

A family of Uranium-carboxylic acid hybrid materials: synthesis, structure and mixed-dyes selective adsorption

*Xue Gao,^a Jian Song,^a Li Xian Sun,^b Yong Heng Xing,^a Feng Ying Bai^{*a} and Zhan Shi^c*

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Table S2 Hydrogen bond lengths (Å) and angles (°) for complexes **1-4***

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Complex 1			
U(1)-O(1)	1.755(3)	U(1)-O(2)	1.770(3)
U(1)-O(5)	2.355(2)	U(1)-O(6)	2.363(2)
U(1)-O(7) ^{#1}	2.364(3)	U(1)-O(3)	2.393(2)
U(1)-O(4)	2.408(3)		
O(1)-U(1)-O(2)	179.15(13)	O(1)-U(1)-O(5)	87.95(11)
O(2)-U(1)-O(5)	91.67(11)	O(1)-U(1)-O(6)	95.35(10)
O(2)-U(1)-O(6)	85.21(10)	O(5)-U(1)-O(6)	67.76(8)
O(1)-U(1)-O(7) ^{#1}	86.64(11)	O(2)-U(1)-O(7) ^{#1}	94.12(11)
O(5)-U(1)-O(7) ^{#1}	142.52(9)	O(6)-U(1)-O(7) ^{#1}	75.87(9)
O(1)-U(1)-O(3)	93.35(11)	O(2)-U(1)-O(3)	86.49(11)
O(5)-U(1)-O(3)	142.83(9)	O(6)-U(1)-O(3)	148.56(9)
O(7) ^{#1} -U(1)-O(3)	74.55(9)	O(1)-U(1)-O(4)	89.41(12)
O(2)-U(1)-O(4)	89.75(12)	O(5)-U(1)-O(4)	71.24(9)
O(6)-U(1)-O(4)	138.49(9)	O(7) ^{#1} -U(1)-O(4)	145.64(9)
O(3)-U(1)-O(4)	71.63(9)		
Complex 2			
U(1)-O(1)	1.775(3)	U(1)-O(1) ^{#1}	1.775(3)
U(1)-O(2)	2.452(3)	U(1)-O(2) ^{#1}	2.452(2)
U(1)-O(4)	2.472(3)	U(1)-O(4) ^{#1}	2.472(3)
U(1)-O(3)	2.503(3)	U(1)-O(3) ^{#1}	2.503(3)
O(1)-U(1)-O(1) ^{#1}	179.999(1)	O(1)-U(1)-O(2)	92.80(13)
O(1)-U(1)-O(2) ^{#1}	87.20(13)	O(1) ^{#1} -U(1)-O(2)	87.20(13)
O(1) ^{#1} -U(1)-O(2) ^{#1}	92.80(13)	O(2)-U(1)-O(2) ^{#1}	180.0
O(1) ^{#1} -U(1)-O(4) ^{#1}	93.60(13)	O(1)-U(1)-O(4) ^{#1}	86.40(13)
O(2)-U(1)-O(4) ^{#1}	62.80(10)	O(2) ^{#1} -U(1)-O(4) ^{#1}	117.20(10)
O(1) ^{#1} -U(1)-O(4)	86.40(13)	O(1)-U(1)-O(4)	93.60(13)
O(2)-U(1)-O(4)	117.20(10)	O(2) ^{#1} -U(1)-O(4)	62.80(10)
O(4)-U(1)-O(4) ^{#1}	180.0	O(1) ^{#1} -U(1)-O(3) ^{#1}	87.77(13)
O(1)-U(1)-O(3) ^{#1}	92.23(13)	O(2)-U(1)-O(3) ^{#1}	114.15(10)

O(2) ^{#1} -U(1)-O(3) ^{#1}	65.85(10)	O(4) ^{#1} -U(1)-O(3) ^{#1}	52.12(10)
O(4)-U(1)-O(3) ^{#1}	127.88(10)	O(1) ^{#1} -U(1)-O(3)	92.23(13)
O(1)-U(1)-O(3)	87.76(13)	O(2)-U(1)-O(3)	65.85(10)
O(2) ^{#1} -U(1)-O(3)	114.15(10)	O(4) ^{#1} -U(1)-O(3)	127.88(10)
O(4)-U(1)-O(3)	52.12(10)	O(3)-U(1)-O(3) ^{#1}	180.0

