

## New Journal of Chemistry

# Microwave-Assisted Synthesis and Photophysical Studies of Novel Fluorescent N-acylhydrazone- and Semicarbazone-7-OH- Coumarin Dyes

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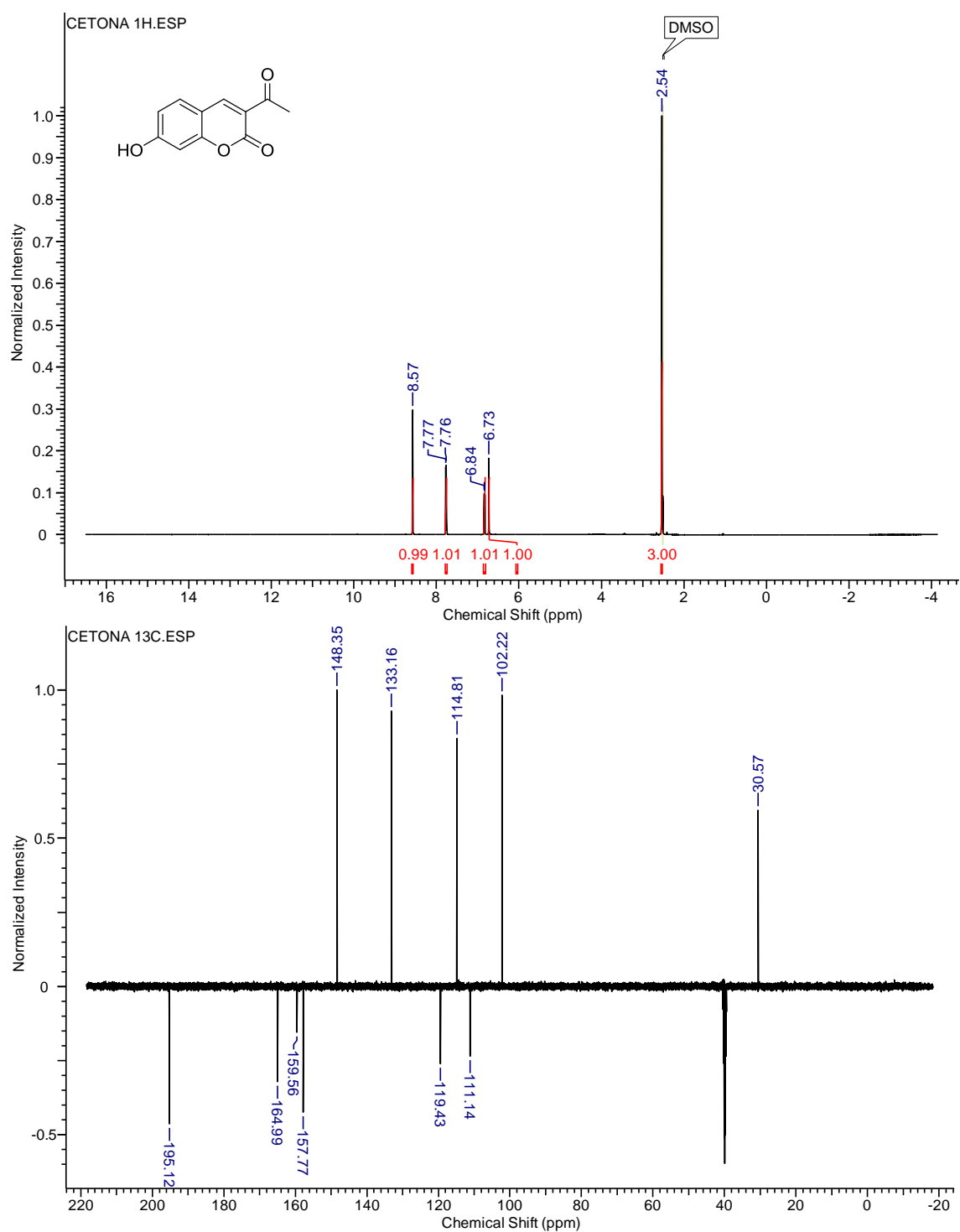
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## Supporting Information

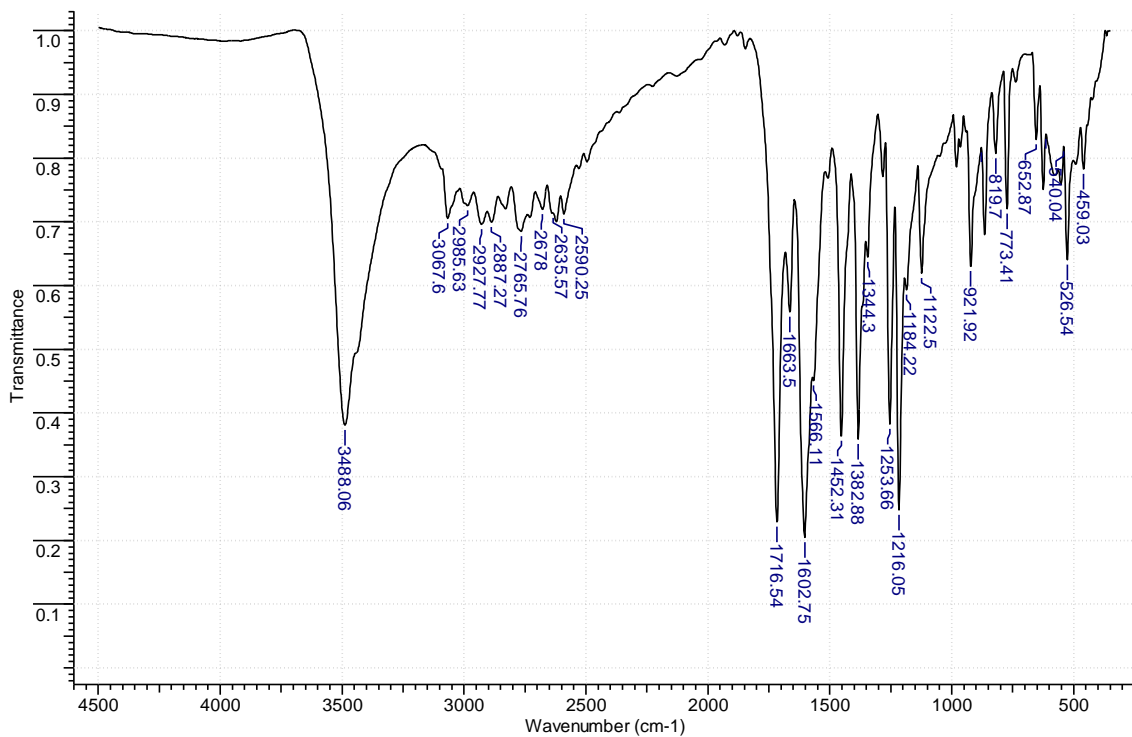
### Contents

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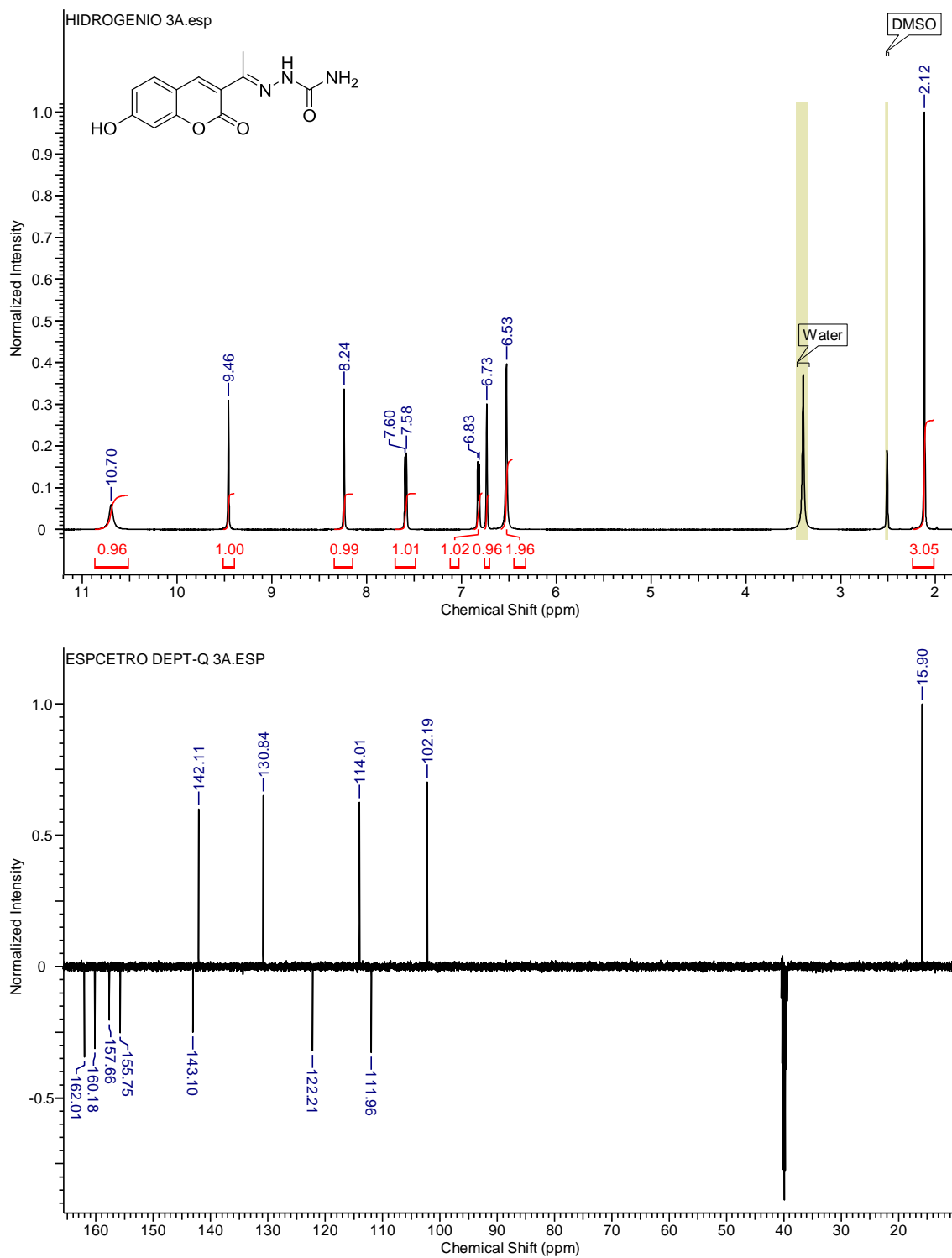
# $^1\text{H}$ NMR, $^{13}\text{C}$ NMR, IV and Mass Spectra



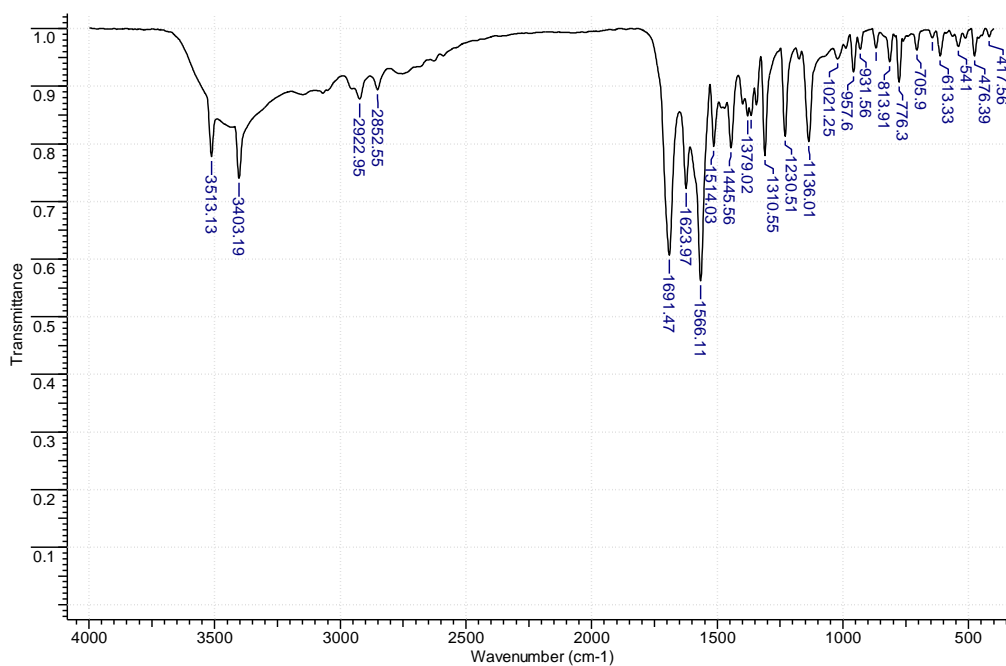
**Fig. S1.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **1** in  $\text{DMSO-}d_6$ .



