

Supplementary Information for:
**Bioconjugatable, PEGylated Hydroporphyrins for Photochemistry and
 Photomedicine. Narrow-Band, Near-Infrared-Emitting Bacteriochlorins**

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1. Scale-Up Procedures

Several bacteriochlorins were prepared at somewhat larger scale than in the procedures described in the body of the paper. Bacteriochlorin **B2** was prepared in a larger scale (~40 mg) starting from bacteriochlorin precursor **10**. The Suzuki coupling reaction of bacteriochlorin **1** (270 mg) with **9** afforded **10** (465 mg, 90% yield, to be compared to 85% yield for the smaller scale synthesis). A portion of the sample of bacteriochlorin **10** (200 mg) was consumed for the reactions of 15-bromination and 15-acylation (Suzuki coupling) to afford bacteriochlorin **13** (60 mg, 25% yield; to be compared to a 31% yield for the smaller scale synthesis). The subsequent deprotection (97%) and PEGylation (77%) were conducted to afford the target bacteriochlorin **B2** (~40 mg). **B4** was also prepared in a larger scale (20 mg). Summaries and comparisons of the smaller- and larger-scale synthesis for **B2** and **B4** are listed in Table S1.

Table S1. Summaries and comparisons for the smaller- and larger-scale synthesis of **B2** and **B4**.

Cmpd	Smaller scale			Larger scale		
	Purification ^a	Time	Total yield	Purification ^a	Time	Total yield
B2	Column chromatography (2 steps), solvent wash (2 steps)	5 d	22% ^b	Column chromatography (2 steps), solvent wash (2 steps)	7 d	17% ^b
B4	Column chromatography (3 steps), solvent wash (2 steps)	6 d	3% ^c	Column chromatography (3 steps), solvent wash (2 steps)	8 d	6% ^c

^aThe aqueous/organic wash is not included. ^bCalculated from four continuous reaction steps for the conversion of **10** to **B2**. ^cCalculated from four continuous reaction steps for the conversion of **15** to **B4**.

Synthesis protocols:

Bacteriochlorin B2. A mixture of **14** (25.0 mg, 30.5 μmol), Cs_2CO_3 (200 mg, 613 μmol) and **PEG₄-NHS** (410 mg, 1.23 mmol) in DMF (375 μL) was stirred under argon for 2.5 h. The reaction mixture was diluted with saturated aqueous NaHCO_3 (2.5 mL) and stirred for 4 h. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (40.0 mg, 77% yield).

Bacteriochlorin B4. A solution of **18** (17.0 mg, 13.0 μmol) in CH_2Cl_2 (2.80 mL) was stirred under argon for 2 min, followed by addition of TFA (0.280 mL). After 1 h, the reaction mixture was diluted with CHCl_3 (~3 mL) and dried under vacuum to thoroughly remove the CH_2Cl_2 and TFA. This procedure (CHCl_3 addition – solvent removal) was repeated twice, whereupon the mixture was dried under vacuum. Tributylamine (20 μL) was added to the solid residue, and the resulting suspension was sonicated for 3 min. A mixture of THF and hexanes (1:2) was then added, and the resulting suspension was sonicated (5 min) and centrifuged. The supernatant was discarded to afford a greenish solid. A mixture of the resulting bacteriochlorin (11.8 mg, 13.0 μmol), Cs_2CO_3 (20.0 mg, 64.7 μmol , 5 equiv) and **PEG₄-NHS** (134 mg, 0.410 mmol, 34 equiv) in DMF (2.8 mL) was stirred under argon for 2 h. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (20.0 mg, 86% yield).

Bacteriochlorin 13. A solution of **10** (200 mg, 187 μmol) in THF (375 mL) was treated with NBS (40.0 mg, 224 μmol) in THF (2.25 mL) at room temperature for 1.5 h. The reaction mixture was diluted with CH_2Cl_2 and washed with saturated aqueous NaHCO_3 . The organic layer was dried (Na_2SO_4) and concentrated to afford a red solid, which was transferred to a Schlenk flask. Samples of **12** (154 mg, 0.463 mmol), $\text{Pd}(\text{PPh}_3)_4$ (43.6 mg, 37.7 μmol), and Cs_2CO_3 (187 mg, 0.572 mmol) were placed in the Schlenk flask, and the mixture was dried and deaerated under high vacuum for 30 min. Toluene/DMF [9.5 mL (2:1), deaerated by bubbling with argon for 30 min] was added to the Schlenk flask under argon, and the resulting mixture was deaerated by three freeze-pump-thaw cycles. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (60.0 mg, 25% yield).

Bacteriochlorin 14. A solution of **13** (50.0 mg, 39.2 μmol) in CH_2Cl_2 (2.35 mL) was stirred under argon for 2 min followed by the addition of TFA (1.51 mL). After 1 h, the reaction mixture was diluted with CHCl_3 and dried under an argon flow. Tributylamine (642 μL , 2.70 mol) was added to the solid residue, and the mixture was sonicated for 3 min. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (31.0 mg, 97% yield).

Bacteriochlorin 15. A mixture of bacteriochlorin **2** (298 mg, 443 μmol), **9** (450 mg, 964 μmol), $\text{Pd}(\text{PPh}_3)_4$ (306 mg, 265 μmol), Cs_2CO_3 (432 mg, 1.33 mmol) and toluene/DMF [45 mL

(2:1), deaerated by bubbling with argon for 1 h] was added to a Schlenk flask and deaerated by three freeze-pump-thaw cycles. The remaining reaction protocol, and purification procedure and characterization data (ESI-MS were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (326 mg, 62% yield).

Bacteriochlorin 16. A solution of **15** (326 mg, 275 μmol) in THF (63.0 mL) was treated with NBS (54.0 mg, 303 μmol) in THF (3.00 mL) at room temperature under argon for 40 min. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (250 mg, 72% yield).

Bacteriochlorin 18. A mixture of **16** (250 mg, 198 μmol), $\text{Pd}_2(\text{dba})_3$ (75.0 mg, 90.0 μmol , 0.45 equiv) and $\text{P}(o\text{-tol})_3$ (150 mg, 475 μmol , 2.4 equiv) was dried under high vacuum in a Schlenk flask for 1 h. Toluene/DMF [19.5 mL (2:1), deaerated by bubbling with argon for 30 min] was added to the Schlenk flask under argon, and the resulting mixture was deaerated by three freeze-pump-thaw cycles. 6-Heptynoic acid (**17**, 250 μL , 2.00 mmol, 10 equiv) was added after the second freeze-pump-thaw cycle. The remaining reaction protocol, purification procedure and characterization data (ESI-MS data were not obtained) were essentially identical with those in the body of the paper to afford the title bacteriochlorin (45.0 mg, 17% yield). A mixture of debrominated bacteriochlorin **15** and starting bacteriochlorin **16** (1:1 determined by ^1H NMR spectroscopy, ~ 130 mg) was recovered after chromatography.

2. Exploratory Syntheses Towards PEGylated Bacteriochlorins

Several other target bacteriochlorins were also designed with expectation to tune the wavelength or to explore complementary synthetic methodologies. Three target compounds shown in Chart S1 were attempted. None of these target compounds was obtained due to synthetic difficulties encountered with the intermediates. Here, the “S” terminology is included to denote that these compounds are part of the Supplementary Information section and do not appear in the body of the paper.

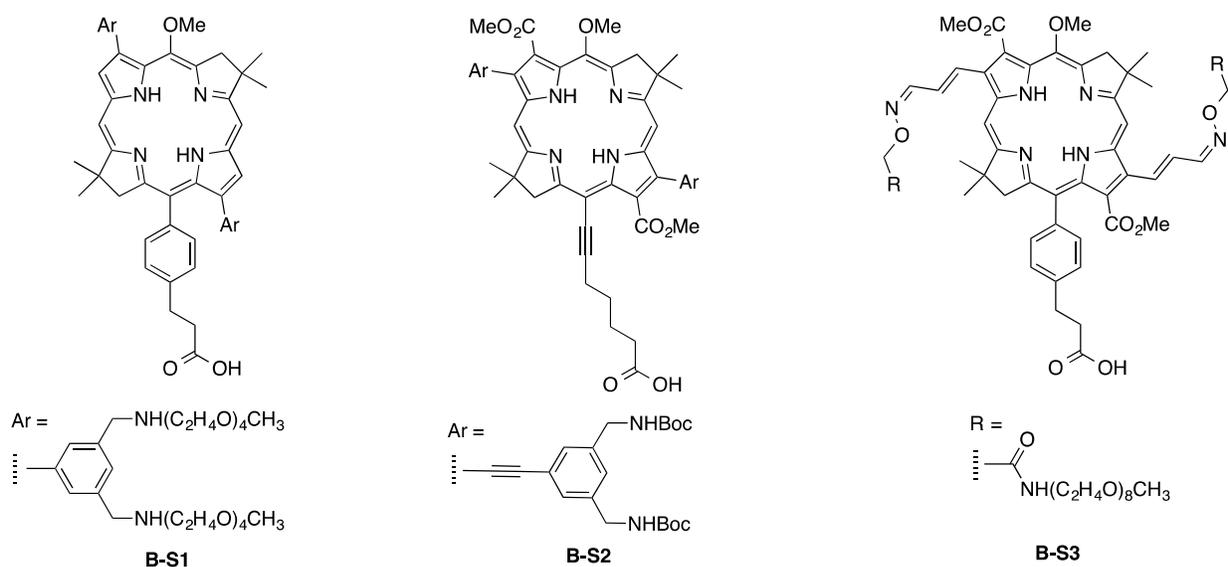
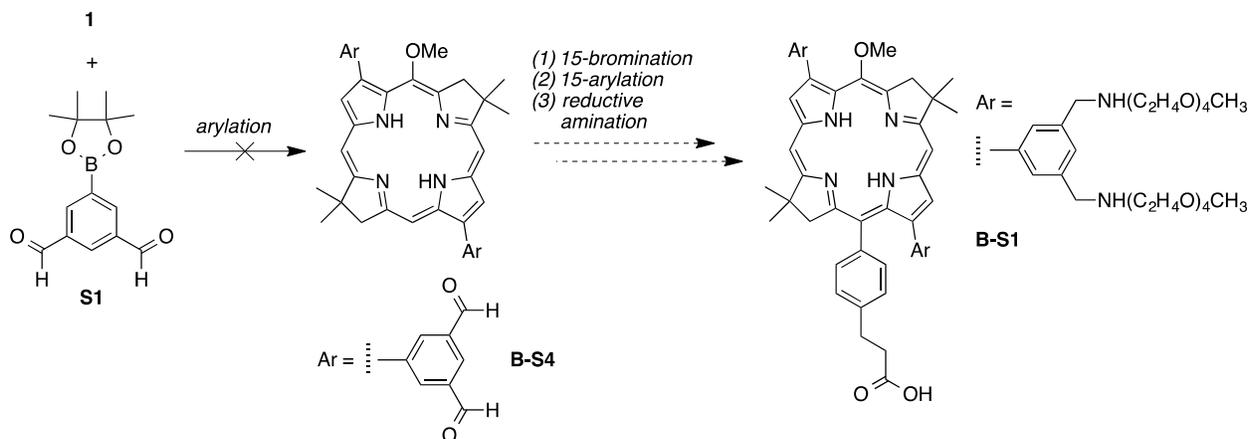


Chart S1. Attempted target bacteriochlorins.

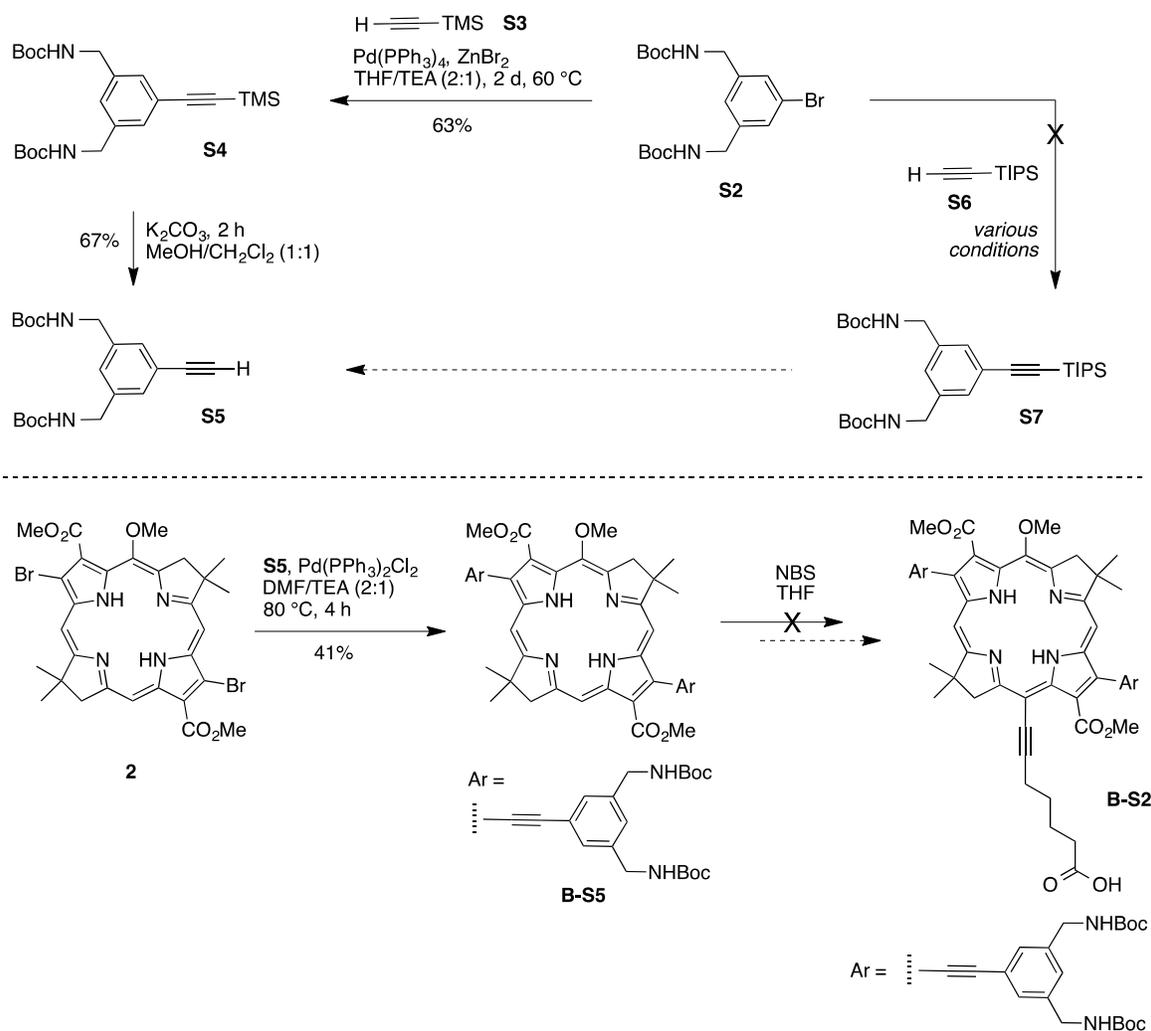
Bacteriochlorin **B-S1** was designed with expectation of a Q_y absorption band at ~ 730 nm. The Suzuki coupling reaction of bacteriochlorin **1** with compound **S1**¹¹ failed to afford the tetraformylbacteriochlorin **B-S4** (Scheme S1). This was a surprise, since a similar Suzuki coupling reaction with 15-bromo-2,12-bis(4-methylphenyl)bacteriochlorin using the same coupling partner successfully afforded the corresponding 15-(3,5-diformyl)-2,12-bis(4-methylphenyl)bacteriochlorin in 81% yield.¹¹ This synthesis failure may stem from a difference in reactivity of the bacteriochlorin *meso* (15-) and β (3- and 13-) positions.¹¹



Scheme S1. Attempted synthesis of a PEGylated bacteriochlorin.

Previous studies revealed that attachment of 3,13-diethynyl groups shifted the Q_y absorption band of the bacteriochlorin to 749 nm, to be compared with 729 nm for the parent 3,13-dibromobacteriochlorin.³⁷ In a separate study,¹¹ a bathochromic shift of 18 nm was achieved by incorporating an ethynyl group at the 15-position of a 2,12-di-*p*-tolyl-5-methoxybacteriochlorin. Accordingly, incorporation of three ethynyl groups (2-, 12-, and 15-positions) with a 3,13-dicarboethoxybacteriochlorin was expected to cause a bathochromic shift of the Q_y band from 749 nm to ~ 795 nm.

Treatment of Boc-protected 3,5-diamino-1-bromobenzene **S2** with trimethylsilylacetylene (**S3**) under Pd-catalysis conditions afforded **S4** in 63% yield (Scheme S2).^{S1} The resulting compound **S4** was subjected to trimethylsilyl (TMS) group removal by potassium carbonate in a mixed solvent of methanol and dichloromethane^{S2} to give the Sonogashira coupling partner **S5** in 67% yield. The use of triisopropylsilylacetylene (**S6**) instead of trimethylsilylacetylene (**S3**) failed to afford the desired product **S7** under various conditions. All conditions examined for both routes are summarized in Table S2.



Scheme S2. Attempted synthesis of a PEGylated bacteriochlorin.

Table S2. Reaction conditions examined for the Sonogashira reaction.

Entry	Reaction conditions ^{a,b}	Observed	Detected by
1	Pyridine, Pd(PPh ₃) ₂ Cl ₂ (0.01 eq), S6 (2 eq), CuI (0.2 eq), 80 °C, 16 h	unknown	¹ H NMR
2	THF/TEA (3:2), Pd(PPh ₃) ₂ Cl ₂ (0.05 eq), S6 (3 eq), CuI (0.4 eq), 80 °C, 16 h	unknown	¹ H NMR
3	Pyridine, Pd(PPh ₃) ₂ Cl ₂ (0.05 eq), S3 (3 eq), CuI (0.4 eq), 80 °C, 36 h	unknown	¹ H NMR
4	THF/ <i>i</i> -Pr ₃ N (3:2), Pd(PPh ₃) ₂ Cl ₂ (0.05 eq), S3 (3 eq), CuI (0.4 eq), 80 °C, 36 h	unknown	¹ H NMR
5	DMF/TEA (3:2), Pd(PPh ₃) ₂ Cl ₂ (0.05 eq), S3 (3 eq), CuI (0.4 eq), 80 °C, 24 h	unknown	¹ H NMR
6	THF/TEA (2:1), Pd(PPh ₃) ₄ (0.03 eq), S3 (3 eq), ZnBr ₂ (0.67 eq), 60 °C, 2 d	product S4	¹ H NMR, ¹³ C NMR, ESI-MS

^aConcentrations of S2 are 0.40 M for entry 1, 0.24 M for entries 2–5, and 0.30 M for entry 6. ^bAll equivalents in the brackets are relative to compound S2.

Sonogashira coupling reaction of bacteriochlorin **2** with coupling partner S5 under copper-free Sonogashira conditions^{11,35} gave Boc-protected bacteriochlorin B-S5 in 41% yield (Scheme S2). Treatment of bacteriochlorin B-S5 with NBS, however, failed to give the 15-brominated bacteriochlorin under various neutral and basic conditions, with the starting bacteriochlorin B-S5 unreacted or decomposed (Table S3).

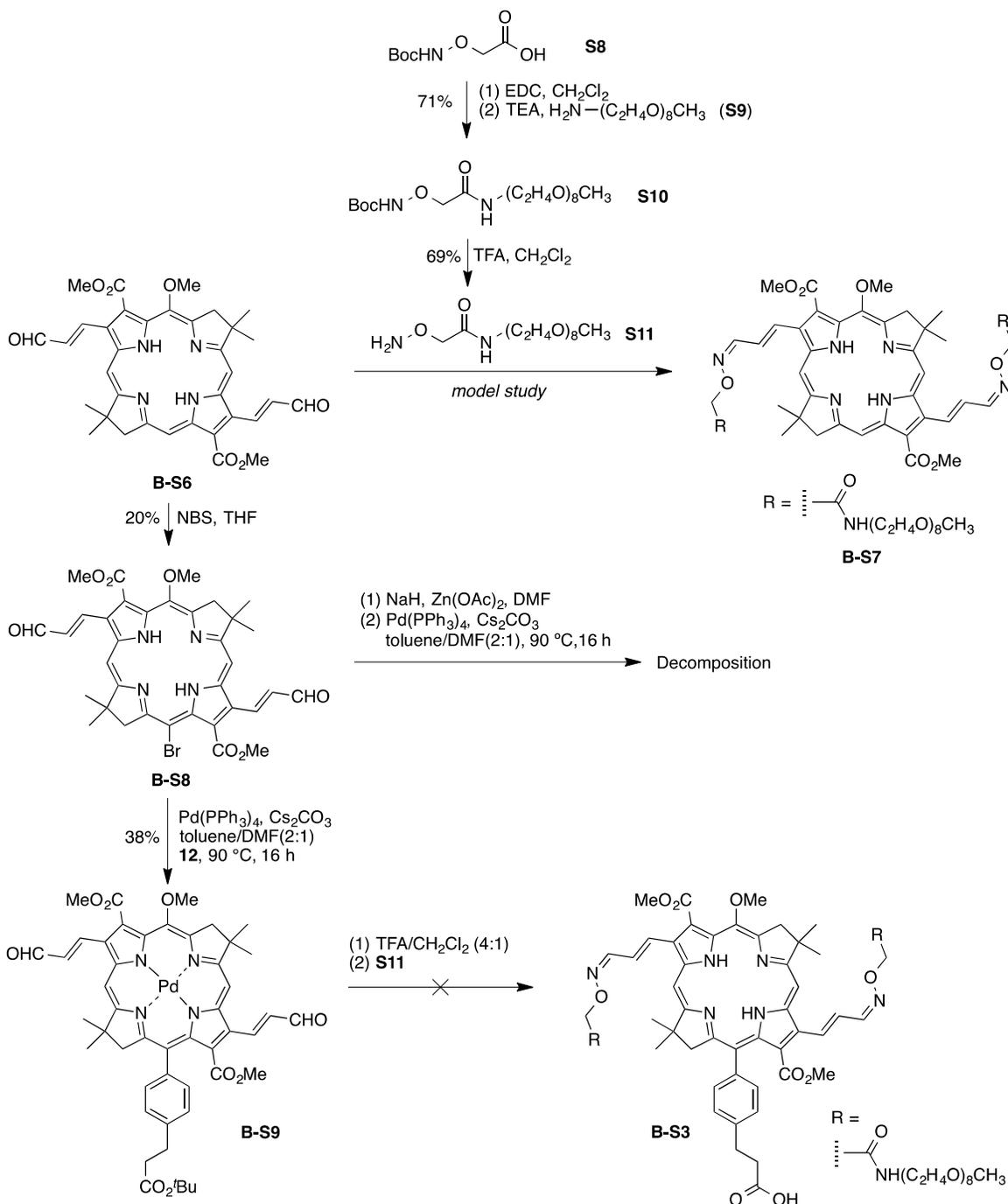
Table S3. Reaction conditions for 15-bromination.

Entry	Reaction condition ^a	Observation ^b
1	NBS (1.0 eq), THF, 40 min, rt	SM ^c
2	NBS (2.0 eq), THF, 2 h, rt	SM ^c
3	NBS (3.0 eq), THF, 20 min, rt	SM ^c
4	NBS (6.0 eq), THF, 10 min, rt	decomposed
5	NBS (1.0 eq), CH ₃ CN, 16 h, rt	SM ^c
6	NBS (1.1 eq), CH ₂ Cl ₂ , pyridine (150 eq), 50 min, rt	SM ^c

^aThe concentration of B-S5 in THF was 4 mM for entries 1–5 and 5.5 mM for entry 6. A stock solution of NBS (0.1 M in THF) was used for all entries. ^bConfirmed by MALDI-MS and absorption spectroscopy. ^cStarting bacteriochlorin B-S5 was detected.

Bacteriochlorin B-S3 bearing an extended π -electron conjugation was designed to tune the wavelength to ~800 nm. The PEG unit was incorporated at the final step via Schiff's base formation. This route entailed the preparation of a PEGylated aminoxy compound S11 as described below (Scheme S3): (i) EDC-mediated coupling reaction of a Boc-protected aminoxy

carboxylic acid (**S8**) with PEG₈-amine **S9** to afford compound **S10** in 71% yield, and (ii) cleavage of the Boc group to give **S11** in 69% yield. Both **S10** and **S11** were used directly for the next step without column chromatography purification.



Scheme S3. Synthesis of a PEGylated bacteriochlorin.

A model study of the reactivity of the 3-oxopropenyl substituent toward the aminoxy group was examined with known bacteriochlorin **B-S6**^{S3} and **S11**. The resulting mixture consisted of the starting bacteriochlorin **B-S6**, mono-oxime (structure not shown) and di-oxime product **B-S7** (Scheme S3 and Table S4). Variation of the reaction conditions did not improve

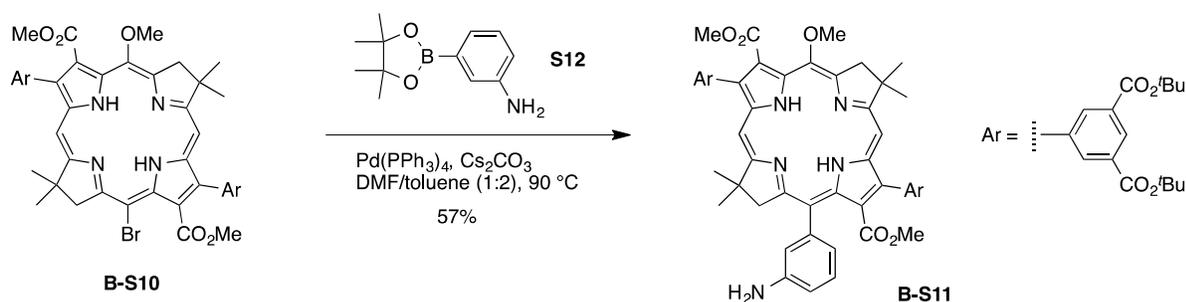
the product yield (Table S4). Silica column chromatography was not effective in separating the three components, either. A second problem for this route was the unexpected Pd-insertion during the Suzuki coupling reaction at the bacteriochlorin 15-position for the introduction of a bioconjugatable tether. The Pd-metallated bacteriochlorin is undesirable because of a short-lived fluorescence lifetime, precluding any further photophysical studies or applications. Treatment with 80% TFA in dichloromethane was not successful in removing the chelated palladium. By contrast with the Pd-metallated bacteriochlorin, zinc bacteriochlorins typically exhibit satisfactory fluorescence properties. Accordingly, the bacteriochlorin was subjected to zinc metalation, following by Suzuki coupling at the 15-position; however, the bacteriochlorin decomposed during the Suzuki coupling step as determined by MALDI-MS and absorption spectroscopy. In short, derivatization of the 3-oxopropenyl substituent was not achieved.

Table S4. Reaction conditions for oxime formation.^a

Entry	Catalyst	Solvent ^c	Temp.	Results ^d
1	aniline (100 eq)	CH ₂ Cl ₂ /DMSO/NH ₄ OAc = 2:1:0.2	rt	1:1:1
2	aniline (500 eq)	CH ₂ Cl ₂	rt to 60 °C	1:2:3
3	N/A	CH ₂ Cl ₂ /C ₂ H ₅ OH = 1:1	80 °C	1:1:1
4 ^b	N/A	NH ₄ OAc/CH ₂ Cl ₂ = 1:1	rt	1:2:3

^a10 mM of bacteriochlorin **B-S6** and ~20 equiv of **S11**. ^bStarting bacteriochlorin is the product mixture from entry 2. ^cNH₄OAc buffer: pH 4.5, 100 mM. ^d**B-S6**/mono-oxime/**B-S7**.

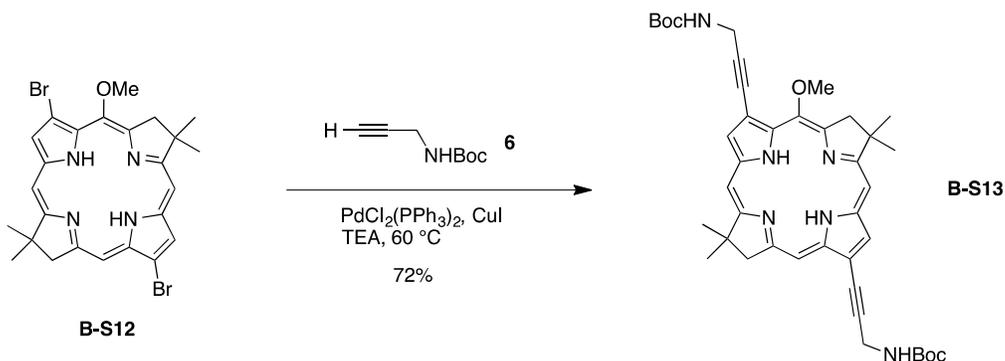
Bacteriochlorin **B-S11** was initially designed as an intermediate for elaboration to a water-soluble bioconjugatable bacteriochlorin (Scheme S4). The Suzuki coupling reaction of dibromobacteriochlorin **B-S10**³⁸ and **S12** proceeded smoothly. The position of the Q_y absorption band of **B-S11** (742 nm) was quite close to several other bacteriochlorins,³⁸ however, so further exploration was not pursued.



Scheme S4. Synthesis of a versatile bacteriochlorin building block.

Bacteriochlorin **B-S13** was designed to prepare a water-soluble bacteriochlorin by quaternization of two primary amines (Scheme S5). The Sonogashira coupling reaction of dibromobacteriochlorin **B-S12**¹² and ethyne **6** proceeded smoothly. However, studies in

parallel revealed that even four ammonium units on a bacteriochlorin did not afford the desirable water-solubility.¹⁵



Scheme S5. Synthesis of a versatile bacteriochlorin building block.

Non-commercial compounds. Compounds **S1**,¹¹ **S2**,^{S4} **B-S6**,^{S3} **B-S10**,³⁸ and **B-S12**¹² were prepared as described in the literature.

***tert*-Butyl 6-heptynoate.** A mixture of 6-heptynoic acid (**17**, 1.0 g, 7.9 mmol), di-*tert*-butyl dicarbonate (2.3 g, 10 mmol) and MgCl₂ (75 mg, 0.79 mmol) in *tert*-butyl alcohol (4.0 mL) was stirred under argon for 16 h. The crude reaction mixture was diluted with water and extracted with ethyl acetate. The organic extract was dried (Na₂SO₄), concentrated and chromatographed [silica, hexanes/ethyl acetate (24:1)] to afford a viscous colorless liquid (0.98 g, 68%): ¹H NMR (400 MHz, CDCl₃) δ 1.48 (s, 9H), 1.50–1.59 (m, 2H), 1.67–1.74 (m, 2H), 1.95–1.96 (m, 1H), 2.19–2.26 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 18.4, 24.4, 28.1, 28.3, 35.2, 68.7, 80.3, 84.3, 173.0; ESI-MS obsd 205.1199, calcd 205.1196 [(M + Na)⁺, M = C₁₁H₁₈O₂].

