Supplement Materials

Brønsted Acid Catalyzed Transoximation Reaction: synthesis of Aldoximes and Ketoximes without use of hydroxylamine salts

Kengo Hyodo,^{*} Kosuke Togashi, Naoki Oishi, Genna Hasegawa and Kingo Uchida Department of Material Chemistry, Faculty of Science and Technology, Ryukoku University, 1-5 Yokotani, Oe-cho,Seta, Otsu, Shiga 520-2194, Japan Fax: +81-77-543-7483; Tel: +81-77-543-7462; E-mail: hyodo@rins.ryukoku.ac.jp

Content

1. General Methods	S2
2. Optimization of Reaction Conditions	\$3
3. Synthetic Procedures	S4
4. Observation Experiments about Existence of Hydroxylamine Derivatives in-situ	S17
5. NMR Experiments for Hydrolysis of 2g	S23
6. Kinetic Studies	S28
7. Cross-over Reaction	S31
8. Retro Reaction	S33
9. References	S35
10. NMR Chart	S36

1. General Methods:

All reactions were performed in oven-dried glassware under a positive pressure of argon. Solvents were transferred via syringe and were introduced into the reaction vessels through a rubber septum. All reactions were monitored by thin-layer chromatography (TLC) carried out on 0.25 mm Merck silica-gel (60-F254). The TLC plates were visualized with UV light and 7% phosphomolybdic acid or basic KMnO₄ in H₂O/heat. Column chromatography was carried out on a column packed with silica-gel 60N spherical neutral size 63-210 µm. ¹H NMR (400 MHz), ¹⁹F NMR (376 MHz) and ¹³C NMR (100 MHz) spectra for solution in CDCl₃, or CD₂Cl₂ or CD₃CD d₆-DMSO were recorded on a JEOL RESONANCE JNM-ECS-400. Chemical shifts (δ) are expressed in ppm downfield from internal TMS or CHCl₃ or CH₂Cl₂ or CH₃CN or DMSO or C₆F₆. EI-MS were recorded on a SHIMADZU GC-MS QP-2010 Plus (Column: Frontier Lab UltraALLOY-5(MS/HT) (30 m, 0.25 mmID, 0.25 µm film), He: 30 psi mL/min, condition: 70 °C to 120 °C, 5 °C/min). ESI-MS were recorded on a SHIMADZU LC-MS-2020 or Agilent G1956B LCMS system with 1100 Series HPLC. Infrared spectra were recorded on a JASCO FT/IR-660 Plus. CH₂Cl₂ was used after distillation from CaH₂. The amount of water contented in distilled CH₂Cl₂ was measured by Karl-Fisher Moisture Titrator MKC 510N, Kyoto Electronics Manufacturing Co., Ltd. All aldehydes and ketones are commercial available. 70 wt% perchloric acid aqueous solution was purchased from Nacalai Tesque, Inc. Iron (II) perchlorate hexahydrate was purchased from Wako Pure Chemical Industries, Ltd. 2-Butanone oxime and ethyl acetohydroxamate were purchased from Tokyo Chemical Industry Co., Ltd. Anhydrous HClO4 NH2OH ethanol solution was purchased from Nacalai Tesque, Inc. All aldoximes and ketoximes were characterized by ¹H NMR to compare to previous report.¹⁻¹¹

2. Optimization of Reaction Conditions

Solvent effect

Table S1. Screening of solvents^a

	+ HO N - Me Me	HClO ₄ (10 mol%) Solvent, r.t., 24 h		он н
Entry	Solvent	E/Z (-) ^b	Yield (%) ^c	
1	CH_2Cl_2	99/1	65	
2	THF	-	0	
3	Toluene	96/4	37	
4	CH ₃ CN	92/8	52	
5	Et ₂ O	97/3	33	
6	Acetone	-	31	
7	EtOH	87/13	47	
8	H_2O	99/1	90	

a) Reaction conditions: 1a (0.16 mmol), 2a (0.24 mmol), 70 wt% HClO₄ (10 mol%) in CH₂Cl₂ at r.t. for 24 h. b) E/Z ratio was determined by ¹H NMR from crude mixtures. c) Isolated yield.

Catalyst effect

Table S2. Screening of perchloric acid and metal perchlorate salts^a



	Periodic table of the $M(ClO_4)_n \cdot xH_2O$													
Μ	1	2	3	4	5	6	7	8	9	10	11	12	13	14
1	H 96% yield E/Z=97/3 n=1, x=2.4		_											
2	Li 0% yield n=1, x=3	Be										В	С	NH ₄ 0% yield n=1, x=0
3	Na	Mg										Al 90% yield E/Z=96/4 n=3, x=9	Si	Р
4	K 0% yield n=1, x=0	Ca 65% yield E/Z=93/7 n=2, x=4	Sc 98% E/Z= 97/3 n=3, x=6	Ti	V	Cr	Mn 20% yield E/Z= 89/11 n=2, x=6	Fe 99% yield E/Z=96/4 n=2, x=6	Co 83% yield E/Z=99/1 n=2, x=6	Ni 62% yield E/Z=100/0 n=2, x=6	Cu 36% yield E/Z=86/14 n=2, x=6	Zn 70% yield E/Z= 99/1 n=2, x=6	Ga	Ge
5	Rb	Sr	Y	Zr	Nb	Мо	Тс	Ru	Rh	Pd	Ag 0% yield n=1, x=0	Cd	In	Sn

a)Reaction conditions: 1a (0.16 mmol), 2g (0.24 mmol), catalyst (10 mol%) in CH₂Cl₂ at r.t. for 24 h. E/Z ratio was determined by ¹H NMR from crude mixtures. Isolated yield.

3. Synthetic Procedures

General procedure for aldoxime synthesis via transoxoimation



To a 50 mL flask were added to aldehyde or ketone (11 mmol, 1.0 eq.), H₂O or EtOH (0.2 M), 70 wt% HClO₄ aq (5-30 mol%) under N₂ at indicated temperature in Table 3. Then methyl ethyl ketone (1.5 equiv.) was added to stirred reaction mixtures. After reaction mixtures turned to white suspension for 24 h, the white solid was collected by filtration and washed using water (10 mL). The collected product was dried over P_2O_5 under vacuum for 12 h to remove water.

If white suspension was not observed in the reaction mixtures after 24 hours, the reaction mixtures were extracted with AcOEt (50 mL×3). The combined organic layer was washed with brine (50 mL) and dried over Na₂SO₄. After removal of Na₂SO₄ from organic layer, solvent was evaporated under reduced pressure. The remained crude mixtures was purified by silica gel column chromatography (hexane/Ethyl acetate = 90/10 or hexane/Ethyl acetate = 50/50 or hexane/dichloromethane = 2/8 to dichloromethane/Et₂O = 95/5) and white solid product or colorless oil was obtained.