Complex 3

U(1)-O(1)	1.755(4)	U(1)-O(2)	1.763(4)
U(1)-O(8)	2.392(5)	U(1)-O(3)	2.436(5)
U(1)-O(4)	2.464(4)	U(1)-O(5)	2.477(4)
U(1)-O(7)	2.493(5)	U(1)-O(6)	2.494(5)
O(1)-U(1)-O(2)	178.3(2)	O(1)-U(1)-O(8)	92.4(2)
O(2)-U(1)-O(8)	88.7(2)	O(1)-U(1)-O(3)	90.8(2)
O(2)-U(1)-O(3)	90.9(2)	O(8)-U(1)-O(3)	67.9(2)
O(1)-U(1)-O(4)	91.5(2)	O(2)-U(1)-O(4)	89.1(2)
O(8)-U(1)-O(4)	121.2(2)	O(3)-U(1)-O(4)	53.4(2)
O(1)-U(1)-O(5)	90.4(2)	O(2)-U(1)-O(5)	88.4(2)
O(8)-U(1)-O(5)	172.4(2)	O(3)-U(1)-O(5)	119.2(2)
O(4)-U(1)-O(5)	65.8(2)	O(1)-U(1)-O(7)	89.5(2)
O(2)-U(1)-O(7)	90.1(2)	O(8)-U(1)-O(7)	52.8(2)
O(3)-U(1)-O(7)	120.7(2)	O(4)-U(1)-O(7)	174.0(2)
O(5)-U(1)-O(7)	120.1(2)	O(1)-U(1)-O(6)	88.2(2)
O(2)-U(1)-O(6)	90.1(2)	O(8)-U(1)-O(6)	121.0(2)
O(3)-U(1)-O(6)	171.1(2)	O(4)-U(1)-O(6)	117.8(2)
O(5)-U(1)-O(6)	52.0(1)	O(7)-U(1)-O(6)	68.2(2)

Complex 4

U(1)-O(1)	1.768(5)	U(1)-O(2)	1.772(5)
U(1)-O(6)	2.289(5)	U(1)-O(5)	2.306(5)
U(1)-O(8)	2.460(5)	U(1)-O(12)	2.482(5)
U(1)-O(7)	2.487(5)	U(2)-O(4)	1.762(5)
U(2)-O(3)	1.769(5)	U(2)-O(5) ^{#1}	2.296(5)
U(2)-O(6) ^{#2}	2.308(5)	U(2)-O(9)	2.468(5)

U(2)-O(11)	2.479(5)	U(2)-O(10)	2.487(5)
O(1)-U(1)-O(2)	178.1(2)	O(1)-U(1)-O(6)	86.9(2)
O(2)-U(1)-O(6)	92.9(2)	O(1)-U(1)-O(5)	94.9(2)
O(2)-U(1)-O(5)	87.0(2)	O(6)-U(1)-O(5)	83.50(17)
O(1)-U(1)-O(8)	89.9(2)	O(2)-U(1)-O(8)	89.1(2)
O(6)-U(1)-O(8)	142.53(17)	O(5)-U(1)-O(8)	133.96(16)
O(1)-U(1)-O(12)	94.6(2)	O(2)-U(1)-O(12)	83.6(2)
O(6)-U(1)-O(12)	74.64(17)	O(5)-U(1)-O(12)	155.62(17)
O(8)-U(1)-O(12)	68.43(16)	O(1)-U(1)-O(7)	86.9(2)
O(2)-U(1)-O(7)	94.1(2)	O(6)-U(1)-O(7)	152.03(17)
O(5)-U(1)-O(7)	69.91(16)	O(8)-U(1)-O(7)	64.66(15)
O(12)-U(1)-O(7)	133.06(16)	O(4)-U(2)-O(3)	179.7(2)
O(4)-U(2)-O(5) ^{#1}	89.1(2)	O(3)-U(2)-O(5) ^{#1}	90.7(2)
O(4)-U(2)-O(6) ^{#2}	90.5(2)	O(3)-U(2)-O(6) ^{#2}	89.7(2)
O(5) ^{#1} -U(2)-O(6) ^{#2}	81.39(17)	O(4)-U(2)-O(9)	87.7(2)
O(3)-U(2)-O(9)	92.6(2)	O(5) ^{#1} -U(2)-O(9)	154.44(16)
O(6) ^{#2} -U(2)-O(9)	73.29(17)	O(4)-U(2)-O(11)	85.9(2)
O(3)-U(2)-O(11)	93.8(2)	O(5) ^{#1} -U(2)-O(11)	73.09(16)
O(6) ^{#2} -U(2)-O(11)	154.27(17)	O(9)-U(2)-O(11)	131.86(16)
O(4)-U(2)-O(10)	91.3(2)	O(3)-U(2)-O(10)	88.8(2)
O(5) ^{#1} -U(2)-O(10)	140.18(16)	O(6) ^{#2} -U(2)-O(10)	138.41(17)
O(9)-U(2)-O(10)	65.28(15)	O(11)-U(2)-O(10)	67.22(16)

*Symmetry transformations used to generate equivalent atoms: complex 1: #1: -x+2, -y, -z; complex 2: #1: -x+1, -y+1, -z+1; complex 4: #1: x+1, -y+1/2, z+1/2, #2: -x+1, y-1/2, -z+3/2.

