## **Supporting Information**

## One Small Molecule as Theranostic Agent: Naphthalimide Dye for Subcellular Fluorescent Localization and Photodynamic Therapy *in vivo*

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#### 1. Materials and Instruments

All chemical reagents and solvents are analytic grade and chemicals were purchased from TCI and J&K reagent Co. Ltd without further purification. Compounds were purified through column chromatography with silica gel (HaiYang, Qingdao) 300-400 mesh.

<sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were collected by Bruker AM-400 or AM-500 spectrometer (in CDCl<sub>3</sub>, CD<sub>3</sub>OD and DMSO- $d_6$ , TMS as the internal standard), HRMS was conducted in the Analysis and Test Center of East China University of Science and Technology (ECUST). The purifications of final compounds were tested by HPLC performed on a Hewlett-Packard 1100 system chromatograph with a photodiode array detector using a Zorbax RX-C18, 250 mm × 4.6 mm column. The mobile phase was a gradient of 30-100% acetonitrile (solvent 1) and 10 mMNH<sub>4</sub>OAc in water (pH 6.0) (solvent 2) at a flow rate of 1.0 mL/min (0-15.0 min, 30-100% solvent 1). The melting points of compounds were obtained by WRS-1B-digital melting point apparatus. Spectroscopic properties were measured on All Varian Cary 100 UV spectrophotometer and Varian Cary Eclipse FL spectrophotometer and the singlet oxygen quantum yields were accomplished by monochromatic light system composed of CM 110 1/8mmonochromator and ASB-XE-175 Xenon light source from Spectral Products. Optical imaging was performed through fluorescence microscope and confocal microscopy (Nikon, Japan).

#### 2. Synthesis

#### 4-bromo-3-nitro-1,8-naphthalic anhydride (2).

4-bromo-1,8-naphthalic anhydride (5.54 g, 20 mmol) was stirred in 20 mL concentrated sulphuric acid at 0 °C, sodium nitrate (2.10 g, 24 mmol) was then added slowly within 30 min and the mixture was stirred at room temperature for 3 hours. The mixture was then poured into 200 mL ice water accompanied with stirring fierce and filtrated to get residue. The residue was purified through recrystallization with acetic acid to give out **2**, light yellow solid, yields: 72%. mp> 300 °C.<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$ 8.91 (s, 1H), 8.82 (d, *J* = 8.4 Hz, 1H), 8.73 (d, *J* = 7.2 Hz, 1H), 8.18 (t, *J* = 7.6 Hz, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>): 160.10, 159.50, 149.83, 135.69, 135.37, 131.40, 130.96, 130.93, 126.03, 122.25, 121.54, 120.90. MS (EI) calcd for C<sub>12</sub>H<sub>4</sub>BrNO<sub>5</sub>[M]<sup>+</sup>: 320.9, found: 320.9.

#### 4-((4-bromophenyl)thio)-3-nitro-1,8-naphthalic anhydride (3a).

Compound **2** (2.00 g, 6.21 mmol) was dissolved in 25 mL ethanol, 4-bromobenzenethiol (1.47 g, 7.76 mmol) was then added and the reaction was heated at 80 °C for 5 hours. The mixture was evaporated in vacuum and purified on silica gel chromatography (PE/DCM = 3/1, v/v) to perform**3a**, yellow solid, yield: 82%. mp 194.5-195.4 °C.<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  8.97 (s, 1H), 8.82 (dd,  $J_I = 0.8$  Hz,  $J_2 = 8.4$  Hz, 1H), 8.67 (dd,  $J_I = 1.2$  Hz,  $J_2 = 7.2$  Hz, 1H), 8.06 (t, J = 7.6 Hz, 1H), 7.51 (dd,  $J_I = 2.0$  Hz,  $J_2 = 8.8$  Hz, 2H), 7.16 (dd,  $J_I = 2.0$  Hz,  $J_2 = 8.8$  Hz, 2H). <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ): 160.10, 159.74, 153.44, 134.90, 134.49, 134.10, 133.07, 132.36, 131.39, 131.29, 131.04, 130.97, 125.41, 123.29, 121.47, 121.26. HRMS (ESI) calcd for C<sub>18</sub>H<sub>9</sub>BrNO<sub>5</sub>S [M+H]<sup>+</sup>: 429.9379, found: 429.9384

#### 4-phenylthio-3-nitro-1,8-naphthalic anhydride (3b).<sup>S1</sup>

Compound **2** (2.00 g, 6.21 mmol) was dissolved in 25 mL ethanol, benzenethiol (0.86 g, 7.76 mmol) was then added and the reaction was heated at 80 °C for 5 hours. The mixture was evaporated in vacuum and purified on silica gel chromatography (PE/DCM = 3/1, v/v) to give out **3b**, yellow solid, yield: 75%. mp 177.6-178.9 °C.<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>+CD<sub>3</sub>OD):  $\delta$  8.92 (d, J = 8.4 Hz, 1H), 8.82 (s, 1H), 8.73 (d, J = 7.2 Hz, 1H), 7.90 (t, J = 8.0 Hz, 1H), 7.26-7.19 (m, 3H), 7.11 (d, J = 7.6 Hz, 2H).<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>+CD<sub>3</sub>OD): 159.15, 158.88, 152.24, 136.43, 136.19, 135.63, 135.40, 130.62, 130.19, 129.80, 129.70, 129.36, 128.76, 128.34, 127.09, 126.52.

