9, 112.1, 108.6, 101.5, 81.6, 81.1, 74.0, 72.4; **HRMS** calcd for C₁₇H₁₀O₂: 246.0681, found: 246.0677.

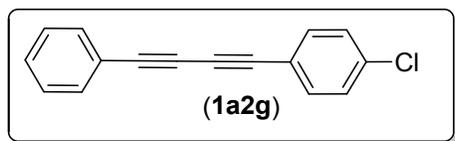
1-fluoro-4-(phenylbuta-1, 3-diyn-1-yl) benzene (1a2f)^{s6}



Colorless solid; m.p. 121-125 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.52-7.47 (m, 4 H), 7.36-7.30 (m, 3 H), 7.04-6.99 (m, 2 H); ¹³C NMR (100 MHz, CDCl₃): δ 164.3, 164.2, 161.8, 161.7, 134.5,

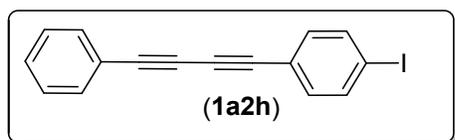
134.48, 134.47, 132.4, 129.1, 128.44, 128.43, 121.7, 121.6, 116.0, 115.9, 115.79, 115.77, 81.5, 80.4, 73.9, 73.7; **HRMS** calcd for C₁₆H₉F: 220.0688, found: 220.0689.

1-chloro-4-(phenylbuta-1,3-diyn-1-yl)benzene (1a2g)^{s7}



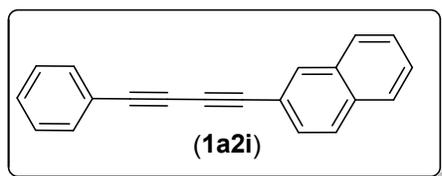
Yellow oil; m.p.= 131-135°C; **¹H NMR** (400 MHz, CDCl₃): δ 7.51-7.50 (m, 2 H), 7.44-7.42 (m, 2 H), 7.36-7.29 (m, 5H); **¹³C NMR** (100 MHz, CDCl₃): δ 135.3, 133.6, 132.5, 129.3, 128.8, 128.4, 121.5, 120.3, 82.1, 80.2, 74.8, 73.6; **HRMS** calcd for C₁₆H₉Cl: 236.0393, found: 236.0401.

1-iodo-4-(phenylbuta-1,3-diyn-1-yl) benzene (1a2h)^{s7}



Yellow oil; m.p.= 138-142°C; **¹H NMR** (400 MHz, CDCl₃): δ 7.67-7.65 (m, 2 H), 7.52-7.50 (m, 2 H), 7.36-7.30 (m, 3 H), 7.22 (d, *J*= 8.0 Hz, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 137.6, 133.7, 132.4, 129.3, 128.4, 121.5, 121.2, 95.4, 82.2, 80.4, 75.2, 73.6; **HRMS** calcd for C₁₆H₉I: 327.9749, found: 327.9782.

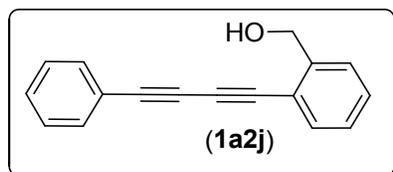
2-(phenylbuta-1,3-diyn-1-yl)naphthalene (1a2i)^{s7}



Yellow solid; m.p.= 97-101°C; **¹H NMR** (400 MHz, CDCl₃): δ 8.04-7.75 (m, 5H), 7.54-7.46 (m, 3 H) 7.36-7.30 (m, 4H); **¹³C NMR** (100 MHz, CDCl₃): δ 133.1, 133.0, 132.7, 132.5, 132.4,

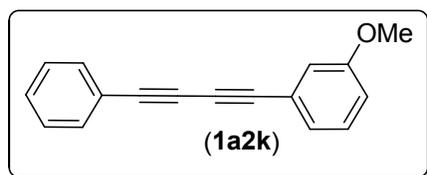
129.2, 128.4, 128.1, 127.8, 127.2, 126.7, 121.7, 118.9, 82.0, 81.7, 74.2; **HRMS** calcd for C₂₀H₁₂: 252.0939, found: 252.0951.

(2-(phenylbuta-1, 3-diyn-1-yl) phenyl) methanol (1a2j)



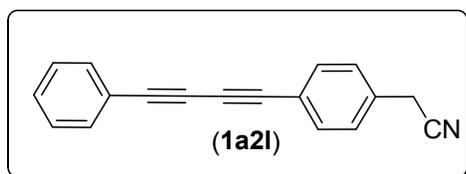
Yellow oil; m.p.= 82-86 °C; **¹H NMR** (400 MHz, CDCl₃): δ 7.54-7.50 (m, 4 H), 7.48-7.33 (m, 4 H), 7.29-7.25 (m, 1 H), 4.87 (s, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 143.9, 133.2, 132.4, 129.5, 129.3, 128.4, 127.4, 127.2, 121.5, 119.7, 82.7, 78.8, 78.2, 73.6, 63.5; **HRMS** calcd for C₁₇H₁₂O: 232.0888, found: 232.0890.

1-methoxy-3-(phenylbuta-1, 3-diyn-1-yl) benzene (1a2k)^{s8}



White solid; m.p.= 50-54 °C; **¹H NMR** (400 MHz, CDCl₃): δ 7.55-7.52 (m, 2 H), 7.38-7.34 (m, 3 H), 7.27-7.05 (m, 3 H), 6.95-6.92 (m, 1 H); **¹³C NMR** (100 MHz, CDCl₃): δ 159.2, 132.4, 129.5, 129.2, 128.4, 125.0, 122.6, 121.6, 117.0, 115.9, 81.5, 81.4, 73.8, 73.6, 55.2; **HRMS** calcd for C₁₇H₁₂O: 232.0888, found: 232.0891.

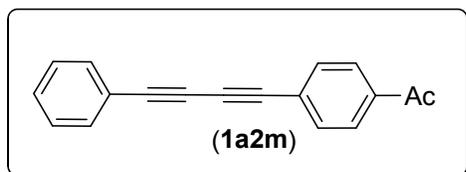
1-(4-(phenylbuta-1,3-diyn-1-yl)phenyl)ethanone (1a2l)



Pale white solid; m.p.= 77-81 °C; **¹H NMR** (600 MHz, CDCl₃): δ 7.52-7.50 (m, 4 H), 7.36-7.28 (m, 5 H), 3.75 (s, 2 H); **¹³C NMR** (150 MHz, CDCl₃): δ 133.1, 132.5, 130.8, 129.3, 128.4, 128.0,

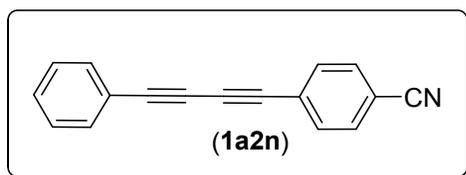
121.8, 121.5, 117.1, 82.0, 80.4, 74.7, 73.6 23.6; **HRMS** calcd for C₁₈H₁₁N: 241.0891, found: 241.0893.

1-(4-(phenylbuta-1, 3-diyn-1-yl) phenyl)ethanone (1a2m)



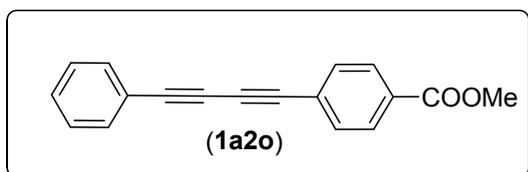
Pale white solid; m.p.= 111-115 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.90 (d, *J*= 4.0 Hz, 2 H), 7.58 (d, *J*= 4.0 Hz, 2 H) 7.52 (d, *J*= 4.0 Hz, 2 H), 7.38-7.33 (m, 3 H), 2.58 (s, 3 H); ¹³C NMR (100 MHz, CDCl₃): δ 197.0, 136.6, 132.5, 132.4, 129.4, 128.4, 128.1, 126.4, 121.2, 83.1, 80.3, 76.6, 73.4, 26.5; **HRMS** calcd for C₁₈H₁₂O: 244.0888, found: 244.0893.

4-(phenylbuta-1, 3-diyn-1-yl)benzonitrile (1a2n)^{s9}



Yellow solid; m.p.= 131-135 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.62-7.56 (m, 4 H), 7.52 (d, *J*= 4.0 Hz, 2 H), 7.39-7.31(m, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 133.8, 132.5, 132.0, 129.7, 128.5, 126.7, 121.1, 118.2, 112.3, 83.9, 79.2, 78.1, 73.2; **HRMS** calcd for C₁₇H₉N: 227.0735, found: 227.0738.

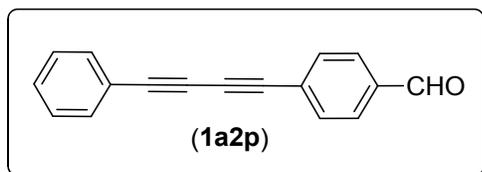
Methyl 4-(phenylbuta-1,3-diyn-1-yl)benzoate (1a2o)^{s10}



Yellow solid; m.p.= 105-109 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.98 (d, *J*= 4.0 Hz, 2 H), 7.57-7.50 (m, 4 H), 7.39-7.30 (m, 3 H) 3.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 166.2, 132.5,

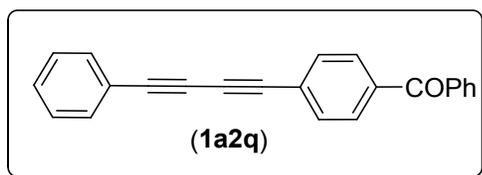
132.3, 130.2, 129.5, 129.4, 128.4, 126.4, 121.3, 82.9, 80.4, 76.6, 73.5, 52.3; **HRMS** calcd for $C_{18}H_{12}O_2$: 260.0837, found: 260.0840.

4-(phenylbuta-1,3-diyne-1-yl)benzaldehyde (1a2p)



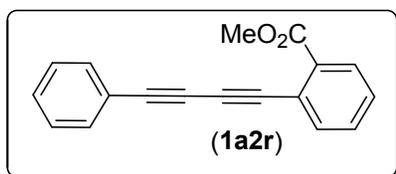
Pale white solid; m.p.= 97-101 °C; **1H NMR** (400 MHz, $CDCl_3$): δ 9.90 (s, 1 H), 7.83 (d, $J= 8.0$ Hz, 2 H), 7.64 (d, $J= 8.0$ Hz, 2 H), 7.53 (d, $J= 8.0$ Hz, 2 H), 7.39-7.31 (m, 3 H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 191.2, 135.9, 132.9, 132.5, 129.6, 129.5, 128.5, 128.0, 121.2, 83.6, 80.2, 77.6, 73.4; **HRMS** calcd for $C_{17}H_{10}O$: 230.0732, found: 230.0728.

phenyl (4-(phenylbuta-1,3-diyne-1-yl)phenyl)methanone (1a2q)



Pale white solid; m.p.= 129-133°C; **1H NMR** (400 MHz, $CDCl_3$): δ 7.76 (d, $J= 4.0$ Hz, 4 H), 7.59 (d, $J= 8.0$ Hz, 2 H), 7.54-7.45 (m, 4 H), 7.37-7.31 (m, 4 H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 195.6, 137.5, 137.1, 132.6, 132.5, 132.4, 132.2, 129.9, 129.4, 129.1, 128.4, 125.9, 121.4, 83.0, 80.4, 76.8, 73.5; **HRMS** calcd for $C_{23}H_{14}O$: 306.1045, found: 306.1044.

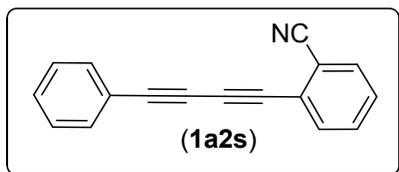
methyl 2-(phenylbuta-1,3-diyne-1-yl)benzoate (1a2r)^{s3}



Pale white solid; m.p.= 91-93°C **1H NMR** (400 MHz, $CDCl_3$): δ 7.95 (d, $J= 6.3$, 1 H), 7.64 (d, $J= 6.3$, 1 H), 7.52-7.51 (m, 2 H), 7.48-7.46 (m, 1 H), 7.40-7.37 (m, 1 H), 7.35-7.30 (m, 3 H); **^{13}C**

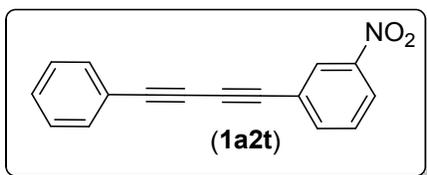
NMR (100 MHz, CDCl₃): δ 166.0, 135.1, 132.4, 131.7, 130.5, 129.2, 128.6, 128.3, 122.5, 121.7, 83.2, 79.7, 78.7, 74.0, 52.3; **HRMS** calcd for C₁₈H₁₂O₂: 260.0837, found: 260.0831.

2-(phenylbuta-1,3-diyne-1-yl)benzotrile (1a2s)



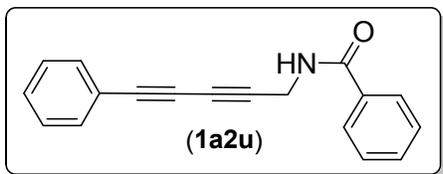
Pale white solid; m.p.= 124-129 °C; **¹H NMR** (400 MHz, CDCl₃): δ 7.65-7.60 (m, 2 H), 7.57-7.52 (m, 3 H), 7.45-7.32 (m, 4 H); **¹³C NMR** (100 MHz, CDCl₃): δ 133.3, 132.8, 132.6, 132.4, 129.7, 129.0, 128.4, 125.7, 121.0, 117.1, 115.8, 84.3, 79.9, 76.6, 73.2; **HRMS** calcd for C₁₇H₉N: 227.0735, found: 227.0732.

1-nitro-3-(phenylbuta-1,3-diyne-1-yl)benzene (1a2t)^{s6}



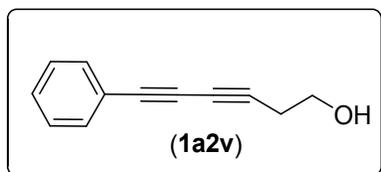
Pale white solid; m.p.= 147-151 °C; **¹H NMR** (400 MHz, CDCl₃): δ 8.32-7.77 (m, 3 H), 7.53-7.48 (m, 3 H), 7.40-7.31 (m, 3 H); **¹³C NMR** (100 MHz, CDCl₃): δ 148.3, 138.2, 132.8, 129.9, 129.8, 128.7, 127.4, 124.0, 121.3, 83.3, 78.8, 77.6, 73.4; **HRMS** calcd for C₁₆H₉NO: 247.0633, found: 247.0635.

N-(5-phenylpenta-2,4-diyne-1-yl)benzamide (1a2u)^{s11}



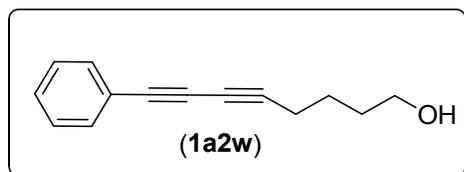
Pale white solid; m.p.= 106-110°C; **¹H NMR** (400 MHz, CDCl₃): δ 7.81-7.79 (m, 2 H), 7.49-7.25 (m, 8 H), 6.98 (s, 1 H), 4.38 (s, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 167.3, 133.4, 132.5, 131.7, 129.2, 128.5, 128.2, 127.0, 121.2, 78.5, 77.3, 73.3, 68.0, 30.5; **HRMS** calcd for C₁₈H₁₃NO: 259.0997, found: 259.1000.

6-phenylhexa-3, 5-diyne-1-ol (1a2v)^{s12}



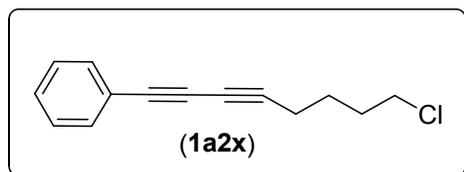
Colorless oil; **¹H NMR** (400 MHz, CDCl₃): δ 7.45 (d, *J*= 12.0 Hz, 2 H), 7.32-7.28 (m, 3 H), 3.77 (t, *J*= 12.0 Hz, 2 H), 2.62 (t, *J*= 12.0 Hz, 3 H); **¹³C NMR** (100 MHz, CDCl₃): δ 132.5, 129.0, 128.3, 121.7, 80.9, 75.3, 73.9, 66.8, 60.7, 23.9; IR (KBr, cm⁻¹) 3054, 2365, 1776, 1604, 1474, 979; **HRMS** calcd for C₁₂H₁₀O: 170.0732, found: 170.0729.

8-phenylocta-5,7-diyne-1-ol (1a2w)^{s12}



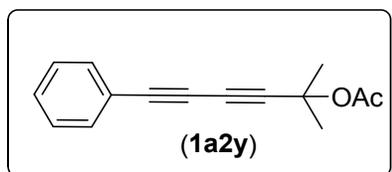
Colorless oil; **¹H NMR** (400 MHz, CDCl₃): δ 7.45-7.43 (m, 2 H), 7.31-7.25 (m, 3 H), 3.65 (t, *J*= 4.0 Hz, 2 H), 2.38 (t, *J*= 8.0 Hz, 3 H) 1.69-1.65(m, 3 H); **¹³C NMR** (100 MHz, CDCl₃): δ 132.4, 128.8, 128.2, 121.9, 84.2, 74.8, 74.1, 65.3, 62.1, 31.6, 24.4, 19.3; **HRMS** calcd for C₁₄H₁₄O: 198.1045, found: 198.1039.

(8-chloroocta-1,3-diyne-1-yl)benzene (1a2x)



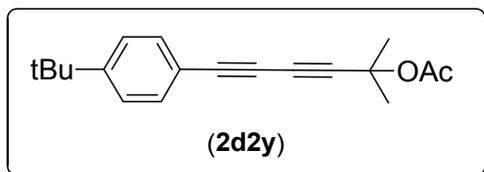
Colorless oil; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.46-7.44 (m, 2 H), 7.32-7.26 (m, 3 H), 3.56 (t, $J=8.0$ Hz, 2 H), 2.40 (t, $J=8.0$ Hz, 2 H) 1.95-1.88 (m, 2 H), 1.76-1.68 (m, 2 H), ; $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 132.4, 128.9, 128.3, 121.9, 83.5, 75.0, 74.1, 65.7, 44.3, 31.4, 25.3, 18.8; **HRMS** calcd for $\text{C}_{14}\text{H}_{13}\text{Cl}$: 216.0706, found: 216.0800.