Table S2 Hydrogen bond lengths (Å) and angles (°) for complexes **1-4***

D-H...A	d-D(H)/Å	d(H...A)/Å	d(D...A)/Å	∠D-H...A/°
Complex 1				
O(4)-H(4B)...O(9)	0.85	1.82	2.530(4)	139.2
N(1)-H(1)...O(8) ^{#2}	0.86	1.87	2.686(4)	157.2
O(4)-H(4A)...O(8) ^{#3}	0.85	1.93	2.682(4)	146.0
C(12)-H(12)...O(2) ^{#4}	0.93	2.58	3.347(5)	139.8
Complex 2				
O(2)-H(2A)...O(1) ^{#2}	0.85	2.50	3.028(4)	120.9
Complex 3				
N(1)-H(1A)...O(5)	0.90	1.87	2.757(7)	166.3
N(1)-H(1B)...O(2) ^{#1}	0.90	2.11	2.915(7)	148.8
Complex 4				
O(1W)-H(1WB)...O(8)	0.85	1.97	2.818(9)	175.2
O(1W)-H(1WA)...O(12) ^{#5}	0.85	2.30	3.144(11)	173.7
O(12)-H(12A)...O(9) ^{#4}	0.85	1.95	2.769(7)	161.0

*Symmetry transformation used to generate equivalent atoms: complex **1**: #2: -x+1, -y, -z, #3: -x+3/2, y+1/2, -z+1/2, #4: x-1/2, -y-1/2, z-1/2; complex **2**: #2: -x, -y+1, -z+1; complex **3**: #1: x, -y+1/2, z+1/2; complex **4**: #4: -x+1, y+1/2, -z+3/2, #5: x+1, y, z.

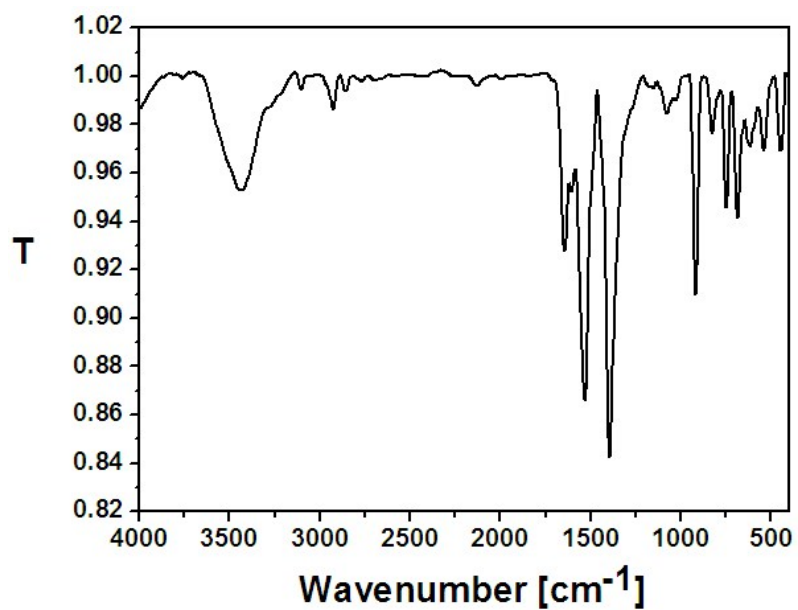


Fig. S1 The Solid-state IR spectra of complex **1** at a room temperature.

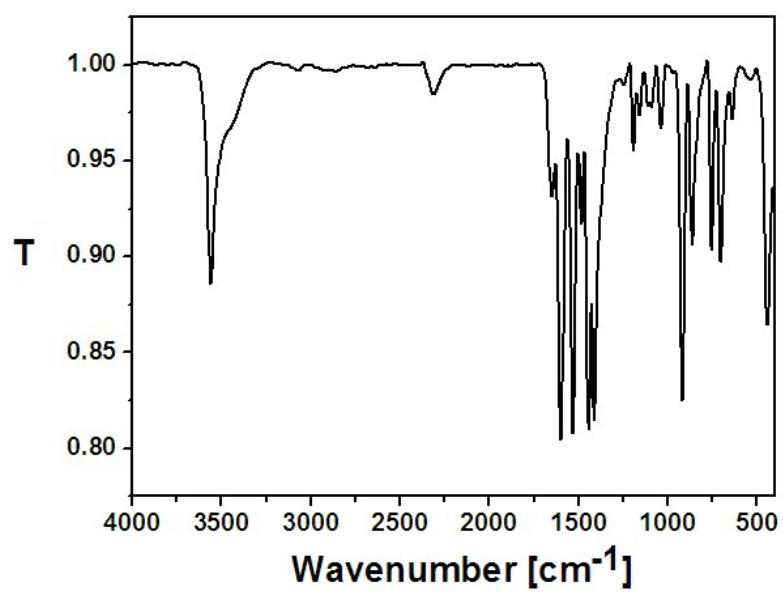


Fig. S2 The Solid-state IR spectra of complex **2** at a room temperature.

