Supporting Information

One Small Molecule as Theranostic Agent: Naphthalimide Dye for Subcellular Fluorescent Localization and Photodynamic Therapy *in vivo*

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Table of Contents:

1. Materials and Instruments

2. Synthesis

3. Method

- 3.1 Spectroscopic materials and methods
- 3.2 Determination of fluorescent quantum yield
- 3.3Determination of singlet oxygen quantum yield
- 3.4 Cell culture
- 3.5 Cell viability assay
- 3.6 In vitro imaging of 1a
- 3.7 In vivo experiments
- 3.8 H&E staining
- 3.9 Mechanistic study
- 3.10 Statistical analysis

4. Data

5. NMR spectra

6. References

1. Materials and Instruments

All chemical reagents and solvents are analytic grade and chemicals were purchased from TCI and J&K reagent Co. Ltd without further purification. Compounds were purified through column chromatography with silica gel (HaiYang, Qingdao) 300-400 mesh.

¹H NMR and ¹³C NMR spectra were collected by Bruker AM-400 or AM-500 spectrometer (in CDCl₃, CD₃OD and DMSO- d_6 , TMS as the internal standard), HRMS was conducted in the Analysis and Test Center of East China University of Science and Technology (ECUST). The purifications of final compounds were tested by HPLC performed on a Hewlett-Packard 1100 system chromatograph with a photodiode array detector using a Zorbax RX-C18, 250 mm × 4.6 mm column. The mobile phase was a gradient of 30-100% acetonitrile (solvent 1) and 10 mMNH₄OAc in water (pH 6.0) (solvent 2) at a flow rate of 1.0 mL/min (0-15.0 min, 30-100% solvent 1). The melting points of compounds were obtained by WRS-1B-digital melting point apparatus. Spectroscopic properties were measured on All Varian Cary 100 UV spectrophotometer and Varian Cary Eclipse FL spectrophotometer and the singlet oxygen quantum yields were accomplished by monochromatic light system composed of CM 110 1/8mmonochromator and ASB-XE-175 Xenon light source from Spectral Products. Optical imaging was performed through fluorescence microscope and confocal microscopy (Nikon, Japan).

2. Synthesis

4-bromo-3-nitro-1,8-naphthalic anhydride (2).

4-bromo-1,8-naphthalic anhydride (5.54 g, 20 mmol) was stirred in 20 mL concentrated sulphuric acid at 0 °C, sodium nitrate (2.10 g, 24 mmol) was then added slowly within 30 min and the mixture was stirred at room temperature for 3 hours. The mixture was then poured into 200 mL ice water accompanied with stirring fierce and filtrated to get residue. The residue was purified through recrystallization with acetic acid to give out **2**, light yellow solid, yields: 72%. mp> 300 °C.¹H NMR (400 MHz, DMSO-*d*₆): δ 8.91 (s, 1H), 8.82 (d, *J* = 8.4 Hz, 1H), 8.73 (d, *J* = 7.2 Hz, 1H), 8.18 (t, *J* = 7.6 Hz, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): 160.10, 159.50, 149.83, 135.69, 135.37, 131.40, 130.96, 130.93, 126.03, 122.25, 121.54, 120.90. MS (EI) calcd for C₁₂H₄BrNO₅[M]⁺: 320.9, found: 320.9.

4-((4-bromophenyl)thio)-3-nitro-1,8-naphthalic anhydride (3a).

Compound **2** (2.00 g, 6.21 mmol) was dissolved in 25 mL ethanol, 4-bromobenzenethiol (1.47 g, 7.76 mmol) was then added and the reaction was heated at 80 °C for 5 hours. The mixture was evaporated in vacuum and purified on silica gel chromatography (PE/DCM = 3/1, v/v) to perform**3a**, yellow solid, yield: 82%. mp 194.5-195.4 °C.¹H NMR (400 MHz, DMSO- d_6): δ 8.97 (s, 1H), 8.82 (dd, $J_I = 0.8$ Hz, $J_2 = 8.4$ Hz, 1H), 8.67 (dd, $J_I = 1.2$ Hz, $J_2 = 7.2$ Hz, 1H), 8.06 (t, J = 7.6 Hz, 1H), 7.51 (dd, $J_I = 2.0$ Hz, $J_2 = 8.8$ Hz, 2H), 7.16 (dd, $J_I = 2.0$ Hz, $J_2 = 8.8$ Hz, 2H). ¹³C NMR (100 MHz, DMSO- d_6): 160.10, 159.74, 153.44, 134.90, 134.49, 134.10, 133.07, 132.36, 131.39, 131.29, 131.04, 130.97, 125.41, 123.29, 121.47, 121.26. HRMS (ESI) calcd for C₁₈H₉BrNO₅S [M+H]⁺: 429.9379, found: 429.9384

4-phenylthio-3-nitro-1,8-naphthalic anhydride (3b).^{S1}

Compound **2** (2.00 g, 6.21 mmol) was dissolved in 25 mL ethanol, benzenethiol (0.86 g, 7.76 mmol) was then added and the reaction was heated at 80 °C for 5 hours. The mixture was evaporated in vacuum and purified on silica gel chromatography (PE/DCM = 3/1, v/v) to give out **3b**, yellow solid, yield: 75%. mp 177.6-178.9 °C.¹H NMR (400 MHz, CDCl₃+CD₃OD): δ 8.92 (d, J = 8.4 Hz, 1H), 8.82 (s, 1H), 8.73 (d, J = 7.2 Hz, 1H), 7.90 (t, J = 8.0 Hz, 1H), 7.26-7.19 (m, 3H), 7.11 (d, J = 7.6 Hz, 2H).¹³C NMR (100 MHz, CDCl₃+CD₃OD): 159.15, 158.88, 152.24, 136.43, 136.19, 135.63, 135.40, 130.62, 130.19, 129.80, 129.70, 129.36, 128.76, 128.34, 127.09, 126.52.

