

Electronic Supplementary Information

Design, synthesis and optical properties of small molecules based on dithieno[3,2-*b*:2',3'-*d*]stannole and stannafluorene

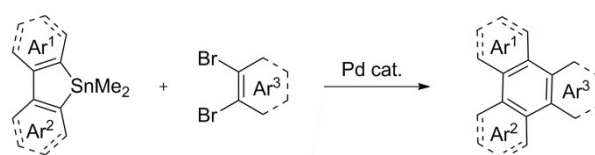
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Scheme S1. The reaction of dialkylstannole derivatives with 1,2-dihaloarenes ^[1].

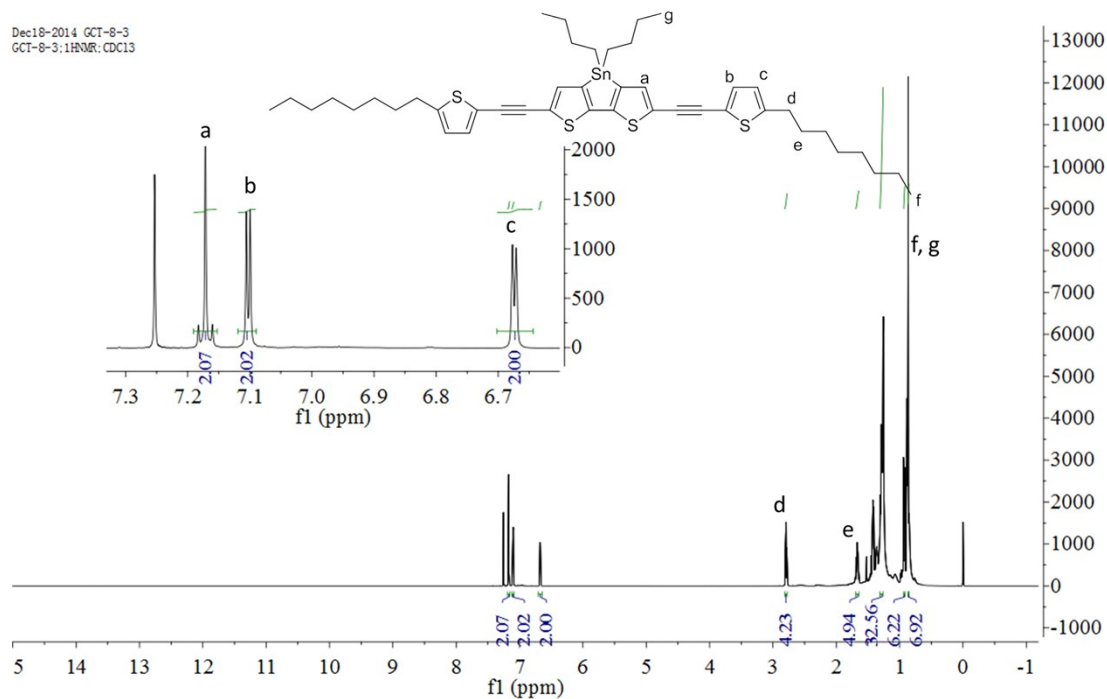


Figure S1. ¹H NMR spectrum of **DTSn-1** in CDCl₃.

