

## Solvent Extraction of Americium(III) and Europium(III) with Tridentate N,N-dialkyl-1,10-Phenanthroline-2-Amides-Derived Ligands: Extraction, Complexation and Theoretical Study

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### Electronic supplement information

#### 1. FT-IR

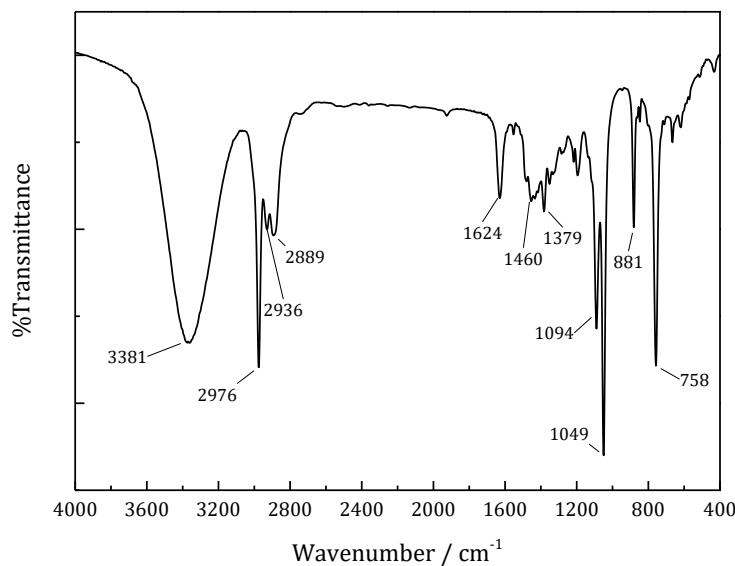


Fig. S1 FT-IR spectrum of C2-PTA

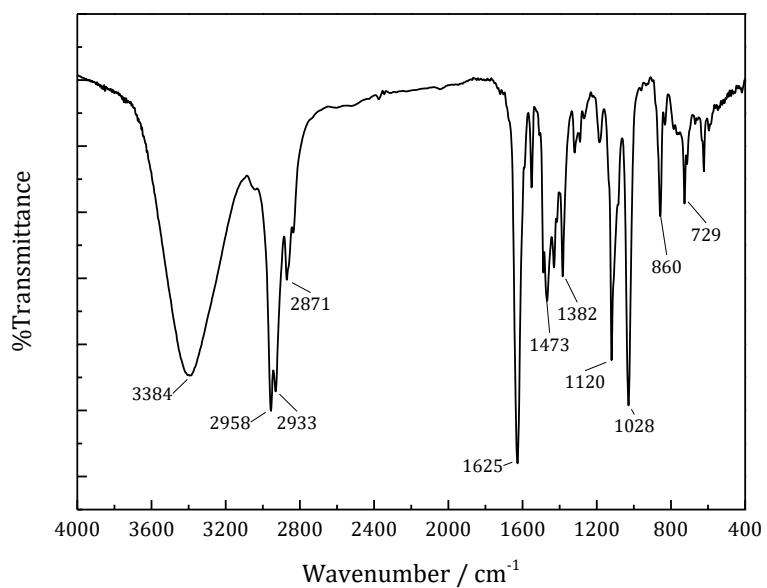