2-[3,5-Bis(*tert*-butoxycarbonylaminoethyl)phenyl]-1-trimethylsilylacetylene (S4**).** Following a reported procedure,^{S1} a mixture of **S2** (1.0 g, 2.4 mmol), Pd(PPh₃)₄ (80 mg, 0.07 mmol, 0.03 equiv) and ZnBr₂ (0.40 g, 1.8 mmol, 0.67 equiv) was dried and deaerated under high vacuum for 30 min in a Schlenk flask. A mixture of THF/triethylamine [12 mL (2:1), bubbled with argon for 30 min] was added, and the resulting mixture was deaerated by two freeze-pump-thaw cycles. Trimethylsilylacetylene (**S3**, 1.0 mL) was added, and the reaction mixture was subjected to a third freeze-pump-thaw cycle. The reaction mixture was stirred at 60 °C for 2 days. The resulting mixture was allowed to cool to room temperature, washed with brine and water, and dried over Na₂SO₄. After removal of the solvent under vacuum, hexanes (10 mL) was added. The resulting suspension was sonicated (2 min) and centrifuged. The supernatant was discarded to afford a brown solid (0.70 g, 63%): mp 169–170 °C; ¹H NMR (400 MHz, CDCl₃) δ 0.24 (s, 9H), 1.46 (s, 18H), 4.28 (s, 4H), 4.84 (s, 2H), 7.16 (s, 1H), 7.30 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 28.5, 44.2, 80.0, 95.0, 104.5, 124.0, 127.1, 130.2, 140.2, 156.3; ESI-MS obsd 455.2335, calcd 455.2337 [(M + Na)⁺, M = C₂₃H₃₆N₂O₄Si].

[3,5-Bis(*tert*-butoxycarbonylaminoethyl)phenyl]acetylene (S5**).** Following a reported procedure,^{S2} a mixture of **S4** (0.66 g, 1.5 mmol) and K₂CO₃ (0.50 g, 3.6 mmol, 2.4 equiv) in MeOH/CH₂Cl₂ (6.6 mL, 1:1) was stirred at room temperature for 2 h. Water was added to the reaction mixture, and the resulting solution was extracted with CH₂Cl₂. The combined extract was washed with water, dried (Na₂SO₄), concentrated under vacuum, and

chromatographed [silica, CH₂Cl₂/ethyl acetate (10:1 to 5:1)] to yield a pale white solid (0.37 g, 67%): mp 132–133 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.45 (s, 18H), 3.05 (s, 1H), 4.26 (d, *J* = 5.6 Hz, 4H), 4.92 (s, 2H), 7.16 (s, 1H), 7.29 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 29.0, 44.2, 79.2, 80.0, 83.8, 123.0, 127.2, 130.2, 140.3, 156.3; ESI-MS obsd 383.1939, calcd 383.1941 [(M + Na)⁺, M = C₂₀H₂₈N₂O₄].

2-*tert*-Butoxycarbonylaminoxy-*N*-(2,5,8,11,14,17,20,23-octaoxapentacosan-25-yl)-acetamide (S10). A mixture of carboxylic acid **17** (287 mg, 1.50 mmol) and EDC (288 mg, 1.50 mmol) in CH₂Cl₂ (3.00 mL) was stirred for 5 min, followed by addition of amine **S9** (575 mg, 1.5 mmol) and TEA (209 μL, 1.50 mmol). The reaction mixture was stirred at room temperature for 16 h. The reaction mixture was diluted with CH₂Cl₂ and washed one time with saturated NaHCO₃ solution and then with 2 N HCl solution. The organic phase was separated and dried (MgSO₄) to afford a colorless oil (595 mg, 71%): ¹H NMR (300 MHz, CDCl₃) δ 1.48 (s, 9H), 3.38 (s, 3H), 3.50–3.66 (m, 32H), 4.33 (s, 2H), 7.92 (br, 1H), 8.32 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 28.4, 39.1, 59.2, 69.7, 70.2, 70.6, 70.69, 70.74, 72.1, 76.0, 82.5, 157.7, 169.3; ESI-MS obsd 557.3272, calcd 557.3280 [(M + H)⁺, M = C₂₄H₄₈N₂O₁₂].

2-Aminoxy-*N*-(2,5,8,11,14,17,20,23-octaoxapentacosan-25-yl)acetamide (S11). A sample of **S10** (162 mg, 0.290 mmol) in CH₂Cl₂ (1.00 mL) was stirred under argon for 2 min, followed by the addition of TFA (1.00 mL). After 1 h, the solvent was removed with a stream of argon, and the reaction residue was washed with saturated aqueous NaHCO₃. The organic layer was separated, dried (Na₂SO₄) and concentrated to afford a colorless oil (92.0 mg, 69%): ¹H NMR (300 MHz, CDCl₃, the two amine protons were not observed) δ 3.38 (s, 3H), 3.51–3.66 (m, 31H), 4.14 (s, 2H), 5.95 (s, 1H), 6.99 (br, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 38.8, 59.3, 69.9, 70.2, 70.5, 70.8, 72.2, 75.2, 76.3, 170.4; ESI-MS obsd 457.2756, calcd 457.2756 [(M + H)⁺, M = C₁₉H₄₀N₂O₁₀].

3,13-Dimethoxycarbonyl-2,12-bis[2-(3,5-bis(*tert*-butoxycarbonylaminoethyl)-phenyl)ethynyl]-5-methoxy-8,8,18-tetramethylbacteriochlorin (B-S5). Following a reported procedure,³⁵ a mixture of **2** (4.0 mg, 3.2 μmol), Pd(PPh₃)₂Cl₂ (1.0 mg, 1.5 μmol, 0.20 equiv) and **S5** (8.0 mg, 24 μmol, 4.0 equiv) was dried and deaerated under high vacuum in a Schlenk flask for 1 h. DMF (0.80 mL) and triethylamine (0.40 mL) (bubbled with argon for 1 h for each) were added, and the resulting mixture was deaerated by three freeze-pump-thaw cycles. The reaction mixture was stirred at 80 °C for 4 h. The reaction mixture was allowed to cool to room temperature and then washed with brine and water. The organic layer was separated, dried (Na₂SO₄), concentrated to dryness and chromatographed [silica, CH₂Cl₂/ethyl acetate (5:1 to 2:1)] to afford a red solid (3.0 mg, 41%): ¹H NMR (300 MHz, CDCl₃) δ -1.18 (s, 1H), -0.89 (s, 1H), 1.52 (s, 36H), 1.95 (s, 6H), 1.97 (s, 6H), 4.25 (s, 3H), 4.32–4.34 (m, 9H), 4.45 (s, 8H), 5.04 (s, 4H), 7.31 (s, 2H), 7.65 (s, 2H), 7.70 (s, 2H), 8.77 (s, 1H), 8.92 (s, 1H), 9.55 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 28.7, 31.0, 31.2, 44.6, 46.0, 46.3, 47.9, 51.6, 52.4, 53.5, 64.6, 80.0, 82.9, 85.5, 95.8, 97.5, 98.9, 99.2, 101.0, 115.5, 118.0, 120.7, 124.0, 124.5, 127.0, 127.2, 128.8, 130.0, 135.7, 135.7, 135.9, 137.2, 140.3, 156.2, 157.9, 162.8, 165.8, 167.8, 170.6, 173; MALDI-MS obsd 1234.2789; ESI-MS obsd 1233.6212, calcd 1233.6231 [(M + H)⁺, M = C₆₉H₈₄N₈O₁₃]; λ_{abs} (CH₂Cl₂) 383, 553, 791 nm.

15-Bromo-2,12-bis(3-oxoprop-1-enyl)-3,13-dimethoxycarbonyl-5-methoxy-8,8,18-tetramethylbacteriochlorin (B-S8). Following a standard procedure,¹¹ a solution of **B-S6** (5.8 mg, 9.3 μmol) in dry THF (3.1 mL) was treated dropwise (10 min) with a solution of NBS (11 mg, 6.0 equiv) in dry THF (0.56 mL), and the resulting mixture was stirred at room temperature

under argon for 1 h. The reaction mixture was diluted with CH₂Cl₂ and washed with saturated aqueous NaHCO₃. The organic layer was separated, dried (Na₂SO₄) and concentrated. Column chromatography [silica, CH₂Cl₂/ethyl acetate (50:1 to 25:1)] afforded a red solid (1.3 mg, 20%): ¹H NMR (300 MHz, CDCl₃) δ -0.83 (s, 1H), -0.58 (s, 1H), 1.92 (s, 6H), 1.93 (s, 6H), 4.21 (s, 3H), 4.28 (s, 3H), 4.29 (s, 2H), 4.31 (s, 3H), 4.36 (s, 2H), 7.19 (dd, *J* = 16.2, 7.5 Hz, 2H), 8.54 (s, 1H), 8.63 (s, 1H), 8.70 (d, *J* = 16.2 Hz, 2H), 10.07 (t, *J* = 7.5 Hz, 2H); MALDI-MS obsd 702.2023, calcd 703.1762 [(M + H)⁺, M = C₃₅H₃₅BrN₄O₇]; λ_{abs} (CH₂Cl₂) 345, 393, 787 nm.

15-[4-(2-(*tert*-Butoxycarbonyl)ethyl)phenyl]-2,12-bis(3-oxoprop-1-enyl)-3,13-dimethoxy-carbonyl-5-methoxy-8,8,18,18-tetramethylbacteriochlorin (B-S9). Samples of **B-S8** (6.0 mg, 8.5 μmol), Pd(PPh₃)₄ (3.9 mg, 3.4 μmol), Cs₂CO₃ (11 mg, 34 μmol) and **12** (11 mg, 34 μmol) were placed in a Schlenk flask, and the resulting mixture was deaerated under high vacuum for 1 h. Toluene/DMF [0.8 mL (2:1), deaerated by bubbling with argon for 1 h] was added to the Schlenk flask under argon and deaerated by three freeze-pump-thaw cycles. The reaction mixture was stirred at 90 °C for 16 h. The reaction mixture was cooled to room temperature, concentrated to dryness, diluted with ethyl acetate and washed with saturated aqueous NaHCO₃. The organic layer was separated, dried (Na₂SO₄), concentrated and chromatographed [silica, CH₂Cl₂/ethyl acetate (50:1 to 25:1)] to provide a red solid (3.0 mg, 38%): MALDI-MS obsd 931.0782, calcd 933.2685 [(M+H)⁺, M = C₄₈H₅₀N₄O₉Pd]; λ_{abs} (CH₂Cl₂) 339, 380, 561, 800 nm. ¹H NMR spectroscopy was tried, but an informative spectrum was not obtained.

15-(3-Aminophenyl)-2,12-bis[3,5-bis(*tert*-butoxycarbonyl)phenyl]-3,13-dimethoxy-carbonyl-5-methoxy-8,8,18,18-tetramethylbacteriochlorin (B-S11). Following a general procedure,⁵⁵ samples of **B-S10** (9.2 mg, 8.0 μmol), Suzuki coupling partner **S12** (8.8 mg, 40 μmol), Pd(PPh₃)₄ (2.8 mg, 2.4 μmol), K₂CO₃ (13 mg, 96 μmol) and toluene/DMF [0.80 mL (2:1), deaerated by bubbling with argon for 45 min] were added to a Schlenk flask, whereupon the mixture was deaerated by three freeze-pump-thaw cycles. The reaction mixture was stirred at 90 °C for 20 h. The reaction mixture was cooled to room temperature, concentrated to dryness, diluted with CH₂Cl₂ and washed with saturated aqueous NaHCO₃. The organic layer was separated, dried (Na₂SO₄) and concentrated. Column chromatography [twice, silica, CH₂Cl₂/ethyl acetate (47:3)] afforded a red solid (5.5 mg, 62%): ¹H NMR (CDCl₃, 300 MHz) δ -1.40 (br, 1H), -1.09 (br, 1H), 1.64 (s, 18H), 1.66 (s, 18H), 1.74 (s, 3H), 1.77 (s, 3H), 1.83 (s, 3H), 1.85 (s, 3H), 3.52 (s, 3H), 3.86 (s, 2H), 4.20 (s, 3H), 4.30 (s, 3H), 4.37 (s, 2H), 6.93 (d, *J* = 2.4 Hz, 1H), 7.10 (s, 2H), 7.41 (t, *J* = 2.4 Hz, 1H), 8.50 (s, 1H), 8.60 (s, 1H), 8.75 (d, *J* = 1.8 Hz, 2H), 8.86 (s, 1H), 8.88 (s, 1H), 8.90 (d, *J* = 1.8 Hz, 2H), the two pyrrolic protons were not observed; MALDI-MS obsd 1161.8424; ESI-MS MS obsd 1060.5585, calcd 1060.5591 [(M + H)⁺, M = C₆₇H₇₇N₅O₁₃]; λ_{abs} (CH₂Cl₂) 374, 527, 742 nm.

3,13-Bis[3-(*N*-*tert*-butoxycarbonylamino)prop-1-ynyl]-5-methoxy-8,8,18,18-tetramethylbacteriochlorin (B-S13). Following a general procedure,³⁷ samples of **B-S12** (22 mg, 40 μmol), Sonogashira coupling partner **6** (62 mg, 0.40 mmol), PdCl₂(PPh₃)₂ (5.6 mg, 8.0 μmol) and TEA (2.0 mL, deaerated by bubbling with argon for 45 min) were added to a Schlenk flask, whereupon the mixture was deaerated by three freeze-pump-thaw cycles. The reaction mixture was stirred at 60 °C for 18 h. The reaction mixture was cooled to room temperature, concentrated to dryness, diluted with CH₂Cl₂ and washed with saturated aqueous NaHCO₃. The organic layer was separated, dried (Na₂SO₄) and concentrated. Column chromatography [silica, CH₂Cl₂/TEA (99:1 to 49:1)] afforded a green solid (5.5 mg, 72%): ¹H NMR (CDCl₃, 300 MHz) δ -1.86 (br, 1H), -1.66 (br, 1H), 1.56 (s, 9H), 1.57 (s, 9H), 1.92 (s, 12H), 4.38–4.41 (m,

7H), 4.57 (br, 4H), 5.12 (br, 2H), 8.49 (s, 1H), 8.50 (s, 1H), 8.68 (d, $J = 1.5$ Hz, 1H), 8.70 (d, $J = 1.5$ Hz, 1H), 8.80 (s, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 28.7, 31.0, 31.2, 45.7, 45.9, 51.9, 64.7, 78.3, 80.7, 89.6, 92.8, 96.5, 97.3, 97.6, 111.9, 116.5, 125.0, 125.8, 131.4, 134.3, 135.6, 135.9, 138.4, 154.8, 161.4, 170.0, 170.6; MALDI-MS obsd 708.1287; ESI-MS obsd 706.3841, calcd 706.3837 [$\text{M} = \text{C}_{61}\text{H}_{80}\text{N}_8\text{O}_9$]; λ_{abs} (CH_2Cl_2) 354, 365, 376, 518, 743 nm.

3. Additional Photophysics Studies

A relatively low quantum yield of **B5** (0.017) was observed in neat water, compared with a value of 0.075 in DMSO. This diminution upon examination in aqueous solution was also observed with hydrophilic chlorins^{S6} and bacteriochlorins.⁴³ Herein, **B5** was examined in seven different solvents with distinct dielectric constants, and quantum yields were calculated using 2,12-di-*p*-tolyl-5-methoxybacteriochlorin in toluene ($\Phi_f = 0.18$) as the standard. The results are listed in Table S5. The highest Φ_f value (0.082) was obtained in toluene, which is the least polar solvent among those screened, while the lowest Φ_f (0.016) was obtained in water.

Table S5. Solvent effects on the fluorescence properties of **B5**.

Solvent	Dielectric constant	Absorption		Emission		Φ_f
		λ_{max} (nm)	fwhm (nm)	λ_{max} (nm)	fwhm (nm)	
Toluene	2.4	811	33	820	30	0.082
CH_2Cl_2	8.9	810	35	821	30	0.054
CH_3OH	33	806	36	819	38	0.044
DMSO	47	810	34	821	34	0.075
H_2O	80	819	42	831	34	0.017
D_2O	80	819	43	832	36	0.025
Formamide	111	816	39	827	34	0.043

The absorption and emission spectra of **8** in DMF are shown in Figure S2.

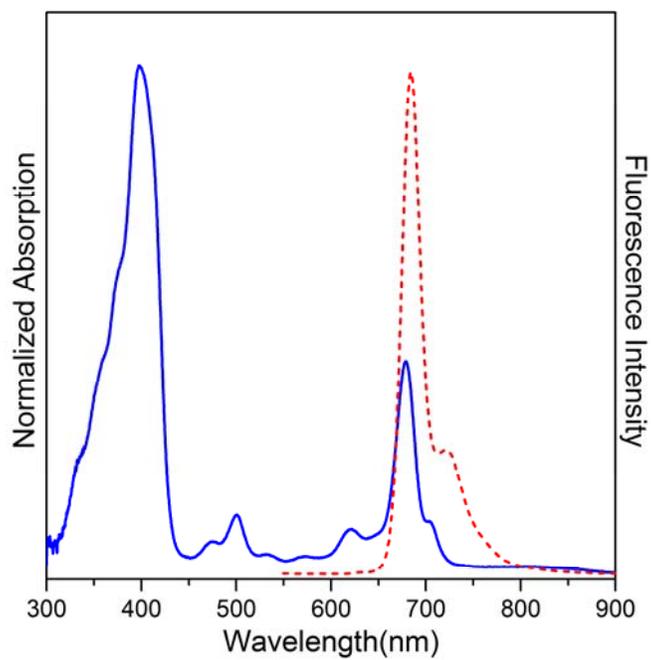


Figure S2. Absorption (solid) and fluorescence (dashed) spectra for **8** in DMF. The fluorescence was obtained using excitation at the Soret maximum.

4. Flow Cytometry Data

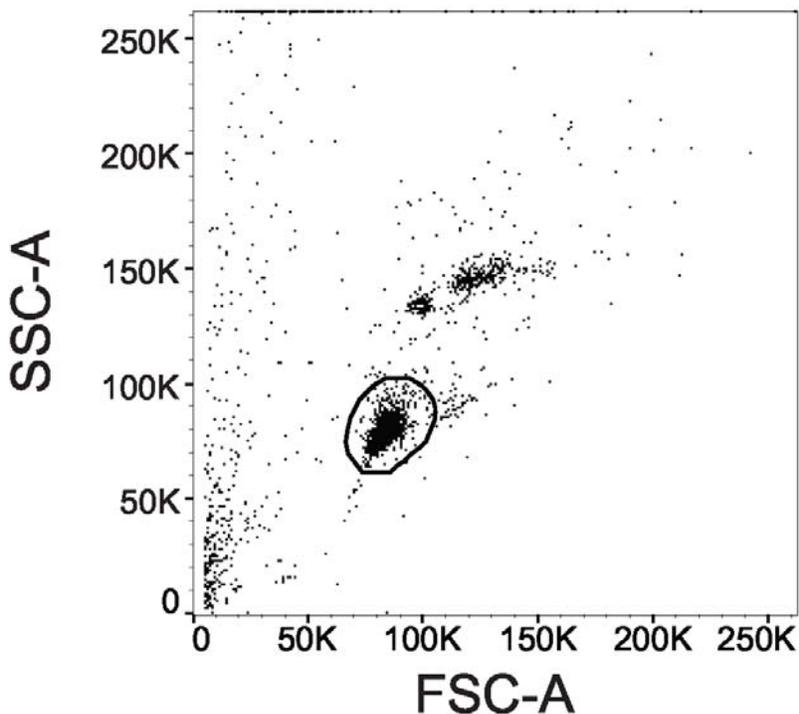
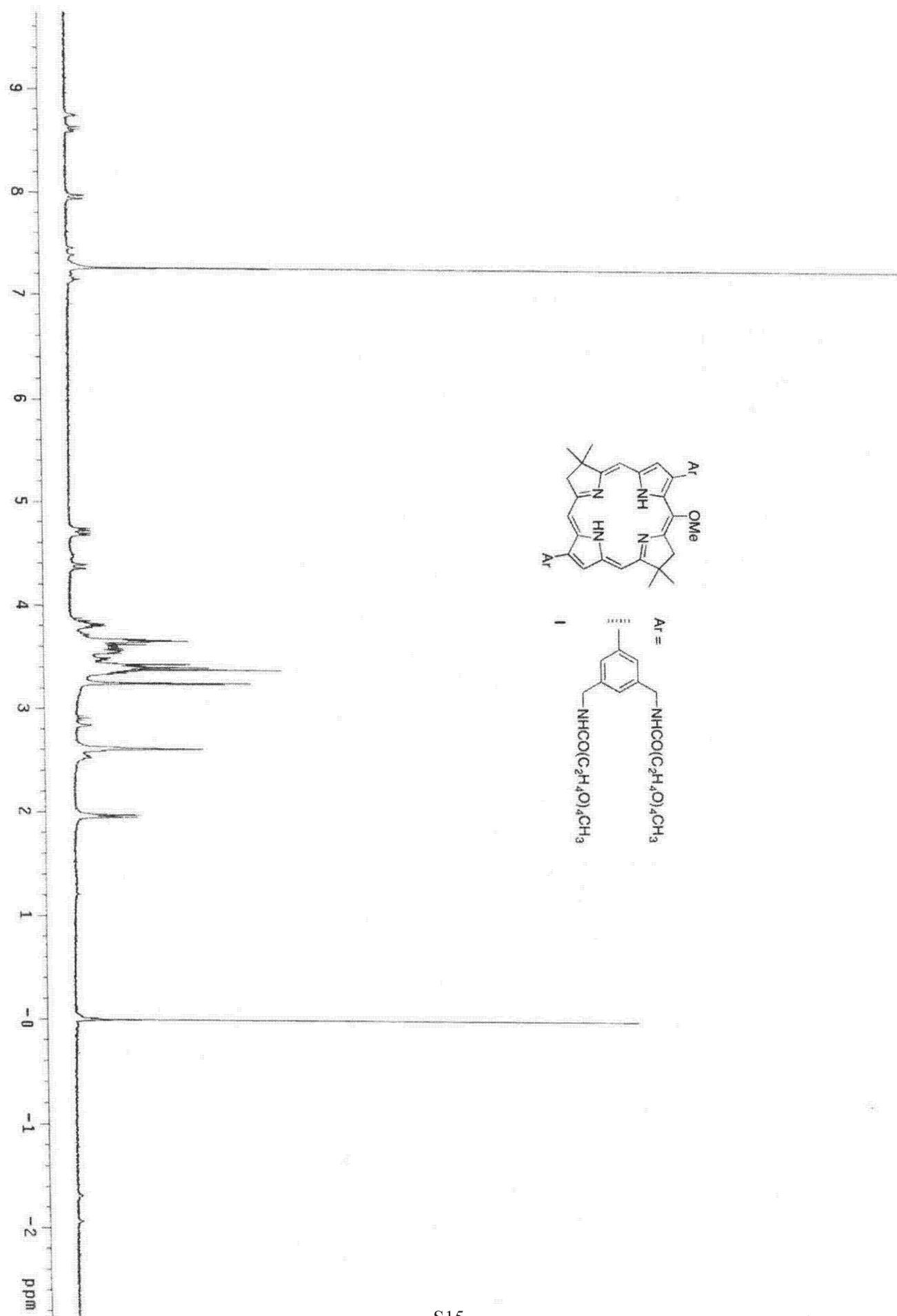


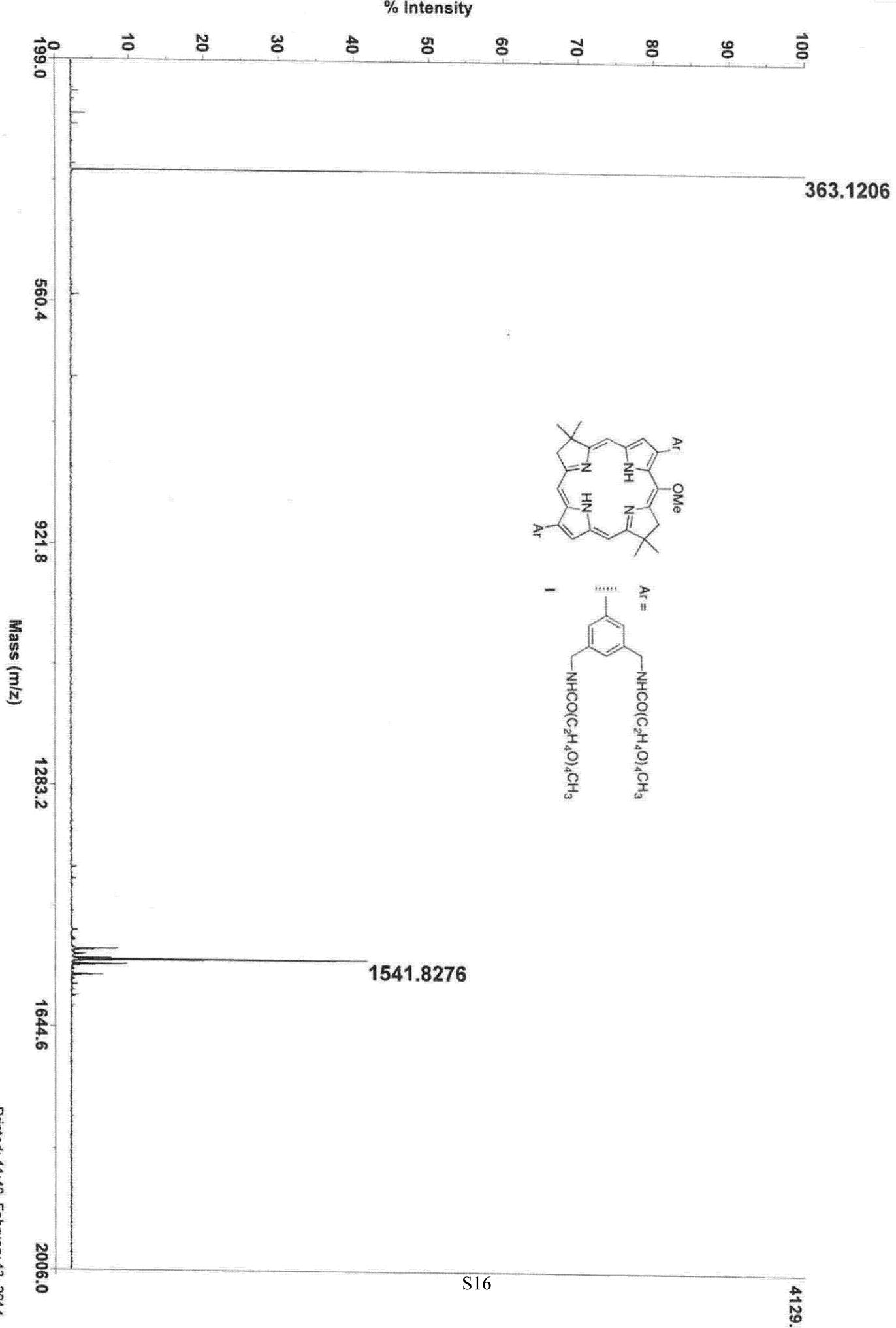
Figure S3. Dot plot of side-scattered light (SSC) versus forward-scattered light (FSC), which provide a measure of granularity and size, respectively. Selection of beads for subsequent analyses was achieved by gating around the major population of single beads, as shown.