O II		N ^{°OH}	HCIO ₄ or H ₂ S	O ₄ (5 mol%)	N ^{°OH}	O II
R ³ R	⁴ ⁺ Et	儿 Me	H ₂ O, r.t.	., 24 h	R ³ R ⁴ Ι	≣t Me
1		2d			3	4d
Entry	1	3	HClO ₄ E/Z ^b	HClO ₄ Yield (%) ^c	H_2SO_4 E/Z^b	H_2SO_4 Yield (%) ^c
1	1a	3 a	99/1	95	99/1	94
2	1b	3 b	99/1	91	99/1	94
3	1c	3c	92/8	99	99/1	81
4	1d	3d	99/1	92	99/1	91
5	1e	3e	99/1	90	99/1	87
6	1f	3f	99/1	90	99/1	89
7	1g	3 g	99/1	97	99/1	90
8	1h	3h	99/1	95	99/1	86
9	1i	3i	52/48	98	19/81	91
10	1j	3j	99/1 ^e	93	99/1 ^e	85
11 ^d	1k	3k	54/46 ^e	85	32/68 ^e	99
12	1 l	31	99/1	95	99/1	85
13 ^f	1m	3m	99/1	74(93) ^g	99/1	58
14 ^h	1n	3n	76/24 ^e	82	71/29 ^e	76
15 ^f	10	30	88/12 ^e	70(97) ^g	89/11 ^e	52
16 ^f	1p	3p	-	$41(97)^{i}$	-	31
$17^{\rm h}$	1q	3q	-	93	-	90
18	1r	3r	64/36	84	66/34	80

Table S3. Scope of Substrates catalyzed by HClO₄ or H₂SO₄

a) Reaction conditions: **1** (11 mmol, gram scale), **2d** (1.5 equiv.), HClO₄ (5 mol%) in H₂O (0.20 M) at r.t. b) E/Z ratio was determined by ¹H NMR. c) Isolated yield. d) 10 mol% catalyst was used at 40°C. e) This means the ratio of major and minor isomer. f) 30 mol% catalyst was used in EtOH at reflux. g) **2g** was used instead of **2d** with 5 mol% HClO₄ at r.t. h) 10 mol% catalyst was used at reflux. i) **2g** was used instead of **2d** at 40°C.

(*E*)-2- Naphthaldoxime $(3a)^{11}$



¹H NMR (400 MHz, CDCl₃) δ 7.48-7.52 (m, 2H), 7.84-7.86 (m, 4H), 7.87 (s, 1H), 8.29 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 122.9, 126.8, 127.2, 128.0, 128.5, 128.8, 128.9, 129.8, 133.3, 134.3, 150.7; MS (ESI, negative) m/z calcd. for C₁₁H₈NO [M-H]: 170.06, Found 170.10.

(E)-Benzaldoxime (3b)²⁾

N_OH

¹H NMR (400 MHz, CDCl₃) δ 7.37-7.41 (m, 3H), 7.57-7.59 (m, 2H), 8.16 (s, 1H), 8.38 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 127.2, 128.9, 130.2, 132.0, 150.5; MS (APCI, negative) m/z calcd. for C₇H₆NO [M-H]: 120.05, Found 120.05.

(*E*)-4-Anisaldoxime $(3c)^{2}$



¹H NMR (400 MHz, CDCl₃) δ 3.83 (s, 3H), 6.90 (d, J = 8.5 Hz, 2H), 7.52 (d, J = 8.5 Hz, 2H), 8.11 (s, 1H), 8.67 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 55.4, 114.4, 124.6, 128.7, 150.0, 161.2; MS (ESI, negative) m/z calcd. for C₈H₈NO₂ [M-H]: 150.06, Found 150.10.

(*E*)- α , α , α -Trifluoro-4-tolubenzaldoxime (3d)³⁾



¹H NMR (400 MHz, CDCl₃) δ 7.65 (d, J = 8.5 Hz, 2H), 7.70 (d, J = 8.5 Hz, 2H), 8.18 (br, 1H), 8.19 (s, 1H); ¹⁹F NMR (376 MHz, CDCl₃) δ -66.0 (s, 3F); ¹³C NMR (100 MHz, CDCl₃) δ 124.0 (q, $J_{C-F} = 271$ Hz), 125.9 (q, $J_{C-F} = 3.8$ Hz), 127.4, 131.9 (q, $J_{C-F} = 32$ Hz), 135.4, 149.3; MS (ESI, negative) m/z calcd. for C₈H₃F₃NO [M-H]: 188.03, Found 188.05.

(*E*)-4-Fluorobenzaldoxime $(3e)^{2}$



¹H NMR (400 MHz, CDCl₃) δ 7.08 (m, 2H), 7.57 (m, 2H), 8.13 (s, 1H), 8.29 (br, 1H); ¹⁹F NMR (376 MHz, CDCl₃) δ -113.23 (s, 1F); ¹³C NMR (100 MHz, CDCl₃) δ 116.1 (d, $J_{C-F} = 21.9$ Hz), 128.2 (d, $J_{C-F} = 3.3$ Hz), 129.1 (d, $J_{C-F} = 8.3$ Hz), 149.5 (d, $J_{C-F} = 1.2$ Hz), 164.0 (d, $J_{C-F} = 249$ Hz); MS (ESI, negative) m/z calcd. for C₇H₅FNO [M-H]: 138.04, Found 138.10.

(E)-4-Chlorobenzaldoxime (3f)²⁾



¹H NMR (400 MHz, CDCl₃) δ 7.36 (d, J = 8.5 Hz, 2H), 7.51 (d, J = 8.5 Hz, 2H), 8.12 (s, 1H), 8.40 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 128.4, 129.2, 130.4, 136.2, 149.5; MS (ESI, negative) m/z calcd. for C₇H₅ClNO [M-H]: 154.01, Found 154.05.

(E)-3-Chlorobenzaldoxime (3g)³⁾



¹H NMR (400 MHz, CDCl₃) δ 7.30-7.38 (m, 2H), 7.44 (d, *J* = 7.3 Hz, 1H), 7.59 (s, 1H), 8.10 (s, 1H), 8.24 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 125.4, 127.0, 130.2, 130.2, 133.7, 135.0, 149.4; MS (ESI, negative) m/z calcd. for C₇H₅ClNO [M-H]: 154.01, Found 154.00.

(E)-2-Chlorobenzaldoxime (3h)²⁾



¹H NMR (400 MHz, CDCl₃) δ 7.25-7.34 (m, 2H), 7.39 (d, *J* = 8.0 Hz, 1H), 7.82 (dd, *J* = 1.8, 7.7 Hz, 1H), 8.42 (br, 1H), 8.58 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 127.2, 127.3, 129.7, 130.1, 131.2, 134.1, 147.8; MS (ESI, negative) m/z calcd. for C₇H₅ClNO [M-H]: 154.01, Found 154.00.

trans-Cinnamaldoxime (3i)²⁾



(*E*)-**3i**: ¹H NMR (400 MHz, CDCl₃) δ 6.84 (d, *J* = 5.4 Hz, 2H), 7.28-7.38 (m, 3H), 7.46 (d, *J* = 7.0 Hz, 2H), 7.95 (t, *J* = 4.6 Hz, 1H), 8.50 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 121.8, 127.1, 128.9, 129.1, 135.9, 139.2, 152.1; MS (ESI, negative) m/z calcd. for C₉H₈NO [M-H]: 146.06, Found 146.10.