2-methyl-6-phenylhexa-3,5-diyne-2-yl acetate (1a2y)



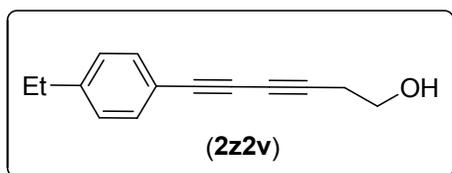
Pale white solid; m.p.= 55-59 °C; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.45-7.43 (m, 2 H), 7.33-7.26 (m, 3 H), 2.02 (s, 3H), 1.68 (s, 6 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 169.2, 132.4, 129.1, 128.3, 121.5, 82.7, 79.2, 73.2, 71.8, 68.9, 28.6, 21.7; **HRMS** calcd for $\text{C}_{15}\text{H}_{14}\text{O}_2$: 226.0994, found: 226.0998.

6-(4-(tert-butyl)phenyl)-2-methylhexa-3,5-diyne-2-yl acetate (2d2y)



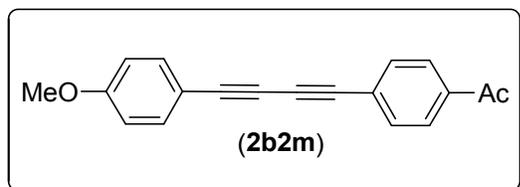
Pale white solid; m.p.= 77-81 °C; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.39 (d, $J=12.0$ Hz, 2H), 7.30 (d, $J=8.0$ Hz, 2 H), 2.02 (s, 3 H), 1.68 (s, 6 H), 1.27 (s, 9 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 169.2, 152.5, 132.2, 125.3, 118.3, 82.2, 79.4, 72.5, 71.8, 69.1, 34.8, 31.0, 28.6, 21.7.; **HRMS** calcd for $\text{C}_{19}\text{H}_{22}\text{O}_2$: 282.1620, found: 282.1619.

6-(4-ethylphenyl)hexa-3,5-diyne-1-ol (2z2v)



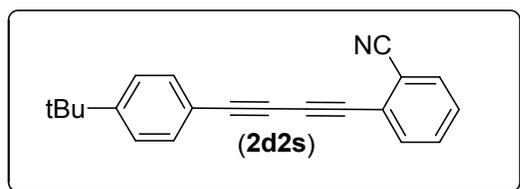
Colorless oil; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.37 (d, $J= 8.0$ Hz, 2 H), 7.11 (d, $J= 8.0$ Hz, 2 H), 3.77 (t, $J= 4.0$ Hz, 2 H), 2.61 (q, $J= 12.0$ Hz, 2H), 2.01 (s, 1 H), 1.19 (t, $J= 12.0$ Hz, 3 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 145.6, 132.5, 127.9, 118.7, 80.5, 75.6, 73.2, 66.9, 60.7, 28.8, 23.9, 15.1; **HRMS** calcd for $\text{C}_{14}\text{H}_{14}\text{O}$: 198.1045, found: 198.1039.

1-((4-methoxyphenyl)buta-1,3-diyn-1-yl)phenyl)ethanone (2b2m)



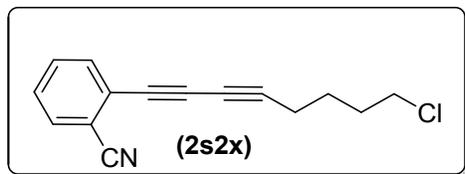
Pale white solid; m.p.= 134-138 °C; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.88 (d, $J= 8.0$ Hz, 2 H), 7.55 (d, $J= 8.0$ Hz, 2 H), 7.44 (d, $J= 8.0$ Hz, 2 H), 6.83 (d, $J= 8.0$ Hz, 2 H), 3.79 (s, 3 H), 2.56 (s, 3 H), ; $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 197.1, 160.5, 136.5, 134.1, 132.4, 128.1, 126.8, 114.1, 113.1, 83.5, 79.9, 77.3, 72.4, 55.3, 26.5; **HRMS** calcd for $\text{C}_{19}\text{H}_{14}\text{O}_2$: 274.0994, found: 274.0992.

2-((4-(tert-butyl)phenyl)buta-1,3-diyn-1-yl)benzonitrile (2d2s)



Pale white solid; m.p.= 140-144 °C; $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 7.64-7.52 (m, 3 H), 7.47-7.42 (m, 3 H), 7.36-7.34 (m, 2 H), 1.30 (s, 9 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 153.2, 133.3, 132.7, 132.4, 132.3, 128.9, 125.9, 125.5, 117.9, 117.1, 115.7, 84.7, 80.2, 76.4, 72.6, 34.9, 31.0; **HRMS** calcd for $\text{C}_{21}\text{H}_{17}\text{N}$: 283.1361, found: 283.1363.

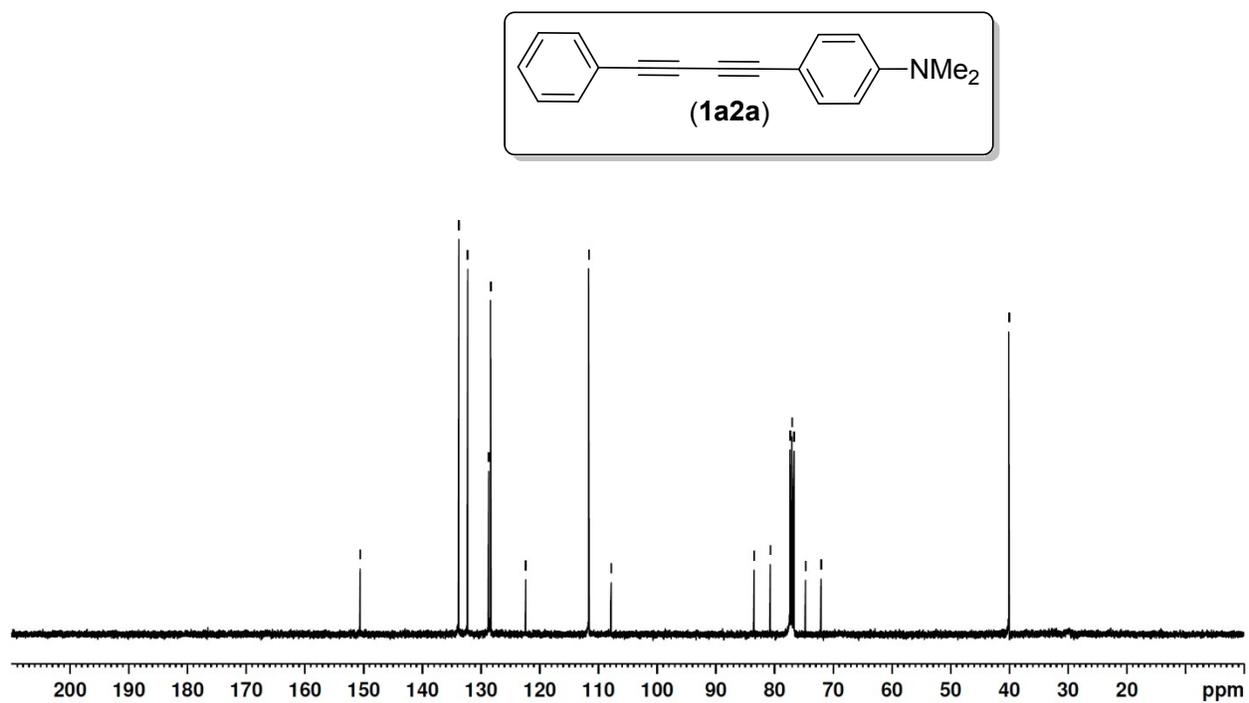
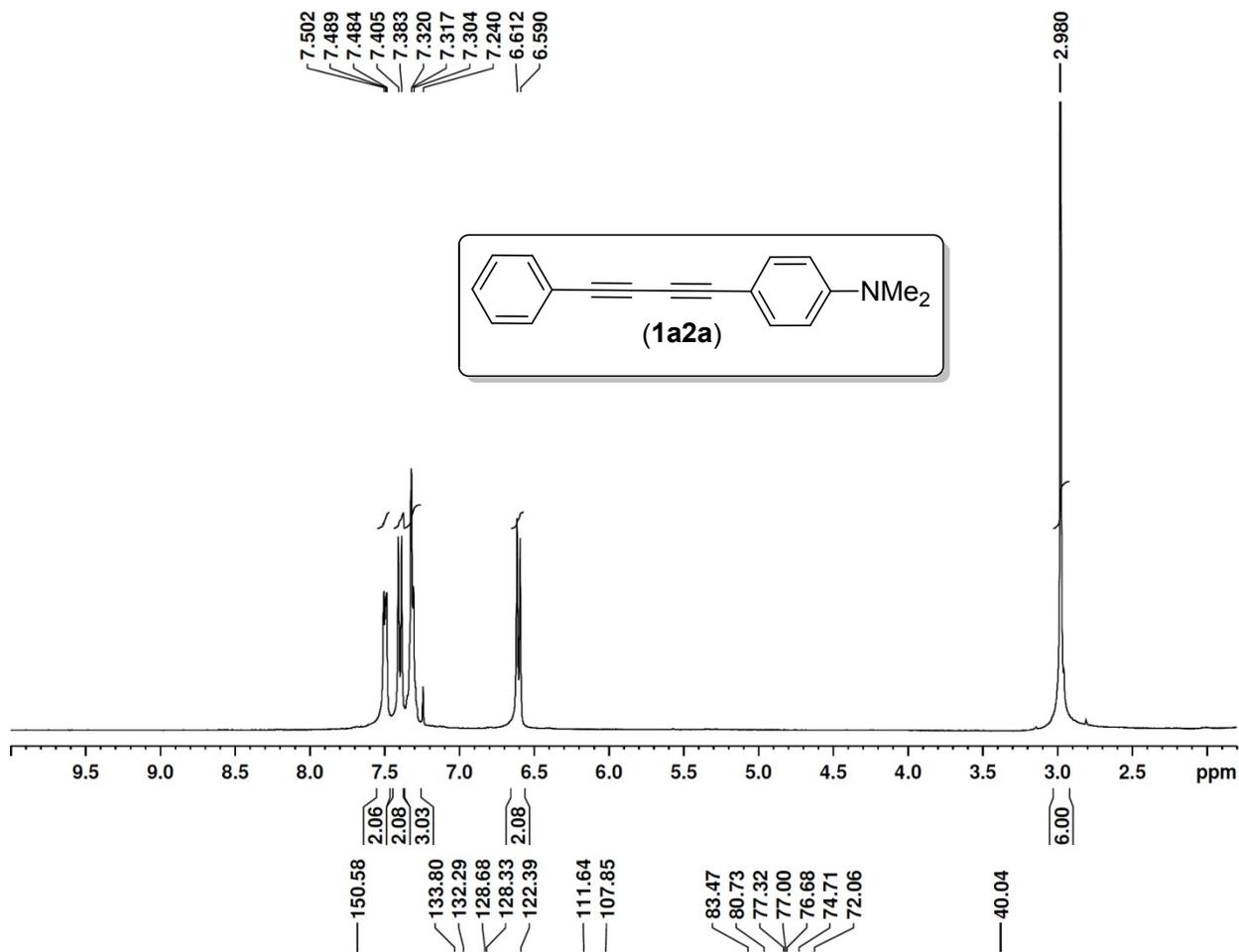
2-(8-chloroocta-1,3-diyn-1-yl)benzonitrile (2s2x)

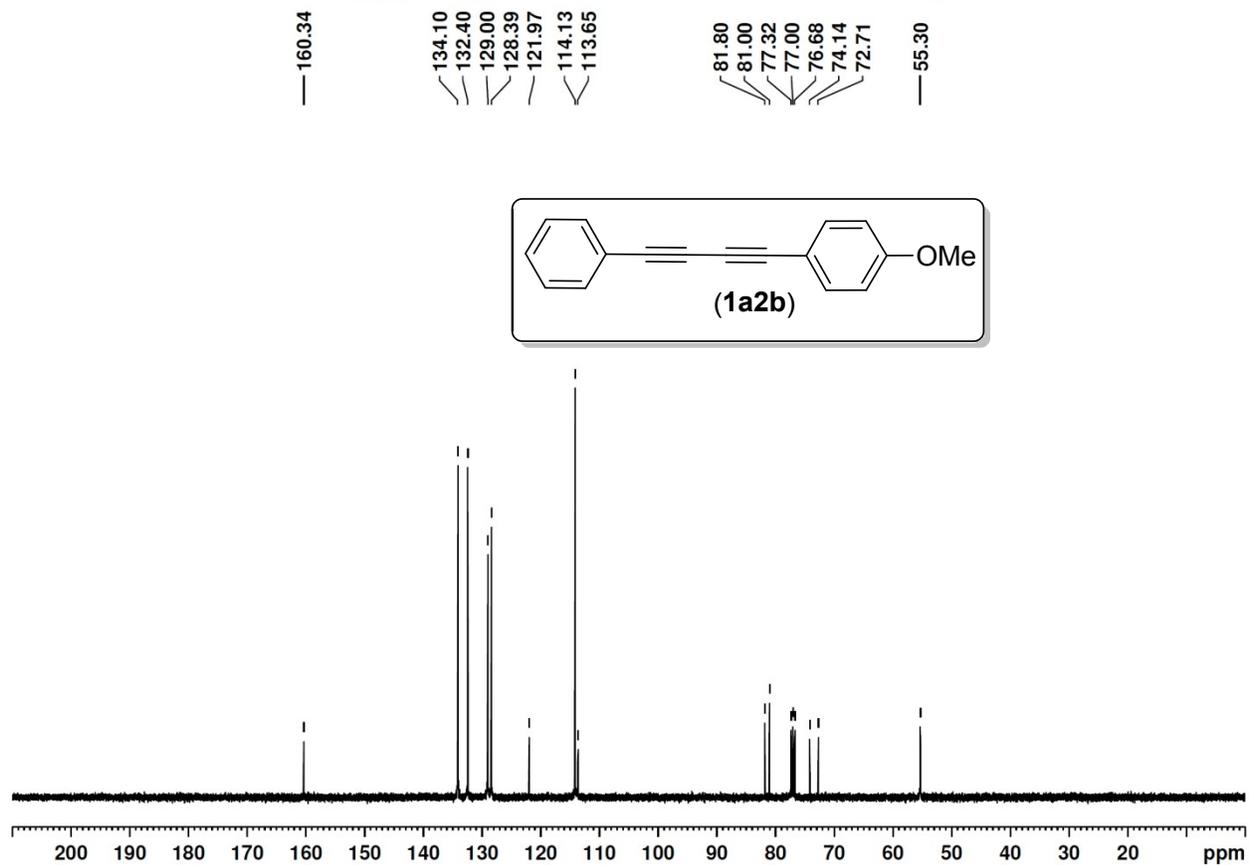
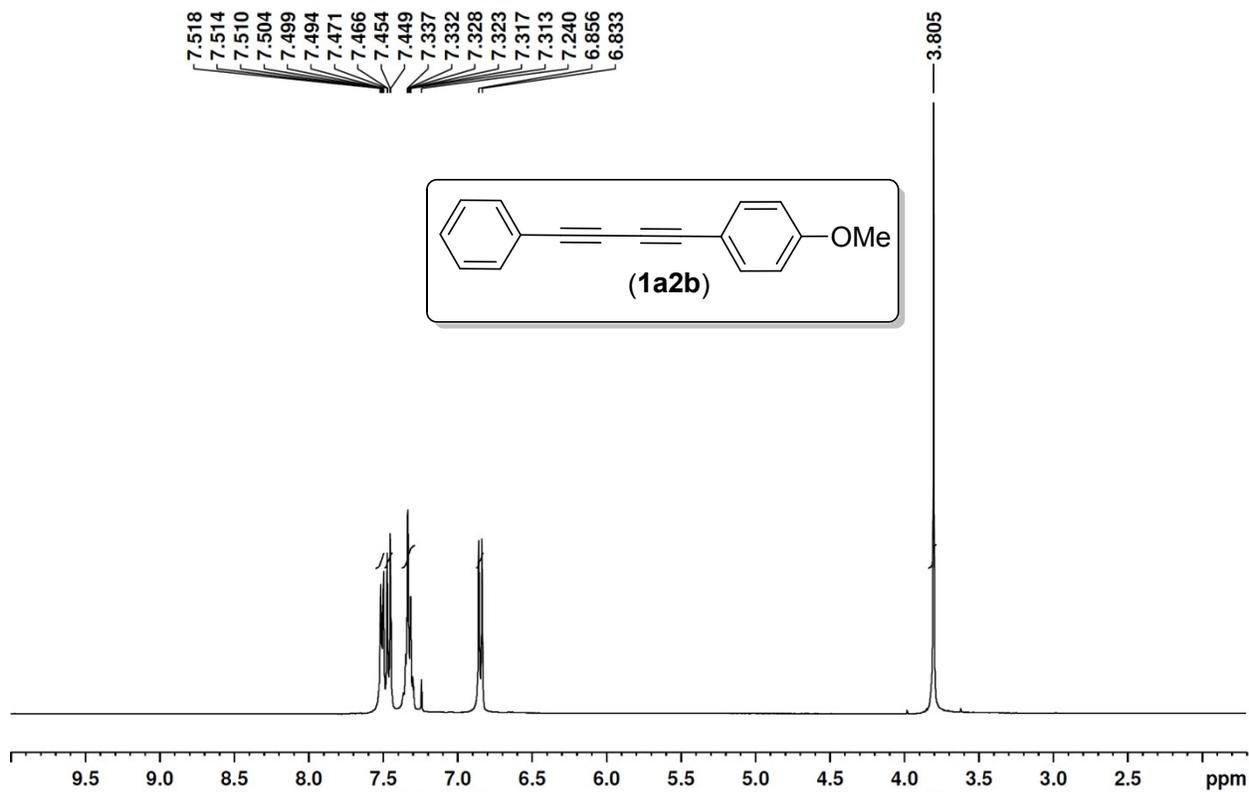


