

Supplementary Information

Accessible bidentate diol functionality within highly ordered composite periodic mesoporous organosilicas

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1. Solid-state NMR characterization of materials

2. NMR characterization of precursors

1. Solid-state NMR characterization of materials

1.1 Deprotection of solid-state MeO- and MEMO-PMOs

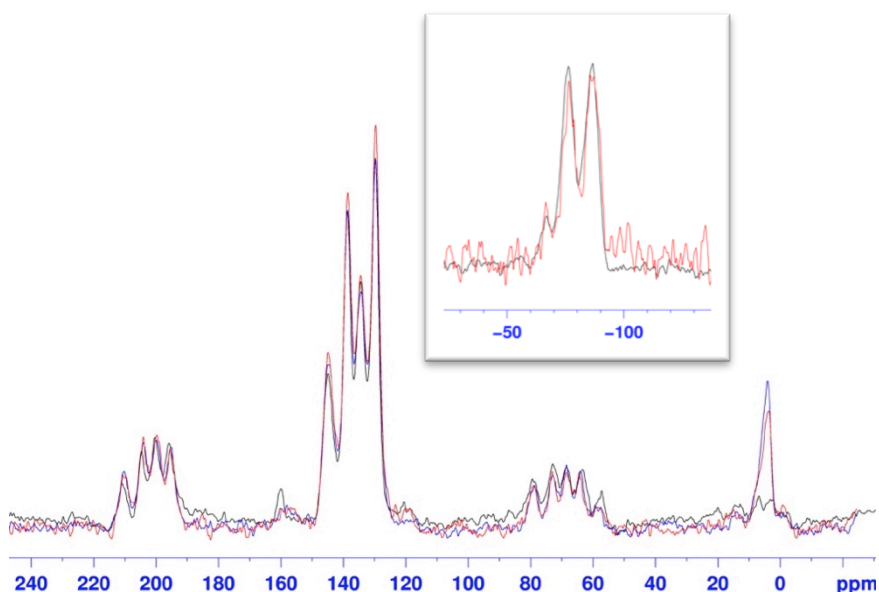


Figure S1. Solid-state ^{13}C CP MAS NMR of MeO-cPMO before deprotection (black), passivated with TMS and subsequently deprotected with boron tribromide over 24 hours at room temperature (blue), and subjected to deprotection a second time using the same conditions (red); and ^{29}Si NMR (inset) of MeO-cPMO material before (black) and after (red) deprotection

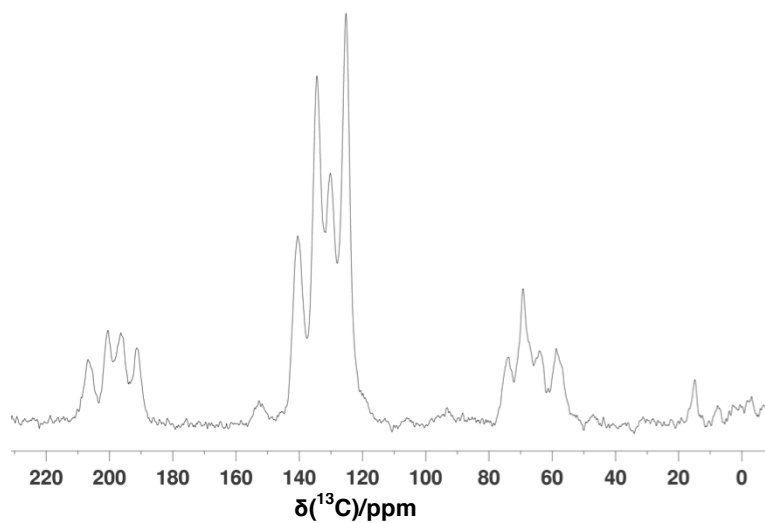


Figure S2. Solid-state ^{13}C CP MAS NMR spectra of $^{20}\text{MEMO-cPMO}$, after extraction of the surfactant with acidified ethanol

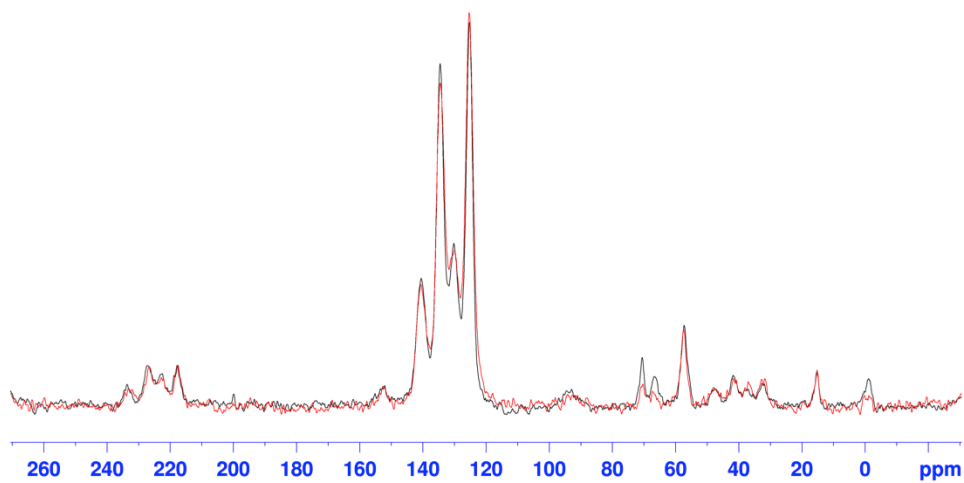


Figure S3. Solid-state ^{13}C CP MAS NMR of TMS-passivated MEMO-PMO treated with acidified ethanol at $55\text{ }^{\circ}\text{C}$ for 6 hours (black) and subsequently treated for 24 hours (red) to cleave MEM groups

1.2 Preparation of phosphate ester materials

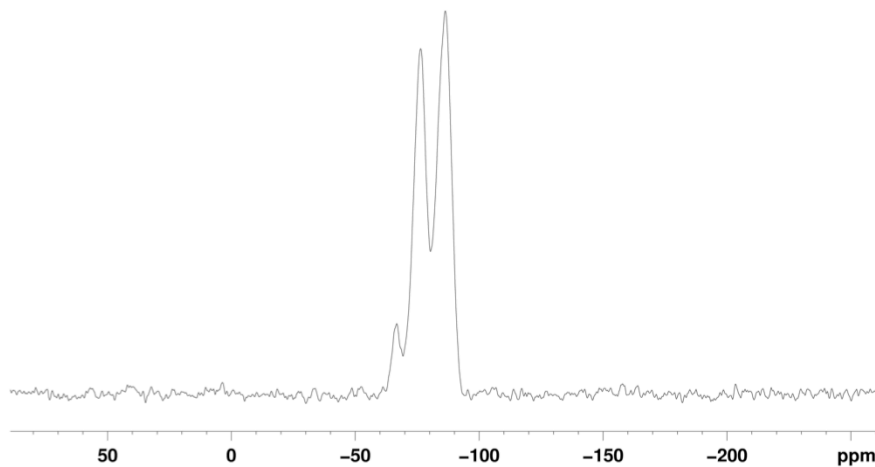


Figure S4. Solid-state ^{29}Si CP MAS NMR of BTESBp PMO treated with phosphorus oxychloride

