

**Supporting Information**

# **Electrocatalytic upgrading of itaconic acid to methylsuccinic acid using fermentation broth as substrate solution**

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- 1. Calculations**
- 2. Cyclic voltammograms**
- 3. NMR spectra**

## 1. Calculations

Calculation of charge chronoamperometry ( $Q$  = charge,  $I$  = current,  $t$  = time):

$$Q = \int_x^y I dt$$

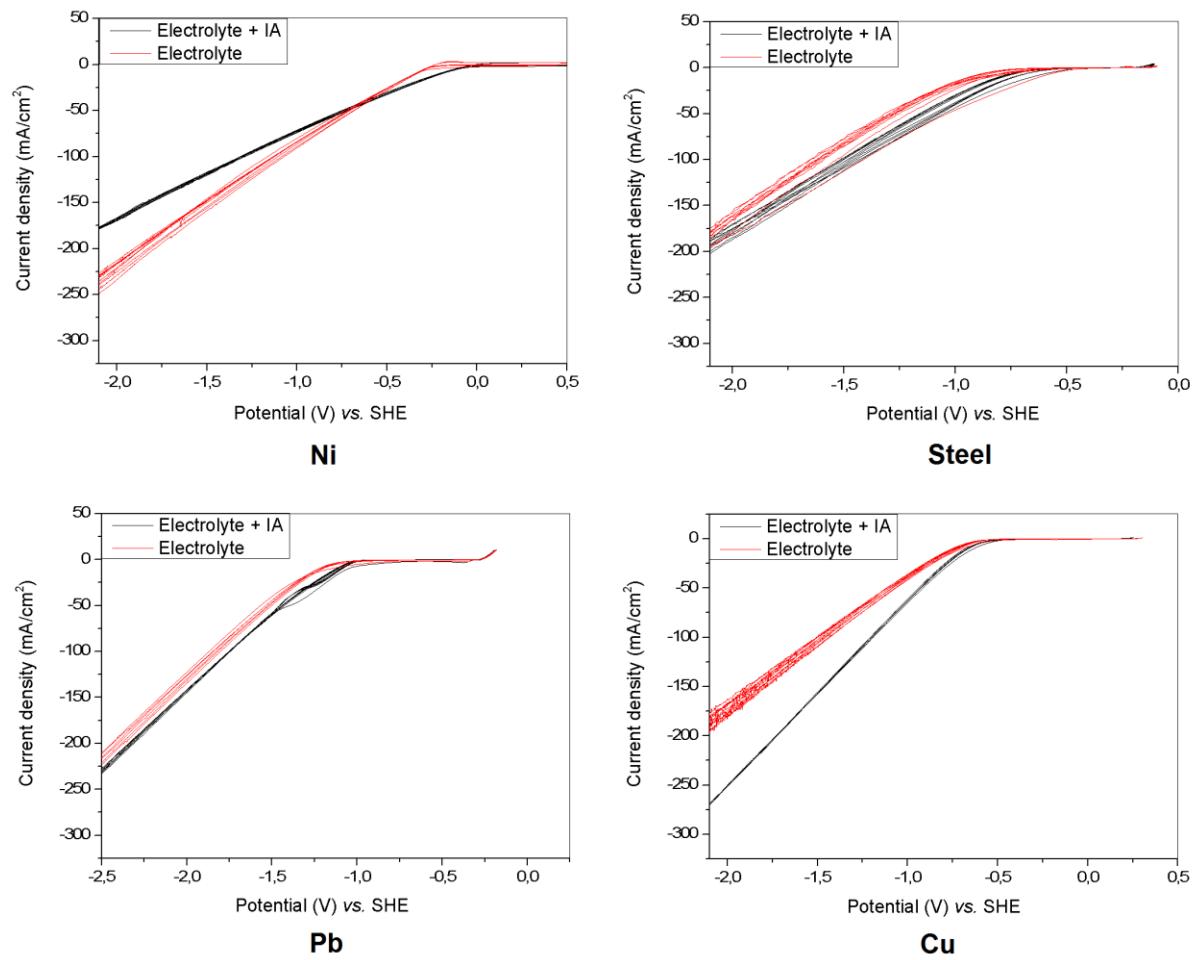
Calculation of charge potentiometry:

$$Q = It$$

Calculation of the faradiac efficiency ( $n_{mol}$  = amount of product in mol,  $n$  = number of electrons transferred,  $F$  = Faraday constant):

$$F_{eff} = \frac{n_{mol}nF}{Q} \cdot 100\%$$

## 2. Cyclic voltammograms



**Figure 1:** Cyclic voltammograms of aqueous sulfuric acid ( $0.5 \text{ mol L}^{-1}$ ) mixtures and aqueous sulfuric acid ( $0.5 \text{ mol L}^{-1}$ ) mixtures with  $0.1 \text{ mol L}^{-1}$  itaconic acid with different electrode materials.

### 3. NMR spectra