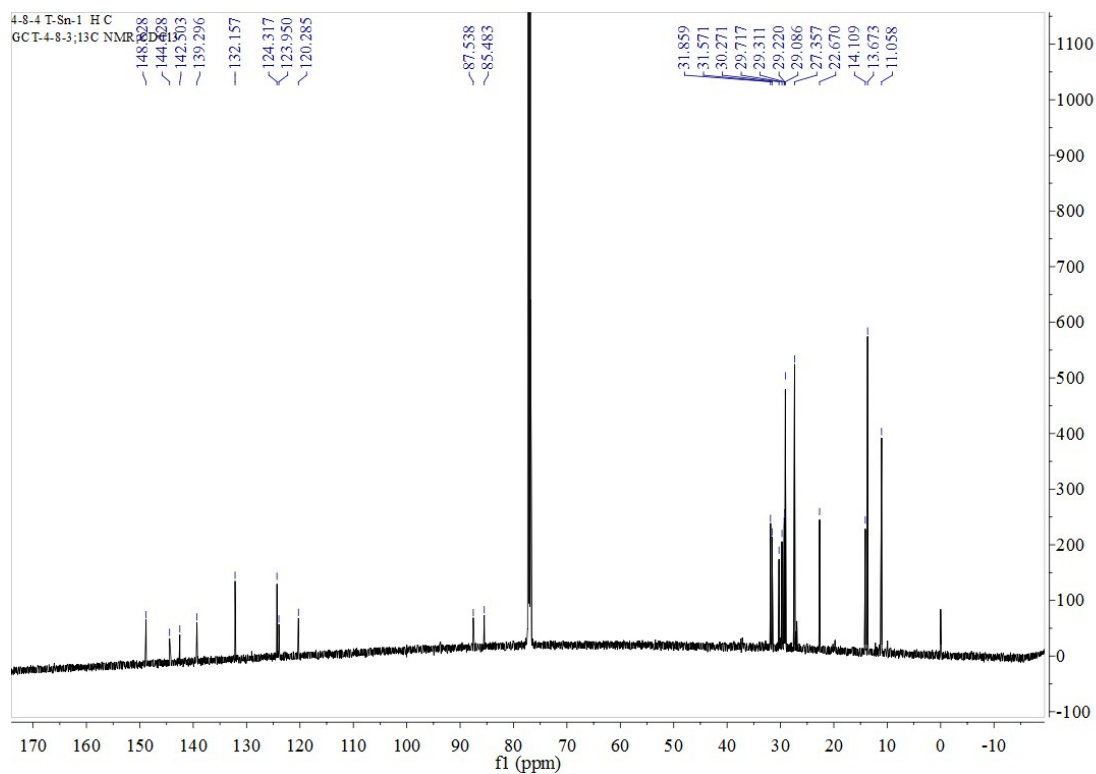


Figure S2. ^{13}C NMR spectrum of **DTSn-1** in CDCl_3 .

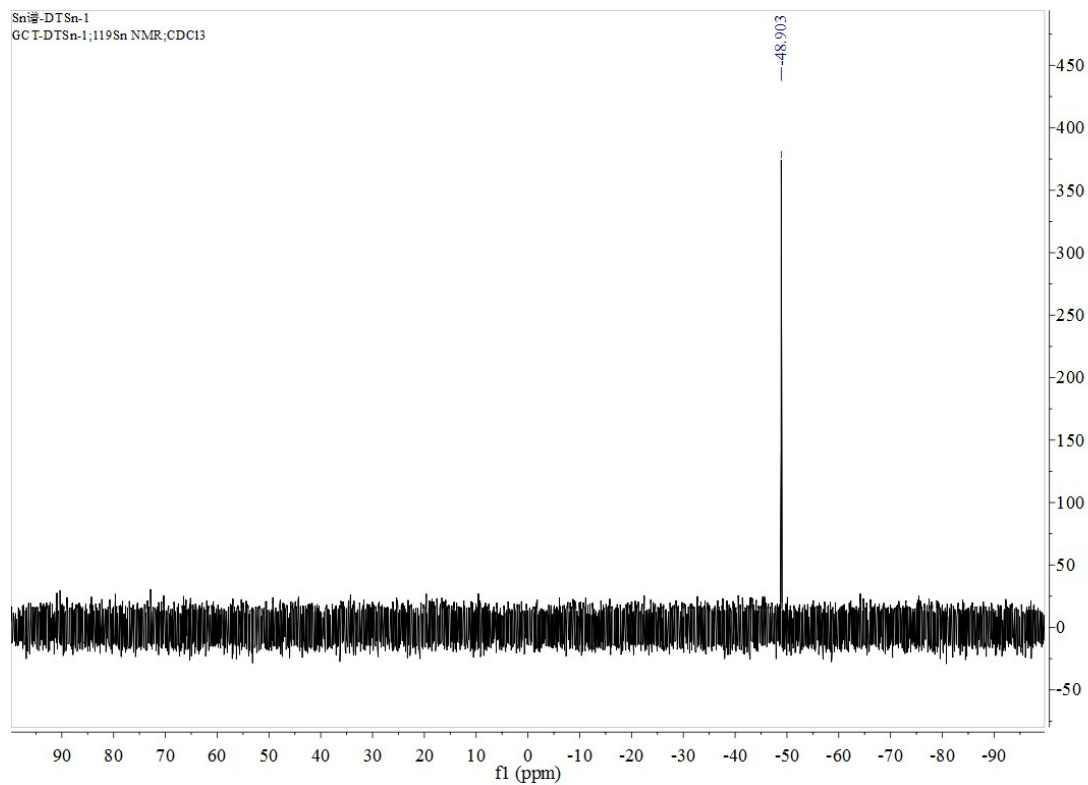


Figure S3. ^{119}Sn NMR spectrum of **DTSn-1** in CDCl_3 .