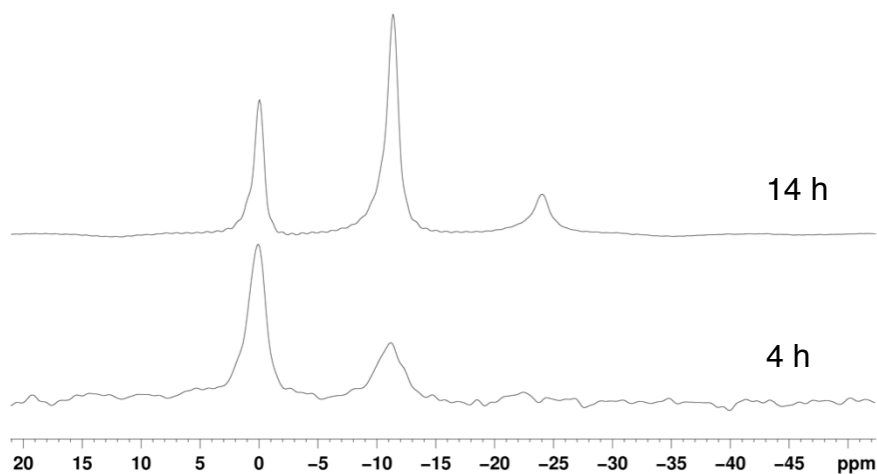


Figure S5. Solid-state ^{31}P MAS NMR of MCM-41 treated with POCl_3 for 4 h or 14 h to prepare HOP(O)-MCM showed an increase in the -12, -22 ppm signals with longer treatment times, indicating a higher degree of $[\text{O}=\text{P}-(\text{OSi})(\text{OH})_2]$ and $[\text{O}=\text{P}-(\text{OSi})_2(\text{OH})]$ formation at longer exposures

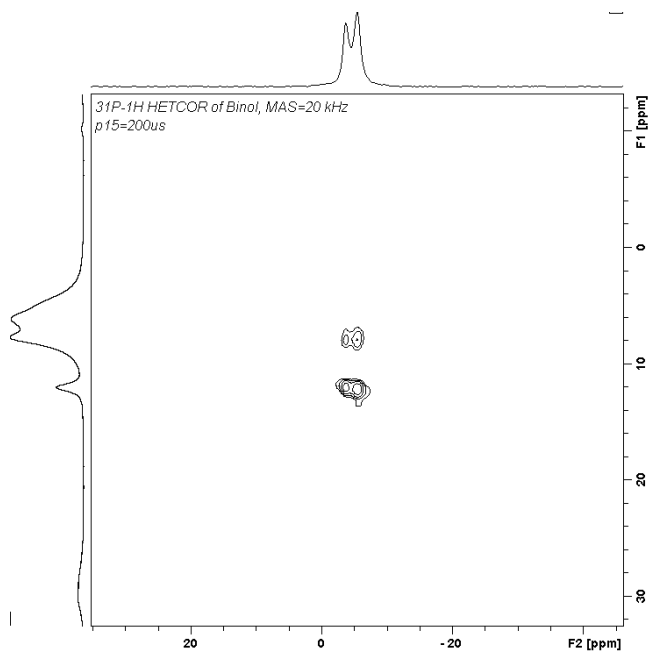


Figure S6. High-field solid -state ^{31}P - ^1H HETCOR spectra of BINOL hydrogen phosphate reference

1.3 Deconvolution and line fitting of solid-state ^{13}C CP MAS NMR spectra

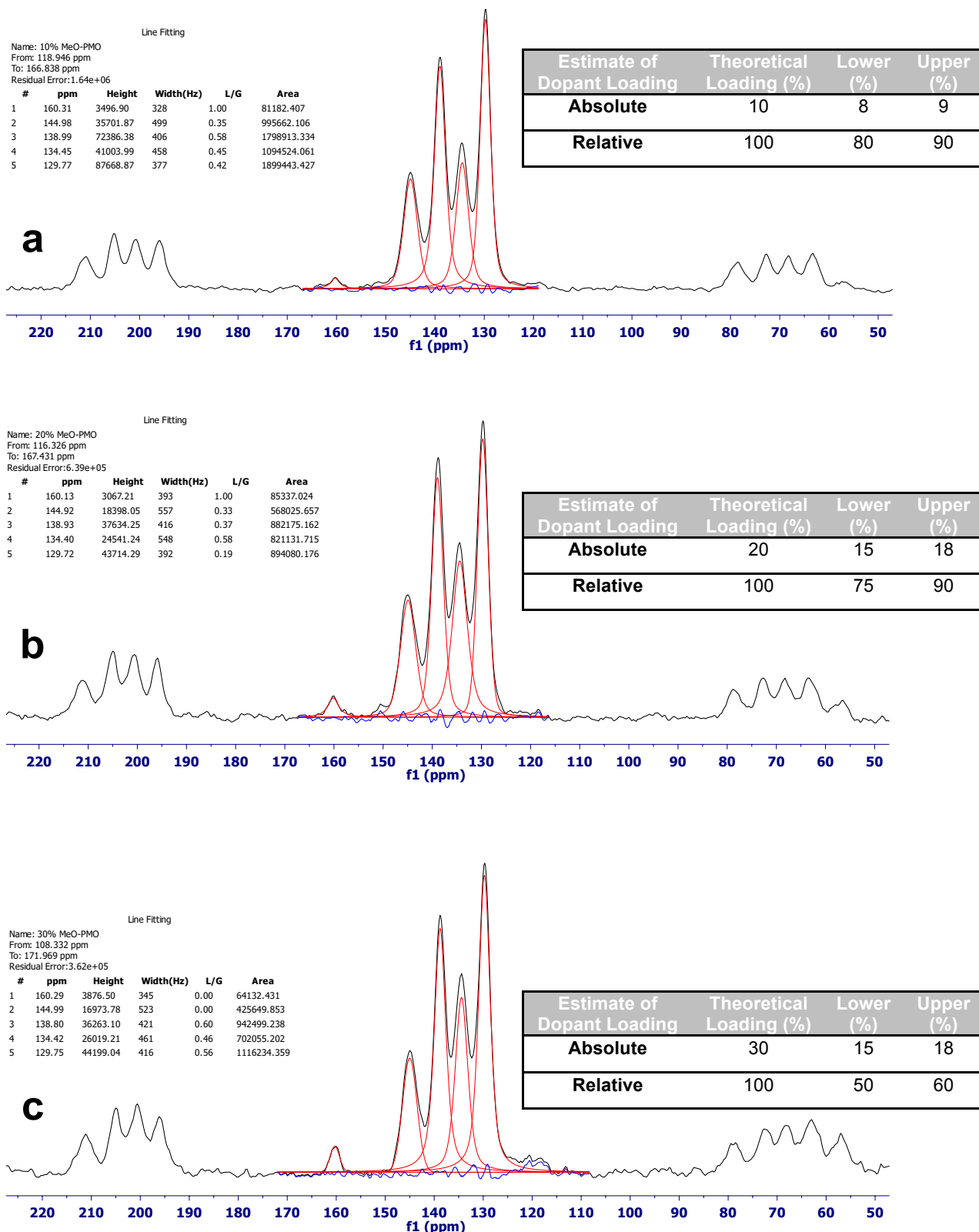


Figure S 7. Deconvolution and line fitting of ^{13}C CP MAS NMR spectra of a) $^{10}\text{MeO-PMO}$, b) $^{20}\text{MeO-PMO}$ and; c) $^{30}\text{MeO-PMO}$ showing fitted peak area report and calculated estimates of dopant loadings relative to the theoretical loading