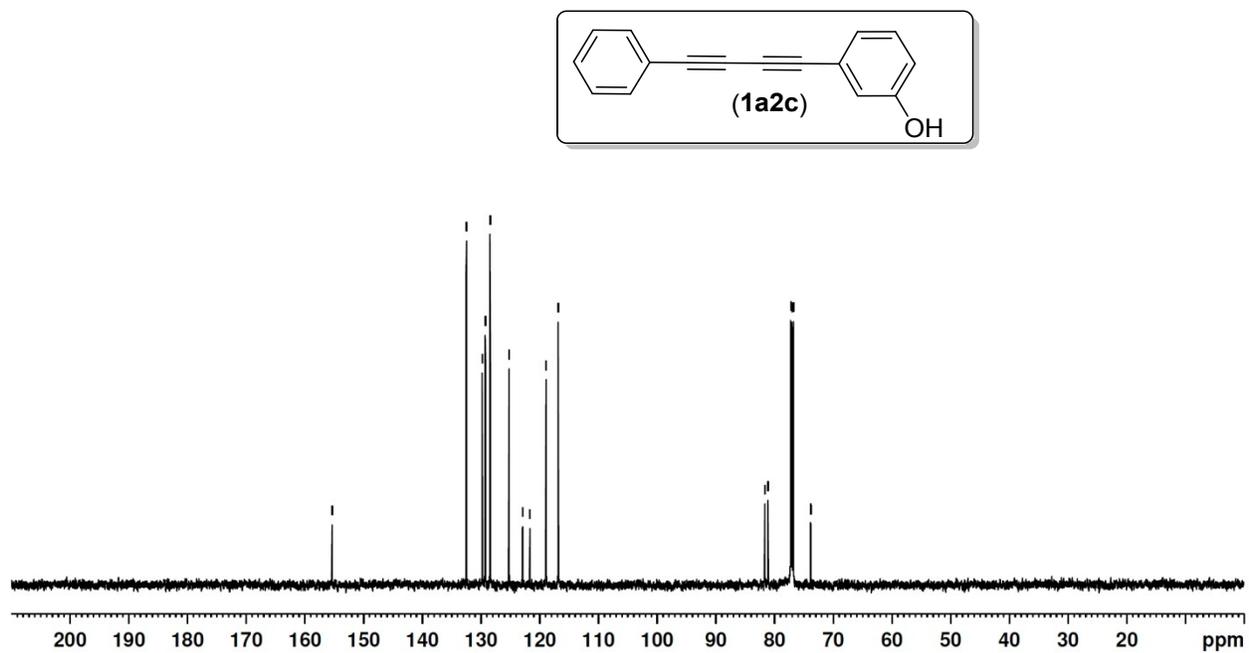
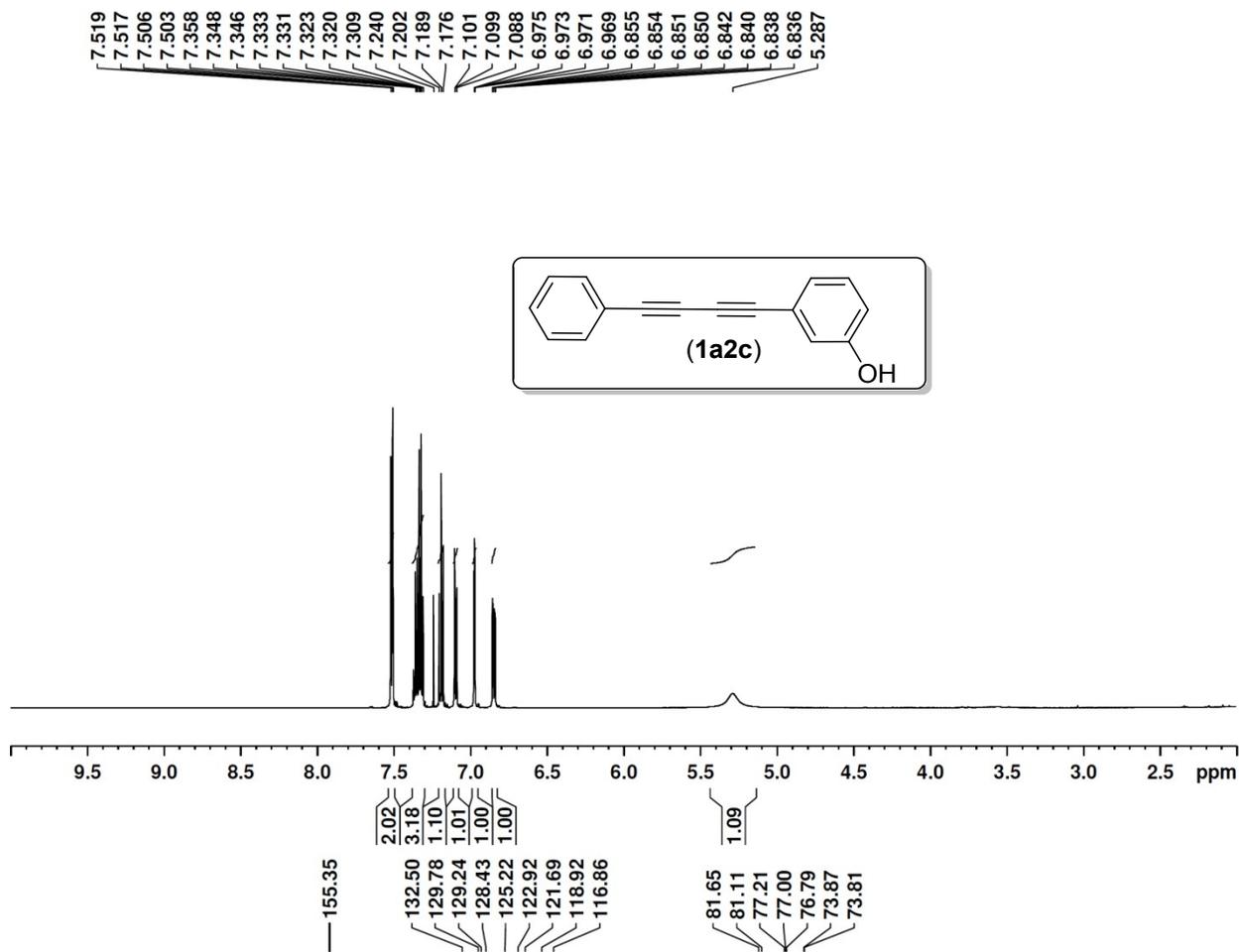
Pale white solid; m.p.= 120-124 °C ¹H NMR (400 MHz, CDCl₃): δ 7.61 (d, *J*= 8.0 Hz, 1 H), 7.57-7.49 (m, 2 H), 7.40 (d, *J*= 8.0 Hz, 1 H), 3.56 (t, *J*= 8.0 Hz, 2 H), 2.43 (t, *J*= 4.0 Hz, 2 H), 1.93-1.89 (m, 2 H), 1.77-1.71 (m, 2 H) ; ¹³C NMR (100 MHz, CDCl₃): δ 133.4, 132.7, 132.3, 128.8, 126.0, 117.1, 116.0, 86.8, 80.4, 70.5, 65.2, 44.2, 31.3, 25.1, 18.9; HRMS calcd for C₁₅H₁₂CN: 241.0658, found: 247.0656.

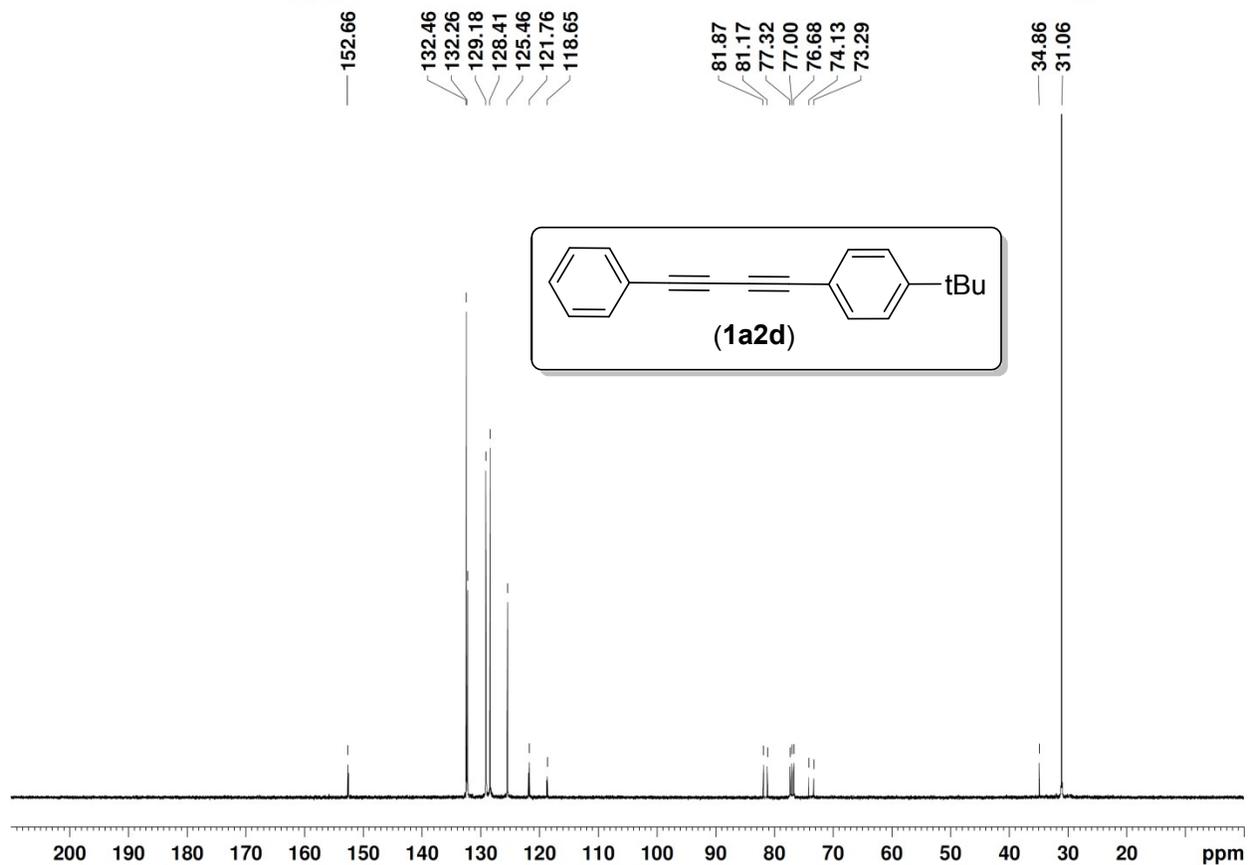
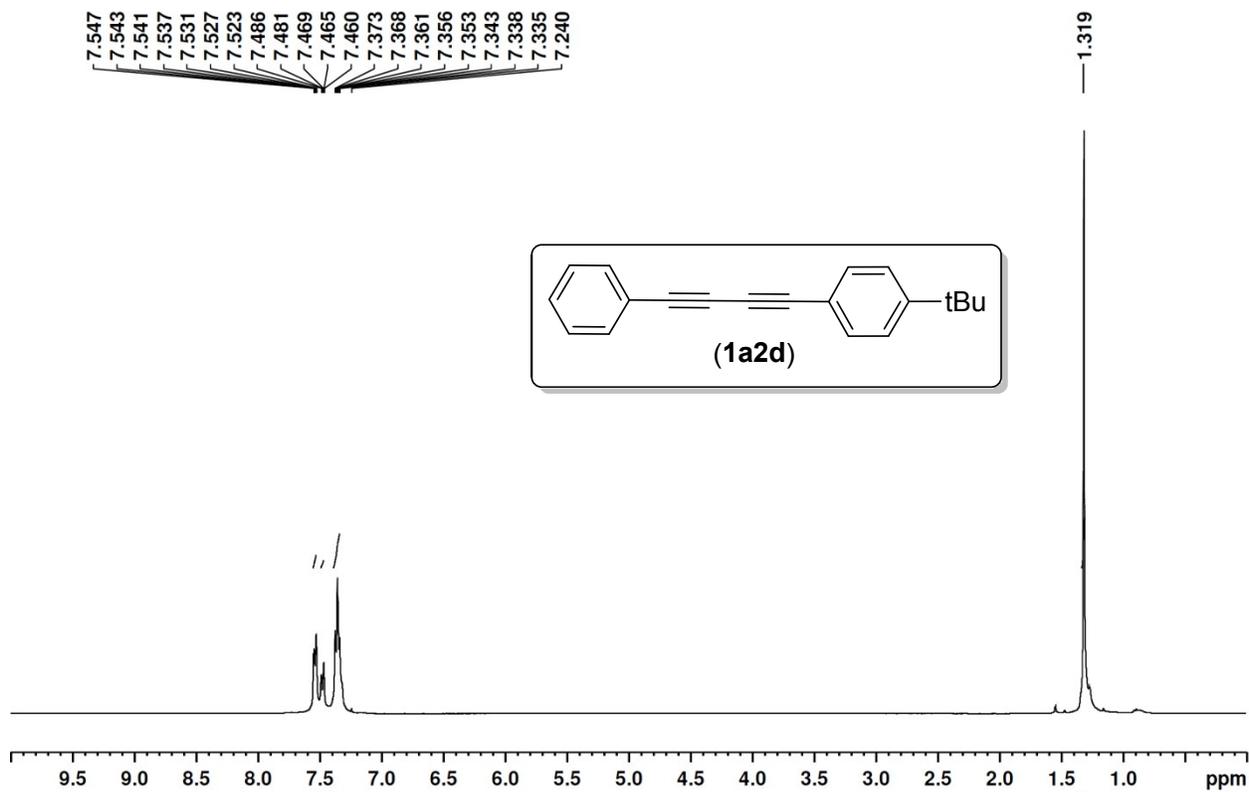
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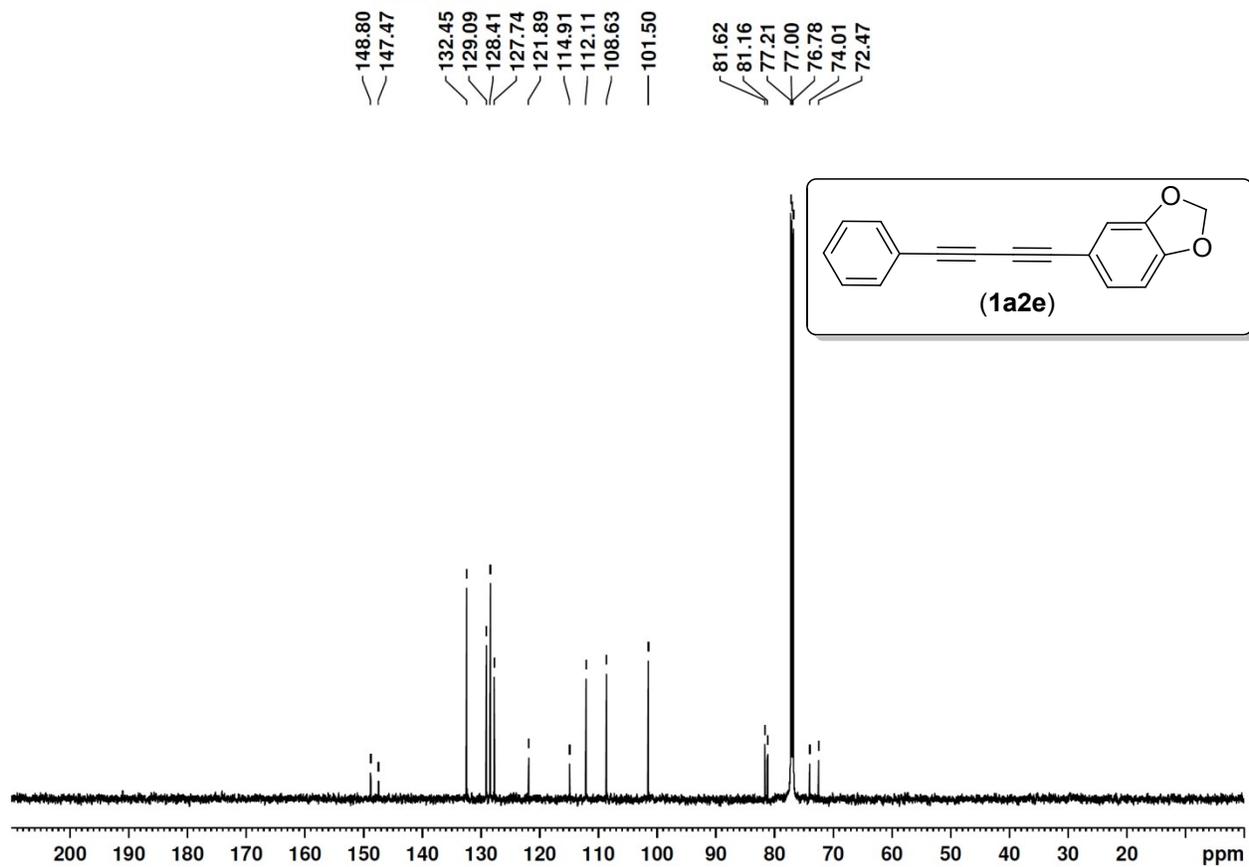
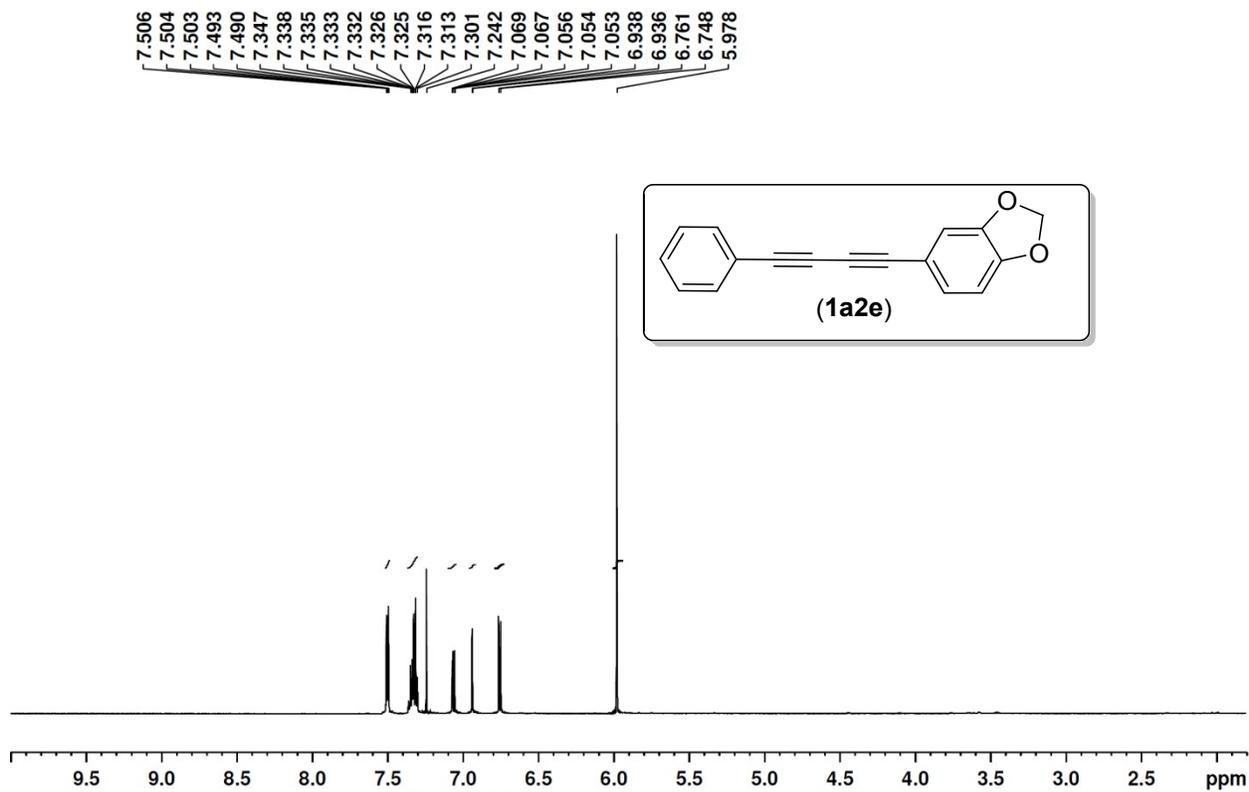
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- S5. Y. Liu, C. Wang, X. Wang, and J. Wan, *Tetrahedron Lett.*, 2013, **54**, 3953.
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- S9. Y. Weng, B. Cheng, C. He, and A. Lei, *Angew. Chem. Int. Ed.*, 2012, **51**, 9547.
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- S12. H. Li, L. Wang, M. Yang, and Y. Qi, *Catal. Commun.*, 2012, **17**, 179.