#### 4-((4-bromophenyl)thio)-3-amino-1,8-naphthalic anhydride (4a).

Stannous chloride dihydrate (2.26 g, 10 mmol) was stirred in 15 mL concentrated hydrochloric acid at room temperature for 10 mins, **3a** was added slowly and the reaction was then heated at 85 °C for 3 hours, the mixture was filtered and washed with water until the filtrate was neutral, and residue was dried and purified on silica gel chromatography (PE/DCM = 1/1, v/v) to give out **4a**, yellow green solid, yield: 85%. mp> 300 °C.<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.63 (d, *J* = 8.4 Hz, 1H), 8.37 (d, *J* = 7.6 Hz, 1H), 8.15 (s, 1H), 7.73 (t, *J* = 7.6 Hz, 1H), 7.32 (d, *J* = 8.8 Hz, 2H), 6.90 (d, *J* = 8.8 Hz, 2H), 5.06 (s, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): 168.68, 160.55, 149.37, 139.26, 135.63, 133.74, 132.35, 131.72, 129.10, 128.87, 128.04, 123.55, 120.85, 119.87, 118.85, 112.26. HRMS (ESI) calcd for C<sub>18</sub>H<sub>11</sub>BrNO<sub>3</sub>S [M+H]<sup>+</sup>: 399.9638, found: 399.9639.

#### 4-phenylthio-3-amino-1,8-naphthalic anhydride (4b). <sup>S1</sup>

Stannous chloride dehydrate (2.26 g, 10 mmol) was stirred in 15 mL concentrated hydrochloric acid at room temperature for 10 mins, **3b** was added slowly and the reaction was then heated at 85 °C for 3 hours, the mixture was filtered and washed with water until the filtrate was neutral, and

residue was dried and purified on silica gel chromatography (PE/DCM = 1/1, v/v) to give out **4b**, yellow green solid, yield: 80%. mp 225.9-226.7 °C.<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>+CD<sub>3</sub>OD):  $\delta$  8.68 (d, J = 8.4 Hz, 1H), 8.36 (d, J = 7.2 Hz, 1H), 8.16 (s, 1H), 7.72 (t, J = 8.0 Hz, 1H), 7.22-7.15 (m, 3H), 7.03 (d, J = 7.6 Hz, 2H), 5.09 (s, 2H).<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>+CD<sub>3</sub>OD): 160.80, 160.74, 149.42, 135.73, 134.47, 132.01, 129.30, 128.94, 128.62, 126.55, 126.12, 124.37, 123.56, 120.40, 118.64, 113.15. MS (ESI) calcd for C<sub>18</sub>H<sub>11</sub>NO<sub>3</sub>S [M+H]<sup>+</sup>: 322.0, found: 322.0.

## 9-bromo-*N*-(2-(2-hydroxyethoxy)ethyl)-benzo[k,l]thioxanthene-3,4-naphthalimide (1a) and 9bromo-*N*-(2-(2-hydroxyethoxy)ethyl)-benzo[b]furano[2,1-c]naphthalimide (1b).

Sodium nitrite (0.38 g, 5.5 mmol) was added to 32 mL concentrated sulphuric acid and 12 mL acetic acid in three times at -5 °C, 4a(2.00 g, 5 mmol) was then added slowly within 45 mins, the mixture was stirred for another 1 hour to give out black viscous liquid. At the same time, anhydrous cupric sulfate (34.00 g, 213 mmol) was dissolved in 350 mL deionized water and 45 mL acetic acid, and the blue solution was heated under reflux. Then the black liquid was added dropwise to the blue solution, the mixture was sequentially warmed at 100 °C for another 6 hours, and filtered to give out the brown solid. The solid was then stirred in 10 mL anhydrous ethanol withdiglycolamine (0.22 g, 0.20 mL) at 50 °C for 3 hours. The mixture was evaporated in vacuum and residue was purified on silica gel chromatography (PE/DCM = 1/1, v/v and then DCM/MeOH = 100/1) to give out **1b**, light yellow solid, yield: 70% and 1a, orange solid, yield: 18%. mp (1a) 217.7-218.4 °C. <sup>1</sup>H NMR  $(500 \text{ MHz}, \text{CDCl}_3)$ :  $\delta$  8.47 (d, J = 8.0 Hz, 1H), 8.30 (d, J = 8.0 Hz, 1H), 8.23 (s, 1H), 8.05 (d, J = 8.0 Hz, 1H, 7.46 (d, J = 9.0 Hz, 1H), 7.43 (d, J = 8.0 Hz, 1H), 7.20 (d, J = 9.0 Hz, 1H), 4.37 (t, J = 9.0 Hz, 1H5.5 Hz, 2H), 3.79 (t, J = 5.5 Hz, 2H), 3.62 (t, J = 4.5 Hz, 2H), 3.58 (t, J = 4.5 Hz, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): 164.15, 163.69, 140.11, 135.53, 132.91, 132.71, 131.08, 130.51, 130.30, 129.65, 128.92, 127.70, 125.35, 121.46, 121.39, 120.69, 119.56, 117.90, 72.50, 68.43, 61.52, 39.80. HPLC purity: 95.3%, Retention time = 13.73 min. HRMS (EI) calcd for  $C_{22}H_{16}BrNO_4S$  [M]<sup>+</sup>: 468.9983, found: 468.9982. mp (**1b**) 201.3-201.7 °C. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.67 (s, 1H), 8.47 (d, J = 7.2 Hz, 1H), 8.14 (d, J = 8.0 Hz, 1H), 8.02 (d, J = 1.6 Hz, 1H), 7.70 (t, J = 8.0 Hz, 1H), 7.70 (t, J = 8.0 Hz, 1H), 8.14 (d, J = 8.07.64 (d, J = 8.4 Hz, 1H), 7.51 (dd,  $J_I = 1.6$  Hz,  $J_2 = 8.4$  Hz, 1H), 4.41 (t, J = 5.6 Hz, 2H), 3.91 (t, J= 5.6 Hz, 2H), 3.75~3.73 (m, 4H), 2.36 (brs, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): 164.12, 163.91, 143.92, 137.13, 137.10, 131.80, 131.02, 130.43, 129.75, 127.50, 126.62, 126.58, 124.81, 124.63, 124.04, 123.25, 119.70, 119.62, 72.38, 68.49, 61.90, 39.72. HPLC purity: 99.5%, Retention time = 14.34 min. HRMS (EI) calcd for C<sub>22</sub>H<sub>16</sub>BrNO<sub>4</sub>S [M]<sup>+</sup>: 468.9983, found: 468.9981.

