Highly chemoselective reduction of nitroarenes over non-noble metal

nickel-molybdenum oxide catalysts

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1. Tables and Figures

1.1. Tables S1-S2

Table S1. Textural properties of the prepared catalysts

Sample	$\mathbf{S}_{\mathrm{BET}}$	\mathbf{V}_{p}	D_p	Ni crystallite
	$(m^2 g^{-1})$	$(cm^3 g^{-1})$	(nm)	size (nm)
SBA-15	678	0.95	5.5	-
CN@SBA-15	358	0.55	5.2	-
Ni/CN@SBA-15	339	0.46	4.8	22.4
6.0%MoO ₃ /CN@SBA-15	338	0.44	4.9	_
Ni-1.5%MoO ₃ /CN@SBA-15	326	0.42	4.9	18.0
Ni-3.0%MoO ₃ /CN@SBA-15	316	0.39	4.8	14.8
Ni-4.5%MoO ₃ /CN@SBA-15	297	0.37	4.9	10.1
Ni-6.0%MoO ₃ /CN@SBA-15	285	0.29	4.8	8.0
Ni-7.5%MoO ₃ /CN@SBA-15	253	0.23	4.9	10.4

Table S2. Actual composition of Mo, Ni, C, and N elements in the Ni-xMoO₃/CN@SBA-15 and MoO₃/CN@SBA-15 catalysts

Sample	Ni ^a (wt%)	Mo ^a	Total content of	C/N molar
		(wt%)	$(C + N)^{b} (wt%)$	ratio ^c
Ni/CN@SBA-15	3.2	_	26.4	7.8
MoO ₃ /CN@SBA-15	_	3.5	9.4	2.2
Ni-1.5%MoO3 /CN@SBA-15	3.1	0.6	22.3	6.9
Ni-3.0%MoO3 /CN@SBA-15	3.2	1.3	24.4	6.7
Ni-4.5%MoO3 /CN@SBA-15	3.2	2.2	32.3	6.5
Ni-6.0%MoO3 /CN@SBA-15	2.7	3.8	32.9	6.1
Ni-7.5%MoO ₃ /CN@SBA-15	2.7	4.7	33.5	6.6

^a Determined by ICP-AES. ^b Determined by TG. ^c Determined by XPS.



1.2. Figures S1-S6

Figure S1. (a) N_2 sorption isotherms, and (b) BJH pore size distributions of the NixMoO₃/CN@SBA-15 and MoO₃/CN@SBA-15 catalysts.



Figure S2. Low-angle XRD patterns of the Ni-*x*MoO₃/CN@SBA-15 and MoO₃/CN@SBA-15 catalysts.



Figure S3. Wide-angle XRD patterns of the Ni-*x*MoO₃/CN@SBA-15 and MoO₃/CN@SBA-15 catalysts.





Figure 4. TEM images of the Ni- $xMoO_3/CN@SBA-15$ and $MoO_3/CN@SBA-15$ catalysts. (a) x = 0%, (b) x = 1.5%, (c) x = 3.0%, (d) x = 4.5%, (e) x = 6.0%, (f) x = 7.5%, and (g) $MoO_3/CN@SBA-15$.



Figure 5. XPS spectra of the Ni-*x*MoO₃/CN@SBA-15 and MoO₃/CN@SBA-15 catalysts: (a) Ni 2p, (b) Mo 3d, (c) C 1s, (d) N 1s, and (e) the survey scan.



Figure S6. (a.b), XRD, (c,d), XPS and (e), TEM results of the spent Ni-6.0%MoO₃/CN@SBA-15 catalyst for the reduction of 2,4-dichloronitrobenzene after the tenth run.