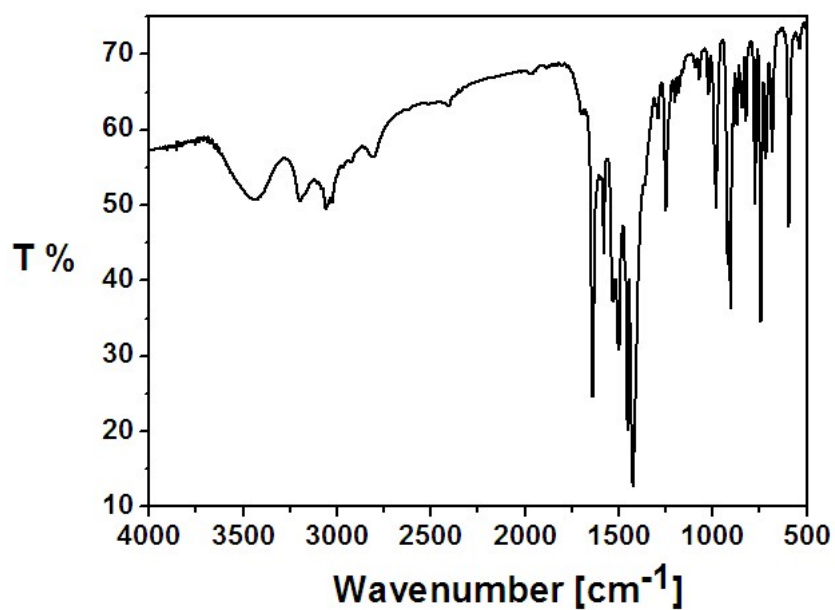


Fig. S3 The Solid-state IR spectra of complex 3 at a room temperature.

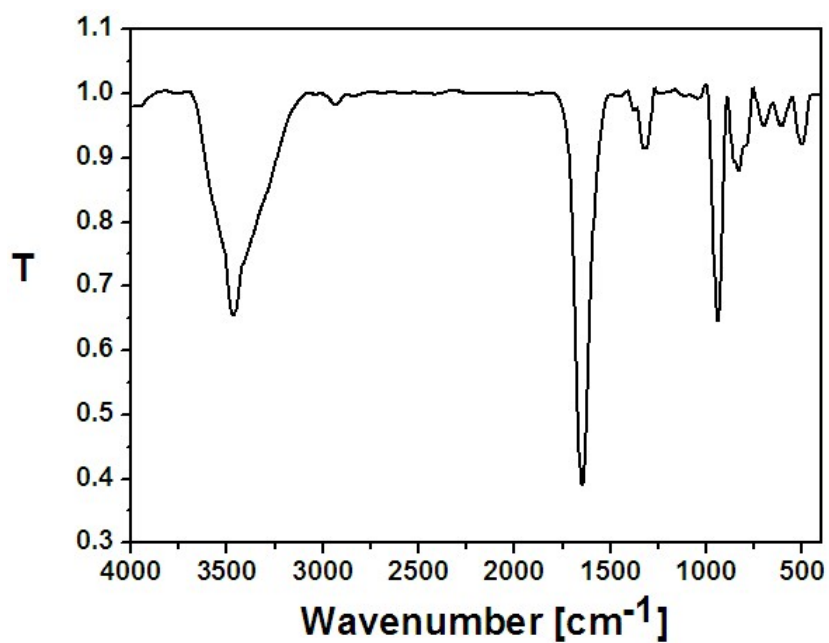


Fig. S4 The Solid-state IR spectra of complex 4 at a room temperature.

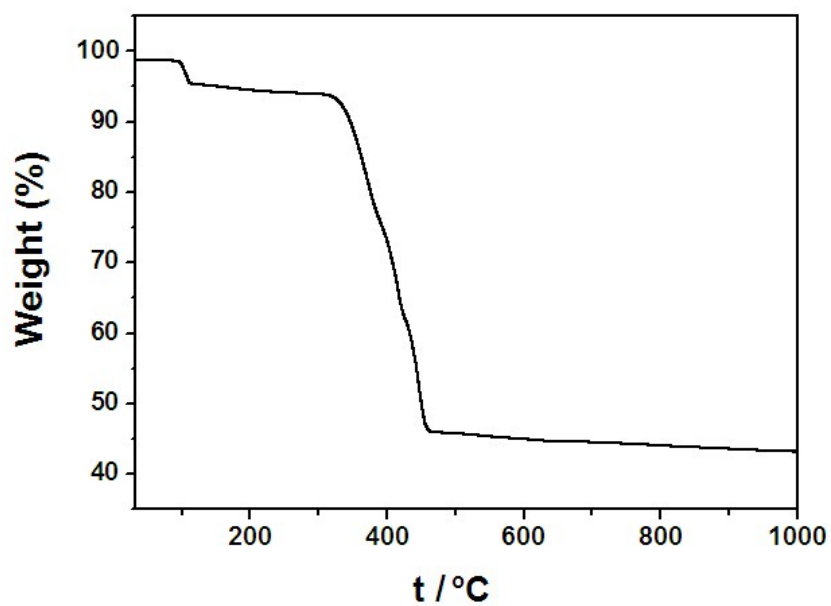


Fig. S5 The TG curve for complex 1.