**Fig. S2.** IR spectra of **1** in KBr.



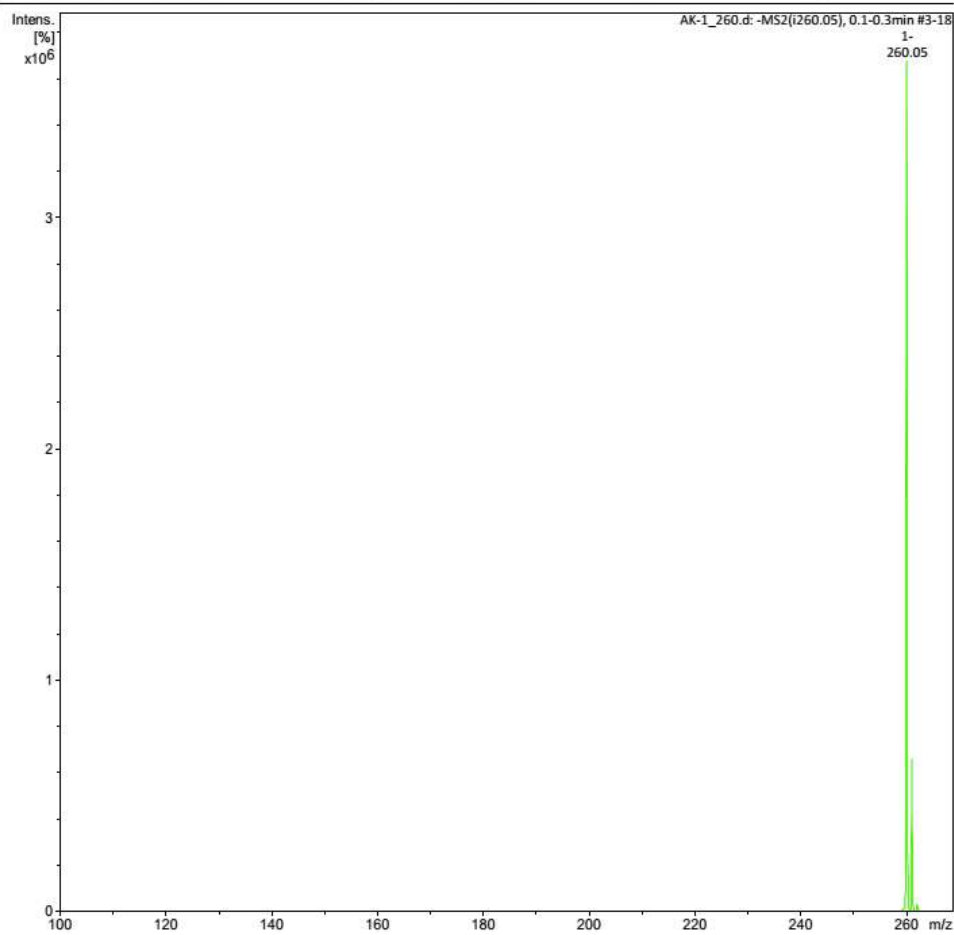
**Fig. S3.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3a** in  $\text{DMSO-}d_6$ .



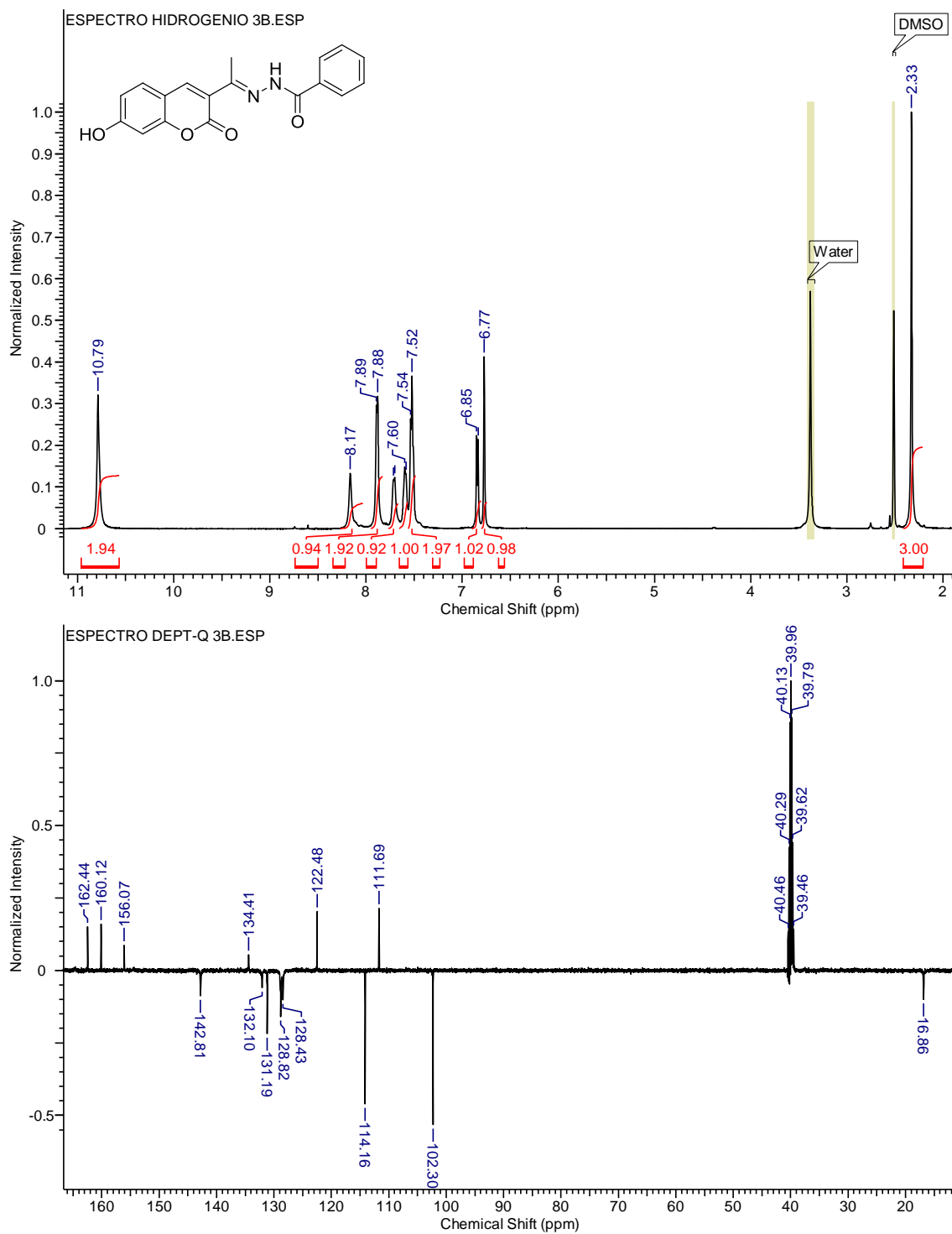
**Fig. S4.** IR spectra of **1** in KBr.

**Acquisition Parameter**

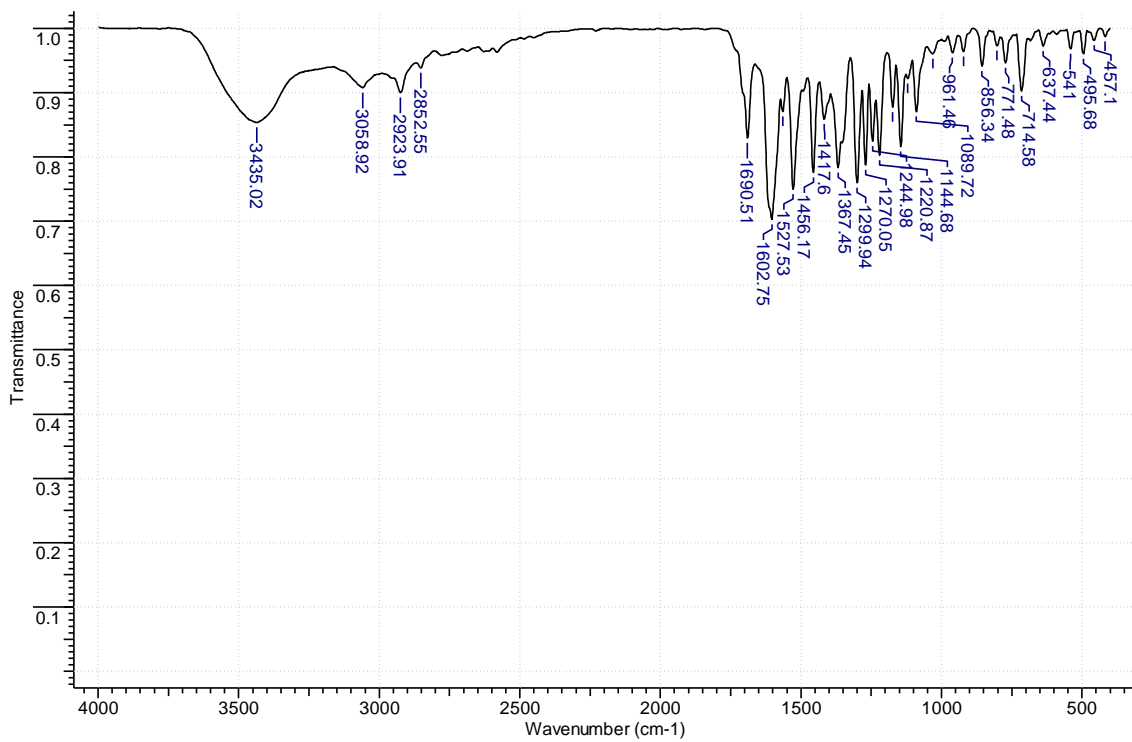
Ion Source Type	ESI	Ion Polarity	Negative	Alternating Ion Polarity	off
Mass Range Mode	Enhanced Resolution	Scan Begin	100 m/z	Scan End	800 m/z
Accumulation Time	591 $\mu$ s	RF Level	71 %	Trap Drive	51.1
SPS Target Mass	500 m/z	Averages	5 Spectra	n/a	n/a



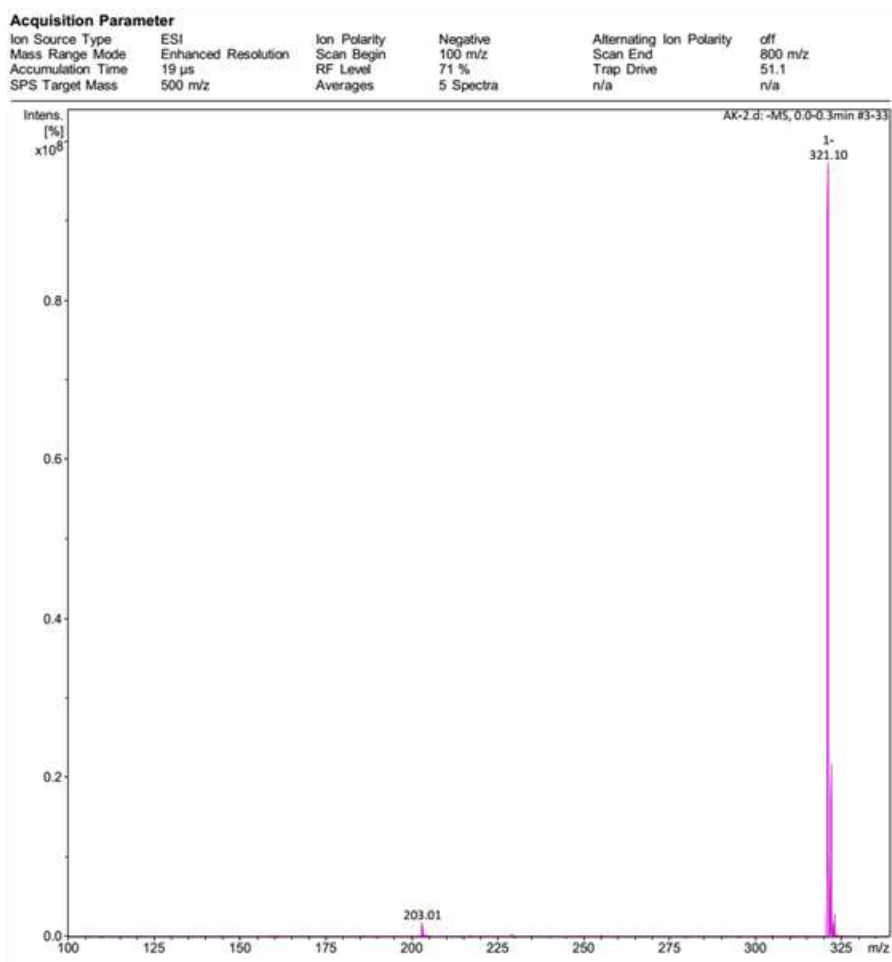
**Fig. S5.** EM spectra of **3a**.



