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Electronic Supplementary Information

Ligand-directed assembly of trinuclear and one-dimensional
heterotrimetallic $\text{Cu}^{\text{II}}\text{Ln}^{\text{III}}\text{Fe}^{\text{III}}$ complexes. Unusual antiferromagnetic
 $\text{Cu}^{\text{II}}\text{Fe}^{\text{III}}$ coupling via cyano bridges

Mei-Jiao Liu,^a Kong-Qiu Hu,^a Cai-Ming Liu,^b Ai-Li Cui^a and Hui-Zhong Kou^{a,*}

^a Department of Chemistry, Tsinghua University, Beijing 100084, P. R. China.

^b Beijing National Laboratory for Molecular Sciences, Center for Molecular Science,
Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, P. R. China

Table S1 Selected bond distances (Å) and bond angles (°) for complexes **1-3**^a

Complex	1	2	3	
Ln1–O1	2.330(3)	2.373(3)	Dy1–O1	2.311(3)
Ln1–O2	2.591(3)	2.552(3)	Dy1–O2	2.456(3)
Ln1–O3	2.335(3)	2.317(3)	Dy1–O3	2.453(3)
Ln1–O4	2.576(3)	2.587(3)	Dy1–O4	2.290(2)
Ln1–O2W	2.387(3)	2.353(3)	Dy1–O1W	2.355(3)
Ln1–O3W	2.371(3)	2.294(3)	Dy1–O2W	2.352(3)
Ln1–O4W	2.432(4)	2.411(4)	Dy1–O6W	2.398(3)
Ln1–N8	2.473(4)	2.461(4)	Dy1–N8	2.412(3)
Ln1–O1W'	2.753(10)	2.733(13)		
Cu1–O1	1.940(3)	1.917(3)	Cu1–O1	1.970(2)
Cu1–O3	1.952(3)	1.998(3)	Cu1–O4	1.976(3)
Cu1–N1	1.952(4)	1.940(5)	Cu1–N6	1.965(3)
Cu1–N6	1.962(4)	2.006(4)	Cu1–N7	1.982(4)
Cu1–O1W	2.683(16)	2.605(13)	Cu1–O1W	2.447(3)
Cu1–O1W'	2.714(7)	2.657(9)		
Fe1–C1	1.937(4)	1.935(4)	Fe1–C7	1.923(4)
Fe1–C9	1.940(5)	1.948(5)	Fe1–C14	1.939(4)
Fe1–C14	1.947(4)	1.953(4)	Fe1–C15	1.936(4)
Fe1–C16	1.941(4)	1.950(4)	Fe1–C20	1.940(4)
Fe1–C19	1.932(5)	1.929(5)	Fe1–C21	1.951(4)
Fe1–C23	1.937(5)	1.946(5)	Fe1–C22	1.948(4)
Fe1–C≡N	174.7(4) - 177.9(3)	174.9(4) – 177.9(4)	Fe1–C≡N	171.1(5) – 179.0(4)
Ln1–N8–C1	168.9(3)	169.6(3)	Dy1–N8–C7	155.3(3)
Cu1–O1–Ln1	106.10(12)	106.90(12)	Cu1–O1–Dy1	106.42(11)
Cu1–O3–Ln1	105.53(12)	106.27(12)	Cu1–O4–Dy1	106.99(11)