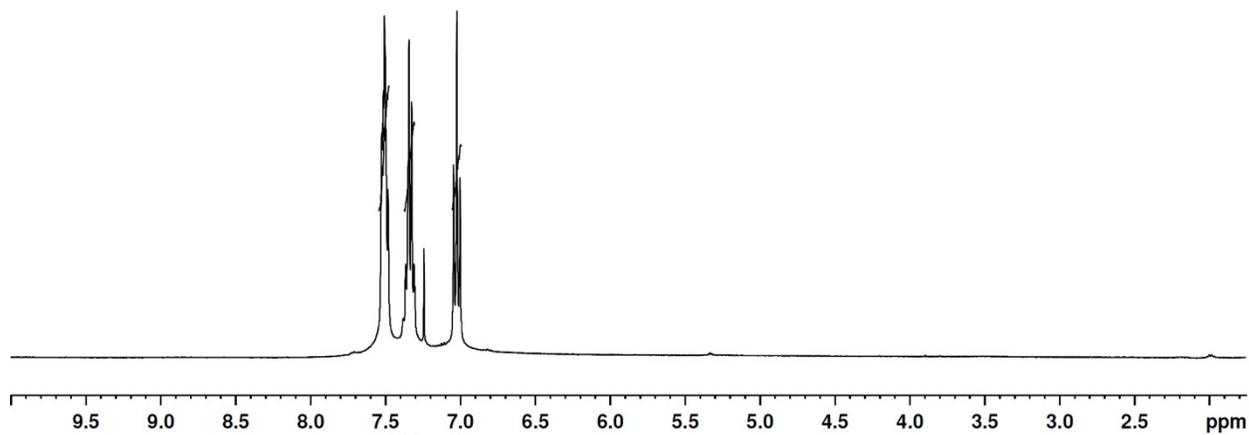
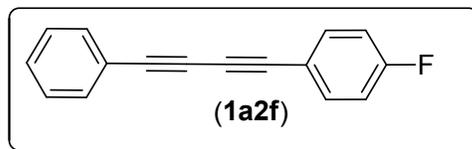




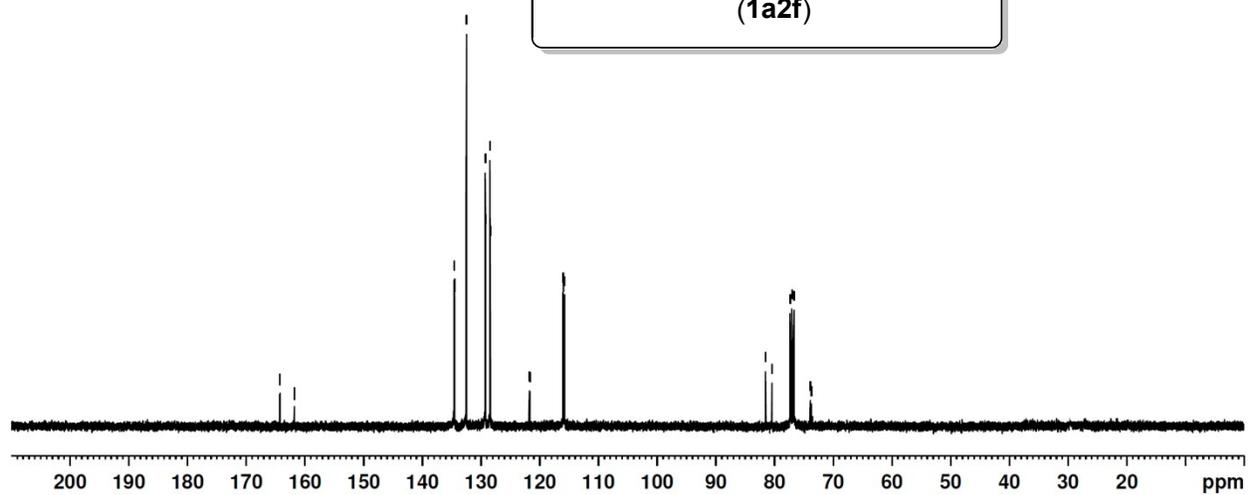
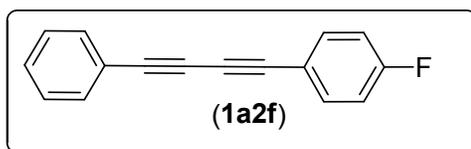




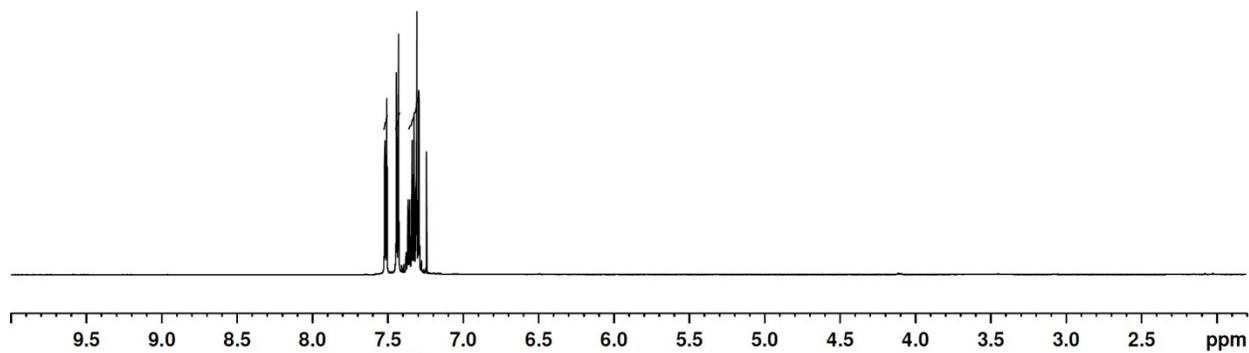
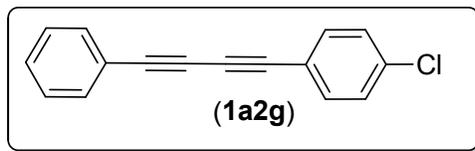
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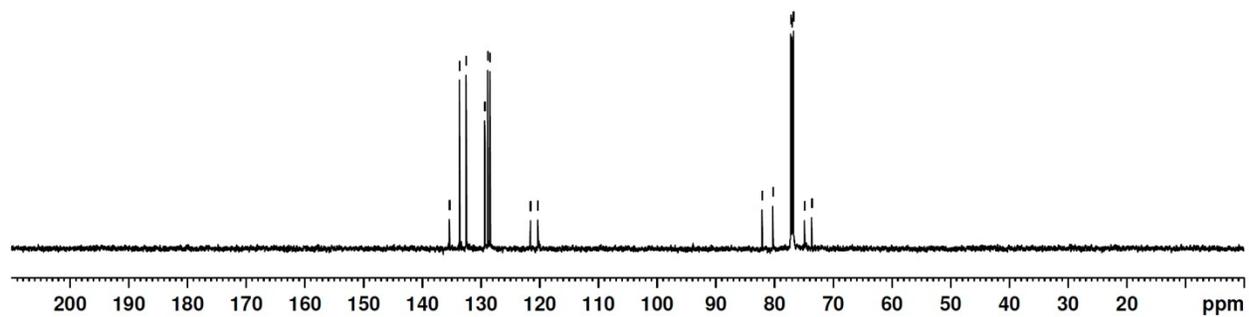
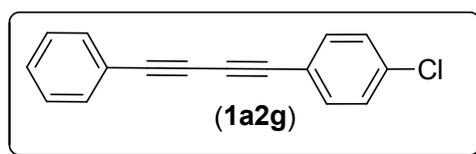
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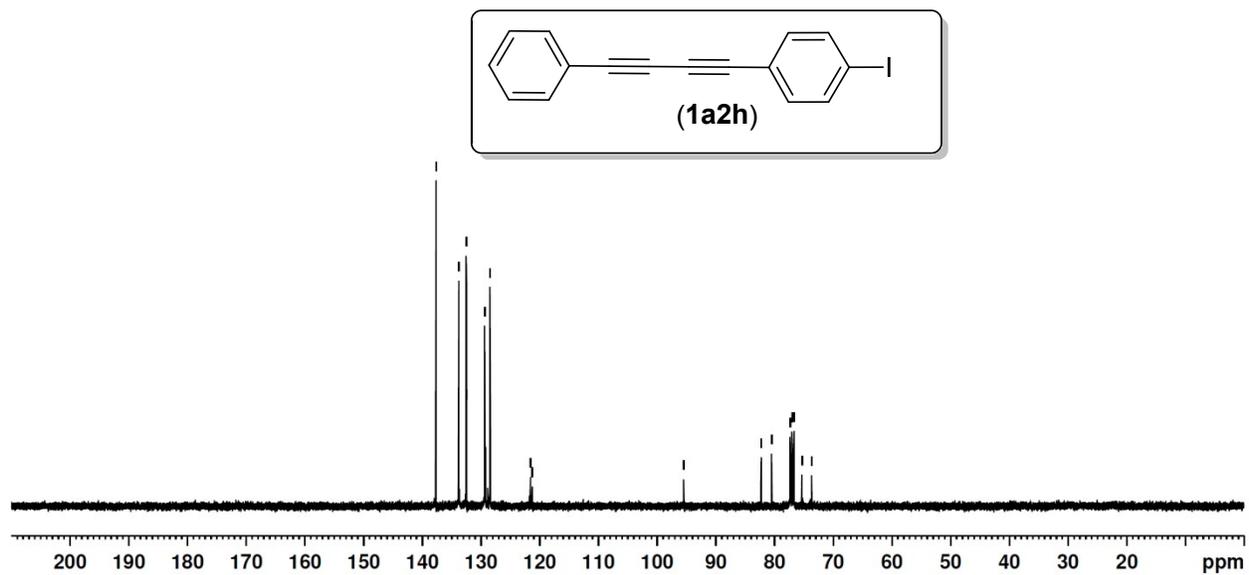
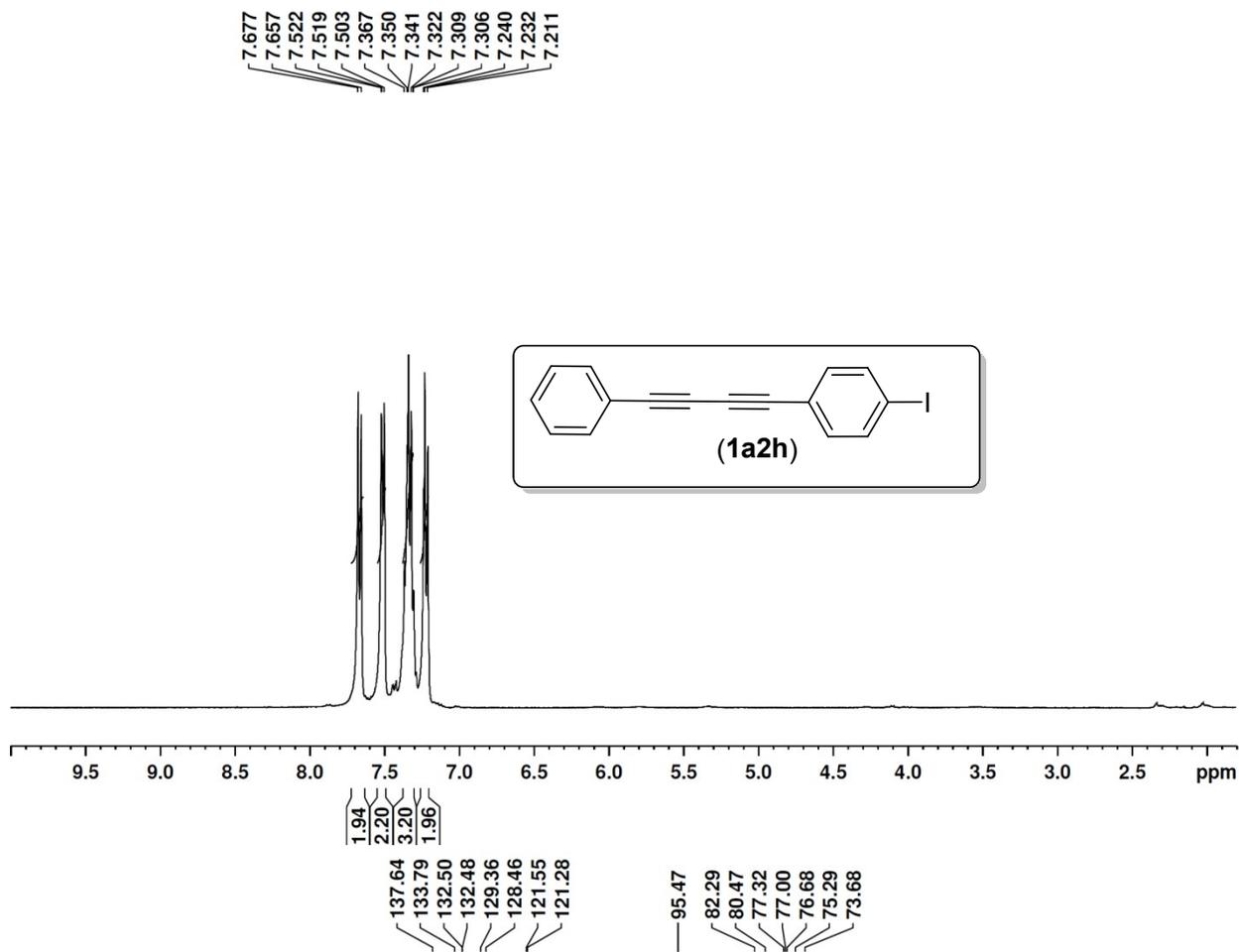