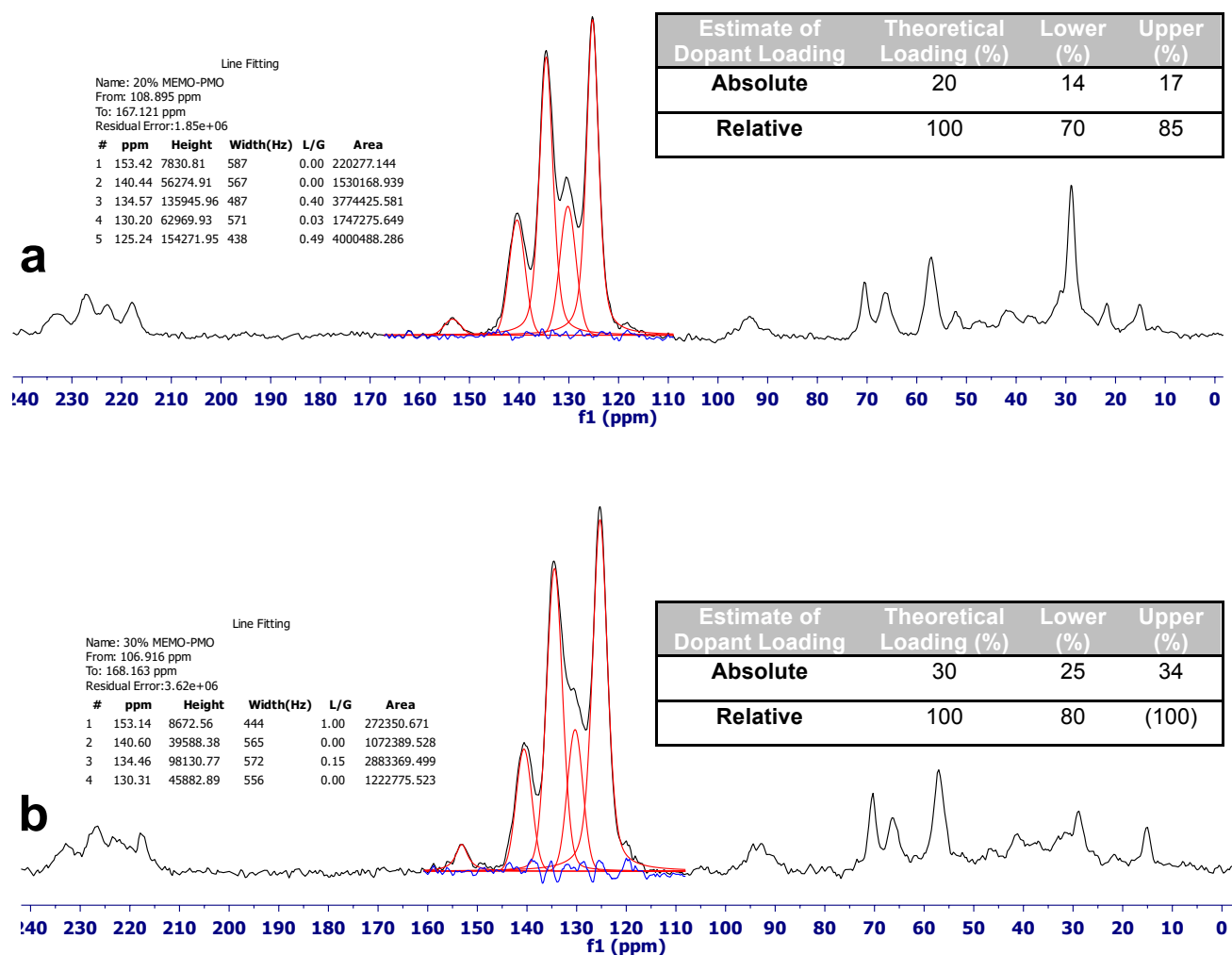


Figure S 8. Deconvolution and line fitting of ^{13}C CP MAS NMR spectra of a) $^{20}\text{MEMO-PMO}$ and; b) $^{30}\text{MEMO-PMO}$ showing fitted peak area reports and calculated estimates of dopant loadings relative to the theoretical loading

Estimate calculations

Deconvolution was achieved using the line-fitting feature of MestRec Nova NMR processing software. Estimates were calculated based on the ratio of areas of peaks 1 and 2, representing quaternary carbon signals from (Ar)C-O (dopant) and (Ar)C-C(Ar) (bulk and dopant), respectively. The dopant (Ar)C-C(Ar) overlaps completely here and thus (Ar)C-C(Ar) (bulk) cannot be evaluated independently. Upper estimates subtract Area_1 of (Ar)C-O (dopant) from Area_2 (bulk + dopant overlapped with bulk). Lower limits assume Area_1 is negligible to Area_2 and the overlapped signal is not subtracted. See example calculation for $^{10}\text{MeO-PMO}$:

$$\text{Lower estimate limit: } \frac{\text{mol\% dopant}}{\text{mol\% bulk}} = \frac{\text{Area}_1}{\text{Area}_2} \times 100\% = \frac{81182.407}{995662.106} \times 100\% = 8\%$$

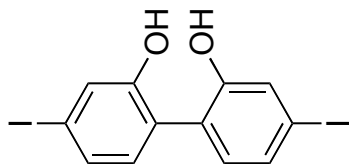
$$\text{Upper estimate limit: } \frac{\text{mol\% dopant}}{\text{mol\% bulk}} = \frac{\text{Area}_1}{(\text{Area}_2 - \text{Area}_1)} \times 100\% = \frac{81182.407}{(995662.106 - 81182.407)} \times 100\% = 9\%$$

2. NMR characterization of precursors



Proton 300 MHz
dihydroxybiphenyl

7.424
7.400
7.281
6.981
6.954

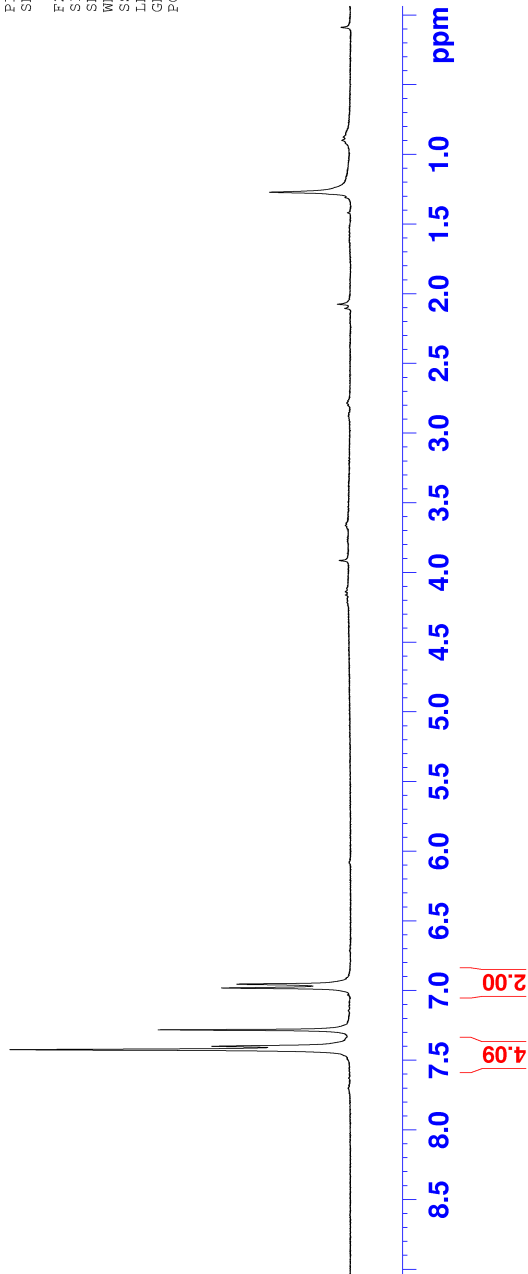


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FIDRES     0.1094190 Hz
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RG         645.1
DW         81.000 use
DE         6.00 use
TE         300.0 K
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TD0        1