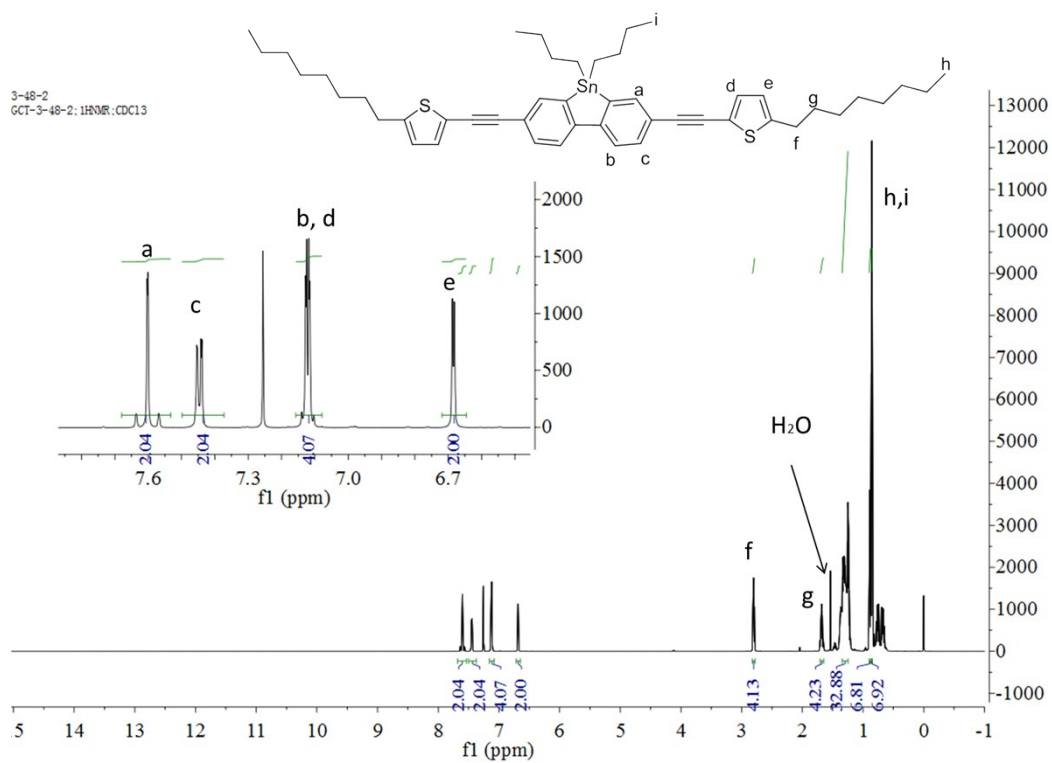


Figure S4. 1H NMR spectrum of **SnF-1** in $CDCl_3$.