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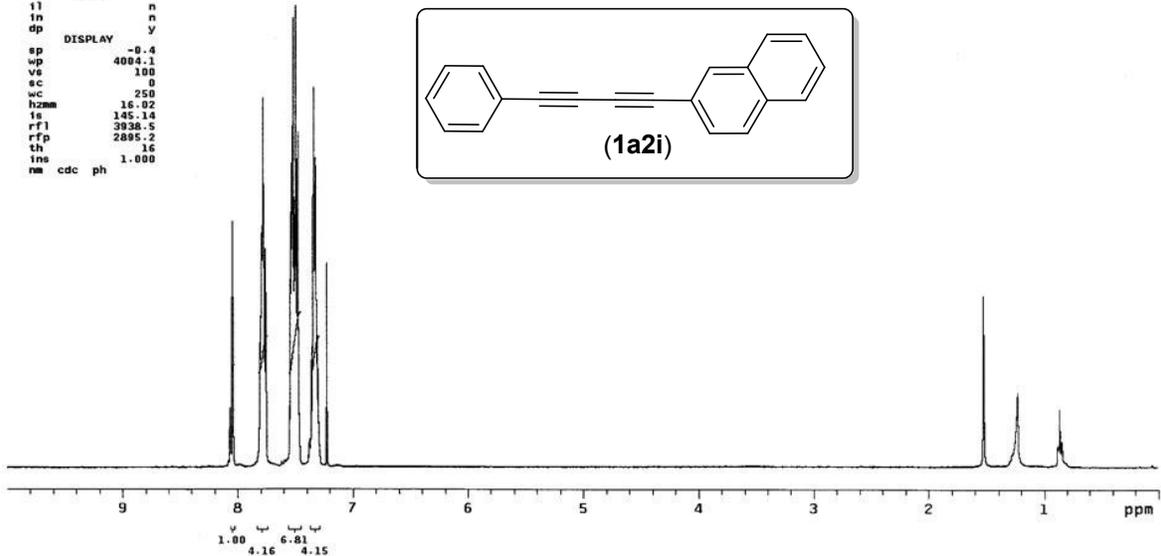
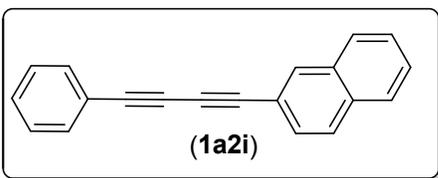
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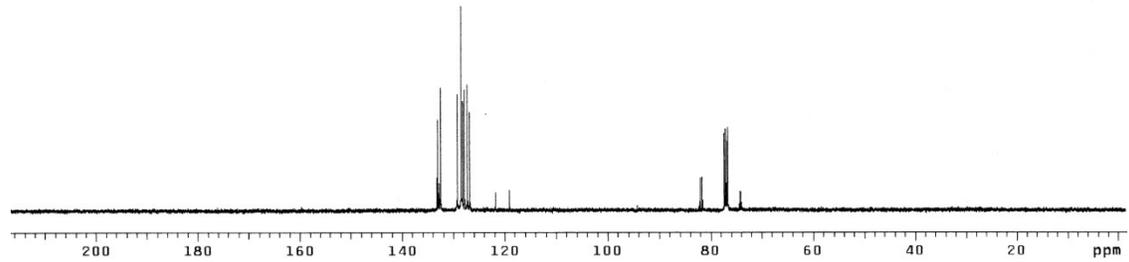
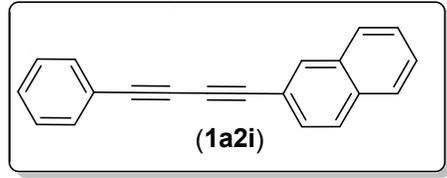
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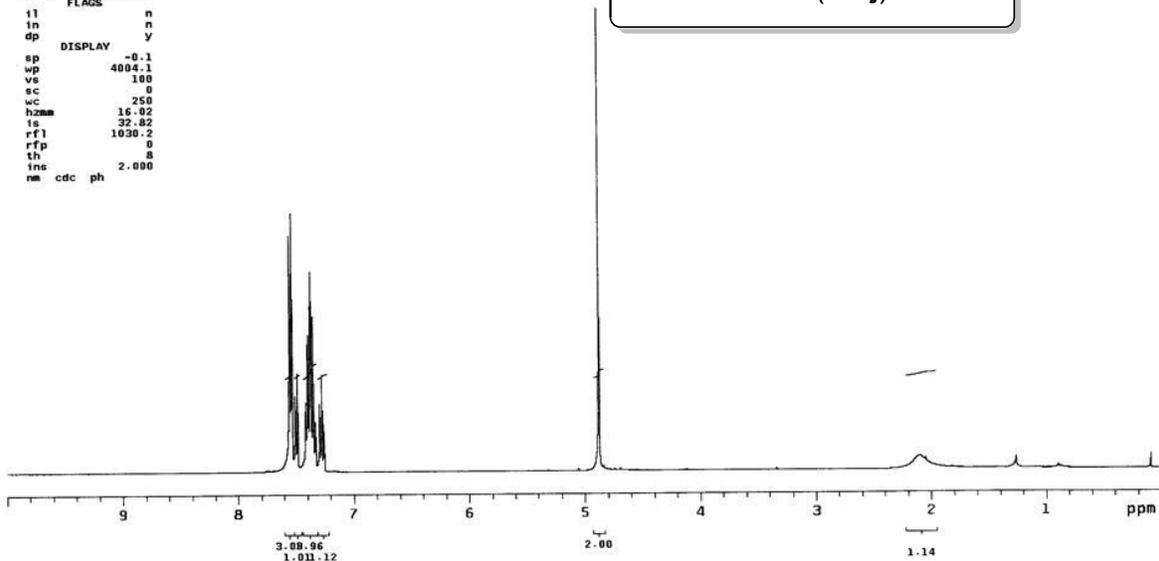
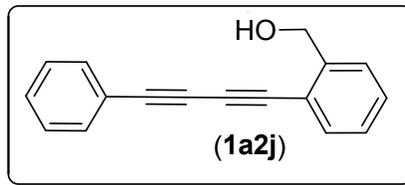


```

1344-b
expl stdlh
SAMPLE
date Jan 21 2013 dfrq 400.444
solvent CDC13 dn H1
file exp dpwr 30
ACQUISITION dof 0
sfrq 400.444 da nm
tn H1 dm c
at 1.995 dmf PROCESSING 200
np 23064
sw 6086.0 wtfile ft
fb 3400 proc
bs 4 fn not used
tpwr 57
pw 7.0 werr
dl 1.000 wexp
tof 0 wbs
nt 100 wnt
ct 28
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -0.1
wp 4004.1
vs 100
sc 0
wc 250
hzmm 16.02
is 32.02
rf1 1030.2
rfp 0
th 8
ins 2.000
nm cdc ph

```

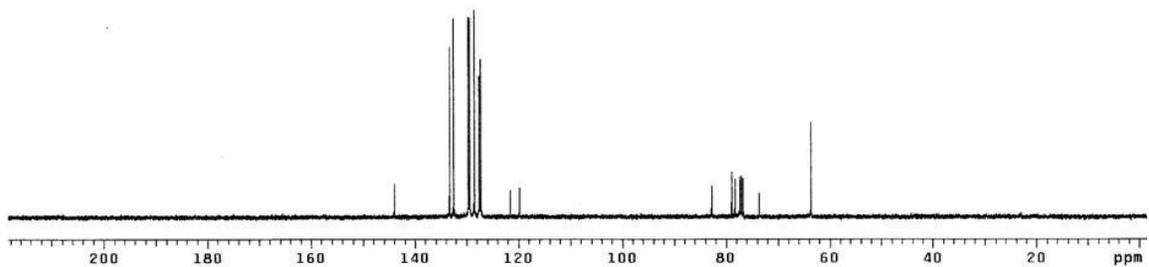
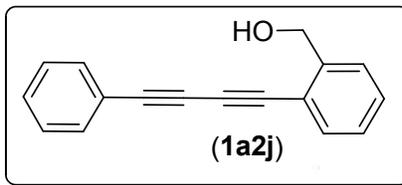
4.875

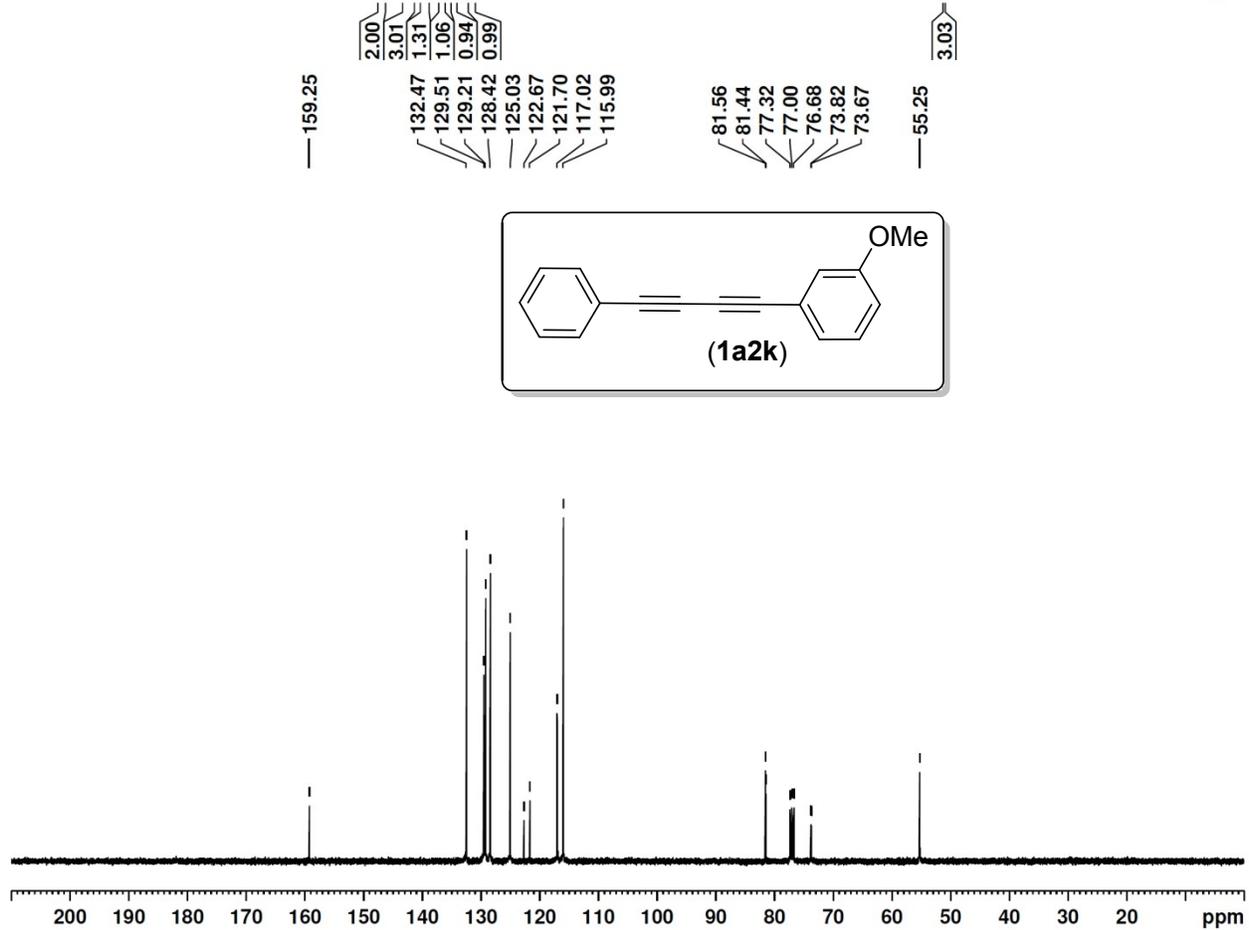
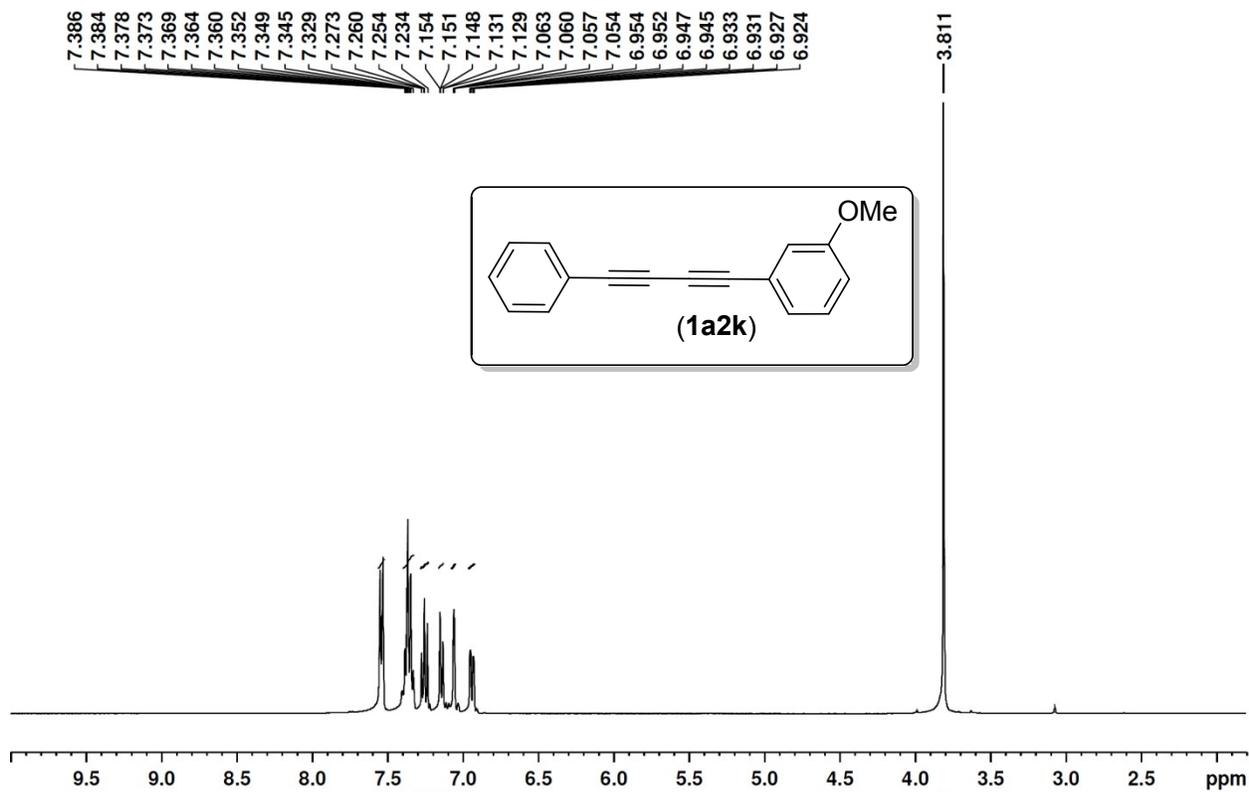


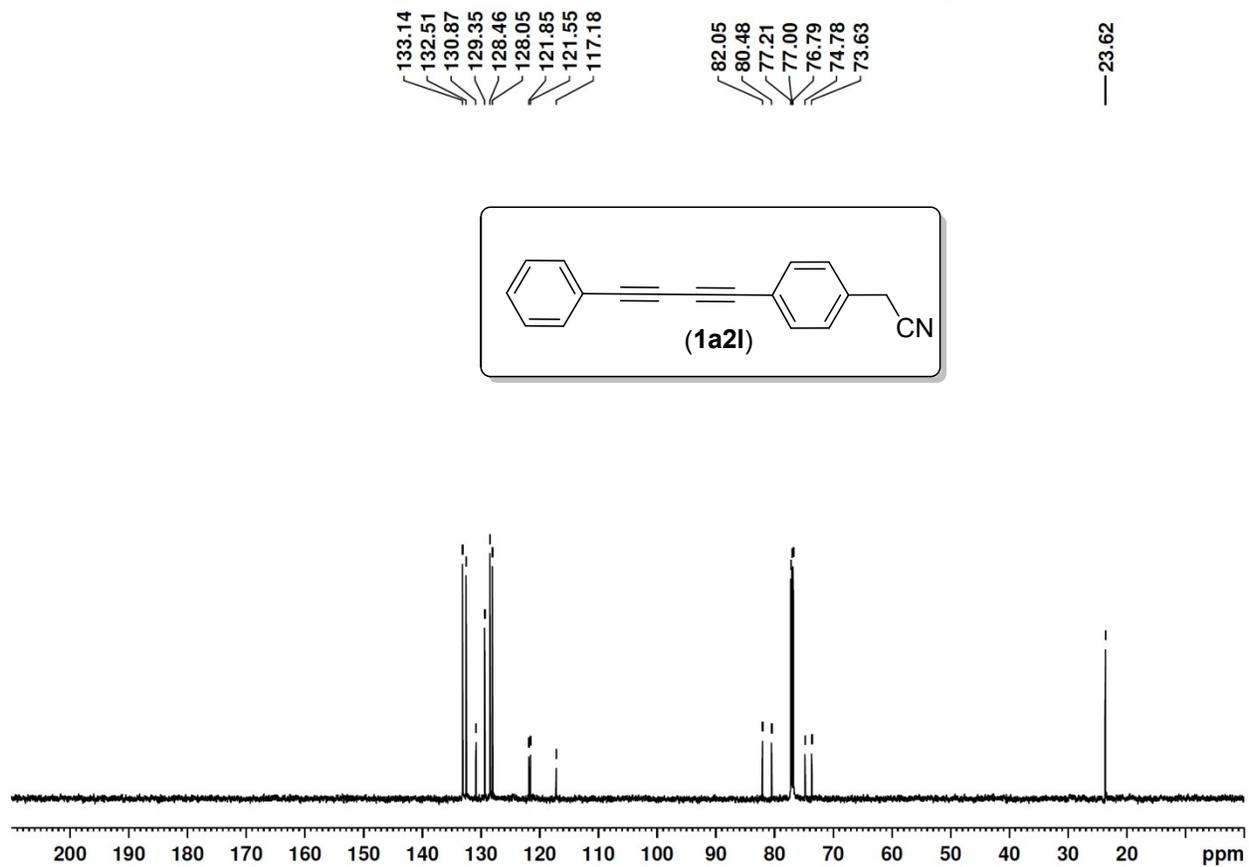
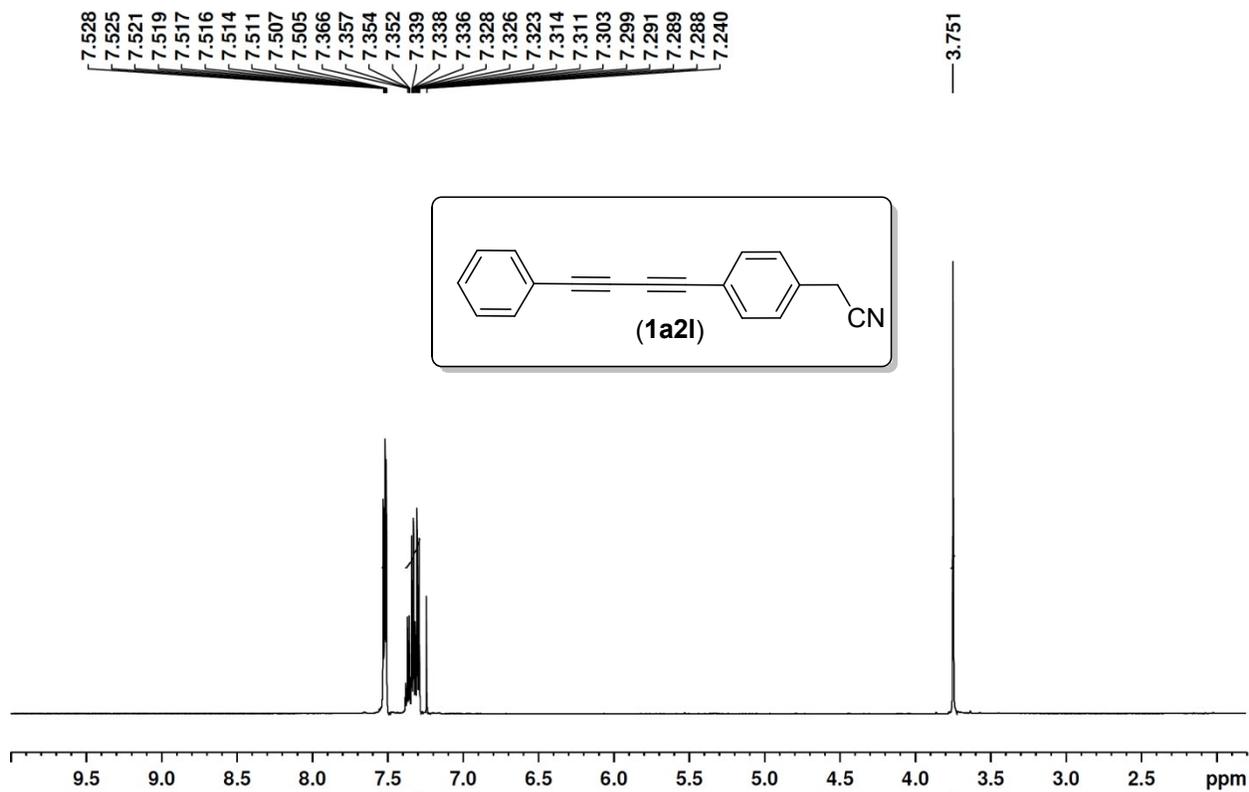
```

1344-b
exp2 std13c
SAMPLE
date Jan 21 2013 dfrq 400.444
solvent CDC13 dn H1
file exp dpwr 30
ACQUISITION dof 0
sfrq 100.701 da yyy
tn C13 dm 9000
at 1.189 dmf PROCESSING 1.00
np 58968
sw 25000.0 lb
fb 13800 wtfile
bs 4 proc ft
tpwr 58 fn not used
pw 8.7
dl 0 werr
tof 100000 wexp
ct 828 wnt
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -161.5
wp 22151.8
ve 45
sc 0
wc 250
hzmm 88.61
is 500.00
rf1 10752.8
rfp 7753.2
th 3
ins 100.000
nm no ph