(*Z*)-**3i**: ¹H NMR (400 MHz, CDCl₃) δ 6.88 (d, *J* = 16.0 Hz, 1H), 7.27-7.47 (m, 5H), 7.50-7.54 (m, 2H), 8.52 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 115.8, 127.7, 128.9, 129.5, 135.9, 140.3, 149.1; MS (ESI, positive) m/z calcd. for C₉H₁₀NO [M+H]: 148.08, Found 148.10.

3-Phenylpropionaldoxime (**3j**)⁶⁾



3j (minor): ¹H NMR (400 MHz, CDCl₃) δ 2.51-2.56 (m, 2H), 2.81-2.84 (m, 2H), 7.19-7.31 (m, 5H), 7.46 (dt, J = 1.8, 5.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 31.3, 32.9, 126.3, 128.5, 128.6, 140.6, 151.5. These NMR data for **3j** was obtained from E and Z mixtures NMR chart because we cannot isolate only **3j** (minor) which was easily isomerize **3j** (major).

3j (major): ¹H NMR (400 MHz, CDCl₃) δ 2.68-2.73 (m, 2H), 2.80-2.84 (m, 2H), 6.75 (t, *J* = 5.3 Hz, 1H), 7.19-7.32 (m, 5H), 8.93 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 26.5, 32.1, 126.4, 128.4, 128.7, 140.8, 151.9; MS (APCI, positive) m/z calcd. for C₉H₁₂NO [M+H]: 150.08, Found 150.10.

Decanal oxime (3k)⁷⁾



3k (major): ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H), 1.22-1.39 (m, 12H), 1.49 (quin, *J* = 7.4 Hz, 2H), 2.20 (td, *J* = 7.4, 6.3 Hz, 2H), 7.42 (t, *J* = 6.3 Hz, 1H), 8.38 (br, 1H).

3k (minor): ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, *J* = 6.6 Hz, 3H), 1.22-1.39 (m, 12H), 1.49 (quin, *J* = 7.4 Hz, 2H), 2.38 (td, *J* = 7.4 Hz, 5.5 Hz, 2H), 6.72 (t, *J* = 5.5 Hz, 1H), 8.81(br, 1H).

¹³C NMR (100 MHz, CDCl₃) δ 14.2, 22.8, 25.1, 26.2, 29.40, 29.44, 29.5, 29.6, 32.0, 153.1

MS (APCI, positive) m/z calcd for $C_{10}H_{22}NO$ [M+H]: 172.16, Found 172.20.

2-Thiophenecarbaldoxime (3l)²⁾

(*E*)-**31**: ¹H NMR (400 MHz, d_6 -DMSO) δ 7.12 (dd, J = 1.3, 0.9 Hz, 1H), 7.47 (d, J = 0.9 Hz, 1H), 7.73 (dd, J = 1.3, 0.2 Hz, 1H), 7.84 (s, 1H), 11.9 (s, 1H); ¹³C NMR (100 MHz, d_6 -DMSO) δ 139.8, 131.1, 131.08, 131.04, 126.2; MS (APCI, positive) m/z calcd for C₅H₆NOS [M+H]: 128.01, Found 128.10.

(*Z*)-**31**: ¹H NMR (400 MHz, CDCl₃) δ 7.11 (dd, *J* = 1.3, 3.6 Hz, 1H), 7.42 (d, *J* = 3.6 Hz, 1H), 7.59 (d, *J* = 5.1 Hz, 1H), 7.76 (s, 1H), 9.38 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 126.4, 131.0, 131.9, 132.1, 141.3; MS (ESI, negative) m/z calcd. for C₅H₄NOS [M-H]: 126.002, Found 126.05.

Acetophenoxime (3m)⁸⁾

N_OH

(*E*)-**3m**: ¹H NMR (400 MHz, CDCl₃) δ 2.31 (s, 3H), 7.37-7.41 (m, 3H), 7.60-7.63 (m, 2H), 9.69 (br, 1H); MS (ESI, positive) m/z calcld. for C₈H₁₀NO [M+H]: 136.08, Found 136.1.

2-Undecanoxime (3n)⁹⁾

N~OH

3n (major): ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, *J* = 7.0 Hz, 3H), 1.26 (m, 12H), 1.50 (t, *J* = 7.2 Hz, 2H), 1.88 (s, 3H), 2.18 (t, *J* = 7.8 Hz, 2H), 9.19 (br, 1H): ¹³C NMR (100 MHz, CDCl₃) δ 13.5, 14.2, 22.8, 26.4, 29.3, 29.4, 29.5, 29.6, 32.0, 36.0, 158.8; MS (ESI, positive) m/z calcld. for C₁₁H₂₄NO [M+H]: 186.19, Found 186.2. **3n** (minor): ¹H NMR (400 MHz, CDCl₃) δ 0.88 (t, *J* = 7.1 Hz, 3H), 1.26-1.30 (m, 12H), 1.50 (t, *J* = 7.7 Hz, 2H), 1.86 (s, 3H), 2.36 (t, *J* = 7.6 Hz, 2H), 8.36 (br, 1H): ¹³C NMR (100 MHz, CDCl₃) δ 14.3, 20.0, 22.8, 25.7, 28.7, 29.44, 29.54, 29.63, 29.86, 32.0, 159.5; MS (ESI, positive) m/z calcld. for C₁₁H₂₄NO [M+H]: 186.19, Found 186.2.

Phenacyl Chloride oxime (30)¹⁰⁾



30 (major) : ¹H NMR (400 MHz, CDCl₃) δ 4.62 (s, 2H), 7.42-7.44 (m, 3H), 7.67-7.70 (m, 2H), 9.50 (br, 1H): ¹³C NMR (100 MHz, CDCl₃) δ 32.4, 126.4, 128.9, 130.2, 133.4, 154.4; MS (ESI, positive) m/z calcld. for C₈H₉ClNO [M+H]: 170.04, Found 170.0.

Benzophenoxime (**3p**)¹¹⁾



¹H NMR (400 MHz, CDCl₃) δ 7.29-7.50 (m, 10H), 8.50 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 128.0, 128.4, 128.5, 129.26, 129.35, 129.7, 132.8, 136.3, 158.2; MS (ESI, positive) m/z calcld. for C₁₃H₁₂NO [M+H]: 198.09, Found 198.1.

Cyclohexanone oxime (3q)⁸⁾



¹H NMR (400 MHz, CDCl₃) δ 1.61-1.67 (m, 6H), 2.22 (t, *J* = 5.8 Hz, 2H), 2.51 (t, *J* = 5.8 Hz), 9.32 (br, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.5, 25.7, 25.9, 26.9, 32.2, 160.8; MS (ESI, positive) m/z calcld. for C₆H₁₂NO [M+H]: 114.09, Found 114.20.

Benzylideneacetone oxime (3r)¹²⁾



3r (major) : ¹H NMR (400 MHz, CDCl₃) δ 7.48 (d, J = 7.3, 2H), 7.35 (t, J = 7.7, 2H), 7.30-7.25 (m, 1H), 6.91 (d, J = 16.4 Hz, 1H), 6.86 (d, J = 16.4 Hz, 1H), 2.16 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 156.9, 136.4, 133.6, 128.9, 128.8, 128.6, 127.0, 125.9, 9.9; MS (ESI, positive) m/z calcld. for C₁₀H₁₂NO [M+H]:162.09, Found 162.10.