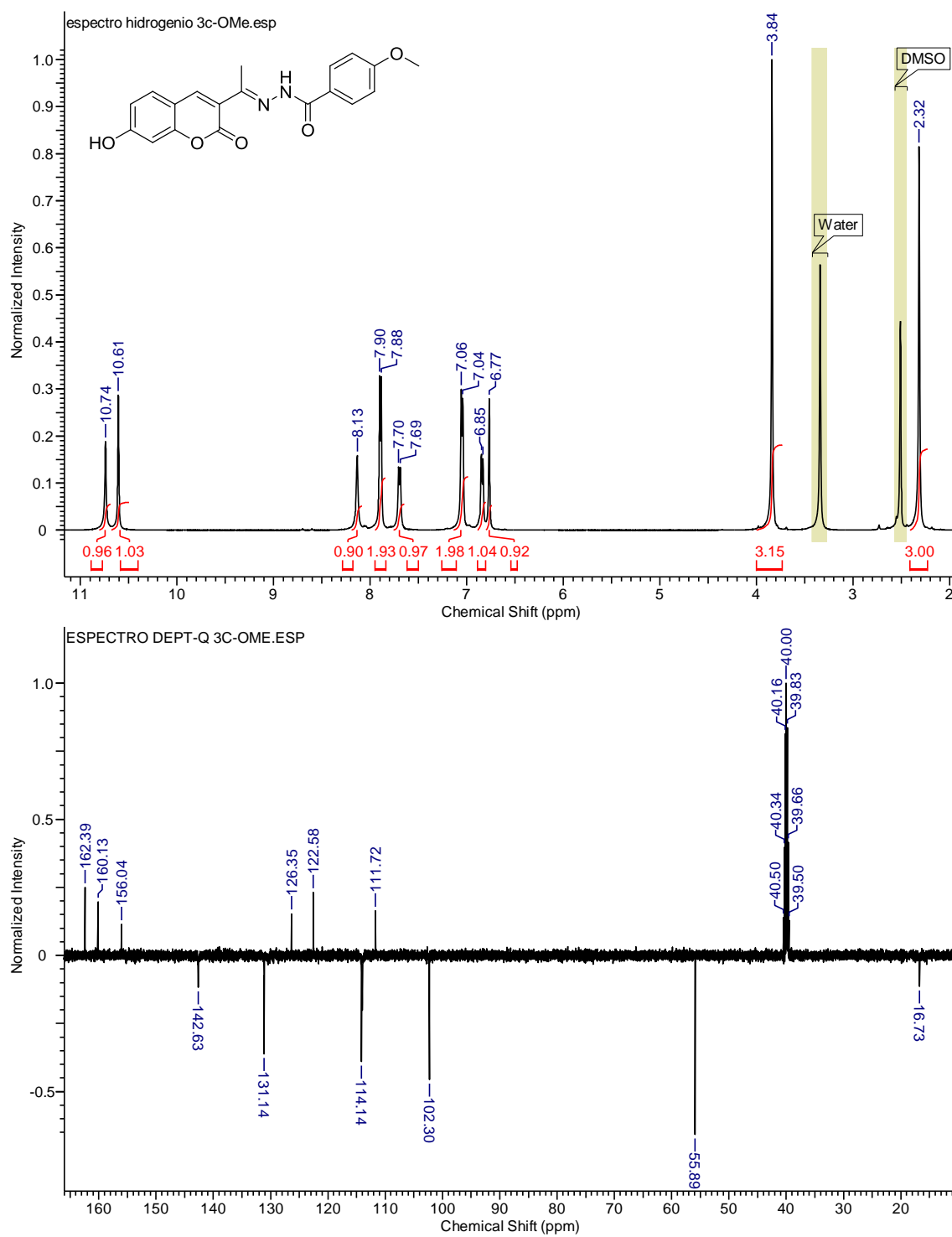
**Fig. S6.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3b** in  $\text{DMSO-}d_6$ .



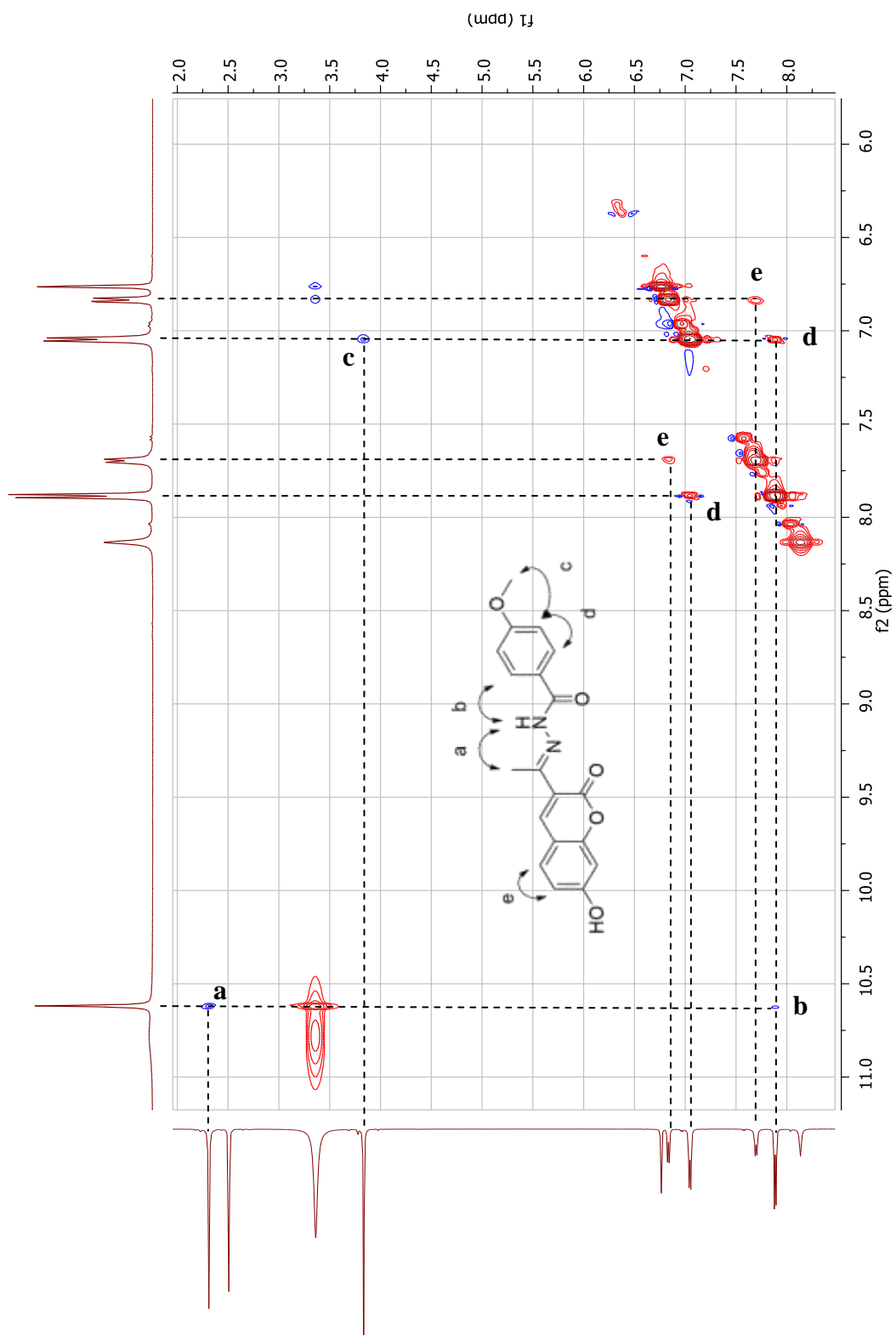
**Fig. S7.** IR spectra of **3b** in KBr.



**Fig. S8.** EM spectra of **3b**.







**Fig. S10.** NOESY spectrum of **3c** in DMSO- $d_6$  and correlations indicating the (*E*)-isomer.

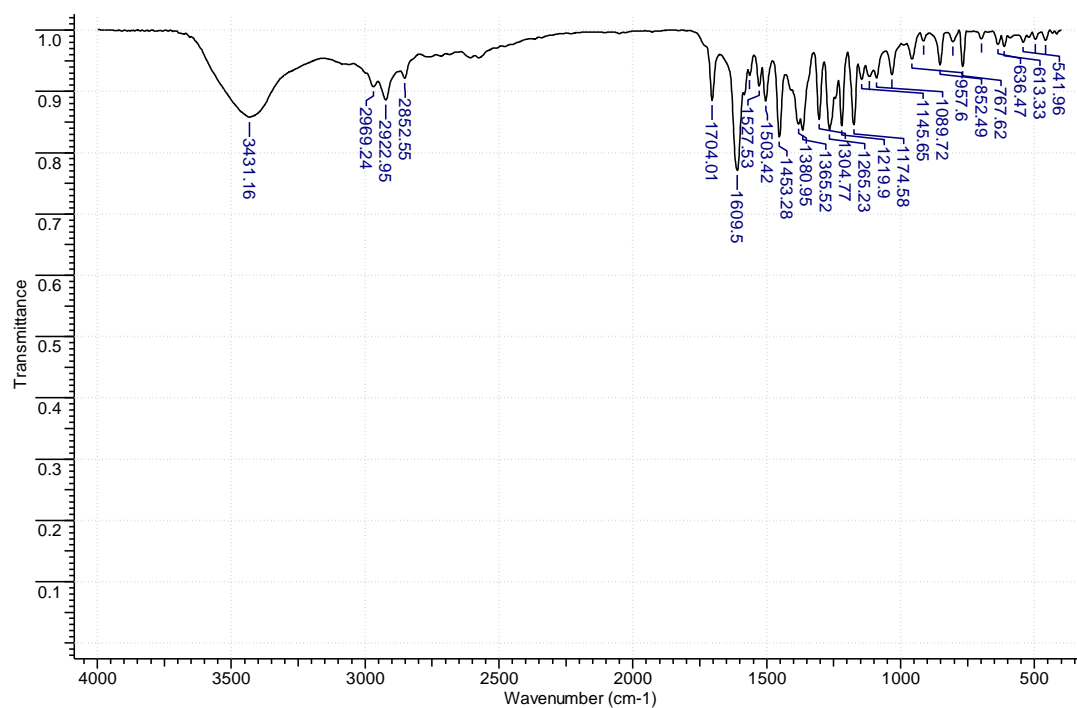


Fig. S11. IR spectra of **3c** in KBr.