4-((4-bromophenyl)thio)-3-amino-1,8-naphthalic anhydride (4a).

Stannous chloride dihydrate (2.26 g, 10 mmol) was stirred in 15 mL concentrated hydrochloric acid at room temperature for 10 mins, **3a** was added slowly and the reaction was then heated at 85 °C for 3 hours, the mixture was filtered and washed with water until the filtrate was neutral, and residue was dried and purified on silica gel chromatography (PE/DCM = 1/1, v/v) to give out **4a**, yellow green solid, yield: 85%. mp> 300 °C.¹H NMR (400 MHz, CDCl₃): δ 8.63 (d, *J* = 8.4 Hz, 1H), 8.37 (d, *J* = 7.6 Hz, 1H), 8.15 (s, 1H), 7.73 (t, *J* = 7.6 Hz, 1H), 7.32 (d, *J* = 8.8 Hz, 2H), 6.90 (d, *J* = 8.8 Hz, 2H), 5.06 (s, 2H). ¹³C NMR (100 MHz, CDCl₃): 168.68, 160.55, 149.37, 139.26, 135.63, 133.74, 132.35, 131.72, 129.10, 128.87, 128.04, 123.55, 120.85, 119.87, 118.85, 112.26. HRMS (ESI) calcd for C₁₈H₁₁BrNO₃S [M+H]⁺: 399.9638, found: 399.9639.

4-phenylthio-3-amino-1,8-naphthalic anhydride (4b). ^{S1}

Stannous chloride dehydrate (2.26 g, 10 mmol) was stirred in 15 mL concentrated hydrochloric acid at room temperature for 10 mins, **3b** was added slowly and the reaction was then heated at 85 °C for 3 hours, the mixture was filtered and washed with water until the filtrate was neutral, and

residue was dried and purified on silica gel chromatography (PE/DCM = 1/1, v/v) to give out **4b**, yellow green solid, yield: 80%. mp 225.9-226.7 °C.¹H NMR (400 MHz, CDCl₃+CD₃OD): δ 8.68 (d, J = 8.4 Hz, 1H), 8.36 (d, J = 7.2 Hz, 1H), 8.16 (s, 1H), 7.72 (t, J = 8.0 Hz, 1H), 7.22-7.15 (m, 3H), 7.03 (d, J = 7.6 Hz, 2H), 5.09 (s, 2H).¹³C NMR (100 MHz, CDCl₃+CD₃OD): 160.80, 160.74, 149.42, 135.73, 134.47, 132.01, 129.30, 128.94, 128.62, 126.55, 126.12, 124.37, 123.56, 120.40, 118.64, 113.15. MS (ESI) calcd for C₁₈H₁₁NO₃S [M+H]⁺: 322.0, found: 322.0.

9-bromo-*N*-(2-(2-hydroxyethoxy)ethyl)-benzo[k,l]thioxanthene-3,4-naphthalimide (1a) and 9bromo-*N*-(2-(2-hydroxyethoxy)ethyl)-benzo[b]furano[2,1-c]naphthalimide (1b).

Sodium nitrite (0.38 g, 5.5 mmol) was added to 32 mL concentrated sulphuric acid and 12 mL acetic acid in three times at -5 °C, 4a(2.00 g, 5 mmol) was then added slowly within 45 mins, the mixture was stirred for another 1 hour to give out black viscous liquid. At the same time, anhydrous cupric sulfate (34.00 g, 213 mmol) was dissolved in 350 mL deionized water and 45 mL acetic acid, and the blue solution was heated under reflux. Then the black liquid was added dropwise to the blue solution, the mixture was sequentially warmed at 100 °C for another 6 hours, and filtered to give out the brown solid. The solid was then stirred in 10 mL anhydrous ethanol withdiglycolamine (0.22 g, 0.20 mL) at 50 °C for 3 hours. The mixture was evaporated in vacuum and residue was purified on silica gel chromatography (PE/DCM = 1/1, v/v and then DCM/MeOH = 100/1) to give out **1b**, light yellow solid, yield: 70% and 1a, orange solid, yield: 18%. mp (1a) 217.7-218.4 °C. ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 8.47 (d, J = 8.0 Hz, 1H), 8.30 (d, J = 8.0 Hz, 1H), 8.23 (s, 1H), 8.05 (d, J = 8.0 Hz, 1H, 7.46 (d, J = 9.0 Hz, 1H), 7.43 (d, J = 8.0 Hz, 1H), 7.20 (d, J = 9.0 Hz, 1H), 4.37 (t, J = 9.0 Hz, 1H5.5 Hz, 2H), 3.79 (t, J = 5.5 Hz, 2H), 3.62 (t, J = 4.5 Hz, 2H), 3.58 (t, J = 4.5 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃): 164.15, 163.69, 140.11, 135.53, 132.91, 132.71, 131.08, 130.51, 130.30, 129.65, 128.92, 127.70, 125.35, 121.46, 121.39, 120.69, 119.56, 117.90, 72.50, 68.43, 61.52, 39.80. HPLC purity: 95.3%, Retention time = 13.73 min. HRMS (EI) calcd for $C_{22}H_{16}BrNO_4S$ [M]⁺: 468.9983, found: 468.9982. mp (**1b**) 201.3-201.7 °C. ¹H NMR (400 MHz, CDCl₃): δ 8.67 (s, 1H), 8.47 (d, J = 7.2 Hz, 1H), 8.14 (d, J = 8.0 Hz, 1H), 8.02 (d, J = 1.6 Hz, 1H), 7.70 (t, J = 8.0 Hz, 1H), 7.70 (t, J = 8.0 Hz, 1H), 8.14 (d, J = 8.07.64 (d, J = 8.4 Hz, 1H), 7.51 (dd, $J_1 = 1.6$ Hz, $J_2 = 8.4$ Hz, 1H), 4.41 (t, J = 5.6 Hz, 2H), 3.91 (t, J= 5.6 Hz, 2H), 3.75~3.73 (m, 4H), 2.36 (brs, 1H). ¹³C NMR (100 MHz, CDCl₃): 164.12, 163.91, 143.92, 137.13, 137.10, 131.80, 131.02, 130.43, 129.75, 127.50, 126.62, 126.58, 124.81, 124.63, 124.04, 123.25, 119.70, 119.62, 72.38, 68.49, 61.90, 39.72. HPLC purity: 99.5%, Retention time = 14.34 min. HRMS (EI) calcd for C₂₂H₁₆BrNO₄S [M]⁺: 468.9983, found: 468.9981.