N-(2-(2-hydroxyethoxy)ethyl)-benzo[k,l]thioxanthene-3,4-naphthalimide (1c) and N-(2-(2-hydroxyethoxy)ethyl)-benzo[b]furano[2,1-c]naphthalimide (1d). <sup>S1</sup>

Sodium nitrite (0.38 g, 5.5 mmol) was added to 32 mLconcentrated sulphuric acid and 12 mL acetic acid in three times at -5 °C,4b (1.60 g, 5 mmol) was then added slowly within 45 mins, the mixture was stirred for another 1 hour to give out black viscous liquid. At the same time, anhydrous cupric sulfate (34.00 g, 213 mmol) was dissolved in 350 mL deionized water and 45 mL acetic acid, and the blue solution was heated under reflux. Then the black liquid was added dropwise to the blue solution, the mixture was sequentially warmed at 100 °C for another 6 hours, and filtered to give out the brown solid. The solid was then stirred in 10 mL anhydrous ethanol withdiglycolamine (0.22 g, 0.20 mL) at 50 °C for 3 hours. The mixture was evaporated in vacuum and residue was purified on silica gel chromatography (PE/DCM = 1/1, v/v and then DCM/MeOH = 100/1) to give out 1d, light yellow solid, yield: 55% and 1c, orange solid, yield: 20%. mp (1c) 192.5-193.8 °C.<sup>1</sup>H NMR  $(500 \text{ MHz}, \text{CDCl}_3)$ :  $\delta$  8.41 (d, J = 8.0 Hz, 1H), 8.24 (d, J = 8.0 Hz, 1H), 8.02 (d, J = 7.5 Hz, 1H),  $7.94 (d, J = 8.0 Hz, 1H), 7.25 \sim 7.34 (m, 4H), 4.39 (t, J = 5.0 Hz, 2H), 3.87 (t, J = 5.0 Hz, 2H), 3.72$ 3.73 (m, 2H), 3.70-3.71 (m, 2H).<sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): 164.12, 163.69, 140.74, 136.77, 132.68, 131.62, 130.88, 130.32, 130.00, 127.84, 127.61, 126.40, 126.10, 125.34, 120.92, 120.35, 119.14, 117.76, 72.30, 68.49, 61.90, 39.58.HPLC purity: 98.3%, Retention time = 11.61 min. HRMS (EI) calcd for C<sub>22</sub>H<sub>17</sub>NO<sub>4</sub>S [M]<sup>+</sup>: 391.0878, found: 391.0875. mp (1d) 160.0-160.7°C.<sup>1</sup>H NMR (400 MHz, CDCl3):  $\delta$  8.89 (s, 1H), 8.45 (d, J = 7.2 Hz, 1H), 8.19 (d, J = 8.0 Hz, 1H), 8.09  $(dd, J_1 = 2.0 Hz, J_2 = 5.2 Hz, 1H), 7.84 (dd, J_1 = 2.8 Hz, J_2 = 5.2 Hz, 1H), 7.67 (t, J = 8.0 Hz, 1H),$  $7.48 \sim 7.50$  (m, 2H), 4.41 (t, J = 5.2 Hz, 2H), 3.91 (t, J = 5.2 Hz, 2H), 3.72 - 3.74 (m, 4H), 2.76 (brs, 1H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>): 162.78, 162.71, 142.03, 137.75, 134.69, 131.99, 129.75, 129.36, 127.37, 127.34, 125.64, 125.34, 125.23, 123.77, 122.82, 122.24, 122.01, 118.64, 72.08, 66.88, 60.21, 38.71. HPLC purity: 99.5%, Retention time = 12.17 min. HRMS (EI) calcd for C<sub>22</sub>H<sub>17</sub>NO<sub>4</sub>S[M]<sup>+</sup>: 391.0878, found: 391.0881.

#### 3. Methods

#### 3.1 Spectroscopic materials and methods

Spectroscopic properties were measured on All Varian Cary 100 UV spectrophotometer and Varian Cary Eclipse FL spectrophotometer and the singlet oxygen quantum yields were accomplished by monochromatic light system composed of CM 110 1/8mmonochromator and ASB-XE-175 Xenon light source from Spectral Products. Optical imaging was performed through fluorescence microscope and confocal microscopy (Nikon, Japan).

#### 3.2 Determination of fluorescent quantum yield

All compounds were dissolved in ethanol at room temperature. Absorbance spectra and relevant fluorescence spectra were collected by a Varian Cary 100 UV-Vis spectrophotometer and a Varian Cary Eclipse Fluorescence spectrophotometer. The molar extinction coefficients were obtained by

Lambert-Beer's Law and fluorescence quantum yields were calculated as below:

$$\Phi_F = \Phi_R \left(\frac{S}{S_R}\right) \left(\frac{\eta}{\eta_R}\right)_{(1)}$$

where  $\Phi$  is the fluorescence quantum yield, S is the slope of integrated fluorescence intensity vs. absorbance,  $\eta$  is the relatively refractive index of solvent, subscript <sub>R</sub> indicate the reference standard of 4-*N*-butylamide-*N*-butyl-1,8-naphthalimide, whose fluorescent quantum yield ( $\Phi_R$ ) was 0.81 in ethanol.

