Three β-Octamolybdate-based Supramolecular Hybrids Constructed from a Bis-imidazolyl-bis-amide Ligand: Fast and Selective Adsorption Activities of Organic Dyes

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Table S1 Selected hydrogen bonding geometry (Å, °) for compound 1.

D–H…A	D–H / Å	H…A / Å	D…A / Å	D–H…A / o
O(1)– H(2WA)····O(2W)	0.85	2.20	3.021	160
$N(4)-H(4C)\cdots O(2W)$	0.86	2.16	2.903	145
N(6)−H(6B)····O(8)	0.86	2.19	3.009	160

Table S2 Selected hydrogen bonding geometry (Å, °) for compound 3.

D–H···A	D–H / Å	H…A / Å	D…A / Å	D–H…A / o
N(10)-H(6C)···O(24)	0.86	2.10	2.949	169
N(11)-H(5C)···O(32)	0.86	2.05	2.910	176
N(9)-H(9A)····O(19)	0.86	2.13	2.965	162

Table S3 Selected bond distances (Å) and angles (°) for compounds 1–3.

Compound 1			
Mo(1)-O(9)	1.696(2)	Mo(2)-O(13)	1.918(2)
Mo(1)-O(8)	1.759(2)	Mo(2)-O(8)	2.264(2)
Mo(1)-O(16)#1	1.926(2)	Mo(2)-O(11)#1	2.513(2)
Mo(1)-O(15)	1.962(2)	O(16)-Mo(1)#1	1.926(2)
Mo(1)-O(11)	2.109(2)	O(16)-Mo(3)	2.014(2)
Mo(1)-O(11)#1	2.371(2)	O(16)-Mo(4)	2.397(2)
Mo(1)-Mo(4)	3.2064(4)	Mo(3)-O(5)	1.692(3)
Co(1)-N(2)	2.049(3)	Mo(3)-O(4)	1.706(3)
Co(1)-N(1)	2.099(3)	Mo(3)-O(13)	1.886(2)
Co(1)-O(1W)	2.125(3)	Mo(3)-O(11)#1	2.371(2)
Co(1)-O(3)	2.128(3)	Mo(4)-O(10)	1.694(3)
Co(1)-O(2)	2.132(2)	Mo(4)-O(14)	1.713(3)
Co(1)-O(1)	2.189(3)	Mo(4)-O(6)#1	1.897(2)
O(15)-Mo(4)	1.998(2)	Mo(4)-O(11)	2.286(2)
O(15)-Mo(3)	2.311(2)	O(6)-Mo(4)#1	1.897(2)
Mo(2)-O(7)	1.682(3)	O(11)-Mo(3)#1	2.371(2)

Mo(2)-O(2)	1.727(2)	O(11)-Mo(1)#1	2.371(2)
Mo(2)-O(6)	1.910(2)	O(11)-Mo(2)#1	2.513(2)
N(2)-Co(1)-N(1)	176.77(12)	O(3)-Co(1)-O(2)	90.67(10)
N(2)-Co(1)-O(1W)	90.19(12)	N(2)-Co(1)-O(1)	77.49(11)
N(1)-Co(1)-O(1W)	87.68(12)	N(1)-Co(1)-O(1)	105.04(11
N(2)-Co(1)-O(3)	103.94(11)	O(1W)-Co(1)-O(1)	92.85(10)
N(1)-Co(1)-O(3)	78.31(11)	O(3)-Co(1)-O(1)	87.77(10)
O(1W)-Co(1)-O(3)	165.63(10)	O(2)-Co(1)-O(1)	165.13(10)
N(2)-Co(1)-O(2)	88.56(11)		
N(1)-Co(1)-O(2)	89.09(11)		
O(1W)-Co(1)-O(2)	92.31(10)		
Symmetry codes for 1:	#1 - x + 1, -y + 1, -z	-1.	
	Co	mpound 2	
Mo(1)-O(11)	1.693(4)	Mo(2)-O(14)	2.520(3)
Mo(1)-O(12)	1.764(3)	Mo(3)-O(8)	1.695(4)
Mo(1)-O(20)#1	1.926(3)	Mo(3)-O(7)	1.706(4)
Mo(1)-O(13)	1.964(4)	Mo(3)-O(15)	1.892(4)
Mo(1)-O(14)#1	2.109(3)	Mo(3)-O(20)	2.018(3)
Mo(1)-O(14)	2.372(3)	Mo(3)-O(13)	2.312(3)
Mo(1)-Mo(4)	3.2133(7)	Mo(3)-O(14)	2.355(3)
Zn(1)-N(1)	2.017(4)	Mo(4)-O(10)	1.693(4)
Zn(1)-N(3)	2.073(4)	Mo(4)-O(9)	1.707(4)
Zn(1)-O(1W)	2.149(4)	Mo(4)-O(17)#1	1.894(4)
Zn(1)-O(1)	2.173(4)	Mo(4)-O(13)	2.001(3)
Zn(1)-O(6)	2.204(4)	Mo(4)-O(14)#1	2.299(3)
Zn(1)-O(2)	2.217(4)	Mo(4)-O(20)	2.405(3)
Mo(2)-O(16)	1.689(4)	O(14)-Mo(1)#1	2.109(3)
Mo(2)-O(6)	1.724(4)	O(14)-Mo(4)#1	2.299(3)
Mo(2)-O(17)	1.910(4)	O(17)-Mo(4)#1	1.894(4)
Mo(2)-O(15)	1.919(3)	O(20)-Mo(1)#1	1.926(3)
Mo(2)-O(12)	2.262(3)	N(3)- $Zn(1)$ - $O(2)$	106.01(16)
N(1)-Zn(1)-N(3)	176.03(18)	O(1W)-Zn(1)-O(2)	94.11(15)
N(1)-Zn(1)-O(1W)	90.76(16)	O(1)-Zn(1)-O(2)	88.44(15)
N(3)-Zn(1)-O(1W)	88.30(16)	O(6)-Zn(1)-O(2)	163.25(13)
N(1)-Zn(1)-O(1)	102.33(16)		
N(3)-Zn(1)-O(1)	78.66(16)		
O(1W)-Zn(1)-O(1)	166.91(14)		
N(1)-Zn(1)-O(6)	86.41(15)		
N(3)-Zn(1)-O(6)	89.77(16)		
O(1W)-Zn(1)-O(6)	91.91(14)		
	89.23(15)		
O(1)-Zn(1)-O(6)			