Fig. S2 FT-IR spectrum of C4-PTA

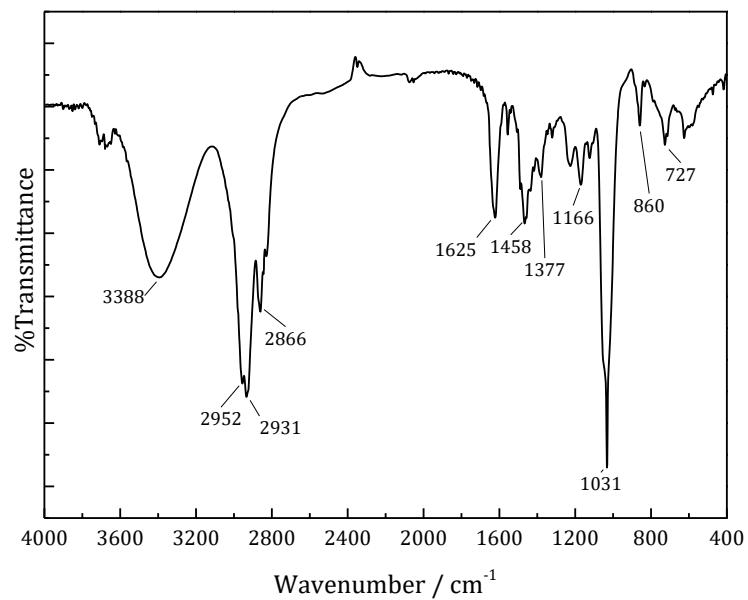


Fig. S3 FT-IR spectrum of C6-PTA

## 2. <sup>1</sup>H NMR

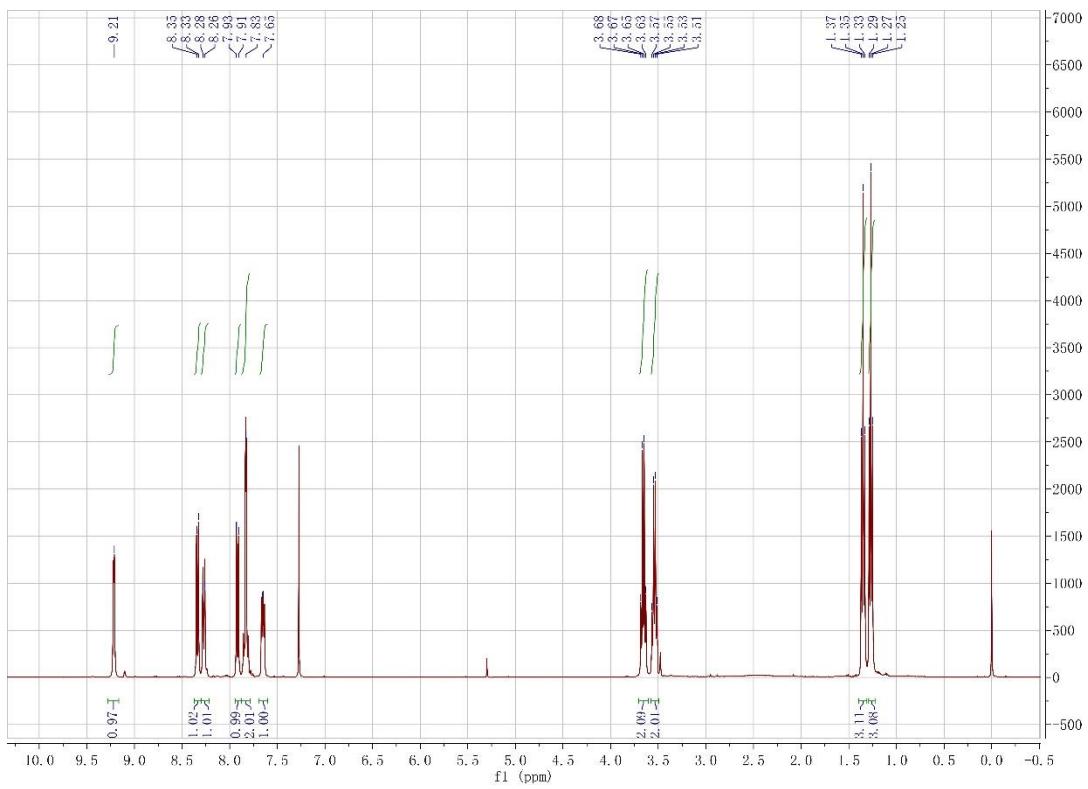


Fig. S4  $^1\text{H}$  NMR spectrum of C2-PTA

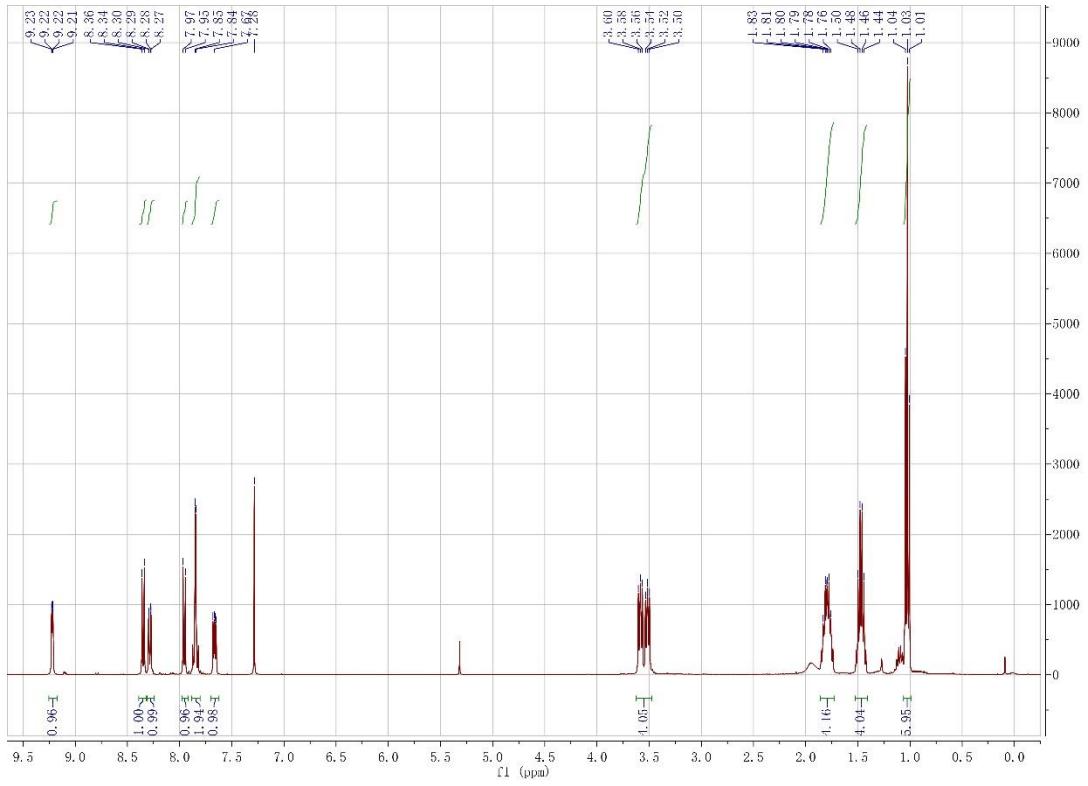


Fig. S5  $^1\text{H}$  NMR spectrum of C4-PTA

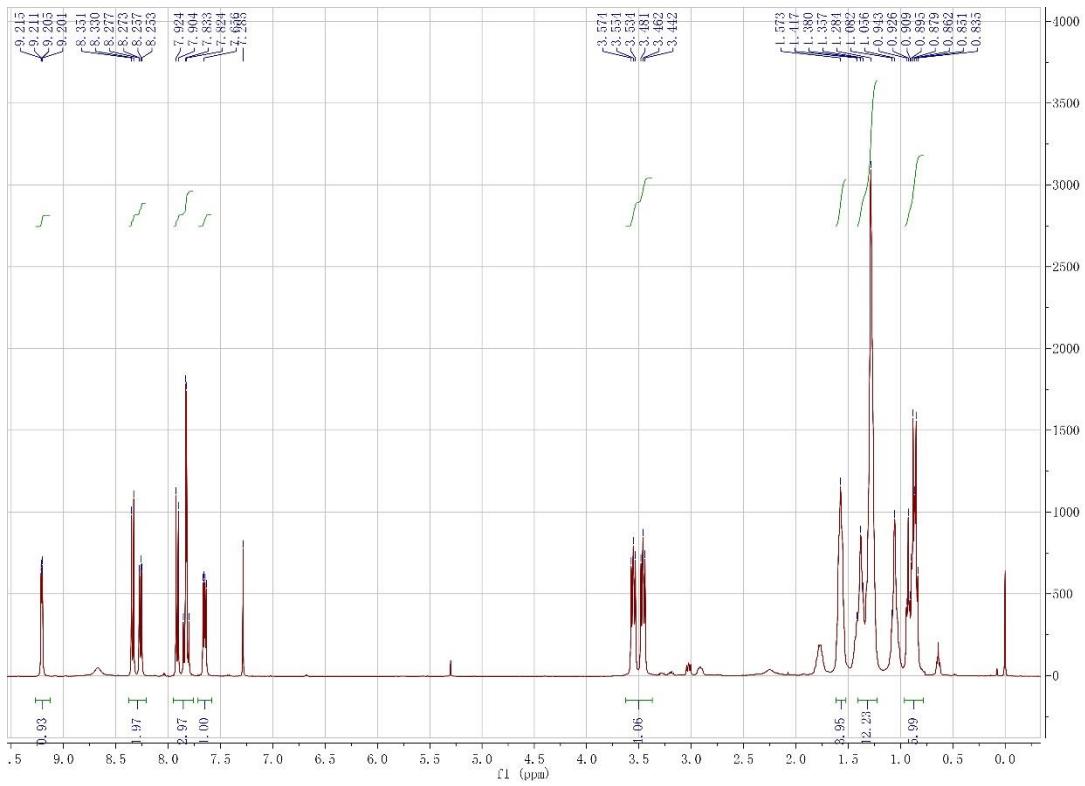