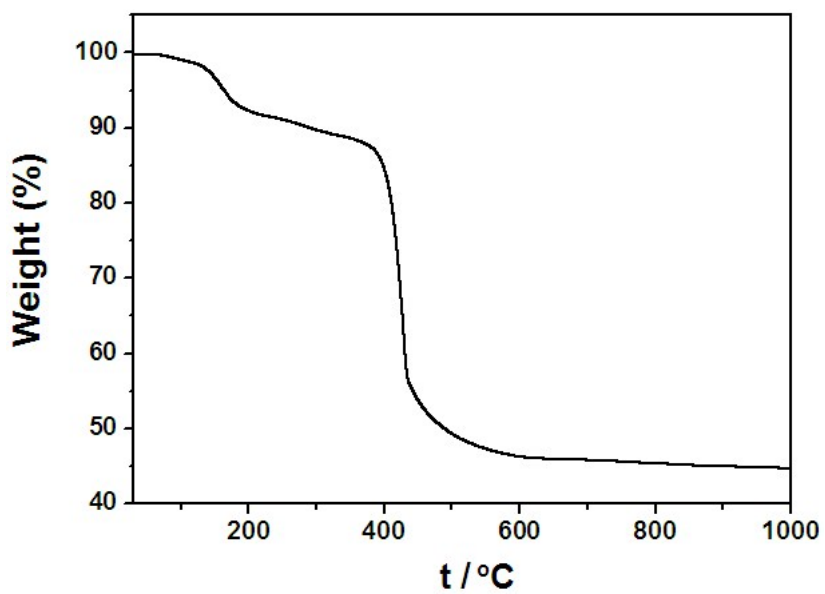


Fig. S6 The TG curve for complex 2.

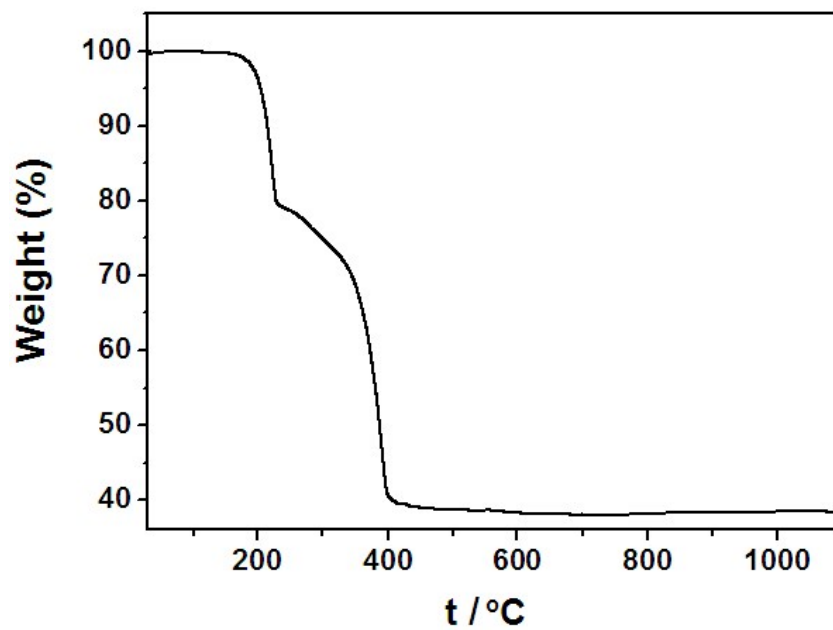


Fig. S7 The TG curve for complex 3.

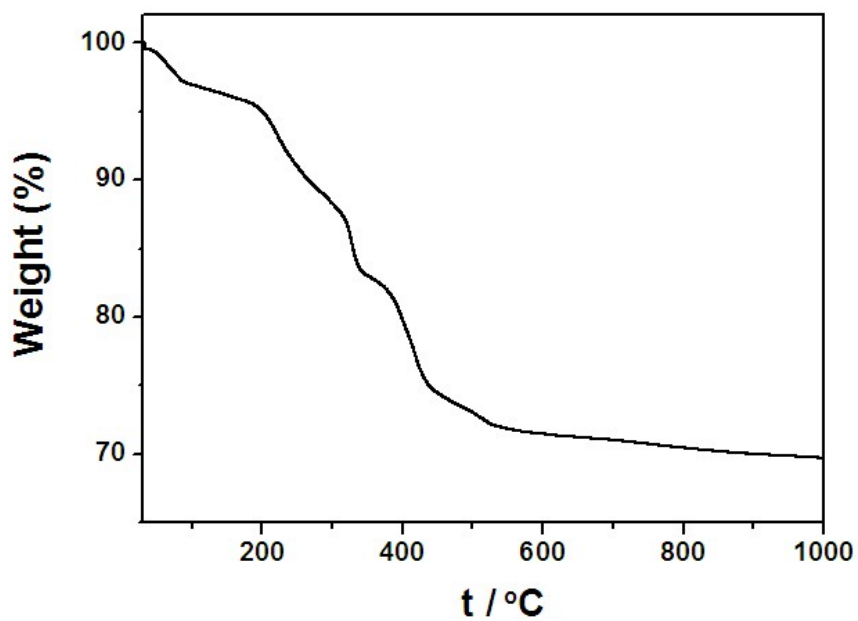


Fig. S8 The TG curve for complex 4.