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proton NMR on bbfo probe 400 MHz

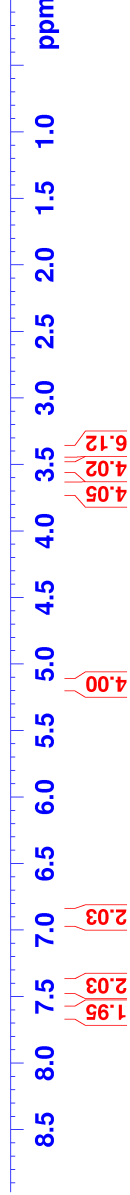
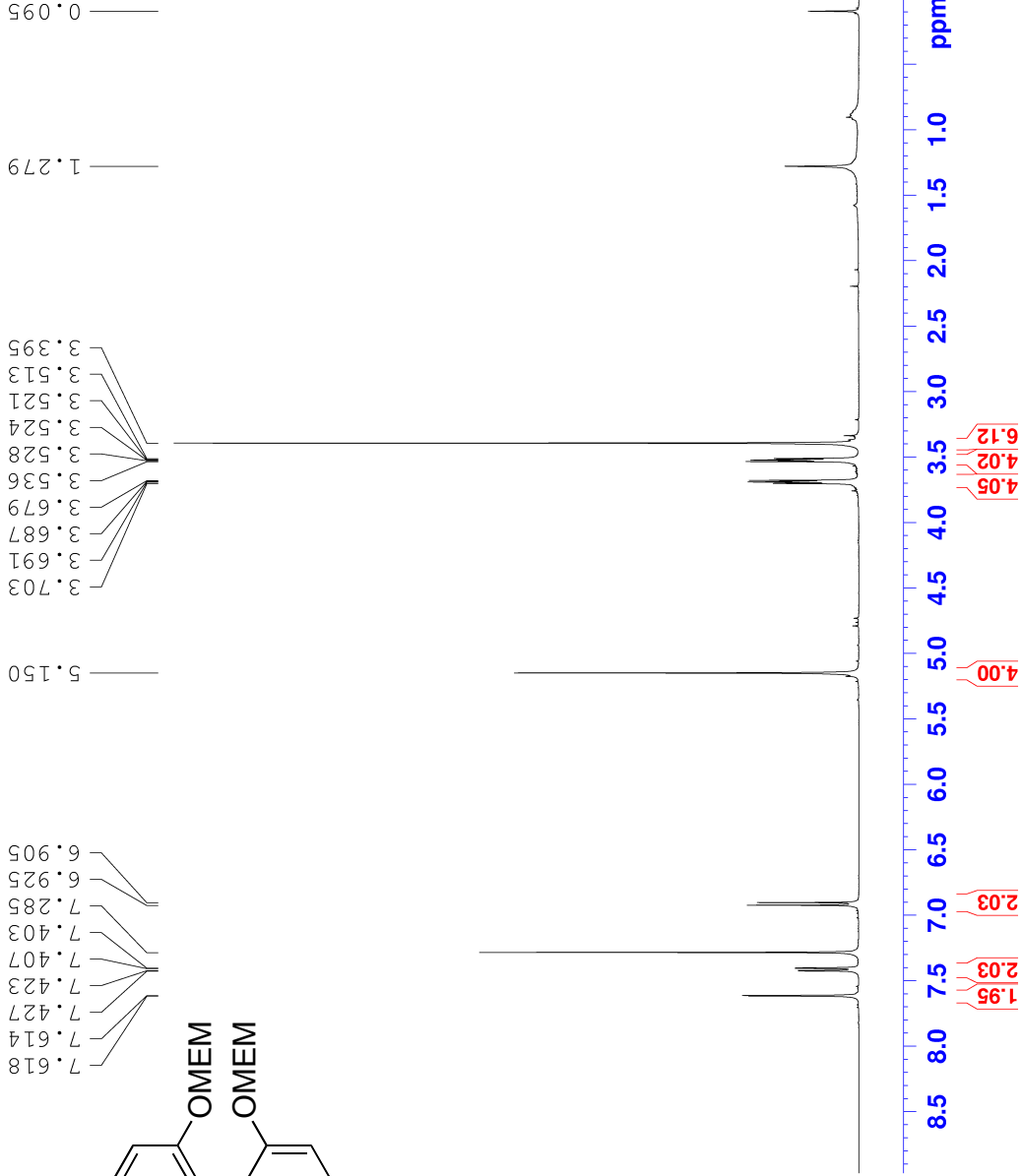
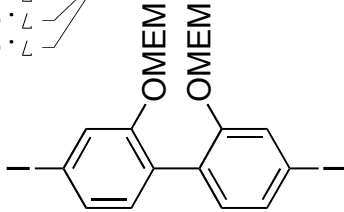
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3.513
3.521
3.524
3.528
3.536
3.679
3.687
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5.150
6.905
6.925
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7.403
7.407
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7.618

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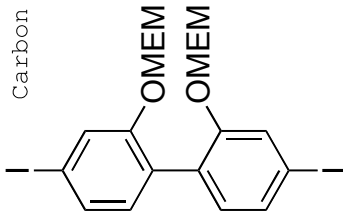
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Carbon NMR on bbfo probe