Fig. S6 <sup>1</sup>H NMR spectrum of C6-PTA

### 3. <sup>13</sup>C NMR

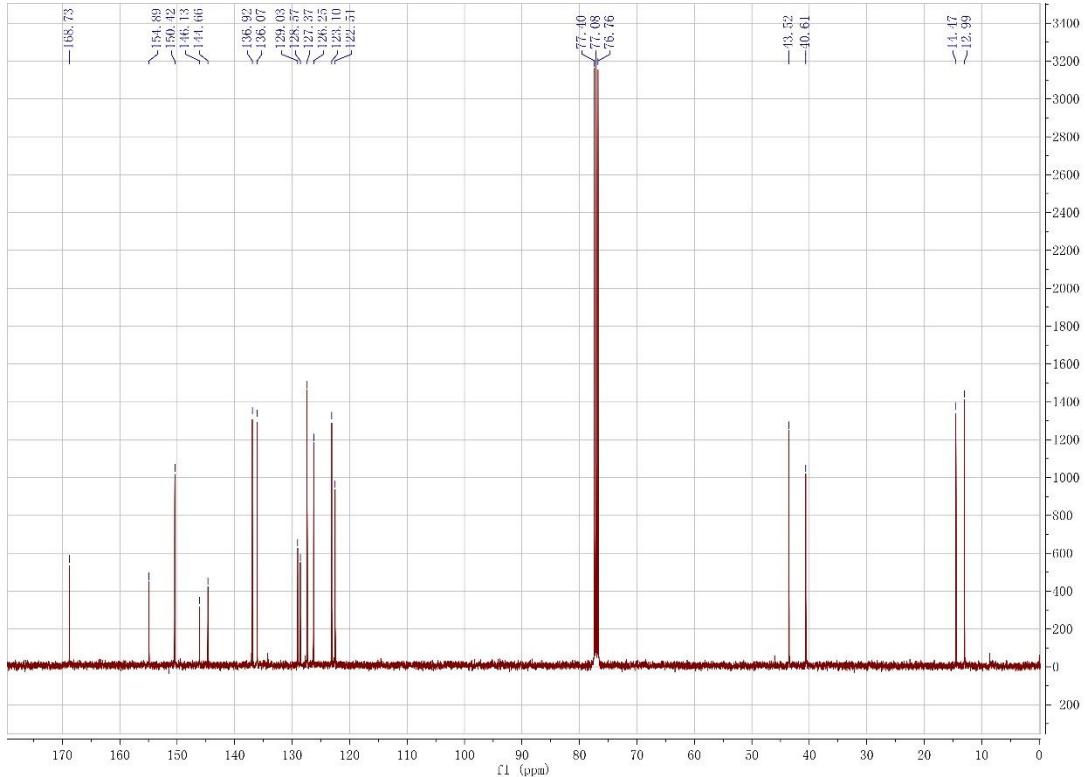


Fig. S7 <sup>13</sup>C NMR spectrum of C2-PTA

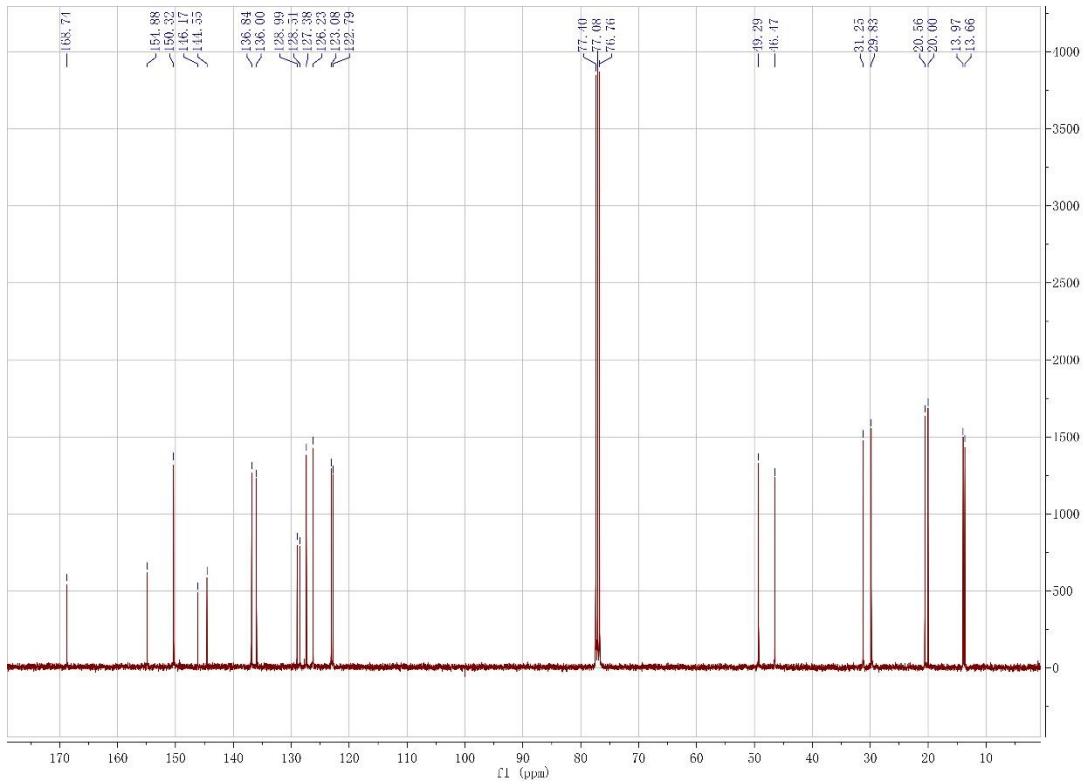


Fig. S8  $^{13}\text{C}$  NMR spectrum of C4-PTA

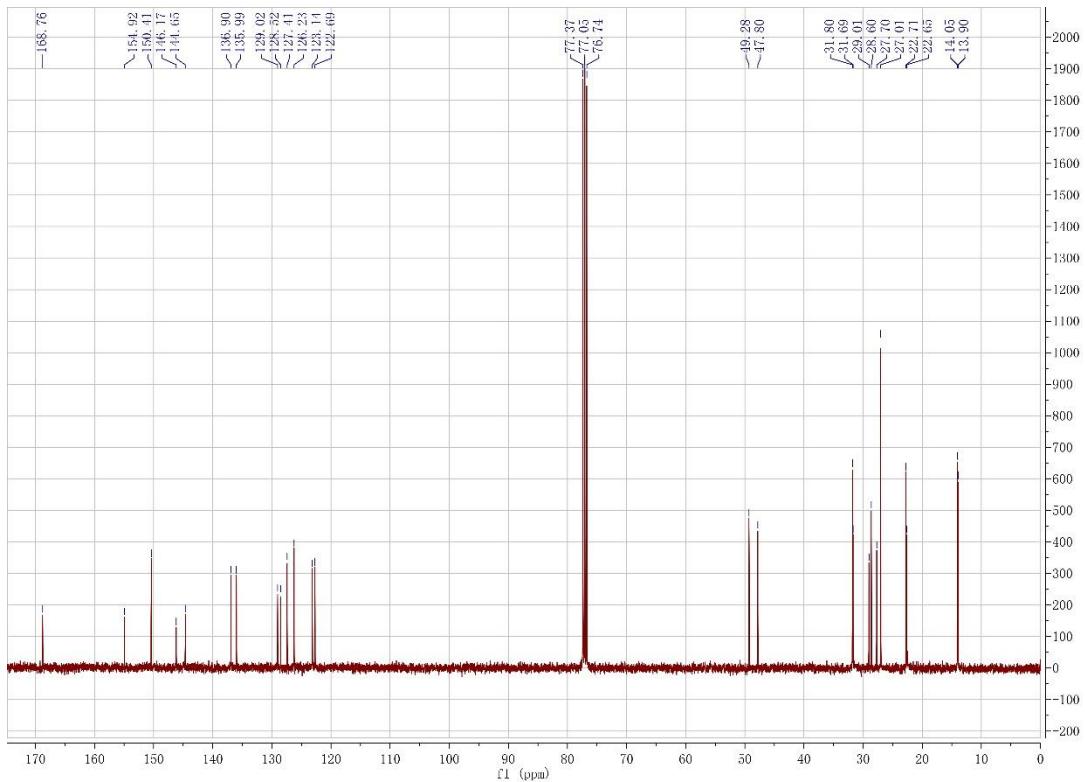


Fig. S9  $^{13}\text{C}$  NMR spectrum of C6-PTA

#### 4. ESI-MS

Acquisition Parameter