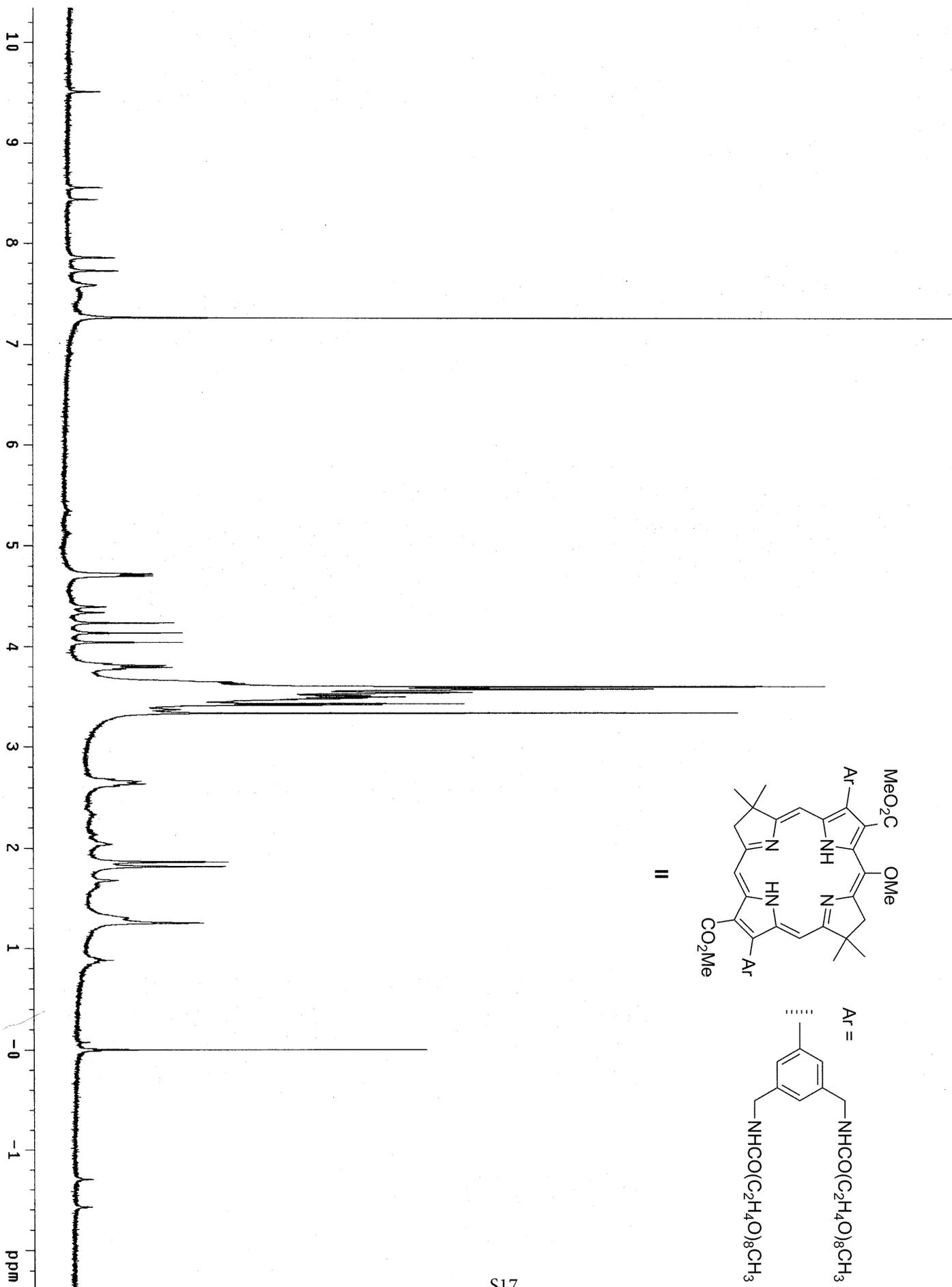
5. References

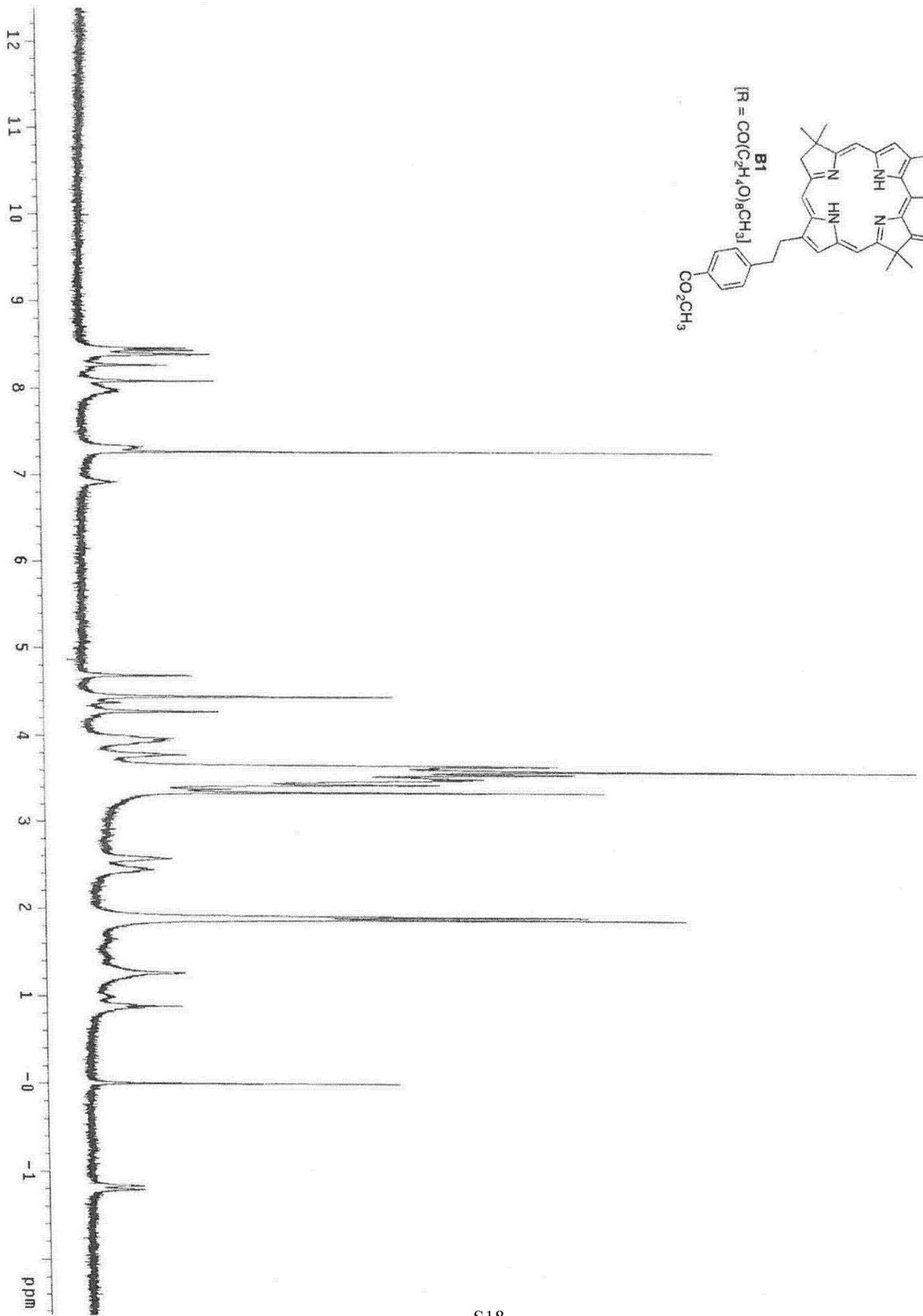
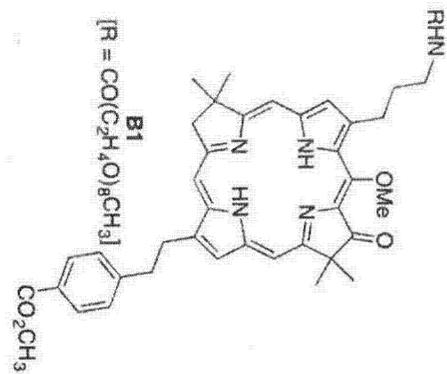
- S1 A. Florian, M. J. Mayoral, V. Stepanenko and G. Fernandez, *Chem. Eur. J.*, 2012, **18**, 14957–14961.
- S2 H. N. Nakayama, T. Morita and S. Kimura, *J. Phys. Chem. C*, 2010, **114**, 4669–4674.
- S3 E. Yang, N. Zhang, M. Krayner, M. Taniguchi, J. R. Diers, C. Kirmaier, J. S. Lindsey, D. F. Bocian and D. Holten, *Photochem. Photobiol.*, 2015, **92**, 111–125.
- S4 B. Sookcharoenpinyo, E. Klein, Y. Ferrand, D. B. Walker, P. R. Brotherhood, C. Ke, M. P. Crump and A. P. Davis, *Angew. Chem. Int. Ed.*, 2012, **51**, 4586–4590.
- S5 K. R. Reddy, J. Jiang, M. Krayner, M. A. Harris, J. W. Springer, E. Yang, J. Jiao, D. M. Niedzwiedzki, D. Pandithavidana, P. S. Parkes-Loach, C. Kirmaier, P. A. Loach, D. F. Bocian, D. Holten and J. S. Lindsey, *Chem. Sci.* 2013, **4**, 2036–2053.
- S6 K. E. Borbas, V. Chandrasher, C. Muthiah, H. L. Kee, D. Holten and J. S. Lindsey, *J. Org. Chem.*, 2008, **73**, 3145–3158.

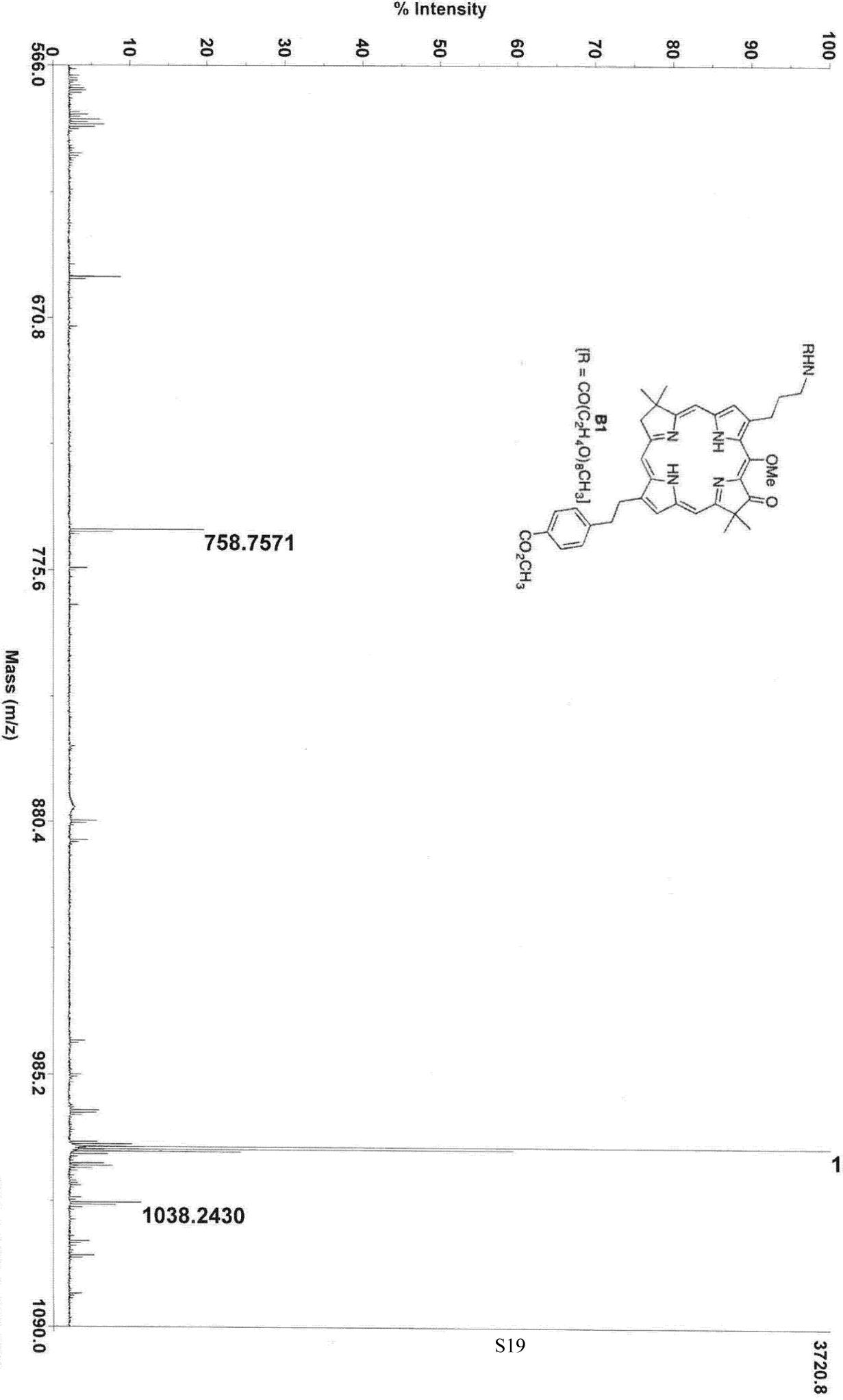
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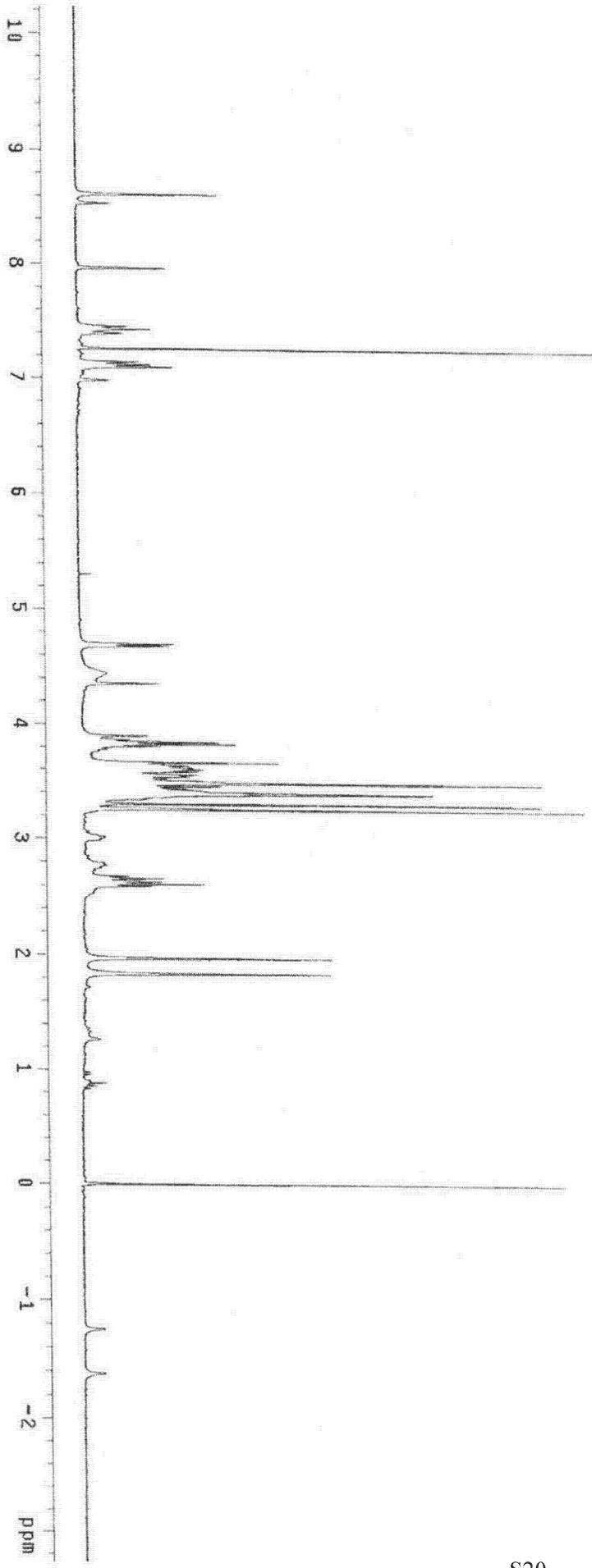
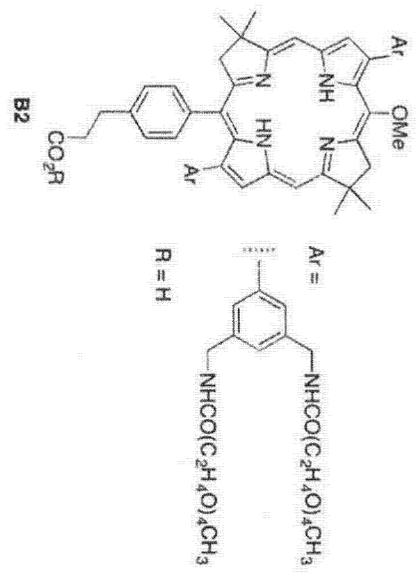


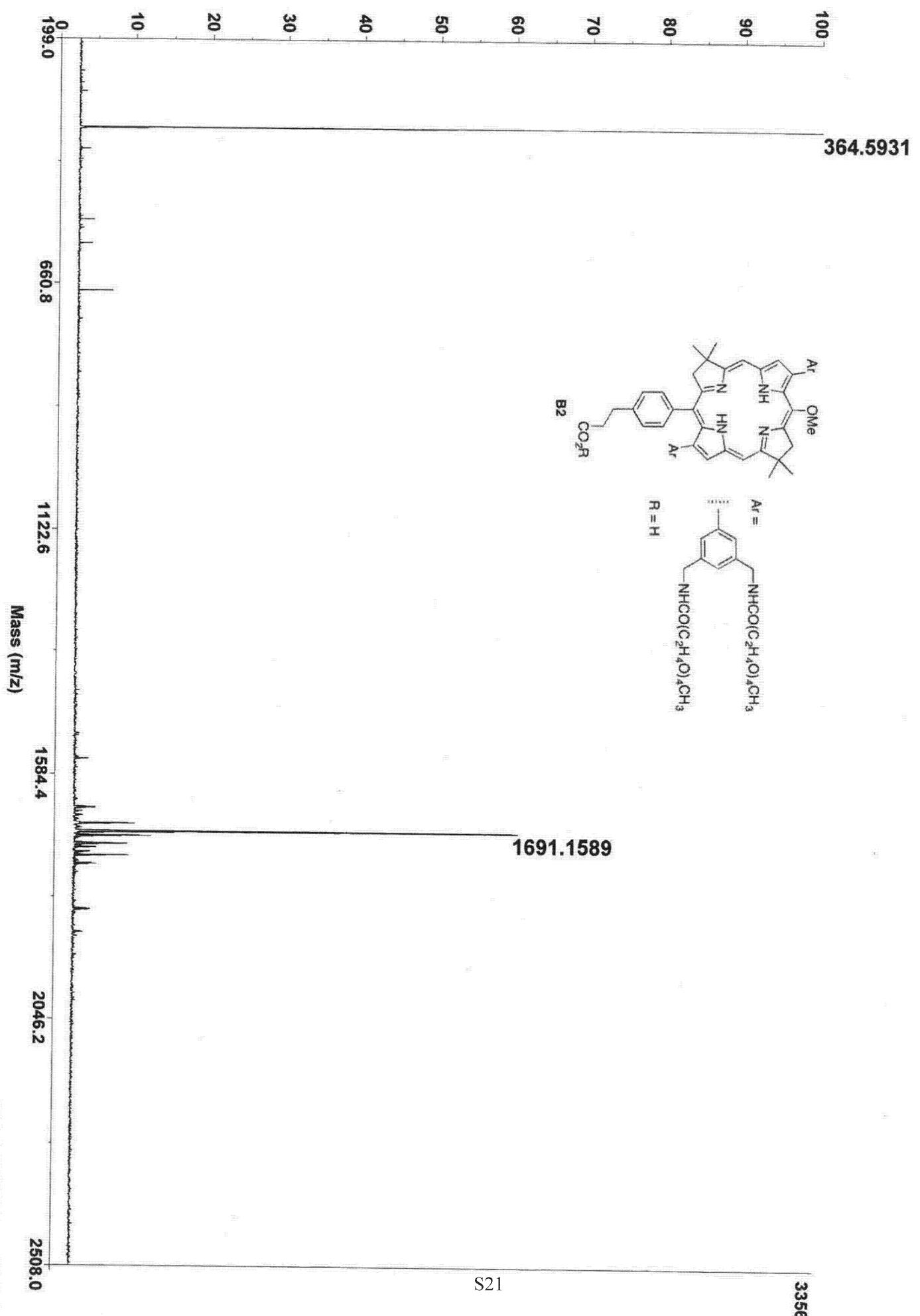






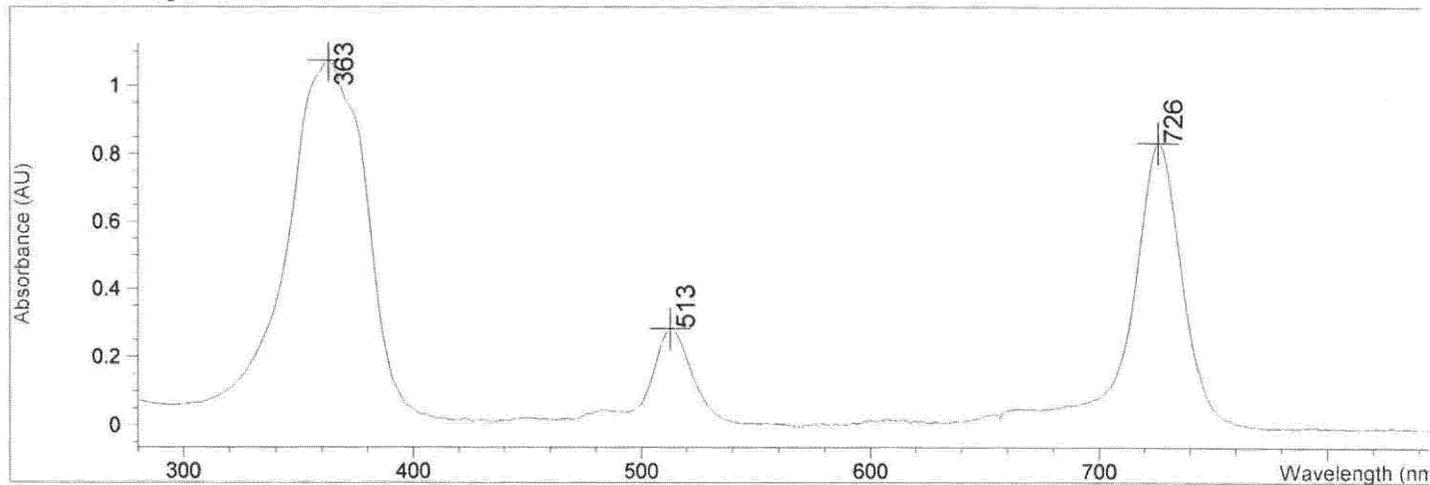






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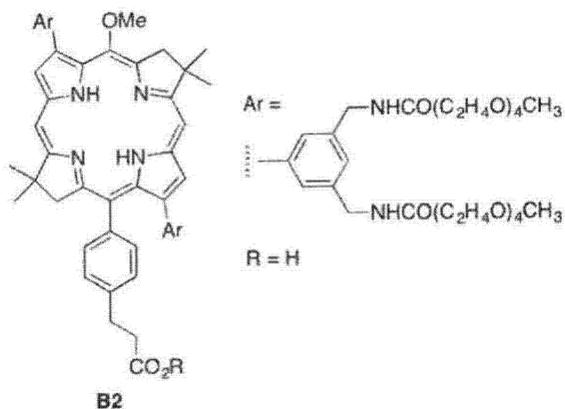


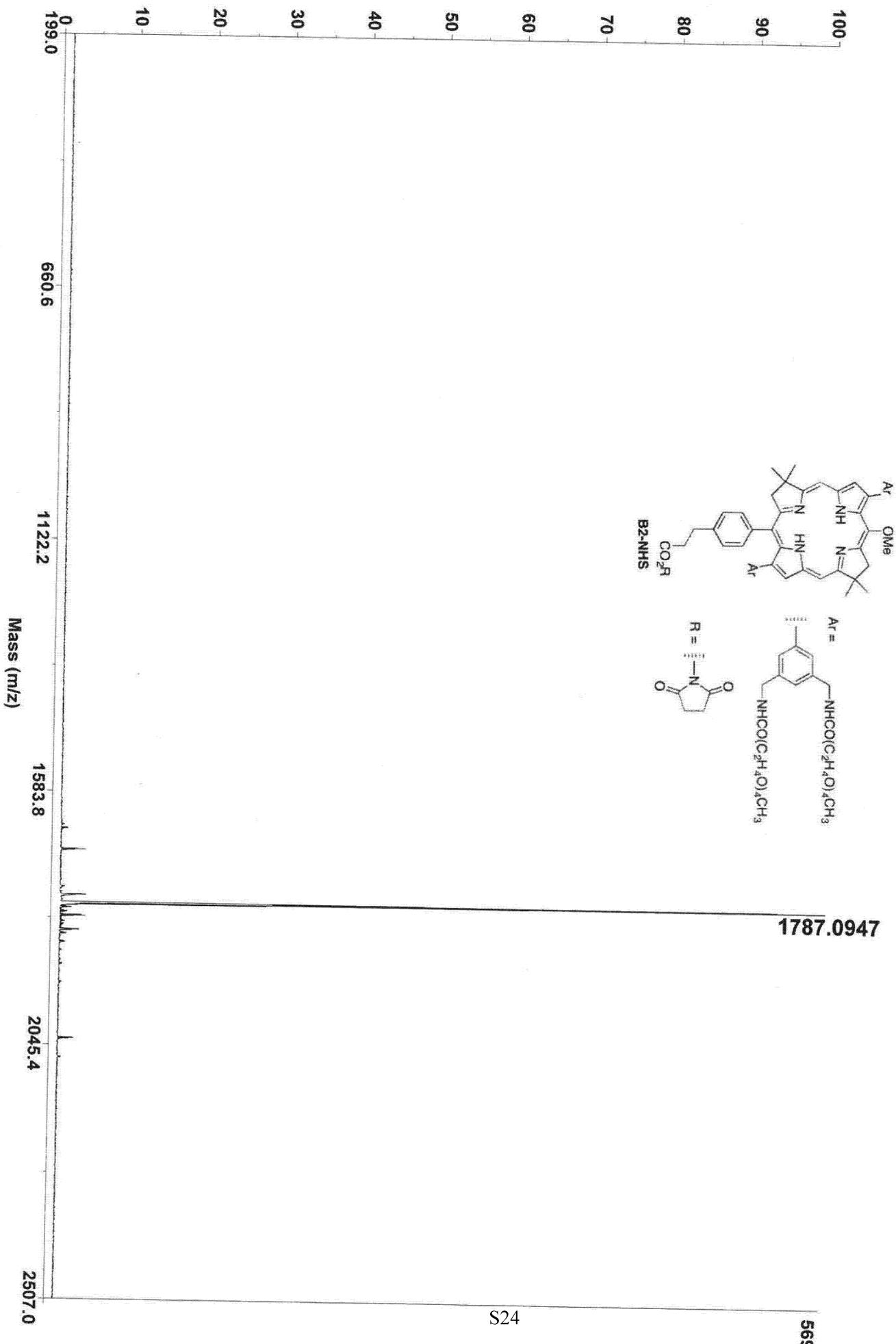
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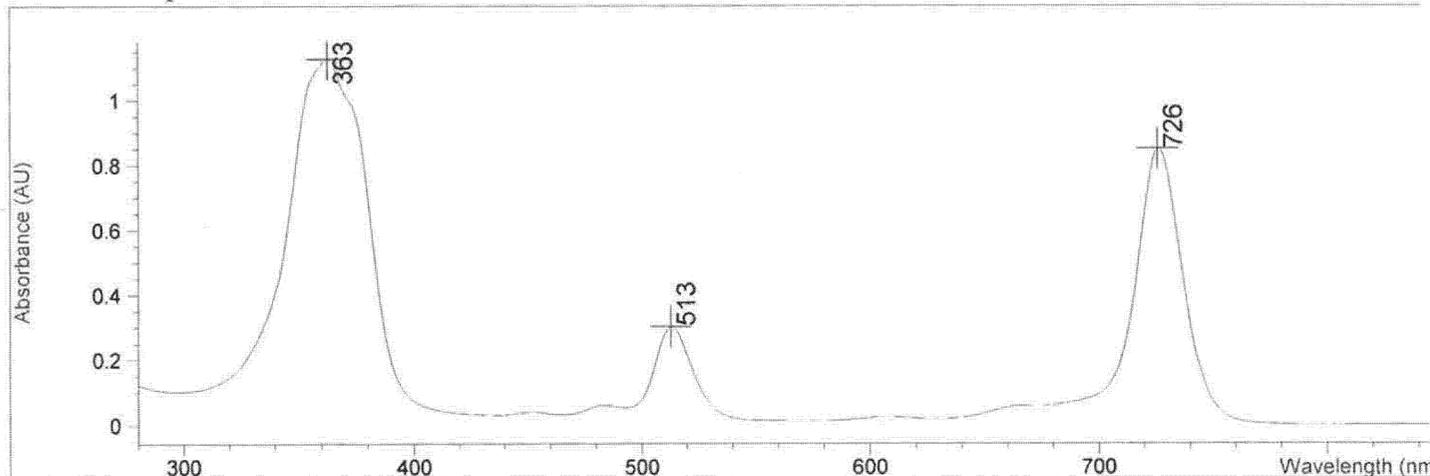
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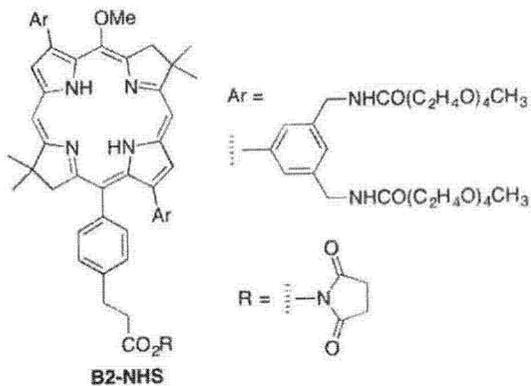