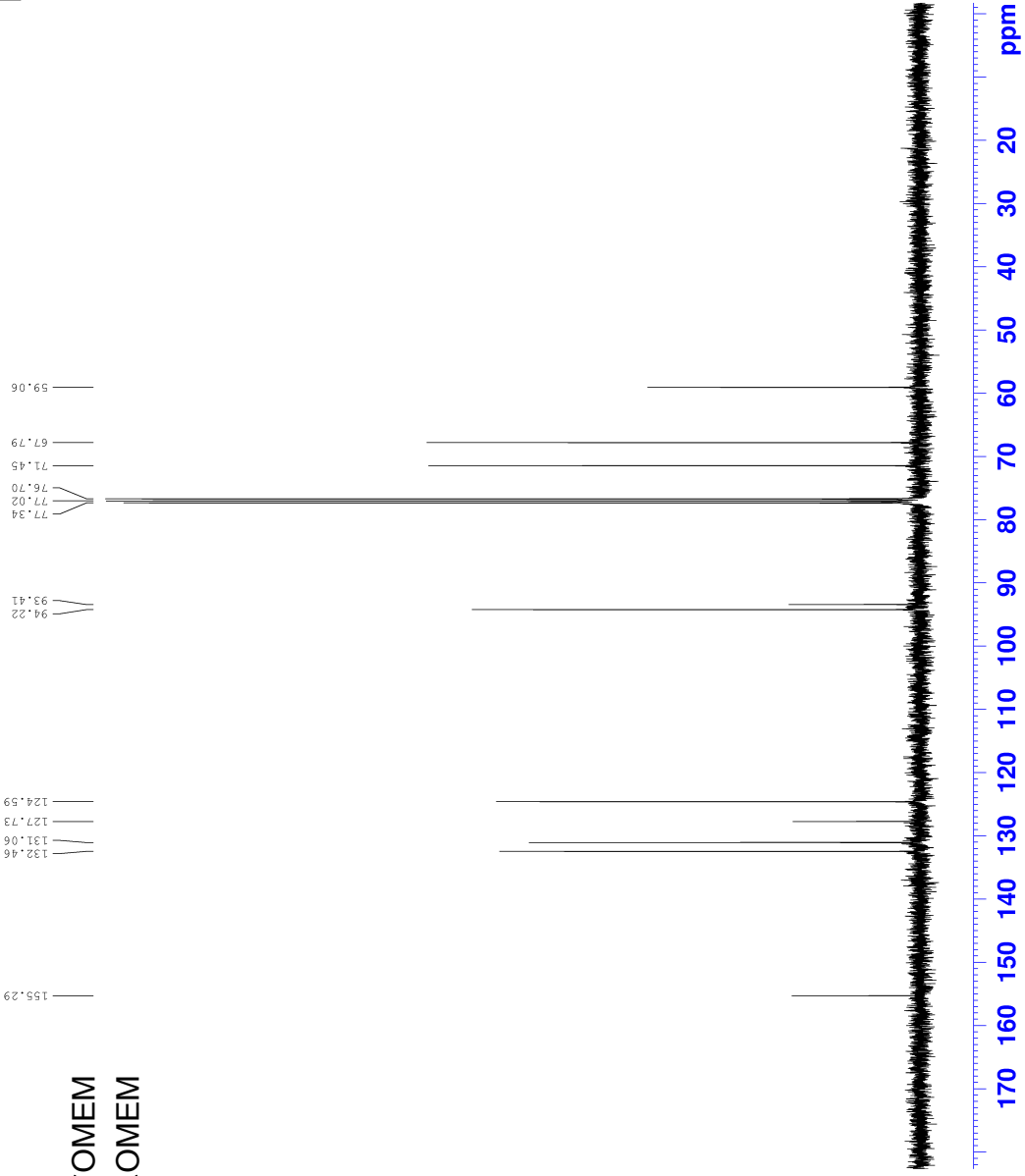
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d11 0.03000000 se
TD0 1

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PL1 -3.00 dB
SFO1 100.6228298 MH

===== CHANNEL f2 =====
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P2 100.00 us
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PL12 14.48 dB
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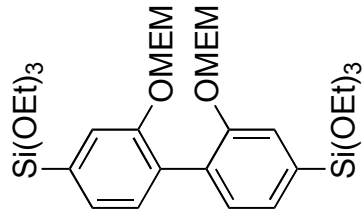


proton NMR on bbfo probe 400 MHz

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7.372
7.285
7.264

3.961
3.944
3.926
3.909
3.641
3.630
3.626
3.618
3.618
3.474
3.466
3.462
3.450
3.353

1.317
1.299
1.282

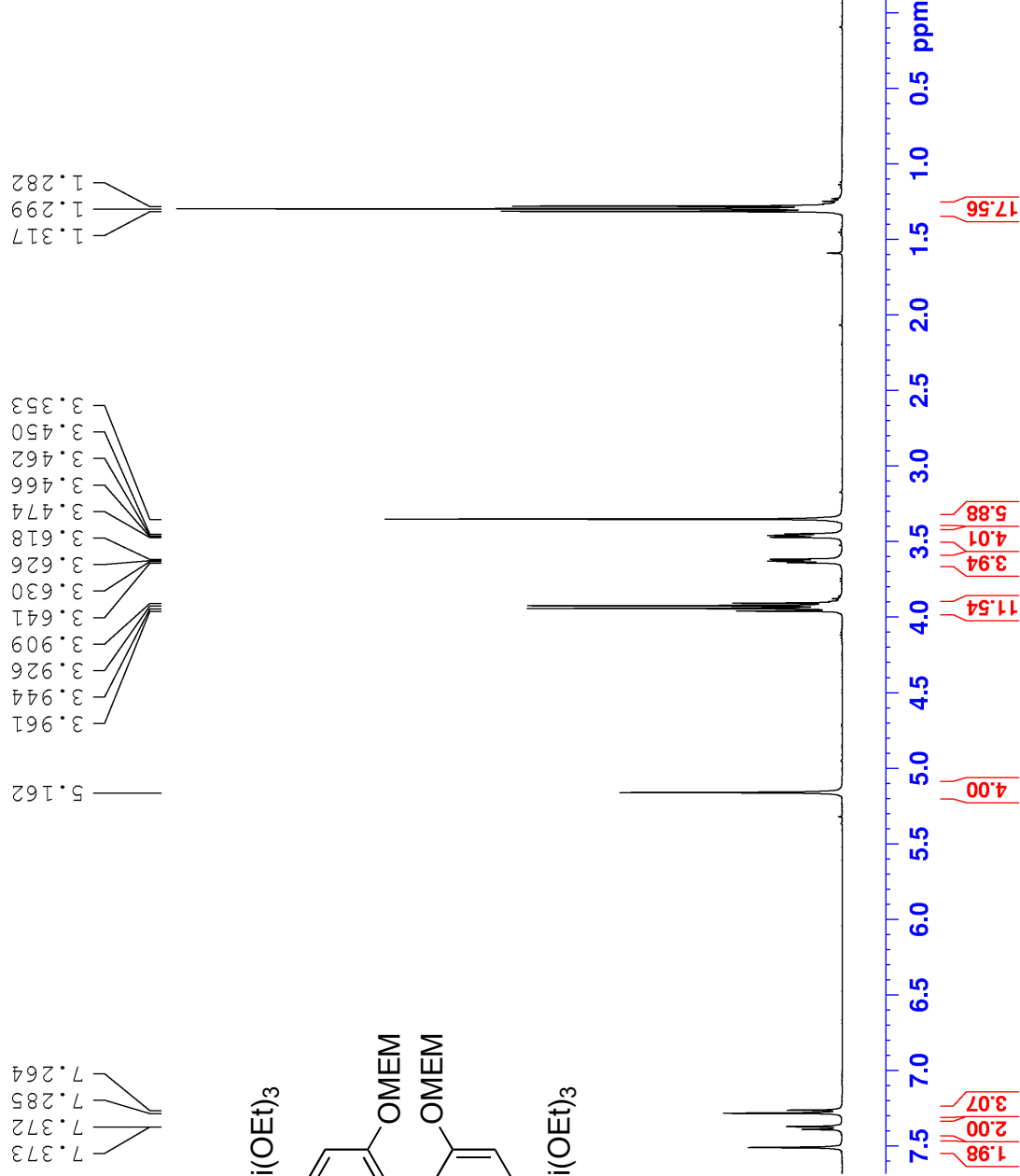


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RG 16
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1.98

