

ESI - Supplementary material

**Surfactant free synthesis of high surface area Pt@PdM₃ (M = Mn, Fe, Co, Ni, Cu)
core/shell with enhanced electrocatalytic activity and durability for PEM fuel cell
application.**

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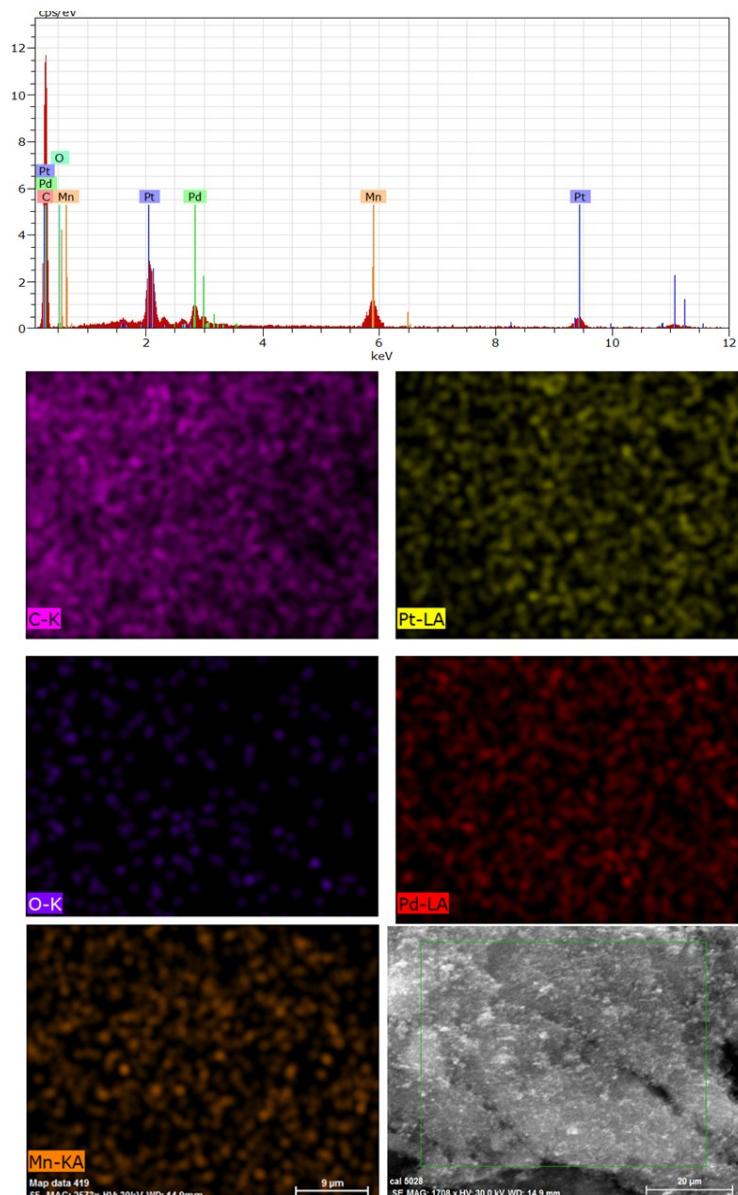


Fig. S1 SEM EDAX and elemental mapping images of Pt@PdMn₃/C electrocatalyst.

