

**A nitroolefin functionalized DPP fluorescent probe for selective detection of  
hydrogen sulfide**

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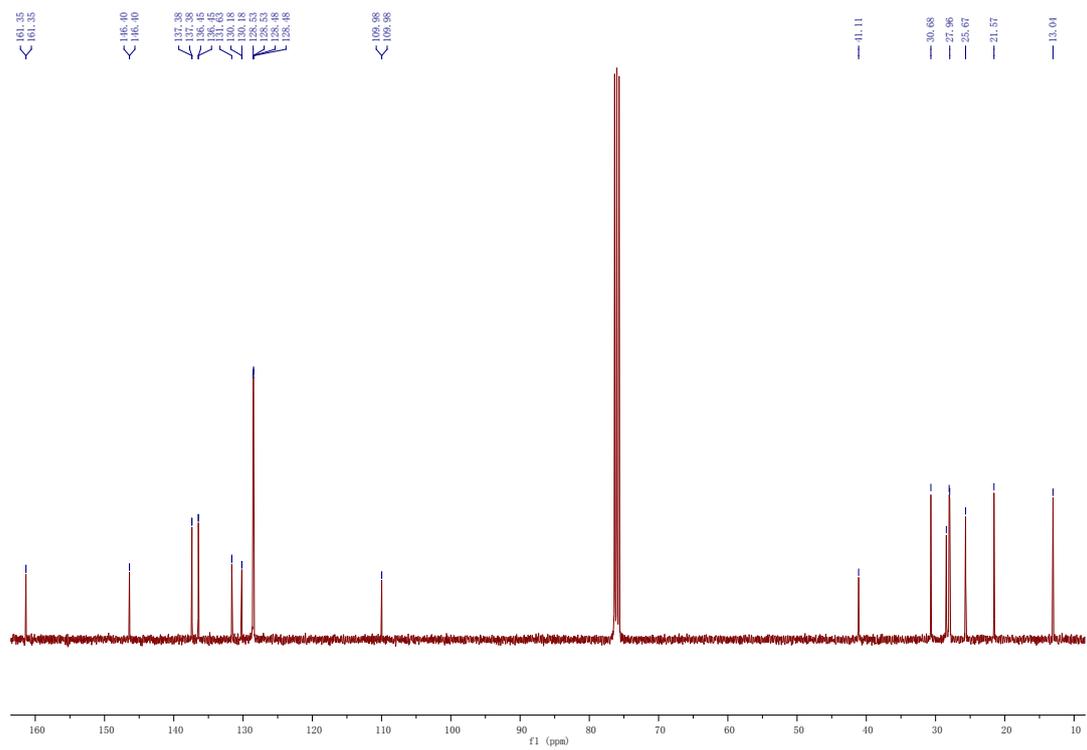


Figure S2.  $^{13}\text{C}$  NMR spectrum of probe **DPP-NO<sub>2</sub>** ( $\text{CDCl}_3$ , 100 MHz).

## Generic Display Report

### Analysis Info

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Method esi\_pos\_50-1000\_with calibration\_for 1min.m  
Sample Name 16082911  
Comment