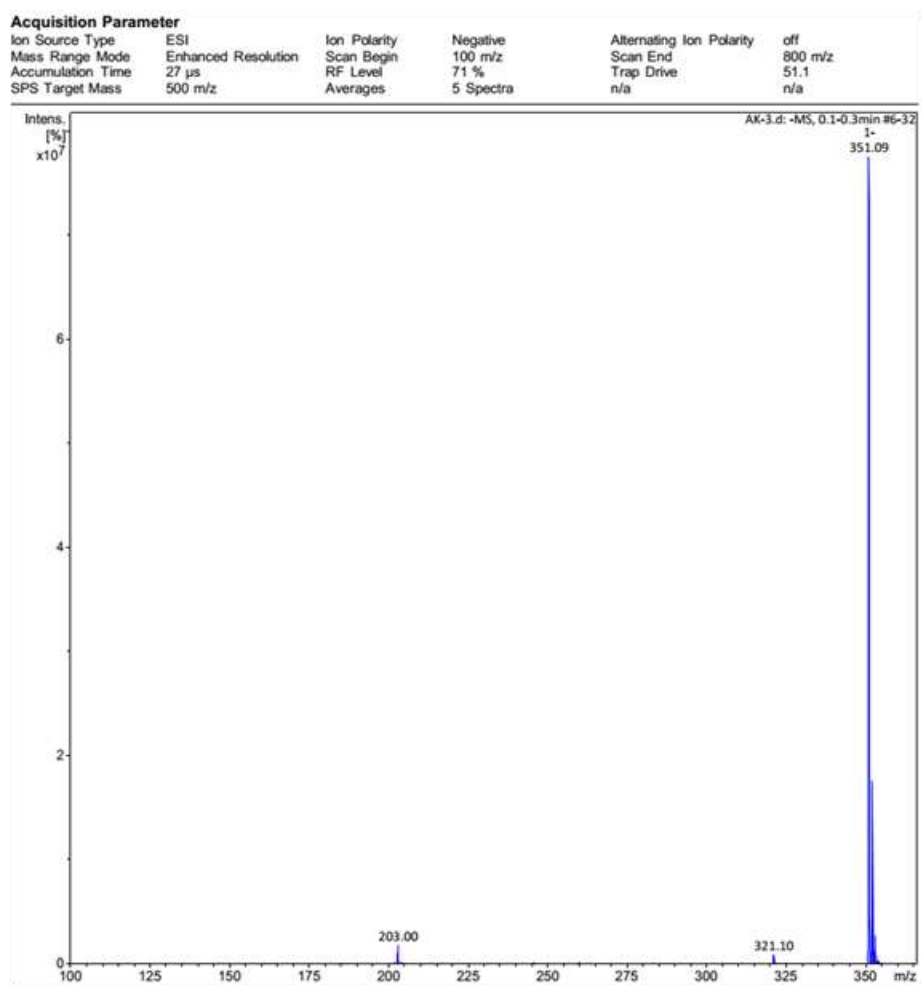
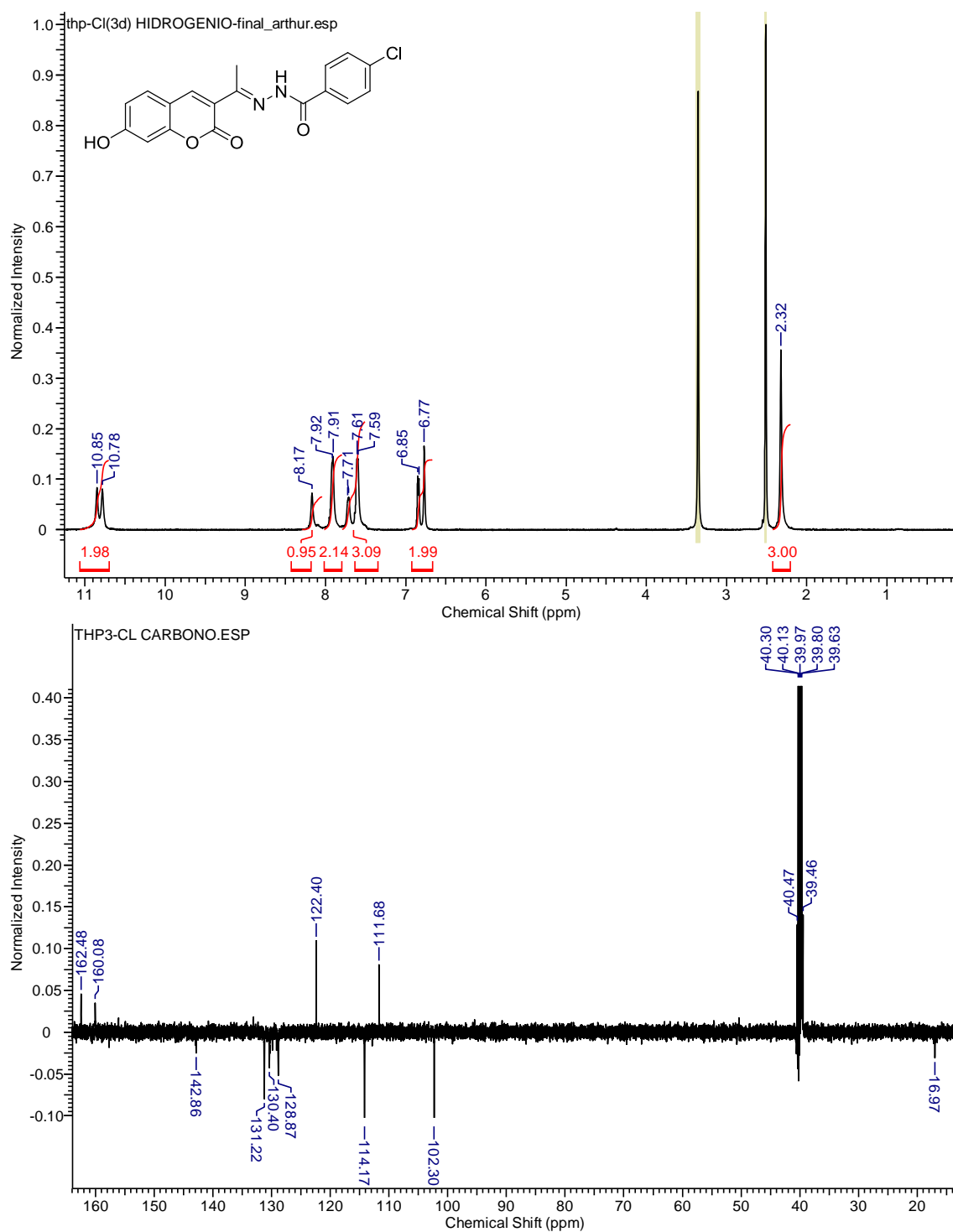
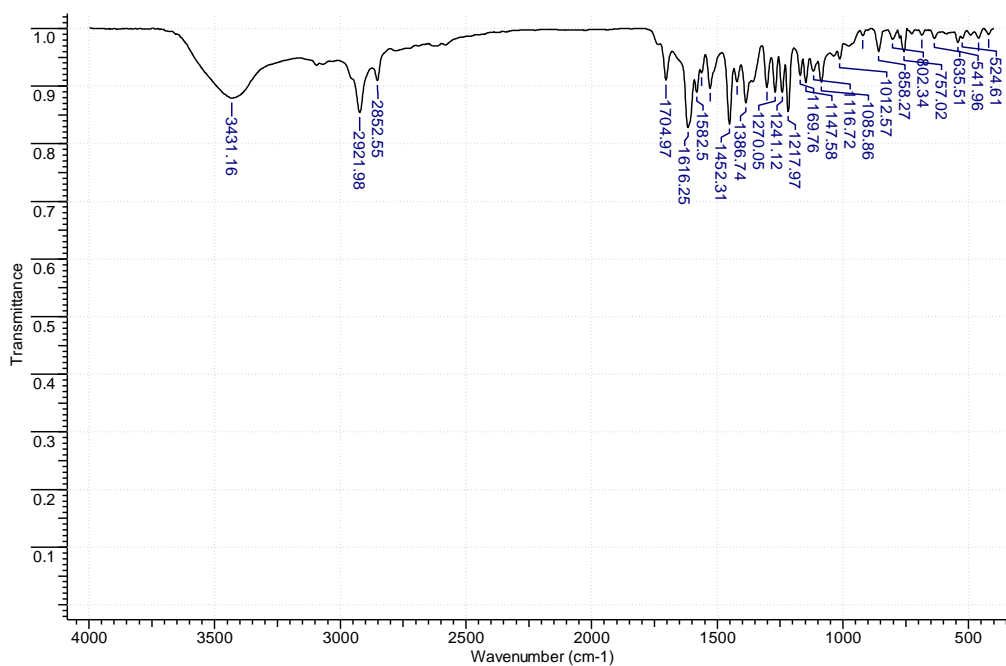


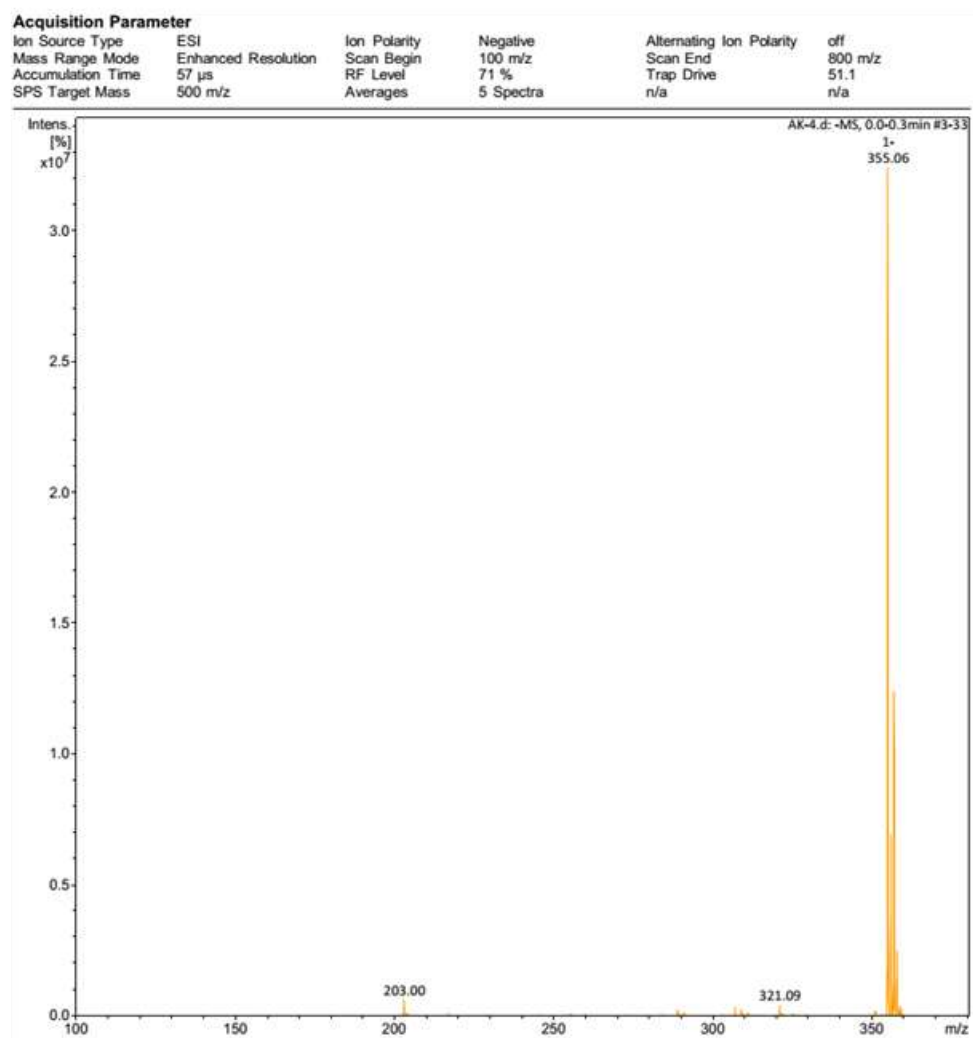
Fig. S12. EM spectra of **3c**.



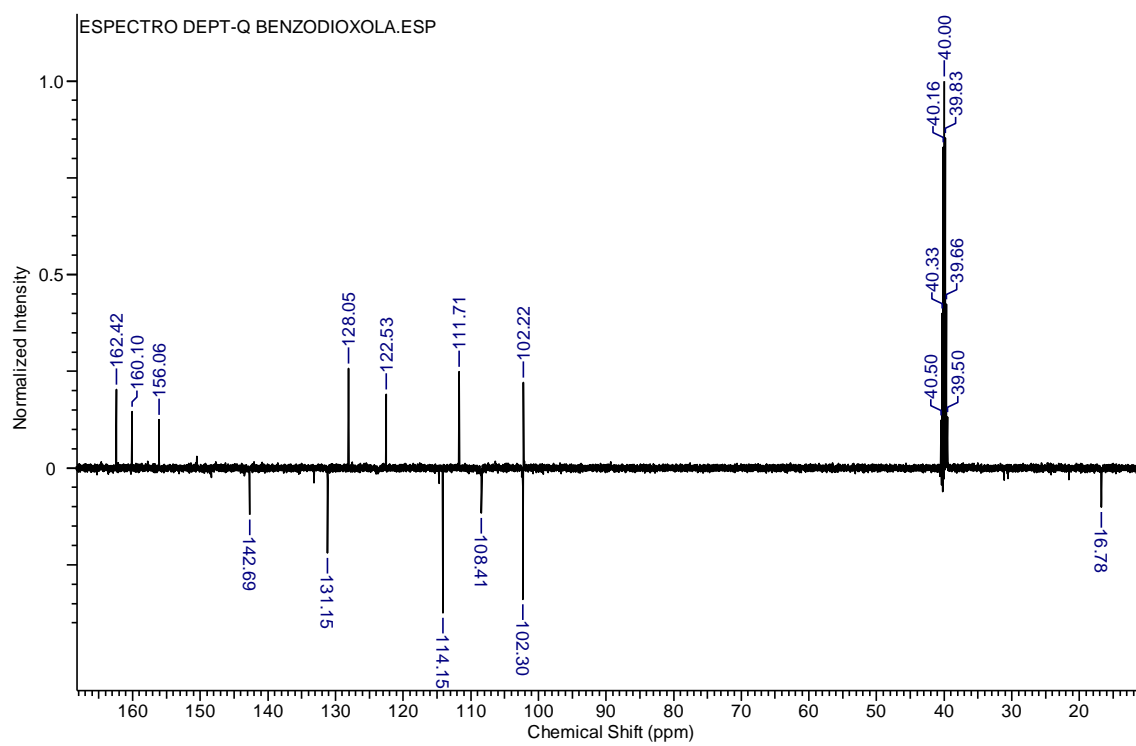
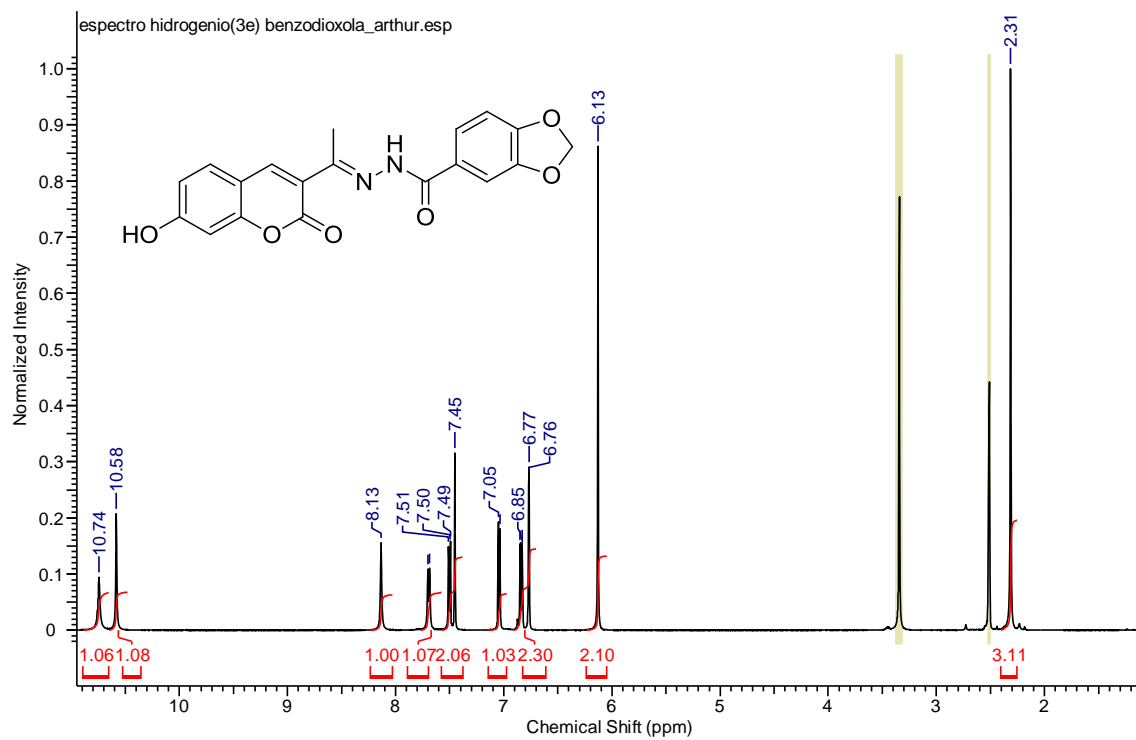
**Fig. S13.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3d** in  $\text{DMSO-}d_6$ .