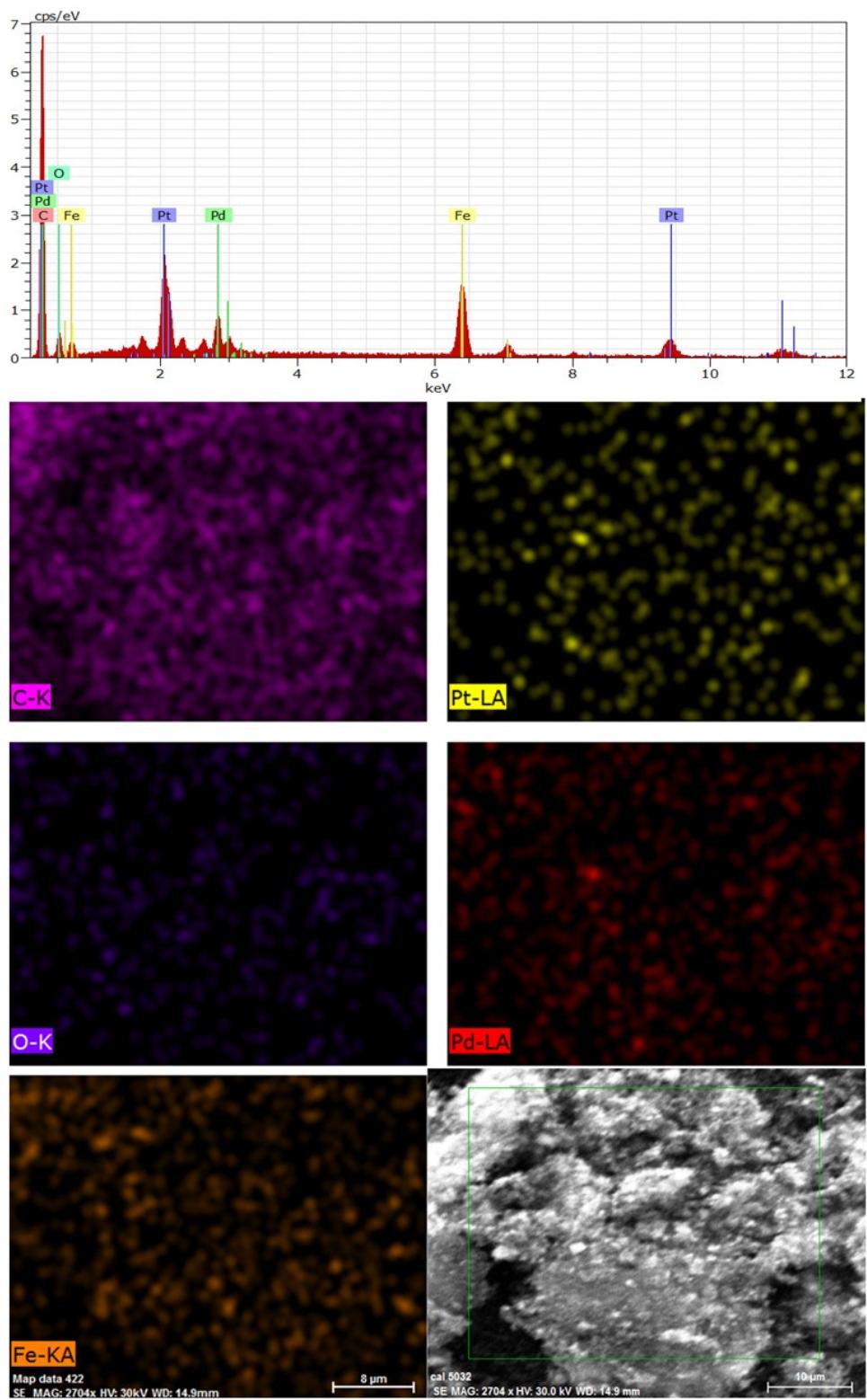


Fig. S2 SEM EDAX and elemental mapping images of Pt@PdFe₃/C electrocatalyst.