N-(2-(2-hydroxyethoxy)ethyl)-benzo[k,l]thioxanthene-3,4-naphthalimide (1c) and N-(2-(2-hydroxyethoxy)ethyl)-benzo[b]furano[2,1-c]naphthalimide (1d). ^{S1}

Sodium nitrite (0.38 g, 5.5 mmol) was added to 32 mLconcentrated sulphuric acid and 12 mL acetic acid in three times at -5 °C,4b (1.60 g, 5 mmol) was then added slowly within 45 mins, the mixture was stirred for another 1 hour to give out black viscous liquid. At the same time, anhydrous cupric sulfate (34.00 g, 213 mmol) was dissolved in 350 mL deionized water and 45 mL acetic acid, and the blue solution was heated under reflux. Then the black liquid was added dropwise to the blue solution, the mixture was sequentially warmed at 100 °C for another 6 hours, and filtered to give out the brown solid. The solid was then stirred in 10 mL anhydrous ethanol withdiglycolamine (0.22 g, 0.20 mL) at 50 °C for 3 hours. The mixture was evaporated in vacuum and residue was purified on silica gel chromatography (PE/DCM = 1/1, v/v and then DCM/MeOH = 100/1) to give out 1d, light yellow solid, yield: 55% and 1c, orange solid, yield: 20%. mp (1c) 192.5-193.8 °C.¹H NMR $(500 \text{ MHz}, \text{CDCl}_3)$: δ 8.41 (d, J = 8.0 Hz, 1H), 8.24 (d, J = 8.0 Hz, 1H), 8.02 (d, J = 7.5 Hz, 1H), $7.94 (d, J = 8.0 Hz, 1H), 7.25 \sim 7.34 (m, 4H), 4.39 (t, J = 5.0 Hz, 2H), 3.87 (t, J = 5.0 Hz, 2H), 3.72$ 3.73 (m, 2H), 3.70-3.71 (m, 2H).¹³C NMR (125 MHz, CDCl₃): 164.12, 163.69, 140.74, 136.77, 132.68, 131.62, 130.88, 130.32, 130.00, 127.84, 127.61, 126.40, 126.10, 125.34, 120.92, 120.35, 119.14, 117.76, 72.30, 68.49, 61.90, 39.58.HPLC purity: 98.3%, Retention time = 11.61 min. HRMS (EI) calcd for C₂₂H₁₇NO₄S [M]⁺: 391.0878, found: 391.0875. mp (1d) 160.0-160.7°C.¹H NMR (400 MHz, CDCl3): δ 8.89 (s, 1H), 8.45 (d, J = 7.2 Hz, 1H), 8.19 (d, J = 8.0 Hz, 1H), 8.09 $(dd, J_1 = 2.0 Hz, J_2 = 5.2 Hz, 1H), 7.84 (dd, J_1 = 2.8 Hz, J_2 = 5.2 Hz, 1H), 7.67 (t, J = 8.0 Hz, 1H),$ $7.48 \sim 7.50$ (m, 2H), 4.41 (t, J = 5.2 Hz, 2H), 3.91 (t, J = 5.2 Hz, 2H), 3.72 - 3.74 (m, 4H), 2.76 (brs, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆): 162.78, 162.71, 142.03, 137.75, 134.69, 131.99, 129.75, 129.36, 127.37, 127.34, 125.64, 125.34, 125.23, 123.77, 122.82, 122.24, 122.01, 118.64, 72.08, 66.88, 60.21, 38.71. HPLC purity: 99.5%, Retention time = 12.17 min. HRMS (EI) calcd for C₂₂H₁₇NO₄S[M]⁺: 391.0878, found: 391.0881.

3. Methods

3.1 Spectroscopic materials and methods

Spectroscopic properties were measured on All Varian Cary 100 UV spectrophotometer and Varian Cary Eclipse FL spectrophotometer and the singlet oxygen quantum yields were accomplished by monochromatic light system composed of CM 110 1/8mmonochromator and ASB-XE-175 Xenon light source from Spectral Products. Optical imaging was performed through fluorescence microscope and confocal microscopy (Nikon, Japan).

3.2 Determination of fluorescent quantum yield

All compounds were dissolved in ethanol at room temperature. Absorbance spectra and relevant fluorescence spectra were collected by a Varian Cary 100 UV-Vis spectrophotometer and a Varian Cary Eclipse Fluorescence spectrophotometer. The molar extinction coefficients were obtained by

Lambert-Beer's Law and fluorescence quantum yields were calculated as below:

$$\Phi_F = \Phi_R \left(\frac{S}{S_R}\right) \left(\frac{\eta}{\eta_R}\right)_{(1)}$$

where Φ is the fluorescence quantum yield, S is the slope of integrated fluorescence intensity vs. absorbance, η is the relatively refractive index of solvent, subscript _R indicate the reference standard of 4-*N*-butylamide-*N*-butyl-1,8-naphthalimide, whose fluorescent quantum yield (Φ_R) was 0.81 in ethanol.