3r (minor) : ¹H NMR (400 MHz, CDCl₃) δ 7.64 (d, J = 16.6 Hz, 1H), 7.59-7.53 (d, J = 7.4, 2H), 7.38-7.30 (m, 3H), 6.95 (d, J = 16.6 Hz, 1H), 2.15 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.5, 136.7, 136.4, 129.3, 128.9, 128.6, 127.7, 117.0, 17.0.MS (ESI, positive) m/z calcld. for C₁₀H₁₂NO [M+H]:162.09, Found 162.10.

Scalability and synthetic application

Preparation of the 2-Chlorobenzaldoxime (3h) on 100 g scale



100 g scale

To a three necked 2 L flask were added to 2-chloroaldehyde (100 g, 0.711 mol, 1.0 eq.), H_2O (0.71 L, 1.0 M), methyl ethyl ketone (76.4 mL, 0.853 mol, 1.2 eq.) under argon at room temperature (23–25°C). Then, 70 wt% HClO₄ aq (2.92 mL, 36.0 mmol, 5 mol%) was dropwised to stirred reaction mixtures vigorously. After 24 hours, the reaction mixture was cooling to 15°C in order to solidify product, the white solid appeared was collected by filtration and washed using water (0.75 L). The collected product was dried over KOH under vacuum at r.t. for 24 h to remove water. The white product was obtained 103 g (94% yield). The remained filtrate aqueous solution was distilled by evaporator (pressure: 77 mmHg, bath temp.: 40°C) and methyl ethyl ketone was recovered 24 mL (37% recovery).



Figure S1. 100 g scale reaction in 2 L flask

Preparation of ethyl *O*-allyl-acetohydroxamate (6)¹³⁾



Allyl bromide (0.93 mL, 9.70 mmol, 1.1 equiv.) was added to a solution of ethyl acetohydroxamate (7.5 g, 72.8 mmol) and K₂CO₃ (10.2 g, 72.8 mmol, 1.0 equiv.) in Acetone (14.6 ml, 5.0 M) with stirring at r.t. After addition was complete, the reaction mixtures was stirred under reflux for 24 h. After the starting materials was consumed, the reaction mixtures was cooling to r.t. and water (3.9 mL) was added. After removal of acetone by evaporator, water phase was extracted with Et₂O (20 mL×3). The combined organic layer was washed with 10% NaOH aq. (20 mL) and dried over K₂CO₃. After removal of K₂CO₃ from organic layer, solvent was evaporated under reduced pressure. The remained crude mixtures was purified by silica gel column chromatography (pentane only to pentane /Et₂O = 90/10) and yellow solid product was obtained (4.13 g, 64%).

¹H NMR (CDCl₃, 400 MHz, ppm) δ 1.27 (t, *J* = 7.1 Hz, 3H), 1.95 (s, *J* = 7.1 Hz, 3H), 4.01 (q, *J* = 7.1 Hz, 2H),

4.41 (d, J = 4.4 Hz, 2H), 5.18 (d, J = 15.4 Hz, 1H), 5.27 (d, J = 15.4 Hz, 1H), 5.94-6.03 (m, 1H); ¹³C NMR (CDCl₃, 100 MHz, ppm) δ 13.9, 14.5, 62.3, 74.6, 117.3, 134.8, 162.7; MS (ESI, positive) m/z calcd for C₇H₁₄NO₂ [M+H]: 144.10, Found 144.20.

3-Hydroxy-17-*O***-allyl-oximino-estra-1**, **3**, **5**(10)-triene (7)¹⁴⁾



To a 50 mL flask were added to estrone **5** (2.5 g, 9.24 mmol, 1.0 eq.), EtOH (9.2 mL, 1.0 M), 70 wt% HClO₄ aq. (0.24 mL, 2.77 mmol, 30 mol%) under N₂ at room temperature. Then ethyl *O*-allyl-acetohydroxamate **6** (1.73 g, 13.9 mmol, 1.5 equiv.) was dropwised to stirred reaction mixtures at 40 °C. After 18 hours, the reaction mixtures was passing through silica/celite® pad and washed with Et₂O (20 mL). The collected solution was evaporated and dried under vacuum. The crude mixture was purified by silica gel column chromatography (hexane/Et₂O = 70/30 to 60/40) and white solid product was obtained (2.6 g, 93% yield).

¹H NMR (CDCl₃, 400 MHz, ppm) δ 0.94 (s, 3H), 2.83-1.41 (m, 15H), 4.55 (d, *J* = 5.6 Hz, 2H), 5.19 (d, *J* = 10.5 Hz, 1H), 5.27 (d, *J* = 17.3 Hz, 1H), 5.68 (s, 1H), 5.93-6.03 (m, 1H), 6.23 (d, *J* = 8.4 Hz, 1H), 6.57 (s, 1H), 7.13 (d, *J* = 8.4 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz, ppm) δ 18.1, 23.8, 26.9, 27.1, 28.1, 30.4, 35.0, 39.0, 44.8, 45.3, 53.8, 113.7, 116.2, 118.0, 127.3, 133.0, 135.3, 138.9, 154.5, 172.3; MS (ESI, positive) m/z calcd for C₂₁H₂₈NO₂ [M+H]: 326.21, Found 326.30.

3-Hydroxy-estra-1,3,5(10)-triene-[17,16-*b***]-pyridine (8)**¹⁴⁾



To a 30 mL flask were added to *O*-allyl oxime **7** (2.00 g, 6.14 mmol, 1.0 eq.) under argon at 230°C and stirred for 24 h. Then the mixtures were cooling to r.t. and dissolved in EtOH. The unsolved solid was removed by filtration and the filtrate solution was evaporated to remove EtOH after the solution was added to silica gel. The crude mixture was purified by silica gel column chromatography (chloroform/ethyl acetate = 90/10) and orange solid product was obtained (282 mg, 15% yield).

¹H NMR (d_6 -DMSO, 400 MHz, ppm) δ 0.89 (s, 3H), 1.39-2.80 (m, 13H), 6.46 (s, 1H), 6.53 (dd, J = 2.2 Hz, 1H), 7.05-7.09 (m, 2H), 7.59 (d, J = 7.4 Hz, 1H), 8.24 (d, J = 4.6 Hz, 1H), 9.02 (s, 1H); ¹³C NMR (d_6 -DMSO,

100 MHz, ppm) δ 17.6, 26.0, 27.0, 29.0, 29.5, 33.6, 37.3, 43.9, 45.5, 54.7, 112.8, 115.0, 121.1, 125.8, 130.3, 132.5, 135.8, 137.1, 146.6, 155.0, 172.6; MS (ESI, positive) m/z calcd for C₂₁H₂₄NO [M+H]: 306.19, Found 306.20.