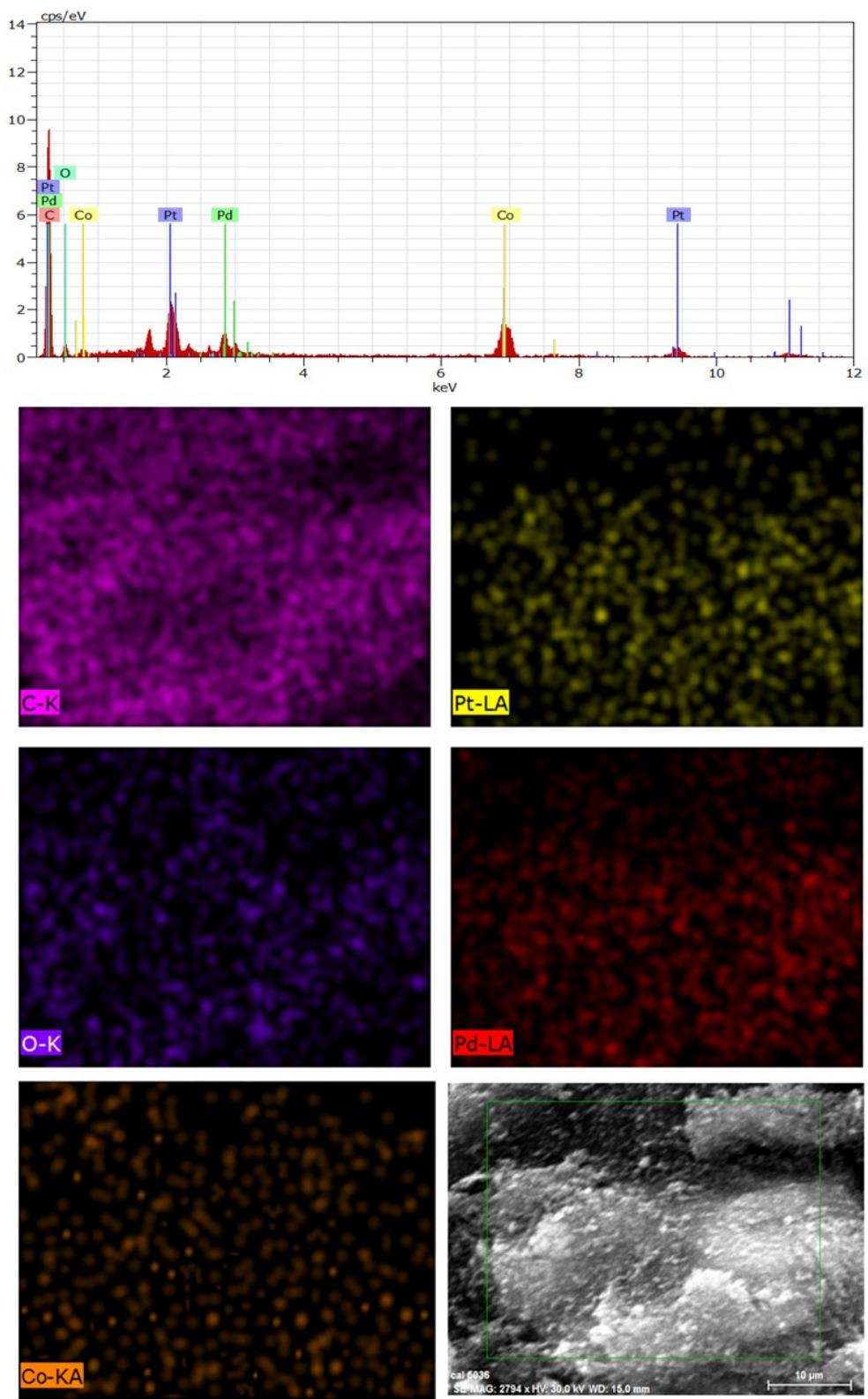


Fig. S3 SEM EDAX and elemental mapping images of Pt@PdCo₃/C electrocatalyst.