```



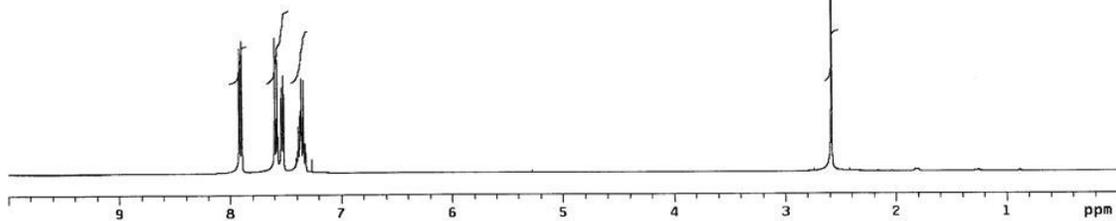
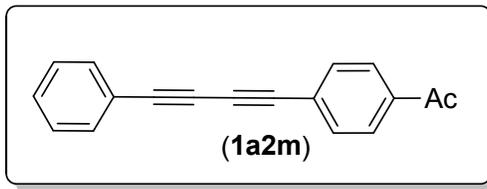




```

1340-a
exp9 std1h
SAMPLE
date Jan 1 2013 dfrq 400.444
solvent CDC13 dn H1
file exp dpwr 39
ACQUISITION exp dof 0
sfrq 400.444 dm nnn
tn H1 dnm C
at 1.935 dmf 200
np 23964 PROCESSING
sw 6006.0 wtfile ft
fb 3400 proc
bs 4 fn not used
tpwr 57
pw 7.0 werr
dl 1.000 wexp
tof 0 wbs
nt 100 wnt
ct 20
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -0.1
wp 4004.1
vs 100
sc 0
wc 250
hzmm 16.02
is 168.16
rfl 1030.2
rfp 0
th 2
ins 3.000
nm cdc ph

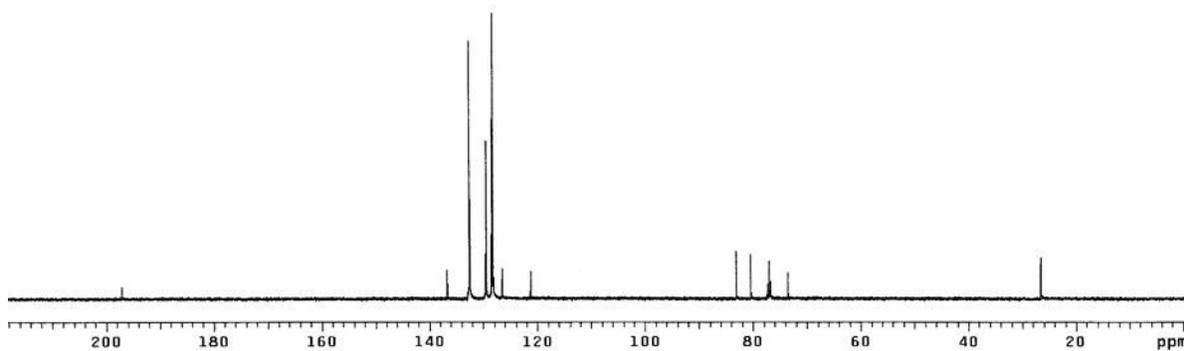
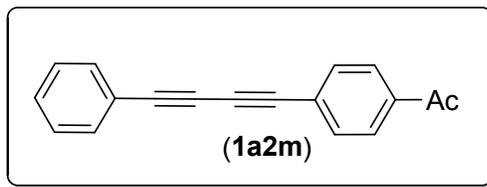
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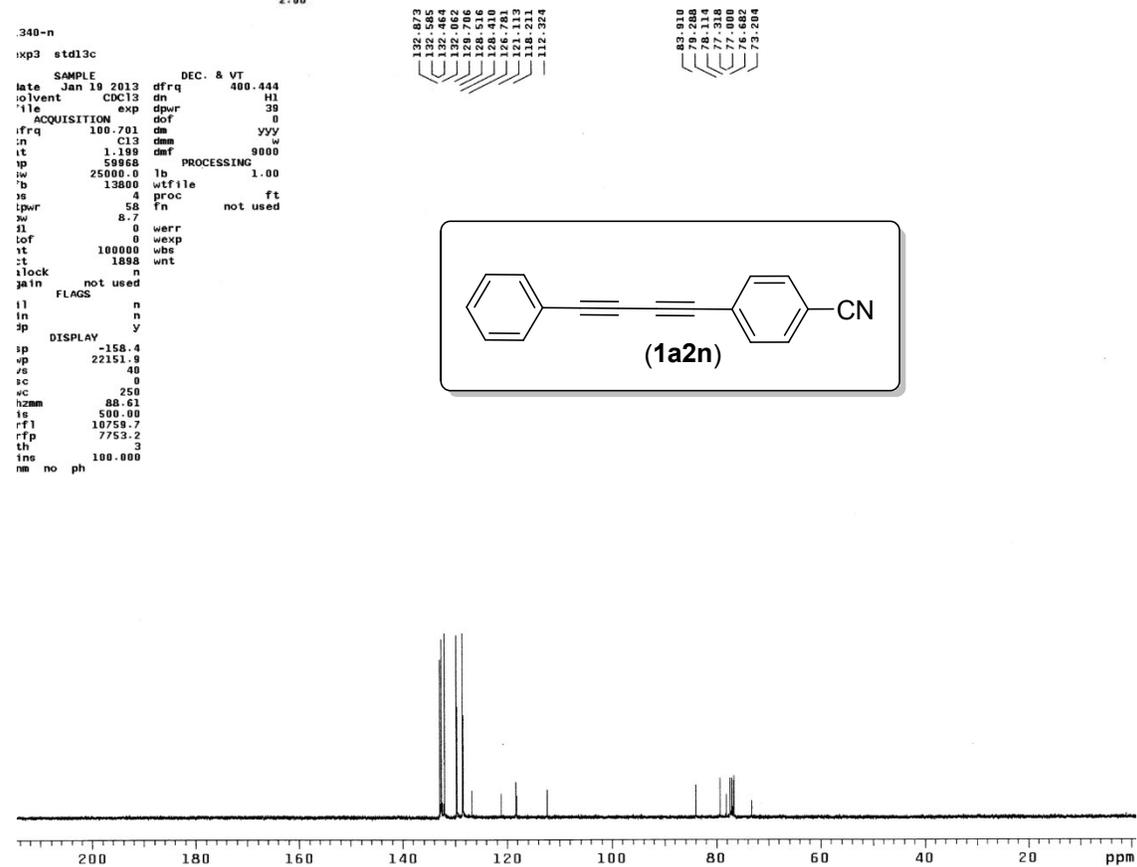
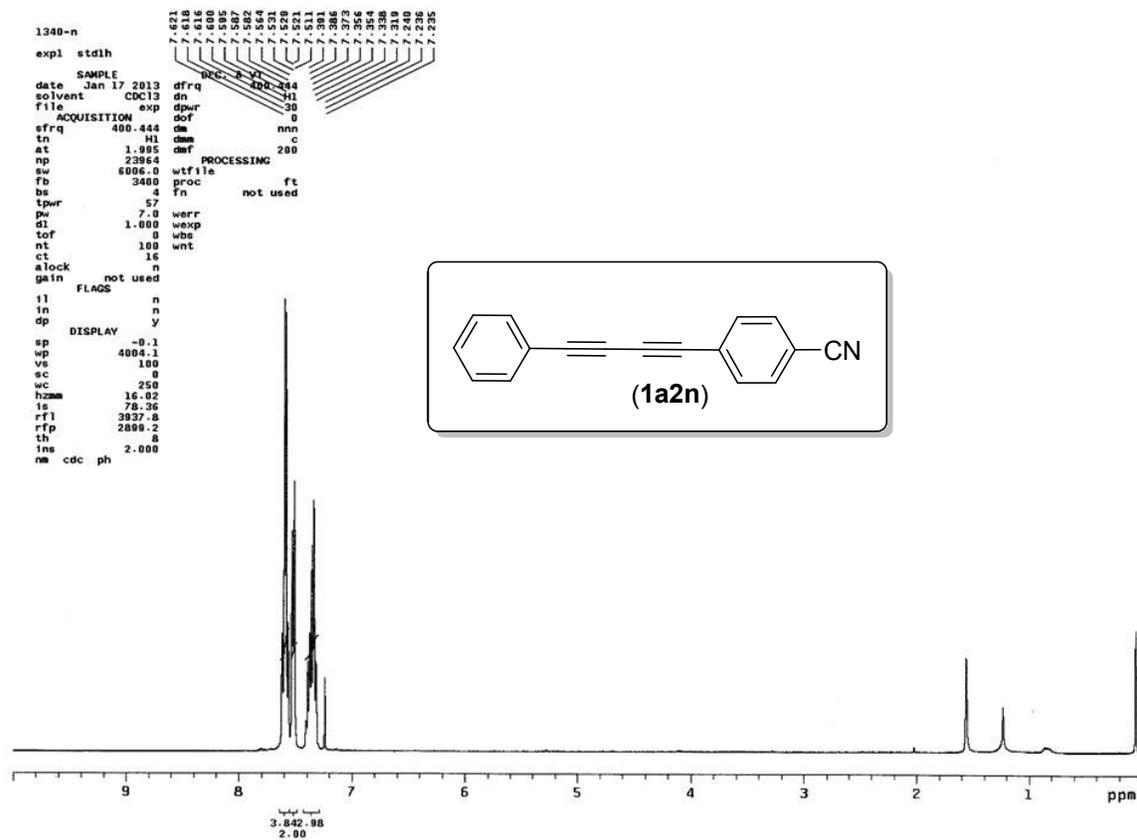


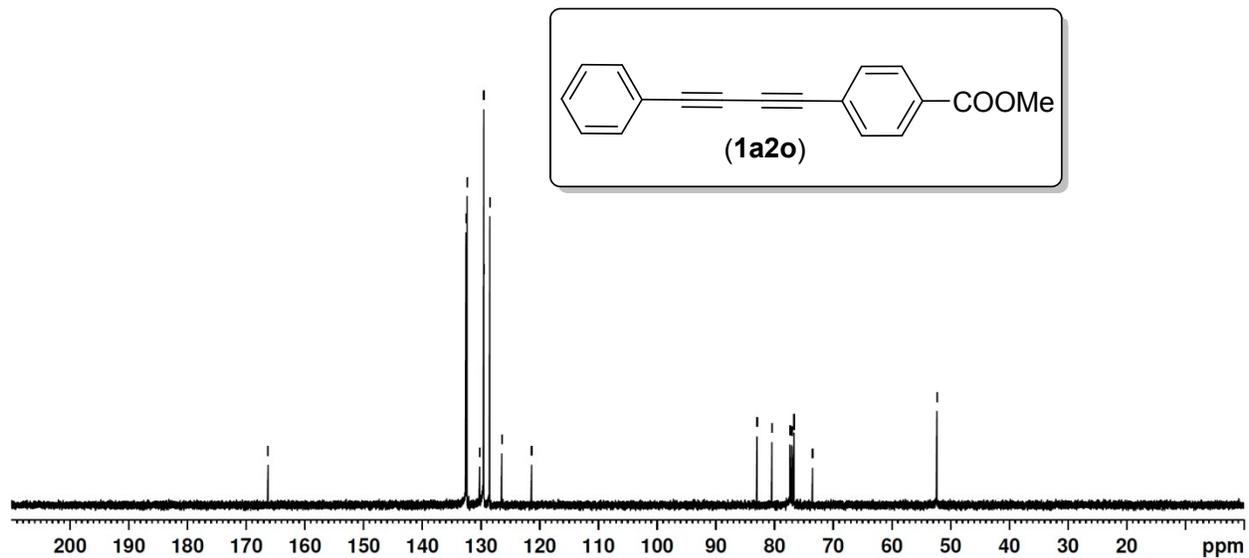
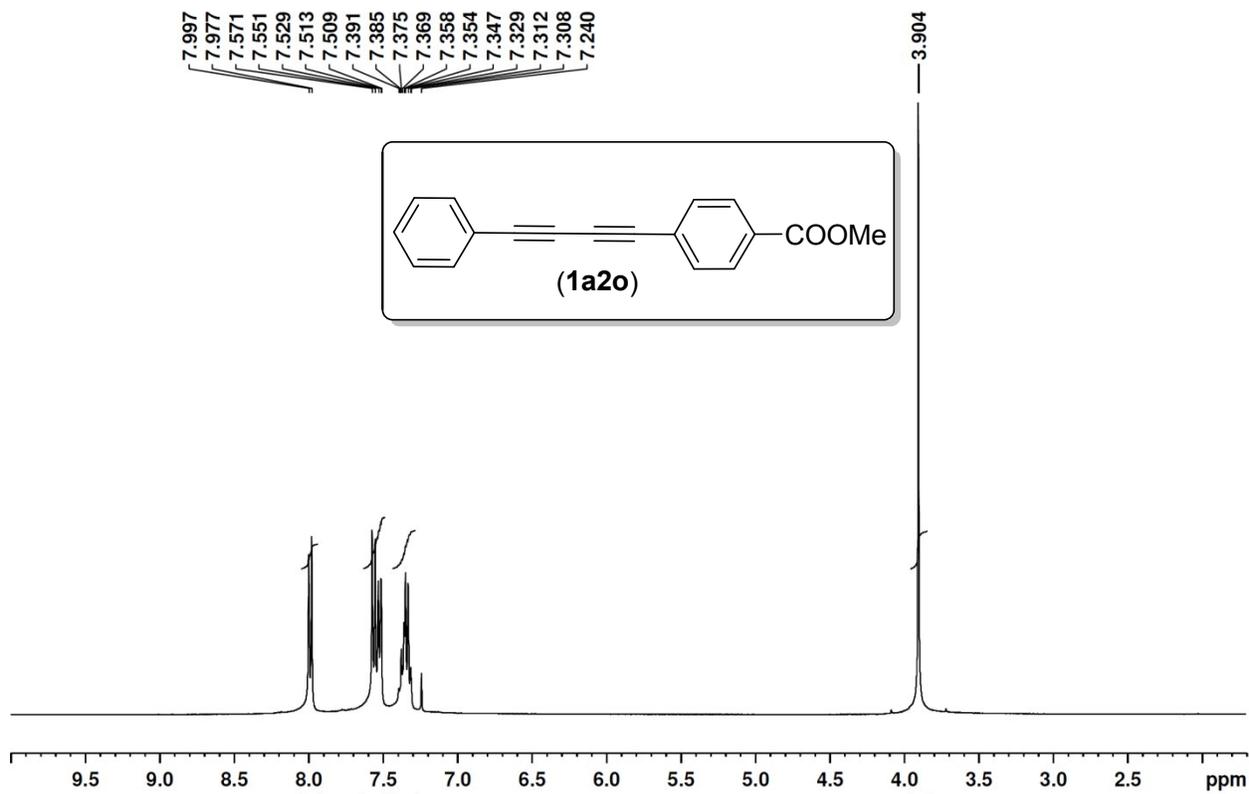
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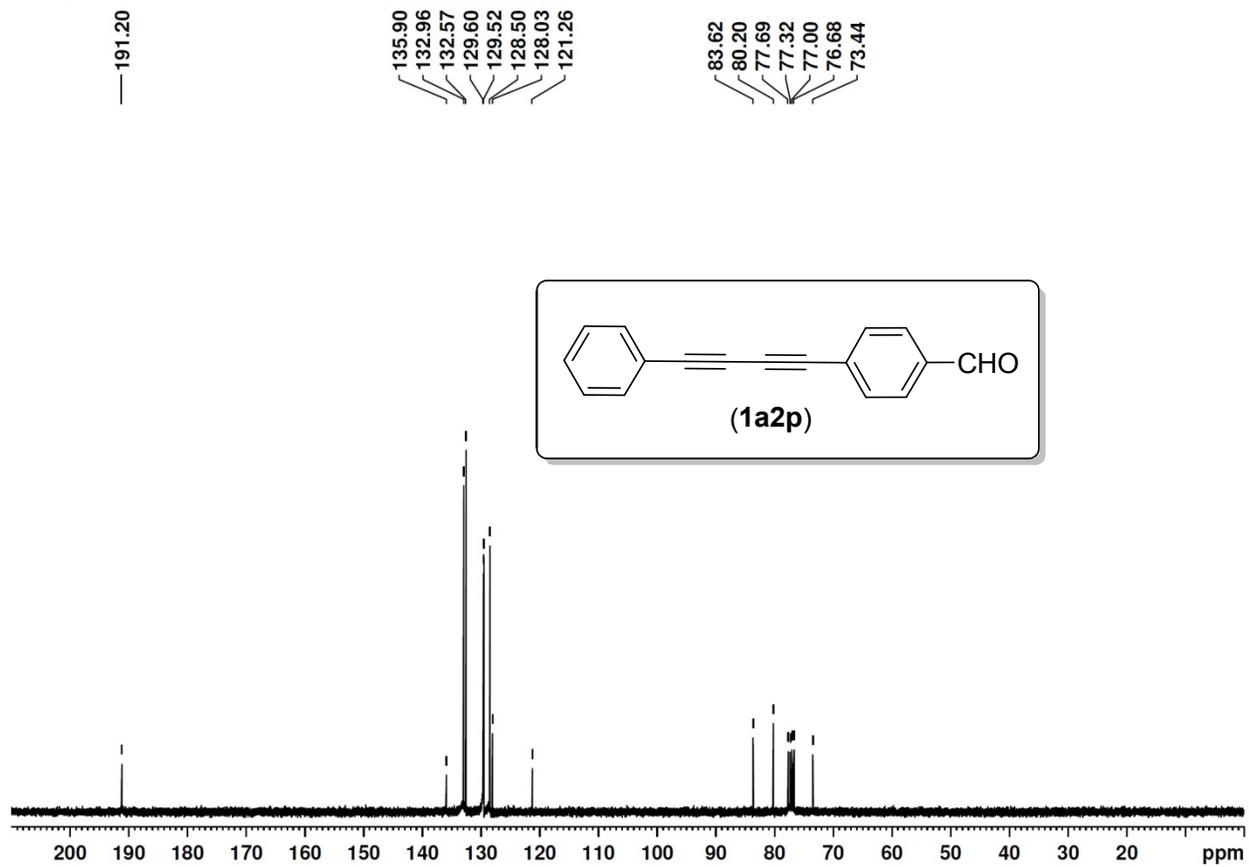
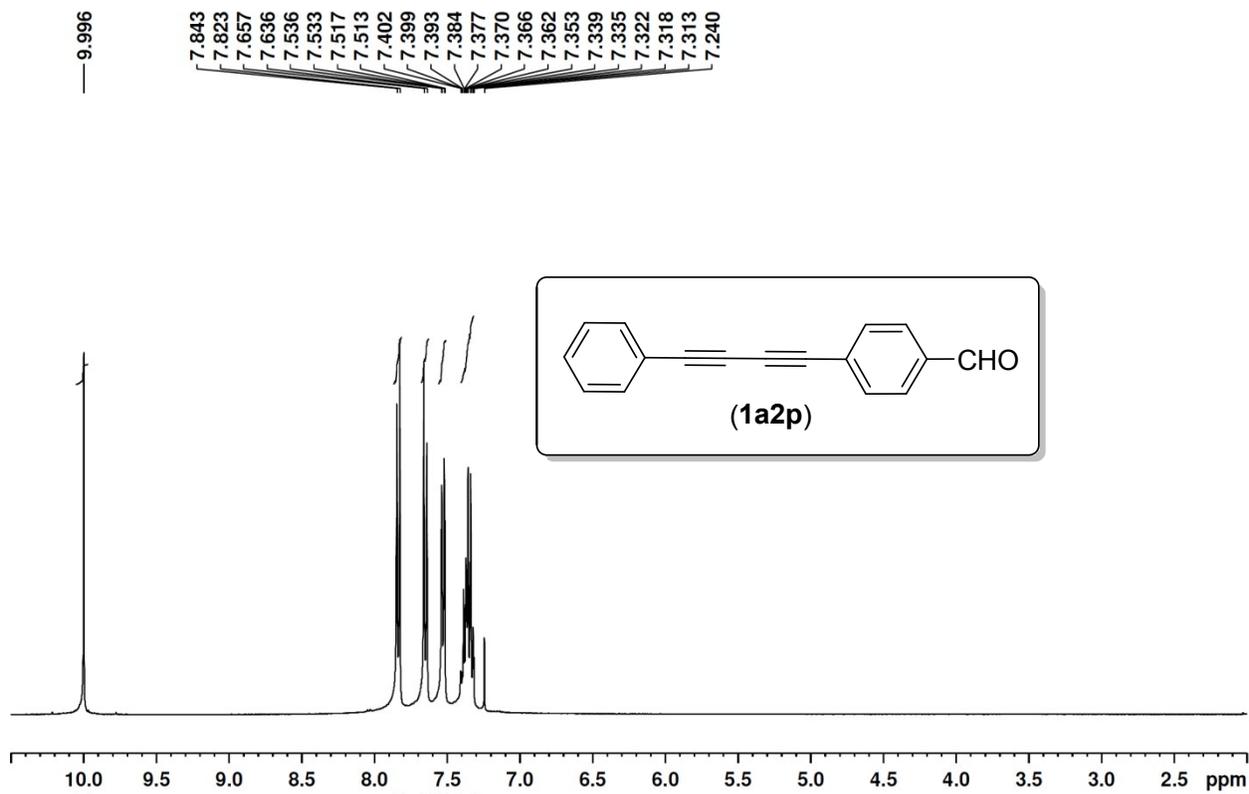
1340-a
exp5 std130
SAMPLE
date Jan 1 2013 dfrq 400.444
solvent CDC13 dn H1
file exp dpwr 39
ACQUISITION exp dof 0
sfrq 100.701 dm yyy
tn C13 dnm w
at 1.199 dmf 9000
np 59968 PROCESSING
sw 25000.0 lb 1.00
fb 13800 wtfile ft
bs 4 proc not used
tpwr 58
pw 8.7 werr
dl 0 wexp
tof 0 wbs
nt 10000 wnt
ct 360
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -167.2
wp 22151.9
vs 60
sc 0
wc 250
hzmm 88.51
is 500.00
rfl 10768.4
rfp 7753.2
th 2
ins 100.000
nm no ph