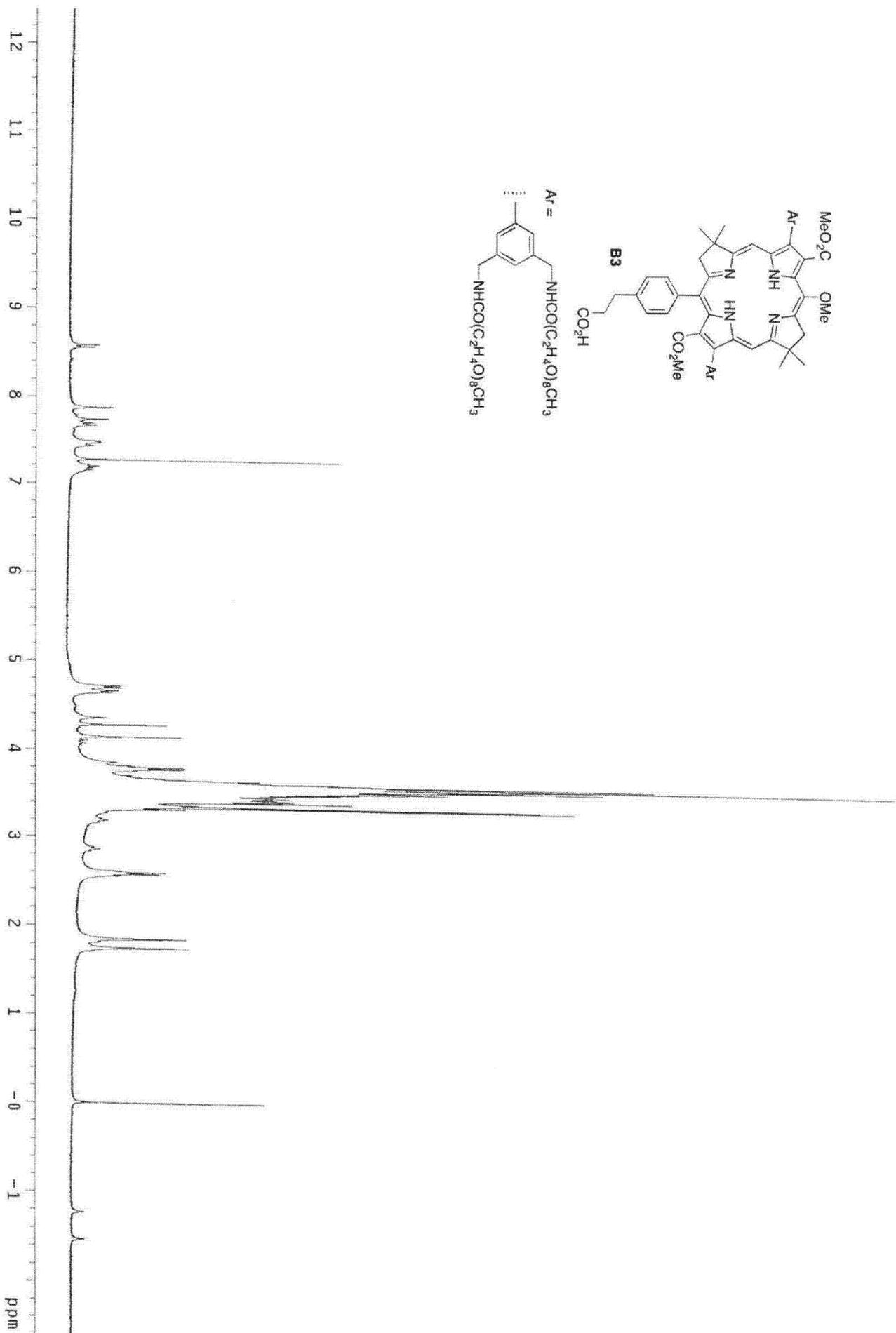
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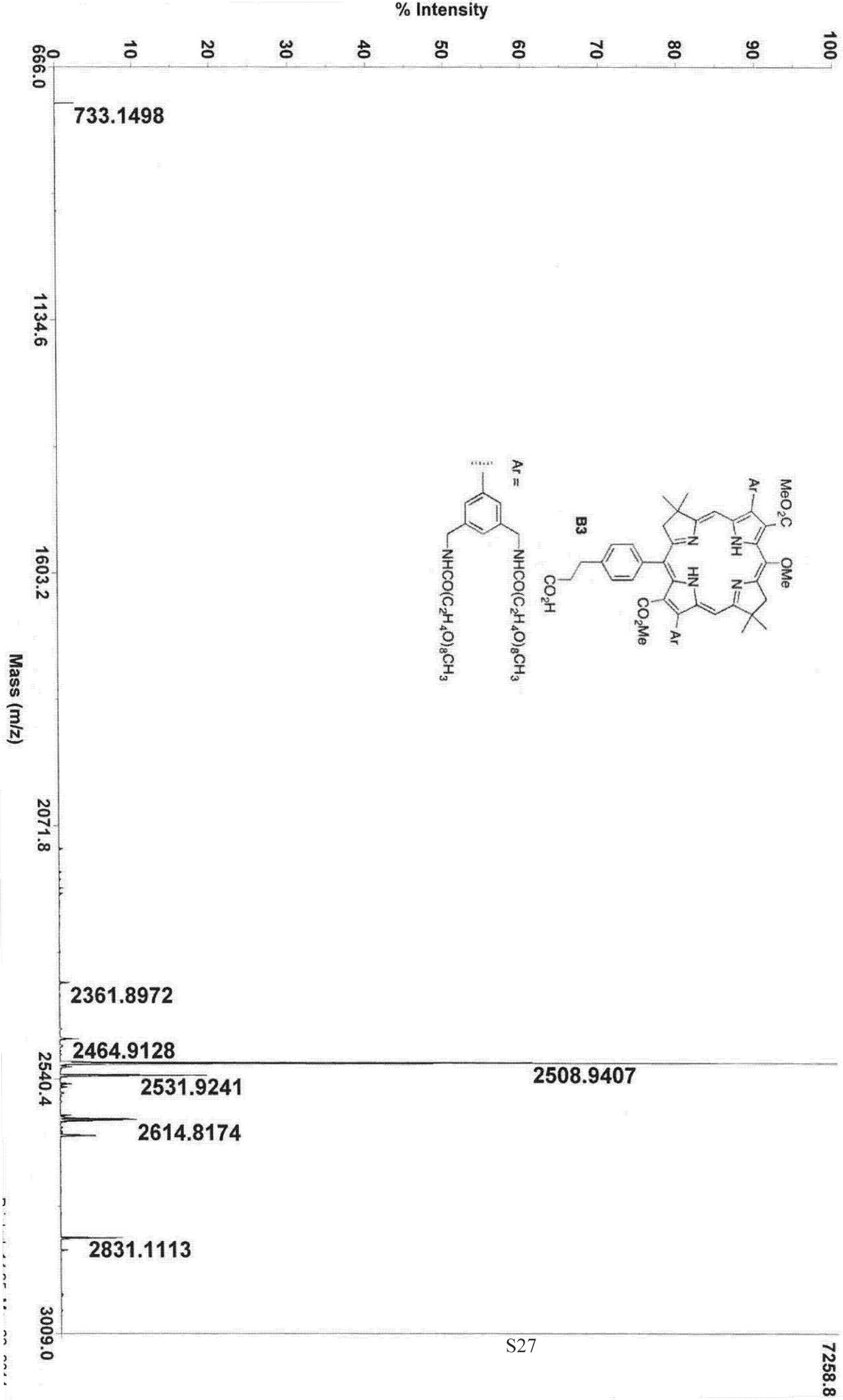
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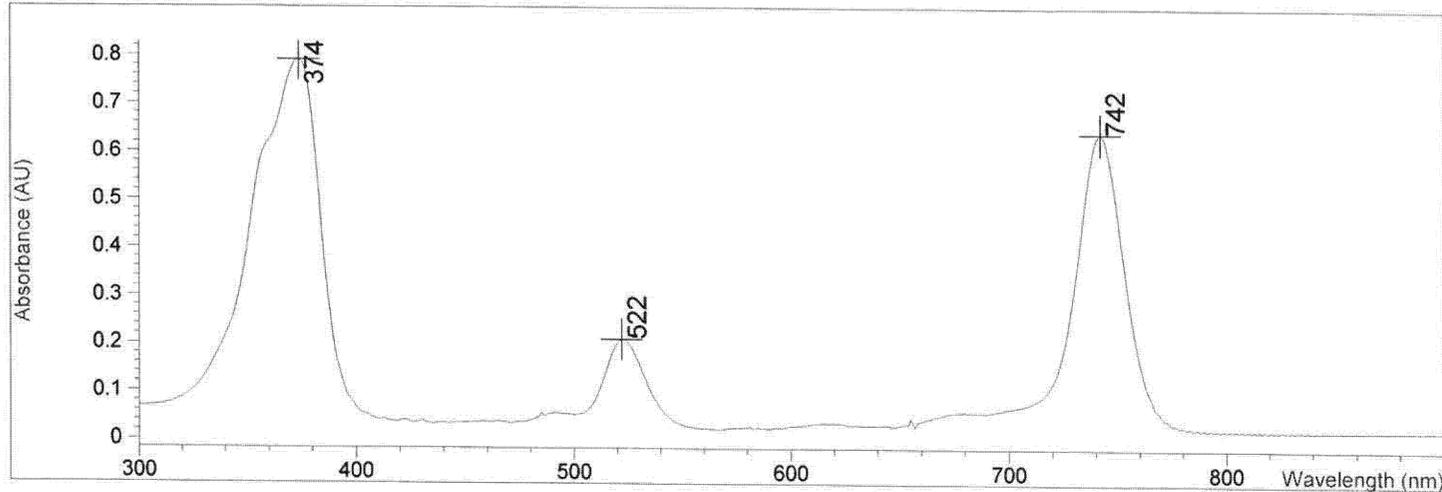






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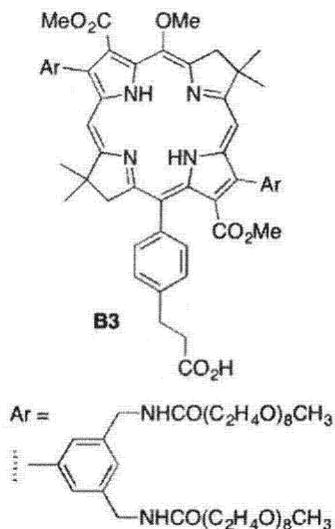


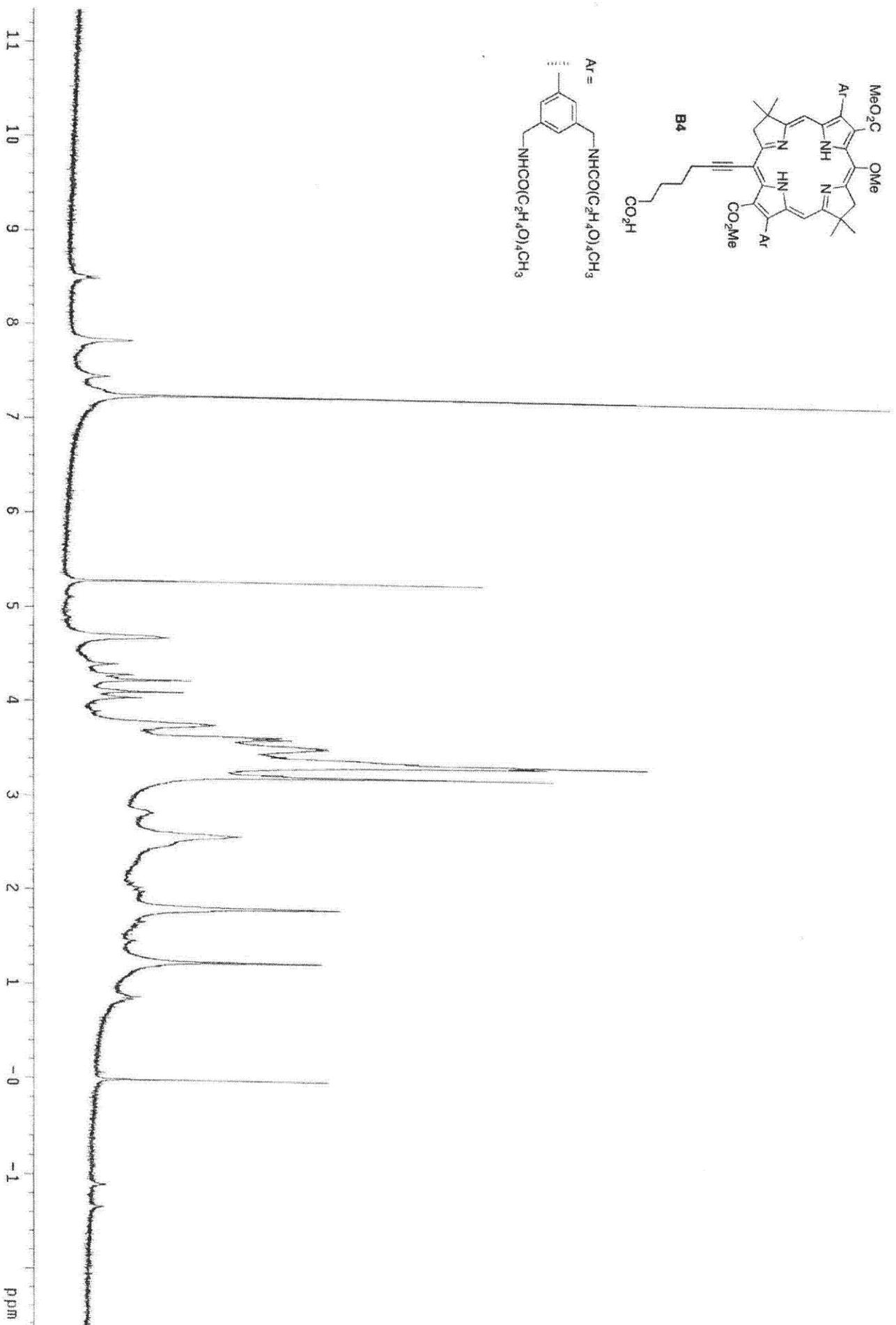
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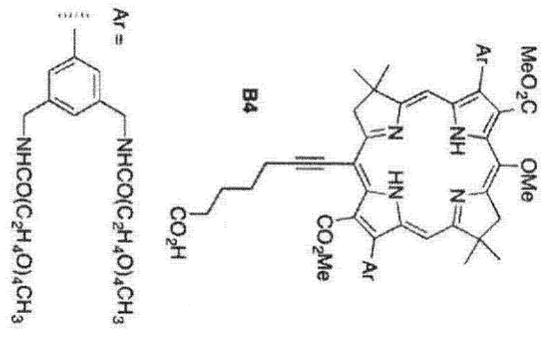
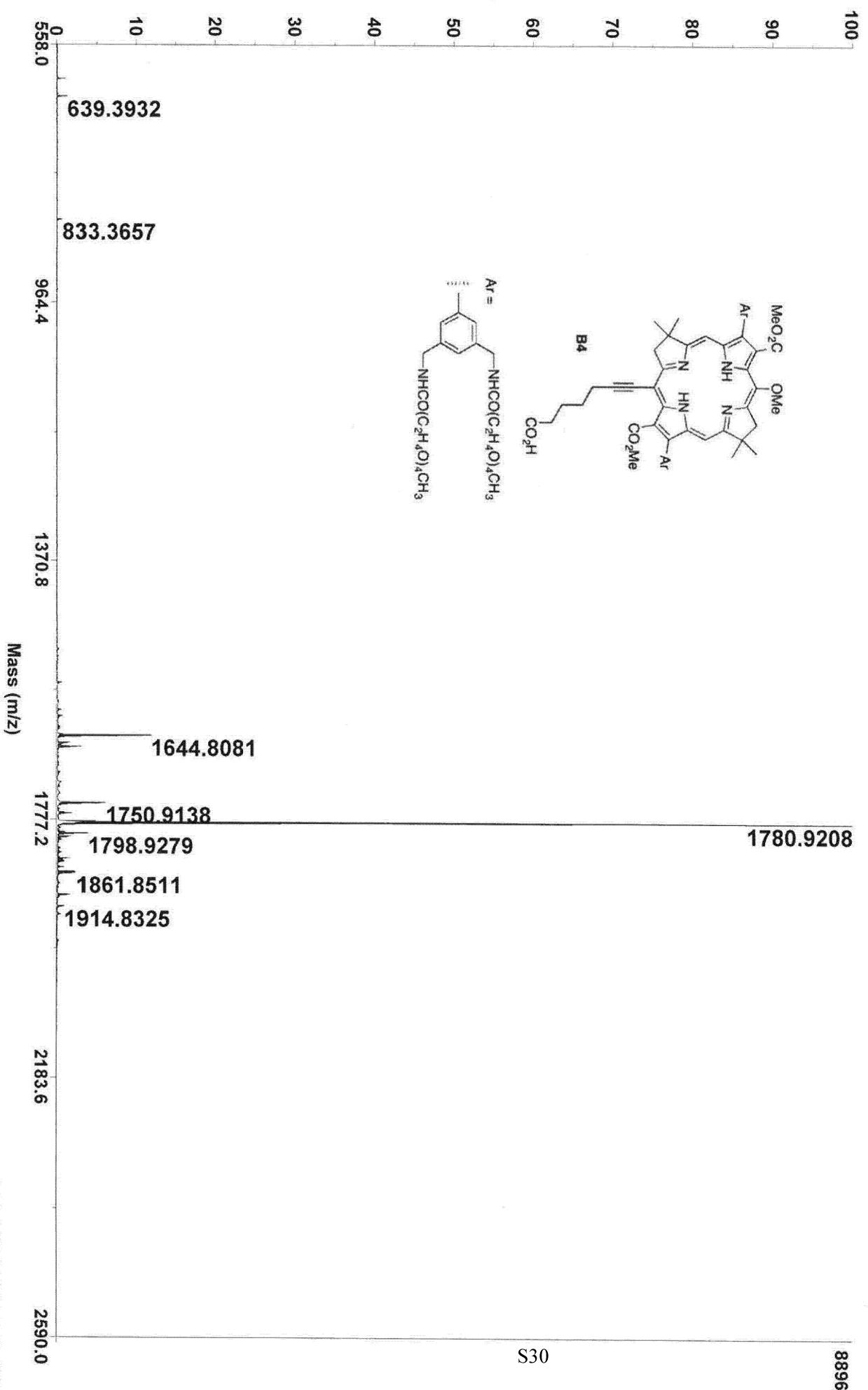
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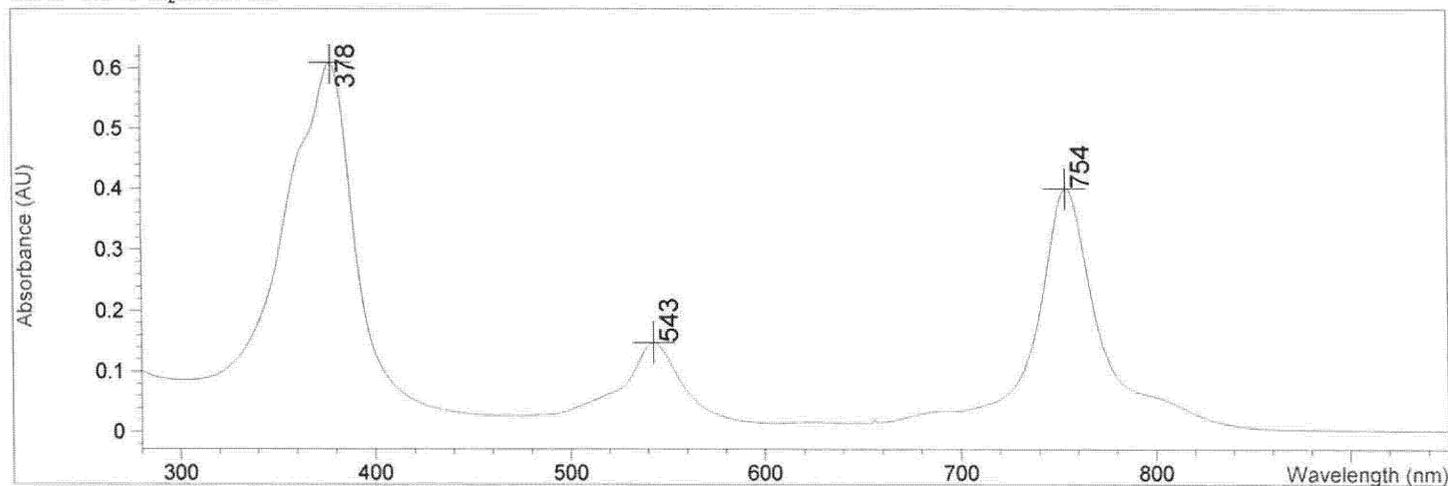






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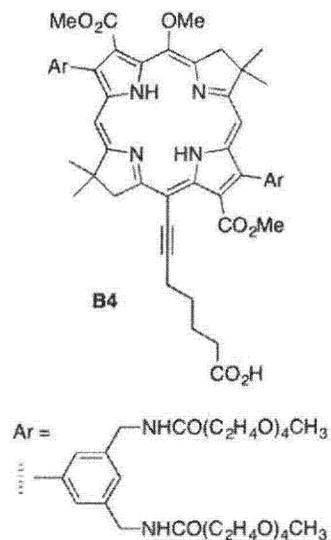


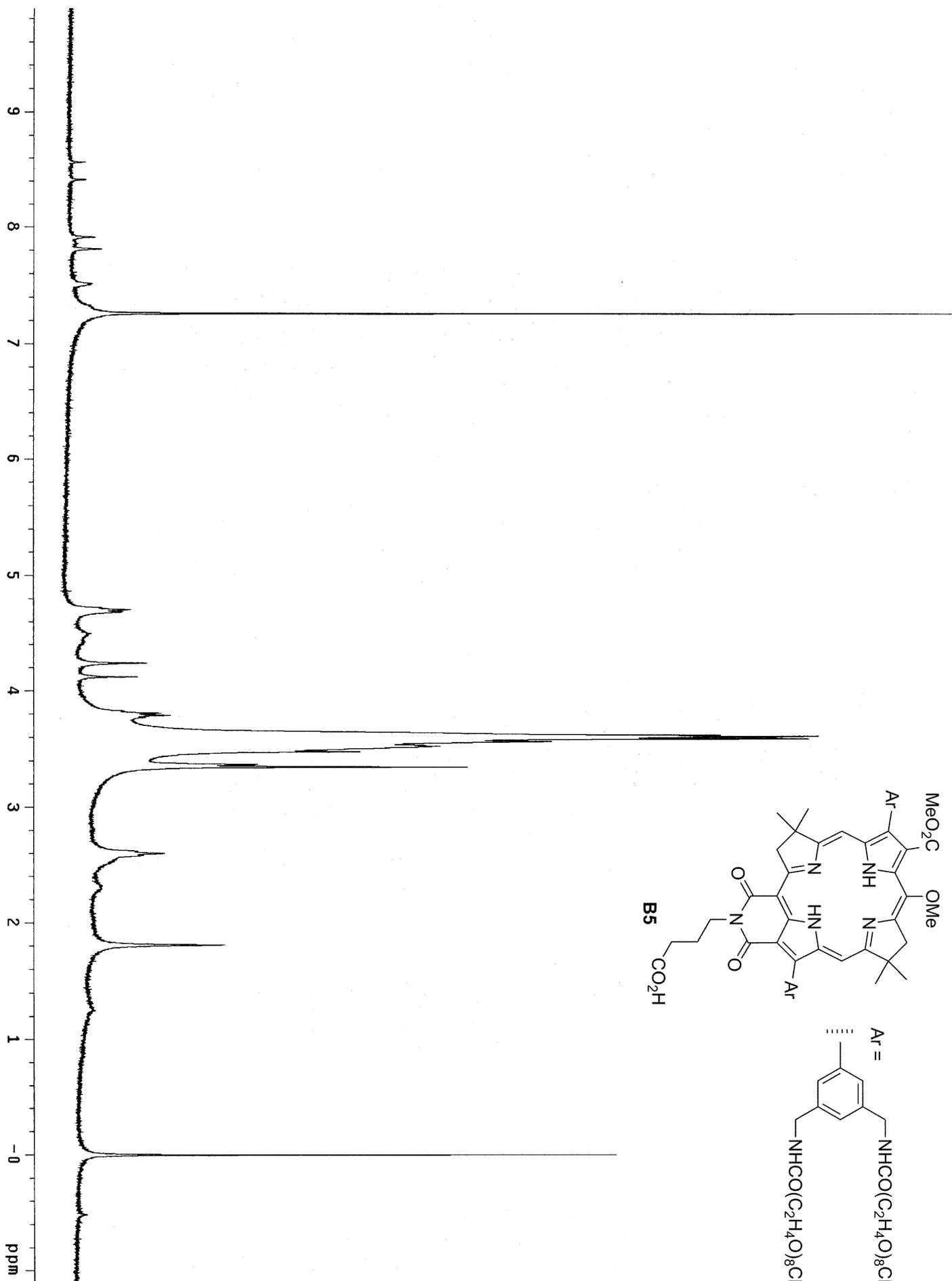
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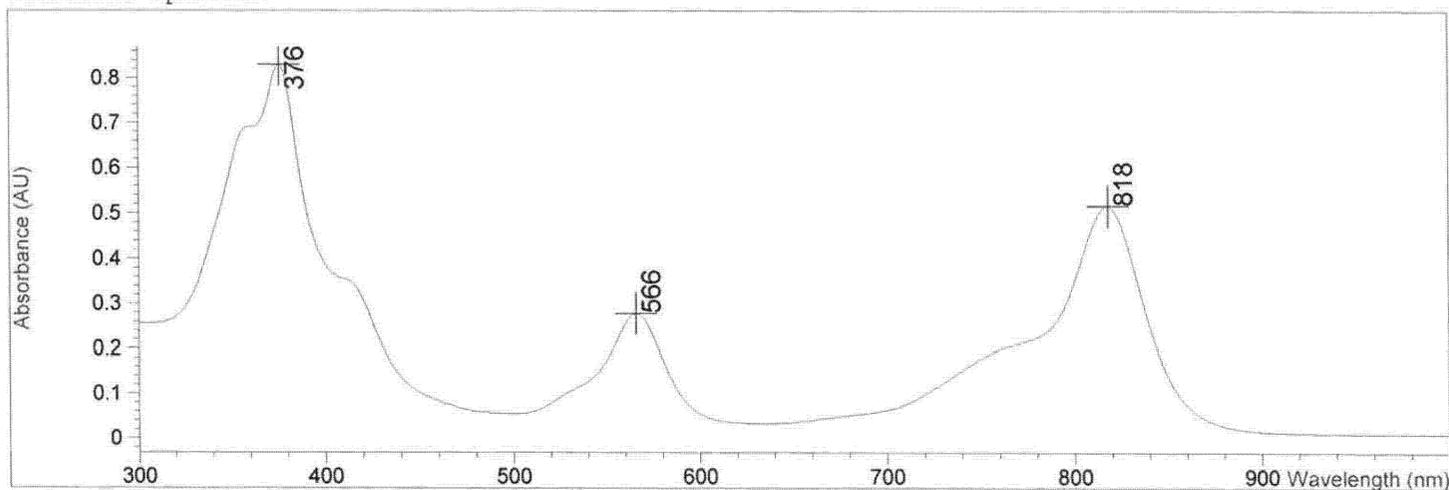
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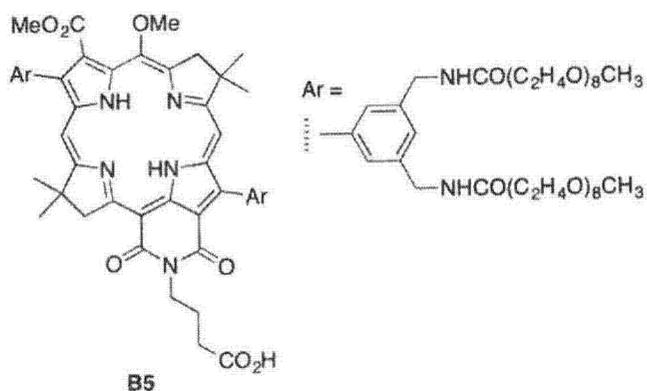


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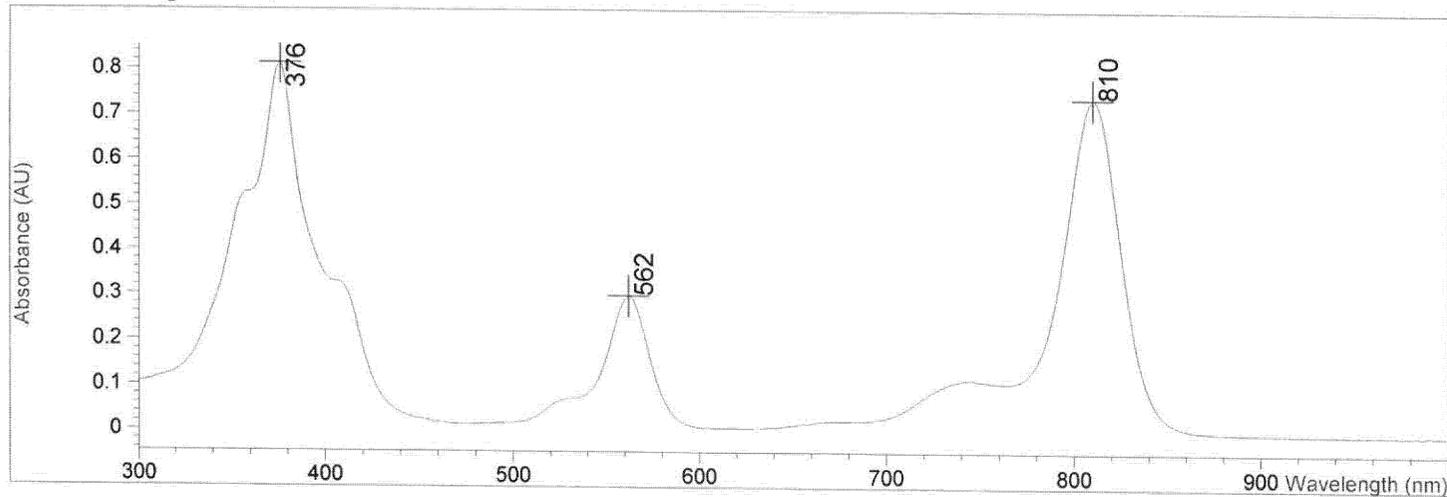
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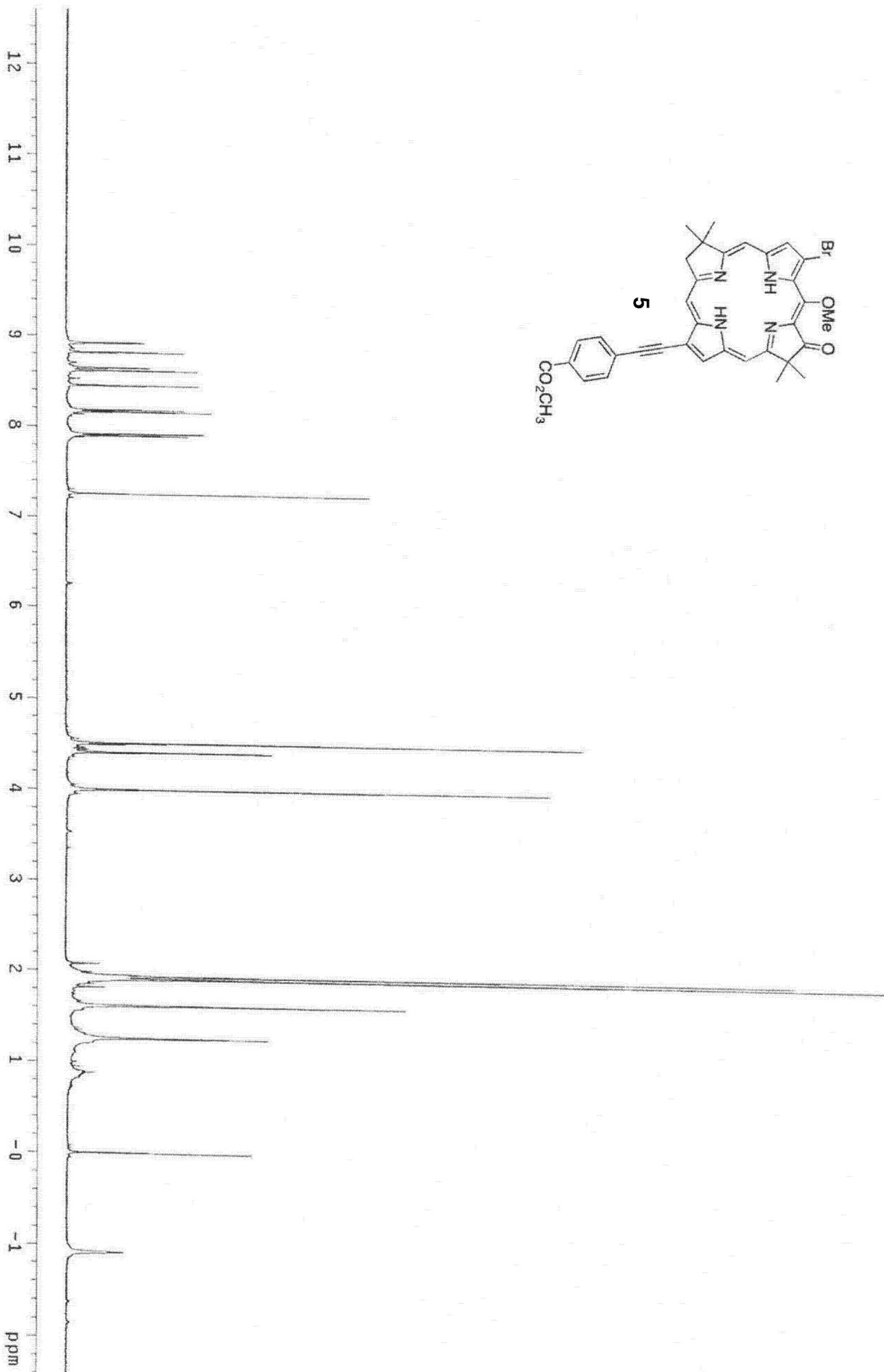
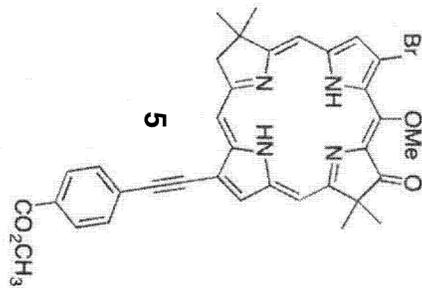