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3.07

4.00

11.54

3.94

4.01

5.88

17.56



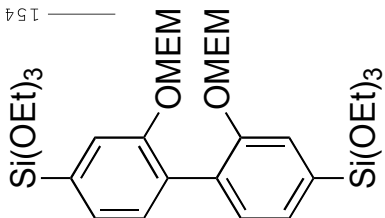
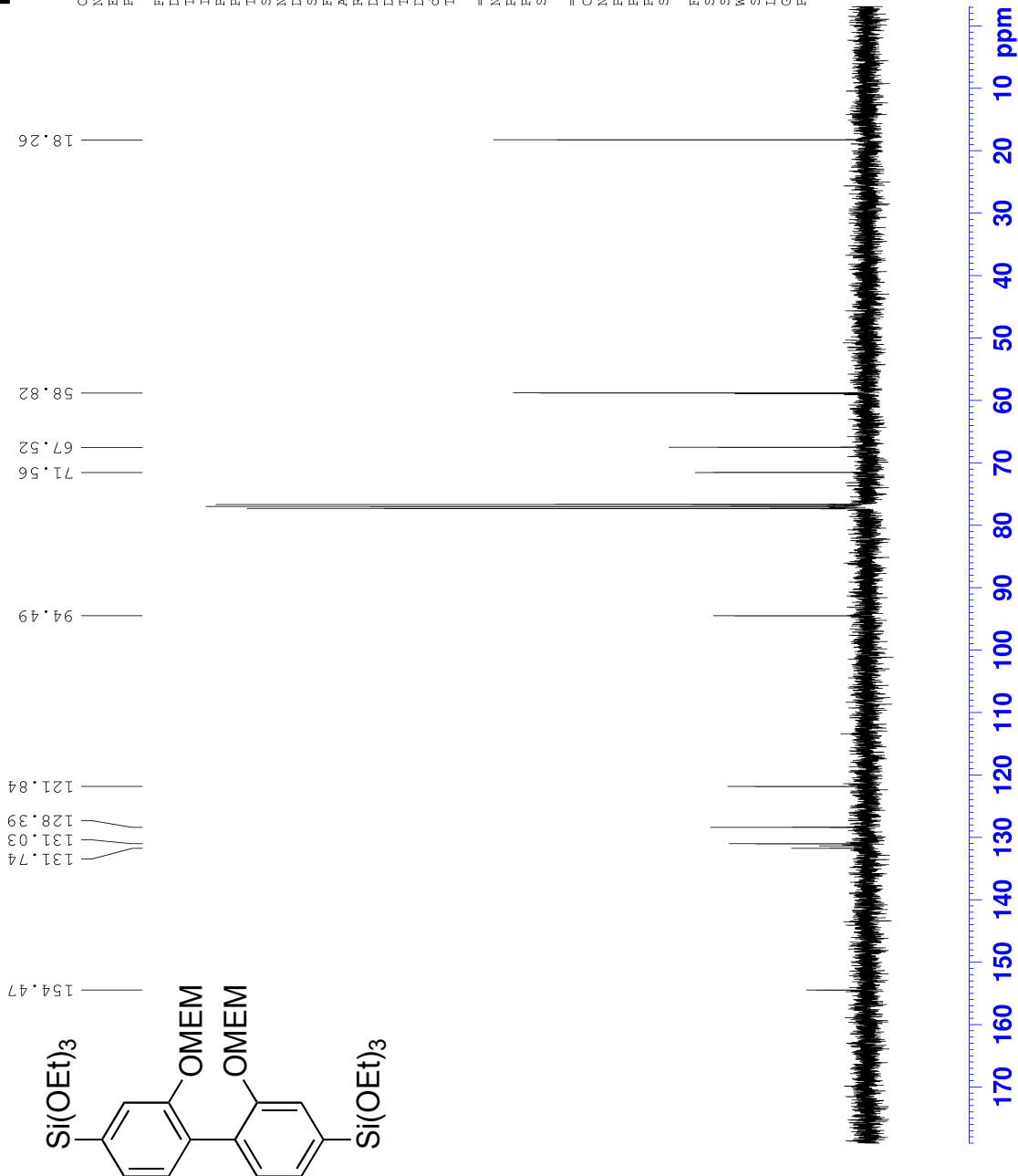
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DE 6.00 us
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d11 0.03000000 sec
TD0 1

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NUC2 1H
PCPD2 100.00 us
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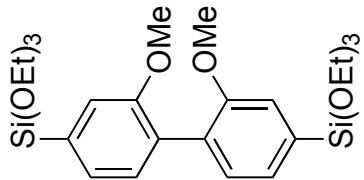
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FIDRES 0.094190 H
AQ 5.3084660 S
RG 1016
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DE 16.000 u
TE 300.0 K
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TD0 1

===== CHANNEL f1 =====
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PL1 0.60 G
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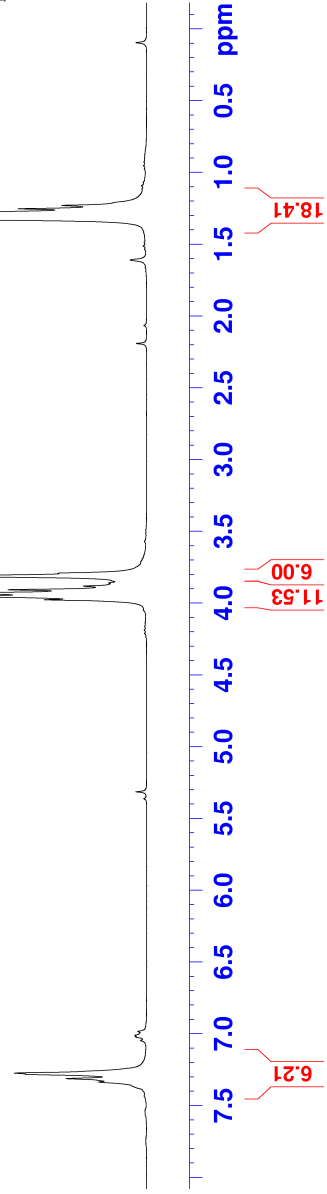
Proton 300 MHz