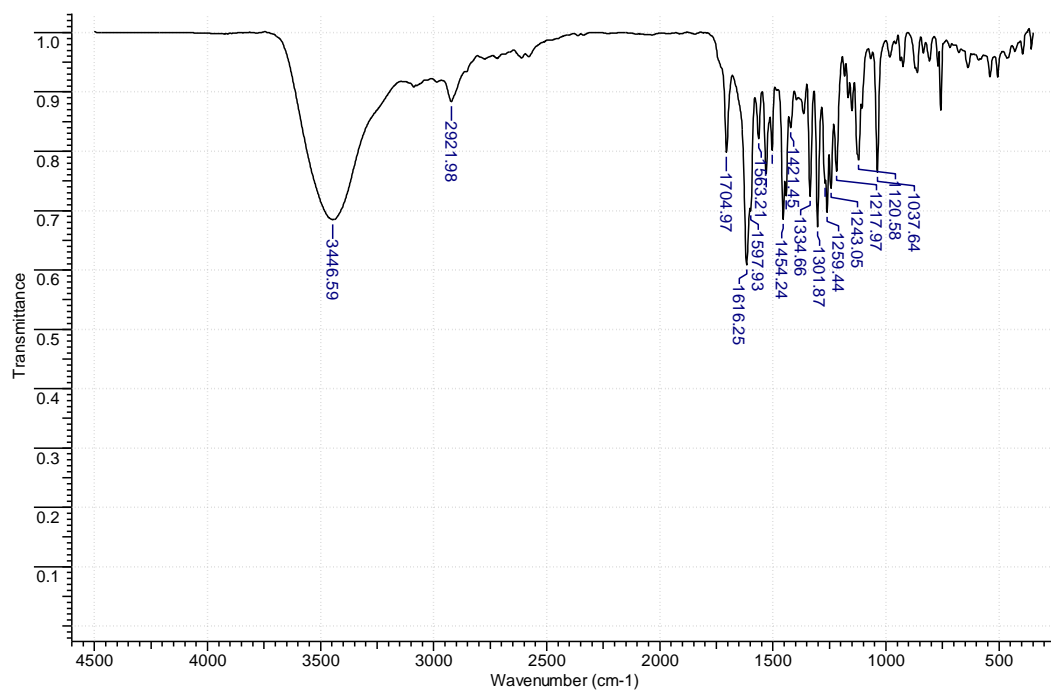
**Fig. S14.** IR spectra of **3d** in KBr.



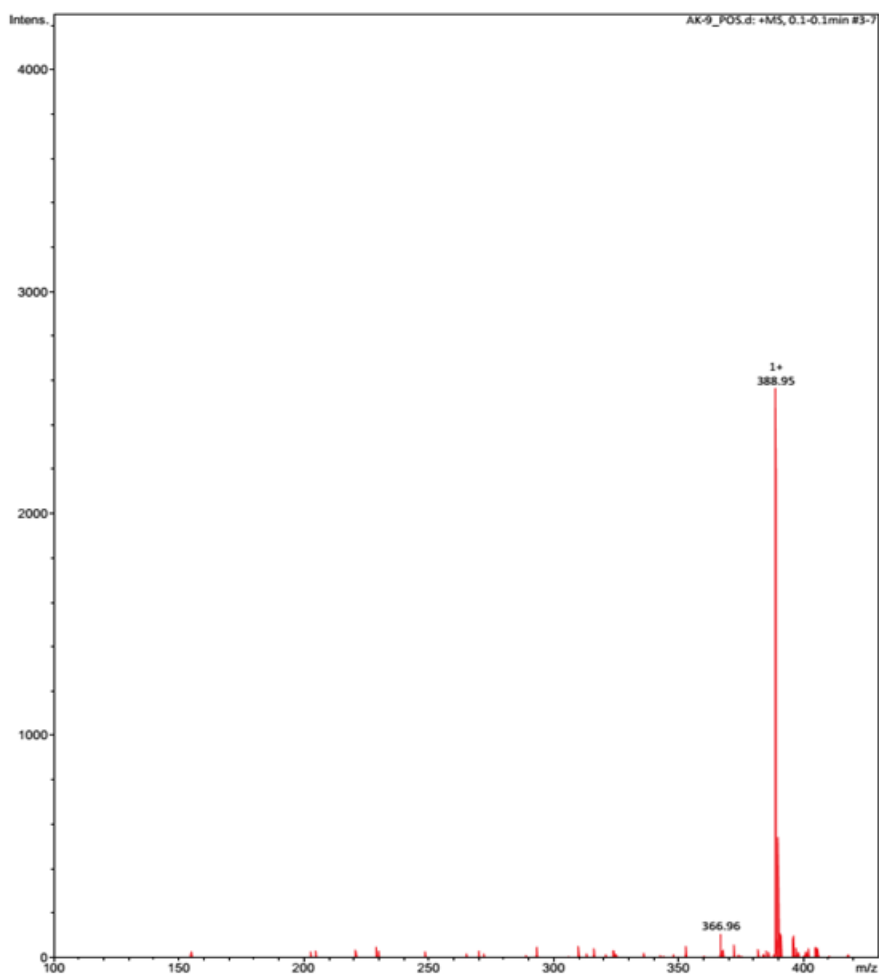
**Fig. S15.** EM spectra of **3d**.



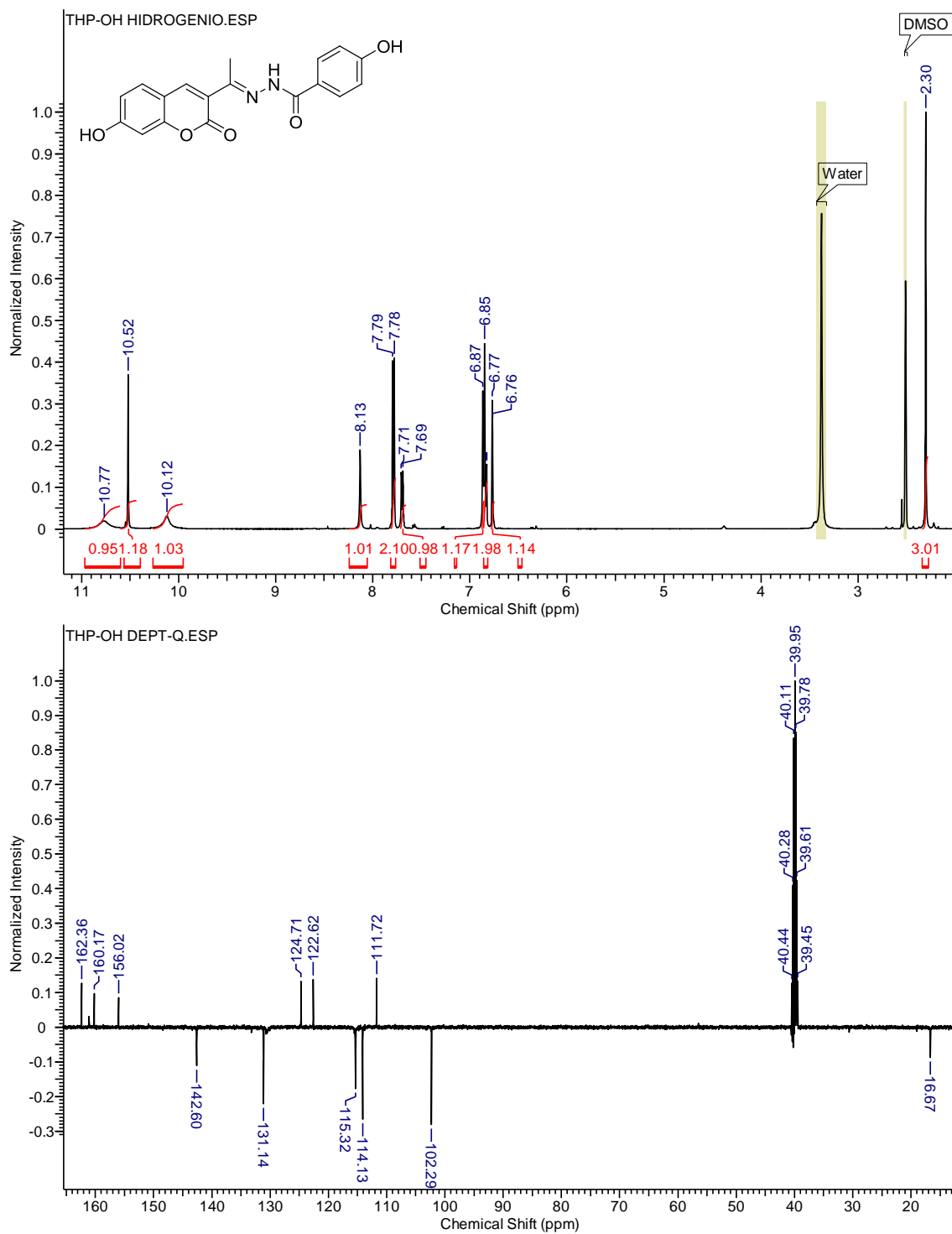
**Fig. S16.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3e** in  $\text{DMSO-}d_6$ .



**Fig. S17.** IR spectra of **3e** in KBr.



**Fig. S18.** EM spectra of **3e**.



**Fig. S19.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3f** in  $\text{DMSO-}d_6$ .

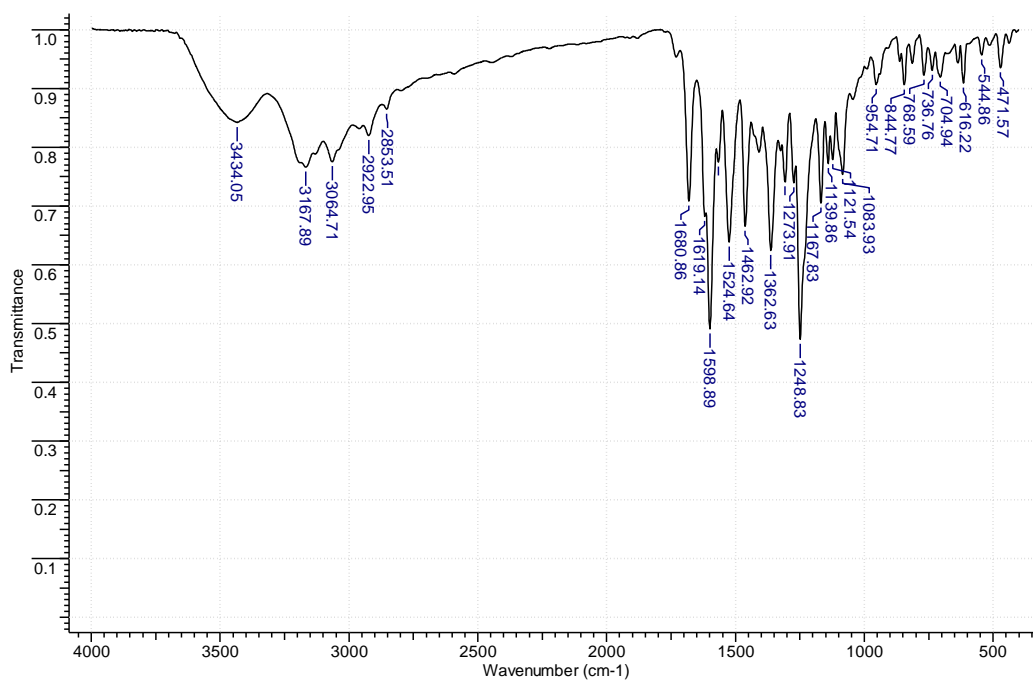


Fig. S20. IR spectra of 3f in KBr.