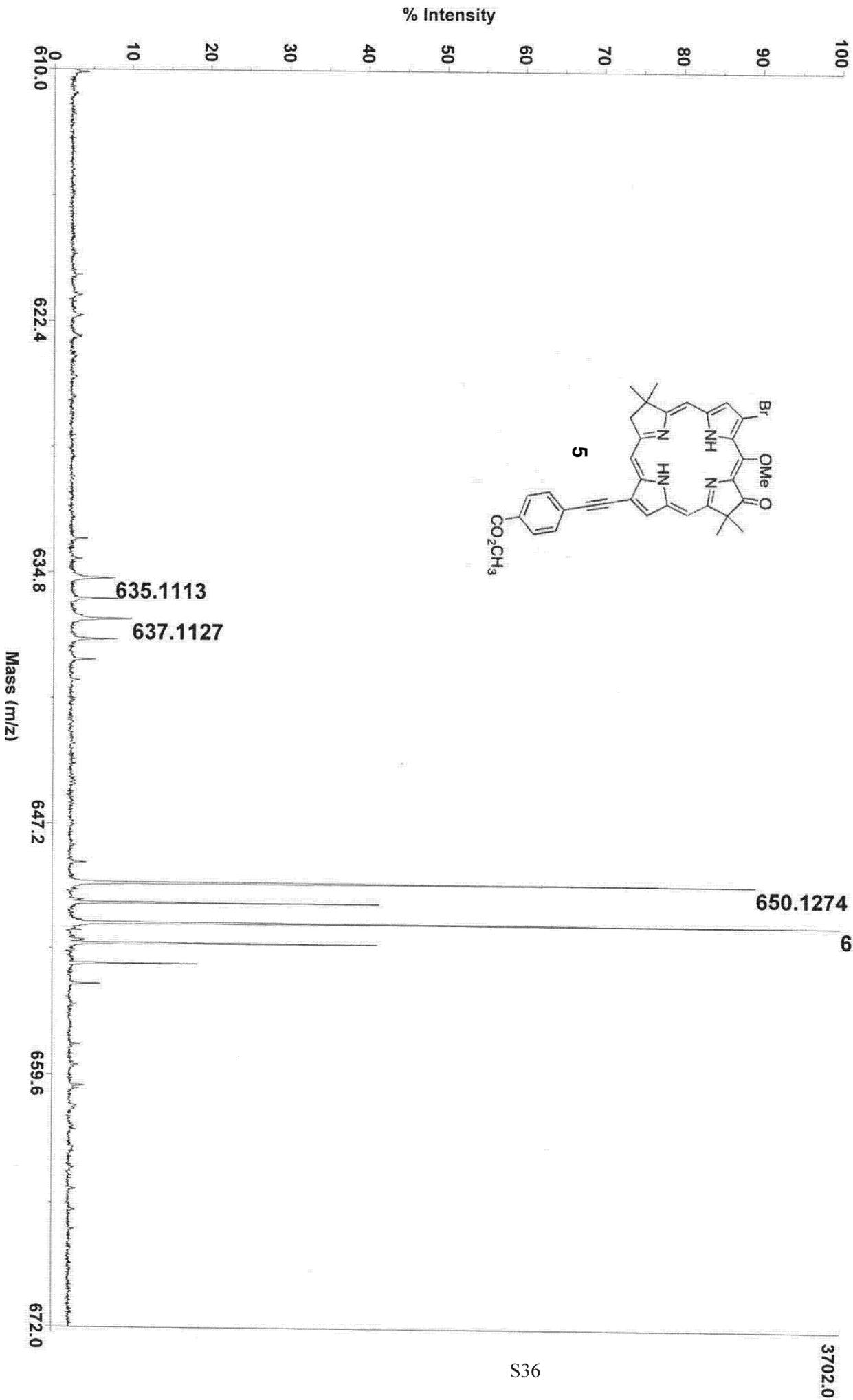
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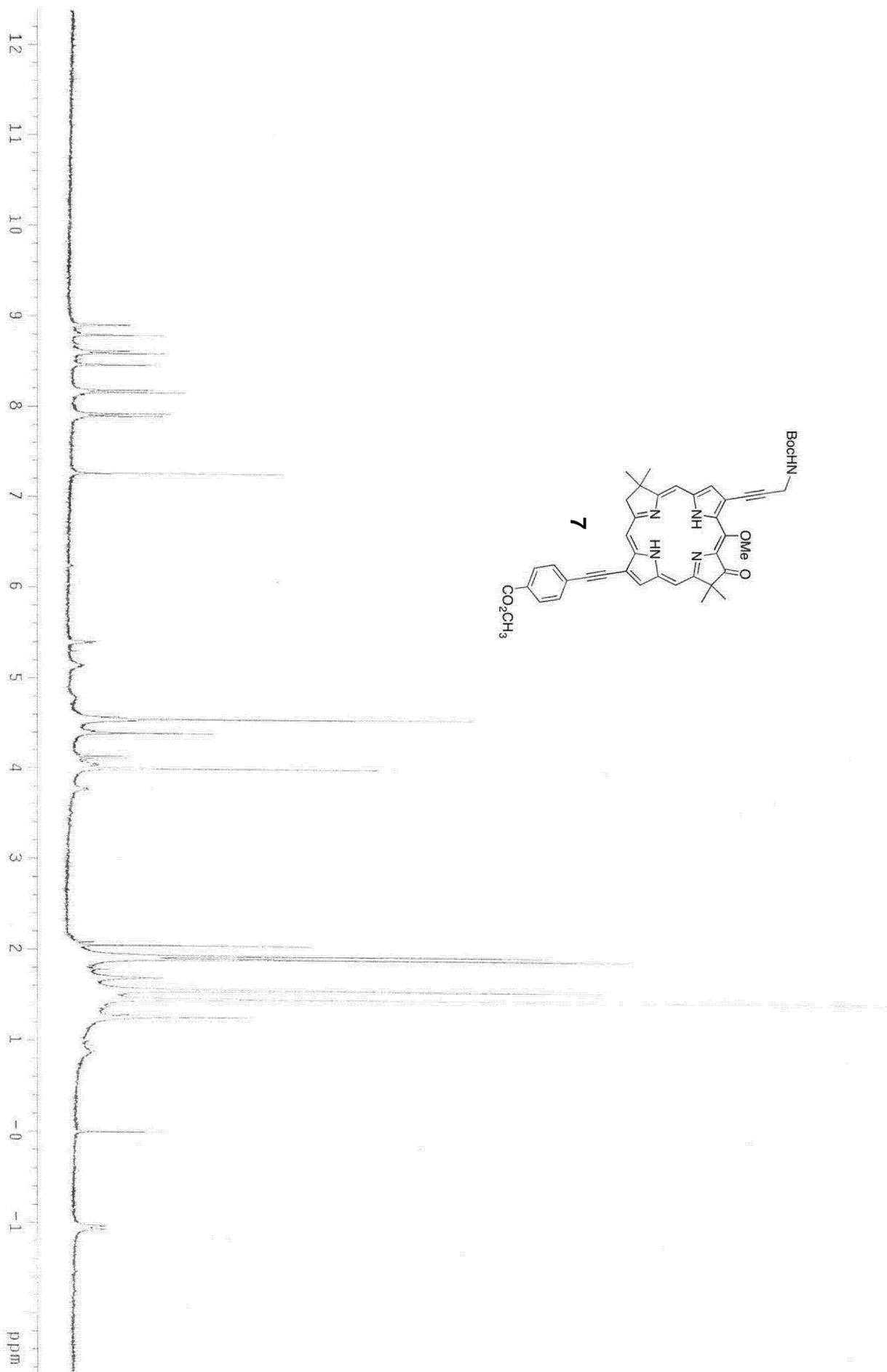
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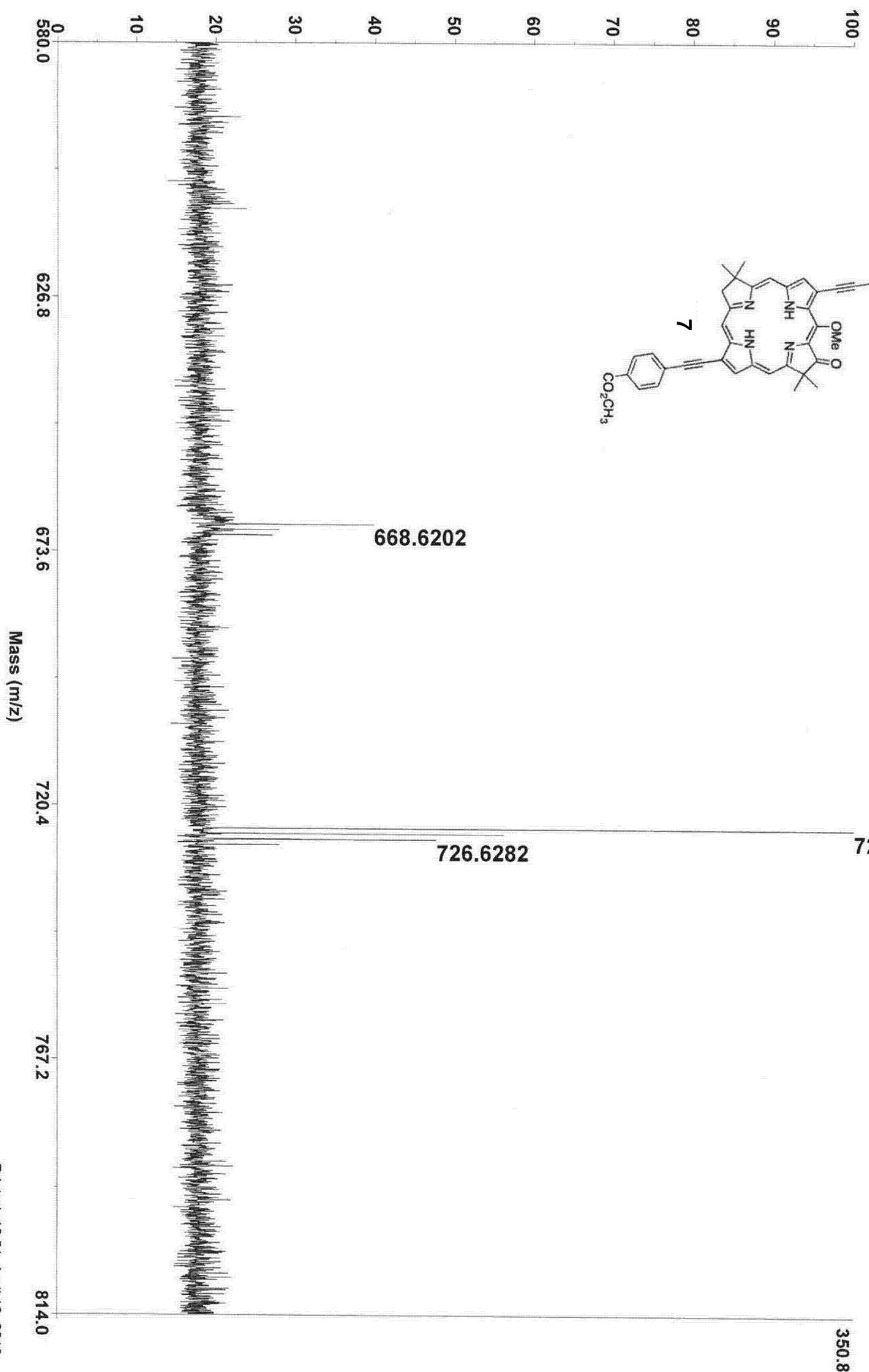
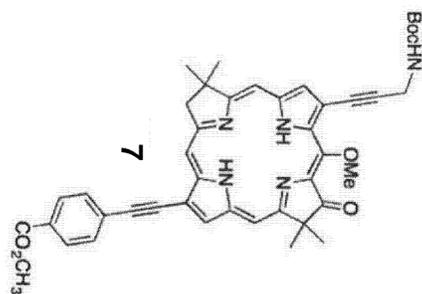
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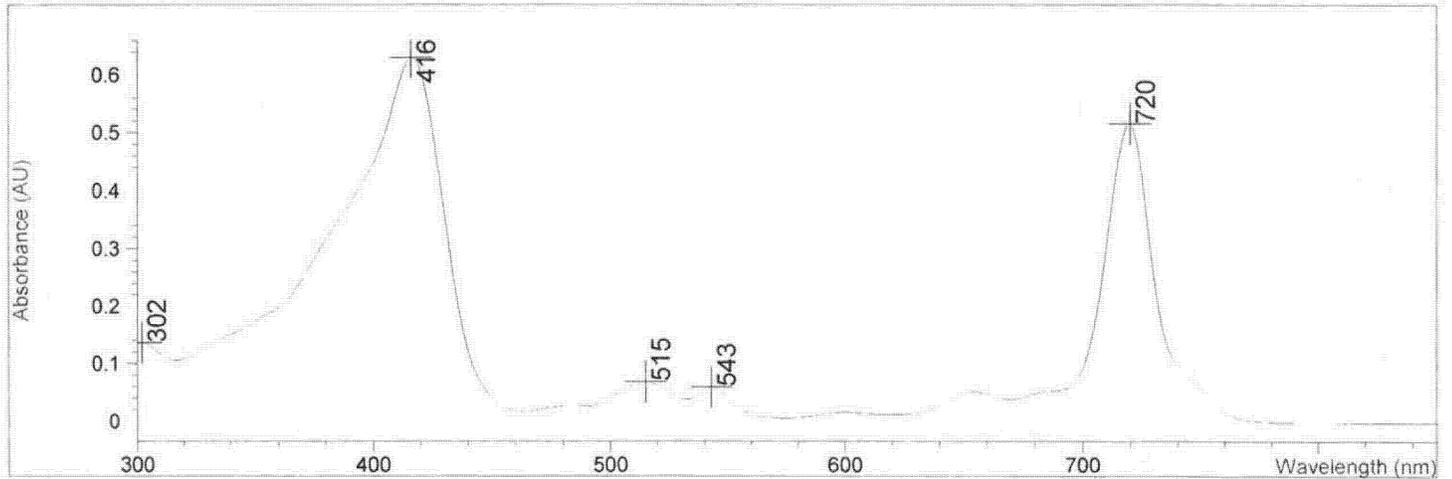






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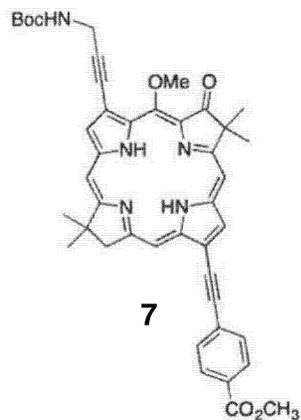


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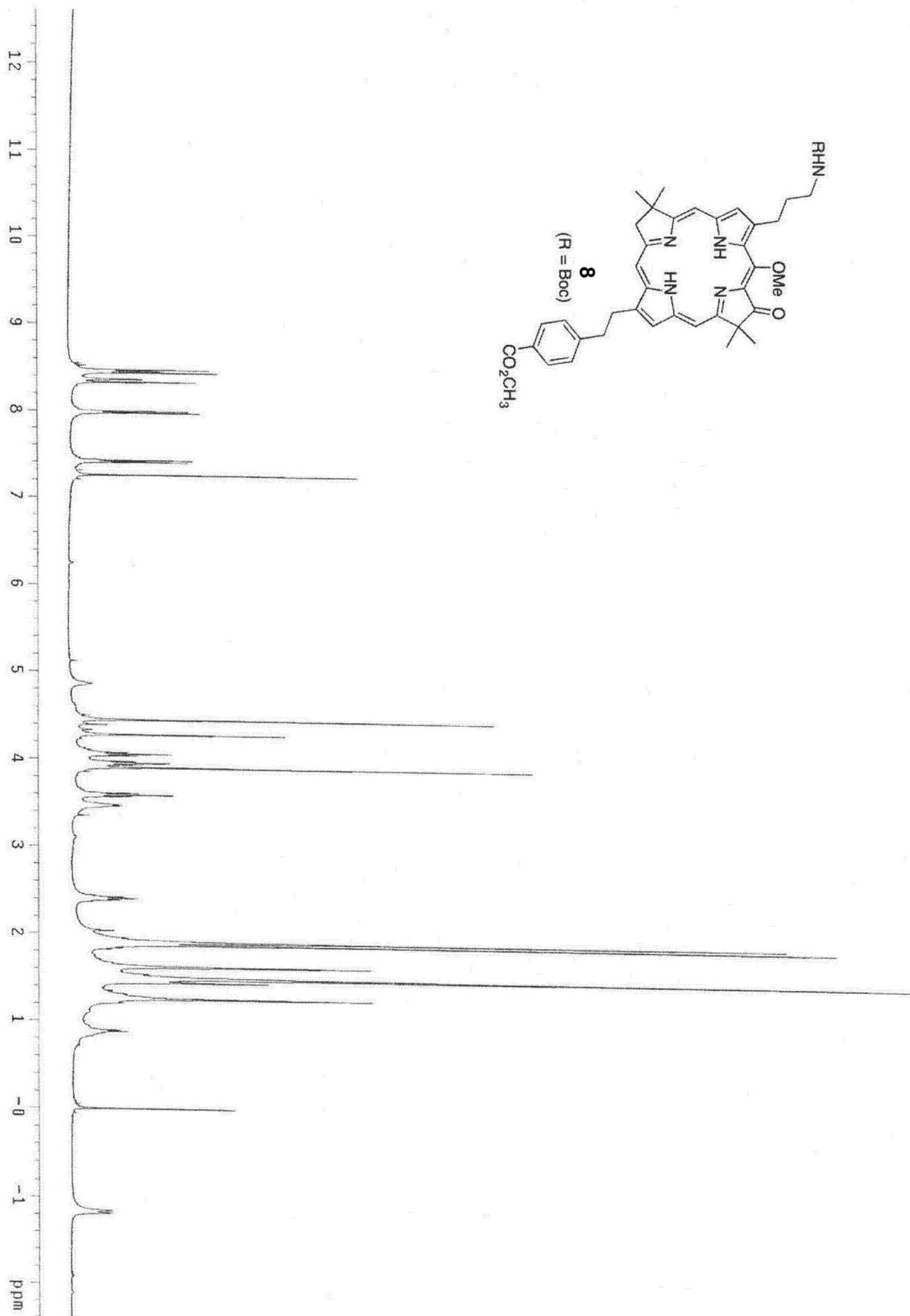
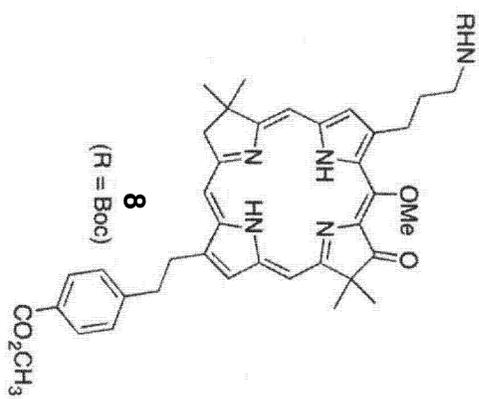
Report generated by : Lindsey Lab

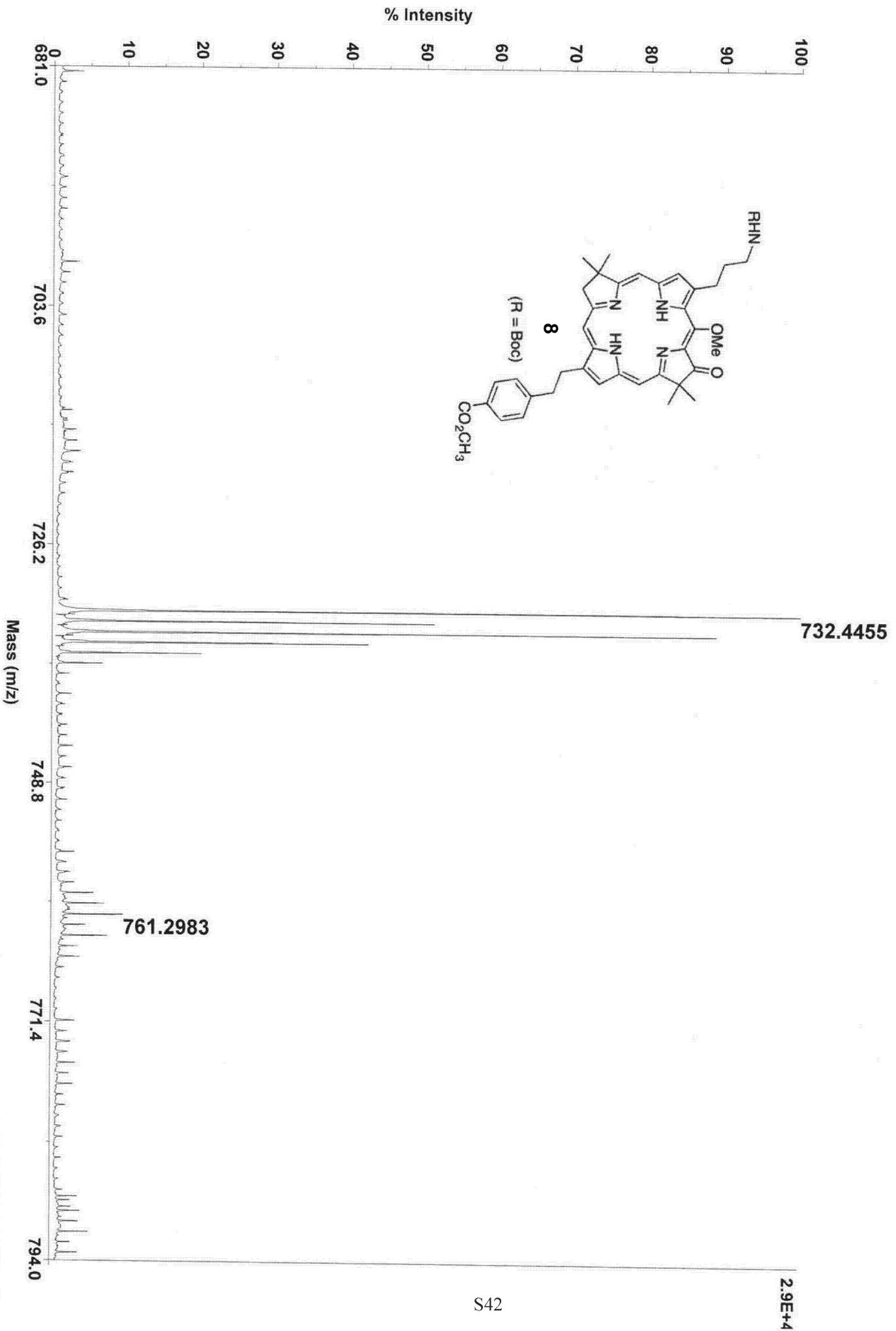
Signature:

*** End Spectrum/Peak Report ***



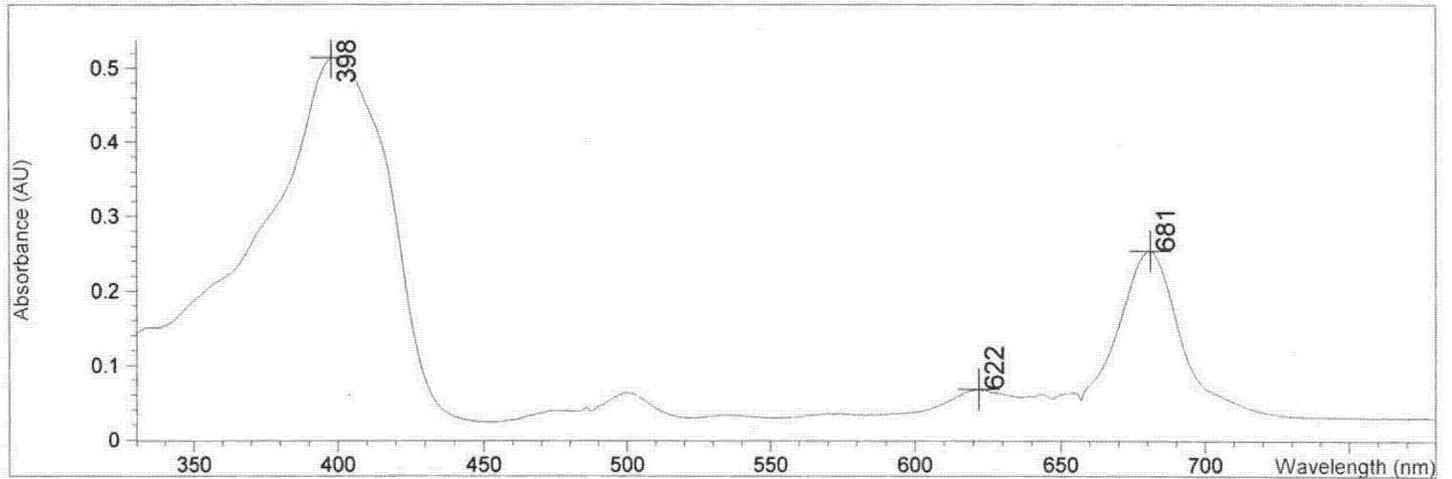
7





Method file : <method not saved>
Information : Default Method
Data File : <data not saved>

Overlaid Spectra:

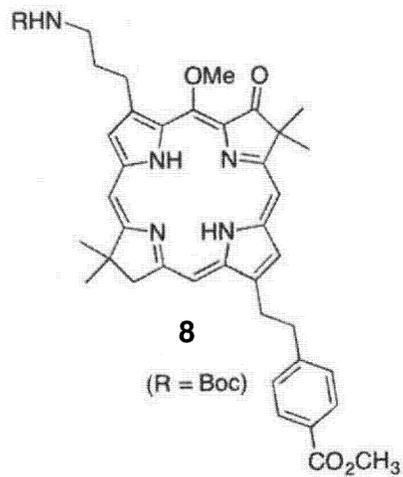


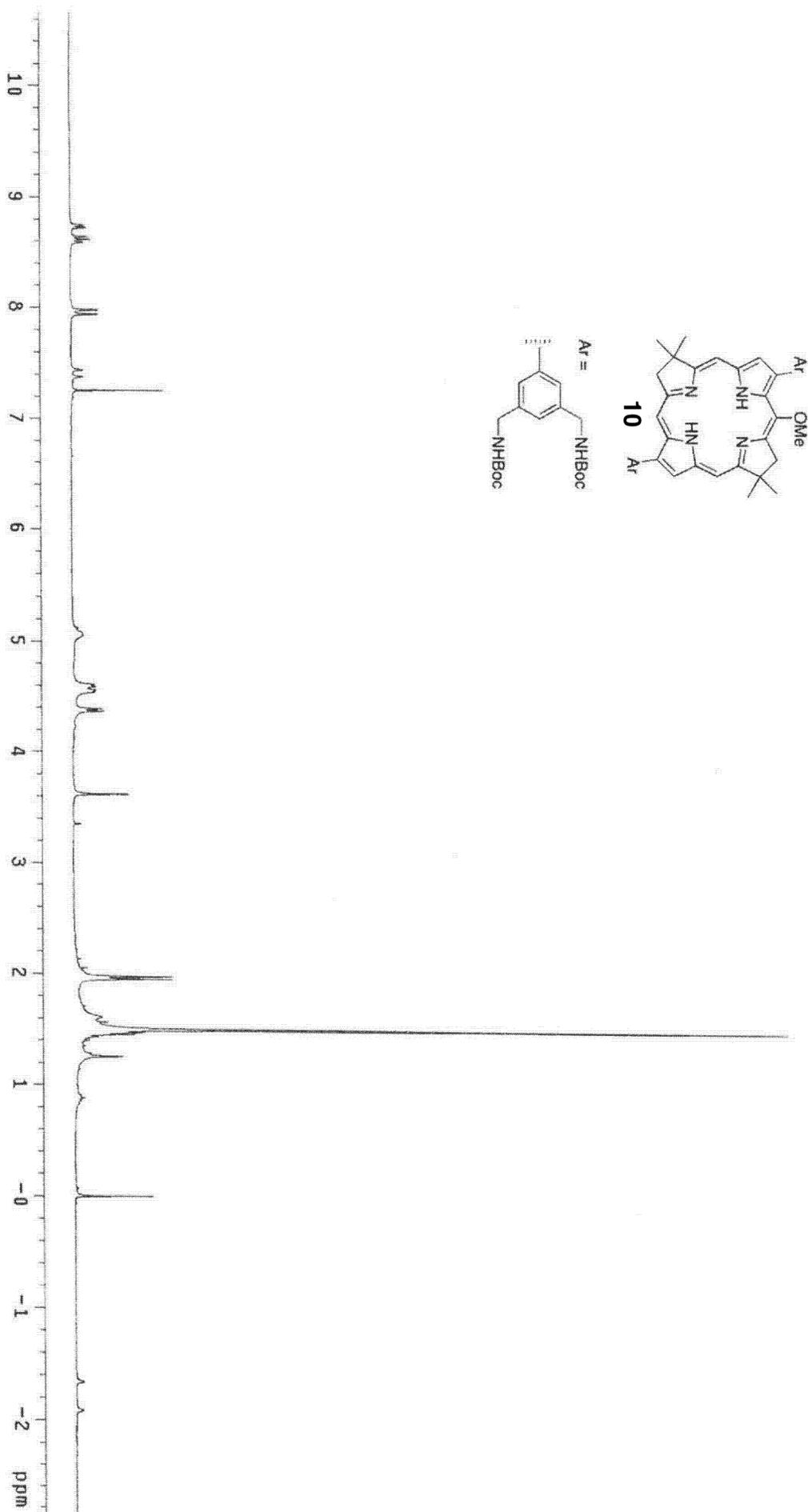
#	Name	Peaks (nm)	Abs (AU)
1		398.0	0.51381
1		681.0	0.25373
1		622.0	6.7577E-2