3.3 Determination of singlet oxygen quantum yield S2

To quantify the irradiation-induced singlet oxygen (${}^{1}O_{2}$), a singlet oxygen trap 1,3-Diphenylisobenzofuran (DPBF) was used in air-saturated acetonitrile. Compounds were dissolved in acetonitrile and added to 100-fold excess DPBF stock solution. After laser irradiation with a monochromator (ASB-XE-175 Xenon light source), the absorbance of DPBF at 410 nm was recorded every half or one minute. Methylene Blue (whose singlet oxygen quantum yield was 0.52 in acetonitrile) was utilized as a contrast. The singlet oxygen quantum yields were determined using the equation:

$$\Phi_{\Delta(T)} = \Phi_{\Delta(MB)} \left(\frac{S_T}{S_{MB}} \right) \left(\frac{F_T}{F_{MB}} \right)_{(2)}$$

Where Φ_{Δ} is the singlet oxygen quantum yield, S is the slope of absorbance of DBPF vs. irradiation time and F is the absorption correction factor.

3.4 Cell culture

Human cell lines (MCF-7, HCT116, U87, MKN45, A375) were purchased from The Cell Bank of Type Culture Collection of the Chinese Academy of Sciences (Shanghai, China). Cells were cultured in RPMI-1640 or DMEM medium, supplemented with 10% fetal bovine serum (FBS, Hyclone, USA) and 1% antibiotic- antimycotic (ABAM Life Technologies, California, USA) in a humidified atmosphere of 95% air and 5% CO₂.

3.5 Cell viability assay

To evaluate the tumor inhibitory efficacy of **1a-1d**, MCF-7 cells (1×10^4) were suspended in medium and plated in 96 well plates. After treatment with indicated concentrations $(0-20 \ \mu\text{M})$ of **1a-1d** for 48 hours, cells were irradiated with the 100-watt quartz-halogen lamp. The fluorescence rate was 240 W/m², measured by a photo radiometer (Delta Ohm, Padua, Italy). After 20 mins of PDT treatment and 4 hours of MTT incubation, the cell viability was determined by MTT assay.^{S3}As the optimal candidate, **1a** was then solely evaluated in other cell lines including HCT116, U87,

MKN45and A375 cells by the same methods above.

3.6 In vitro imaging of 1a

3.6.1 Fluorescence microscopic imaging

With a aim to assess fluorescent imaging results *in vitro*, A375 cells (1×10^5) were treated with **1a** (2.5 μ M, 2 mL) for 48 h. After that, cells were exposed to the light of 28.8 J/cm² within 20 mins for PDT. 4 hours later, the cells were rinsed with PBS for three times. Images were exposed by excitation at 460 nm and emission at 520 nm under an inverted Fluorescence microscope (Nikon, Japan).

3.6.2 Subcellular localization studies

A375 cells (1×10^5) were seeded on a coverslip and co-incubated with **1a** (2.5 µM, 2 mL) for 48 hours. After 20 mins of PDT treatment, the cells were rinsed with PBS for three times and incubated in culture medium with 100 nMMitoTracker Red (invitrogen, USA) for 20 mins, 100 nM ER-Tracker Red for 20 mins (Beyotime, China), 50 nM Lysosome Tracker Red (Beyotime, China) for 30 mins, 1 µM DAPI (Beyotime, China) for 15 mins, respectively. The stained live cells were observed by confocal microscopy (Nikon, Japan).

3.7 In vivo experiments

All six-weeks old nude mice with half males and half females were purchased from Shanghai Slac Laboratory Animal Co. Ltd (The license of experimental animal: SCXK(shanghai) 2012-0002). The mice were maintained under the 12 hour light/dark cycles condition in the specific pathogen free (SPF) cleanroom with fresh air (more than 50%), appropriate temperature (22-26 oC) and humidness (40%-60%). The mice were freely fed with enough sterilized water and food bought from Shanghai Slac Laboratory Animal Co. Ltd. All materials and containers were disinfected and sterilizated before use. After the treatment, the mice were anesthetized and euthanasia in accordance with the international ethical standards. All experimental procedures involving animals in this study were reviewed and approved by the institution of ethics committee in ECUST and strictly conducted according to the guidelines of Care and Use of Laboratory Animals of China (GB14925-2001) for animal experimentation.

Each nude mouse was anesthetized with pelltobarbitalum natricum and subcutaneously inoculated with injection of 1×10^{6} A375 cells. When the volume of the tumor grew to 100 mm³ size, mice were randomizedinto three groups: control group, medicine group and PDT group (each group had 6 mice). **1a** (5 mg/kg) was injected into both medicine group and PDT group of mice in situ while saline in the control group. After 8 hours, 600-micron bare fiber was used to give PDT treatment by white light from ASB-XE-175 Xenon light sourceand it was totally irradiated with a light-dose of 259.2 J/cm² within 60 min. All groups were administered the injection twice a week,

total 6 times during the treatment time. Tumor sizes were measured every 3 days using micrometer calipers. 4 hours after the laser treatment, tumor xenografts were removed and measured. During the experiment, tumor sizes were measured every 3 days using micrometer calipers and tumor volume (TV) was calculated using the following formula: TV (mm³) = D × d²/2, D is the shortest diameters while d is the longest one. Relative tumor volume (RTV) was calculated according to the equation: RTV = V_t/V₀, where V₀ is the tumor volume at day 0 and V_t is the tumor volume at day t.

3.8 H&E staining

Hematoxylin-eosin staining was a conventional method for observing the histological changes of the tissues. After indicated experiments, all of the tumor tissues and several visceral organs tissues were collected and embedded in paraffin. The slices were separately treated with hematoxylin and eosin solution and images were taken with fluorescence microscope (Nikon, Japan).