Preparation of ethyl *O*-(4-fluorobenzoyl)acetohydroxamate and *O*-(4-fluorobenzoyl) -2-naphthaldoxime Ethyl *O*-(4-fluorobenzoyl)acetohydroxamate



4-Fluorobenzoylchloride (155 μ L, 1.31 mmol, 1.35 equiv.) was added to a solution of ethyl acetohydroxamate (100 mg, 0.970 mmol) and triethylamine (196 μ L, 1.41 mmol, 1.45 equiv.) in THF (3.2 ml, 0.30 M) with stirring under ice cooling. After addition was complete, the reaction mixtures was stirred for 15 h at room temperature. Then saturated NH₄Cl aq. (3 mL) was added to quench the reaction and remove THF by evaporator, and the remained aqueous layer was extracted with Et₂O (10 mL×3). The combined organic layer was washed with brine (10 mL) and dried over MgSO₄. After removal of MgSO₄ from organic layer, solvent was evaporated under reduced pressure. The remained crude mixtures was purified by silica gel column chromatography (hexane/Ethyl acetate = 90/10) and yellow solid product was obtained (191 mg, 88%).

¹H NMR (400 MHz, CDCl₃) δ 1.36 (t, *J* = 7.1 Hz, 3H), 2.15 (s, 3H), 4.28 (q, *J* = 7.1 Hz, 2H), 7.11-7.16 (m, 2H), 8.05-8.08 (m, 2H); ¹⁹F NMR (376 MHz, CDCl₃) δ -108.4 (s, 1F); ¹³C NMR (100 MHz, CDCl₃) δ 14.2, 15.1, 63.8, 115.6 (d, *J*_{*C-F*} = 22 Hz), 125.4 (d, *J*_{*C-F*} = 3 Hz), 131.9 (d, *J*_{*C-F*} = 9 Hz), 162.9, 165.8 (d, *J*_{*C-F*} = 253 Hz), 169.6; IR (neat) 3068, 2989, 1747, 1635, 1508, 1379, 1319, 1271, 1159, 1095, 1045, 972, 860, 760, 606 cm⁻¹; MS (ESI, positive) m/z calcd for C₁₁H₁₃FNO₃ [M+H]: 226.09, Found 226.50; Anal. Calcd. for C₁₁H₁₂FNO₃: C, 58.7; H, 5.37; N, 6.22; found C, 58.6; H, 5.45; N, 6.13.

O-(4-fluorobenzoyl) -2-naphthaldoxime



To a flame dried test tube were added to 2-naphthaldehyde (13.3 mg, 0.085 mmol, 1.0 eq.), CH_2Cl_2 (0.9 mL, 0.1 M), 70 wt% HClO₄ aq (0.7 µL, 8.50 µmmol, 10 mol%) under argon at room temperature. Then ethyl *O*-(4-fluorobenzoyl)acetohydroxamate (28.7 mg, 0.128 mmol, 1.5 equiv.) was added to stirred reaction mixtures. After 3 hours, reaction mixtures were filtrated through silica/Celite® pad to remove acid-catalyst. The filtrate solution were evaporated and removed under vacuum. Then, crude mixture was purified by

column chromatography (hexane/benzene = 4/6) giving the corresponding oximes as a white solid (24.4 mg, 98%).

¹H NMR (400 MHz, CDCl₃) δ 7.16 (t, *J* = 8.4 Hz, 2H), 7.51-7.58 (m, 2H), 7.85-7.90 (m, 3H), 8.03-8.09 (m, 2H), 8.17 (dd, J = 8.4, 5.5 Hz, 2H), 8.67 (s, 1H); ¹⁹F NMR (376 MHz, CDCl₃) δ -107.7 (s, 1F); ¹³C NMR (100 MHz, CDCl₃) δ 123.6, 124.9 (d, *J*_{C-F} = 2.9 Hz), 127.0, 127.8, 128.0, 128.1, 128.9 (d, *J*_{C-F} = 27 Hz), 130.9, 132.4 (d, *J*_{C-F} = 9.3 Hz), 133.0, 135.1, 163.1, 166.1 (d, *J*_{C-F} = 253 Hz); IR (neat) 3060, 3022, 1747, 1601, 1504, 1255, 1155, 1078, 974, 893, 850, 758 cm⁻¹; MS (ESI, positive) m/z calcd for C₁₈H₁₃FNO₂ [M+H]: 294.09, Found 294.60; Anal. Calcd. for C₁₈H₁₂FNO₂: C, 73.7; H, 4.12; N, 4.78; found C, 73.5; H, 4.28; N, 4.86.

Procedure for recovery and reuse of reaction filtrate solution



1) To a flame dried 50 mL flask were added to 3-Phenylpropionaldehyde **3i** (0.98 mL (1.0 g), 7.45 mmol, 1.0 eq.), H₂O (37 mL, 0.2 M), 70 wt% HClO₄ aq (32.1 μ L, 0.373 mmol, 5 mol%) under argon at room temperature. Then methylethyl ketone **2d** (1.05 mL, 11.2 mmol, 1.5 equiv.) was dropwised to stirred reaction mixtures. After reaction mixtures turned to white suspension for 12 h, the white solid was collected by filtration and filtrate solution was received to a 100 mL round bottom flask. Furthermore, the above solid was washed using water (10 mL×3) after receiver flask was changed. The collected product was dried over P₂O₅ under vacuum for 12 h to remove water. Aldoxime **3i** was obtained in 79% yield (0.88 g).

2) The previous receiver 100 mL flask which contained filtration reaction solution, was charged argon gas. To the solution was added stirrer bar, 3-Phenylpropionaldehyde **3i** (0.98 mL (1.0 g), 7.45 mmol, 1.0 eq.), methylethyl ketone **2d** (0.70 mL, 7.45 mmol, 1.0 equiv.) and stirred at room temperature for 12 hours. After starting materials was consumed (determined by TLC), the reaction mixtures was filtrated and the filtrate solution was received to another 100 mL round bottom flask. The collected solid was further washed with H_2O (10 mL×3) and dried over P_2O_5 under vacuum at room temperature for 12 h. The desired product was obtained in 86% yield (0.96 g).

3-11) Above procedure 2) was repeated total nine times.



Figure S2. Appearance of catalytic transoximation to 1i from 2d in H₂O. (a) 10 min. (b) 3 hours. (c) 12 hours.

Cycle	Isolated yield (%)				
	Oxime (3i)				
1 st	85				
2 nd	98				
3 th	84				
4 th	80				
5 th	99				
6 th	86				
7 th	71				
8 th	73				
9 th	80				
10 th	83				
11 th	80				

Table S4. The result for synthesis of 3i using reusable reaction filtrate

4. Observation Experiments about Existence of Hydroxylamine Derivatives in-situ Pre-experiments to check hydroxylamine derivatives by ¹H & ¹⁹F NMR¹⁵



To a flame dried test tube ethyl *O*-(4-fluorobenzoyl)acetohydroxamate (15.8 mg, 0.0700 mmol), 1, 1, 2, 2-tetrachloroethane (5.9 mg, 0.0350 mmol) as internal standard for ¹H NMR, perfluorobenzene (2.2 mg, 0.0119 mmol) as internal standard for ¹⁹F NMR, d_3 -CD₃CN (0.7 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then the solution was directly transferred to the NMR tube and taken ¹H & ¹⁹F NMR (before starting reaction: 0 min, Figure S3 (a), S4 (a)). After the measurement of NMR, the solution was returned to the test tube and 70 wt% HClO₄ aq. (23.8 µL, 0.0350 mmol, 0.50 equiv.) was added there under stirring. Further 70 wt% HClO₄ aq. (23.8 µL, 0.0350 mmol, 0.50 equiv.) was added to the reaction mixtures after 1.5 h. Starting material was consumed after 3 h, followed by taken ¹H & ¹⁹F NMR (after finished reaction: 180 min, Figure S3 (b), S4 (b)). We can observe AcOEt and new peaks (8.11, 7.35 ppm) from ¹H NMR, one peak (-102 ppm) from ¹⁹F NMR.