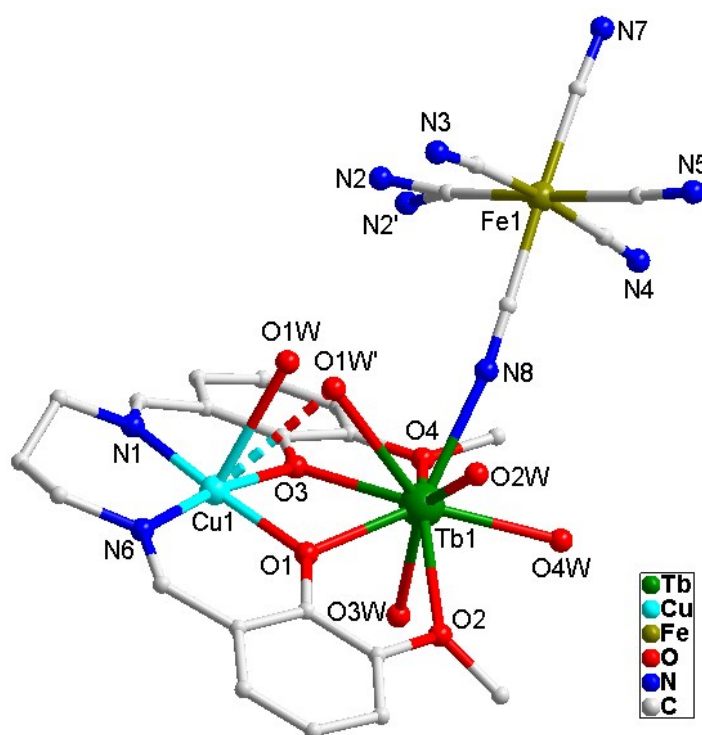
^a In complexes **1** and **2**, the atoms O1W and O1W' are crystallographically disordered, and their occupancies are refined.

Table S2 Selected bond distances (Å) and bond angles (°) for complexes **4-7**.

Complex	4	5	6	7
Ln1-O1	2.386(8)	2.379(5)	2.358(5)	2.350(3)
Ln1-O5	2.343(8)	2.338(4)	2.319(4)	2.307(3)
Ln1-O2	2.376(7)	2.364(5)	2.335(4)	2.325(3)
Ln1-O3	2.637(9)	2.624(5)	2.617(5)	2.621(3)
Ln1-O4	2.549(8)	2.530(5)	2.526(5)	2.502(3)
Ln1-O7	2.573(8)	2.551(5)	2.551(5)	2.530(3)
Ln1-O6	2.597(8)	2.590(5)	2.575(5)	2.588(3)
Ln1-OW1	2.354(8)	2.360(6)	2.327(5)	2.287(3)
Ln1-O8	2.366(8)	2.368(5)	2.348(5)	2.345(3)
Cu2-O5	2.011(8)	1.994(5)	1.999(5)	2.000(3)
Cu2-N1	2.004(10)	1.983(7)	1.988(6)	1.992(3)
Cu2-N2	1.988(10)	1.977(6)	1.969(6)	1.986(3)
Cu2-O8	1.964(8)	1.962(4)	1.964(4)	1.967(3)
Cu2-N10#1	2.167(12)	2.163(8)	2.180(7)	2.193(4)
Cu1-O1	1.992(7)	1.977(5)	1.978(5)	1.978(3)
Cu1-O2	2.001(8)	1.993(5)	1.997(5)	1.997(3)
Cu1-N3	1.981(10)	1.990(6)	1.976(6)	1.988(3)
Cu1-N8	2.158(10)	2.201(7)	2.178(6)	2.182(3)
Cu1-N9	1.983(10)	1.993(6)	1.976(6)	1.987(3)
Fe-C(cyano)	1.919(12) - 1.959(14)	1.904(11) - 1.939(10)	1.914(11) - 1.937(9)	1.921(4) - 1.947(4)
Cu-N≡C	150.4(10), 149.2(11)	151.3(7), 149.2(7)	152.8(6), 149.7(7)	150.0(3), 145.9(3)
Fe-C≡N	172.7(16) - 179.1(13)	175.9(11) - 178.8(9)	175.8(8) - 179.0(8)	175.5(4) - 178.9(5)
Cu1-O1-Ln1	106.9(3)	107.5(2)	107.2(2)	107.22(11)
Cu1-O2-Ln1	107.0(3)	107.5(2)	107.46(19)	107.54(11)
Cu2-O5-Ln1	107.0(4)	107.8(2)	107.4(2)	107.51(11)
Cu2-O8-Ln1	107.8(3)	107.8(2)	107.6(2)	107.20(12)

Table S3 Magnetic data for complexes **1-6**.