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7.312
7.280
7.274

3.976
3.953
3.930
3.907
3.812

1.327
1.304
1.281

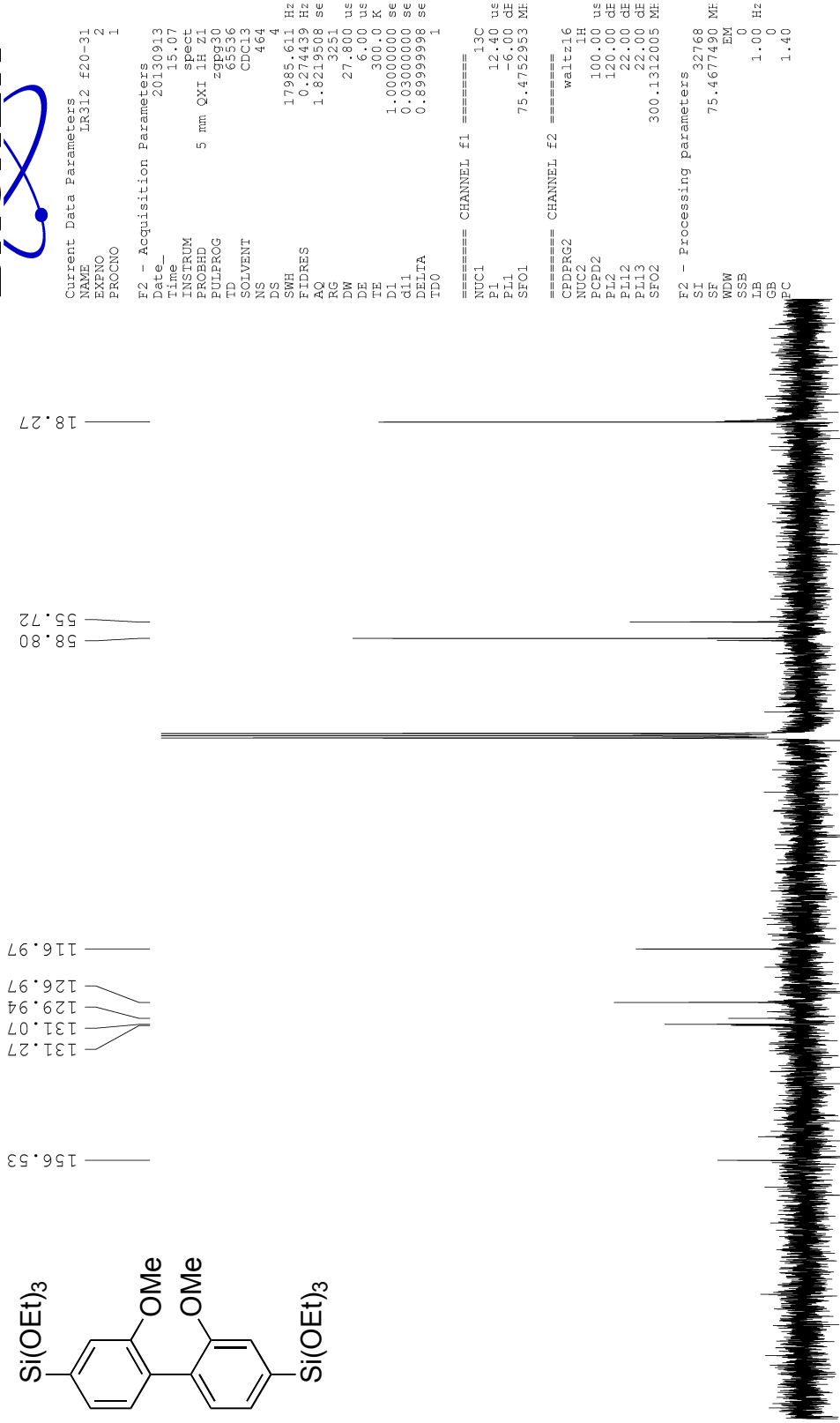
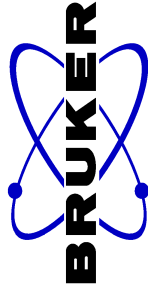
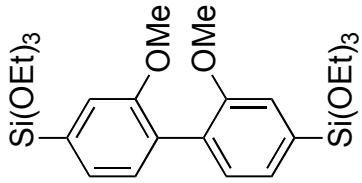


6.21

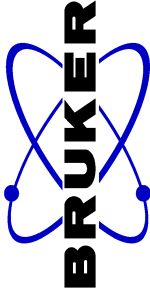
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18.41

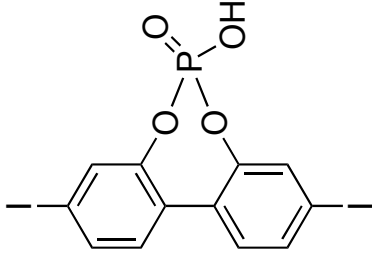
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proton NMR on bbfo probe 400 MHz



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7.362

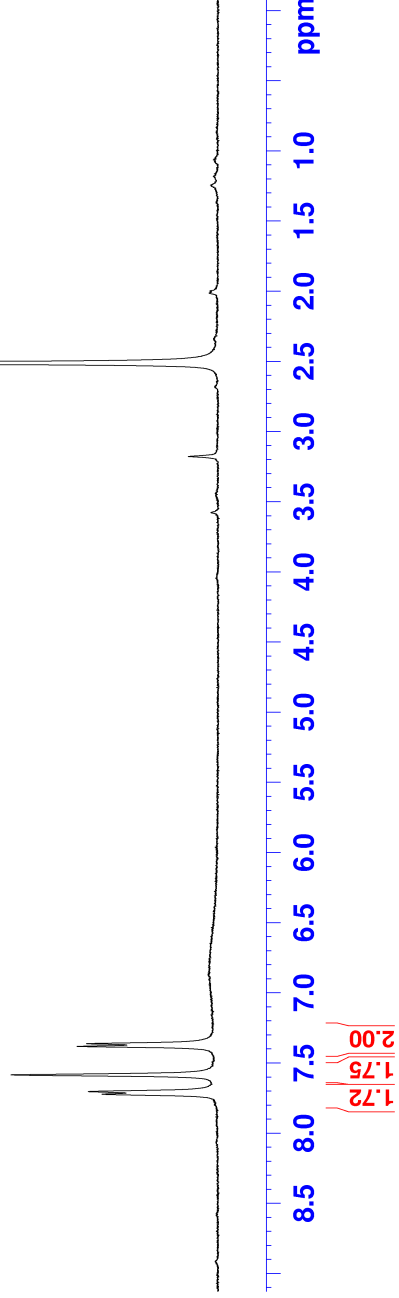


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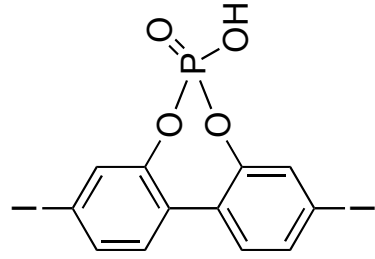
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DE 6.00 u
TE 295.7 K
D1 1.00000000 s
TD0 1