Figure S3. ¹H NMR



Figure S4. ¹⁹F NMR

Experiments for the observation of hydroxylamine derivatives in-situ under transoximation reaction



То flame dried test tube 2-naphthaldehyde **1a** (10.9 mg, 0.0700 1) а mmol), ethyl O-(4-fluorobenzoyl)acetohydroxamate (23.7 mg, 0.105 mmol, 1.5 equiv.), 1, 1, 2, 2-tetrachloroethane (5.9 mg, 0.0350 mmol) as internal standard for ¹H NMR, perfluorobenzene (2.2 mg, 0.0119 mmol) as internal standard for ¹⁹F NMR, d_3 -CD₃CN (0.7 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then the solution was directly transferred to the NMR tube and taken ¹H & ¹⁹F NMR (before starting reaction: 0 min, Figure S5 (a), S6 (a)). After the measurement of NMR, the solution was returned to the test tube and 70 wt% HClO4 aq. (1.2 µL, 0.0140 mmol, 20 mol%) was added there under stirring. After 15, 60 minutes, the reaction mixtures was taken ¹H & ¹⁹F NMR as well as previous methods (Figure S6 (b) (c), S6 (b) (c)).

2) After 60 minutes **1a** was disappeared from ¹H NMR, therefore **1a** (4.1 mg, 0.0263 mmol) was added to the reaction mixtures again.

3) Further 60 minutes, the reaction mixturtes was taken ¹H & ¹⁹F NMR, which showed no peaks of ethyl O-(4-fluorobenzoyl)acetohydroxamate, however two peaks which seemed hydroxylamine derivatives and aldoxime, was confirmed on the ¹⁹F NMR (Figure S5 (d), S6 (d)). Moreover **1a** (2.2 mg, 0.0140 mmol) was added to the reaction mixtures and stirred further 1 hour. As a result, one peak (-102 ppm) was disappeared and another peak (-107 ppm) derived from product was remained on the ¹⁹F NMR chart (Figure S6 (e)).



Figure S5. ¹H NMR



S21



5. NMR Experiments for Hydrolysis of 2g

In *d*₂-CD₂Cl₂ solution (Figure 3 and Table 3 entry 1)



1) To a flame dried test tube ethyl acetohydroxamate **2g** (14.5 μ L, 0.140 mmol), 1, 1, 2, 2-tetrachloroethane (11.7 mg, 0.0697 mmol) as internal standard, d_2 -CD₂Cl₂ (1.4 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (2.4 μ L, 0.0280 mmol, 20 mol%) was added to the reaction mixtures. After 1, 2, 3, 4 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine the yield of AcOEt **4g** from internal standard. After the measurement of ¹H NMR, the solution was returned to the test tube and started the reaction under stirring again.

2) After 4 hours, 70 wt% HClO₄ aq. (2.4 μ L, 0.0280 mmol, 20 mol%) was added to the reaction mixtures (total 40 mol% HClO₄). The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (5, 6, 19 h).

3) After 19 hours, 70 wt% HClO₄ aq. (2.4 μ L, 0.0280 mmol, 20 mol%) was added to the reaction mixtures (total 60 mol% HClO₄) again. The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (20, 21, 22 h).

4) After 22 hours, H₂O (0.5 mg, 0.0278 mmol, 20 mol%) was added to the reaction mixtures (reaction was almost not proceeding). The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (23, 24, 25, 37 h).

5) After 37 hours, 2-naphthaldehyde (4.4 mg, 0.0280 mmol, 20 mol%) **1a** was added to the reaction mixtures (the reaction was restarted). The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (38, 39, 40, 41, 42, 60.5 h).

6) After 60.5 hours (**1a** was consumed by ¹H NMR), 2-naphthaldehyde (4.4 mg, 0.0280 mmol, 20 mol%) **1a** was added to the reaction mixtures again (total 40 mol% **1a**). The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (61.5, 62.5, 66.5 h) (All **2g** disappeared on the ¹H NMR).

In d₂-CD₂Cl₂ solution in the presence of molecular sieve 4A (MS4A) (Figure S7, Table 3 entry 2)



To a flame dried test tube ethyl acetohydroxamate 2g (14.5 µL, 0.140 mmol), 1, 1, 2, 2-tetrachloroethane (13.2 mg, 0.0786 mmol) as internal standard, MS4A (43.3 mg, 300 wt/wt% for 2g, it was dried under 2

mmHg/150°C with P₂O₅ for 1 hour before use), d_2 -CD₂Cl₂ (1.4 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (2.4 µL, 0.0280 mmol, 20 mol%) was added to the reaction mixtures. After 1, 2, 3, 4, 5, 6, 7, 20 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine the yield of AcOEt **4g** from internal standard. After the measurement of ¹H NMR, the solution was returned to the test tube and started the reaction under stirring again.



Figure S7. Hydrolysis of 2g in d_2 -CD₂Cl₂ in the presence of MS4A

In D₂O solution (Figure S8, Table 3 entry 3)



To a flame dried test tube ethyl acetohydroxamate 2g (14.5 µL, 0.140 mmol), 1, 1, 2, 2-tetrachloroethane (11.7 mg, 0.0697 mmol) as internal standard, D₂O (1.4 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (2.4 µL, 0.0280 mmol, 20 mol%) was added to the reaction mixtures. After 1, 2, 3, 17.5, 21 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine the yield of AcOEt **4g** from internal standard. After the measurement of ¹H NMR, the solution was returned to the test tube and started the reaction under stirring again.



Figure S8. Hydrolysis of 2g in D₂O

In H₂¹⁸O solution (Figure S9, Table 3 entry 4)



1) To a flame dried test tube, 70 wt% HClO₄ aq. (2.1 μ L, 0.0250 mmol, 100 mol%) H₂¹⁸O (0.25 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then ethyl acetohydroxamate **2g** (2.6 μ L, 0.025 mmol) was added to the reaction mixtures.

2) After 10 minutes, the reaction mixtures was extracted with d_8 -toluene (1.0 mL). The obtained organic layer was directly measured by ¹H NMR to check the production of AcOEt and the consumption of all **2g**. Next in order to determine whether ¹⁸O-labeled AcOEt **4g** or ¹⁶O-AcOEt **4g**, the solution was directly measured by the GC-MS (Figure S8). We can confirmed that the obtained **4g** had one ¹⁸O labeled oxgen atom. According to the theoretical calculations, the ratio of C₄H₈¹⁶O₂ (MW: 88.05): C₄H₈¹⁶O¹⁸O (MW: 90.06) = 100: 0.20. Following to literature¹⁶), we can calculated the ¹⁸O% of **4g**: (100-5.01*0.20%)/(100-5.01*0.20%+5.01)= 95% (We can conclude the oxygen of carbonyl group of AcOEt mostly came from oxygen of water, not from HClO₄).