	$2S+1L_J$ (Ln^{III})	Theoretical $\chi_m T$ ($cm^3 K mol^{-1}$)	Experimental $\chi_m T$ ($cm^3 K mol^{-1}$)	Weiss constant (K)	constant
CuFeGd (1)	$^8S_{7/2}$	8.64	9.19	1.47	
CuFeTb (2)	7F_6	12.58	12.81	1.30	
CuFeDy (3)	$^6H_{15/2}$	14.93	16.09	14.72	
Cu ₂ FeGd (4)	$^8S_{7/2}$	9.01	9.96	5.55	
Cu ₂ FeTb (5)	7F_6	12.95	13.97	38.96	
Cu ₂ FeDy (6)	$^6H_{15/2}$	15.30	16.97	53.20	

**Fig. S1.** Crystal structure of complex **2**. O1W and O1W' are disordered. Hydrogen and solvent have been omitted for clarity.

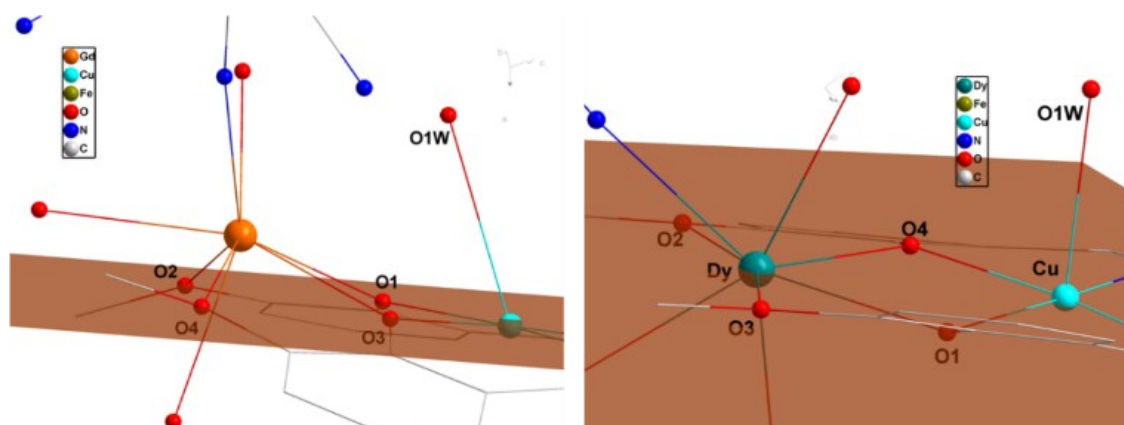


Fig. S2. The relative positions of lanthanide ion deviate from O1O2O3O4 plane in complexes **1** (left) and **3** (right).