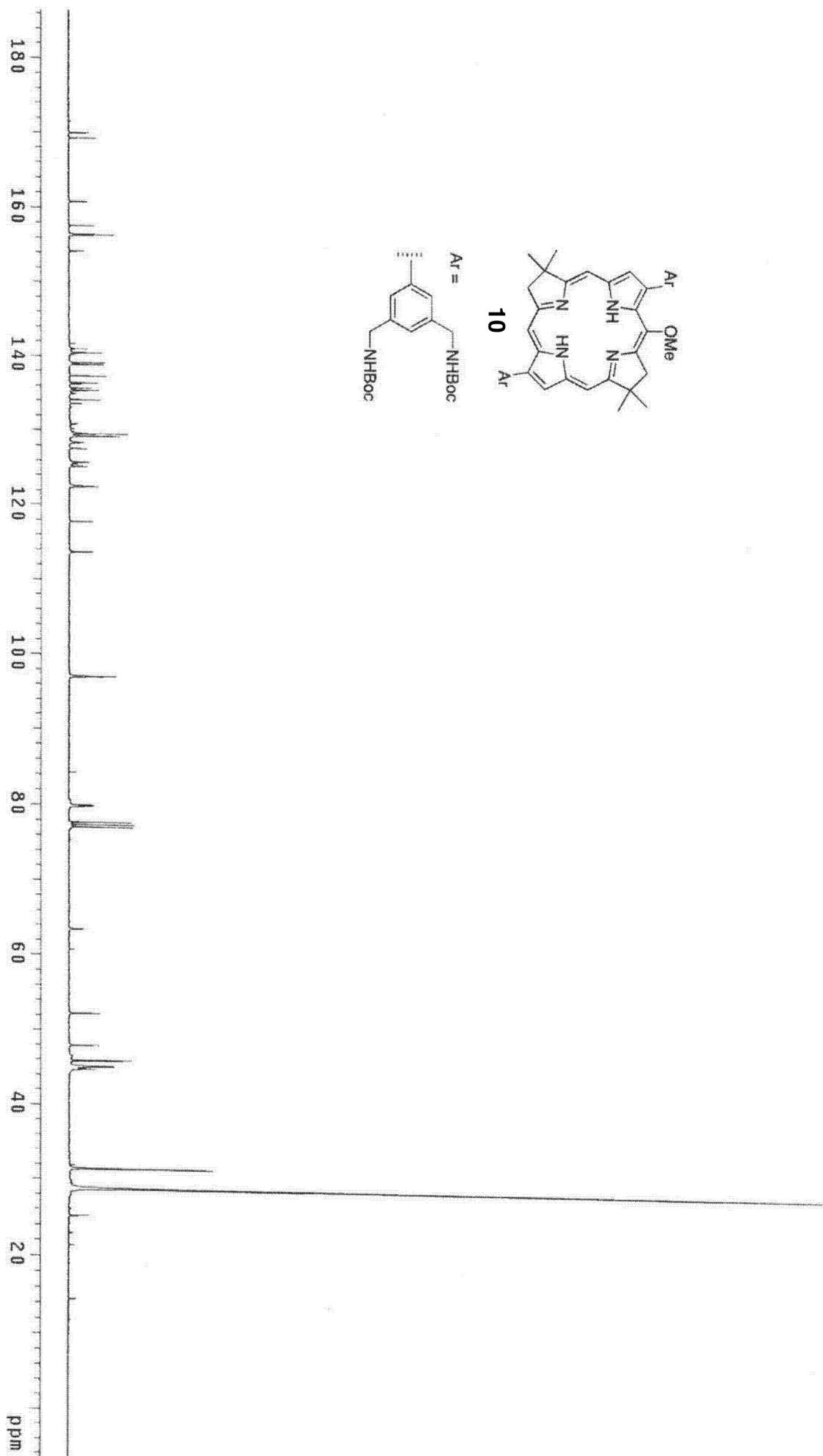
Report generated by : jon

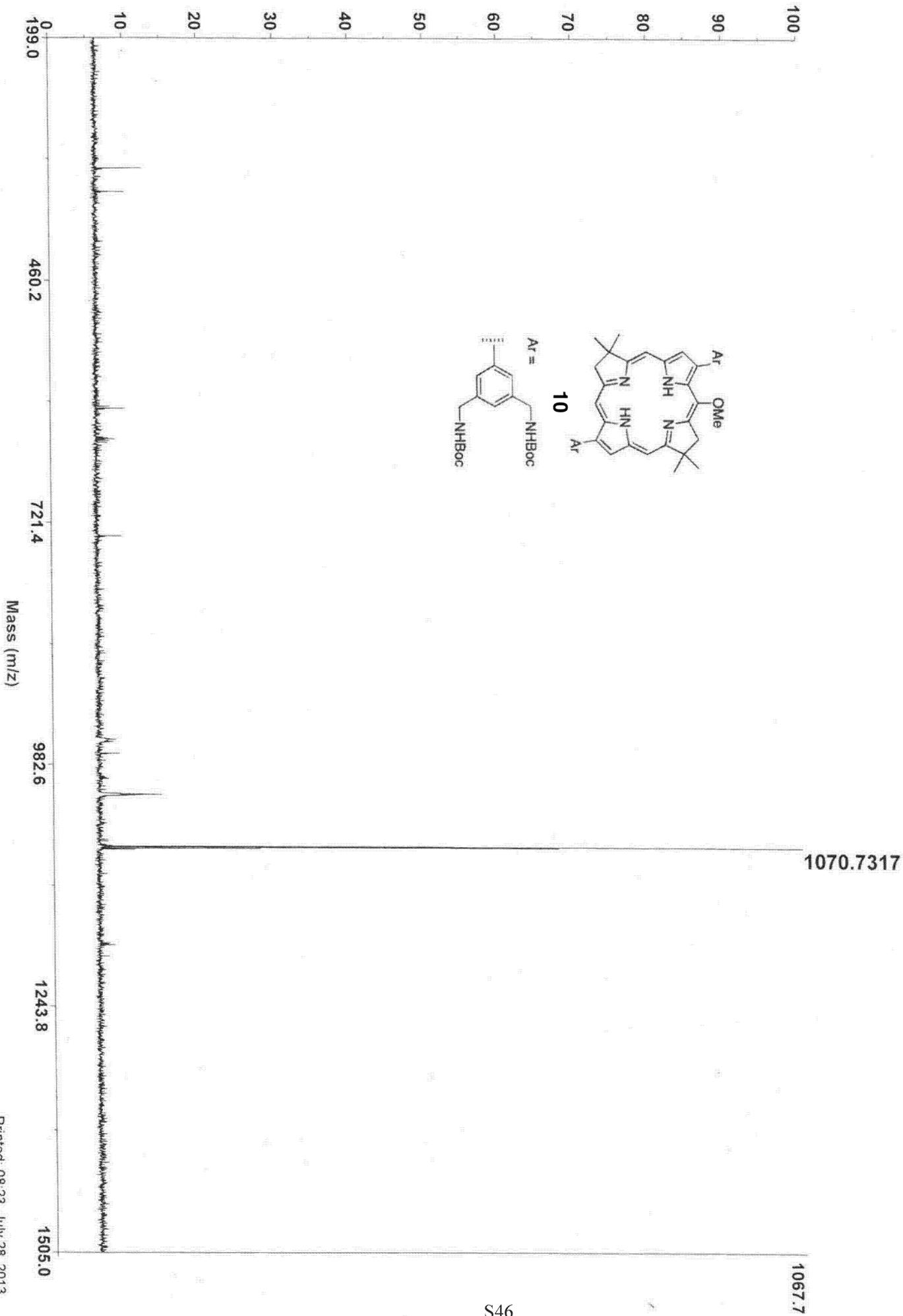
Signature:

*** End Spectrum/Peak Report ***



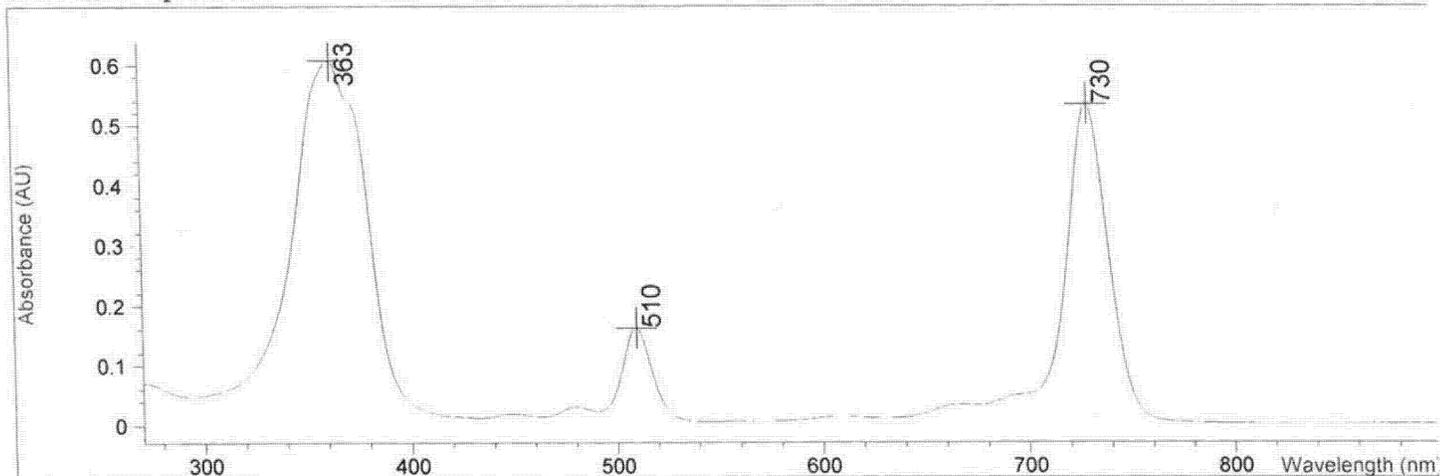






Method file : <untitled>
Information : Default Method
Data File : C:\CHEM32\1\DATA\JIMMY\GROUPED DATA\UVVIS JIMMY A\TEMP\J91, PRO-AM
- RE-TAKEN.SD Created : 4/17/15 14:07:57

Overlaid Spectra:

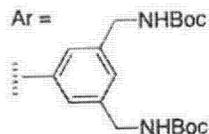
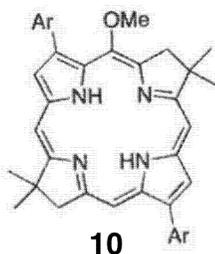


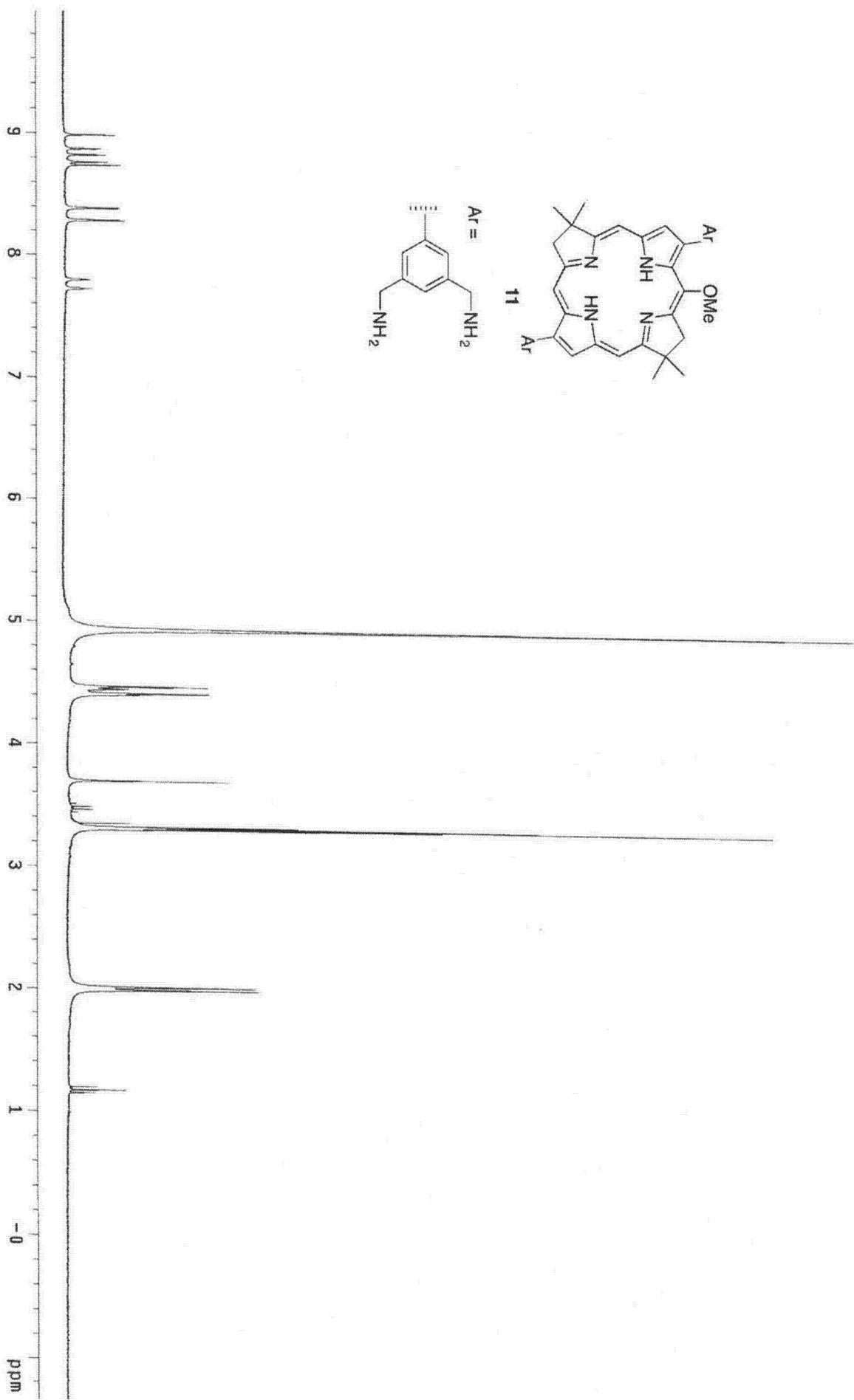
#	Name	Peaks (nm)	Abs (AU)
1		363.0	0.60798
1		730.0	0.53354
1		510.0	0.16280

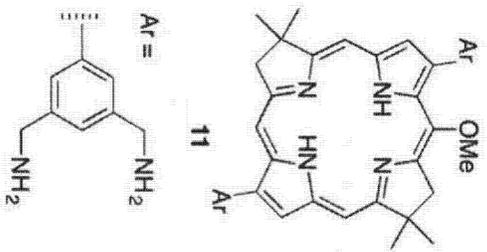
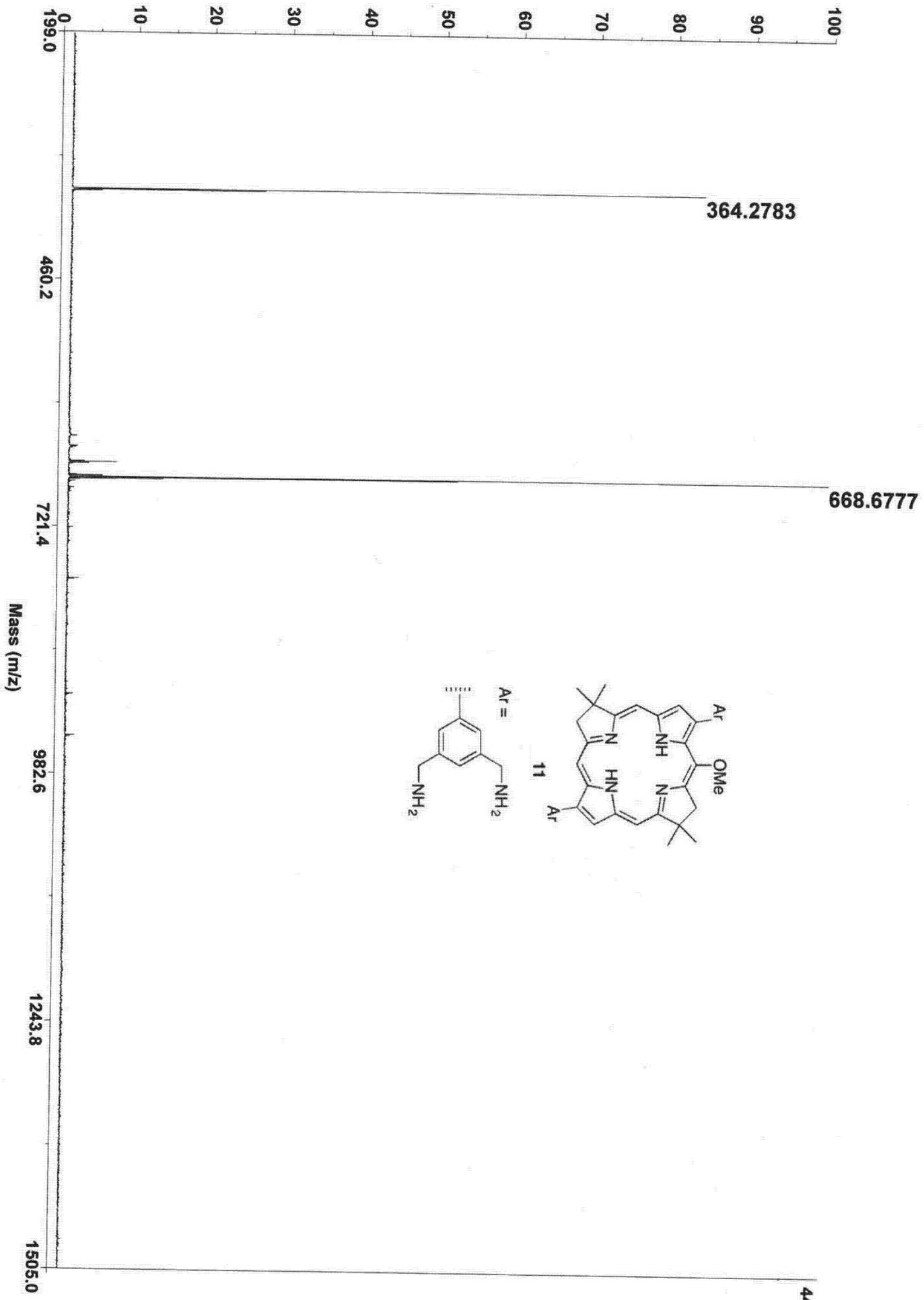
Report generated by : Lindsey Lab

Signature:

*** End Spectrum/Peak Report ***

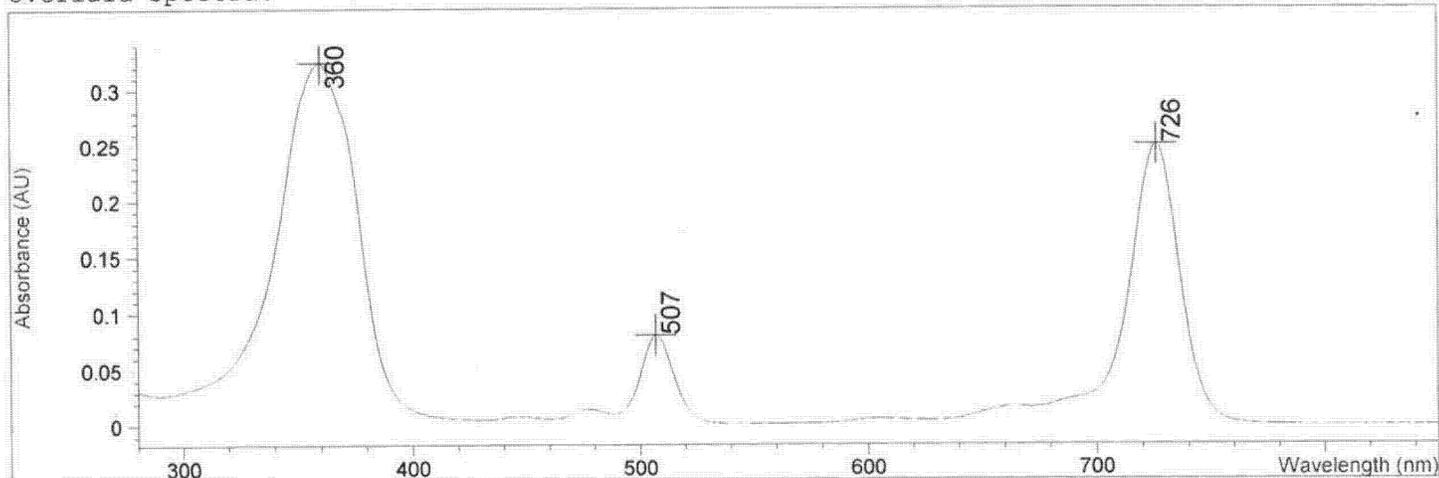






Method file : <untitled>
Information : Default Method
Data File : C:\CHEM32\1\DATA\JIMMY\J93A IN METHANOL, .SD Created : 4/8/15
14:20:55

Overlaid Spectra:

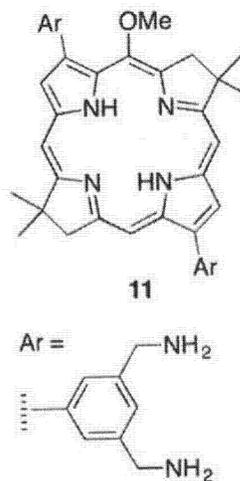


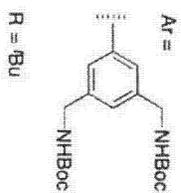
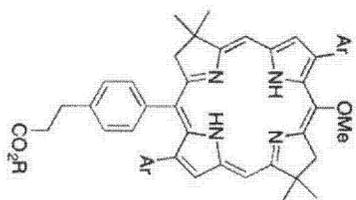
#	Name	Peaks (nm)	Abs (AU)
1		360.0	0.32428
1		726.0	0.24971
1		507.0	8.0328E-2

Report generated by : Lindsey Lab

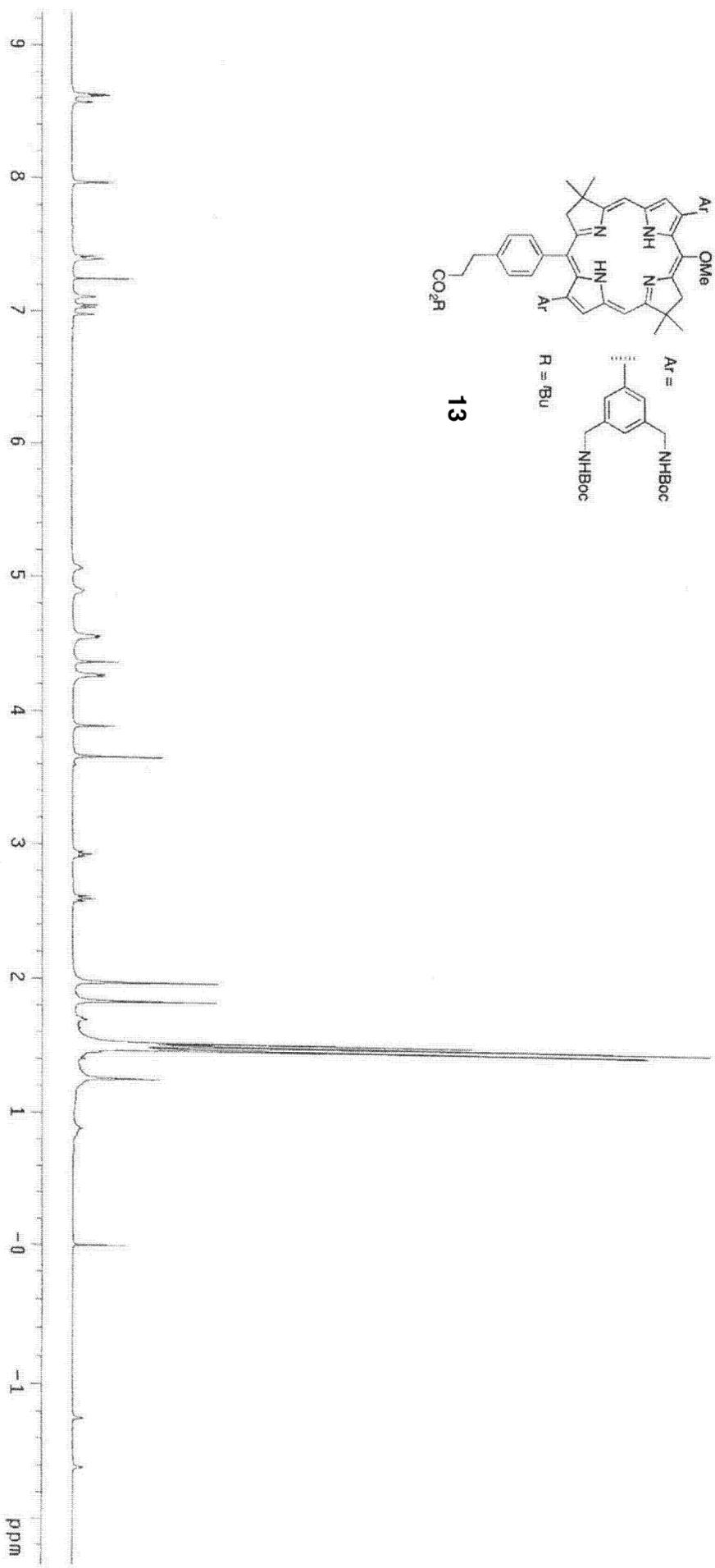
Signature:

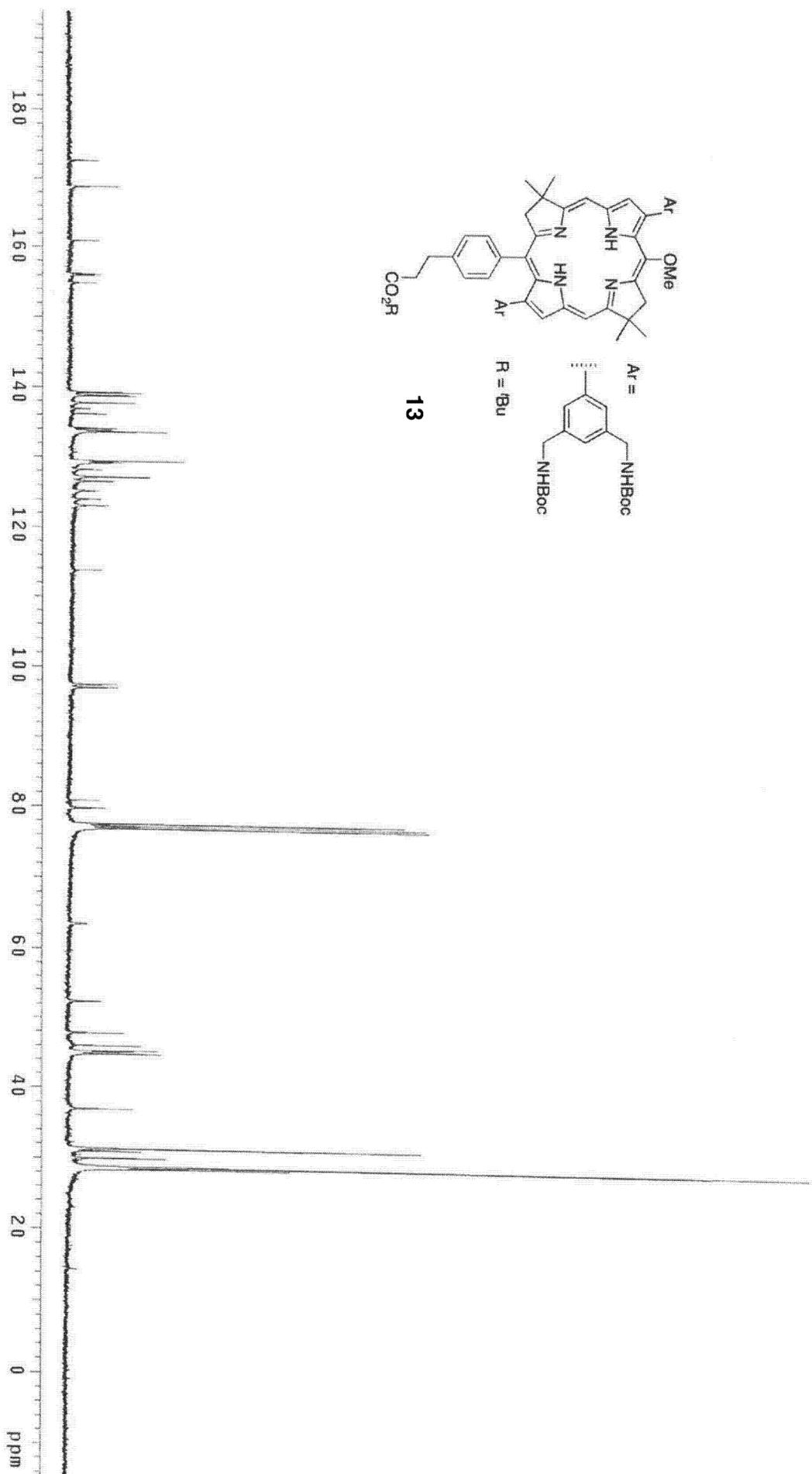
*** End Spectrum/Peak Report ***

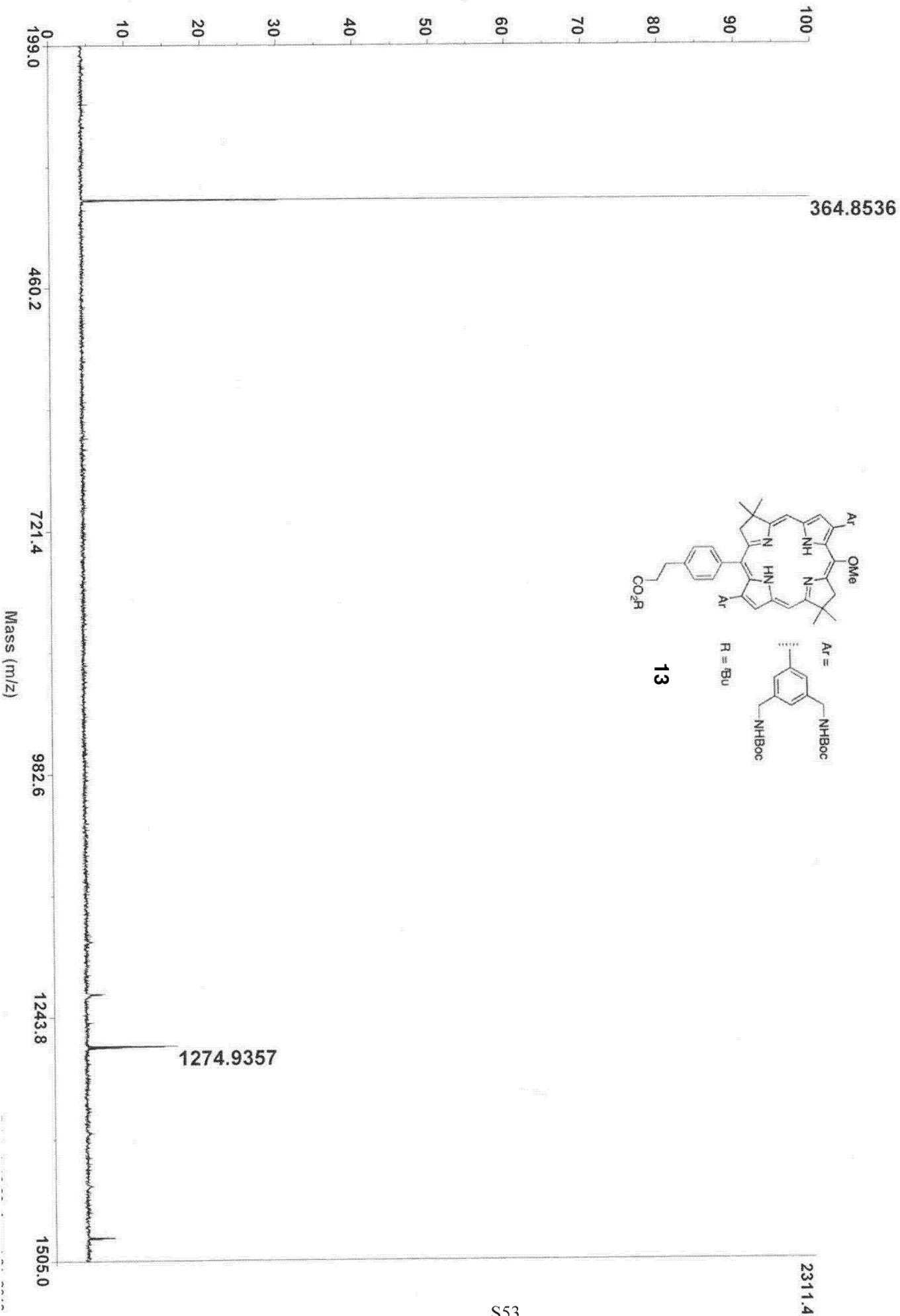




13

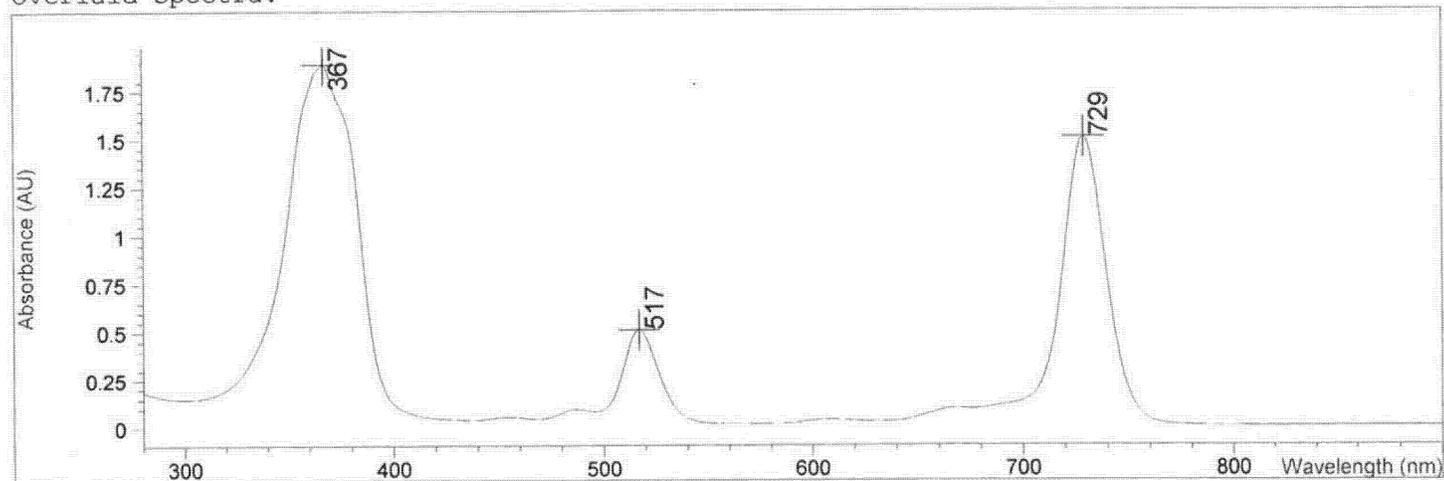






Method file : <untitled>
Information : Default Method
Data File : C:\Chem32\1\DATA\JIMMY\grouped data\J94, PRO-AM-CO2TBU.SD
Created : 8/23/13 14:03:18