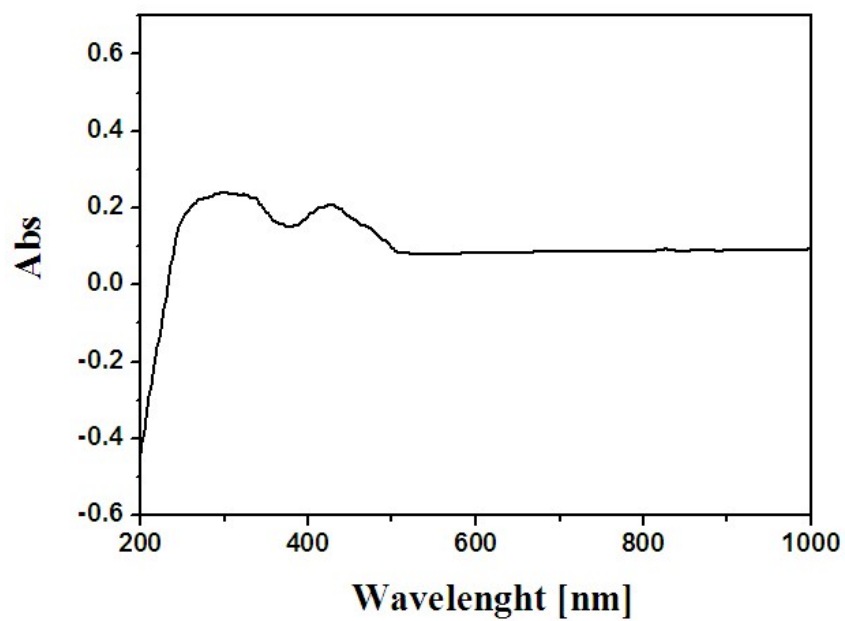


Fig. S9 The UV-Vis spectra for complex 1.

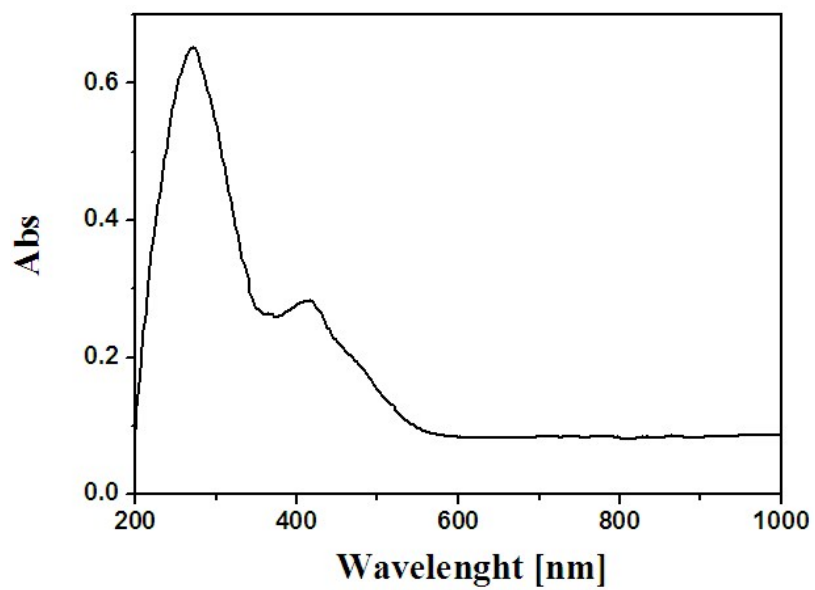


Fig. S10 The UV-Vis spectra for complex 2.