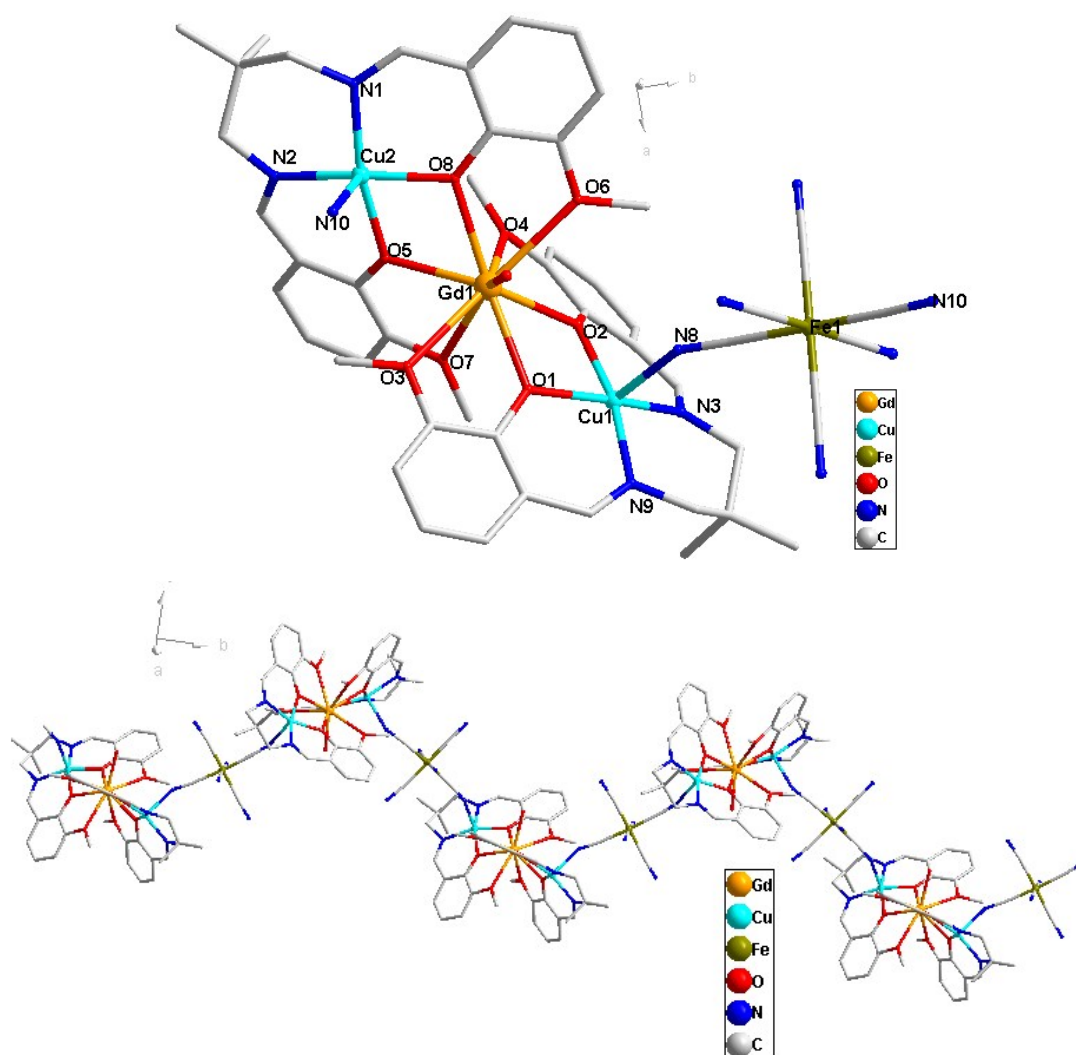


Fig. S3. Crystal structure of complex **4** (above) and the 1D zig-zag chain parallel with b axis (below). Hydrogen and solvent have been omitted for clarity.