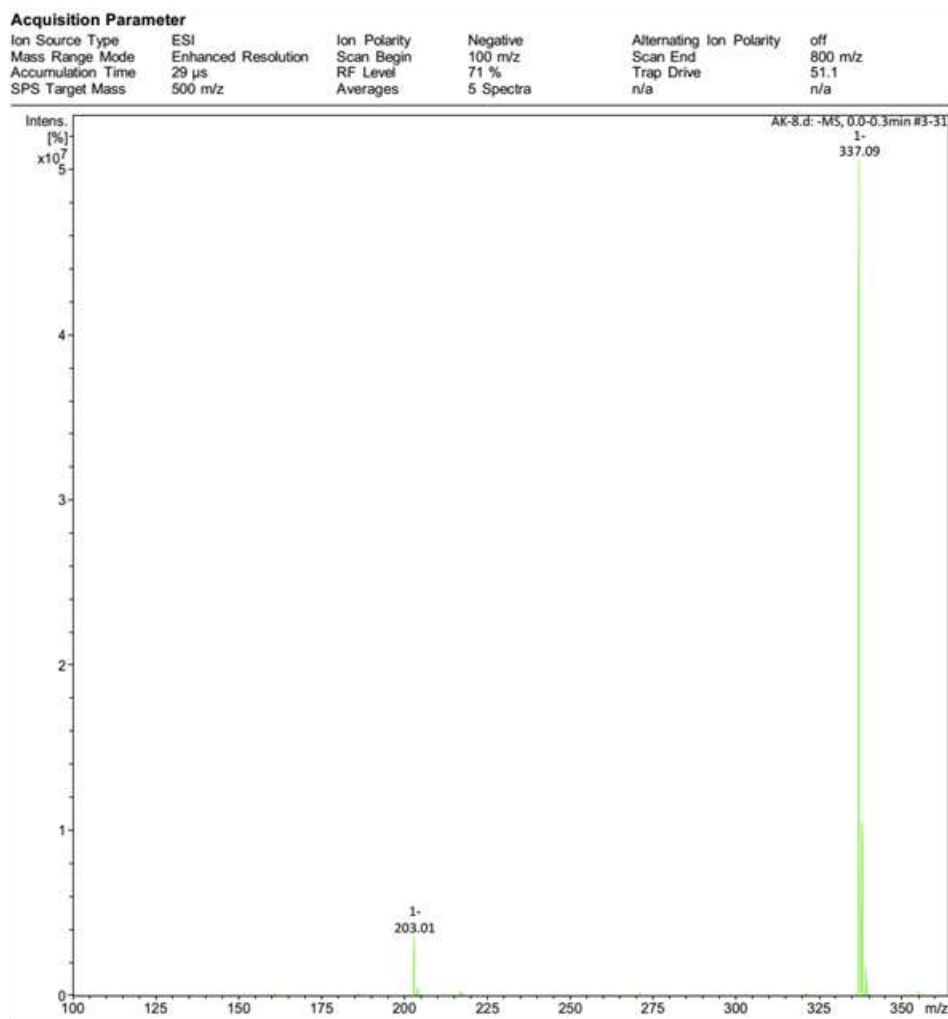
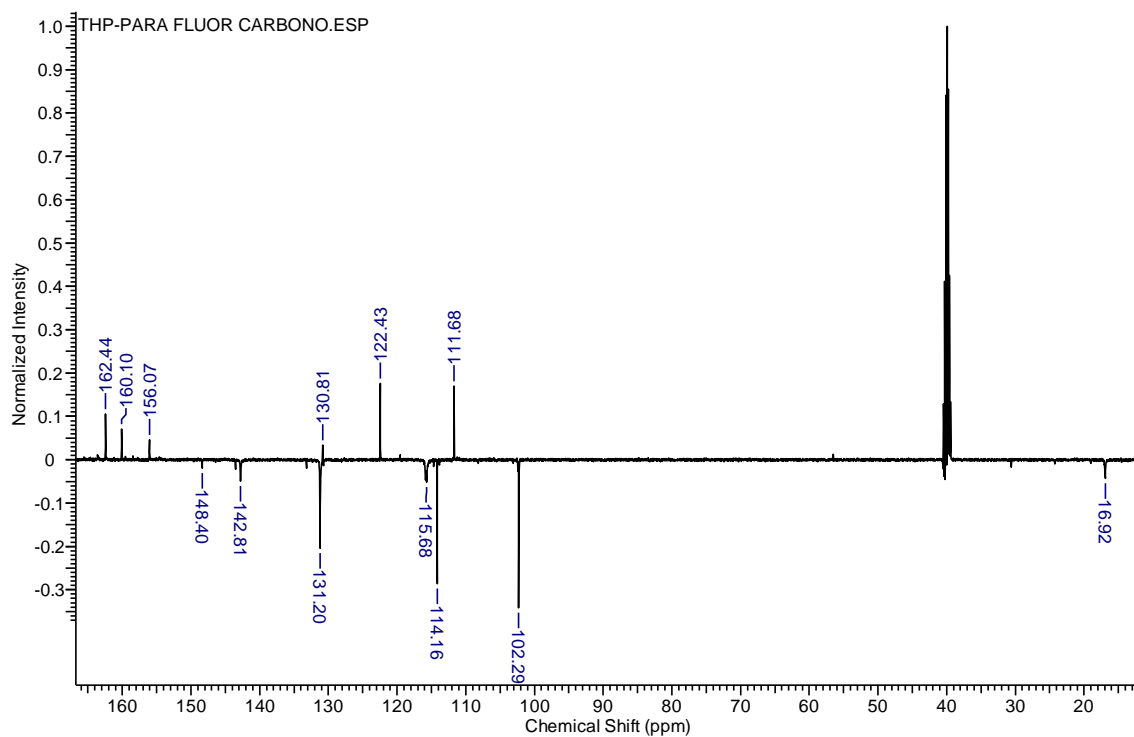
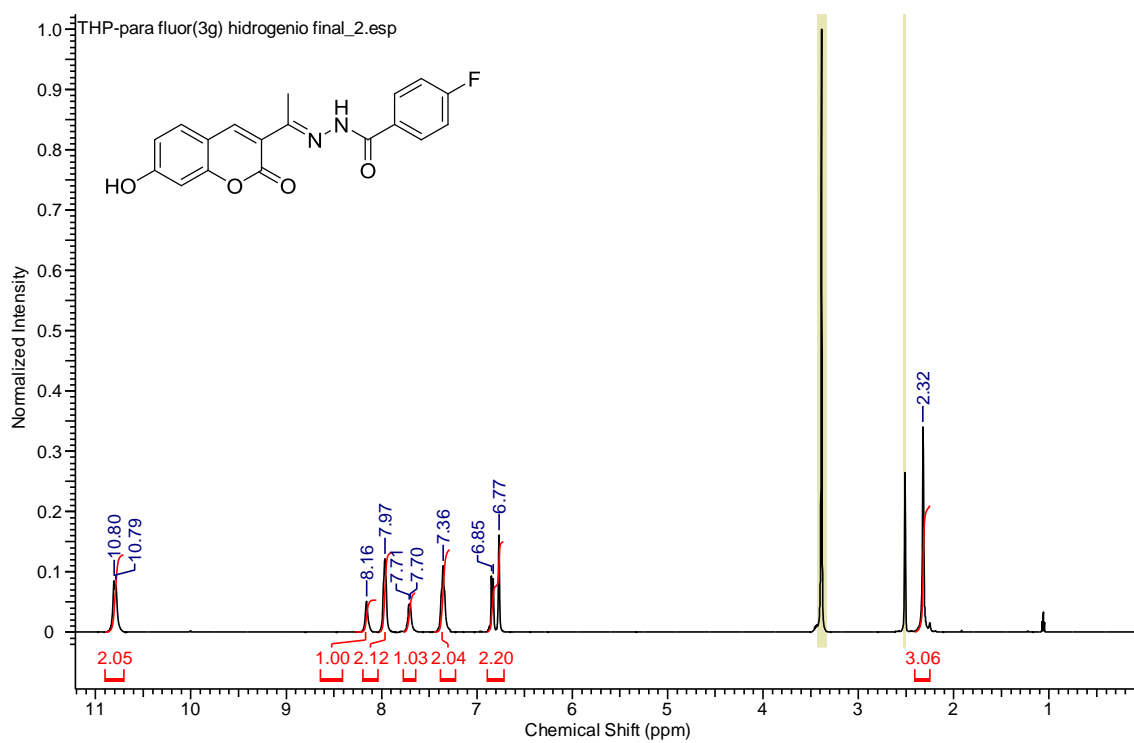


Fig. S21. EM spectra of 3f.





**Fig. S22.**  $^1\text{H}$  NMR (500 MHz),  $^{13}\text{C}$  NMR (125 MHz) spectra of **3g** in  $\text{DMSO-}d_6$ .

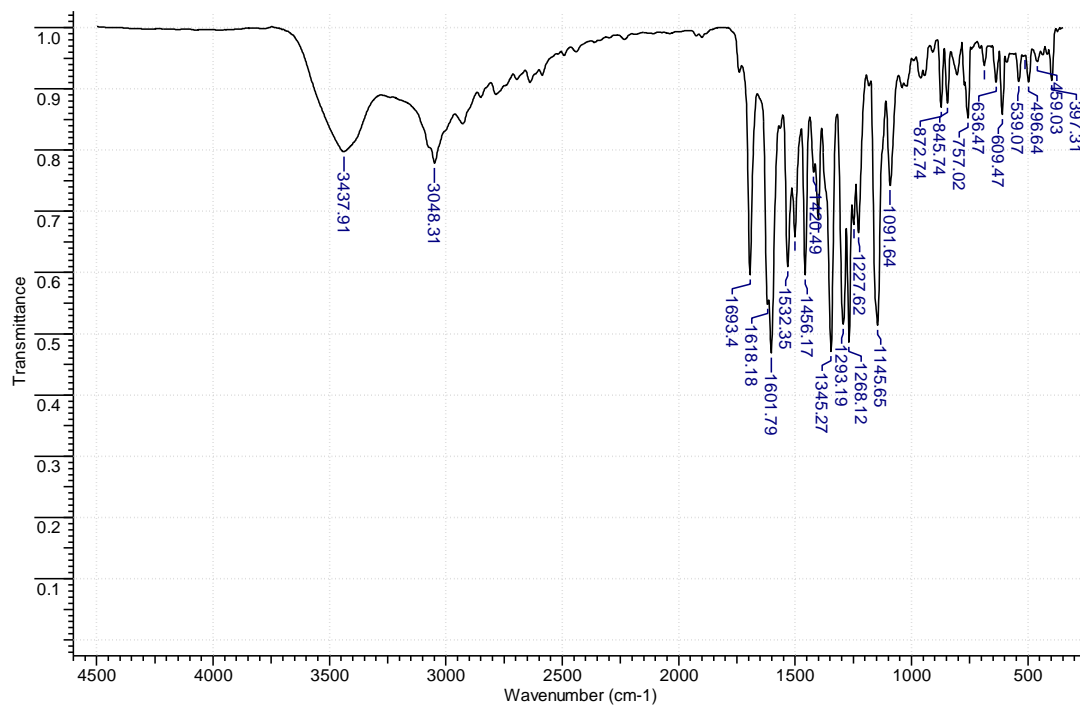


Fig. S23. IR spectra of 3g in KBr.