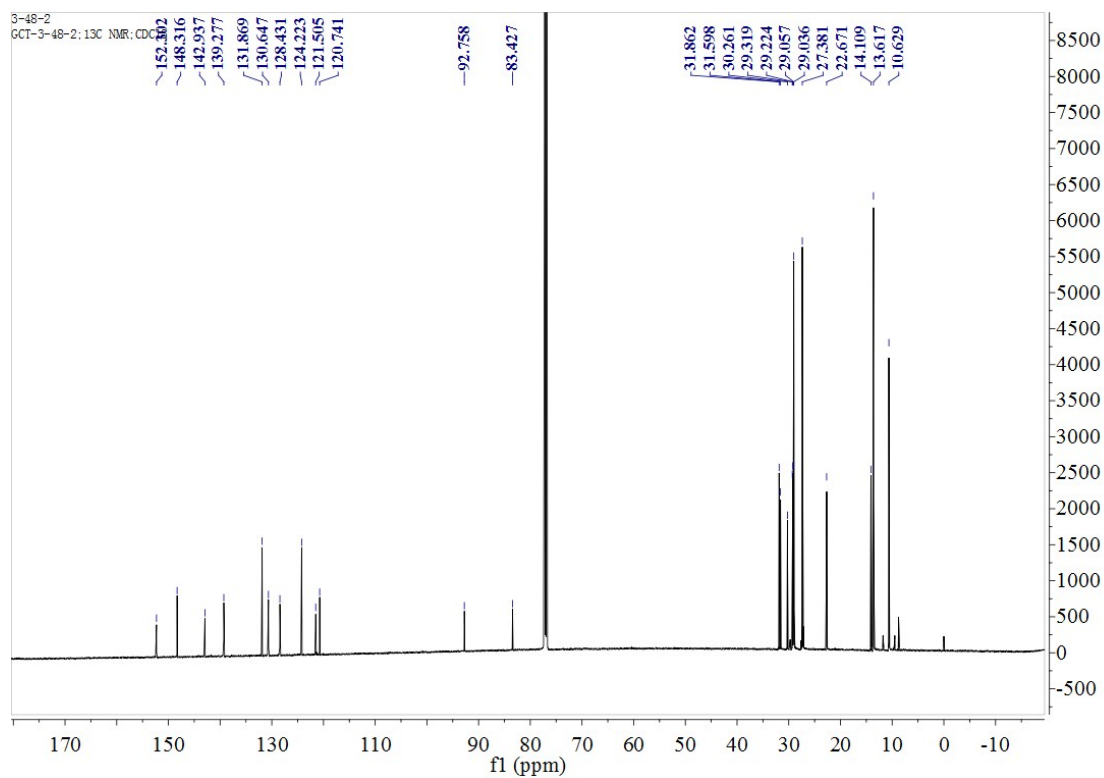
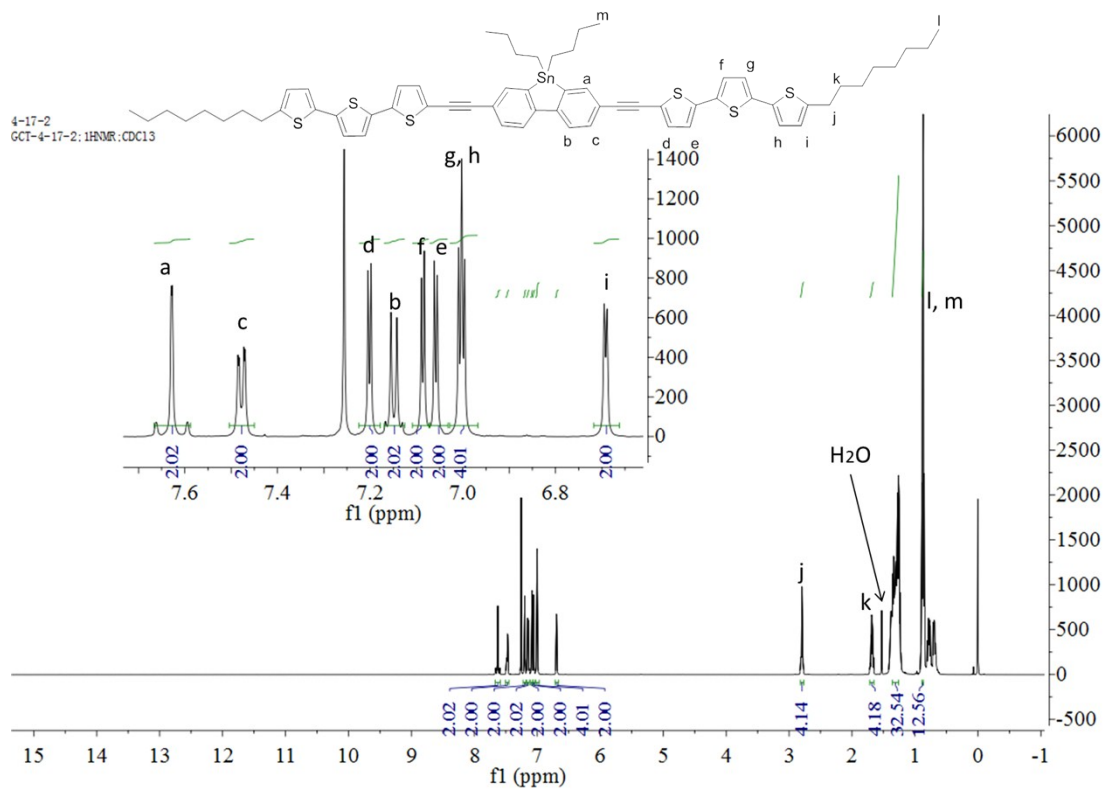
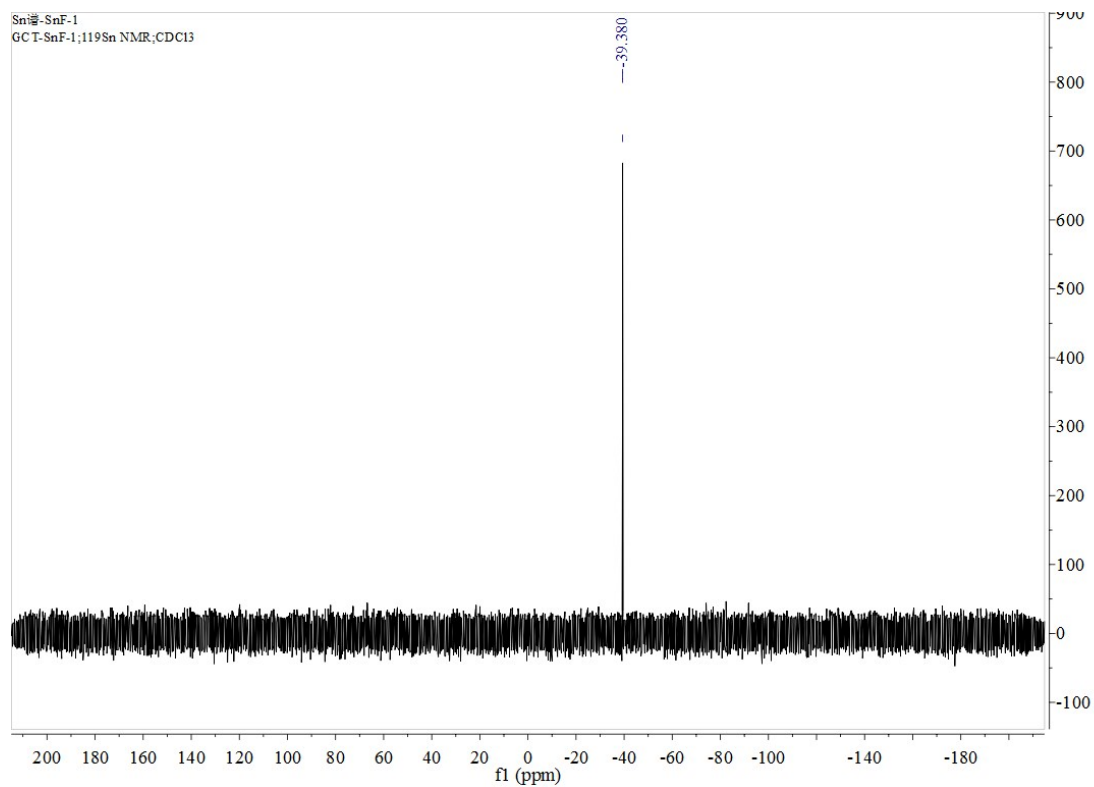


