

Electronic Supplementary Information for

One-pot conversion of furfural to alkyl levulinate over
bifunctional Au-H₄SiW₁₂O₄₀/ZrO₂ without external H₂

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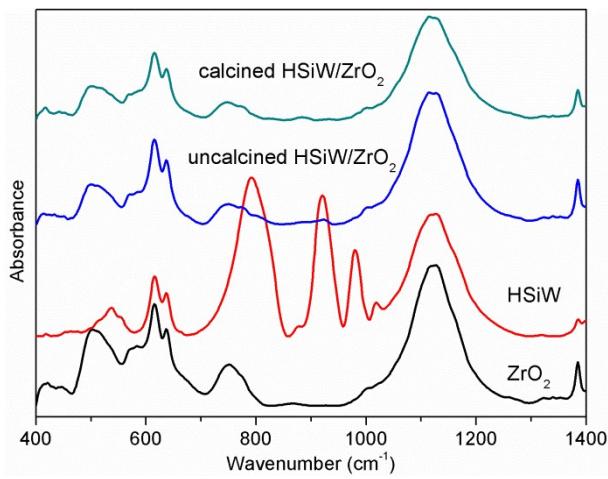


Fig. S1 FTIR spectra of calcined and uncalcined HSiW/ZrO₂.

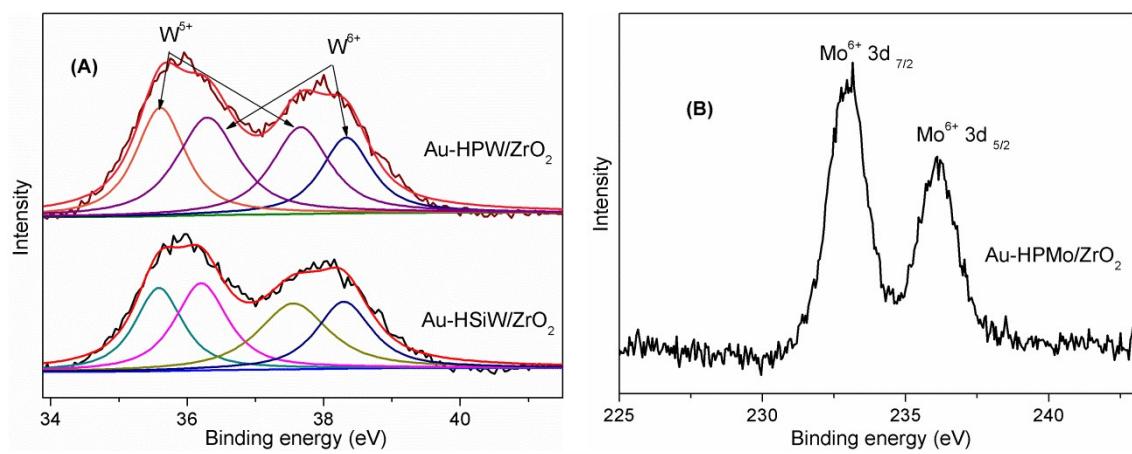


Fig. S2 (A) W^{6+} or W^{5+} 4f and (B) Mo^{6+} 3d XPS spectra of Au-HPA catalysts.

Table S1 The total acidity from titration method

Catalyst	Au/ZrO ₂	Au-HSiW/ZrO ₂	Au-HPW/ZrO ₂	Au-HPMo/ZrO ₂
Acidity (mmol/g)	0.33	1.84	1.51	1.10

The procedure of titration method is described below:^{S1-S2} 0.25 g catalyst was added into 30 mL 0.05 mol/L NaOH aqueous solution at room temperature under stirring for 60 min. After centrifugal separation, the supernatant solution was titrated by 0.05 mol/L HCl aqueous solution with phenolphthalein as an indicative.

Table S2 The turnover frequencies based on active Au or acidic sites

Catalyst	TOF _{Au} (mol _{furfural} ·mol _{Au} ⁻¹ ·h ⁻¹)	TOF _{acidity} (mol _{AL} ·mol _{acid} ·h ⁻¹)
Au-HSiW/ZrO ₂	0.188	0.024
Au-HPW/ZrO ₂	0.195	0.018
Au-HPMo/ZrO ₂	0.174	0.010

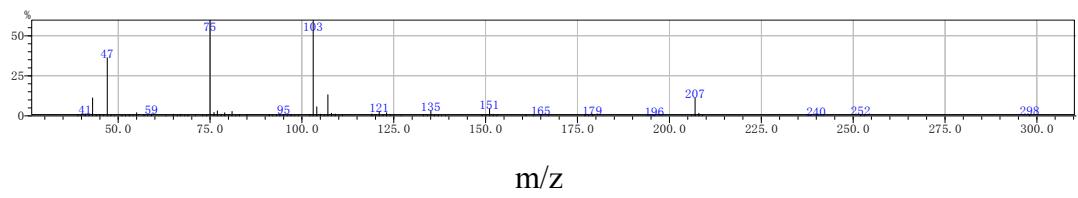


Fig. S3 The possible GC-MS data of TPP.

References:

- S1 J. Wang, W. Xu, J. Ren, X. Liu, G. Lu and Y. Wang, *Green Chem.*, 2011, **13**, 2678–2681.
- S2 P. P. Upare, J. Yoon, M. Y. Kim, H. Kang, D. W. Hwang, Y. K. Hwang, H. H. Kung and J. Chang, *Green Chem.*, 2013, **15**, 2935–2943.