Acquisition Date 8/30/2016 5:19:00 PM

Operator HSJ  
Instrument maXis impact

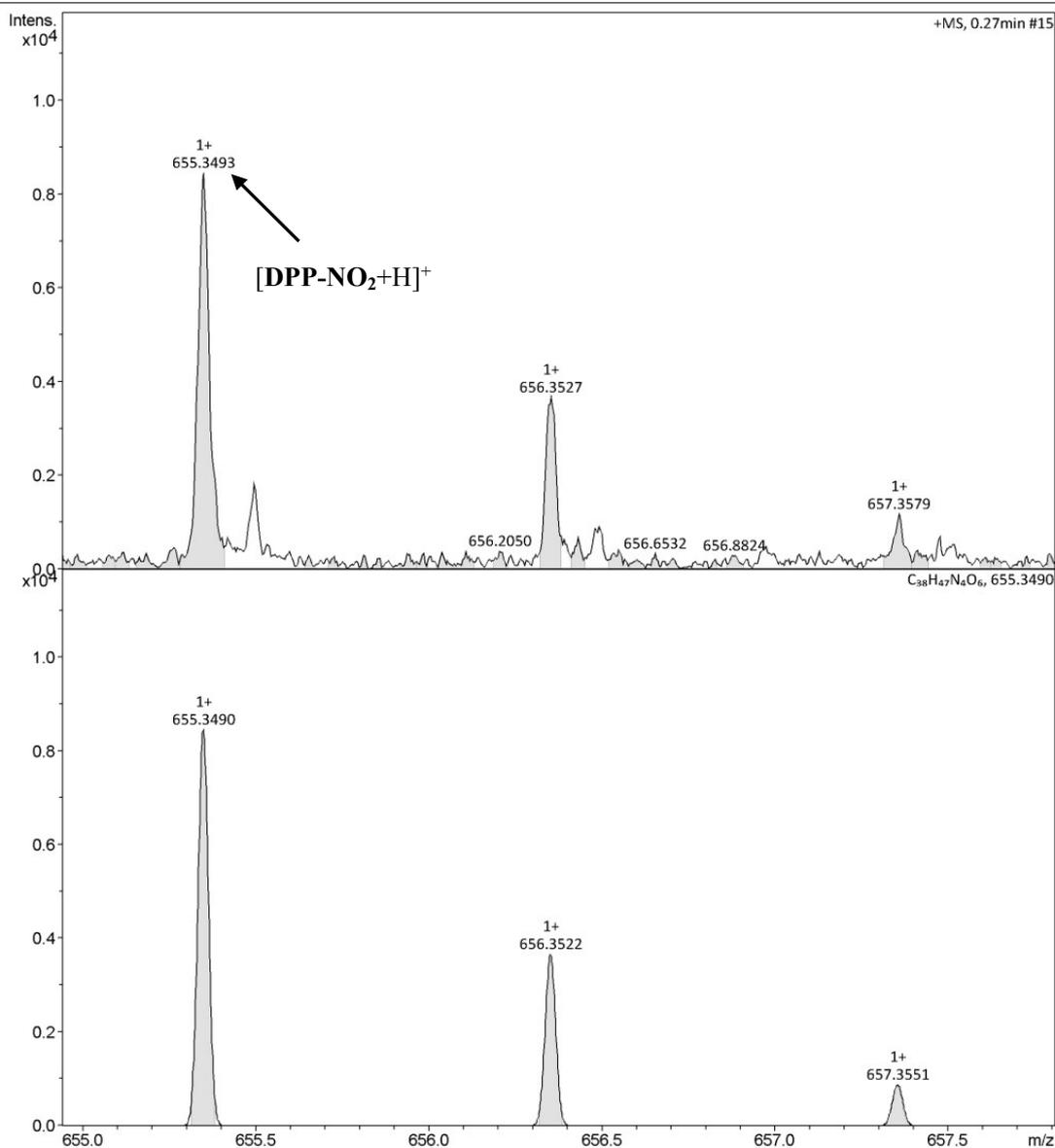


Figure S3. HRMS-ESI spectrum of probe **DPP-NO<sub>2</sub>**.

### Synthesis of **DPP-NH<sub>2</sub>**

65.4 mg (0.1 mmol) **DPP-NO<sub>2</sub>** was dissolved in 30 mL THF, then 112.8 mg (0.5 mmol) stannous chloride and 2 mL hydrochloric acid was added into the reaction

mixture and the reaction mixture was reacted at room temperature for 30 min. Then 30 mL ethyl acetate was added and the mixture were washed with sodium bicarbonate solution to pH=7.0. Then the organic organic layer was separated and dried over Na<sub>2</sub>SO<sub>4</sub>. After concentration under reduced pressure, the crude product was purified by chromatography with a mixture of CH<sub>3</sub>OH and CH<sub>2</sub>Cl<sub>2</sub> (1:20, v:v) to afford the orange solid product **DPP-NH<sub>2</sub>** 15.6 mg in 26.3% yield.