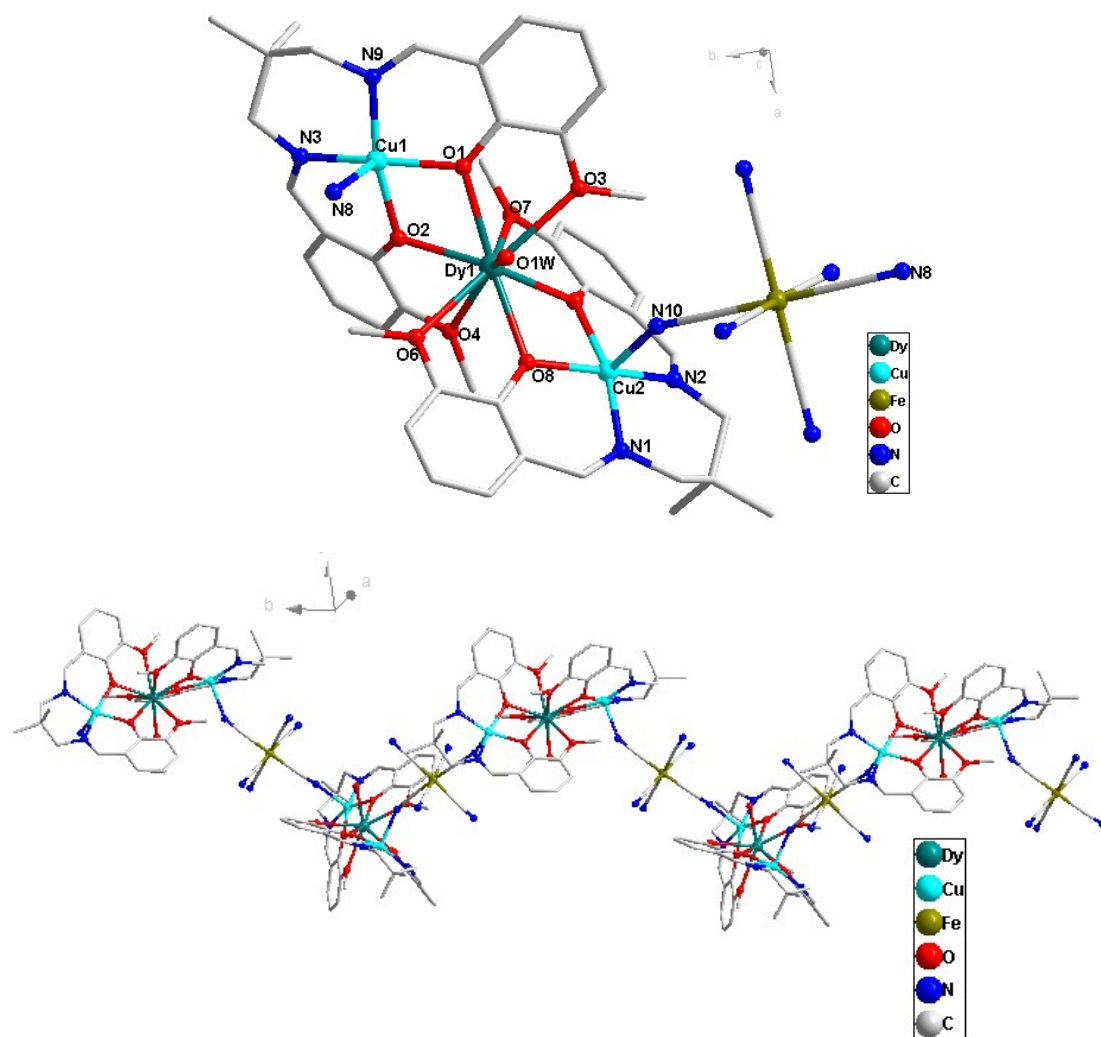


Fig. S4. Crystal structure of complex **6** (above) and the 1D zig-zag chain (below). Hydrogen and solvent have been omitted for clarity.

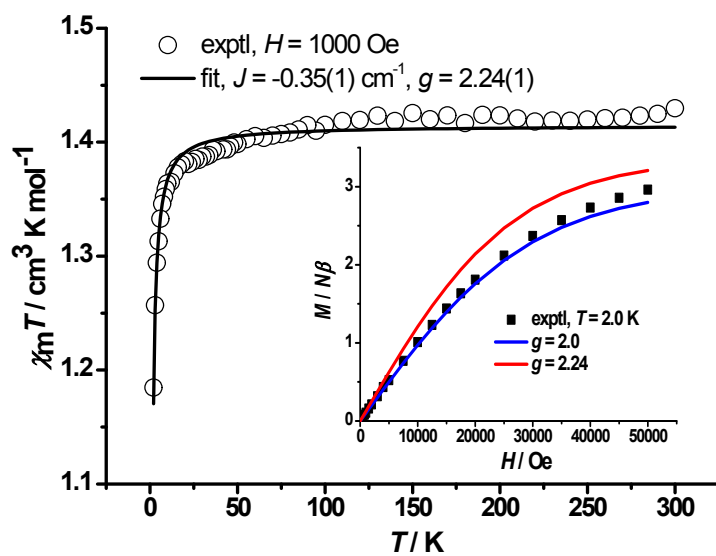


Fig. S5. Temperature dependence of $\chi_m T$ for complex **7**. The solid lines indicate the best fitting results. Inset: Field dependence of magnetization for complex **7** at 2.0 K.