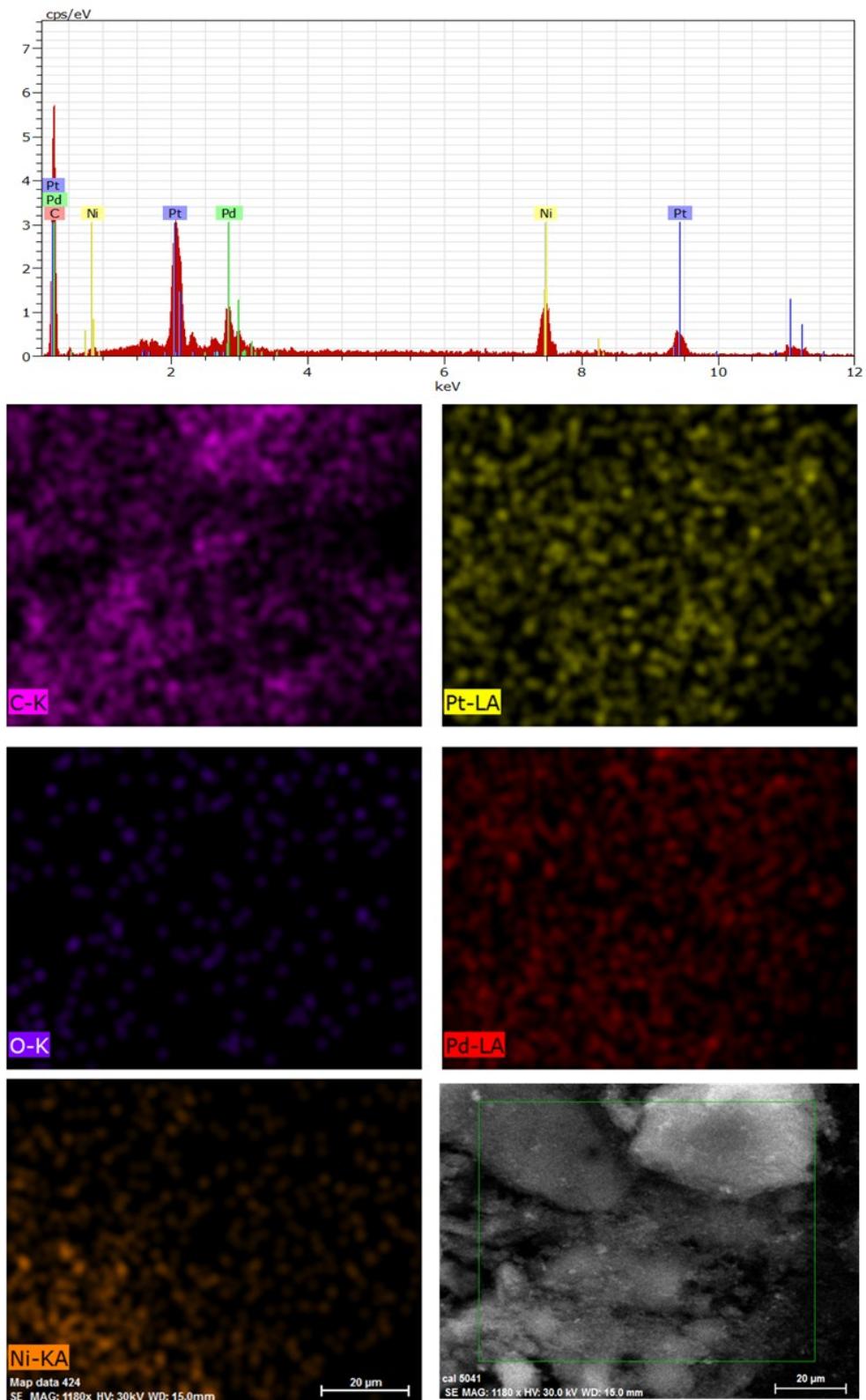


Fig. S4 SEM EDAX and elemental mapping images of Pt@PdNi₃/C electrocatalyst.