#### 3.3 Determination of singlet oxygen quantum yield S2

To quantify the irradiation-induced singlet oxygen ( ${}^{1}O_{2}$ ), a singlet oxygen trap 1,3-Diphenylisobenzofuran (DPBF) was used in air-saturated acetonitrile. Compounds were dissolved in acetonitrile and added to 100-fold excess DPBF stock solution. After laser irradiation with a monochromator (ASB-XE-175 Xenon light source), the absorbance of DPBF at 410 nm was recorded every half or one minute. Methylene Blue (whose singlet oxygen quantum yield was 0.52 in acetonitrile) was utilized as a contrast. The singlet oxygen quantum yields were determined using the equation:

$$\Phi_{\Delta(T)} = \Phi_{\Delta(MB)} \left( \frac{S_T}{S_{MB}} \right) \left( \frac{F_T}{F_{MB}} \right) (2)$$

Where  $\Phi_{\Delta}$  is the singlet oxygen quantum yield, S is the slope of absorbance of DBPF vs. irradiation time and F is the absorption correction factor.

#### 3.4 Cell culture

Human cell lines (MCF-7, HCT116, U87, MKN45, A375) were purchased from The Cell Bank of Type Culture Collection of the Chinese Academy of Sciences (Shanghai, China). Cells were cultured in RPMI-1640 or DMEM medium, supplemented with 10% fetal bovine serum (FBS, Hyclone, USA) and 1% antibiotic- antimycotic (ABAM Life Technologies, California, USA) in a humidified atmosphere of 95% air and 5% CO<sub>2</sub>.

#### 3.5 Cell viability assay

To evaluate the tumor inhibitory efficacy of **1a-1d**, MCF-7 cells  $(1 \times 10^4)$  were suspended in medium and plated in 96 well plates. After treatment with indicated concentrations  $(0-20 \ \mu\text{M})$  of **1a-1d** for 48 hours, cells were irradiated with the 100-watt quartz-halogen lamp. The fluorescence rate was 240 W/m<sup>2</sup>, measured by a photo radiometer (Delta Ohm, Padua, Italy). After 20 mins of PDT treatment and 4 hours of MTT incubation, the cell viability was determined by MTT assay.<sup>S3</sup>As the optimal candidate, **1a** was then solely evaluated in other cell lines including HCT116, U87,

MKN45and A375 cells by the same methods above.

#### 3.6 In vitro imaging of 1a

#### 3.6.1 Fluorescence microscopic imaging

With a aim to assess fluorescent imaging results *in vitro*, A375 cells  $(1 \times 10^5)$  were treated with **1a** (2.5  $\mu$ M, 2 mL) for 48 h. After that, cells were exposed to the light of 28.8 J/cm<sup>2</sup> within 20 mins for PDT. 4 hours later, the cells were rinsed with PBS for three times. Images were exposed by excitation at 460 nm and emission at 520 nm under an inverted Fluorescence microscope (Nikon, Japan).

#### 3.6.2 Subcellular localization studies

A375 cells ( $1 \times 10^5$ ) were seeded on a coverslip and co-incubated with **1a** (2.5 µM, 2 mL) for 48 hours. After 20 mins of PDT treatment, the cells were rinsed with PBS for three times and incubated in culture medium with 100 nMMitoTracker Red (invitrogen, USA) for 20 mins, 100 nM ER-Tracker Red for 20 mins (Beyotime, China), 50 nM Lysosome Tracker Red (Beyotime, China) for 30 mins, 1 µM DAPI (Beyotime, China) for 15 mins, respectively. The stained live cells were observed by confocal microscopy (Nikon, Japan).

#### 3.7 In vivo experiments

All six-weeks old nude mice with half males and half females were purchased from Shanghai Slac Laboratory Animal Co. Ltd (The license of experimental animal: SCXK(shanghai) 2012-0002). The mice were maintained under the 12 hour light/dark cycles condition in the specific pathogen free (SPF) cleanroom with fresh air (more than 50%), appropriate temperature (22-26 oC) and humidness (40%-60%). The mice were freely fed with enough sterilized water and food bought from Shanghai Slac Laboratory Animal Co. Ltd. All materials and containers were disinfected and sterilizated before use. After the treatment, the mice were anesthetized and euthanasia in accordance with the international ethical standards. All experimental procedures involving animals in this study were reviewed and approved by the institution of ethics committee in ECUST and strictly conducted according to the guidelines of Care and Use of Laboratory Animals of China (GB14925-2001) for animal experimentation.

Each nude mouse was anesthetized with pelltobarbitalum natricum and subcutaneously inoculated with injection of  $1 \times 10^{6}$ A375 cells. When the volume of the tumor grew to 100 mm<sup>3</sup> size, mice were randomizedinto three groups: control group, medicine group and PDT group (each group had 6 mice). **1a** (5 mg/kg) was injected into both medicine group and PDT group of mice in situ while saline in the control group. After 8 hours, 600-micron bare fiber was used to give PDT treatment by white light from ASB-XE-175 Xenon light sourceand it was totally irradiated with a light-dose of 259.2 J/cm<sup>2</sup> within 60 min. All groups were administered the injection twice a week,

total 6 times during the treatment time. Tumor sizes were measured every 3 days using micrometer calipers. 4 hours after the laser treatment, tumor xenografts were removed and measured. During the experiment, tumor sizes were measured every 3 days using micrometer calipers and tumor volume (TV) was calculated using the following formula: TV (mm<sup>3</sup>) = D × d<sup>2</sup>/2, D is the shortest diameters while d is the longest one. Relative tumor volume (RTV) was calculated according to the equation: RTV = V<sub>t</sub>/V<sub>0</sub>, where V<sub>0</sub> is the tumor volume at day 0 and V<sub>t</sub> is the tumor volume at day t.

#### 3.8 H&E staining

Hematoxylin-eosin staining was a conventional method for observing the histological changes of the tissues. After indicated experiments, all of the tumor tissues and several visceral organs tissues were collected and embedded in paraffin. The slices were separately treated with hematoxylin and eosin solution and images were taken with fluorescence microscope (Nikon, Japan).