Table S4 Structural and magnetic data for complexes with the CuO₂Gd bridging mode

complexes	$2J_{\text{CuGd}}/\text{cm}^{-1}$	Gd-O ₂ -Cu/°	Cu-O-Gd/°	Gd-Cu/Å	Ref
Complex 1	4.78	19.74	105.82	3.421	/
Complex 4	5.89	18.06	107.18	3.515	/
{[Cu(L ^{α} -Me)] ₂ Gd(H ₂ O)(NO ₃) ₂ }NO ₃	3.62	38.0	97.1,97.9	3.212	20a
[Cu(H ₂ L)(CH ₃ OH)] ₂ Gd(DMF)Fe(CN) ₆	15.94	3.61	107.04	3.450	20b
[Cu(H ₂ L)(CH ₃ OH)] ₂ Gd(DMF)Fe(CN) ₆	9.08	10.35	108.06	3.442	20b
Gd(hfa) ₃ Cu(salen)	0.4	33.78	94.38	3.215	24d
Cu(salabza)Gd(hfac) ₃	0.8	34.28	95.8	3.248	24e
Cu(salen)Gd(pta) ₃	1.21	23.72	97.5	3.288	24f
Cu(acacen)Gd(hfa) ₃	1.25	25.53	97.8	3.301	24f
Gd(hfa) ₃ Cu(salen)(Meim)	1.42	27.72	96.7	3.252	24d
Cu(acacen)Gd(pta) ₃	1.47	29.81	96.3	3.274	24f
Cu(prpen)Gd(hfac) ₃	1.91	30.78	97.1	3.252	24g
CuGd(hmp) ₂ (NO ₃) ₃ (H ₂ O) ₂	3.36	15.32	102.4	3.246	24h
[CuGd(ems)(NO ₃) ₃ ·3H ₂ O]Cu(ems)	3.76	17.77	99.5	3.306	24i
L ^a Cu(C ₃ H ₆ O)Gd(NO ₃) ₃	4.8	13.52	107.8	3.523	24j
L ^b CuGd(NO ₃) ₃	4.98	14.12	103.6	3.443	24k
L ^c Cu(H ₂ O)Gd(NCS) ₃ ·Me ₂ CO	5.5	15.44	106.5	3.454	24l
L ^d CuGd(NO ₃) ₃ ·Me ₂ CO	5.6	9.60	109.4	3.425	24m
L ^a Cu(MeOH)Gd(NO ₃) ₃	6.8	10.00	107.8	3.484	24j
L ^e CuGd(NO ₃) ₃ ·Me ₂ CO	7.0	10.15	106.7	3.428	24n
L ^f CuGd(NO ₃) ₃ ·Me ₂ CO	7.3	12.7	107.1	3.475	20c
L ¹ CuGd(NO ₃) ₃ ·Me ₂ CO	7.4	10.18	107.6	3.475	24o
L ^g CuGd(NO ₃) ₃	7.6	9.22	108.0	3.499	24p
L ^h CuGd(NO ₃) ₃	8.08	3.36	105.2	3.400	24q
L ⁱ CuGd(NO ₃) ₃ ·Me ₂ CO	8.63	11.11	107.0	3.454	24r
L ⁱ CuGd(NO ₃)(N(CN) ₂) ₂ ·Me ₂ CO	8.50	3.27	107.67	3.481	24r
[L ² CuCl ₂ Gd(H ₂ O) ₄]Cl·2H ₂ O	10.1	1.38	108.8	3.512	24s
L ^j CuGd(NO ₃) ₃	10.8	7.1	106.6	3.449	24t
L ^k CuGd(NO ₃) ₃	12.6	2.10	105.0	3.401	24t
[L ^m ₂ Cu ₂ Gd ₂ (NO ₃) ₆ (H ₂ O) ₂]	4.5	20.44	103.0	3.395	24c
[{CuL ¹ }Gd(H ₂ O) ₃ {Co(CN) ₆ }]	7.24	10.19	106.52	3.508	6b
[(bipy)W(CN) ₆ Cu(L ¹)Gd(NO ₃) ₂] ₃	8.67	12.89	107.06	3.466	12c
[{CuL ¹ } ₂ Gd(TCNQ) ₂]·TCNQ	7.7	17.13	107.24	3.477	24u
[GdCu ₃ L ⁿ (hfac) ₃] ⁻	0.86	39.08	96.09	3.393	24v
L ¹ CuGd(hfac) ₂ (ac)	2.6	24.47	104.49	3.437	24a
Gd ₆ Cu ₃ (OH)(pdm) ₃ (O ₂ C ^t Bu) ₉ (CO ₃) ₄	0.8	24.34	102.78	3.491	24x
[CuLn(L ^p)(NO ₃) ₂ (H ₂ O) ₃ MeOH]NO ₃	7.89	2.52	107.94	3.539	24y
[Cu(valaepy) ₂ Gd(O ₂ NO) ₃ (H ₂ O)]	4.94	2.52	108.67	3.506	24b
GdCuL ^a Cl ₃ (CH ₃ OH) ₂	11.2	15.39	/	3.409	24w
[Gd ₂ Cu ₄ L ^r ₄ Cl ₂ (OH)](PF ₆) ₃	9.8	23.36	106.30	3.479	24w

