

Electronic Supplementary Information

Environmental risk – based ranking of solvents by the combination of multimedia model and multi-criteria decision analysis.

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Statement: For the same data point different literature values can be found.* We tried to find the data originating from reliable sources, collecting preferably experimental data not modelled or estimated. Because of the literature data variability, it is possible to find (sometimes significantly) another values in various sources.

ESI Table 1. Parameters used for the assessment procedure.

Chemical group	No.	Compound	molar weight [g/mole]	CAS number	Inhalation LC ₅₀ [ppm]	POCP	fish LC ₅₀ [mg L ⁻¹]	log BCF	Biodegradability half-life [day]	Oral LD ₅₀ [mg kg ⁻¹]	EPA/IACR cancer class
hydrocarbons	1	pentane	72	109-66-0	123000	39.5	10	0	3	5000	0
	2	hexane	86	110-54-3	48000	48.2	2.5	0.35	1	25000	0
	3	cyclohexane	84	110-82-7	34000	23	4.3	2.38	28	12700	0
	4	heptane	100	142-82-5	103000	49.4	375	2.53	7	5000	0
	5	octane	114	111-65-9	25260	46.1	100	3.289	7	5000	0
	6	isooctane	114	540-84-1	3100	45	0.1	2.36	1	5000	0
	7	nonane	128	111-84-2	3200	41.4	300	2.651	7	4000	0
	8	decane	142	124-18-5	2738	38.4	500	2.158	7	5000	0

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	9	undecane	156	1120-21-4	844	38.4	1000	2.722	28	5000	0
	10	dodecane	170	112-40-3	284	31.1	1000	2.497	7	5000	0
	11	benzene	78	71-43-2	13000	21.8	5.9	1.5	6	930	10
	12	toluene	92	108-88-3	26000	63.7	7.63	1.954	3	5500	0
	13	o-xylene	106	95-47-6	6125	105.3	17	1.702	4	4300	0
	14	m-xylene	106	108-38-3	6700	47.4	11	1.764	24	5000	0
	15	p-xylene	106	106-42-3	5000	47.2	16	1.3	7	2100	0
	16	styrene	104	100-42-5	9500	14.2	10	1.13	28	5000	8
alcohols	17	methanol	32	67-56-1	64000	14	15400	0.65	1	1200	0
	18	ethanol	46	64-17-5	20000	39.9	14500	-1.63	2	3450	0
	19	1-propanol	60	71-23-8	68000	56.1	4480	-1.07	3	1870	0
	20	isopropanol	60	67-63-0	16000	18.8	5000	0.5	1	3600	0
	21	1-butanol	74	71-36-3	8000	62	1200	-0.439	7	790	0
	22	isobutanol	74	67-63-0	8000	44.7	1220	0.5	7	2460	0
	23	sec-butyl alcohol	74	78-92-2	16500	44.7	3670	0.5	7	2193	0
	24	tert-butyl alcohol	74	76-65-0	3300	10.6	1940	-1.45	7	2743	0
	25	1-pentanol	88	71-41-0	6000	59.5	370	0.463	4	2200	0
	26	1-hexanol	102	111-27-3	5000	45	98	0.711	5	720	0
	27	1-heptanol	116	111-70-6	9333	45	38	1.09	20	5500	0
	28	1-octanol	130	111-87-5	6500	45	13.3	1.6	30	3200	0
	29	1-nonanol	144	143-08-8	6500	45	18	2.2	40	3560	0
	30	allyl alcohol	58	107-18-6	76	45	0.32	0.5	3	64	0
	31	benzyl alcohol	108	100-51-6	1000	45	10	0.602	5	1230	0
	32	glycerol	92	56-81-5	n.a. [†]	22	5800	0.5	4	25000	0
	33	phenol	94	108-95-2	82	-119	12	2.2	2	270	0
	34	o-cresol	108	95-45-7	40	19	6.2	1.1	2	121	0