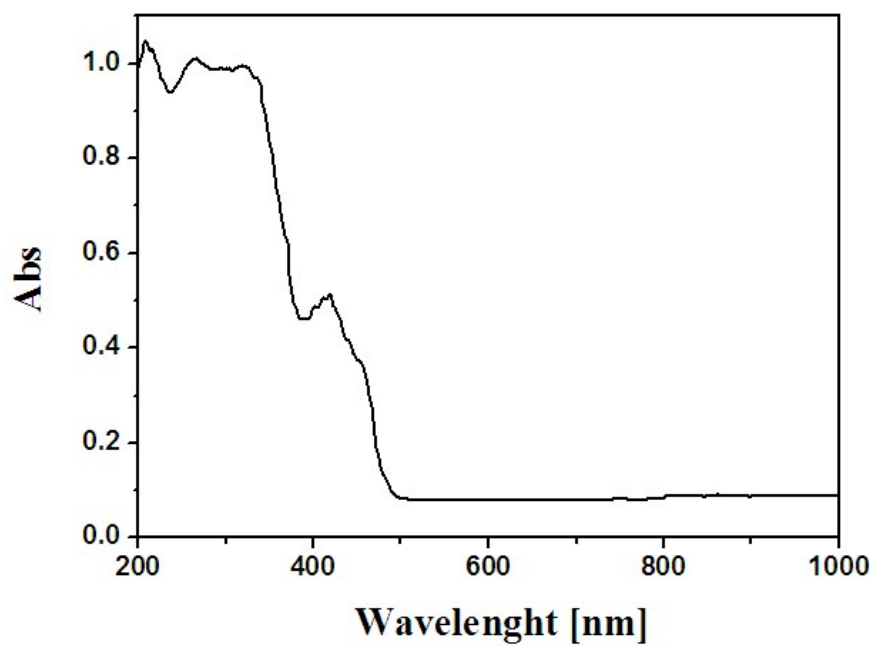


Fig. S11 The UV-Vis spectra for complex 3.

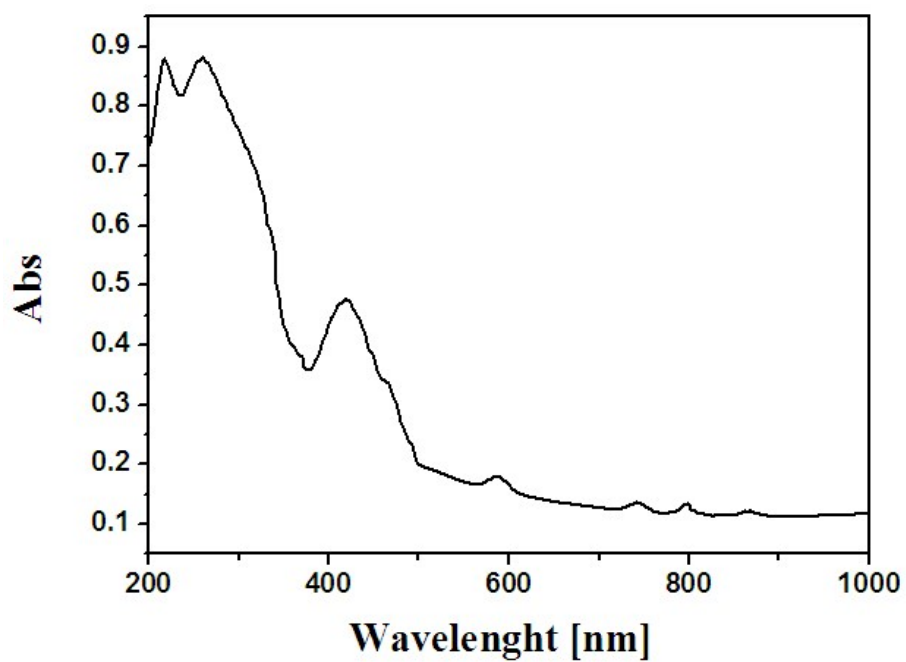


Fig. S12 The UV-Vis spectra for complex 4.

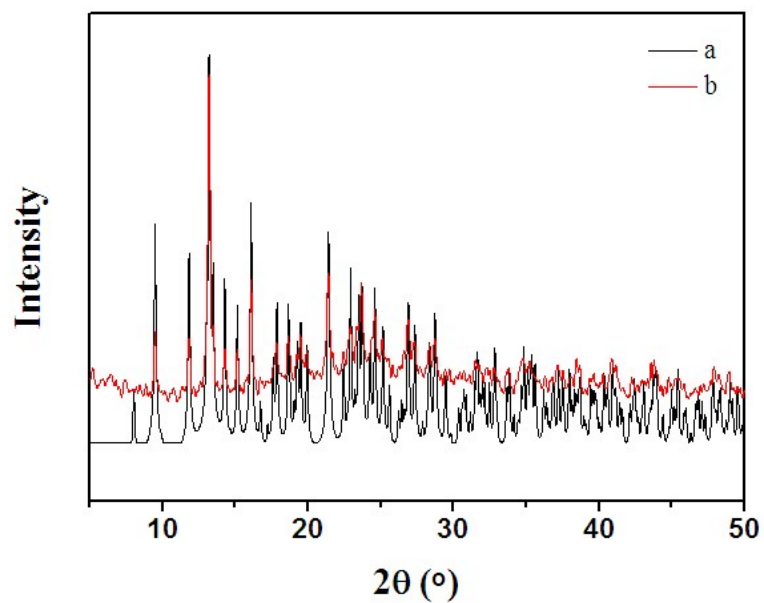


Fig. S13 PXR D powder patterns: (a) the simulated PXR D pattern calculated from single-crystal structure of complex **1**, (b) the experimental PXR D for complex **1**.