#### 3.9 Mechanistic study

#### 3.9.1 Measurement of oxidative damage

DCFH-DA method(2',7'-dichlorofluorescein diacetate) was employed to evaluate intracellular reactive oxygen species (ROS) level.<sup>S4</sup>After 20 mins PDT treatment, cells were collected in serum-free medium and co-incubated with 10  $\mu$ M DCFH-DA at 37 °C for another 20 mins. Cells were then washed three times with PBS and ROS generation was measured using fluorescence intensity of excitation at 488 nm and emission at 525 nm by EnSpire® Multimode Plate Reader (Perkin Elmer, Boston, MA).

#### 3.9.2 Hoechst 33342 staining

A375 cells  $(1 \times 10^5)$  were seeded on glass bottom dishes, cultured and treated with **1a** as described above. The cells were stained with Hoechst 33342 (Sigma, USA) for 20 mins at 37 °C in the dark condition after rinsed with PBS for twice. The images were exposed by the fluorescence microscope (Nikon, Japan).

#### 3.9.3 Western blot analysis

After subjected to the indicated treatment, cells were harvested and lysed with cell lysis solution. Total proteins were quantified using the BCA protein assay kit. Equal amount of protein was run on 10% sodium dodecylsulfate (SDS) polyacrylamide gels and then transferred to PVDF membranes. The membranes were blocked in TBS-Tween 20 solution with non-fat dry milk prior to incubated overnight with primary antibodies PARP (SAB, USA), cleaved-PARP (SAB, USA), Caspase 3 (GenTex, USA) and cleaved-Caspase 3 antibodies (GenTex, USA) (1 mg/mL for each antibody). Following three washes with TBS-T, membranes were incubated with HRP conjugated secondary antibody for 2 h at room temperature. Then membranes were washed by TBS-T three times again,

developed in ECL and visualized with X-ray film. $\beta$ -actin (GenTex, USA) was used as an internal control.

#### 3.9.4 Immunohistochemistry

The expression of Caspase 3 in tumor tissues of nude mice model was assessed by immunohistochemistry assay. The samples were embedded in paraffin and stained by Caspase 3 antibodies (Bioworld, China). <sup>S5</sup> Images were then taken with a fluorescence microscope (Nikon, Japan).

#### 3.9.5 Tunel assay

Tumor apoptosis was performed following the manufacturer's instructions of terminal deoxynucleotidyl transferase Dutp nick end labeling (TUNEL) assay (Roche Applied Science, Germany). The slides were photographed with a fluorescence microscope (Nikon, Japan).

#### 3.10 Statistical analysis

All results shown in the study were obtained in at least three independent experiments and were expressed as mean  $\pm$  standard deviation (SD). Statistical analysis of data was performed using Excel (Microsoft) and GraphPad Prism 5.0 software (GraphPad Software Inc, CA, USA). Significant differences are considered as \*p< 0.05; \*\* p< 0.01; \*\*\* p< 0.001.

4. Data



Fig. S1The UV absorption spectra (blue line) and fluorescent emission spectra (red line) of 1a (25



 $\mu$ M, slit: 2.5, 2.5), **1b**(25  $\mu$ M, slit: 5, 5), **1c** (25  $\mu$ M, slit: 2.5, 2.5) and **1d** (25  $\mu$ M, slit: 5, 5) in EtOH.

Fig. S2 a) MTT assays of 1a-1d exposed with laser irradiation of 0 and 28.8 J/cm<sup>2</sup> in MCF-7 cells;
b) MTT assays of 1a exposed with laser irradiation of 0 and 28.8 J/cm<sup>2</sup> light in A375, HCT116, MKN45 and MCF-7 cells.



**Fig. S3** Fluorescence confocal microscopy imaging of **1a** in A375 cells.Co-localizaiton with Mito-Tracker Red, ER-Tracker Red and Lysosome Tracker Red: a) white light; b) **1a**; c) organelles tracker probes; d) blue DAPI; e) the merge results of b and c; f) the merge results of b and d; g) the merge results of b, c and d. Bars: 10 μm



**Fig. S4** Pathological changes of major organs (including kidney, liver, spleen, heart and lung) and tumor tissues by H&E stained images. Bars: 50 μm.



**Fig. S5** Immunohistochemical assessment of cleaved-Caspase 3 expression of tumor cells after 1ainduced PDT treatment. Scale bars: 50 μm;



**Fig. S6** TUNEL assays of apoptotic cells in the tumor tissues. The nuclei stained with DAPI (bright blue) and apoptotic cells (bright red) were observed using fluorescent microscopy. Bars: 50 μm

#### 5. NMR spectra



Fig. S7The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrum of 1a

#### **Elemental Composition Report**

# Multiple Mass Analysis: 53 mass(es) processed Tolerance = 5.0 mDa / DBE: min = -1.5, max = 60.0 Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions 6283 formula(e) evaluated with 255 results within fimits (all results (up to 1000) for each mass) Elements Used: C: 0-24 H: 0-18 N: 0-1 O; 0-5 \$: 0-1 798r 0-1 818r; 0-1