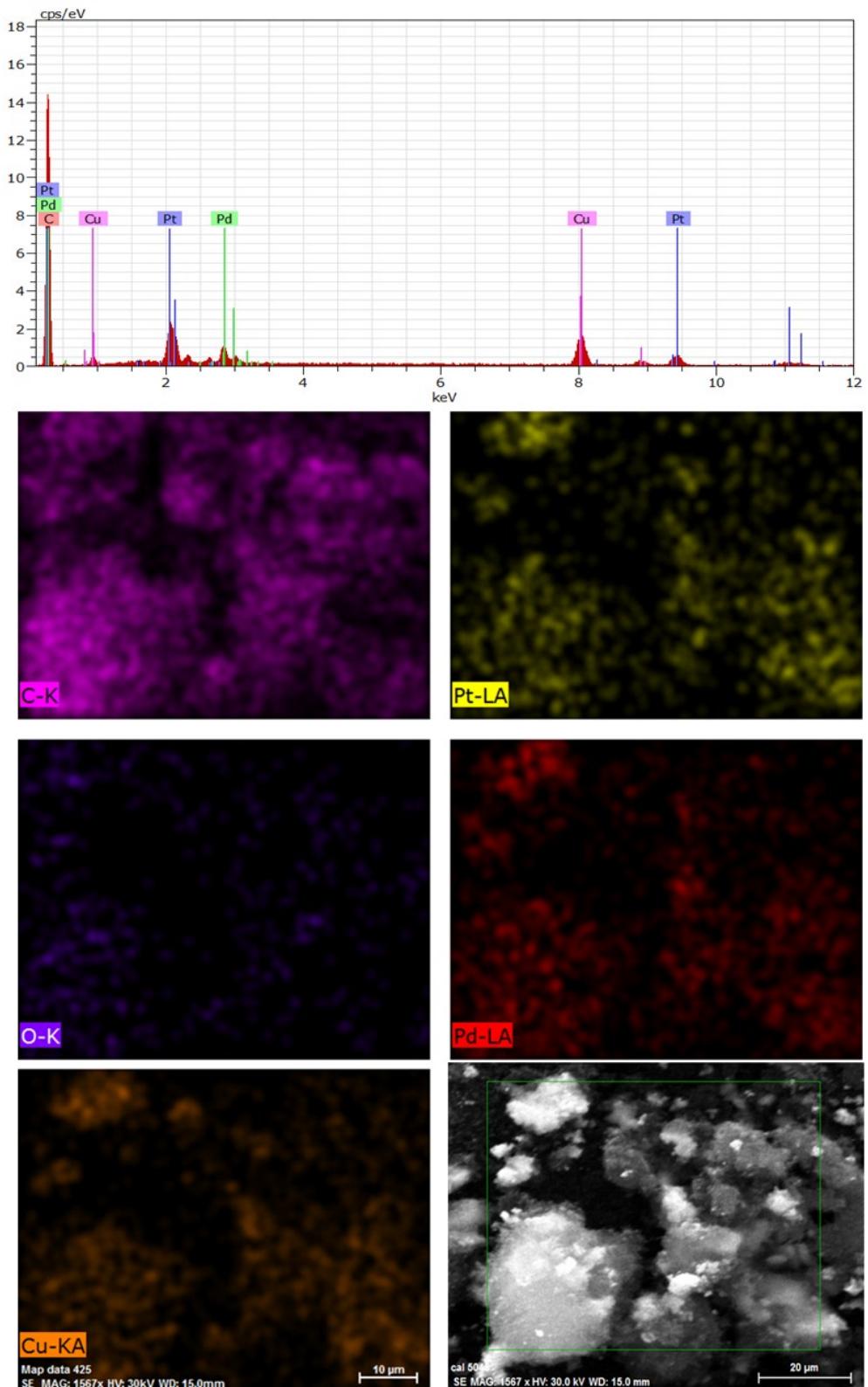
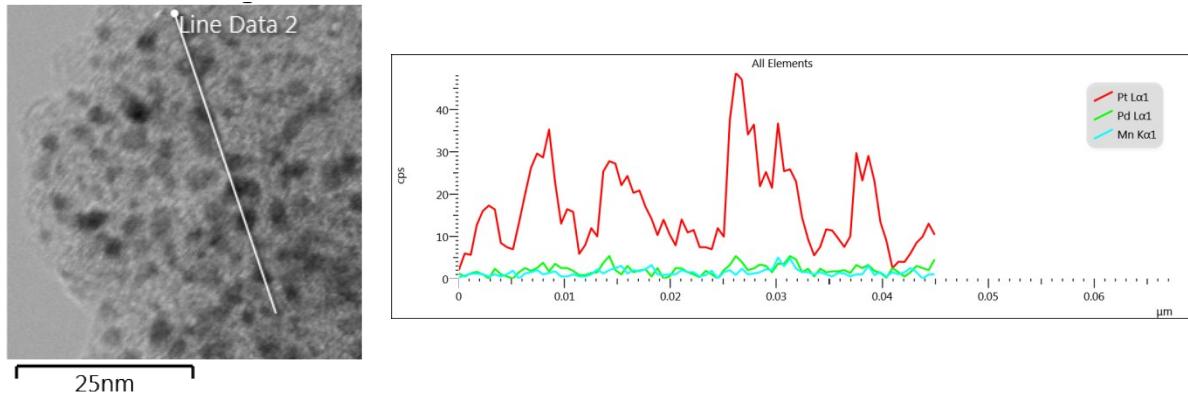
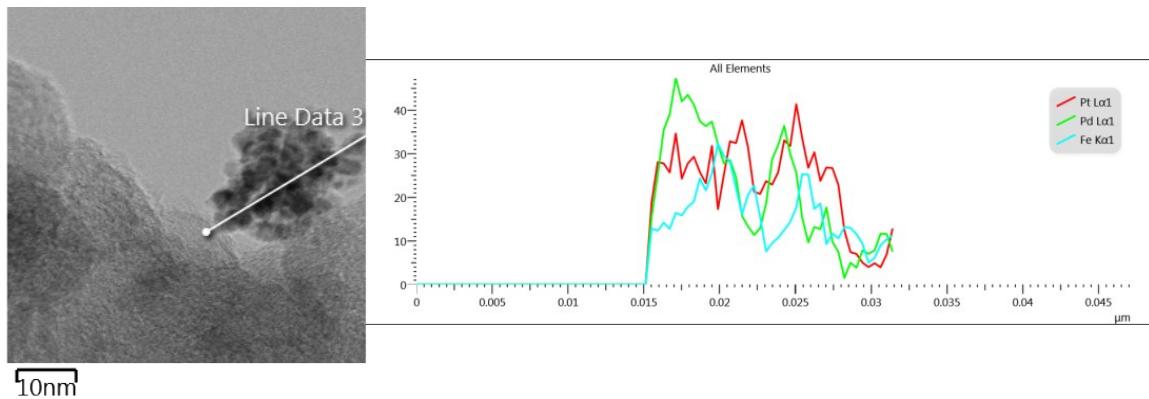


Fig. S5 SEM EDAX and elemental mapping images of Pt@PdCu₃/C electrocatalyst.