Figure S5. ^{13}C NMR spectrum of **SnF-1** in $CDCl_3$.



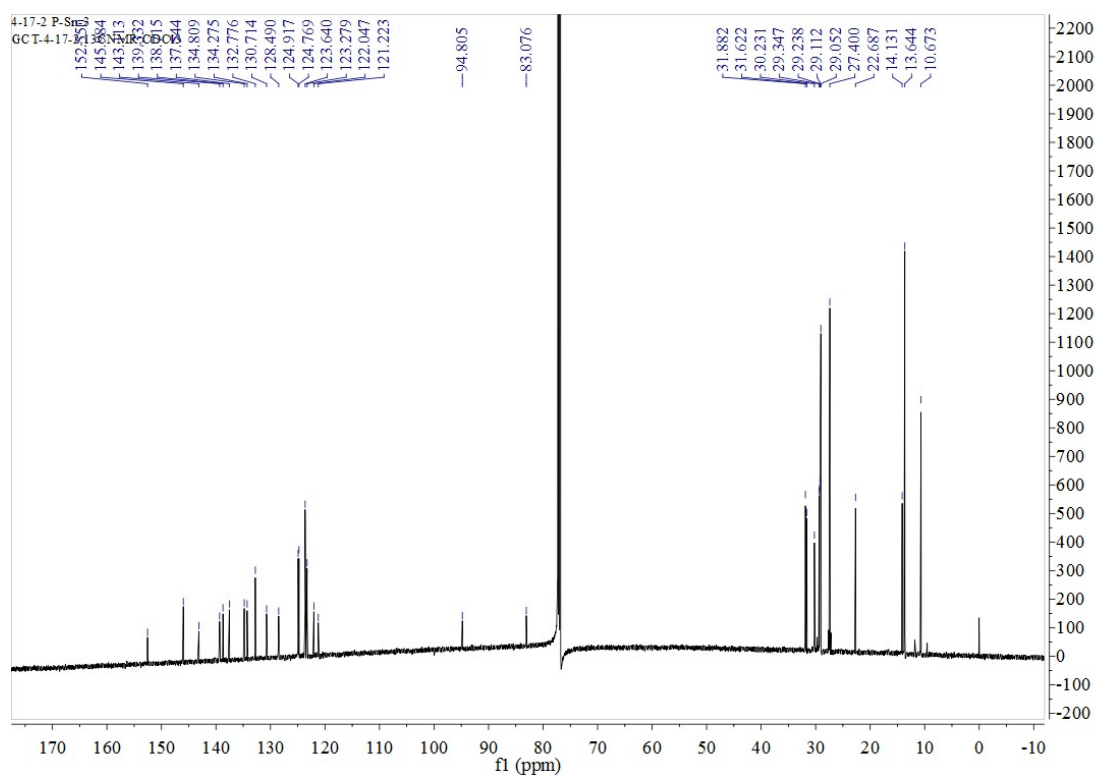


Figure S8. ^{13}C NMR spectrum of **SnF-3** in CDCl_3 .