RLZ-2L-510-071

Waters GCT Premier

20160095 448	(7.467) Cm (4	48-(70+659))		118010 001	P Petriler		TOF MS EI+
100-							382,9445
04							
		· ·			202	0767	383.9495
1				222 0245 2	ουφ. 57.0299	365.944	0 407 9664
71.06	85.102311	3.1343 141.1860	187.0552	202.0040 7		337.9466	468.9982
40 66	a 80 100	120 140 160	180 200	220 240 26	4) 280 30	Hs <del>pt,tinnHatter</del> I0 329 340 36	0 389 400 420 440 460
Minimum: Maximum:	3.00		5.0	10.0	-1.5		
			0.0				
Mass	RA	Calc. Mass	mE a	PEN	DBE	£-ĕIT	Formila
57.0708	3.52	5710704	0.4	7.0	0.5	2773012.9	C4 H9
71.0960	4.59	71.0861	-0.1	-1.4	0.5	654.3	C5 W11
85.1023	3.05	65.10.7	0.0	1.1	0.5	613.3	CO H1.3
99.11/5	3.20	99.11(4	0.1	1.0	0.3	2773016.3	C7 HIS
107 1743	2.30	1.13-1330	See 1.	11.5	015	57.5	ve niz
141 1666	3 24						
155 1798	3.00						
182.9087	3.18	182.9664	0.0	1.6	5.0	156.5	C7 86 N 79Br
	10000000	192.9666	2.1	11.5	0.5	213.8	C5 810 S 815r
		1.92.9657	3.0	1.6.4	0.5	172.7	C4 ¥8 C3 79Br
		182,9717	-3.0	-16.4	0.0	159.3	C4 810 N 9 790r
		182.9718	-3.1	-16.9	1.0	202.0	C4 H8 N C2 813r
187.0552	7.21	187.6548	Ū.d	2.1	12.5	349.4	C1.5 H7
		187.0581	-2.9	-15.3	7.5	130.6	C12 H11 3
		187.0514	3.8	20.3	-1.0	205.6	C4 H13 & O5 S
214.0659	3.37	214.0657	0.2	5.9	1.3.5	2773034.5	C15 HU 3
		214.0664	-0.5	-2.3	4.0	277317E.9	C10 814 03 5
		214.0630	2.9	1.3.5	9-0	2773036.0	
222 0121	5 8 2	214.0690	-3.1	-16.0	3.5	3773165.3	UIS HES N & 70%5
200.0151	5.00	230.0.01	-0.0	-2 6	14 0	t 255 2	C3 H13 H 6 7551
		230.0163	1 3	7.9	-1 0	1000.2	C7 H17 O S B13c
		230 0157	2 4	30.4	19.0	1355.5	C19 112
		230.0154	2.7	11.7	-1.0	589.1	C6 H15 O4 79Br
		230.0214	.3.3	-14.3	-1.5	841.2	C6 H17 N 0 5 79Br
		230.0215	-3.4	-14.3	10.9	1369.3	C12 #6 05
		230.9215	~3.4	-14.8	-0.5	1411.4	C6 H15 N C3 81Er
231.0253	7.37	231.0239	-0.6	-2.6	3.0	939.2	C9 H14 N O 79Br
		231.0241	1.2	5.2	~1.5	997.2	C7 H18 O S B1Br
		231.0268	-1.5	-6.5	13.5	861.7	C16 H7 S
		231.0235	1.3	7.9	18.5	862 - 1	C19 H3
		231.0232	2.1	9.1	-1,5	969.9	C6 M16 C4 79Br
		231.0293	-4.0	-17.3	-1.0	1049.8	C6 £15 N 03 81Br
		231.0293	-4.0	~17.3	9.5	965.9	CI2 #7 05
000.0045	11.07	231.0208	4.5	19,5	3.3	1.017-0	CIV 514 O PIEF CIE 60 9
«34.0345	11.80	232,0347	0.0	2 4	23.6	23.2 G48 7	CLC 20 20 C0 215 X O 7985
		232 A171	-2 5	-11 2	-1 5	176 3	C6 V:7 % 03 81Br
		232.0311	-2.7	-11.6	5.0	90.8	C12 11A C5
		232.0313	3.2	13.8	18.0	39.3	C19 H4
233.0357	3.28	233.0358	-0.1	-0.4	4.0	2773199.8	C8 H11 N D5 3
200.0001	2.00	~~~~~~~~	9.1				

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Fig. S8The HR-MS spectrum of 1a

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Fig. S9The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrum of 1b

olerance =	5.0 mDa	/ DBE: min = -1	.5, max = 50	0.0				
	Mass Odd	and Even Electron	long					
707 formula	a(e) evaluate	d with 202 results	within limits (a	all results (u	p to 1000) for	each mass)		
: 0-22 H:	0-16 N: 0-	1 0:0-4 S:0-1	79Br: 0-1	81Br: 0-1				
LZ-ZL-510-0	41			Waters GC	F Premier			
0160115 427	(7.117) Cm (4	427-(87+93))						1.74e+00
C00							382.9453	
-								
%-							383.9507	
1						227 0476 00	407.9667	
70	0204 115	0000 155 9711	232.0 87.0543	341 257.02	96 303.0345	357.9410 36	5.9424 410.	9739 468.9981
0 60	80 100	120 140 160	180 200 22	0 240 26	0 280 300	320 340 36	50 380 400 420	m/. 440 460
inimum:	3.00				-1.5			
aximum:	100.00		5.0	10.0	50.0			
ass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula	
81.9706	3.88	181.9701	0.5	2.7	11.5	346.1	C10 N O S C8 H7 798r	
82 9697	4 70	181.9674	3.2	17.6	7.0	369.2	C7 H2 O4 S	Br
02.9091	4.70	182.9717	-2.0	-10.9	0.0	278.4	C4 H10 N S	79Br
		182.9/18	-2.1	-11.5	0.5	249.6	C4 H8 N 02 C5 H10 S 8	lBr
87.0543	9.25	182.9657 187.0548	4.0 -0.5	21.9 -2.7	0.5 12.5	293.4 72.1	C4 H8 O3 7 C15 H7	'9Br
30.0202	4.93	187.0581 230.0190	-3.8 1.2	-20.3 5.2	7.5 14.0	75.9 1788.5	C12 H11 S C16 H6 S	
		230.0215 230.0181	-1.3 2.1	-5.7 9.1	-0.5 3.5	1913.8 1055.2	C6 H15 N C C9 H13 N C	03 81Br 0 79Br
		230.0242	-4.0	-17.4	14.5	1812.5	C15 H4 N C C19 H2	02
31 0257	8 47	230.0154	4.8	20.9	-1.0	1113.0	C6 H15 O4	79Br 79Br
51.0257	0.47	231.0268	-1.1	-4.8	13.5	1144.6	C16 H7 S	
		231.0232	2.5	10.8	-1.5	1401.8	C6 H16 O4	79Br
		231.0293	4.9	21.2	3.5	1393.7	C10 H14 O	81Br
32.0341	14.27	232.0337 232.0347	-0.6	-2.6	2.5	6.0	C9 H15 N C C16 H8 S	/9Br
33.0358	3.15	232.0313 233.0364	2.8	12.1	18.0 2.5	37.9 2773048.3	C19 H4 C10 H16 O	81Br
57.0296	15.96	233.0391 257.0299	-3.3	-14.2	17.5 15.0	2773071.3 441.9	C19 H5 C17 H7 N S	,
		257.0272 257.0265	2.4	9.3 12.1	10.5 20.0	502.6 513.4	C14 H9 O3 C20 H3 N	S
58.0358	6.52	257.0263	3.3	12.8	0.0	893.6 417.8	C7 H16 N C C14 H10 O3	04 79Br S
	0.02	258.0344	1.4	5.4	19.5	403.2	C20 H4 N	
		258.0317	4.1	15.9	15.0	439.9	C17 H6 O3	0 918-
59.0420	5.52	259.0422	-0.2	-0.8	19.0	61.0	C20 H5 N	O OIDI
		259.0429 259.0395	-0.9	-3.5	9.5 14.5	77.2	C14 H11 03 C17 H7 03	5
		259.0395 259.0456	2.5 -3.6	9.7 -13.9	4.0 14.0	126.4 50.4	C11 H16 N C17 H9 N S	O 81Br
03.0345	7.09	303.0354	-0.9	-3.0	15.0	34.1	C18 H9 N C	02 S .
							Lingung by the	而後後在多期世
							10 01 01 10 054	何证明作用,