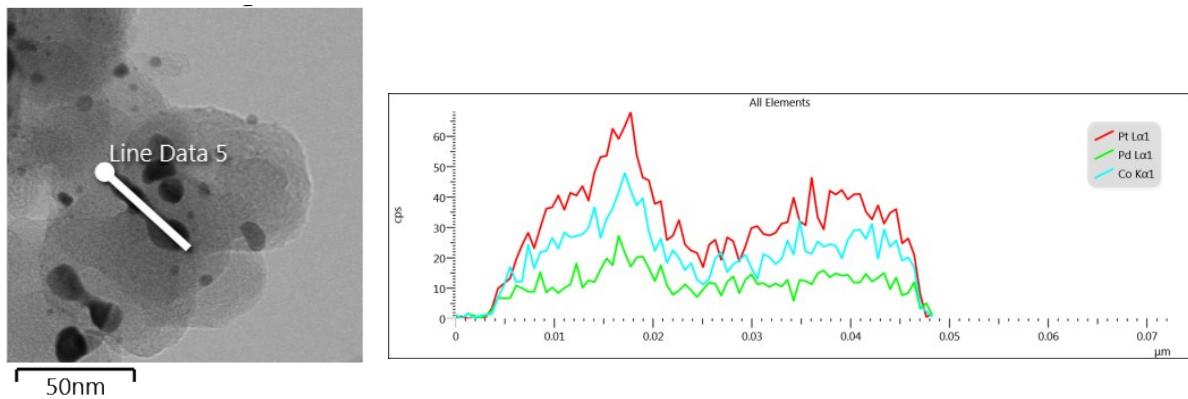
Pt@PdMn₃/C



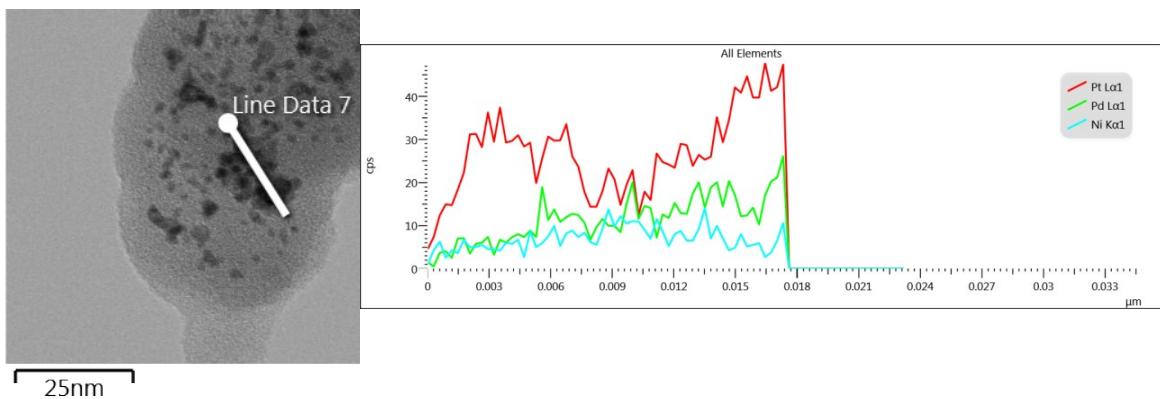
Pt@PdFe₃/C



Pt@PdCo₃/C



Pt@PdNi₃/C



Pt@PdCu₃/C

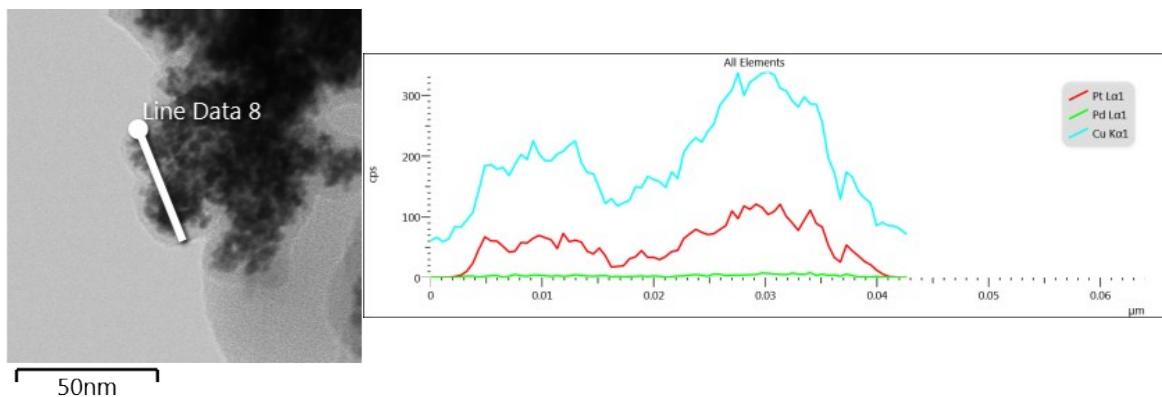


Fig. S6 TEM line elemental mapping images of Pt@PdM₃/C (M = Mn, Fe, Co, Ni, Cu) electrocatalyst