```







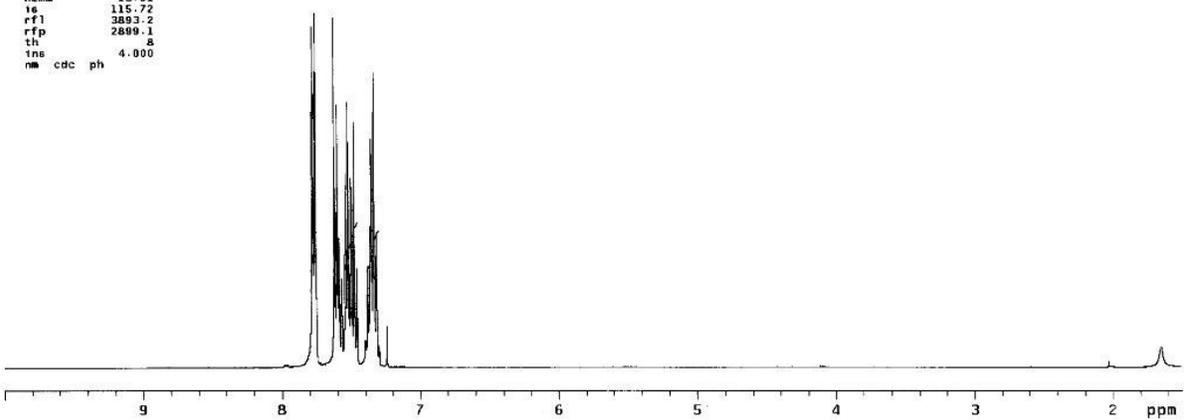
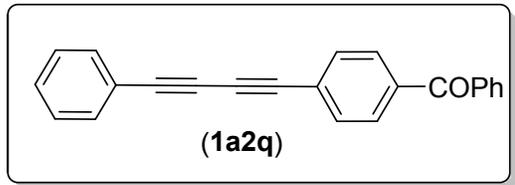


```

1401-1
exp4 std1h
SAMPLE
date Nov 23 2015 dfrq 400.436
solvent CDCl3 dn 200
file exp dpwr 30
ACQUISITION dof 0
sfrq 400.436 dm nnc c
tn H1 dnm 200
at 1.095 dmf PROCESSING
np 2396.4
sw 6006.0 wtf file ft
fb 3400 proc
bs 4 fn not used
tpwr 58
pw 7.2 werr
dl 1.000 wexp
tof 0 wbs
nt 100 wnt
ct 28
alock n
gain not used
FLAGS
ll n
in n
dp y
DISPLAY
sp 608.5
wp 3403.7
vs 75
sc 0
wc 250
hzma 13.61
ie 115.72
rf1 3083.2
rfp 2899.1
th 8
ins 4.000
nm cdc ph

```

7.761
7.756
7.751
7.610
7.588
7.587
7.582
7.571
7.568
7.540
7.537
7.517
7.510
7.503
7.496
7.476
7.467
7.378
7.370
7.365
7.358
7.356
7.351
7.337
7.333

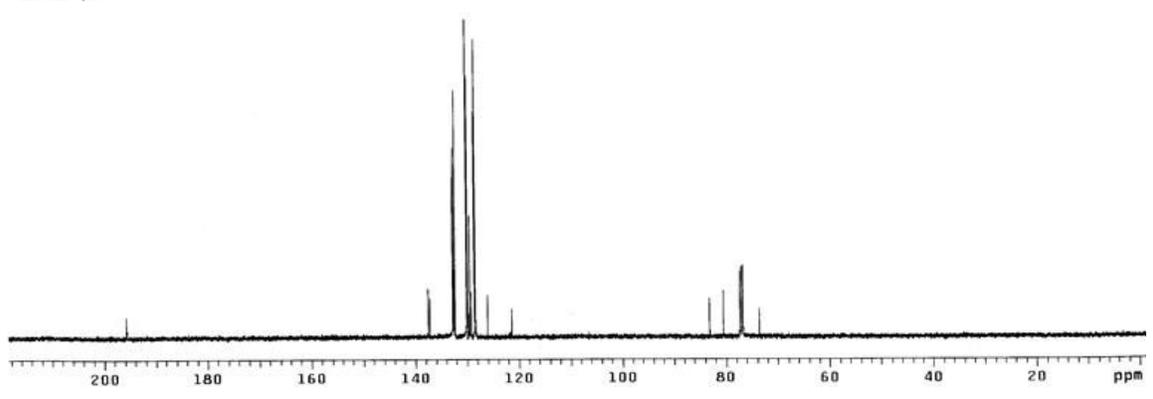
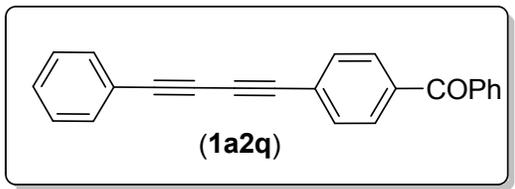