3.9 Mechanistic study

3.9.1 Measurement of oxidative damage

DCFH-DA method(2',7'-dichlorofluorescein diacetate) was employed to evaluate intracellular reactive oxygen species (ROS) level.^{S4}After 20 mins PDT treatment, cells were collected in serum-free medium and co-incubated with 10 μ M DCFH-DA at 37 °C for another 20 mins. Cells were then washed three times with PBS and ROS generation was measured using fluorescence intensity of excitation at 488 nm and emission at 525 nm by EnSpire® Multimode Plate Reader (Perkin Elmer, Boston, MA).

3.9.2 Hoechst 33342 staining

A375 cells (1×10^5) were seeded on glass bottom dishes, cultured and treated with **1a** as described above. The cells were stained with Hoechst 33342 (Sigma, USA) for 20 mins at 37 °C in the dark condition after rinsed with PBS for twice. The images were exposed by the fluorescence microscope (Nikon, Japan).

3.9.3 Western blot analysis

After subjected to the indicated treatment, cells were harvested and lysed with cell lysis solution. Total proteins were quantified using the BCA protein assay kit. Equal amount of protein was run on 10% sodium dodecylsulfate (SDS) polyacrylamide gels and then transferred to PVDF membranes. The membranes were blocked in TBS-Tween 20 solution with non-fat dry milk prior to incubated overnight with primary antibodies PARP (SAB, USA), cleaved-PARP (SAB, USA), Caspase 3 (GenTex, USA) and cleaved-Caspase 3 antibodies (GenTex, USA) (1 mg/mL for each antibody). Following three washes with TBS-T, membranes were incubated with HRP conjugated secondary antibody for 2 h at room temperature. Then membranes were washed by TBS-T three times again,

developed in ECL and visualized with X-ray film. β -actin (GenTex, USA) was used as an internal control.

3.9.4 Immunohistochemistry

The expression of Caspase 3 in tumor tissues of nude mice model was assessed by immunohistochemistry assay. The samples were embedded in paraffin and stained by Caspase 3 antibodies (Bioworld, China). ^{S5} Images were then taken with a fluorescence microscope (Nikon, Japan).

3.9.5 Tunel assay

Tumor apoptosis was performed following the manufacturer's instructions of terminal deoxynucleotidyl transferase Dutp nick end labeling (TUNEL) assay (Roche Applied Science, Germany). The slides were photographed with a fluorescence microscope (Nikon, Japan).

3.10 Statistical analysis

All results shown in the study were obtained in at least three independent experiments and were expressed as mean \pm standard deviation (SD). Statistical analysis of data was performed using Excel (Microsoft) and GraphPad Prism 5.0 software (GraphPad Software Inc, CA, USA). Significant differences are considered as *p< 0.05; ** p< 0.01; *** p< 0.001.

4. Data



Fig. S1The UV absorption spectra (blue line) and fluorescent emission spectra (red line) of 1a (25



 μ M, slit: 2.5, 2.5), **1b**(25 μ M, slit: 5, 5), **1c** (25 μ M, slit: 2.5, 2.5) and **1d** (25 μ M, slit: 5, 5) in EtOH.

Fig. S2 a) MTT assays of 1a-1d exposed with laser irradiation of 0 and 28.8 J/cm² in MCF-7 cells;
b) MTT assays of 1a exposed with laser irradiation of 0 and 28.8 J/cm² light in A375, HCT116, MKN45 and MCF-7 cells.



Fig. S3 Fluorescence confocal microscopy imaging of **1a** in A375 cells.Co-localizaiton with Mito-Tracker Red, ER-Tracker Red and Lysosome Tracker Red: a) white light; b) **1a**; c) organelles tracker probes; d) blue DAPI; e) the merge results of b and c; f) the merge results of b and d; g) the merge results of b, c and d. Bars: 10 μm



Fig. S4 Pathological changes of major organs (including kidney, liver, spleen, heart and lung) and tumor tissues by H&E stained images. Bars: 50 μm.



Fig. S5 Immunohistochemical assessment of cleaved-Caspase 3 expression of tumor cells after 1ainduced PDT treatment. Scale bars: 50 μm;



Fig. S6 TUNEL assays of apoptotic cells in the tumor tissues. The nuclei stained with DAPI (bright blue) and apoptotic cells (bright red) were observed using fluorescent microscopy. Bars: 50 μm

5. NMR spectra



Fig. S7The ¹H-NMR and ¹³C-NMR spectrum of 1a

Elemental Composition Report

Multiple Mass Analysis: 53 mass(es) processed Tolerance = 5.0 mDa / DBE: min = -1.5, max = 60.0 Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions 6283 formula(e) evaluated with 255 results within fimits (all results (up to 1000) for each mass) Elements Used: C: 0-24 H: 0-18 N: 0-1 O; 0-5 S: 0-1 798r 0-1 818r; 0-1