3) After GC-MS measurement, the solution was added to 1, 1, 2, 2-tetrachloroethane (2.0 mg, 0.0119 mmol) as internal standard and taken ¹H NMR to determine the yield of AcOEt **4g** from internal standard (84% yield).



マステープル			
ピーク#:1	保持時	間:1.720(スキ	+ン#:245)
ピーク数:8			
グループ	1-イベ	ント1 スキャン	
#	m/z	絶対強度	相対強度
1	86.00	96	0.13
2	87.00	165	0.23
3	88.05	3572	5.01
4	89.10	5369	7.53
5	90.05	71303	100.00
6	91.05	6968	9.77
7	92.10	386	0.54
8	93.10	3	0.00

(c)

Figure S9. GC-MS chart of the obtained ¹⁸O-labeled AcOEt

(a) total MS chart, (b) expansion MS chart between M.W. 80 and 100, (c) peak intensity

In CD₂Cl₂ with HClO₄·NH₂OH as a catalyst (Figure S10, Table 3 entry 5)



1) To a flame dried test tube ethyl acetohydroxamate **2g** (14.5 μ L, 0.140 mmol), 1, 1, 2, 2-tetrachloroethane (14.0 mg, 0.0834 mmol) as internal standard, d_2 -CD₂Cl₂ (1.4 mL, 0.1 M) was added and stirred under argon atmosphere at room temperature. Then 0.7% anhydrous NH₂OH·HClO₄-EtOH solution (534 mg, 0.0280 mmol, 20 mol%) was added to the reaction mixtures. After 1, 2, 3, 14.5 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine the yield of AcOEt **4g** from internal standard. After the measurement of ¹H NMR, the solution was returned to the test tube and started the reaction under stirring again (The hydrolysis reaction was not almost proceeding).

2) After 14.5 hours, 2-naphthaldehyde **1a** (21.9 mg, 0.140 mmol, 1.0 equiv.) was added to the reaction mixtures. The reaction was monitored by ¹H NMR and calculated the yield of AcOEt **4g** every hours (15.5, 18.5, 19.5 h) (Once aldehyde was added, the reaction was drastically proceeding. This result indicated free HClO₄ was re-generated by aldehyde and started the hydrolysis reaction).



Figure S10. Hydrolysis of 2g in CD₂Cl₂ with HClO₄·NH₂OH as a catalyst

6. Kinetic Studies¹⁷⁾

Experiments for ethyl acetohydroxamate

To every flame dried 30 mL flask aldehyde **1a** (0.80 mmol), tetralin (0.20 mmol, as an internal standard), CH₂Cl₂ (4.0 mL, 0.20 M) was added and stirred under Ar atmosphere. 70 wt% HClO₄ aq. (0.040 mmol, 5 mol%) was added to the reaction mixtures, followed by added ethyl acetohydroxamate **2g** (x [mmol], x = 0.60, 0.80, 1.0, 1.2) at room temperatures. After 20, 40, 60 min, an aliquot part of the reaction mixtures was taken (ca 0.50 mL) and quenched with Et₃N (0.025 mmol). The solvent was evaporated under reduced pressure (ca 200 mmHg) to give the crude mixtures. The yield of aldoxime **3a** was directly determined by ¹H NMR from internal standard.



Figure S11. Kinetic study for ethyl acetohydroxamate

Experiments for 2-naphthaldehyde

To every flame dried 30 mL flask aldehyde **1a** (y [mmol], y = 0.40, 0.60, 0.80, 1.2), tetralin (0.20 mmol, as an internal standard), CH₂Cl₂ (4.0 mL, 0.20 M) was added and stirred under Ar atmosphere. 70 wt% HClO₄ aq. (0.040 mmol) was added to the reaction mixtures, followed by added ethyl acetohydroxamate **2g** (1.2 mmol) at room temperatures. After 20, 40, 60 min, an aliquot part of the reaction mixtures was taken (ca 0.50 mL) and quenched with Et₃N (0.025 mmol). The solvent was evaporated under reduced pressure (ca 200 mmHg) to give the crude mixtures. The yield of aldoxime **3a** was directly determined by ¹H NMR from internal standard.



Figure S12. Kinetic study for 2-naphthaldehyde

Experiments for perchloric acid

To every flame dried 30 mL flask aldehyde **1a** (0.80 mmol), tetralin (0.20 mmol, as an internal standard), CH₂Cl₂ (4.0 mL, 0.20 M) was added and stirred under Ar atmosphere. 70 wt% HClO₄ aq. (z [mmol], z = 0.040, 0.048, 0.060, 0.080) was added to the reaction mixtures, followed by added ethyl acetohydroxamate **2g** (1.2 mmol) at room temperatures. After 20, 40, 60 min, an aliquot part of the reaction mixtures was taken (ca 0.50 mL) and quenched with Et₃N (0.025 mmol). The solvent was evaporated under reduced pressure (ca 200 mmHg) to give the crude mixtures. The yield of aldoxime **3a** was directly determined by ¹H NMR from internal standard.





Figure S13. Kinetic study for perchloric acid

Experiments for water

To every flame dried 30 mL flask aldehyde **1a** (0.80 mmol), tetralin (0.20 mmol, as an internal standard), CH₂Cl₂ (4.0 mL, 0.20 M), H₂O (a [mmol], a = 0.080, 0.16, 0.24, 0.40) was added and stirred under Ar atmosphere. 70 wt% HClO₄ aq. (0.040 mmol, 5 mol%) was added to the reaction mixtures, followed by added ethyl acetohydroxamate **2g** (1.2 mmol) at room temperatures. After 20, 40, 60 min, an aliquot part of the reaction mixtures was taken (ca 0.50 mL) and quenched with Et₃N (0.025 mmol). The solvent was evaporated under reduced pressure (ca 200 mmHg) to give the crude mixtures. The yield of aldoxime **3a** was directly determined by ¹H NMR from internal standard.



Figure S14. Kinetic study for water

7. Cross-over Reaction

Reaction with 1c and 1d, 2g



Scheme S1. Cross-over reaction with 1c and 1d in the presence of 2g

To a flame dried test tube **1c** (20.7 μ L, 0.170 mmol), **1d** (23.2 μ L, 0.170 mmol), CH₂Cl₂ (0.9 mL, 0.20 M) was added and stirred under argon atmosphere. 70 wt% HClO₄ aqueous solution (0.7 μ L, 8.50 μ mol) was added to the reaction mixtures, followed by added **2g** (17.5 μ L, 0.170 mmol) at room temperature. After 24 hours, the reaction mixtures was passed through silica/Celite® pad to remove catalyst and washed with CH₂Cl₂ (10 mL). The collected solution was evaporated to remove solvent and dried over under vacuum pump to give crude mixtures. 1, 1, 2, 2-Tetrachloroethane (14.8 mg, 0.0883 mmol) was added to the crude mixtures as an internal standard. The yield of both **3c** and **3d** were determined by ¹H NMR from the peak of internal standard.