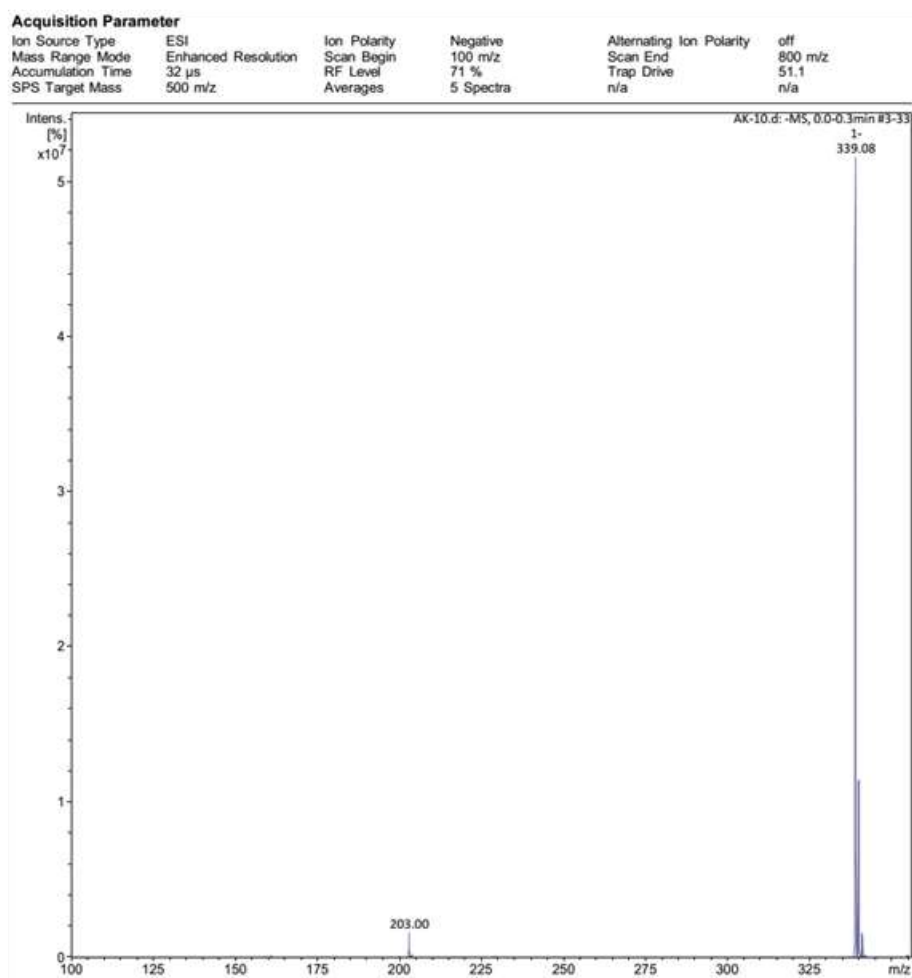
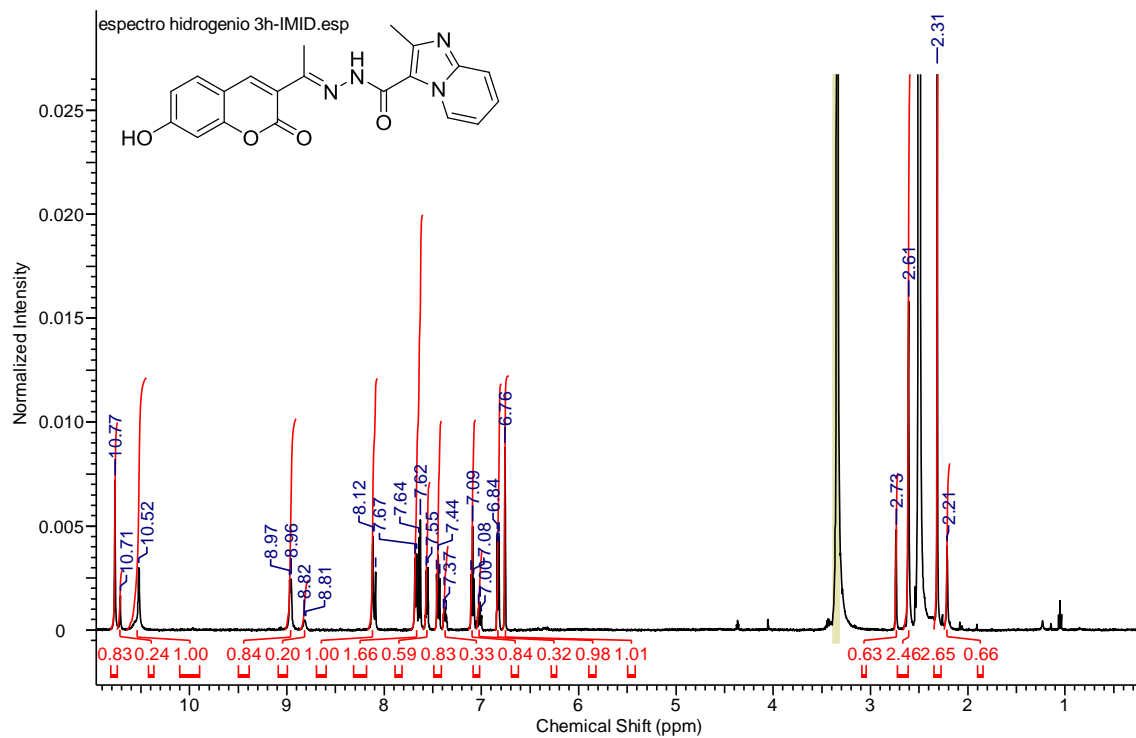
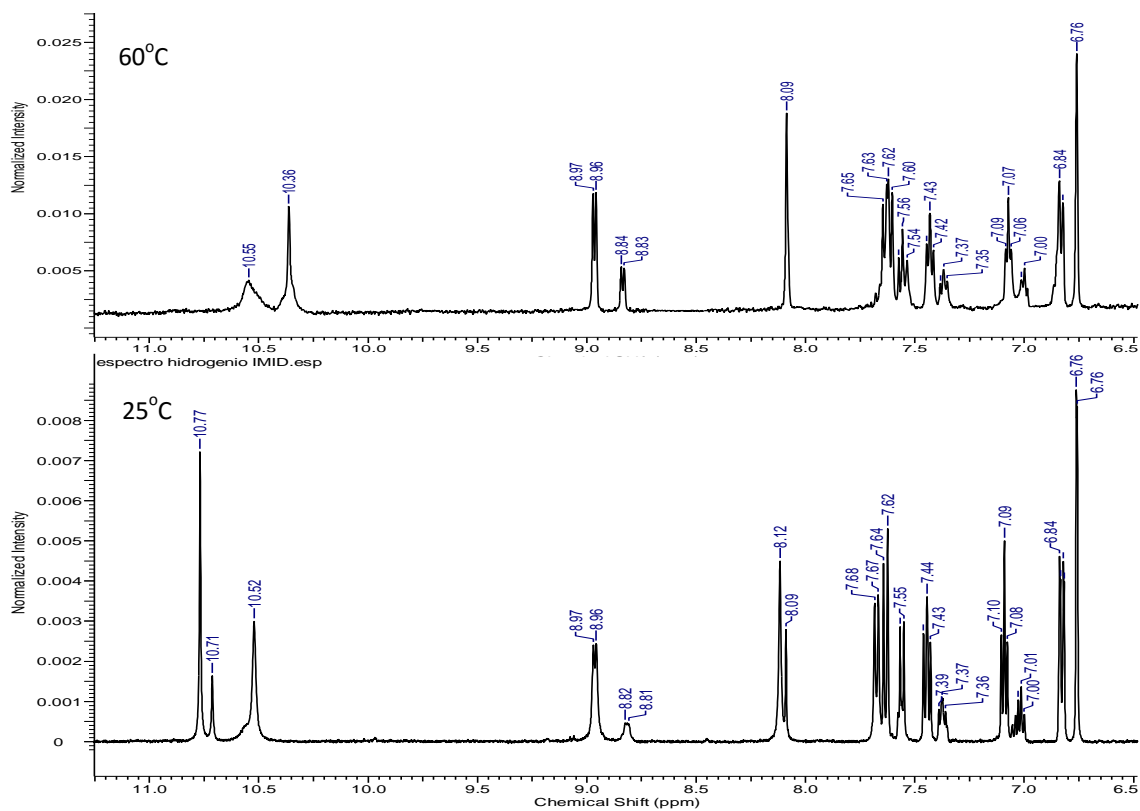


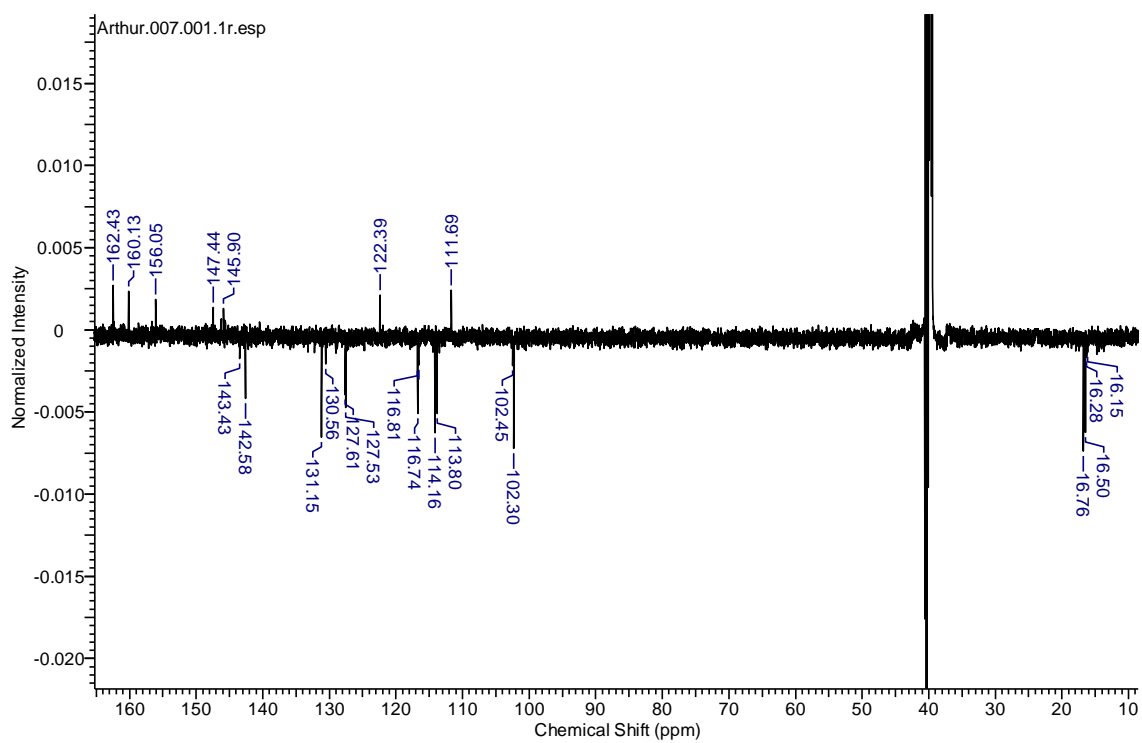
Fig. S24. EM spectra of 3g.



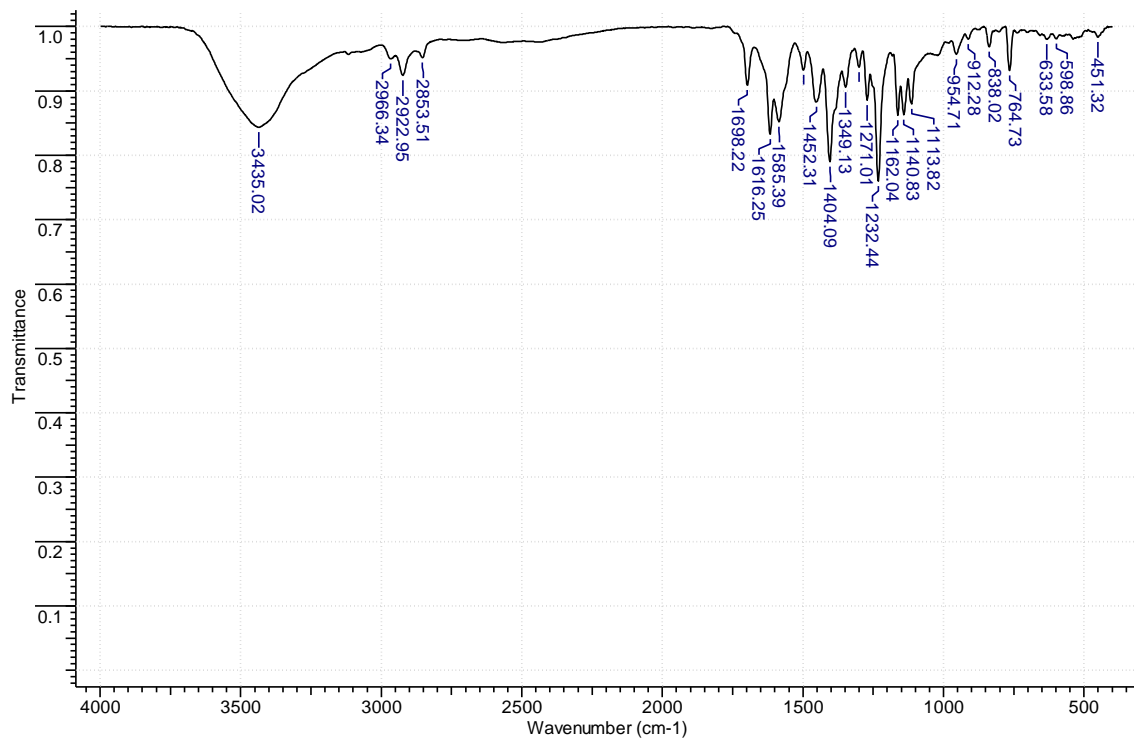
**Fig. S25.**  $^1\text{H}$  NMR (500 MHz), spectra of **3h** in  $\text{DMSO-}d_6$ .