Source Type: ESI  
Focus: Active  
Scan Begin: 100 m/z  
Scan End: 1000 m/z  
Ion Polarity: Positive  
Set Capillary: 4500 V  
Set End Plate offset: -500 V  
Set Collision Cell RF: 280.0 Vpp  
Set Nebulizer: 0.4 Bar  
Set Dry Heater: 180 °C  
Set Dry Gas: 4.0 l/min  
Set Divert Valve: Waste

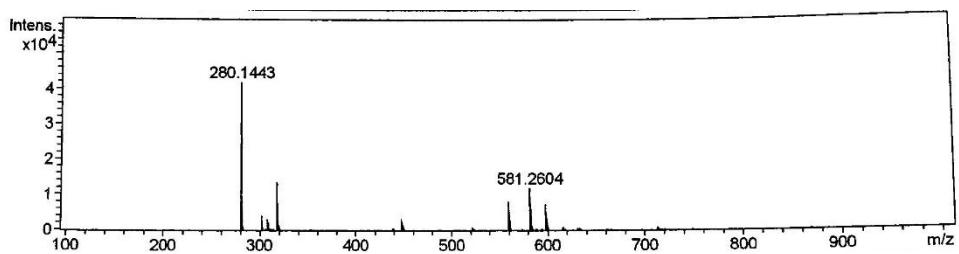


Fig. S10 ESI-MS spectrum of C2-PTA.  