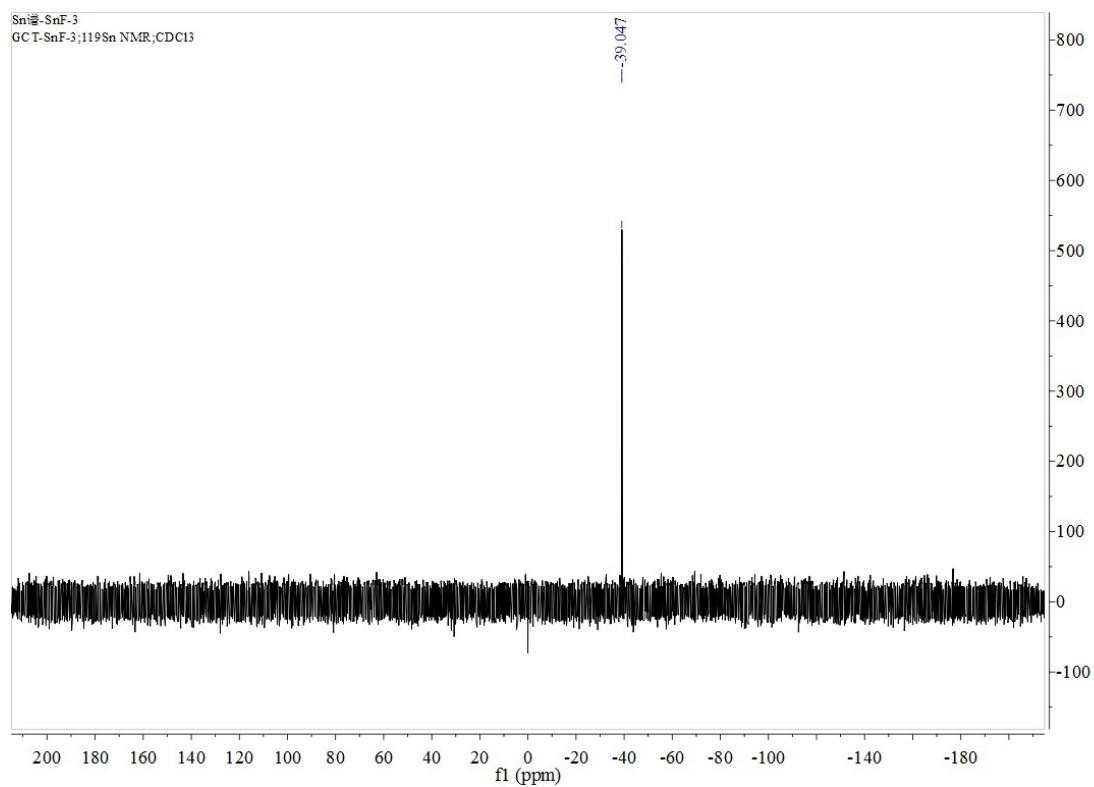


Figure S9. ^{119}Sn NMR spectrum of **SnF-3** in CDCl_3 .

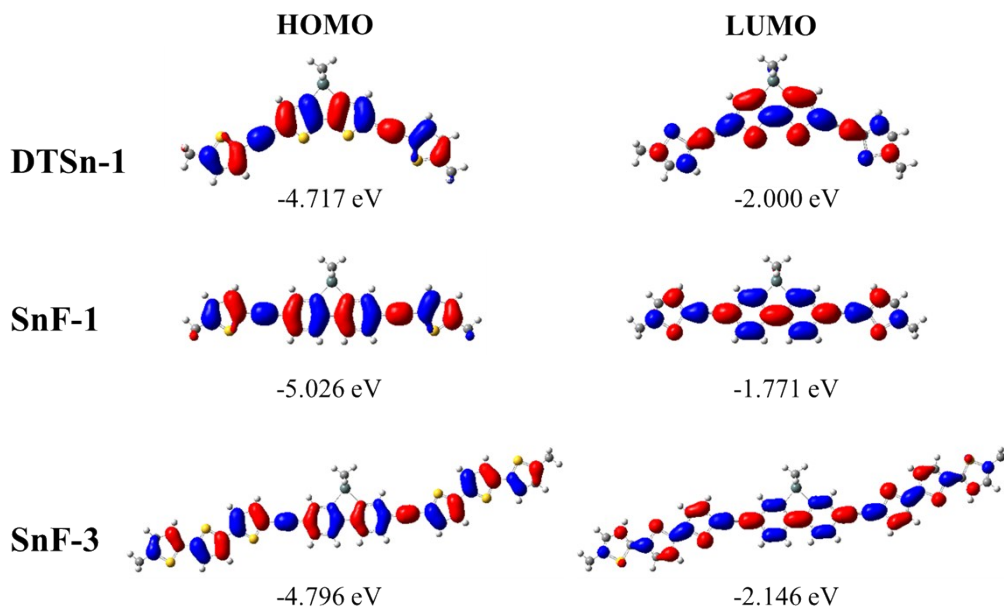


Figure S10. The energies and distributions of the frontier molecular orbitals of the Sn-containing small molecules [2]. The optimized molecular geometries were confirmed to be minimum-energy conformations since there were no imaginary frequencies by vibrational frequencies calculation at the same level.