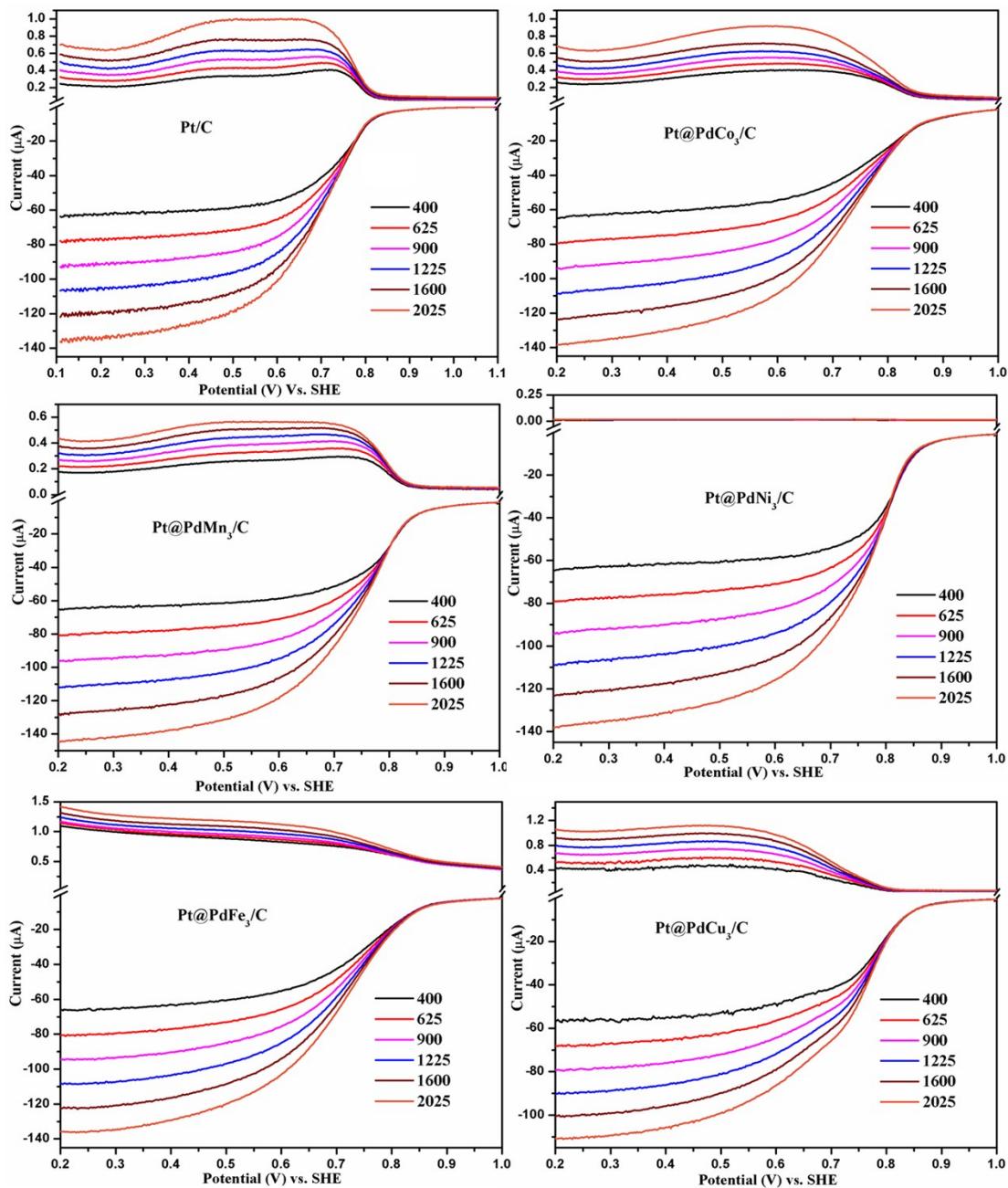


Fig. S7 RRDE measurements (at 400 – 2025 rpm values) of oxygen reduction reaction of Pt/C, Pt@PdM₃/C (M = Mn, Fe, Co, Ni, Cu) electrocatalysts in O₂ saturated 0.5 M H₂SO₄ at a scan rate of 10 mV s⁻¹. Ring electrode potential was fixed at 1.2 V vs. SHE.

