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Supporting information for

Highly selective photocatalytic oxidation of biomass-derived chemicals to carboxyl compounds over Au/TiO_2

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1. Scheme S1

Scheme S1. Light-induced selective oxidation of ethanol to acetic acid in water at ambient temperature by air.

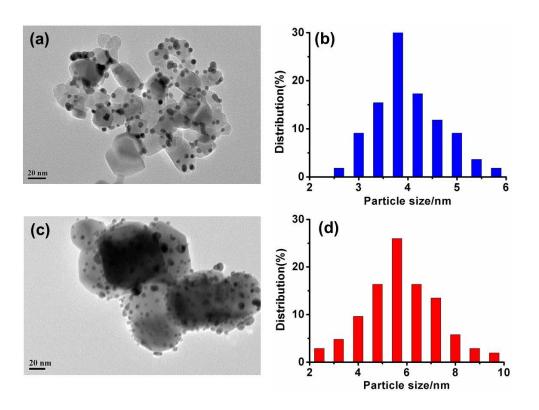


Fig. S1. TEM images and the size distribution of $AuNPs/TiO_2$ (a and b) and $AuNPs/ZrO_2$ (c and d) with 3 wt% AuNPs loading.

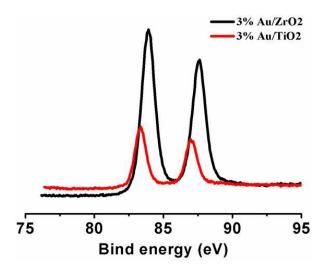


Fig. S2. XPS analysis of gold nanoparticles supported on TiO_2 and ZrO_2 .

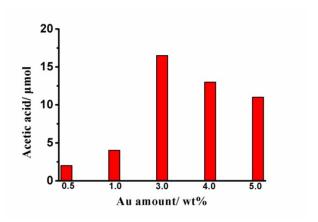


Fig. S3. The effect of the content of Au on the photocatalytic activity. Reaction conditions: ethanol, 100 μ mol; water, 1 mL; AuNPs/TiO₂, 0.025 g; visible light (λ = 420-780 nm, 0.3 W/cm²); Na₂CO₃, 100 μ mol; temperature, 30 °C.

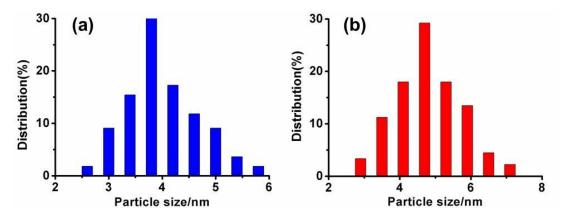


Fig. S4. The size distribution of AuNPs/TiO₂ with different AuNPs loading. 3 wt% (a) and 5 wt% (b).

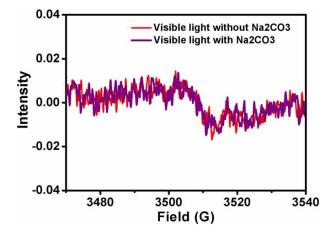


Fig. S5. Typical ESR spectra of DMPO trapped radical species upon photocatalytic oxidation of H₂O over AuNPs/TiO₂ with and without Na₂CO₃ under visible light irradiation.

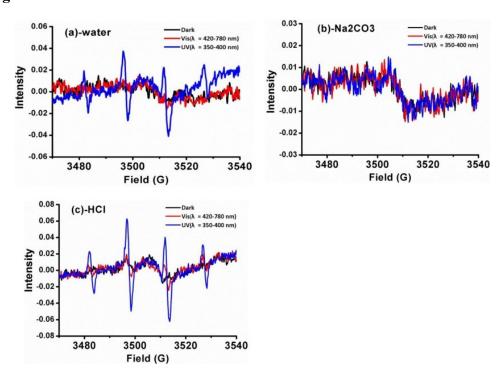


Fig. S6. Dependence of hydroxyl radical formation on the additives under UV and visible light irradiation.

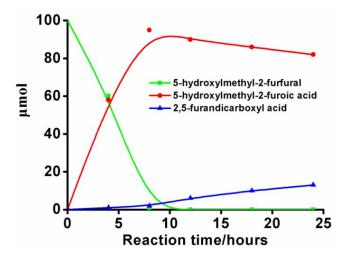
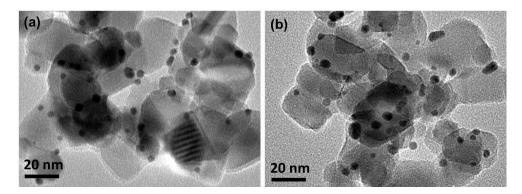


Fig. S7. Time course of the photoxidation of 5-hydroxymethyl-2-furfural over AuNPs/TiO₂ under visible light. Reaction conditions: substrate, 100 μmol; water, 1 mL; AuNPs/TiO₂ with 3 wt% Au, 0.025 g; visible light (λ = 420-780 nm, 0.3 W/cm²); Na₂CO₃, 200 μmol; temperature, 30 °C.



 $\textbf{Fig. S8}. \ TEM \ images \ of \ fresh \ AuNPs/TiO_2 \ and \ AuNPs/TiO_2 \ after \ four \ cycle \ reuse.$