**Fig. S26.**  $^1\text{H}$  NMR, expanded aromatic region of **3h** in  $\text{DMSO-}d_6$  at 25 and 60°C.



**Fig. S27.**  $^{13}\text{C}$  NMR (125 MHz) spectrum of **3h** in  $\text{DMSO-}d_6$  (poor quality due to low solubility).



**Fig. S28.** IR spectra of **3h** in KBr.

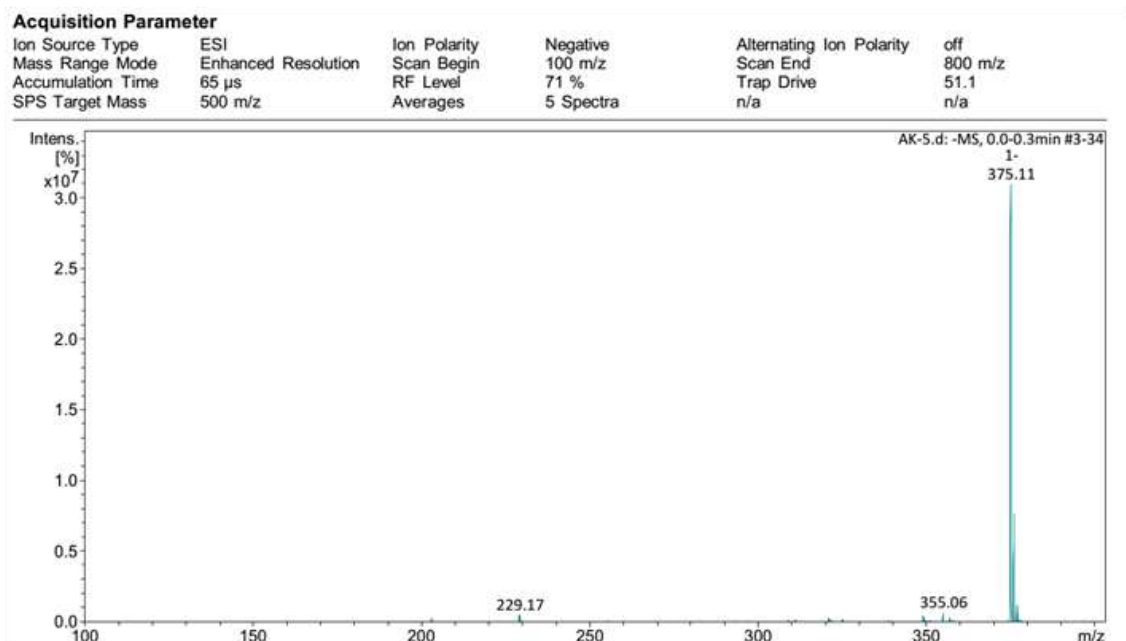


Fig. S29. EM spectra of 3h.

==== Shimadzu LCsolution Analysis Report ====

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 Sample ID :  
 Tray# : 1  
 Vial # : 64  
 Injection Volume : 40  $\mu$ L  
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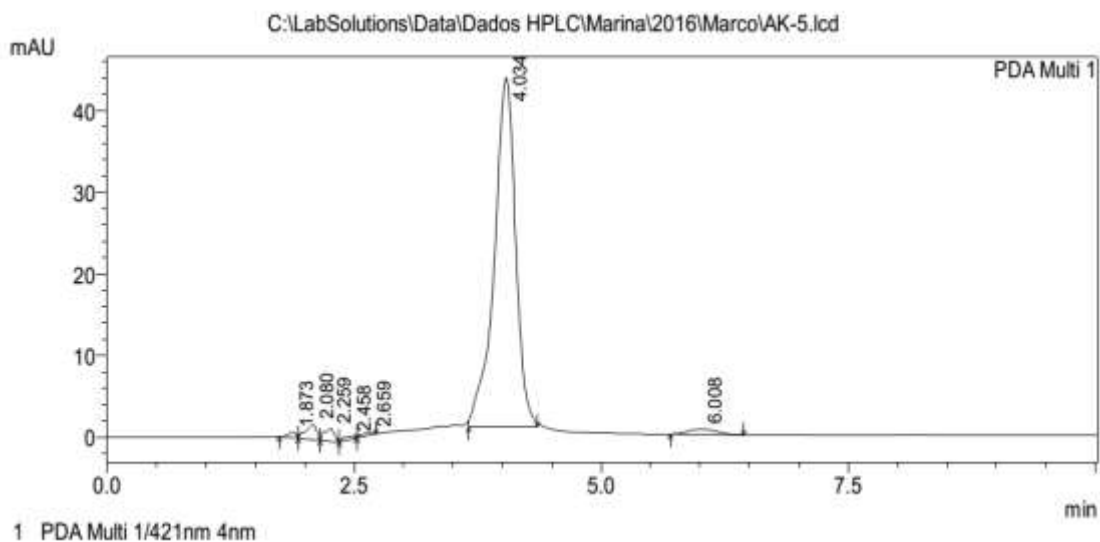
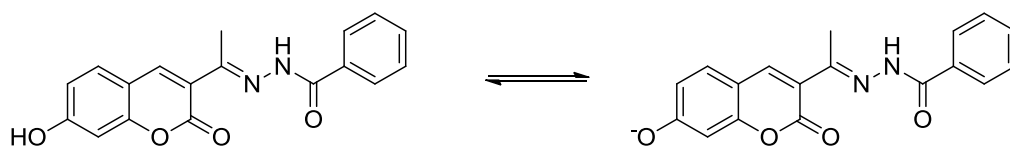
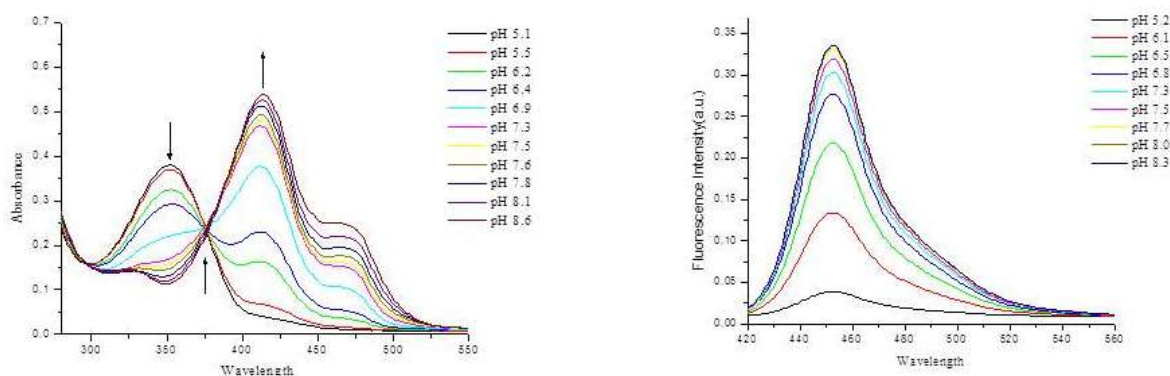


Fig. S30. RP-HPLC analysis of 3h.

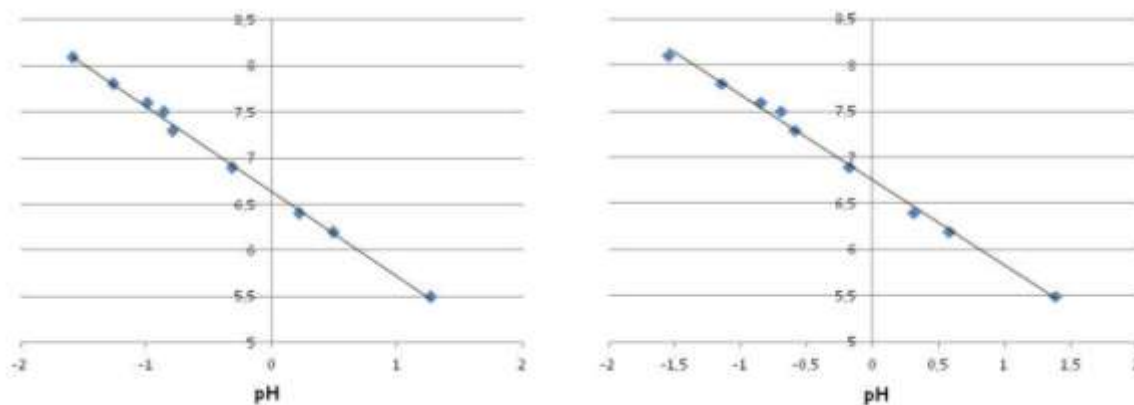
### pKa determination:



The spectral change from pH 5 to pH 8 is ascribed to the deprotonation of the hydroxyl group in the coumarin nucleus, as typical for other hydroxycoumarins, and lead to a well-defined isosbestic point around 375 nm. By the relationship between the pH and  $\log[(A - A_f)/(A_0 - A)]$ , the pKa constant for the deprotonation of the **3b** hydroxycoumarin was calculated to be 6.7.



**Fig.S31.** pH titrations of **3b**. Absorption and emission spectrum.



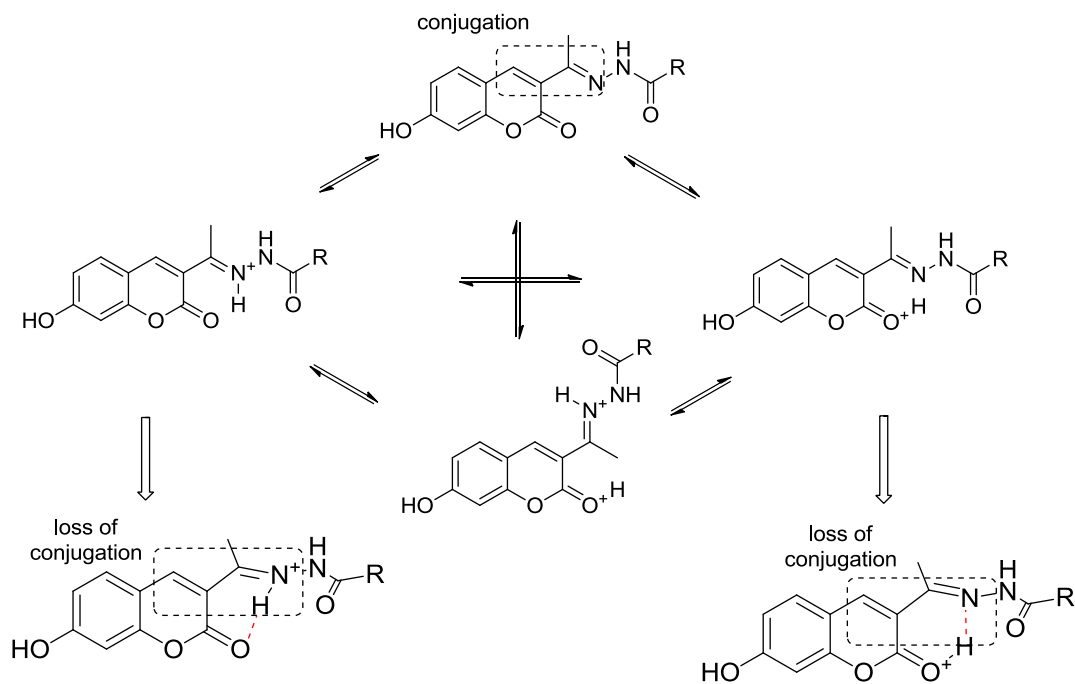
**Fig. S32.** pKa determination of **3b** considering 412nm and 351nm wavelengths.

pKa  $\approx$  6.7.

**Table S1.** Photophysical parameters in water (pH = 3.0) at 298 K.

Compound	$\phi_F^a$
<b>3a</b>	0.048
<b>3b</b>	0.068
<b>3c</b>	0.068
<b>3d</b>	0.066
<b>3h</b>	0.065

<sup>a</sup>Quantum yields were measured by the relative method against the standard compound ethyl-7-OH-coumarin-3-carboxylate ( $\phi = 0.83$  in water) [32].



**Fig. S33.** Proposition of protonated states of *N*-acylhydrazone- and Semicarbazone-7-OH-Coumarins