Formula: C<sub>17</sub>H<sub>18</sub>N<sub>3</sub>O. m/z: 280.1443. Err: 1.8 ppm.

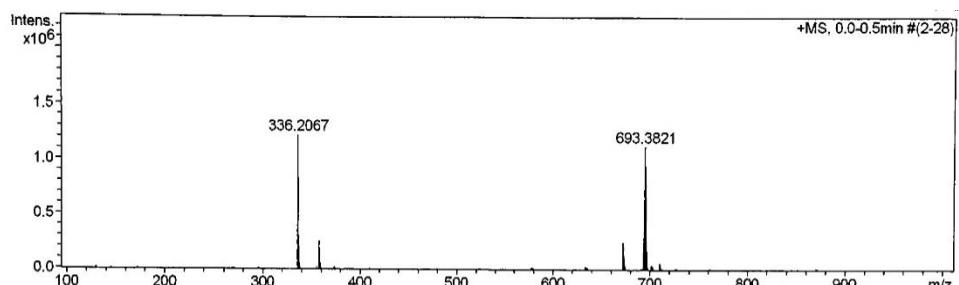


Fig. S11 ESI-MS spectrum of C4-PTA.  
Formula: C<sub>21</sub>H<sub>26</sub>N<sub>3</sub>O. m/z: 336.2070. Err: 0.9 ppm.

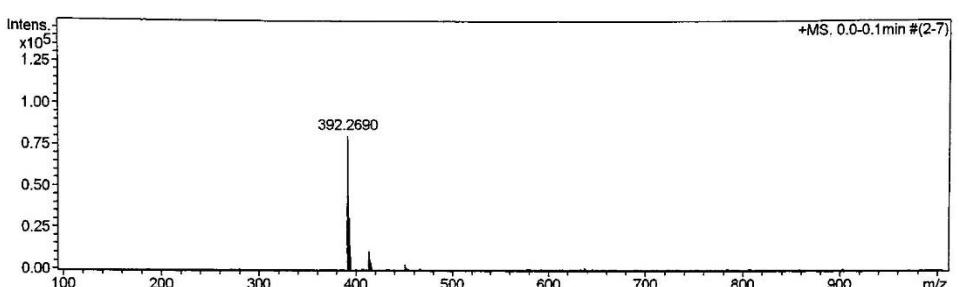


Fig. S12 ESI-MS spectrum of C6-PTA.  
Formula: C<sub>25</sub>H<sub>34</sub>N<sub>3</sub>O. m/z: 392.2696. Err: 1.6 ppm.