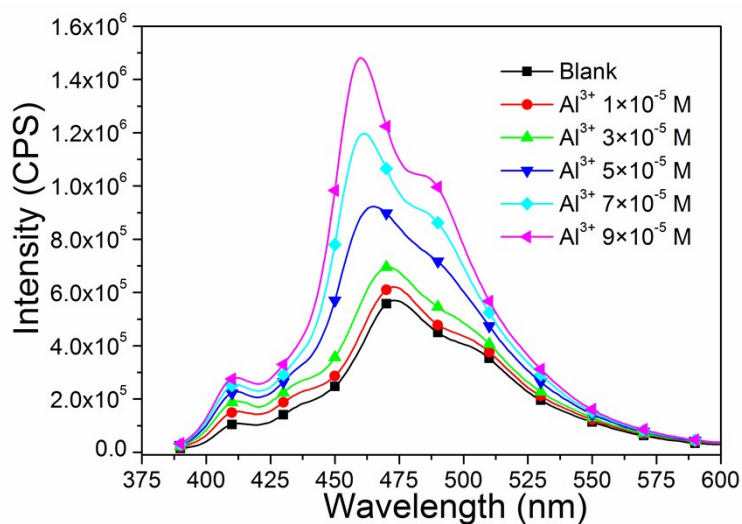


Figure S11. The fluorescence spectra of **DTSn-1** with different concentrations of Al^{3+} (excitation at 360 nm).

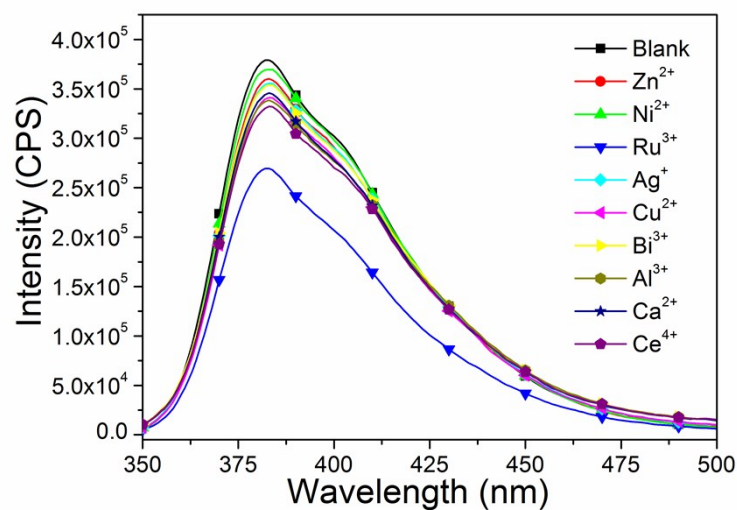


Figure S12. The fluorescence spectra of **SnF-1** in absence and presence of 5×10^{-5} mol/L metal ions (excitation at 320 nm).

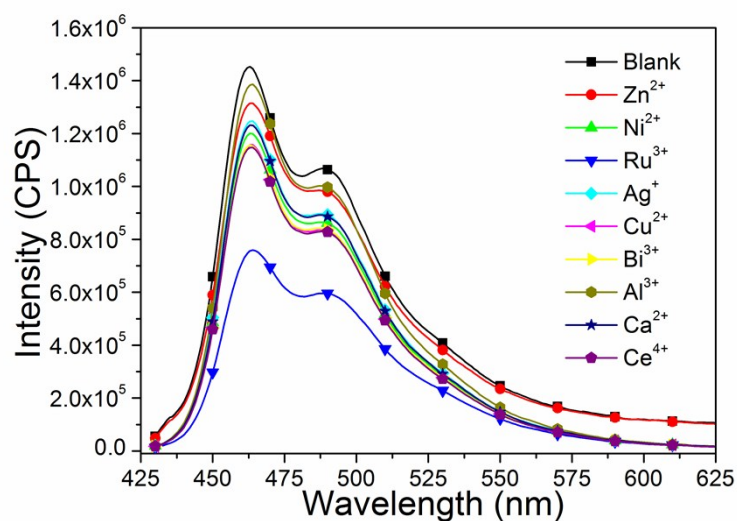


Figure S13. The fluorescence spectra of **SnF-3** in absence and presence of 5×10^{-5} mol/L metal ions (excitation at 400 nm).

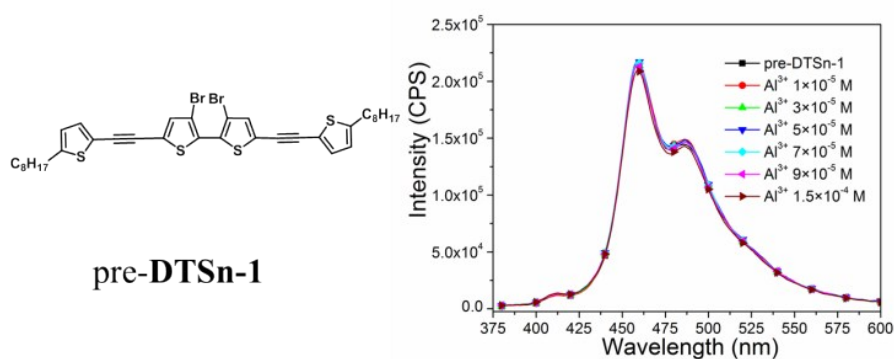


Figure S14. The structure of **pre-DTSn-1** and the fluorescence spectra of **pre-DTSn-1** with different concentrations of Al^{3+} (excitation at 350 nm).