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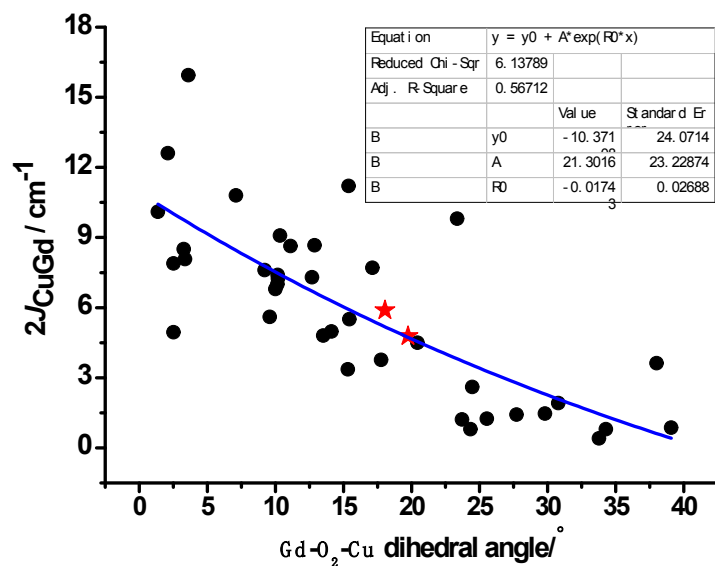


Fig. S6. Magneto-structural correlation for bis-phenoxo-bridged Cu(II)-Gd(III) complexes. The solid line represents the exponential fit to the experimental data. The pentagrams correspond to complexes **1** and **4**.

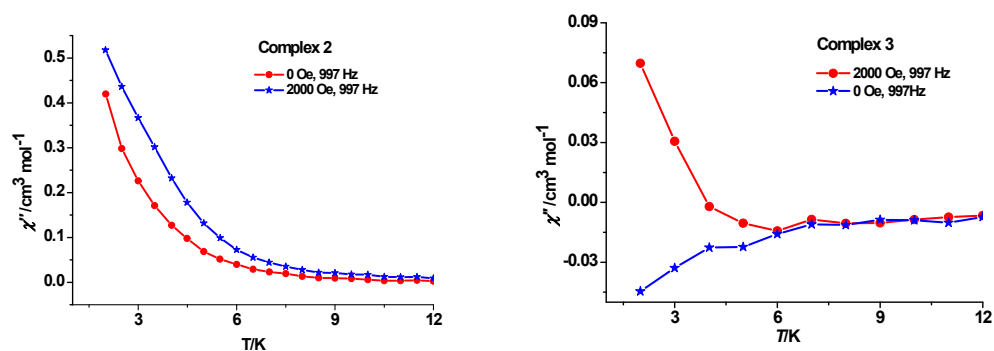


Fig. S7. Temperature dependence of the out-of-phase (χ''_m) parameters for complexes **2** (left) and **3** (right) at 997 Hz.