## 2. Solvent extraction data

### Effect of diluents

Solubilities test shows that C2-PTA is well soluble in cyclohexanone, 1,2-dichloroethane, 2-chrolomethane and chloroform, incompletely soluble in n-octanol and tert-butylbenzene (TBB), and almost insoluble n-hexane and kerosene. The extraction behaviors of Am(III) and Eu(III) with C2-PTA in such diluents were measured, and the results are shown in Table S1.

In spite of the good performance of tetradentate N,N'-diethyl-N,N'-ditolyl-2,9-diamide-1,10-phenanthroline (Et-Tol-DAPhen) in the solvent extraction in cyclohexanone [Xiao, C.-L., et al. (2014). *Inorganic chemistry* 53(3): 1712-1720.], the solubilities of cyclohexanone (8.6 g/100 mL) is much higher than 1,2-dichloroethane (0.81 g/100 mL) and chloroform (0.82 g/100 mL) at 293 K in water, which may lead to a vulnerable organic phase as well as an unstable extraction process. And as is seen from Table S1, the distribution ratios of both Am(III) and Eu(III) with C2-PTA present much higher in chloroform than in cyclohexanone and 1,2-dichloroethane. Such results showed that chloroform is favorable for the extraction of Am(III) and Eu(III) with C2-PTA.

Table S1 Solubilities of C2-PTA, the distribution ratios and the extraction efficient of Am(III) and Eu(III). Organic phase: 0.2 mol/L C2-PTA. Aqueous phase:  $10^{-3.63}$  mol/L  $\text{HNO}_3 + ^{241}\text{Am}/^{152}\text{Eu}$  radiotracers.  $T = 293$  K. The statistical error values were less than 5%.

Diluent	Cyclohexanone	1,2-dichloroethane	Chloroform
$D(\text{Am})$	0.23	2.2	25.2
$D(\text{Eu})$	0.33	1.4	3.3
$E(\text{Am})$	19%	69%	96%
$E(\text{Eu})$	25%	58%	76%

### Effect of extractant concentration

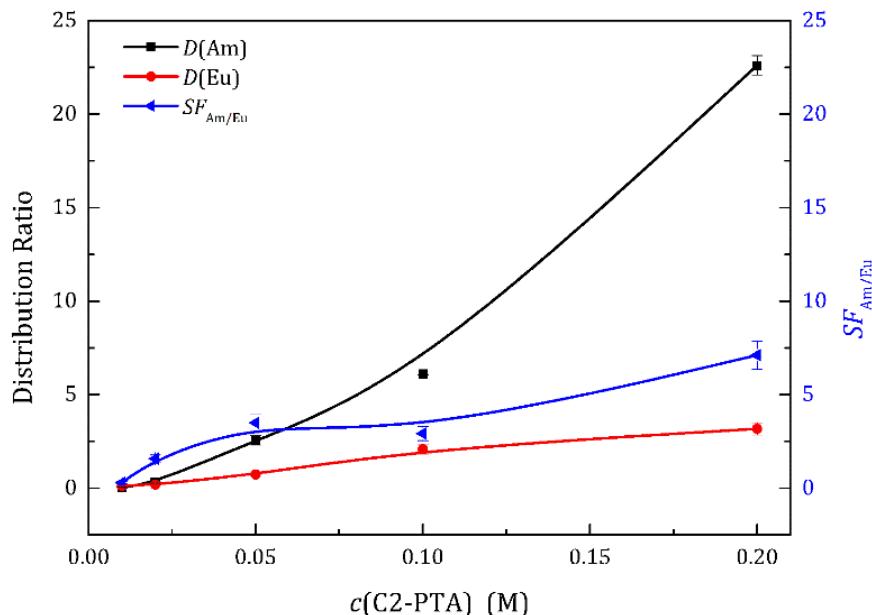


Fig. S13 Effect of extractant concentration. Organic phase: C2-PTA in chloroform. Aqueous phase:  $10^{-3.63}$  mol/L  $\text{HNO}_3 + ^{241}\text{Am}/^{152}\text{Eu}$  tracers.  $T = 293$  K.