RLZ-2L-510-071

Waters GCT Premier

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Minimum	3.00				-1.5			
Maximum:	102.00		5.0	10.0	50.0			
Mass	ЗA	Calc. Mass	mEa	PEN	DBE	1-FIT	Form	1.5
	777	Cerc. Adaa	-00-6	CEM	JUB	1-511	TOTAK	114
57.0708	3.52	5710704	0.4	7.0	0.5	2773012.9		19
71.0860	4.59	71.0861	-0.1	-1.4	0.5	654.3		0.1
85.1023	5.05	85.1017	0.6	7.1	0.5	213.5		11.3
99.1175	3.26	99.1174	0.1	1.0	0.5	2773016.3		115
11.3.1.34.3	3.35	113.1330	1.3	11.5	0.5	57.3	C8 N	117
127,1493	3.09							
141.1660	3.24							
155.1790	3.00							
182.9087	3.18	102.9664	0.0	1.6	5.0	156.5		46 N 79Er
		192.9666	2.1	11.5	0.5	213.8		1.0 S 815r
		182.9657	3.0	1.6.4	0.5	172.7		(8 C3 79Br
		182,9717	-3.0	-16.4	0.0	159.3		110 N S 79Dr
		182.9718	-3.1	-16.9	1.0	202.0		10 N C2 813r
187.0552	7.21	187.6548	0.4	2.1	12.5	349.4		8I7
		107.0501	-2.9	-15.3	7.5	130.6	C12	N11 3
		187.0514	3.8	20.3	-1.6	205.6	C4 F	113 🕺 05 S
214.0659	3.37	214.0657	0.2	5.9	1.3.5	2773034.5	C15	88 N
		214.0664	-0.5	-2.3	4.0	2773176.0	C10	814 03 8
		214.0630	2.9	1.3.5	9.0	2773036.0	C13	H10 C3
		214.0690	-3.1	-14.5	3.5	2773169.3	C13	H12 N S
230.0181	3.53	230.01B1	0.0	0.0	3.5	850.7	C9 E	113 N O 7935
		230.0190	-0.9	-3.9	14.0	1333.2	016	115 3
		230.0 63	1.8	7.8	-1.0	3417.7		0.7 O S 81.3x
		230.0157	2.4	10.4	19.0	1355.5		31/2
		230.0154	2.7	11.7	-1.0	589.1		115 O4 79Br
		230.0214	.3.3	-14.3	-1.5	841.2		117 N O 5 798r
		230.0215	-3.4	-14.3	1.0.9	1369.5		1 6 05
		230.9215	~3.4	-14.3	-C.5	1411.4		15 N 03 816r
231.0253	7.37	231.0239	-0.6	-2.6	3.0	939.2		LIA N O 79Br
231.0203	1.31	231.9241	1.2	5.2	~1.5	997.2		the of S BiBr
			-1.5			861.7		K7 S
		231.0268		-6.5	13.5			нт 5 НЗ
		231.0235	1.0	7.8	18.5	862-1		중성수에서는 · · · · · · · · · · · · · · · · · · ·
		231.0232	2.1	9.1	-1.5	969.9		(16 04 79Br
		231.0293	-4.0	-17.3	-1.0	1049.8		(15 N 03 81Br
		231.0293	-4.0	~17.3	9.5	965.0	C12	H7 05
		231.0208	4.5	19.5	3.5	1017-0	010	EN4 O FIEr
232.0345	11.86	232,6347	~0.2	-0.9	13.0	23.2		HC 2
		232.0337	0.9	3.4	2.5	948.2		415 N O 79Br
		232.0371	-2.6	-11,2	-1.5	176.3		117 🕺 03 815r
		232.0372	-2.7	-11.6	5.0	90.8		11B C5
		232.9313	3.2	13.8	18.0	39.3	C19	H4
233.0357	3.28	233.0358	-0.1	-0.4	4.0	2773199.8	C8 5	0.1 N 05 3
20010001	2100	~~~						

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Fig. S8The HR-MS spectrum of 1a

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Fig. S9The ¹H-NMR and ¹³C-NMR spectrum of 1b

	5.0 mDa	ysis: 50 mass(e / DBE: min = -1. f					
707 formula	(e) evaluate	and Even Electron d with 202 results w		all results (u	p to 1000) fc	or each mass)	
Elements Us C: 0-22 H:		1 O: 0-4 S: 0-1	79Br: 0-1	81Br: 0-1			
RLZ-ZL-510-0	41			Waters GC	F Premier		
20160115 427		427-(87+93))					TOF MS EI+ 1.74e+004
100-							382.9453
-							
%-							383.9507
1							407.9667
-		18	232.0 37.0543	341 257.02	96 303.0345	337.9476 36	55.9424 410.9739 468.9981
0 73.	0304 115 80 100	.0090 155.9711		20 240 26	0 280 300	320 340 3	60 380 400 420 440 460
Minimum: Maximum:	3.00		5.0	10.0	-1.5		
lass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
.81.9706	3.88	181.9701 181.9731	0.5	2.7	11.5 5.0	346.1 601.6	C10 N O S C8 H7 79Br
82.9697	4.70	181.9674 182.9684	3.2 1.3	17.6	7.0	369.2 271.5	C7 H2 O4 S C7 H6 N 79Br
.02.9091	4.70	182.9717	-2.0	-10.9	0.0	278.4	C4 H10 N S 79Br
		182.9718 182.9666	-2.1 3.1	-11.5 16.9	1.0	249.6 255.2	C4 H8 N O2 81Br C5 H10 S 81Br
87.0543	9.25	182.9657 187.0548	4.0	21.9 -2.7	0.5	293.4 72.1	C4 H8 O3 79Br C15 H7
		187.0581	-3.8	-20.3	7.5	75.9	C12 H11 S
230.0202	4.93	230.0190 230.0215	1.2	5.2 -5.7	14.0 -0.5	1788.5 1913.8	C16 H6 S C6 H15 N O3 81Br
		230.0181 230.0242	2.1	9.1 -17.4	3.5 14.5	1055.2 1812.5	C9 H13 N O 79Br C15 H4 N O2
		230.0157	4.5	19.6	19.0	1793.0	C19 H2
231.0257	8.47	230.0154 231.0259	4.8	20.9	-1.0 3.0	1113.0 1365.0	C6 H15 O4 79Br C9 H14 N O 79Br
		231.0268 231.0235	-1.1 2.2	-4.8	13.5 18.5	1144.6 1170.3	C16 H7 S C19 H3
		231.0232	2.5	10.8	-1.5	1401.8	C6 H16 O4 79Br
		231.0293 231.0208	-3.6 4.9	-15.6 21.2	-1.0 3.5	1393.7 1349.9	C6 H16 N O3 81Br C10 H14 O 81Br
232.0341	14.27	232.0337 232.0347	0.4	1.7	2.5	1253.4 6.0	C9 H15 N O 79Br C16 H8 S
		232.0313	2.8	12.1	18.0	37.9	C19 H4
233.0358	3.15	233.0364 233.0391	-0.6 -3.3	-2.6 -14.2	2.5 17.5	2773048.3 2773071.3	
257.0296	15.96	257.0299 257.0272	-0.3 2.4	-1.2 9.3	15.0 10.5	441.9 502.6	C17 H7 N S C14 H9 O3 S
		257.0265	3.1 3.3	12.1 12.8	20.0	513.4 893.6	C20 H3 N C7 H16 N O4 79Br
258.0358	6.52	257.0263 258.0351	0.7	2.7	10.0	417.8	C14 H10 O3 S
		258.0344 258.0377	1.4	5.4	19.5 14.5	403.2 381.5	C20 H4 N C17 H8 N S
		258.0317	4.1	15.9	15.0 4.5	439.9 520.8	C17 H6 O3 C11 H15 N O 81Br
259.0420	5.52	258.0317 259.0422	4.1 -0.2	-0.8	19.0	61.0	C20 H5 N
		259.0429 259.0395	-0.9	-3.5 9.7	9.5 14.5	67.6 77.2	C14 H11 O3 S C17 H7 O3
		259.0395	2.5	9.7	4.0	126.4 50.4	C11 H16 N O 81Br C17 H9 N S
303.0345	7.09	259.0456 303.0354	-0.9	-13.9 -3.0	15.0	34.1	C17 H9 N S C18 H9 N O2 S
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							个共立任何证明作用。