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PL1 -2.00 C
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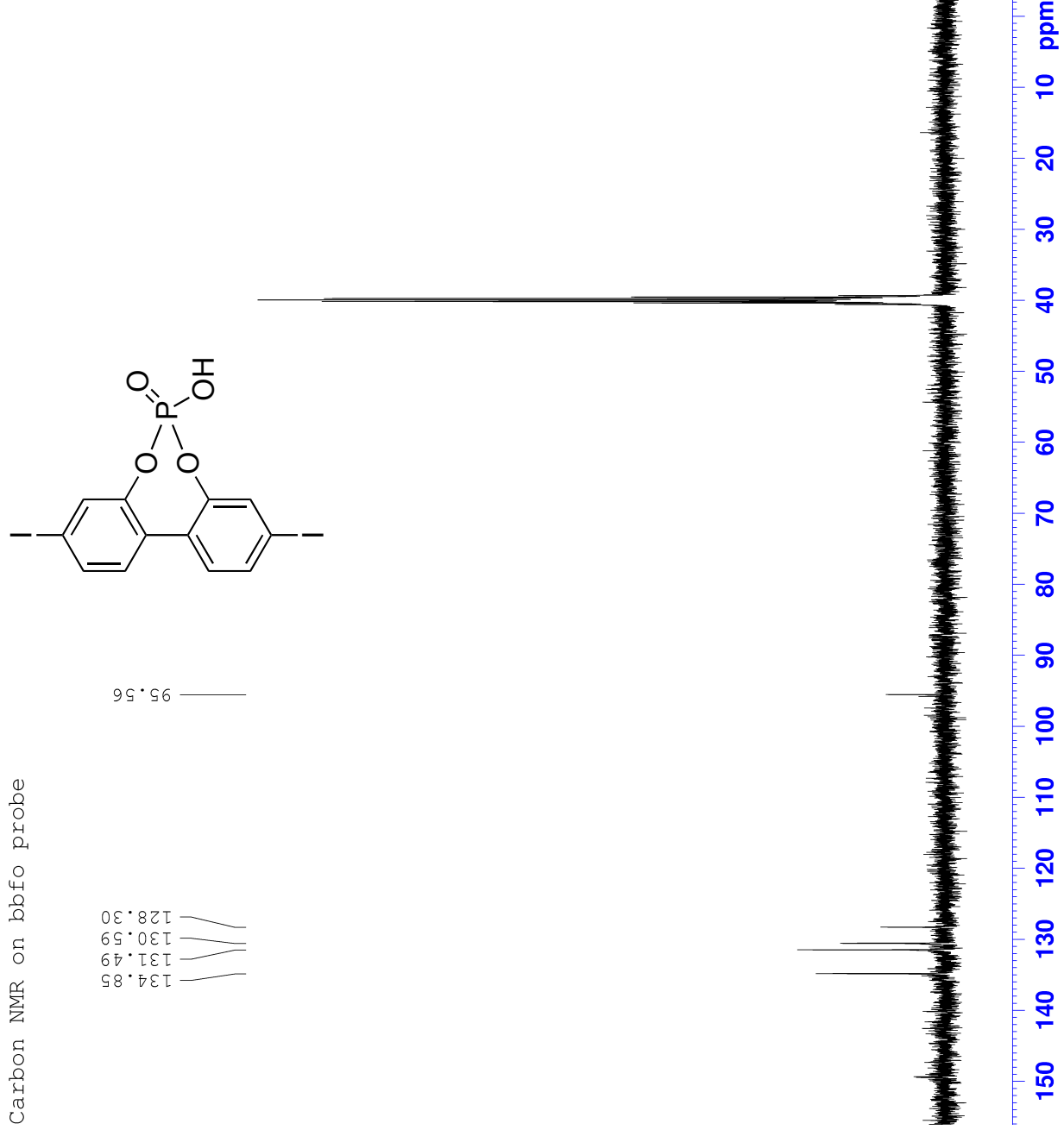


Carbon NMR on bbfo probe



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131.49
130.59
128.30

95.56



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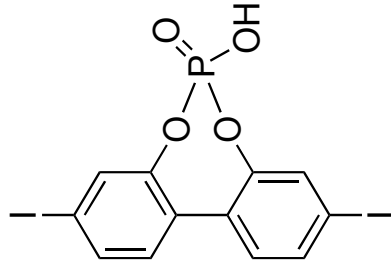
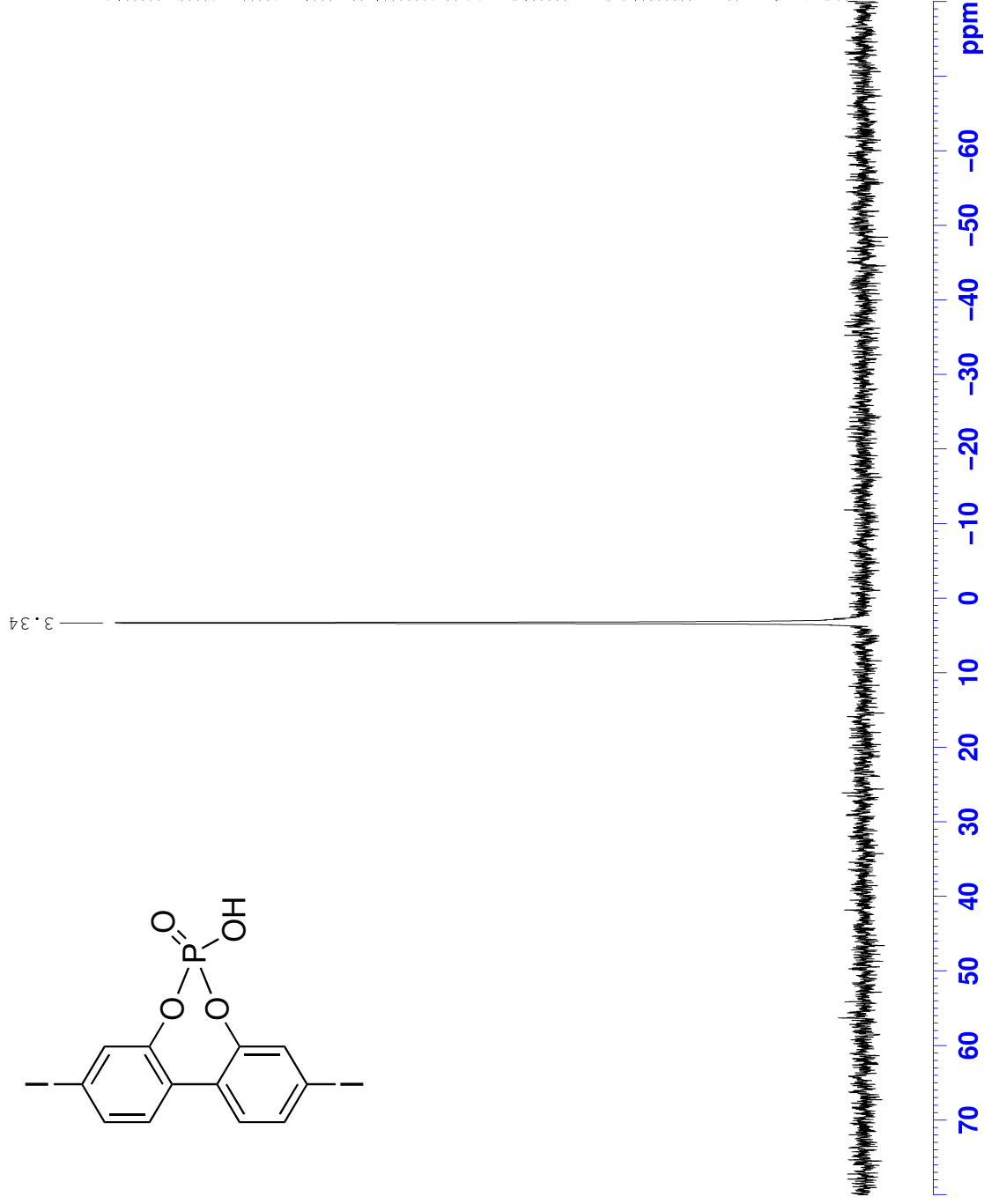
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FIDRES 0.366798 H
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RG 287.4
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DE 6.00 u
TE 296.1 K
D1 0.2000000 s
d11 0.0300000 s
TD0 1

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PL1 -3.00 c
SFO1 100.6228298 M

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Phosphorus NMR on bbfo probe



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PLI2       14.48 dB
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