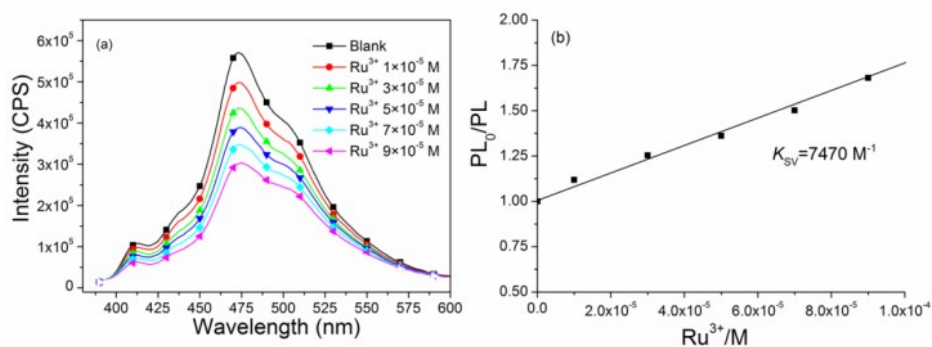


Figure S15. (a) The fluorescence spectra of **DTSn-1** with different concentrations of Ru³⁺ (excitation at 360 nm) and (b) Stern-Volmer plot of the quenching efficiency.

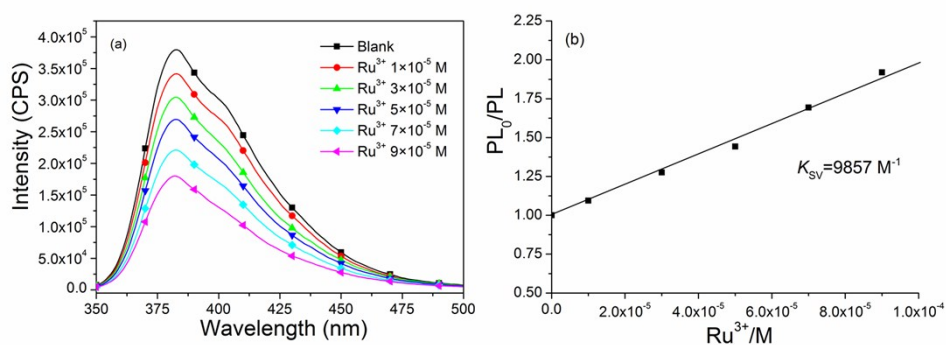


Figure S16. (a) The fluorescence spectra of **SnF-1** with different concentrations of Ru³⁺ (excitation at 320 nm) and (b) Stern-Volmer plot of the quenching efficiency.

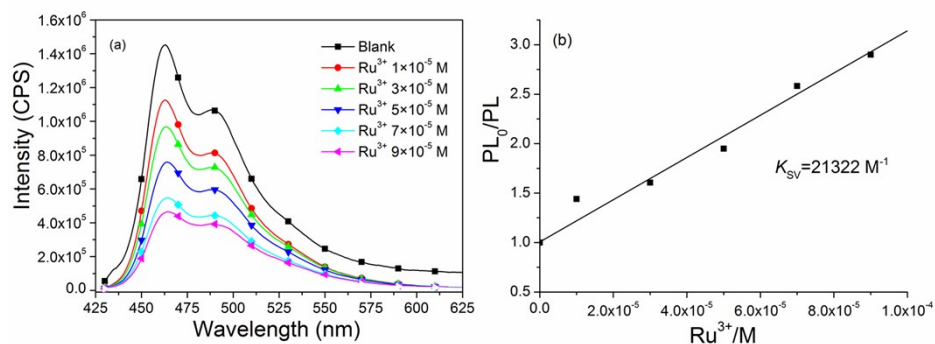


Figure S17. (a) The fluorescence spectra of **SnF-3** with different concentrations of Ru³⁺ (excitation at 400 nm) and (b) Stern-Volmer plot of the quenching efficiency.

The quenching process can be quantitatively described by the Stern-Volmer equation^[3]:

$$\frac{PL_0}{PL} = 1 + K_{SV}[Q] \quad (1)$$

where the PL₀ refers to the overall integrated emission intensity of fluorescence in the

absence of the quencher, PL corresponds to the integrated emission intensity of fluorescence in the presence of the quencher, K_{SV} is the Stern-Volmer constant. Here, the quencher was Ru^{3+} .

References:

1. I. Nagao, M. Shimizu, T. Hiyama, *Angew. Chem. Int. Ed.*, 2009, **48**, 7573-7576.
2. M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, *Gaussian 09*, Revision A.1, Gaussian, Inc., Wallingford CT, 2009.
3. (a) J. Wang, D. Wang, E. K. Miller, D. Moses, G. C. Bazan and A. J. Heeger, *Macromolecules*, 2000, **33**, 5153; (b) S. Wang, B. S. Gaylord and G. C. Bazan, *Adv. Mater.*, 2004, **16**, 2127.