## Highly selective aerobic oxidation of cyclohexane to cyclohexanone and cyclohexanol over $V_2O_5$ (a) TiO\_2 under simulated solar light irradiation

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Figure S1. XPS spectra of Ti2p in the TiO<sub>2</sub> and 0.01-V<sub>2</sub>O<sub>5</sub>@TiO<sub>2</sub>.



**Figure S2.** XPS spectra of O1s in the  $TiO_2$  and  $0.01-V_2O_5$  (a) $TiO_2$ .



Figure S3. HRTEM images of 0.03-V<sub>2</sub>O<sub>5</sub>@TiO<sub>2</sub> (a), 0.05-V<sub>2</sub>O<sub>5</sub>@TiO<sub>2</sub> (b), 0.07-V<sub>2</sub>O<sub>5</sub>@TiO<sub>2</sub> (c).



Figure S4. EDX images of 0.03-V<sub>2</sub>O<sub>5</sub>@TiO<sub>2</sub>.

No obvious aggregation of V species in the catalysts of  $0.03-V_2O_5$  (*i*) TiO<sub>2</sub> is observed in Figure S3a. However, with the increasing of V content, it is found from Figure S 3b and Figure S 3c that aggregation of V species exists in catalyst, which would be detrimental to photocatalytic oxidation of cyclohexane. Meanwhile, to further understand the catalysts of  $0.03-V_2O_5$  (*i*) TiO<sub>2</sub>, the element mapping was performed in Figure S4, from which we could see there are slightly aggregation in the catalysts.