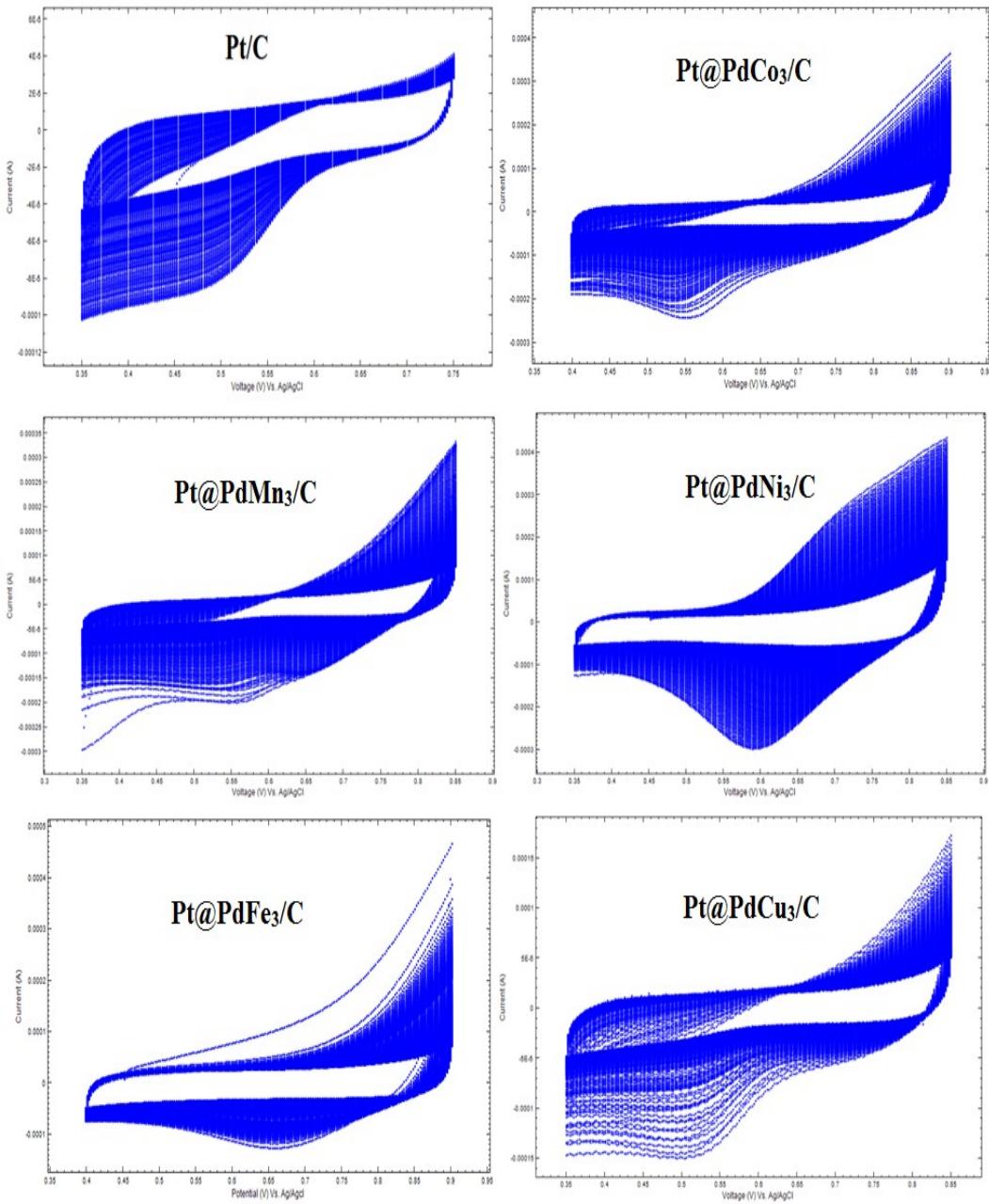


Fig. S8 1000 Potential cycling in the potential window of 0.3 to 0.8 V vs. Ag/AgCl in presence of N₂ saturated 0.5 M H₂SO₄.