Fig. S10The HR-MS spectrum of 1b



Fig. S11The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrum of 1c

#### **Elemental Composition Report**

Multiple Mass Analysis: 22 mass(es) processed Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0 Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions 552 formula(e) evaluated with 43 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-22 H: 0-17 N: 0-1 O: 0-4 S: 0-1 RLZ-ZL-60-023b GCT Premier 20141038 614 (10.234) Cm (611-(5+626))

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20141038 614	(10.234) Cm	(614-(5+826))			0.0010			TOF MS EI+
							• • •	1.71e+004
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73.02	84	112 0154 Lun ou	50 5970	£40 232	2.0343 25	8.0383 286.0326	330.05/9	391.0875
0-111-11	87.0240	140.0104 140.01	no reco		the man	<del></del>		zım miz
60	50 10	0 120 140	160 180	200 220	240	260 280 300	320 340 360	380
Minimum	3.00				-1 5			
Maximum	102.00		5.0	19.0	50.0			
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			~. ~	20.3	~~.~			
Mass	RA	Calc. Mass	πCa	*PM	DBE	1-211	Formula	
115.0104	3.46	116.0170	~1.6	13.8	2.3	2773175.0	C4 H6 M O S	
		116.0135	1.5	10.0	3.0	2773019-0		
187.0549	3.76	187.0548	0.1	0.5	12.5	314.4	C15 M7	
		187.0581	-3.2	-17.1	7.5	303.7	C12 811 S	
216.0656	3.12	219.0057	-0.1	-0.5	13.5	4.8	C16 88 N	
		214.0664	-0.0	-3.7	4.0	13.3	C10 E14 O3 S	
		214.0630	2.6	12.1	9.0	9.1	C13 K10 O3	
		214.0690	-2.4	-15.9	8.5	5.6	C13 H12 N S	
232-0343	9.87	232.0347	-0.4	-1./	13.0	161.1	CID RE S	
233 0405	1 00	232.0313	3.0	£6.2 £ 0	17 5	1 5	CL9 D4 C10 D5	
210.0400	1.02	233.0425	-2.0	-8.6	12.5	15.8	C16 F9 8	
246.0383	3.16	246.0377	0.6	2.4	13.5	2773211.5	C16 H8 N S	
		246.0351	3.2	13.0	9.0	2773217.5	C13 M10 D3 3	
		246.0344	3.9	15.9	18.5	2773062,6	C13 E4 N	
258.0383	9.81	258.0377	0.6	2.3	14.5	429.6	C17 EE N S	
		258,0351	3.2	15 1	-9.0	475.3	CLA ELU OS S	
269 0448	5.52	259.0436	-0.8	-3.1	19.0	289 0	C17 29 N 9	
		259.0423	1.9	7.3	9.5	315.9	C14 E11 03 5	
		259.0422	2.5	10.0	19.0	306.7	C20 1:5 N	
268.0371	4.21	260.0391	-1.0	-3.a	9.5	67.0	C13 H10 N 03	S
100000000000000000000000000000000000000	212 223	260.0348	2.3	0.0	14.5	50.4	C16 E6 N O3	
296.0326	10.1€	286.0327	-G.1	-0.3	15.5	73.3	CIB FR N O S	
		200.0300	2.6	3. 5	20.5	84.0	CI5 HIU 04 3	
303 0340	100.00	303 (1354	-1.6	-4 6	15 0	113.8	CIR FA N O2	5
200.000	100,00	303.0323	2.0	6.6	20.0	285.3	C21 E5 N C2	~
304.0407	27.74	304.0399	0.8	2.6	19.5	43.8	C21 H6 N 02	
		304.0432	-2.5	-8.2	14.5	50.2	C18 H10 N 02	з
305.0390	7.97	305.0425	~3.5	-11.5	18.5	85.9	C22 19 8	
316.0435	7.31	316.0432	0.3	0.9	15.5	308.8	C19 H10 N 02	3
217 0214	6 02	316.0399	.5.0	11	20.5	326.6	CL2 E8 N O2	÷
911.0014	0.07	317 0477	4.3	1 ?	20.0	8 1	C22 FT N 02	<i></i>
328,0451	3.22	328.0432	1.9	5.8	16.5	1336.2	C20 H10 N 02	3
329.0514	3.22	329.0511	0.3	0.9	16.0	1704.8	C20 B11 N 02	3
330.0579	14.27	330.0539	-1.0	-3.0	15.5	226.7	C20 E12 N 02	S
331.0650	7.83	331-0667	-1.7	-5.1	15.0	1.3.5	C20 E13 N 02	3
347.0599	3.87	347.0616	-1.7	-4.9	15.0	1.2.0	C20 M13 N 03	8
373.0806	3-41	3/3.0//3	3.3	8.8	10.0	10.4	C22 HID N DJ	2
351,0879	5.76	291,08(9	-0.3	-0.0	10.0	0.4	CZZ HIF N 94	2