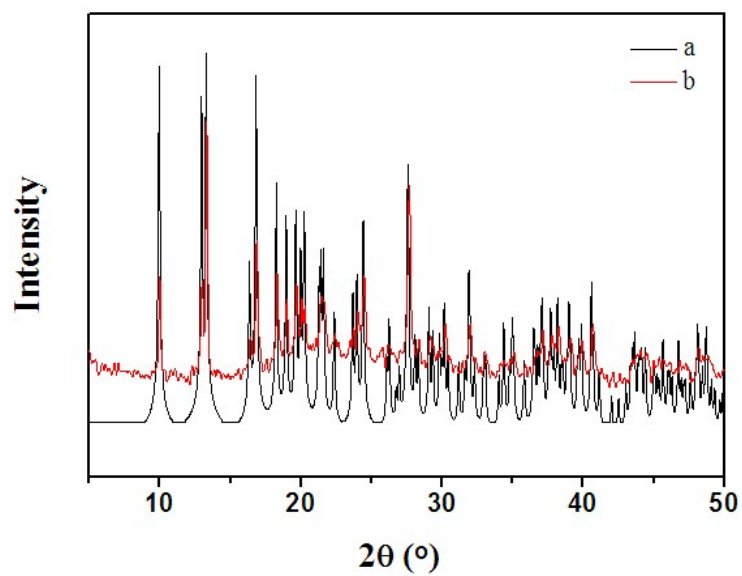


Fig. S14 PXR D powder patterns: (a) the simulated PXR D pattern calculated from single-crystal structure of complex **2**, (b) the experimental PXR D for complex **2**.

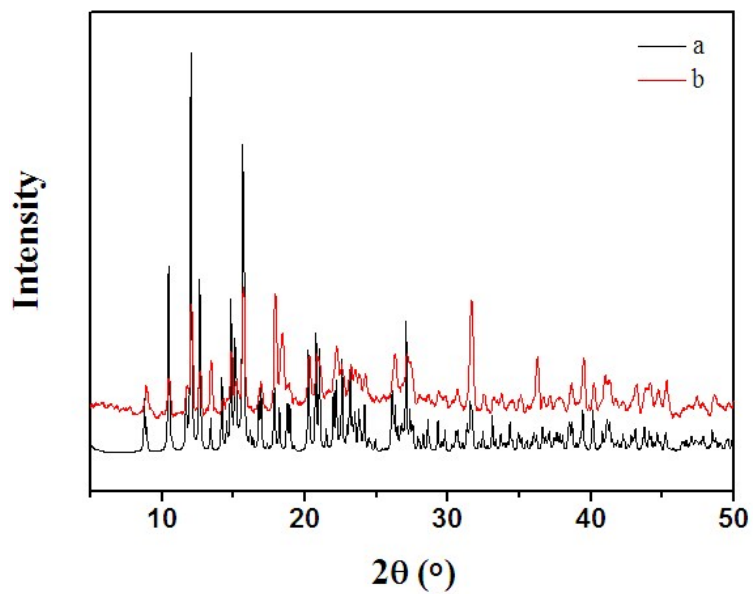


Fig. S15 PXRD powder patterns: (a) the simulated PXRD pattern calculated from single-crystal structure of complex **3**, (b) the experimental PXRD for complex **3**.

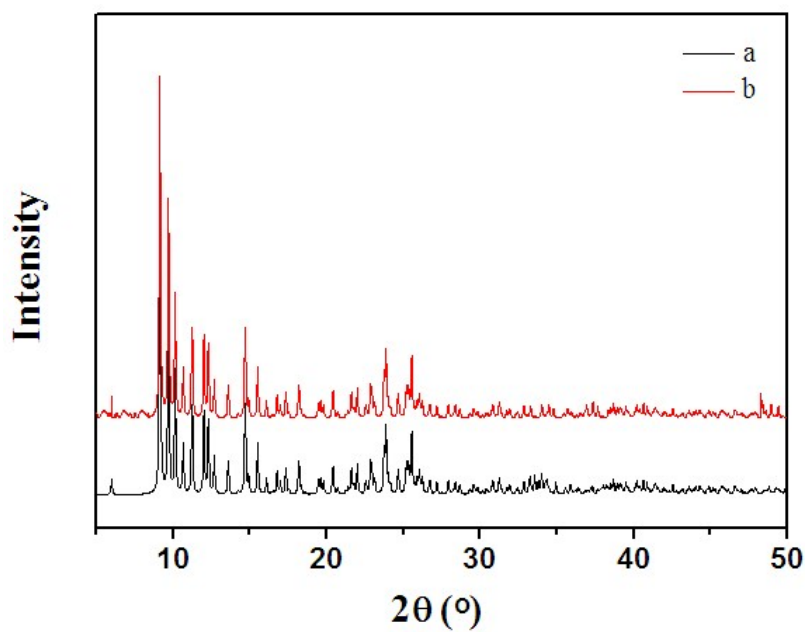


Fig. S16 PXRD powder patterns: (a) the simulated PXRD pattern calculated from single-crystal structure of complex **4**, (b) the experimental PXRD for complex **4**.