### 3. Solution spectroscopy

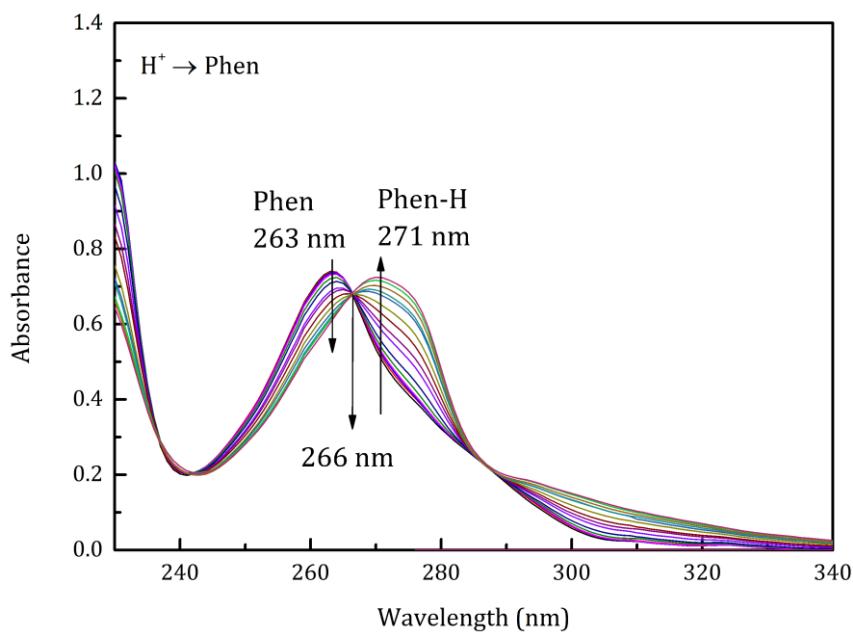


Fig. S14 UV-vis titration of Phen by  $\text{H}^+$  in  $\text{HNO}_3$ /methanol media.  $c_{\text{Phen}} = 2.5 \times 10^{-5}$  mol/L,  $c_{\text{H}^+} = 0-4.0 \times 10^{-5}$  mol/L,  $T = 293$  K.

#### 4. Optimized structures

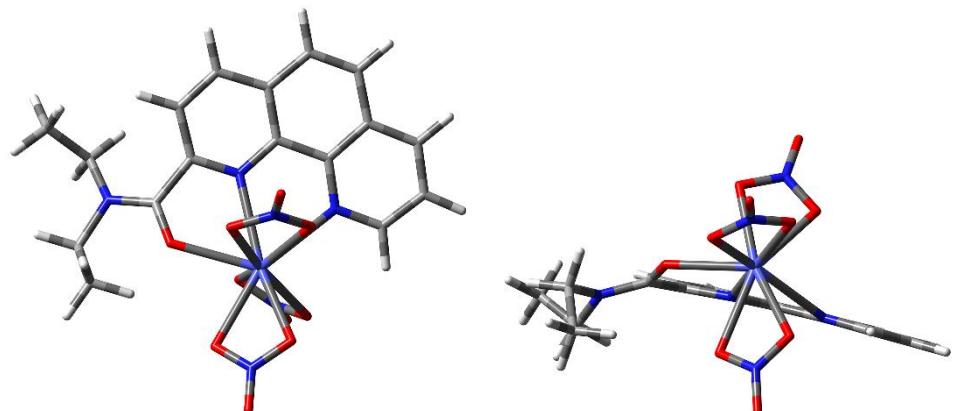


Fig. S15 Optimized structures of  $\text{Am}(\text{C}_2\text{-PTA})(\text{NO}_3)_3$  in front view (left) and lateral view (right).

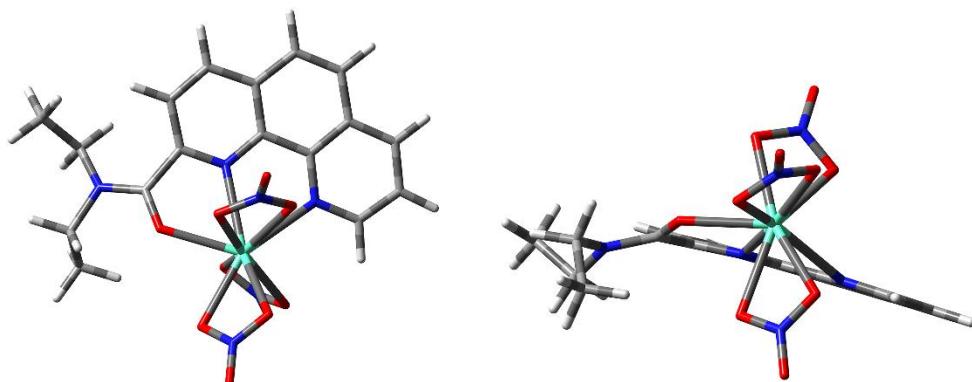
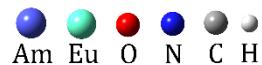


Fig. S16 Optimized structures of Eu (C<sub>2</sub>-PTA)(NO<sub>3</sub>)<sub>3</sub> in front view (left) and lateral view (right).

## 5. Cartesian coordinates in the optimized structures

Table S2 Cartesian coordinates in the optimized structure of Am(C<sub>2</sub>-PTA)(NO<sub>3</sub>)<sub>3</sub>