Reaction with 1a and 3c



Scheme S2. Cross-over reaction with 1a and 3c

To a flame dried test tube **1a** (26.6 mg, 0.170 mmol), **3c** (25.7 mg, 0.170 mmol), CH_2Cl_2 (0.9 mL, 0.20 M) was added and stirred under argon atmosphere. 70 wt% HClO₄ aqueous solution (0.7 µL, 8.50 µmol) was added to the reaction mixtures at room temperature. After 24 hours, the reaction mixtures was passed through silica/Celite® pad to remove catalyst and washed with CH_2Cl_2 (10 mL). The collected solution was evaporated to remove solvent and dried over under vacuum pump to give crude mixtures. 1, 1, 2, 2-Tetrachloroethane (15.5 mg, 0.0925 mmol) was added to the crude mixtures as an internal standard. The yield of both **2a** and **1c** were determined by ¹H NMR from the peak of internal standard.

Reaction with 1a and 3d



Scheme S3. Cross-over reaction with 1a and 3d

To a flame dried test tube **1a** (26.6 mg, 0.170 mmol), **3d** (32.2 mg, 0.170 mmol), CH_2Cl_2 (0.9 mL, 0.20 M) was added and stirred under argon atmosphere. 70 wt% HClO₄ aqueous solution (0.7 µL, 8.50 µmol) was added to the reaction mixtures at room temperature. After 24 hours, the reaction mixtures was passed through silica/Celite® pad to remove catalyst and washed with CH_2Cl_2 (10 mL). The collected solution was evaporated to remove solvent and dried over under vacuum pump to give crude mixtures. 1, 1, 2, 2-Tetrachloroethane (14.3 mg, 0.0852 mmol) was added to the crude mixtures as an internal standard. The yield of both **2a** and **1d** were determined by ¹H NMR from the peak of internal standard.

8. Retro Reaction

Transoximation from aldoxime to ethyl acetate in d₂-CD₂Cl₂



To a flame dried test tube ethyl acetate **4g** (9.1 μ L, 0.0930 mmol, 1.0 equiv.), (*E*)-2- naphthaldoxime **3a** (24.0 mg, 0.140 mmol, 1.5 equiv.), d_2 -CD₂Cl₂ (1.4 mL) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (0.4 μ L, 4.70 μ mol, 5 mol%) was added to the reaction mixtures. After 24 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine **2g**. However the reaction was not occurred and peaks of **2g** was nothing on the ¹H NMR chart.

Transoximation from aldoxime to acetone in *d*₂-CD₂Cl₂



To a flame dried test tube acetone **4a** (10.3 μ L, 0.140 mmol, 1.0 equiv.), (*E*)-2- naphthaldoxime **3a** (36.0 mg, 0.210 mmol, 1.5 equiv.), 1, 1, 2, 2-tetrachloroethane (10.9 mg, 0.0649 mmol) as internal standard d_2 -CD₂Cl₂ (1.4 mL, 0.10 M) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (1.2 μ L, 14.0 μ mol, 10 mol%) was added to the reaction mixtures. After 24 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine yield of **2a** from an internal standard (**2a**: 1% yield). We found retro reaction (**3a** to **4a**) was not almost occurred in this condition.

Transoximation from aldoxime to acetone in D₂O



To a flame dried test tube acetone **4a** (24.9 μ L, 0.340 mmol, 1.0 equiv.), (*E*)-2- benzaldoxime **3b** (61.8 mg, 0.510 mmol, 1.5 equiv.), dioxane (3.7 mg, 0.0425 mmol) as internal standard D₂O (1.7 mL, 0.20 M) was added and stirred under argon atmosphere at room temperature. Then 70 wt% HClO₄ aq. (2.9 μ L, 34.0 μ mol, 10 mol%) was added to the reaction mixtures. After 24 hours, the solution was directly transferred to the NMR tube and taken ¹H NMR to determine yield of **2a** from an internal standard (**2a**: 29% yield). We found retro reaction (**3b** to **4a**) was slightly occurred in this condition.

These results indicated that retro-reaction which **3a** reacted with **4a**, **g** generated from **2a**, **g** in d_2 -CD₂Cl₂ was not occurred under process of transoximation in our catalytic system. Thus retro-reaction which **3b** reacted with **4a** generated from **2a** in D₂O did not occur so much under process of transoximation in our catalytic system. This will be one of the reason our transoximation system gave aldoxime products in high yields.



9. References

- 1) Kainuma, M.; Makishima, M.; Hashimoto, Y.; Miyachi, H. Bioorg. Med. Chem. 2007, 15, 2587-2600.
- 2) Yu, J.; Jin, Y.; Lu, M. Adv. Synth. Catal. 2015, 357, 1175-1180.
- 3) Zhang, L.; Chen, H.; Zha, Z.; Wang, Z. Chem. Commun. 2012, 48, 6574–6576.
- 4) Sharghi, H.; Sarvari, M. H. Synlett, 2001, 99-101.
- 5) Li, J.-T.; Li, X.-L.; Li, T.-S. Ultrasonics Sonochemistry, 2006, 13, 200-202.
- 6) Vo, Q. V.; Trenerry, C.; Rochfort, S.; Wadeson, J.; Leyton, C.; Hughes, A. B. Bioorg. Med. Chem. 2013, 21, 5945-5954.
- 7) Crandall, J. K.; Reix, T. J. Org. Chem. 1992, 57, 6759-6764.
- 8) Furuya, Y.; Ishihara, K.; Yamamoto, H. J. Am. Chem. Soc. 2005, 127, 11240-11241.
- 9) Blay, G.; Benach, E.; Fernández, I, Galletero, S.; Pedro, J. R.; Ruiz, R. Synthesis, 2000, 403-406.
- 10) Pavlishchuk, V.; Birkelbach, F.; Weyhermüller, T.; Wieghardt, K.; Chaudhuri, P. *Inorg. Chem.* **2002**, *41*, 4405-4416.
- 11) Liu, S.; Yu, Y.; Liebeskind, L. S. Org. Lett. 2007, 9, 1947-1950.
- 12) (a) El-Gamal, M.; Bayomi, S. M.; El-Ashry, S. M.; Said, S. A.; Abdel-Aziz, A. A.-M.; Abdel-Aziz, N. I. *Eur. J. Med. Chem.* 2010, 45, 1403-1414; (b) Jerzy, S.; Czeslawa, T. Zesz. Nauk. Politech. Slask. Chem. 1967, 39, 43.
- 13) Khomutov, A. R.; Khurs, E. N.; Khomutov, R. V. Russ. J. Bioorg. Chem. 1988, 14, 385-391.
- 14) Fischer, D. S.; Allan, G. M.; Bubert, C.; Vicker, N.; Smith, A.; Tutill, H. J.; Purohit, A.; Wood, L.; Packham, G.; Mahon, M. F.; Reed, M. J.; Potter, B. V. L. J. Med. Chem. 2005, 48, 5749–5770.
- 15) Parlanti, L.; Discordia, R. P.; Hynes, Jr., J.; Miller, M. M.; O'Grady, H. R.; Shi, Z. Org. Lett. 2007, 9, 3821-3824.
- 16) Kang, Q.-K.; Wang, L.; Liu, Q.-J.; Li, J.-F.; Tang, Y. J. Am. Chem. Soc. 2015, 137, 14594-14597.
- 17) Yukawa, T.; Seelig, B.; Xu, Y.; Morimoto, H.; Matsunaga, S.; Berkessel, A.; Shibasaki, M. J. Am. Chem. Soc. 2010, 132, 11988-11992.

10. NMR chart



































































































S85