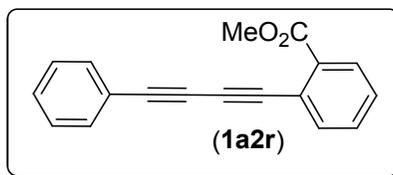
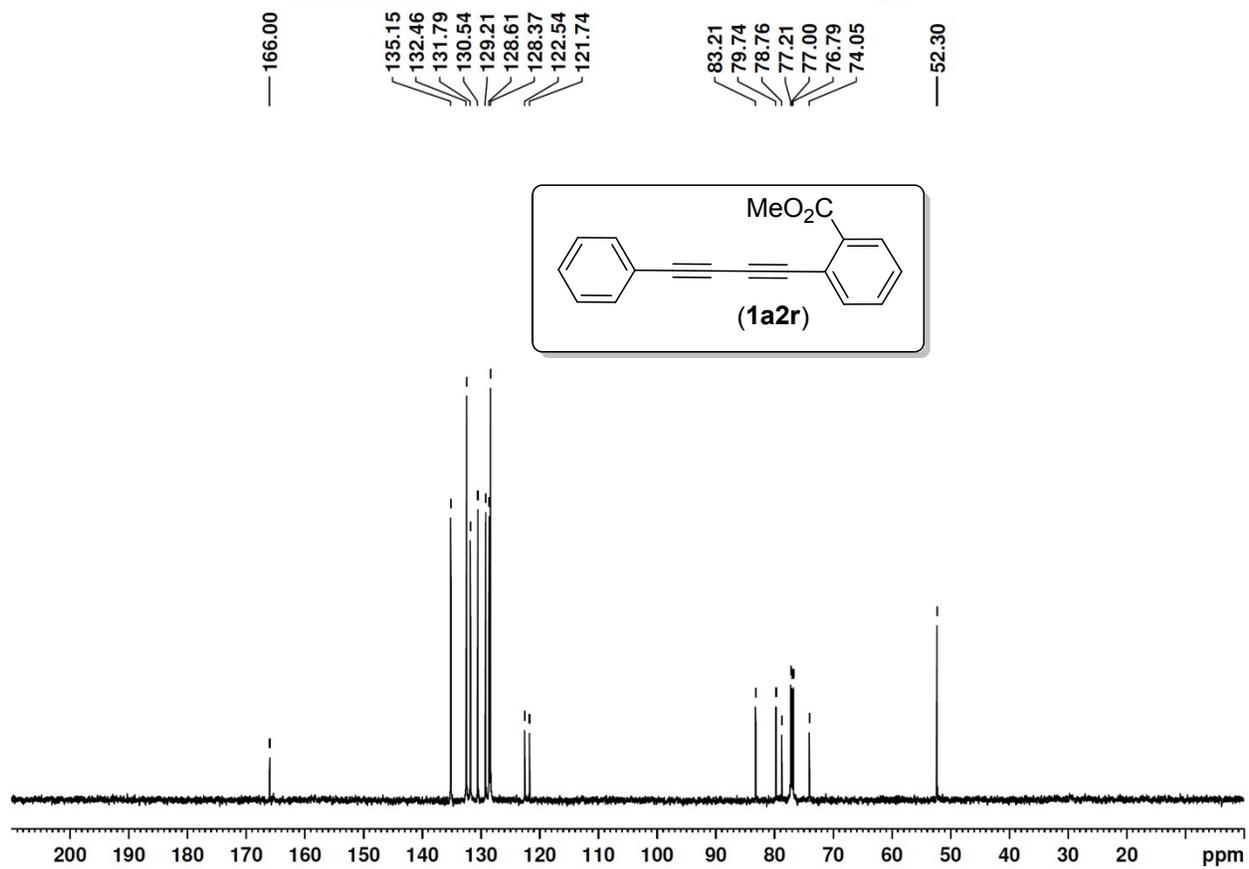
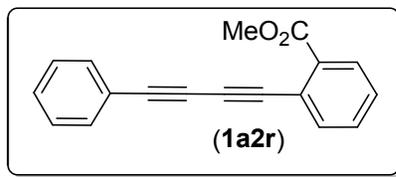
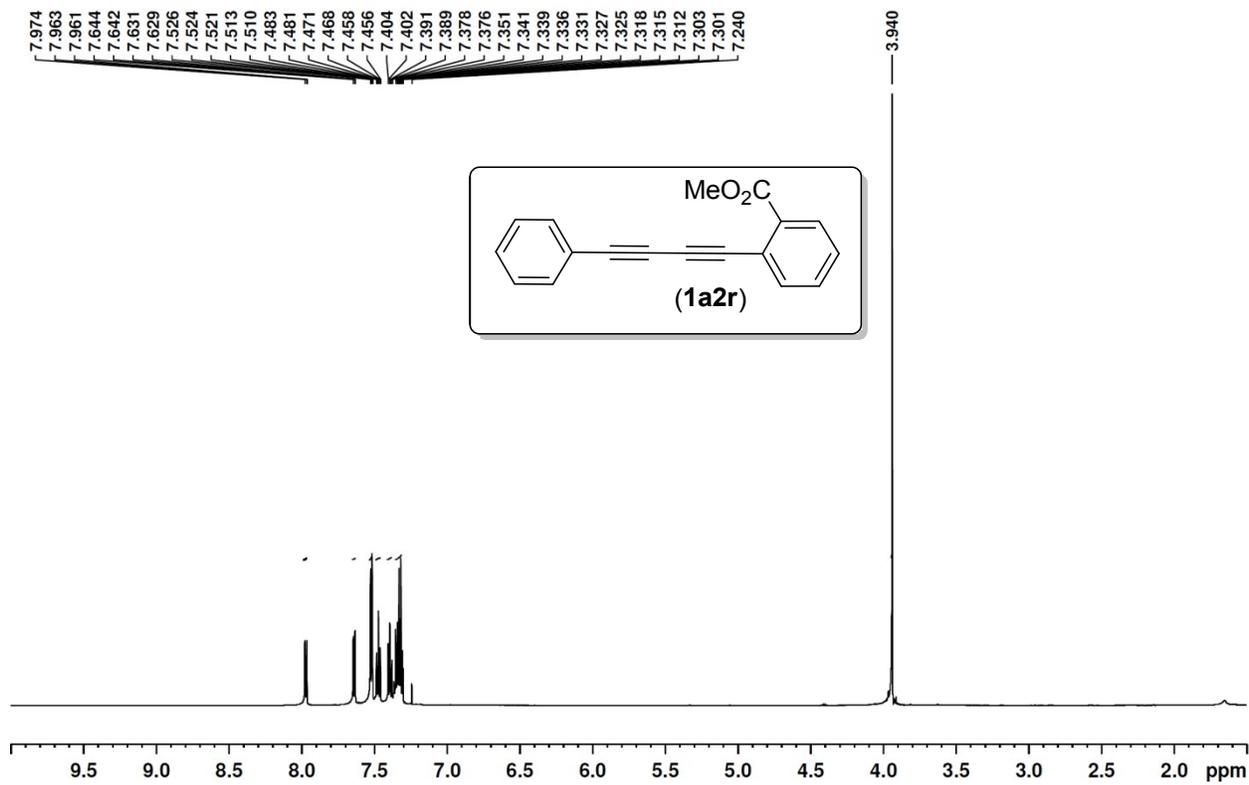
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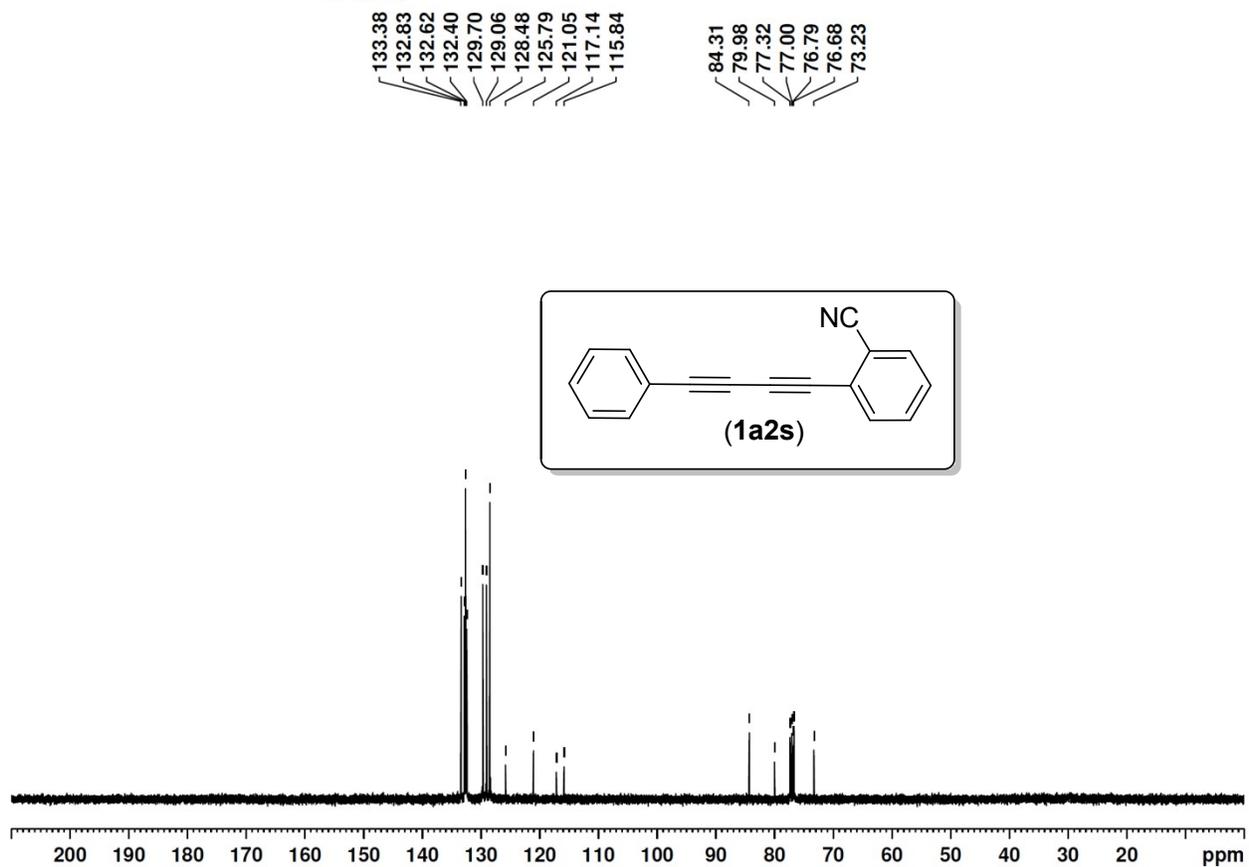
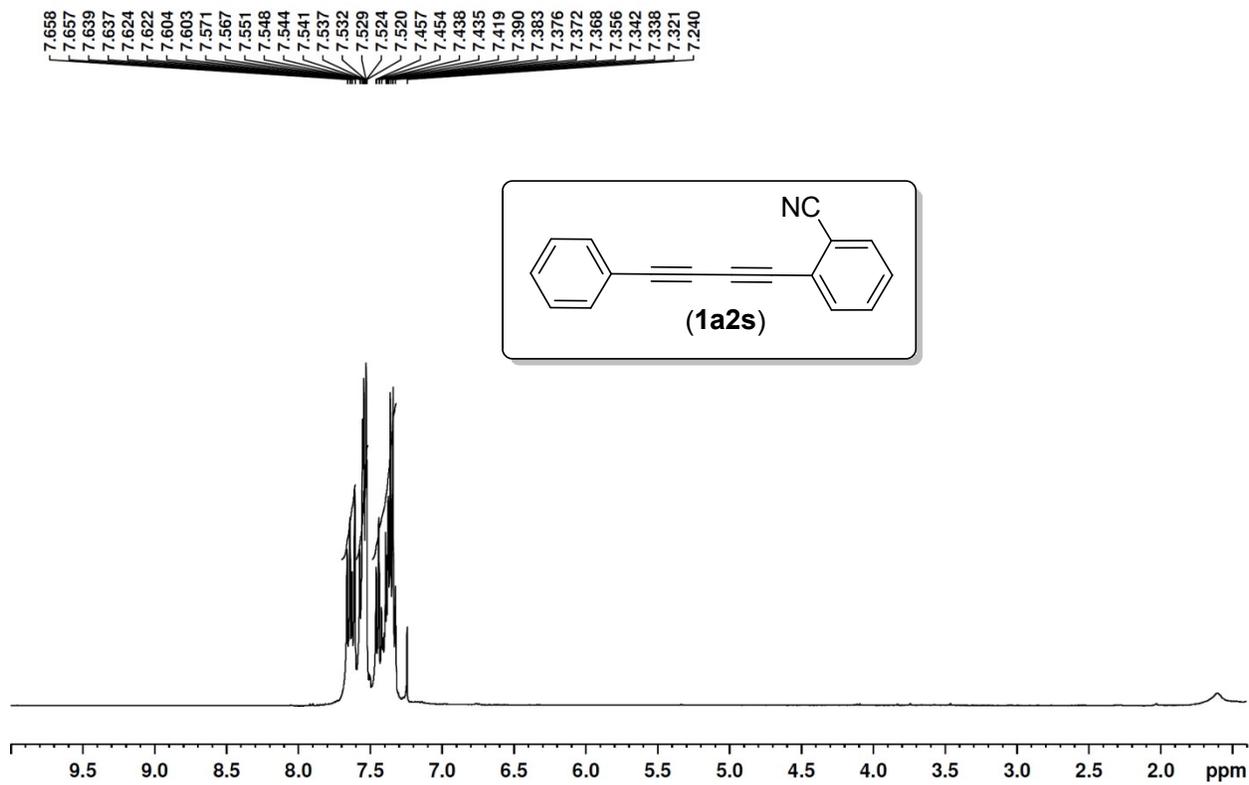
1401-1
exp5 std13c
SAMPLE
date Nov 23 2015 dfrq 400.436
solvent CDCl3 dn H1
file exp dpwr 42
ACQUISITION dof 0
sfrq 100.699 dm nny
tn C13 dnm 8500
at 1.198 dmf PROCESSING
np 5996.6 lb 1.00
sw 25800.0 wtf file ft
fb 13600 proc
bs 4 fn not used
tpwr 58
pw 4.8 werr
dl 0 wexp
tof 100000 wbs
nt 524 wnt
ct n
alock n
gain not used
FLAGS
ll n
in n
dp y
DISPLAY
sp -147.9
wp 22151.2
vs 70
sc 0
wc 250
hzma 88.61
ie 500.00
rf1 10749.8
rfp 7753.1
th 4
ins 100.000
nm no ph

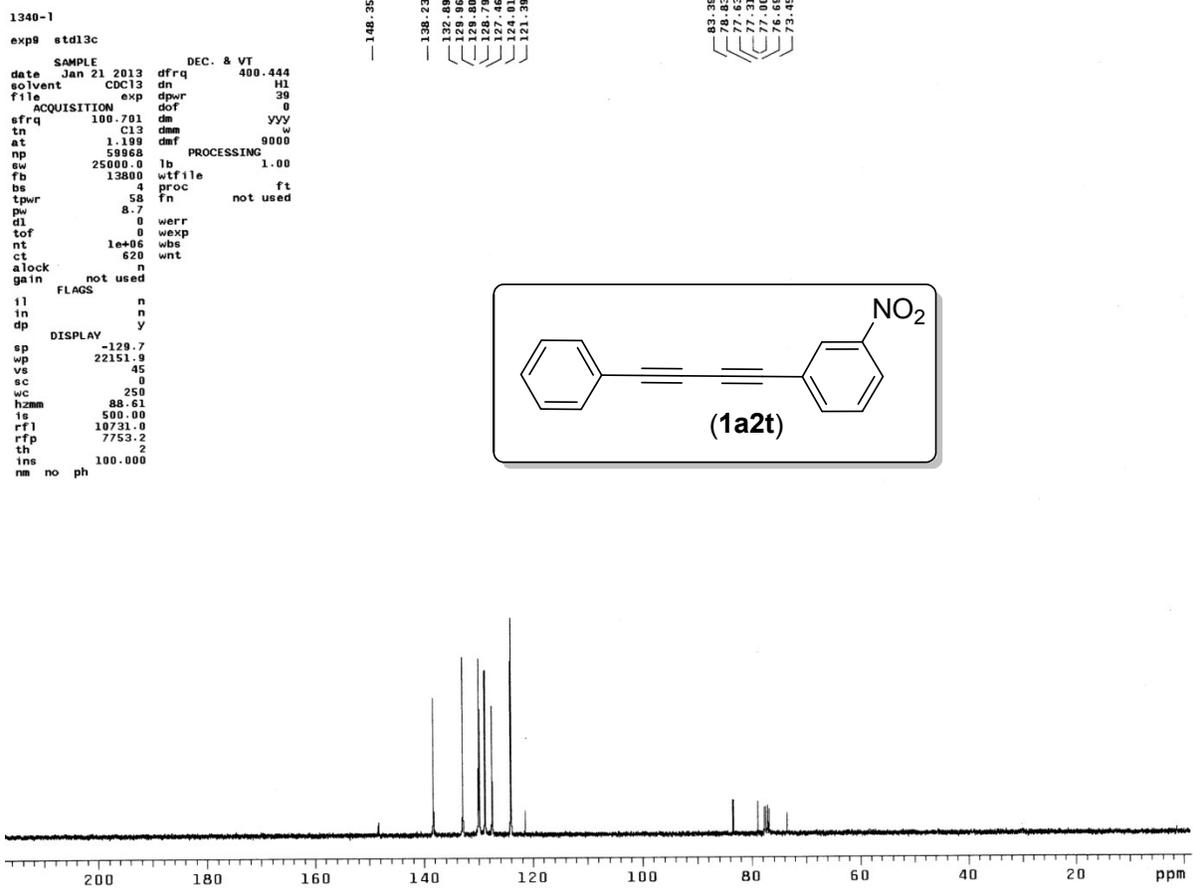
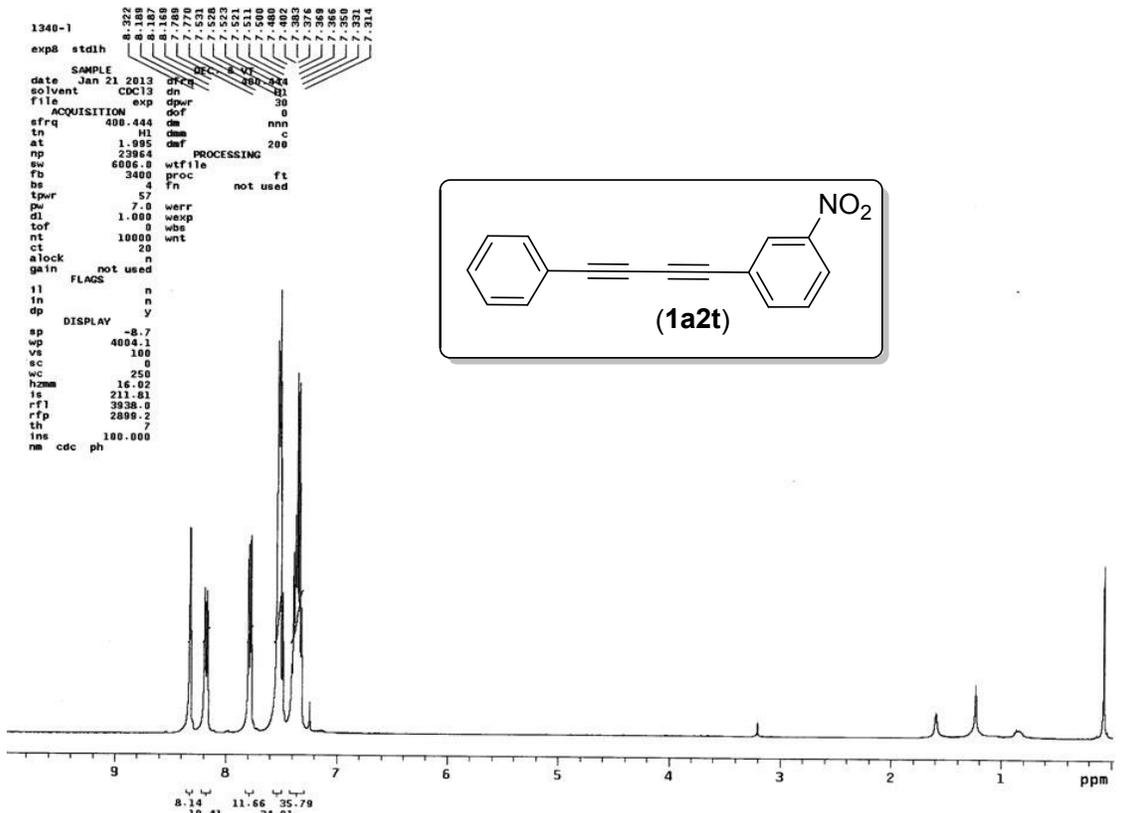
```

137.388
132.647
132.541
132.457
129.905
129.834
129.487
128.176
128.104
128.073
125.971
121.402
83.968
80.476
77.216
77.000
76.685
73.583





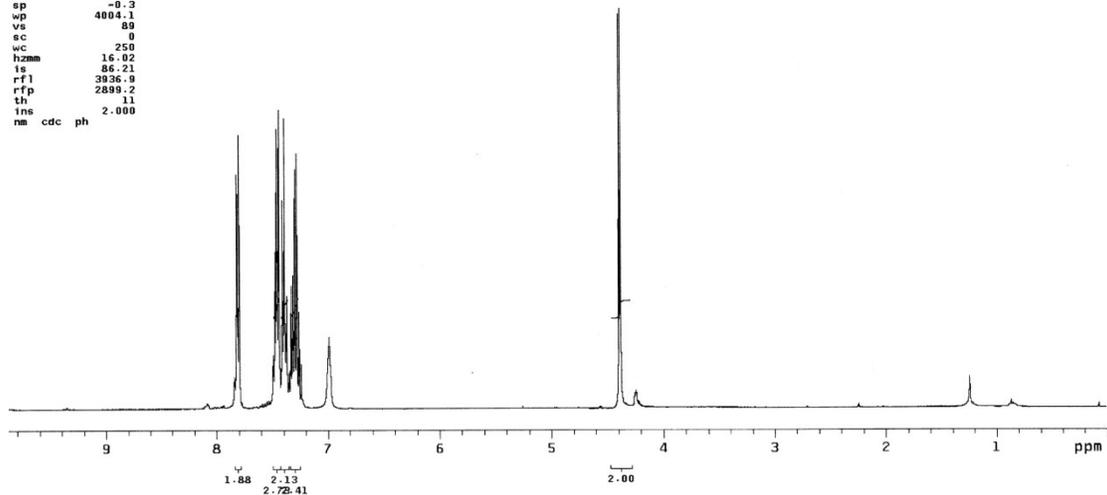
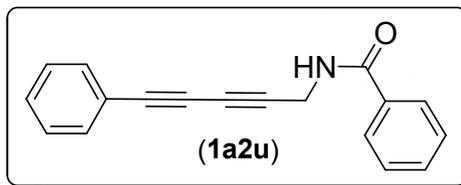




```

1340-s
exp8 std1h
SAMPLE
date Jan 19 2013 dfrq DEC. & VT 400.444
solvent CDC13 dn H1
file exp dpwr 39
ACQUISITION dof 0
sfrq 400.444 dm nmn
tn H1 dmm c
at 1.995 dmf PROCESSING 200
np 23964
sw 6006.0 wtf file
fb 3400 proc ft
bs 4 fn not used
tpwr 57
pw 7.0 werr
d1 1.000 wexp
tof 0 wbs
nt 100 wnt
ct 8
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -0.3
wp 4004.1
vs 89
sc 0
wc 250
hzmm 16.02
is 86.21
rfl 3936.9
rfp 2889.2
th 11
ins 2.000
nm cdc ph

```



```

1340-s
exp9 std13c
SAMPLE
date Jan 19 2013 dfrq DEC. & VT 400.444
solvent CDC13 dn H1
file exp dpwr 39
ACQUISITION dof 0
sfrq 100.701 dm yvy
tn C13 dmm w
at 1.199 dmf PROCESSING 9000
np 59968
sw 25000.0 lb 1.00
fb 13800 wtf file
bs 4 proc ft
tpwr 58 fn not used
pw 8.7
d1 0 werr
tof 0 wexp
nt 100000 wbs
ct 232 wnt
alock n
gain not used
FLAGS
il n
in n
dp y
DISPLAY
sp -3014.9
wp 24999.2
vs 70
sc 0
wc 250
hzmm 6.16
is 500.00
rfl 10768.9
rfp 7753.2
th 3
ins 100.000
nm no ph

```