Center Number	Atomic Number	Atomic Type	Standard orientation		
			X	Y	Z
1	6	0	-0.571877	4.192021	0.634449
2	6	0	-1.646207	3.320528	0.630439
3	6	0	-1.439161	1.984736	0.227706
4	7	0	-0.226233	1.535516	-0.086798
5	6	0	0.824312	2.378282	-0.122889
6	6	0	0.702791	3.751576	0.221240
7	6	0	2.104493	1.851447	-0.548551
8	6	0	3.211060	2.730901	-0.641333
9	6	0	3.053941	4.111920	-0.277189
10	6	0	1.853951	4.603220	0.143953
11	7	0	2.186113	0.534329	-0.861511
12	6	0	3.344689	0.056743	-1.300572
13	6	0	4.499490	0.852974	-1.437101
14	6	0	4.434271	2.188458	-1.097654
15	6	0	-2.477064	0.879104	0.168616
16	8	0	-2.079243	-0.280003	0.399987
17	7	0	-3.770557	1.109424	-0.138807
18	6	0	-4.660540	-0.074518	-0.189516
19	6	0	-4.579996	-0.818599	-1.525315
20	1	0	-0.705240	5.222694	0.952103
21	1	0	-2.620414	3.657136	0.960801
22	1	0	3.920897	4.763704	-0.343217
23	1	0	1.749683	5.648596	0.420298
24	1	0	5.419440	0.405731	-1.798211
25	1	0	5.307940	2.829794	-1.179224
26	1	0	-5.676764	0.284950	-0.001540
27	1	0	-4.377419	-0.734453	0.631170
28	1	0	-5.253950	-1.681711	-1.502626
29	1	0	-3.564793	-1.179381	-1.709507
30	1	0	-4.882005	-0.180256	-2.363732
31	7	0	0.010681	-3.883485	0.872012
32	8	0	-0.085481	-5.061541	1.129349
33	8	0	1.131664	-3.248836	0.938326
34	8	0	-1.002287	-3.166836	0.514680
35	7	0	1.130816	-0.169650	2.825018
36	8	0	-0.110028	-0.357190	2.546298

37	8	0	1.509208	0.188686	3.920046
38	8	0	1.948492	-0.371241	1.847448
39	7	0	0.228089	-1.622440	-2.714469
40	8	0	-0.700470	-0.936836	-2.148223
41	8	0	0.266426	-1.805508	-3.912615
42	8	0	1.123741	-2.100882	-1.917717
43	6	0	-4.319552	2.352001	-0.701869
44	1	0	-4.861363	2.076006	-1.613966
45	1	0	-3.497497	2.994126	-1.021397
46	6	0	-5.260952	3.095599	0.250296
47	1	0	-4.753446	3.399795	1.172035
48	1	0	-5.651385	3.993488	-0.241039
49	1	0	-6.115179	2.473715	0.536115
50	1	0	3.356840	-0.998090	-1.560386
51	95	0	0.261413	-1.085459	0.171003

Table S3 Cartesian coordinates in the optimized structure of Eu(C<sub>2</sub>-PTA)(NO<sub>3</sub>)<sub>3</sub>

Center Number	Standard orientation			Coordinates (Angstroms)		
	Atomic Number	Atomic Type	X	Y	Z	
1	6	0	0.035418	4.107204	0.590091	
2	6	0	-1.151501	3.396046	0.572214	
3	6	0	-1.126155	2.038713	0.189499	
4	7	0	0.018689	1.418947	-0.087988	
5	6	0	1.176807	2.104508	-0.114800	
6	6	0	1.243140	3.486691	0.208777	
7	6	0	2.376167	1.396297	-0.512283	
8	6	0	3.597008	2.109778	-0.592707	
9	6	0	3.630001	3.504084	-0.246391	
10	6	0	2.504212	4.166493	0.144344	
11	7	0	2.273273	0.078683	-0.812848	
12	6	0	3.360342	-0.563113	-1.222863	
13	6	0	4.619762	0.058870	-1.343587	
14	6	0	4.739111	1.394390	-1.019770	
15	6	0	-2.303145	1.085173	0.096056	
16	8	0	-2.067434	-0.122327	0.304421	
17	7	0	-3.549499	1.485623	-0.228393	
18	6	0	-4.579790	0.424525	-0.329054	
19	6	0	-4.554153	-0.294574	-1.680750	
20	1	0	0.041986	5.150603	0.893943	
21	1	0	-2.073531	3.872091	0.879149	
22	1	0	4.581319	4.026381	-0.302464	
23	1	0	2.543788	5.220231	0.405398	

24	1	0	5.473273	-0.519047	-1.681846
25	1	0	5.696219	1.904462	-1.090962
26	1	0	-5.547601	0.905706	-0.158654
27	1	0	-4.405460	-0.283316	0.481948
28	1	0	-5.331238	-1.066201	-1.697788
29	1	0	-3.587214	-0.776578	-1.846667
30	1	0	-4.748922	0.395390	-2.510071
31	7	0	-0.575624	-3.870380	0.953803
32	8	0	-0.847224	-5.014395	1.239526
33	8	0	0.553147	-3.329888	1.261128
34	8	0	-1.397862	-3.101850	0.324195
35	7	0	0.938788	-0.367844	2.841987
36	8	0	-0.298497	-0.528039	2.537668
37	8	0	1.301247	-0.057478	3.957886
38	8	0	1.770097	-0.537499	1.872559
39	7	0	0.188541	-1.848198	-2.610178
40	8	0	-0.622262	-0.966169	-2.150714
41	8	0	0.276218	-2.100310	-3.794254
42	8	0	0.914243	-2.449168	-1.728279
43	63	0	0.091122	-1.213860	0.209394
44	6	0	-3.932807	2.804102	-0.754622
45	1	0	-4.507313	2.624008	-1.670543
46	1	0	-3.035531	3.343523	-1.060700
47	6	0	-4.769888	3.633984	0.223387
48	1	0	-4.223112	3.850046	1.147539
49	1	0	-5.050678	4.584888	-0.242366
50	1	0	-5.691340	3.112953	0.501664
51	1	0	3.224917	-1.611850	-1.472410