Fig. S10The HR-MS spectrum of 1b



Fig. S11The ¹H-NMR and ¹³C-NMR spectrum of 1c

Elemental Composition Report

Multiple Mass Analysis: 22 mass(es) processed Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions 552 formula(e) evaluated with 43 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-22 H: 0-17 N: 0-1 O: 0-4 S: 0-1 RLZ-ZL-60-023b GCT Premier 20141038 614 (10.234) Cm (611-(5+626))

.

LZ-ZL-690-02 0141038 614 ((614-(5+826))		GCT Pre	tulet			TOF MS E
00						303.0	340	1.71e+0
						Į		
1						i		
%-							04.0407	
		2.				ſ	330.0579	
73.028		116.0154 143.015	10 187.0	549	1	0383 286.0326	347.0599	391.0875 استثنیت بر ا
60	50 10		160 180	200 220		60 280 300	320 340 360	380
inimum:	3.00		s 6	3.0.0	-1.5			
aximum:	100100		5.0	10.0	50.0			
938	RA	Calc. Mass	sΩπ	929 Mare	DBE	1-EIL	Formula	
6.0154	3.46	116.0170	~1.6	-13.8	2.5	2773175.0	C4 H6 N O 5	
		116.0136	1.8	15.5	7.5	2773019.0	C7 H2 N 0	
17 05 (0	7 76	116.0110	4.4	37.9	3.0	2773023.8	Cd H4 O4	
7.0549	3,76	197.0548	0.1	0.5	12.5	314.4	C15 M7	
4 0464		187.0581	-3.2	-17.1	7.5	303.7	C12 #11 8	
6.0656	3.12	216.0657	-0.1	-0.5	13.5	4.8	C16 99 N	
		214.0664	-0.6	-3.7	4.0	13.3	C10 E14 03 S	
		214.0630	2.6	12.1	9.0	9.1	013 H10 03	
		214.0690	-2.4	-15.9	8.5	6.6	C13 H12 N S	
2.0343	9.87	232.0347	-0.4	-1.7	13.0	161.1	C15 HE S	
		232.0313	3.0	22.9	19.0	180.7	C19 E4	
3.0405	1.82	233.0391	1.4	6.0	17.5	1,5	C19 E5	
		233.0425	-2.0	-8.6	12.5	13.8	C16 E9 S	
6.0383	3.16	246.0377	0.6	2.4	13.5	2773211.5	С16 Н8 N S	
		246.0351	3.2	13.0	9.0	2773217.5	C13 M10 D3 3	
		246.0344	3.9	15.9	18.5	2773062,6	C13 E4 N	
\$8.0383	9.81	258.0377	0.6	2.3	14.5	429.6	C17 EC N S	
		258.0351	3.2	12.4	13.0	475.3	C14 E10 03 S	
		258.0344	3.9	15.1	19.5	472.3	C20 E4 N	
9.0448	5.52	259.0436	-C.8	-3,1	14.0	289.0	C17 E9 N S	
		259.0423	1.9	7.3	9.5	315.8	C14 H11 03 5	
		259.0422	2.5	10.0	19.0	306.7	C20 1:5 N	
0.0371	4.21	260.0391	-1.0	-3.8	9.5	67.0	C13 E10 N Q3	S
		260.0348	2.3	0.0	14.5	50.4	C16 E6 N 03	
6.0326	10.16	286.0327	-0.1	-0.3	15.5	73.3	CIB FR N G S	1
		286.0300	2.6	9.1	11.0	84.0	C15 H10 04 3	
		286.0293	3.3	11.5	20.5	124.7	C21 H4 N O	
3.0340	100,00	303.0354	-1.4	-4.6	15.0	113.8	C18 E9 N 02	5
	1	303.0323	2.0	6.6	20.0	285.3	C21 E5 N 02	· ·
4.0407	27.74	304.0399	0.8	2.6	19.5	43.8	C21 H6 N 02	
	21414	304.0132	-2.5	-8.2	14.5	50.2	C18 H10 N 02	з
5.0390	7.97	305.0425	-3.5	-11.5	16.5	85.9	C22 H9 D	ч Ч
6.0435	7.31	316.0432	0.3	0.9	15.5	308.8	C19 H10 N 02	з
0-0400	1 + 4.k	316.0399	3.6	11.4	20.5	326.6	C22 E6 N 02	22
7 0514	6 02					3.7		5
17.0514	6.02	317.0511	0.3 3.7	0.9	15.0			8
0.0461	2 22	317.0477		11.7	20.0	8.1		3.
0.0451	3.22	328.0432	1.9	5.8	16.5	1336.2	C20 H10 N 02	3
9.0514	3.22	329.0511	0.3	0.4	76.0	1704.8	C20 B11 N 02	3
30.0579	14.27	330.0599	-1.0	-3.0	15.5	226.7	C20 E12 N 02	3
31.0650	7.83	331-0667	-1-7	-5.1	15.0	13.5	C24) E13 N O2	3
47.0599	3.87	347.0616	-1.7	-4.9	15.0	1.2.0	C20 H13 N 03	8
73.0806	3.47	373.0773	3.3	8.8	16.0	16.4	C22 H15 N 03 C22 H17 N 04	Ξ
91.0875	9.78	391.0879	-0.3	-0.8	15.0	0.4	C22 H17 N 04	3