Overlaid Spectra:

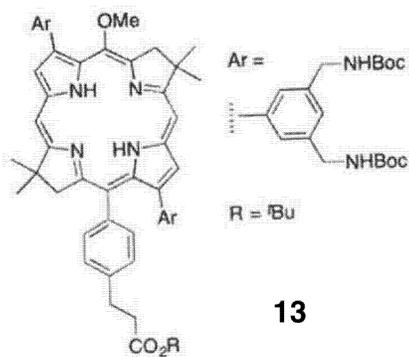


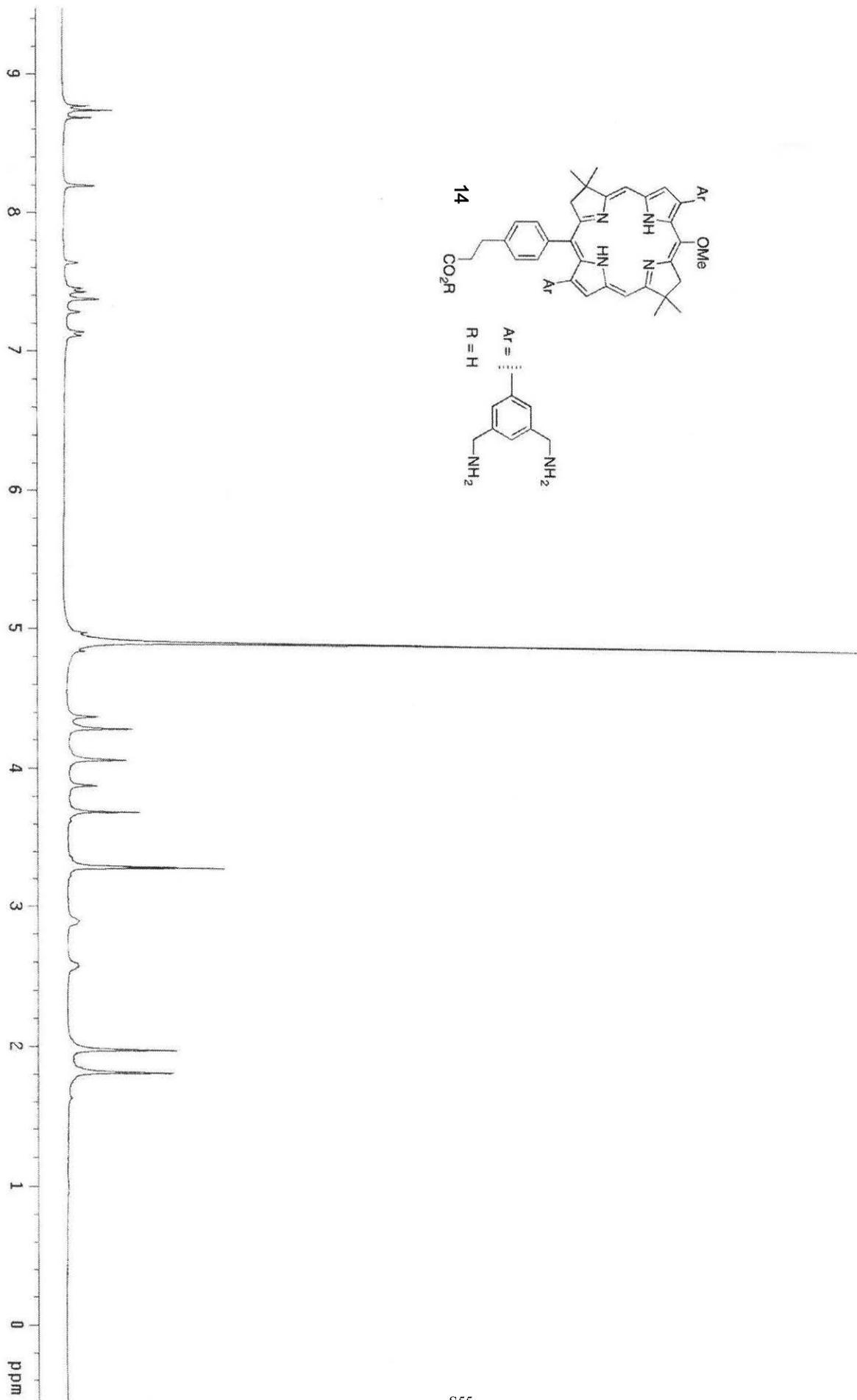
#	Name	Peaks (nm)	Abs (AU)
1		367.0	1.89200
1		729.0	1.50690
1		517.0	0.51017

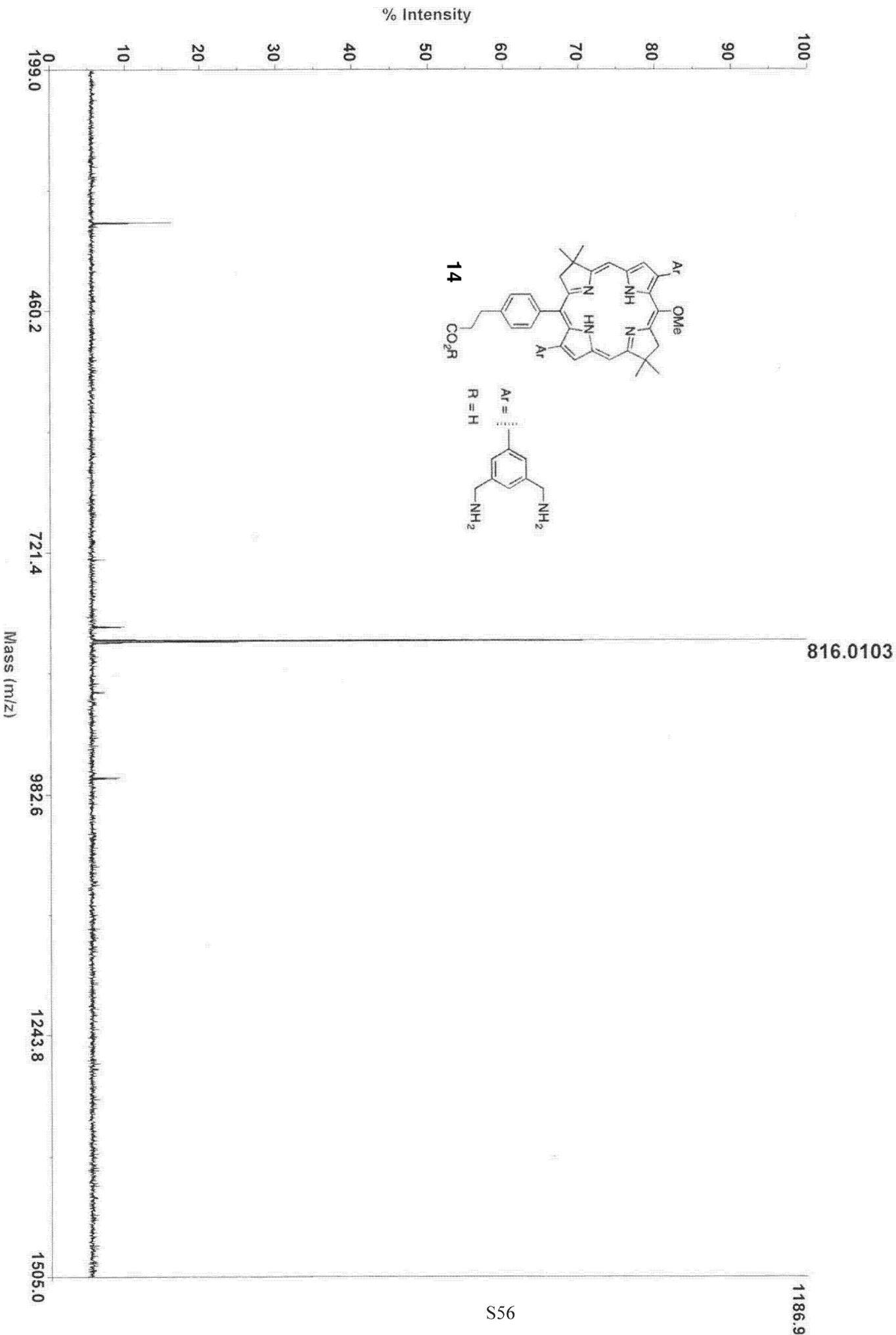
Report generated by : Lindsey Lab

Signature:

*** End Spectrum/Peak Report ***

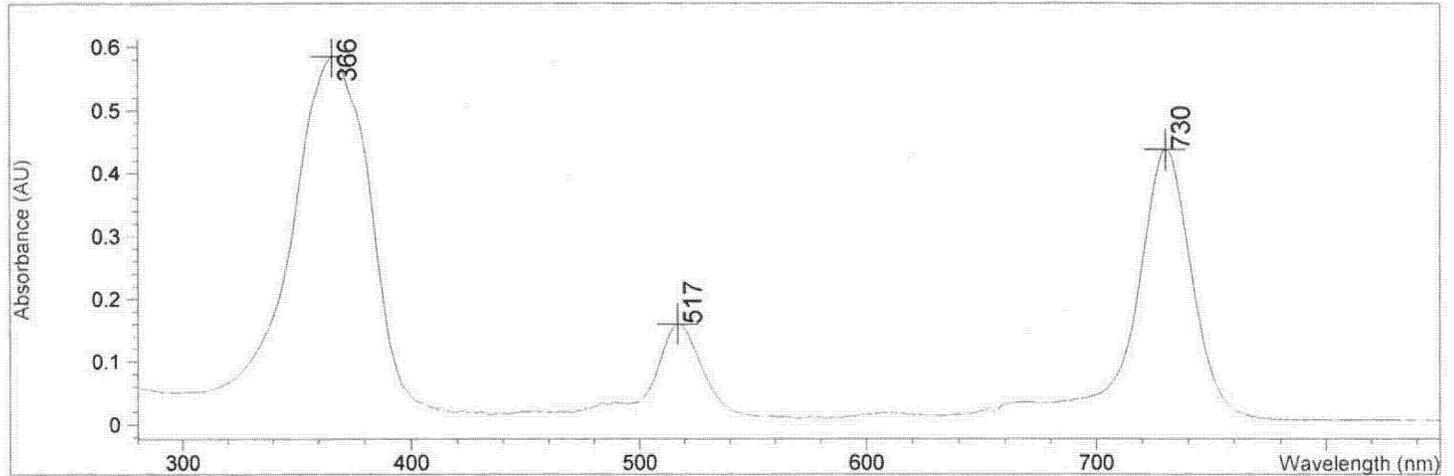






Method file : <untitled>
Information : Default Method
Data File : C:\Chem32\1\DATA\JIMMY\grouped data\J115A, AM-COOH IN METHANOL.SD
Created : 9/5/13 10:31:21

Overlaid Spectra:

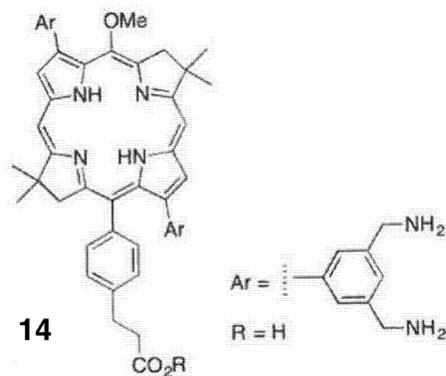


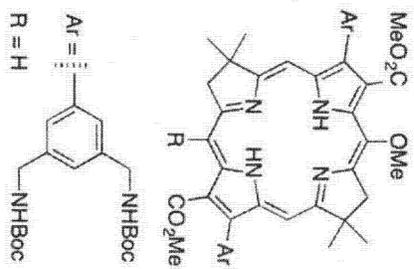
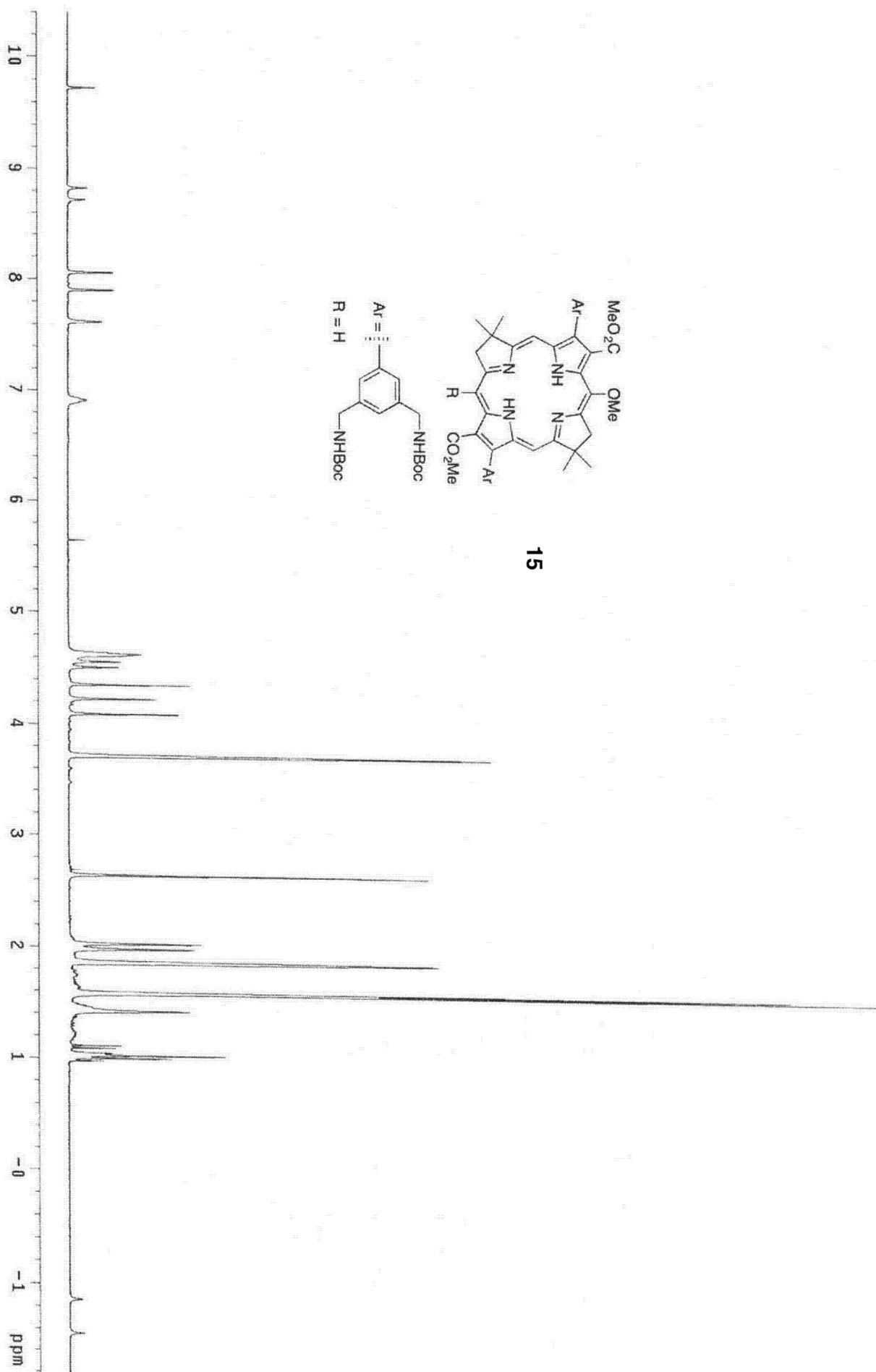
#	Name	Peaks (nm)	Abs (AU)
1		366.0	0.58456
1		730.0	0.43757
1		517.0	0.15903

Report generated by : Lindsey Lab

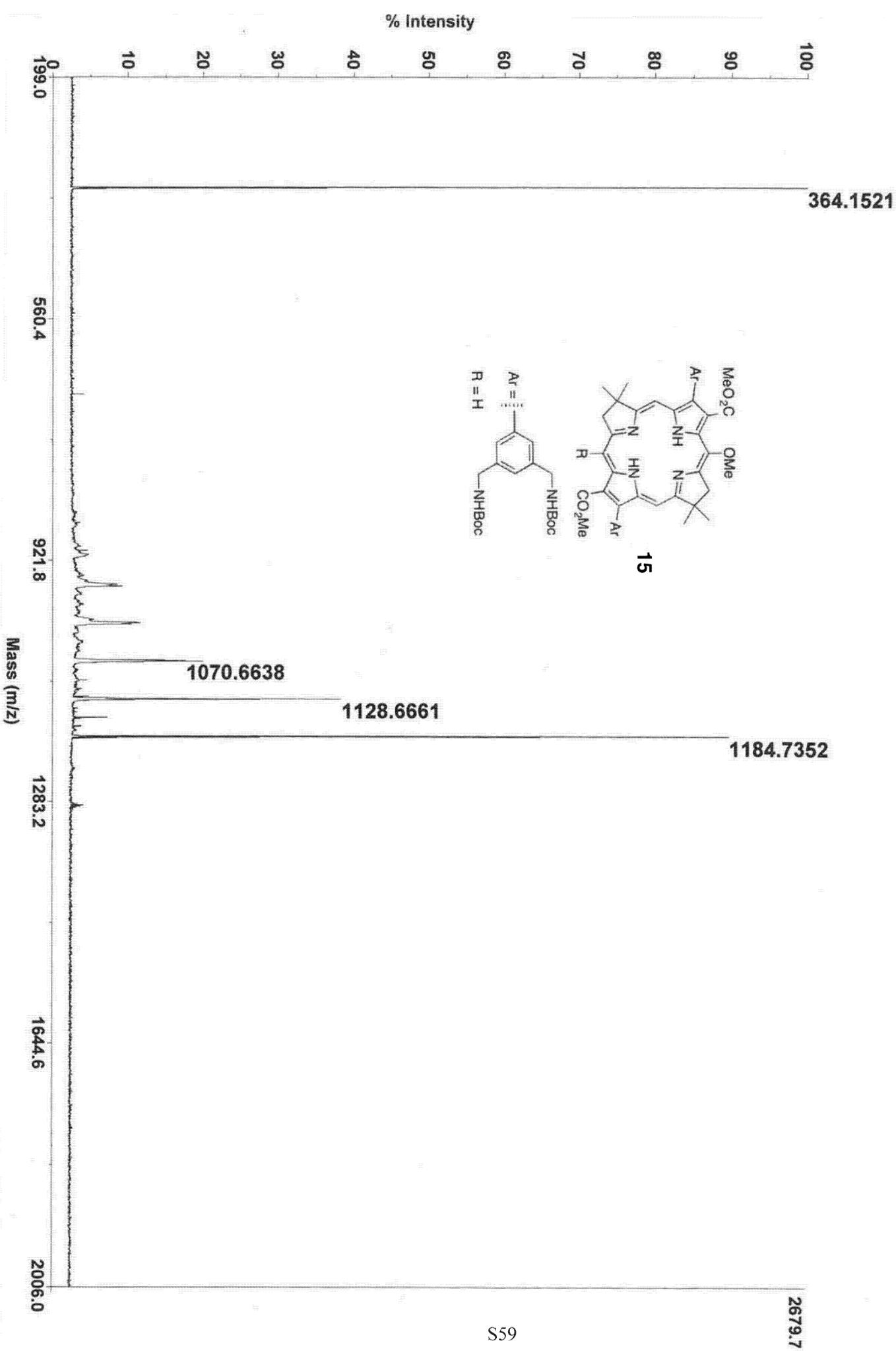
Signature:

*** End Spectrum/Peak Report ***



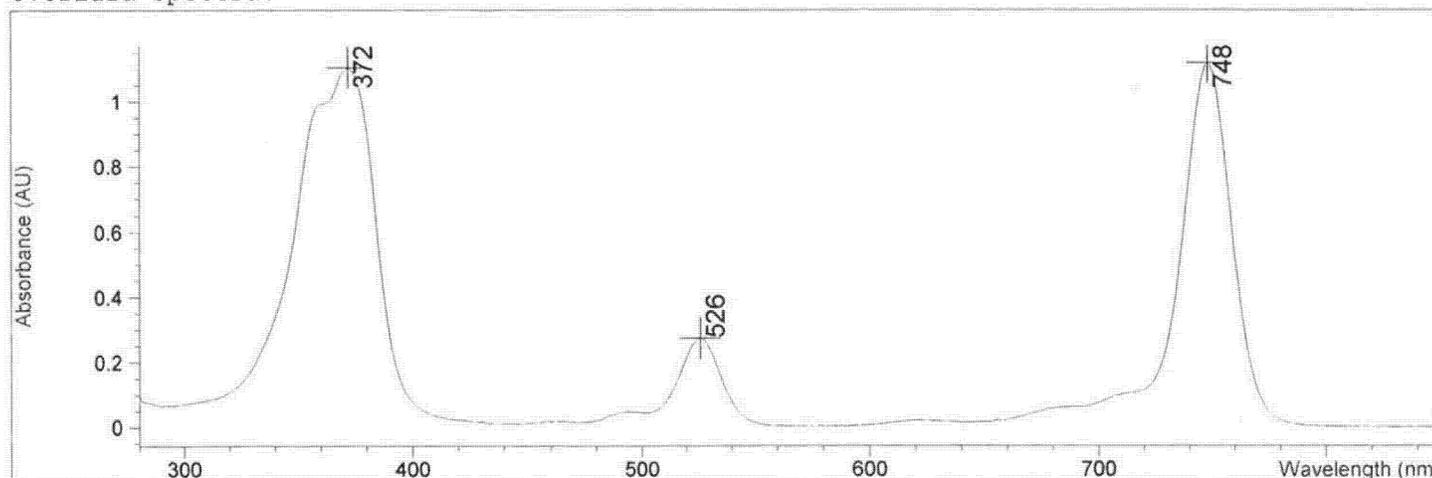


15



Method file : <untitled>
Information : Default Method
Data File : C:\Chem32\1\DATA\JIMMY\grouped data\J123, C-ES-NHBOC IN THF.SD
Created : 2/12/14 15:24:16

Overlaid Spectra:

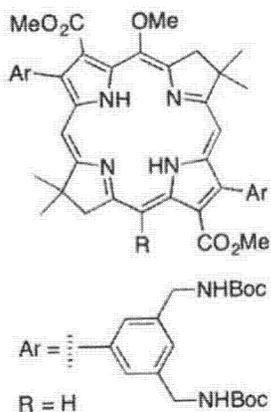


#	Name	Peaks (nm)	Abs (AU)
1		748.0	1.11890
1		372.0	1.10520
1		526.0	0.27391

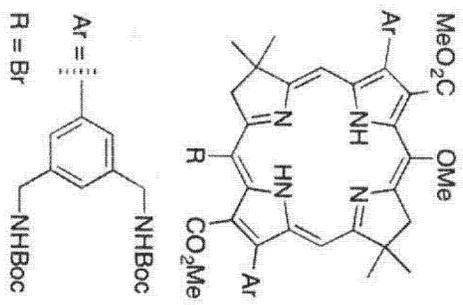
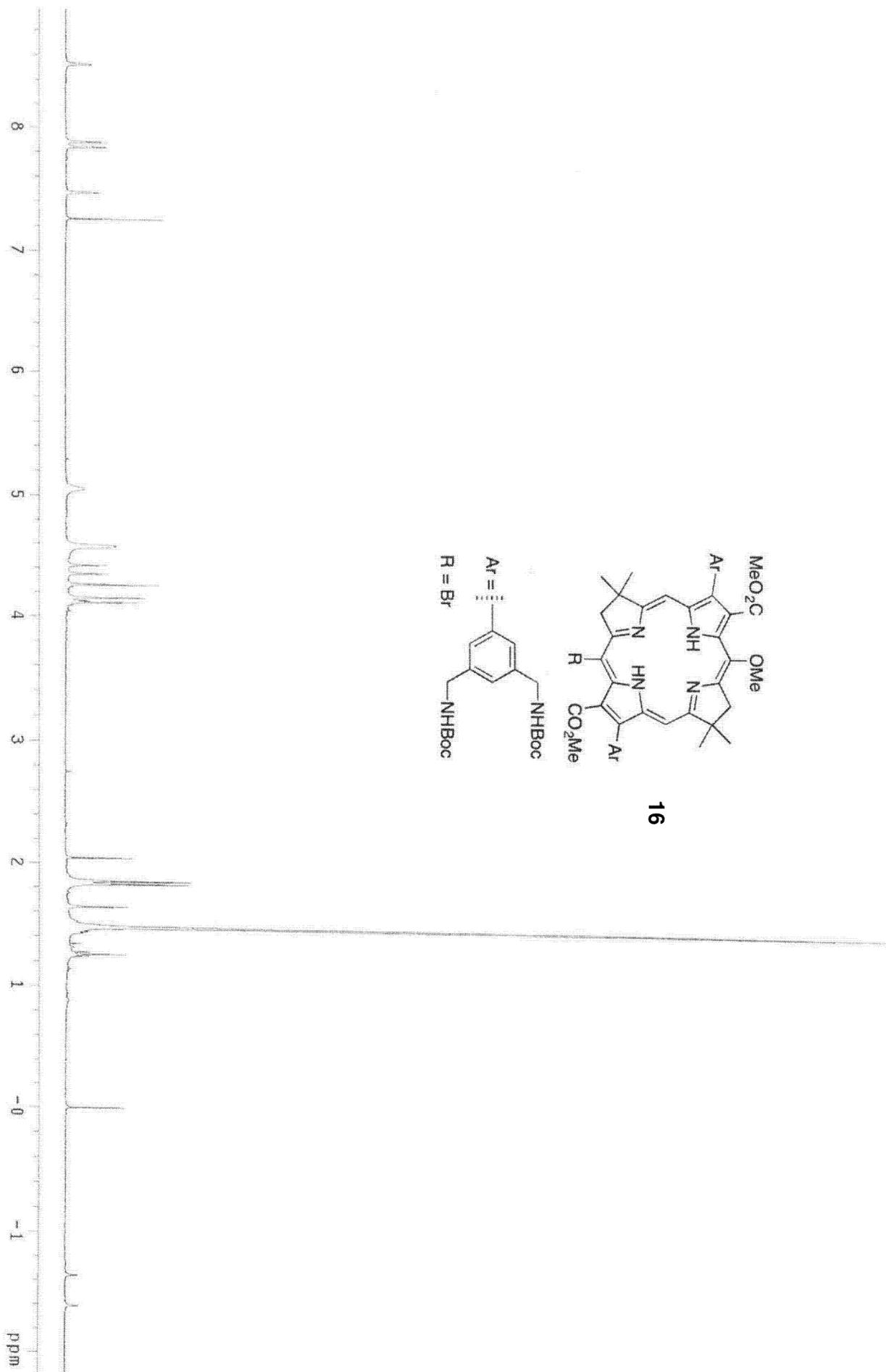
Report generated by : Lindsey Lab

Signature:

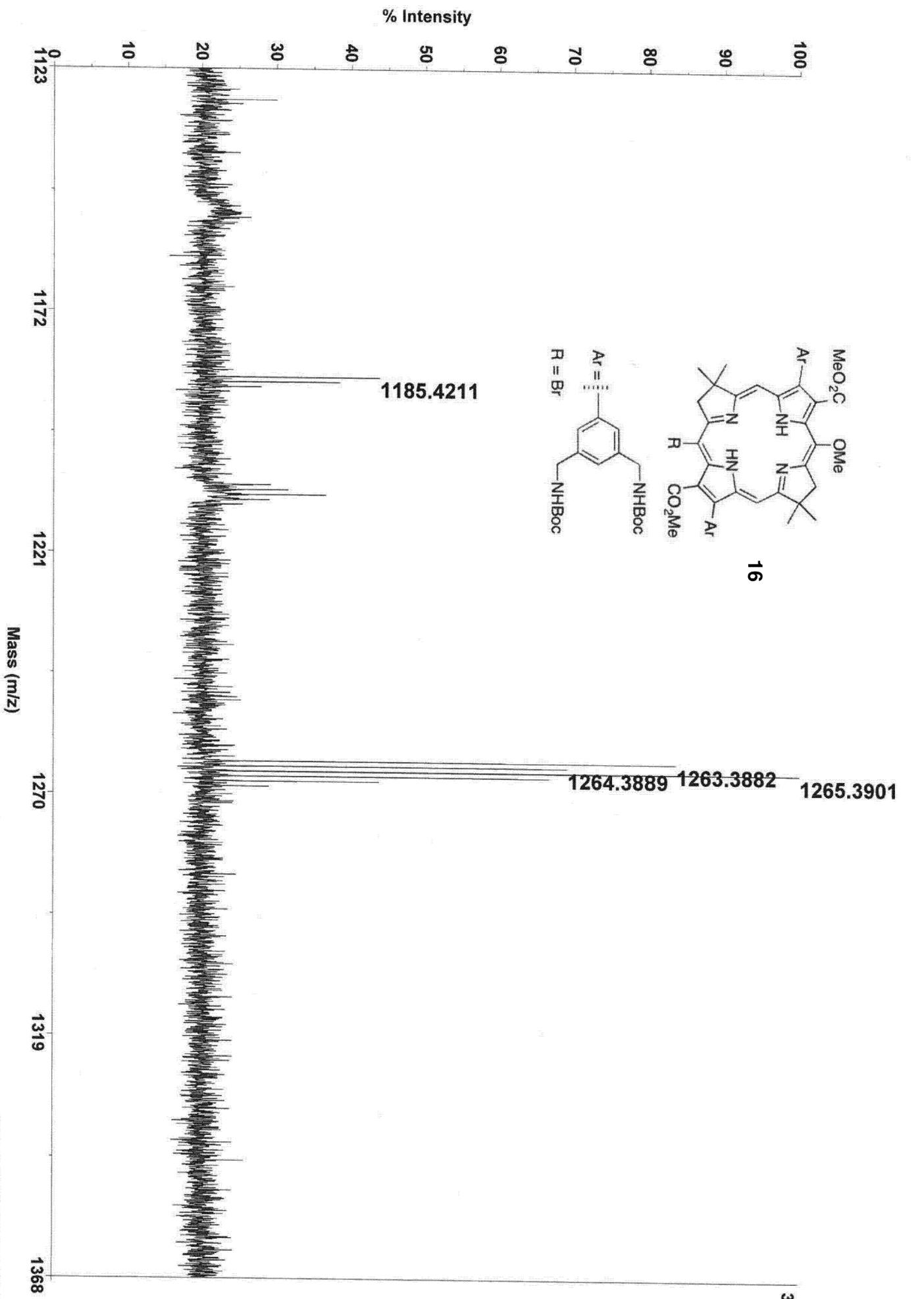
*** End Spectrum/Peak Report ***

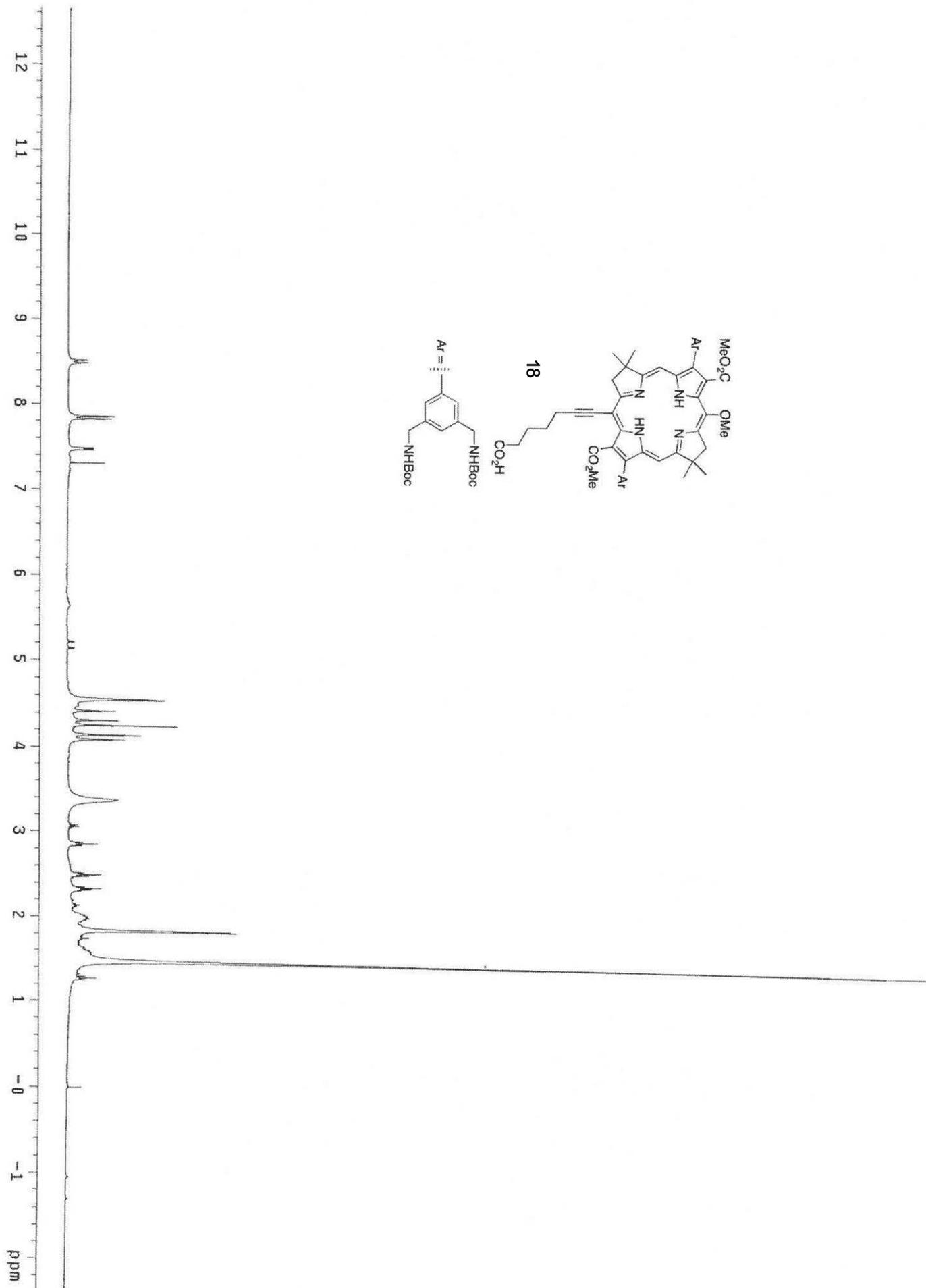


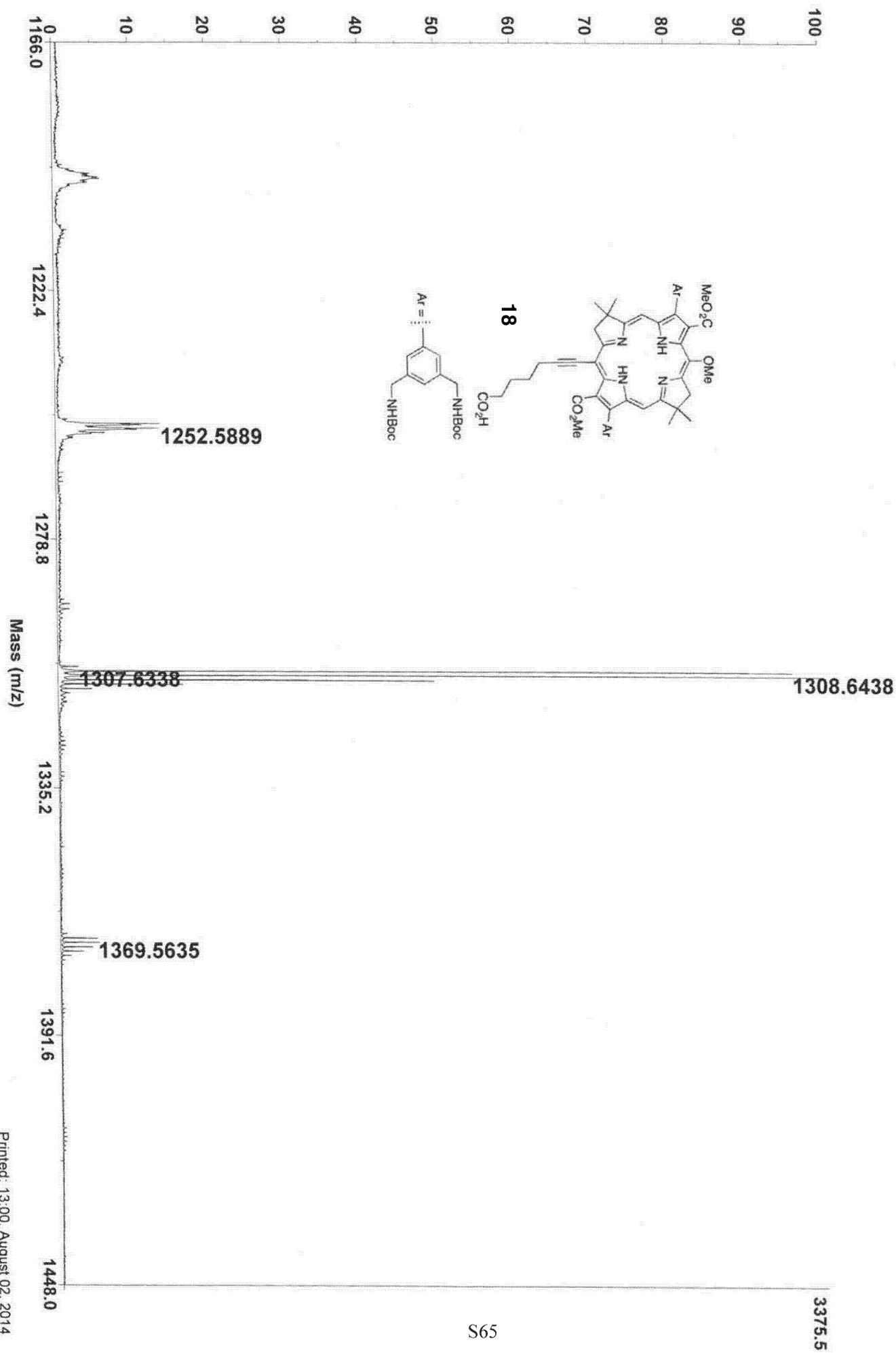
15



16

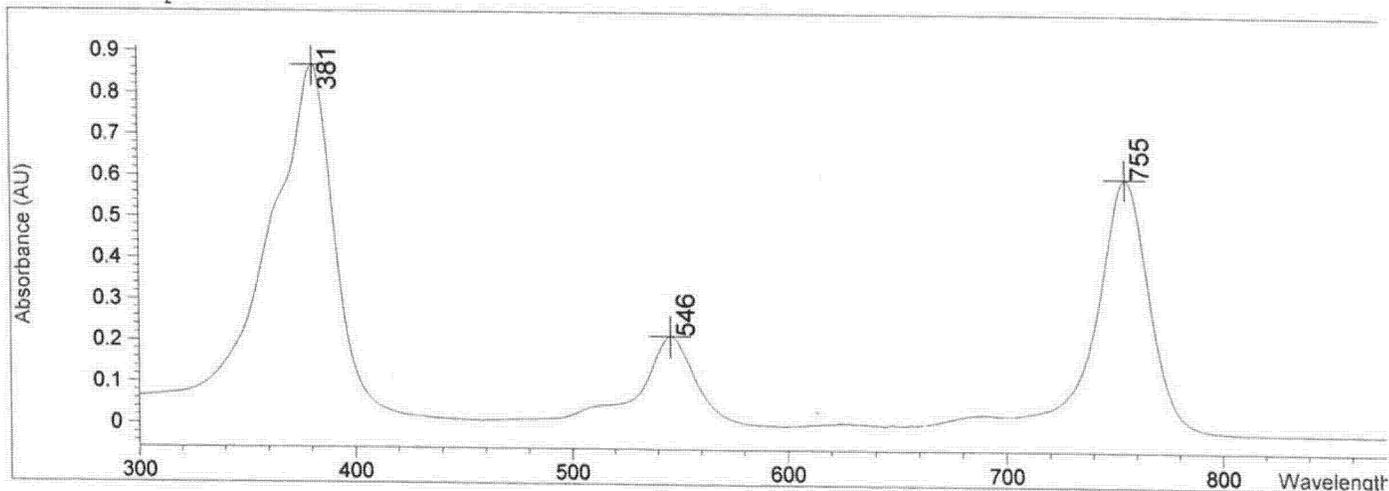






Method file : <method not saved>
Information : Default Method
Data File : C:\Chem32\1\DATA\nuonuo\20140728\PRO-C-ES-YNL-COOH-NHBoc (DCM).SD
Created : 7/29/14 18:15:13

Overlaid Spectra:

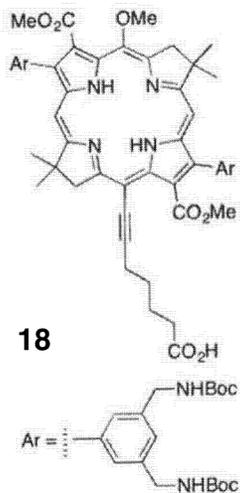


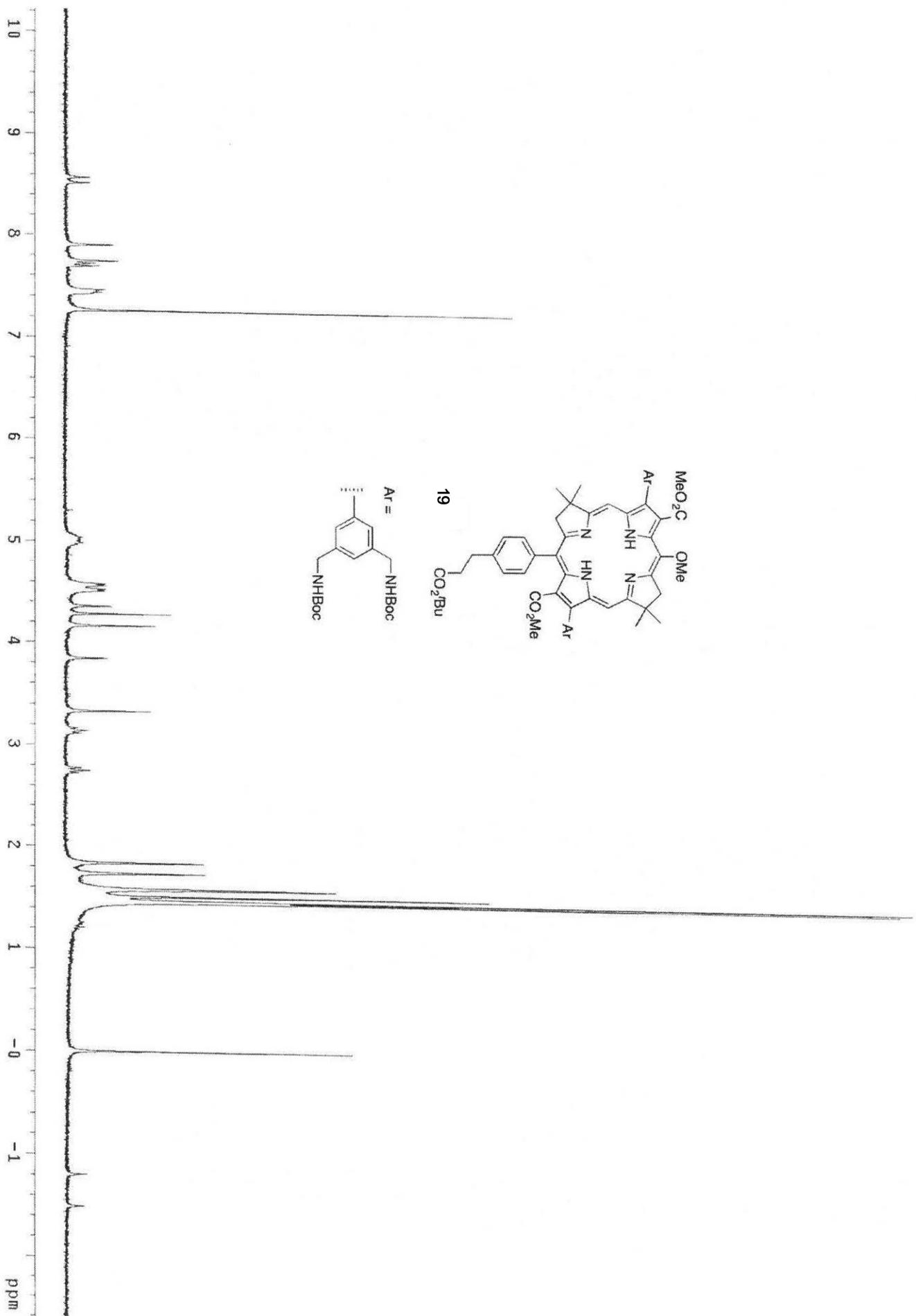
#	Name	Peaks (nm)	Abs (AU)
1		381.0	0.86713
1		755.0	0.60377
1		546.0	0.21494

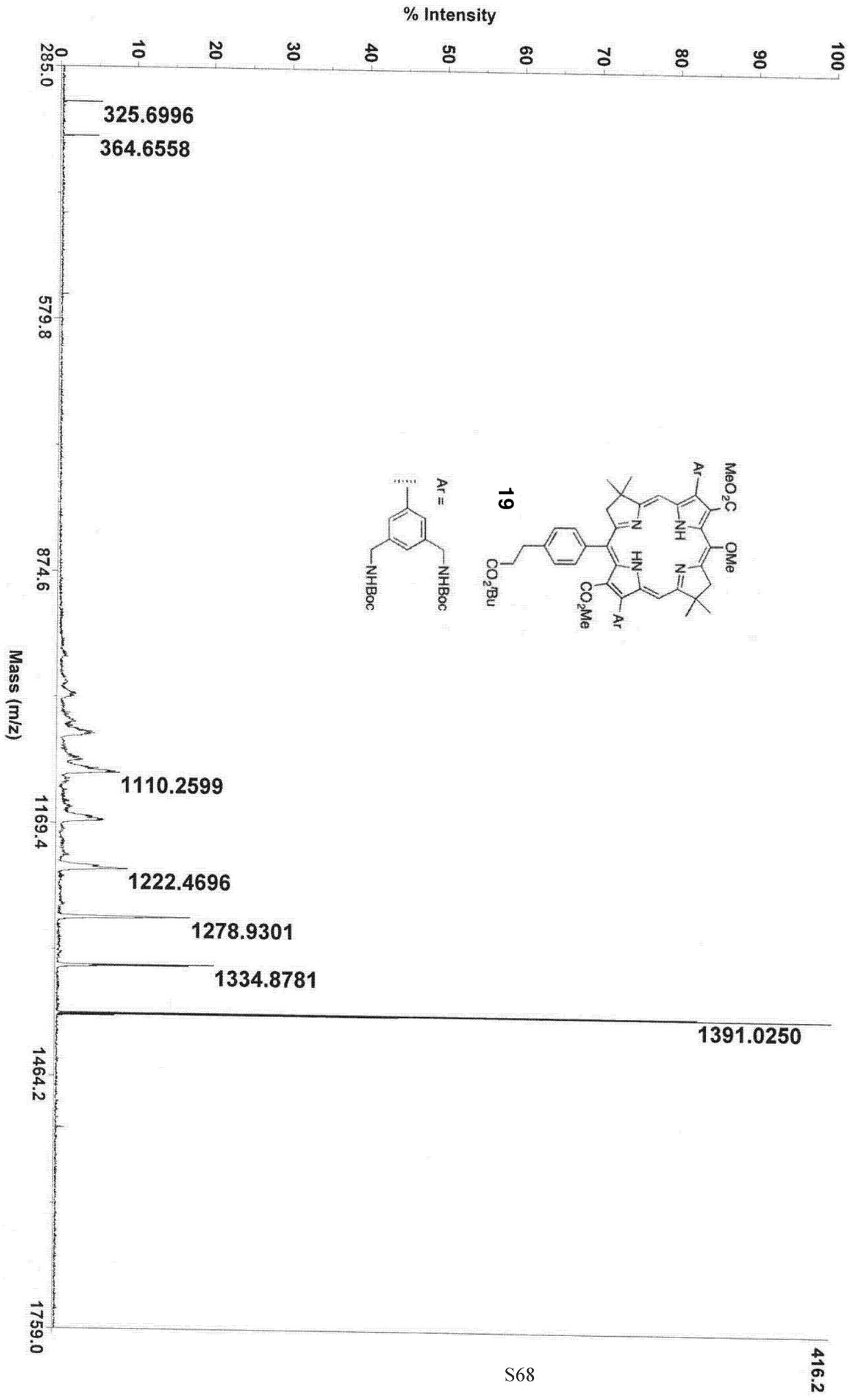
Report generated by : jon

Signature:

*** End Spectrum/Peak Report ***

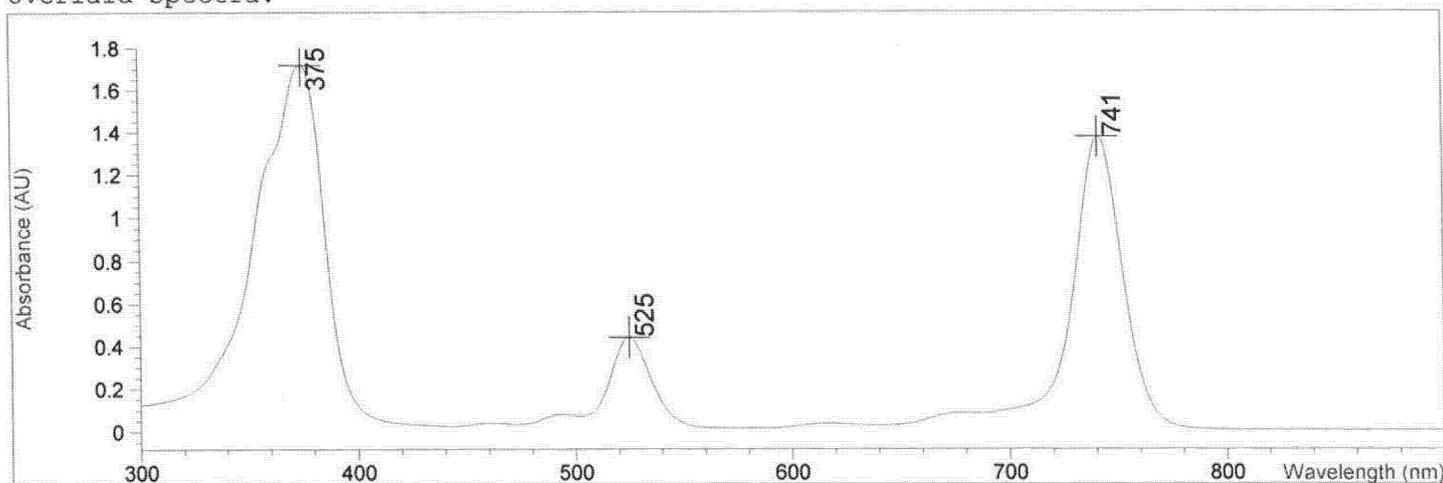






Method file : <method not saved>
Information : Default Method
Data File : C:\CHEM32\1\DATA\NUONUO\20140513\PRO-C-ES-NHBOC-C-COOBUT(DCM).SD
Created : 5/13/14 10:07:17

Overlaid Spectra:

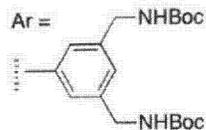
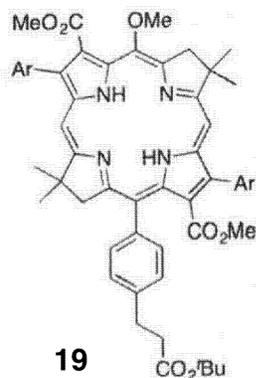


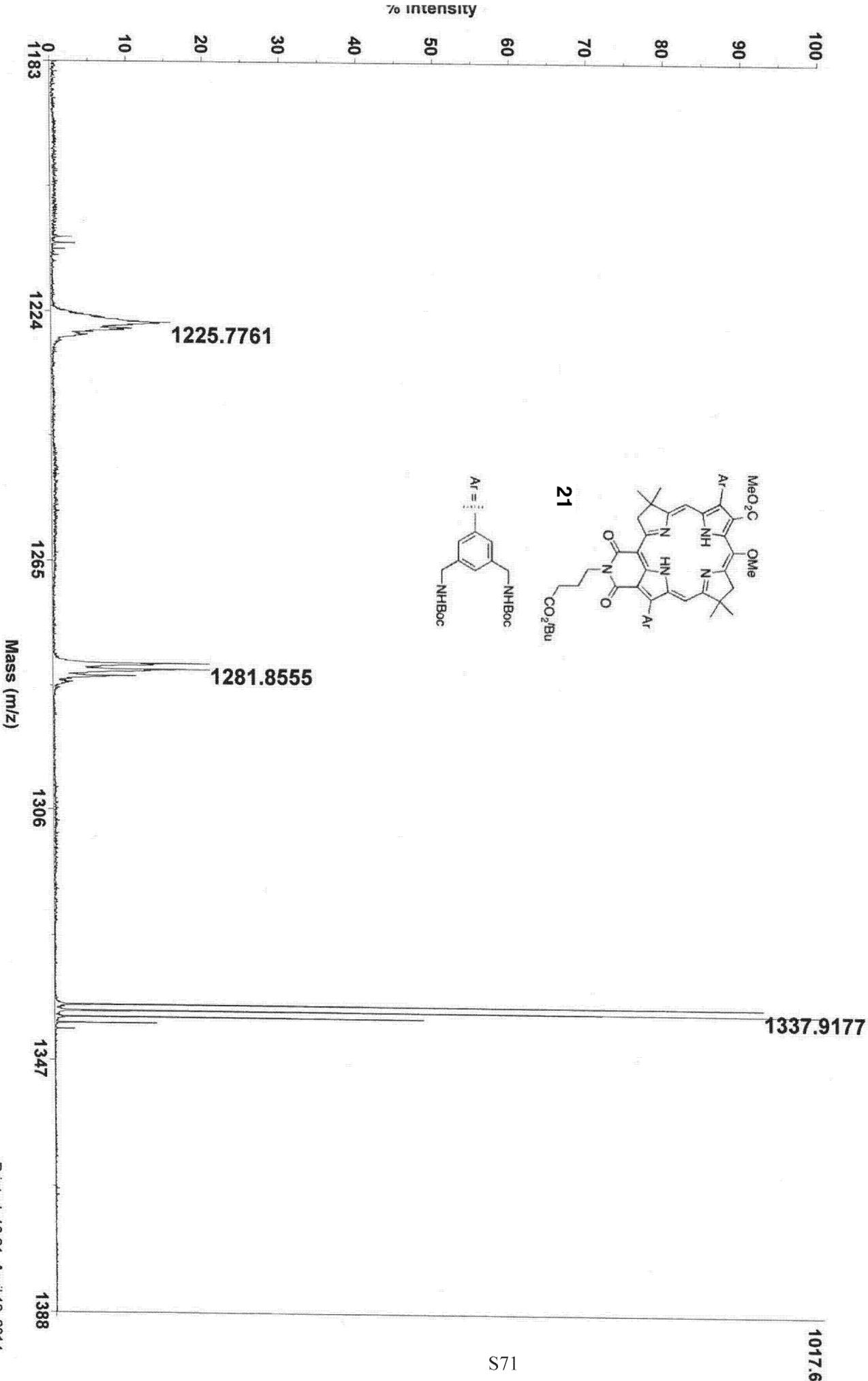
#	Name	Peaks (nm)	Abs (AU)
1		375.0	1.71840
1		741.0	1.37560
1		525.0	0.44172

Report generated by : jon

Signature:

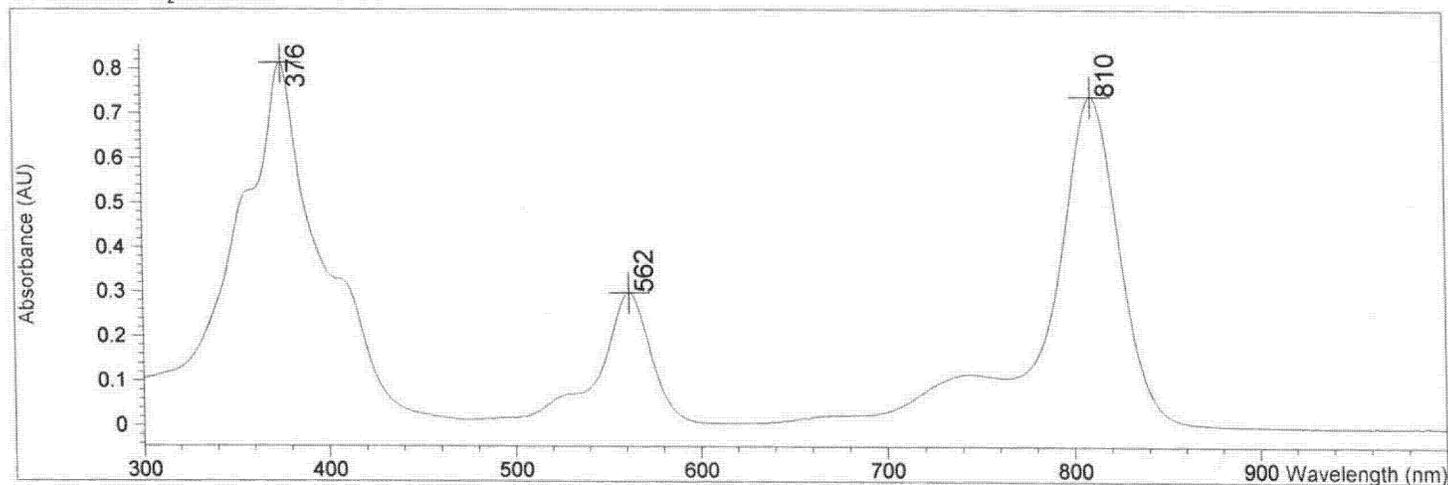
*** End Spectrum/Peak Report ***





Method file : <untitled>
Information : Default Method
Data File : C:\CHEM32\1\DATA\NUONUO\20140411\PRO-C-ES-NHBOC-COOBUT(DCM)-1.SD
Created : 4/11/14 17:58:09

Overlaid Spectra:



#	Name	Peaks (nm)	Abs (AU)
1		376.0	0.81293
1		810.0	0.73882
1		562.0	0.29771

Report generated by : Lindsey Lab

Signature:

*** End Spectrum/Peak Report ***