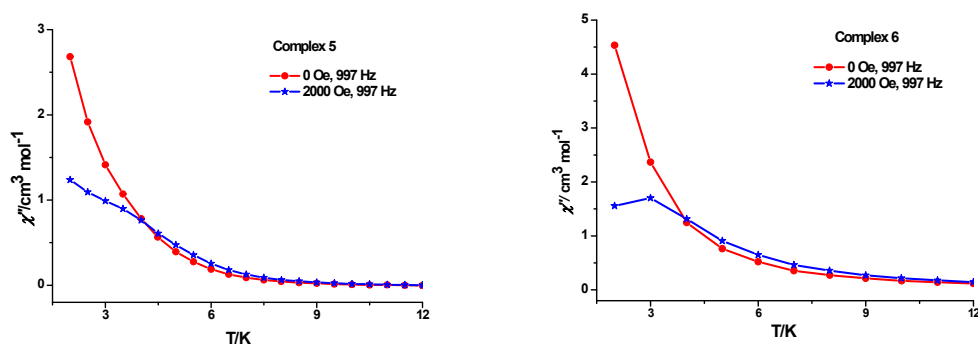


Fig. S8. Temperature dependence of the out-of-phase (χ''_m) parameters for complexes **5** (left) and **6** (right) at 997 Hz.

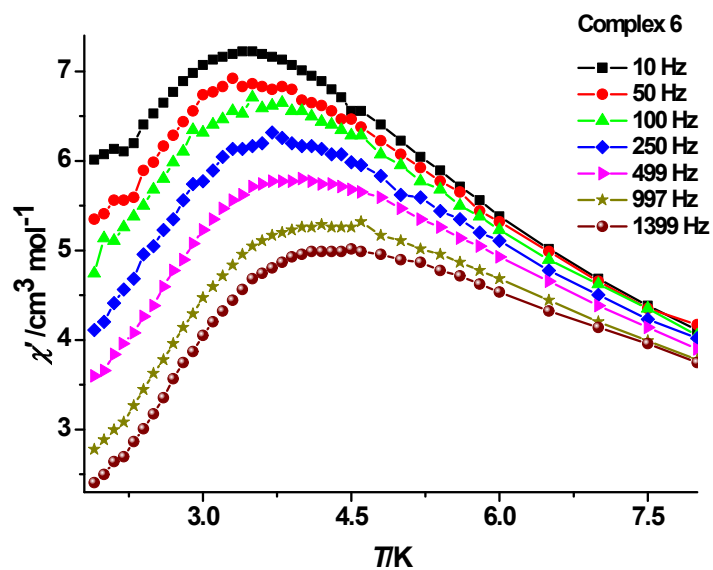


Fig. S9. Temperature dependence of the out-of-phase (χ'_m) parameters for complex **6**.

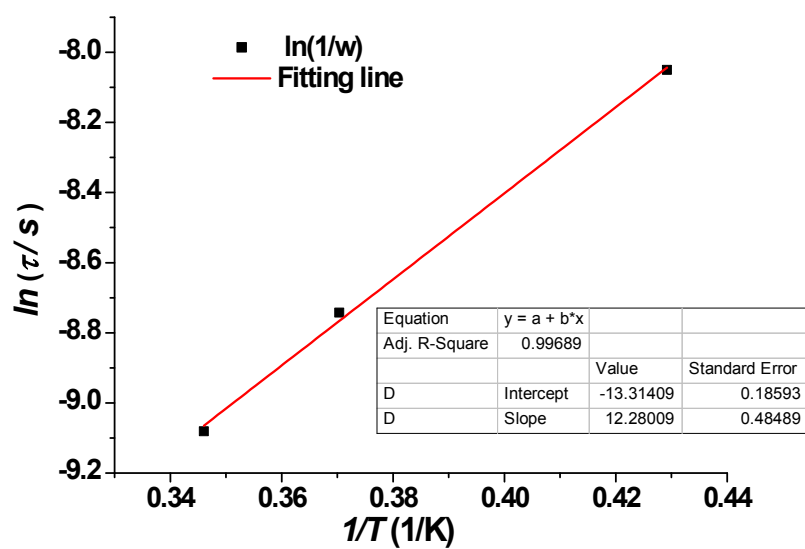


Fig. S10. The $\ln(\tau)$ vs. T^{-1} plots based on the Arrhenius relationship for complex **6**.