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Fig. S12The HR-MS spectrum of 1c

Page 1



Fig. S13The ¹H-NMR and ¹³C-NMR spectrum of 1d

	mula	(e) evalua			Electron lo sults withi		ll resul	ts (up	to 1000) fo	r each r	mass)					
C: 0-2	2 H	I: 0-17	N: 0	-1 0:0	-4 S: 0	-1	Wate	ers GC	T Premier							
			Cm (6	622-54:67)												MS E 18e+0
100											303.03	55				
-											3	04.0421				
%-											ſ					
-										286.03	317	330 305.0413	0.0585			
-		130.0168	143.0	154	189.0698	214.0663	233	.0408	258.0379	2.0426		05.0415	347.0	0599	39	1.088
0-4	00	120	140	160	180	200	220	240	260	280	300	320	340	360	380	-the r
Minim Maxim		3.00	00			5.0	10	0.0	-1.5							
Mass	iunit:	RA		Calc. N	lace	mDa	PE		DBE	i-	FIT	Fo	rmula			
143.0	154	3.78		143.01		-1.3		0.1	3.5		73097			2 S		
214.0		3.13		143.01	33	2.1	14	.7	8.5	27	73016	.5 C9	H3 C	03 S		
				214.06	57	0.6	2.	8	13.5 8.5	27	73021	.0 C1		N N S		
233.0	408	5.09		214.06		3.3	15	3	9.0 17.5		73024	.8 C1 C1		03		
258.0	379	11.31	1	233.042		-1.7 0.2	0.		12.5 14.5		2.5	C1 C1	7 H8	S N S		
				258.03		2.8 3.5		0.9 8.6	10.0 19.5		2.6	C1 C2	0 H4	03 S N		
259.0	448	7.16		259.04	29	-0.8 1.9	7.		14.0 9.5	23	1.4	C1 C1	4 H11	N S 03 S		
261.0	396	5.08		259.04	74	2.6	8.		19.0 13.5	5.		C2 C1	7 H9	N O S		
286.0	317	25.43	3	261.04	27	-3.0	-3	1.5 3.5	14.0		.5	C1 C1	8 H8	N 03 N O S		
				286.03	93	1.7	5.	. 4	11.0	89	.1	C1 C2	1 H4	04 S N O N O		
287.0	367	3.92		287.03	78	-0.4	-3	8.8	20.0	54).3 1.2).1	C2 C1 C1	5 H11	04 S 04		
303.0	355	100.0	20	287.03 287.04 303.03	05	2.3 -3.8 0.1	8. -1 0.	13.2	15.5 15.0 15.0	45	5.3 379.0	C1 C1	8 H9	N O S	s	
304.0		64.4		303.03	20	3.5	11	1.5	20.0		79.5	C2 C1	1 H5	N 02 N 02	s	
305.0		14.80		304.03	99	2.2	7.		19.5 18.5	51	.5	. C2 C2	1 H6	N 02 S		
306.0	412	4.12		306.03	77	3.5		L.4	18.5 15.5	27	73132	.0 C2	1 H8	N S N O2	S	
330.0		36.9		316.03	99	3.6	13	1.4	20.5 15.5	55	646644	.0 C2	2 H6 0 H12	N 02 N 02	S	
331.0	642	7.83		331.06		-2.5	-	7.6	15.0 15.5	7.	4	C2 C2	0 H12	N 02 N 03	S S	
347.0	1599	8.70 3.18		347.06	16	-1.7 -4.1		1.9	15.0 14.5	15	5.8	C2 C2		N 03 N 03	S S	
361.0	777	4.86		361.07	73	0.4		.1.	15.0 16.0	0.	7	C2 C2		N 03 N 03	SS	
391.0	881	11.8		391.08		0.3	. 0		15.0	0.	.7	C2		N 04	S	
552.0		5.25														

Fig. S14The HR-MS spectrum of 1d

Data File C:\HPCHEM\1\DATA\ZL\SEP00004.D

Sample Name: ZL09-1-071

RX C-18 4.6mm*250 Flow=1ml/min 132bar buffer



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Page 1 of 1

Fig. S15The HPLC report of 1a

Data File C:\HPCHEM\1\DATA\ZL\SEP00002.D

RX C-18 4.6mm*250 Flow=1ml/min 132bar buffer



Fig. S16The HPLC report of 1b

Data File C:\HPCHEM\1\DATA\ZL\MAR00098.D

XDB 4.6mm*250 Flow=1.0ml/min 124bar (buffer)



Instrument 1 3/17/2016 10:12:38 AM

Page 1 of 1

Fig. S17The HPLC report of 1c

Data File C:\HPCHEM\1\DATA\ZL\SEP00001.D

RX C-18 4.6mm*250 Flow=1ml/min 132bar buffer



Instrument 1 9/1/2015 3:41:20 PM

Page 1 of 1

Fig. S18The HPLC report of 1d

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