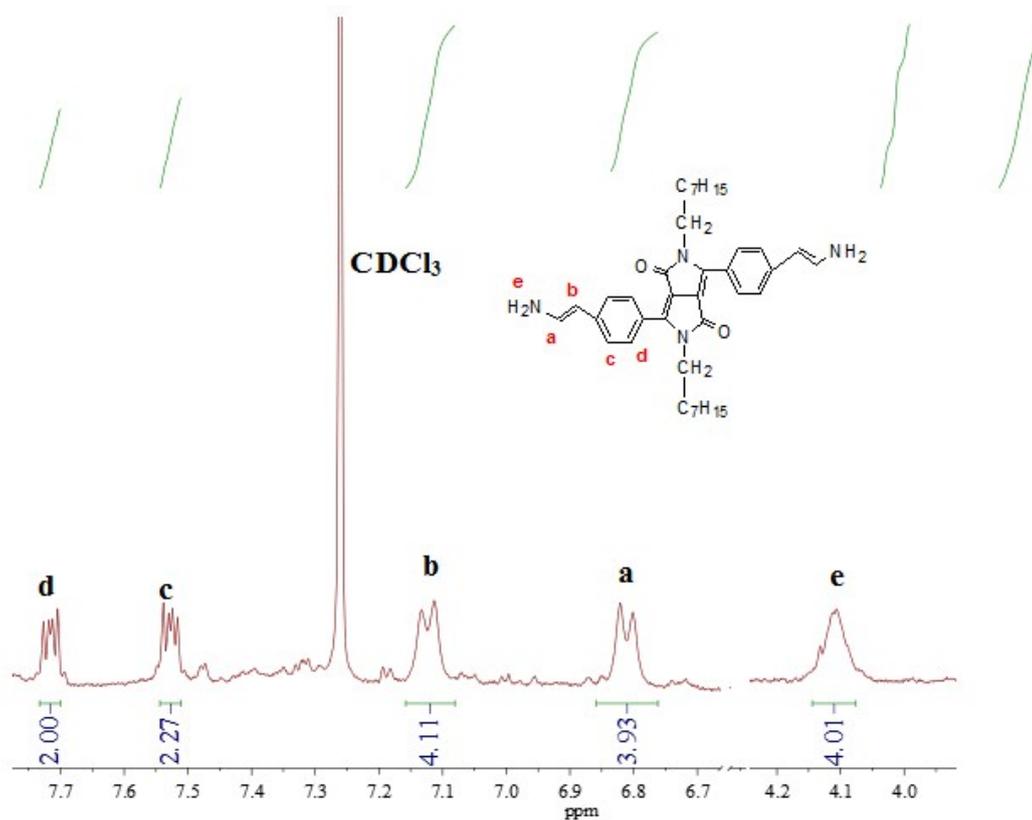


Figure S4. The partial <sup>1</sup>H NMR spectrum of **DPP-NH<sub>2</sub>** (CDCl<sub>3</sub>, 400 MHz).

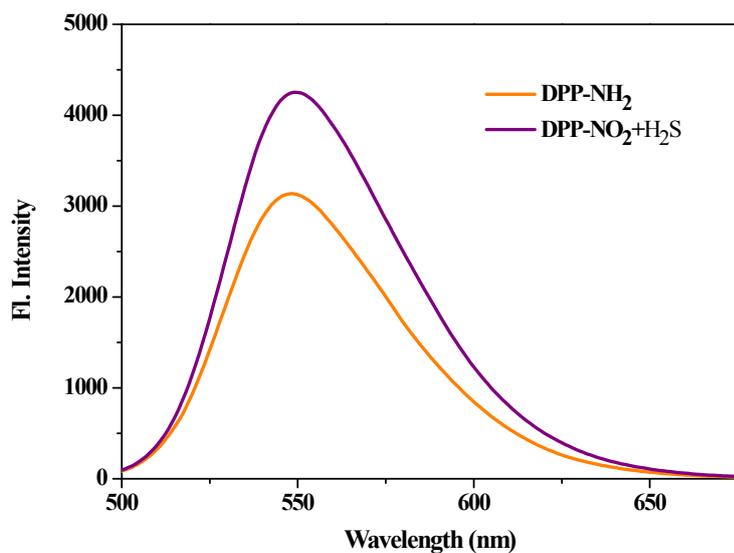


Figure S5. The fluorescence emission spectra comparison between **DPP-NH<sub>2</sub>** (10  $\mu$ M) in CH<sub>3</sub>CN and **DPP-NO<sub>2</sub>** (10  $\mu$ M) + H<sub>2</sub>S in CH<sub>3</sub>CN.

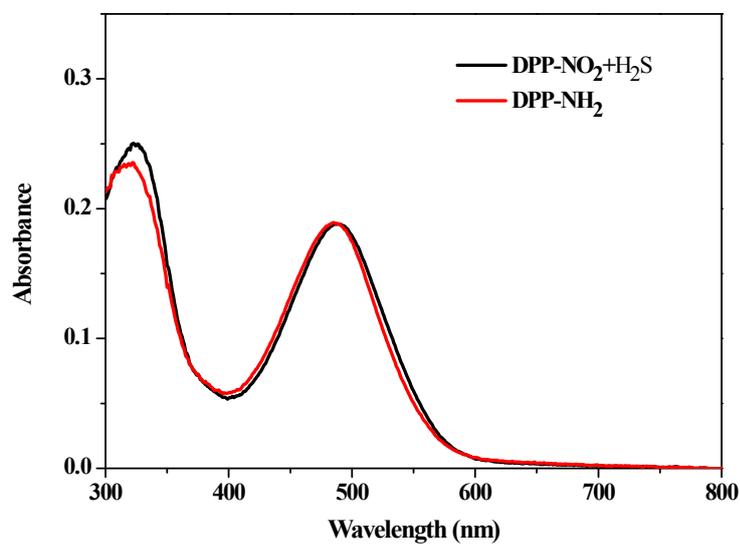


Figure S6. The absorption spectra comparison between **DPP-NH<sub>2</sub>** (10  $\mu$ M) in CH<sub>3</sub>CN and **DPP-NO<sub>2</sub>** (10  $\mu$ M) + H<sub>2</sub>S in CH<sub>3</sub>CN.