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Fig. S12The HR-MS spectrum of 1c

Page 1



Fig. S13The <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrum of 1d

C: 0-22 H RLZ-ZL-510-0 20160116 622	1: 0-17		in limits (a	ll results (up	to 1000) for	r each mass)			
	2 (10.374)	N: 0-1 O: 0-4 S: 0 Cm (622-54:67)	)-1	Waters GC	T Premier				TOF MS E
100						303.0	355		9.18e+0
-							304.0421		
%-]							330.	0585	
}						286.0317	305.0413		
	130.0168	143.0154 189.0698	214.0663	233.0408	258.0379	2.0426		347.0599	391.088
100	120	140 160 180	200	220 240	260	280 300	320	340 360	380
4inimum:	3.00				-1.5				
Maximum:	100.0	0	5.0	10.0	50.0				
lass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	For	nula	
43.0154	3.78	143.0167	-1.3	-9.1	3.5	277309	7.0 C6	H7 02 S	
214.0663	3.13	214.0664	-0.1	-0.5	4.0	277309	L.8 C10	H14 03 S	
		214.0657 214.0690	0.6	2.8	13.5 8.5	277302	6.3 C13	H8 N H12 N S	
33 0409	5 09	214.0630	3.3	15.4	9.0	277302	1.8 C13 C19	H10 O3 H5	
	0.00	233.0425	-1.7	-7.3	12.5	4.5	C16	H9 S	
258.0379	11.31	258.0377	2.8	10.9	14.5	212.5	C14	H10 03 S	
59.0448	7.16	258.0344	3.5	13.6	19.5	233.6	C20 C17	H4 N H9 N S	
		259.0429	1.9	7.3	9.5	233.9	C14	H11 03 S	
261.0396	5.08	261.0374	2.0	8.4	13.5	5.8	C17	H9 O S	
286.0317	25.43	261.0426 286.0327	-3.0	-11.5	14.0 15.5	2.0 28.5	C16 C18	H7 N O3 H8 N O	s
		286.0300	1.7	5.9	11.0	11.1	C15 C21	H10 04 S H4 N O	
287.0367	3.92	287.0371	-0.4	-1.4	20.0	50.3	C21	H5 N O	
		287.0378 287.0344	-1.1 2.3	-3.8 8.0	10.5	54.2 59.1	C15 C18	HT 04 S H7 04	
303.0355	100 0	287.0405	-3.8	-13.2	15.0	45.3	C18 C18	H9 N O H9 N O2	s s
		303.0320	3.5	11.5	20.0	1479.5	C21	H5 N 02	c
304.0421	64.47	304.0432	2.2	7.2	19.5	51.5	. C21	H6 N 02	5
305.0413	14.80	305.0425 306.0377	-1.2 3.5	-3.9 11.4	18.5	32.9 277313	C22 2.0 C21	H9 S H8 N S	
316.0435	13.74	316.0432	0.3	0.9	15.5	554664	1.5 C19	H10 N O2	S
	36.99	330.0589	-0.4	-1.2	15.5	6.6	C20	H12 N 02	S
330.0585	7.83	331.0667 346.0538	-2.5	-7.6	15.0	7.4 444.3	C20 C20	H13 N 02 H12 N 03	S
330.0585 331.0642 346.0541	3.26	347 0616	-1.7	-4.9	15.0	15.8	C20	H13 N 03	S
330.0585 331.0642 346.0541 347.0599	3.26	347.0616	-4.1	-11.8	14.5	0.9	C20	H15 N 03	S
330.0585 331.0642 346.0541 347.0599 348.0653 361.0777	3.26 8.70 3.18 4.86	348.0694 361.0773	0.4						
330.0585 331.0642 346.0541 347.0599 348.0653 361.0777 373.0772	3.26 8.70 3.18 4.86 3.35	347.0616 348.0694 361.0773 373.0773 391.0878	0.4 -0.1 0.3	-0.3	16.0	19.2 0.7	C22 C22	H15 N O3 H17 N O4	S S

Fig. S14The HR-MS spectrum of 1d

Data File C:\HPCHEM\1\DATA\ZL\SEP00004.D

Sample Name: ZL09-1-071

RX C-18 4.6mm\*250 Flow=1ml/min 132bar buffer



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Page 1 of 1

Fig. S15The HPLC report of 1a

Data File C:\HPCHEM\1\DATA\ZL\SEP00002.D

RX C-18 4.6mm\*250 Flow=1ml/min 132bar buffer



Fig. S16The HPLC report of 1b

Data File C:\HPCHEM\1\DATA\ZL\MAR00098.D

XDB 4.6mm\*250 Flow=1.0ml/min 124bar (buffer)



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Page 1 of 1

Fig. S17The HPLC report of 1c

Data File C:\HPCHEM\1\DATA\ZL\SEP00001.D

RX C-18 4.6mm\*250 Flow=1ml/min 132bar buffer



Instrument 1 9/1/2015 3:41:20 PM

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Fig. S18The HPLC report of 1d

#### 6. References

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