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	35	m-cresol	108	108-39-4	160	19	8.9	3	2	242	0
	36	p-cresol	108	106-44-5	160	19	7.9	1.26	2	207	0
ethers	37	diethyl ether	74	60-29-7	32000	44.5	2560	0.73	1	1215	0
	38	methyl – tert butyl ether	88	1634-04-4	23500	15.2	672	1.5	500	4000	0
	39	furan	68	110-00-9	60	20	61	0.332	300	5	8
	40	tetrahydrofuran	72	109-99-9	18000	20	2160	0	40	1650	0
aldehydes	41	ethanal	44	75-07-0	13300	64.1	31	0.5	15	661	8
	42	1-propanal	58	123-38-6	1110	17	14	0.5	11	1410	0
	43	1-butanal	72	123-72-8	6400	17.1	16	0.5	5	5900	0
	44	furfural	96	98-08-1	100	17	32	0.5	5	65	0
ketones	45	acetone	58	67-64-1	21000	-4	5540	-0.187	3	3200	0
	46	2-butanone	72	78-93-3	10900	37.3	3220	0	11	2737	0
	47	2-pentanone	86	107-87-9	2000	17.8	1240	0.477	7	1600	7
	48	3-pentanone	86	96-22-0	5000	16	1530	0.3	7.5	2140	0
	49	methyl isobutyl ketone	100	108-10-1	3000	49	720	0.5	11	2080	7
	50	2-hexanone	100	591-78-6	8000	57.2	428	0.778	5	2590	0
	51	2-heptanone	114	110-43-0	8000	59.9	130	0.39	21	1760	0
	52	cyclohexanone	98	108-94-1	8000	12	527	0.5	7	1620	0
terpenes	53	d-limonene	136	5989-27-5	5000	134	0.72	2.673	7	4400	0
	54	p-cymene	134	99-87-6	1750	100	48	2.457	14	4750	0
	55	α-pinene	136	80-56-8	1110	84	0.28	2.719	21	3700	0
	56	β-pinene	136	127-91-3	3500	70	1	2.503	21	4700	0
organic acids	57	formic acid	46	64-18-6	825	4	46	0	1	730	0
	58	acetic acid	60	64-19-7	16000	6	1000	0	2	3310	0
	59	butyric acid	88	107-92-6	2200	8	66.4	0.5	2	2940	0
	60	isobutyric acid	88	79-31-2	2800	8	228	0.5	3	226	0

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	61	valeric acid	102	109-52-4	1000	8	39	0.5	4	600	0
	62	hexanoic acid	116	142-62-1	865	8	88	0.5	5	1900	0
esters	63	ethyl acetate	88	141-78-6	8000	10	350	1.48	5	5620	0
	64	ethyl acrylate	100	140-88-5	1414	18	4.6	0.316	18	800	8
	65	butyl lactate	146	138-22-7	840	18	75	0.5	10	5000	0
	66	ethyl lactate	188	97-64-3	1120	0	320	0.5	14	2500	0
	67	methyl formate	60	107-31-3	2120	0	120	0.5	2	1622	0
	68	methyl acetate	74	79-20-9	16200	5.9	300	0.5	21	5000	0
	69	methyl laurate	214	111-82-0	571	0	0.52	2.466	5	2000	0
chlorinated	70	dichloromethane	85	75-09-2	16000	2.3	193	0.7	10	1600	8
	71	chloroform	119	67-66-3	15000	0.4	14	0.8	28	908	8
	72	carbon tetrachloride	154	56-23-5	8000	15	24.3	1.47	250	2350	8
	73	trichloroethene	131	79-01-6	8450	32.5	3.1	1.23	180	2400	10
	74	tetrachloro-ethene	166	127-18-4	4100	1.4	9.8	1.69	230	2629	8
	75	1,2-dichloro-ethane	111	542-75-6	950	15	2.94	0.863	28	250	7
	76	1,3-dichloro-propene	261	87-68-3	34	15	0.14	4.7	205	82	7
	77	hexachlorobutadiene	99	107-06-2	1414	15	136	0.3	180	413	7
others	78	acetonitrile	41	75-05-8	3587	25	1640	0.3	28	1320	0

IARC cancer class was translated into the numerical values in the following way: Group 1 (human carcinogen) – 10; Group 2A (probable human carcinogen) – 8; Group 2B (possible human carcinogen) – 7; Groups 3 and 4 (not classifiable and probable human noncarcinogen, respectively) – 0.

Notes and references:

* C.M. Alder, J. D. Hayler, R.K. Henderson, A.M. Redman, L. Shukla, L.E. Shuster, H.F. Sneddon, Green Chem. 2016,18, 3879-3890.

† Not needed as glycerol does not partition to air