Mo(1)-O(18)	1.692(5)	Mo(4)-Mo(3)#2	3.2083(9)
Mo(1)-O(17)	1.753(4)	Mo(5)-O(32)	1.692(5)
Mo(1)-O(21)#1	1.946(4)	Mo(5)-O(16)	1.695(5)
Mo(1)-O(5)	1.950(5)	Mo(5)-O(12)	1.895(5)
Mo(1)-O(15)	2.146(4)	Mo(5)-O(5)	2.005(4)
Mo(1)-O(15)#1	2.403(4)	Mo(5)-O(15)	2.315(4)
Mo(1)-Mo(2)#1	3.2108(9)	Mo(5)-O(21)	2.360(4)
Cu(1)-N(1)	1.902(6)	Mo(6)-O(27)	1.688(5)
Cu(1)-N(2)	1.918(6)	Mo(6)-O(28)	1.699(5)
Cu(1)-O(1)	1.965(5)	Mo(6)-O(26)	1.875(5)
Cu(1)-O(2)	1.967(5)	Mo(6)-O(29)	2.007(4)
Mo(2)-O(13)	1.695(5)	Mo(6)-O(30)	2.393(5)
Mo(2)-O(9)	1.719(5)	Mo(6)-O(25)	2.395(5)
Mo(2)-O(10)	1.888(4)	Mo(7)-O(8)	1.682(5)
Mo(2)-O(21)	1.985(4)	Mo(7)-O(7)	1.707(5)
Mo(2)-O(15)#1	2.315(4)	Mo(7)-O(12)#1	1.933(5)
Mo(2)-O(5)	2.347(5)	Mo(7)-O(10)	1.934(5)
Mo(2)-Mo(1)#1	3.2108(9)	Mo(7)-O(17)	2.296(5)
Cu(2)-N(5)	1.951(6)	Mo(7)-O(15)#1	2.492(4)
Cu(2)-N(6)	1.958(6)	Mo(8)-O(14)	1.690(5)
Cu(2)-O(4)	1.980(5)	Mo(8)-O(20)	1.694(6)
Cu(2)-O(3)	1.981(5)	Mo(8)-O(26)#2	1.905(5)
Cu(2)-O(5W)	2.357(6)	Mo(8)-O(23)	1.964(5)
Mo(3)-O(11)	1.691(5)	Mo(8)-O(19)	2.319(5)
Mo(3)-O(19)	1.765(4)	Mo(8)-O(30)#2	2.474(4)
Mo(3)-O(25)#2	1.935(4)	O(12)-Mo(7)#1	1.933(5)
Mo(3)-O(29)	1.953(4)	O(15)-Mo(2)#1	2.315(4)
Mo(3)-O(30)	2.123(4)	O(15)-Mo(1)#1	2.403(4)
Mo(3)-O(30)#2	2.393(5)	O(15)-Mo(7)#1	2.492(4)
Mo(3)-Mo(4)#2	3.2082(9)	O(21)-Mo(1)#1	1.946(4)
Mo(4)-O(31)	1.706(5)	O(25)-Mo(3)#2	1.935(4)
Mo(4)-O(24)	1.715(5)	O(26)-Mo(8)#2	1.905(5)
Mo(4)-O(23)	1.877(5)	O(30)-Mo(4)#2	2.310(5)
Mo(4)-O(25)	1.989(4)	O(30)-Mo(3)#2	2.393(5)
Mo(4)-O(30)#2	2.310(5)	O(30)-Mo(8)#2	2.474(4)
Mo(4)-O(29)	2.337(5)	O(4)-Cu(2)-O(3)	179.4(3)
N(1)-Cu(1)-N(2)	176.2(3)	N(5)-Cu(2)-O(5W)	103.5(2)
N(1)-Cu(1)-O(1)	83.4(2)	N(6)-Cu(2)-O(5W)	89.3(3)
N(2)-Cu(1)-O(1)	98.6(2)	O(4)-Cu(2)-O(5W)	86.3(2)
N(1)-Cu(1)-O(2)	94.0(2)	O(3)-Cu(2)-O(5W)	94.3(2)
N(2)-Cu(1)-O(2)	83.8(2)		
O(1)-Cu(1)-O(2)	176.1(2)		
N(5)-Cu(2)-N(6)	167.2(3)		
N(5)-Cu(2)-O(4)	96.7(2)		

N(6)-Cu(2)-O(4)	83.1(2)	
N(5)-Cu(2)-O(3)	83.2(2)	
N(6)-Cu(2)-O(3)	96.8(2)	
Symmetry codes for 3: $\#1 - x - y + 1 - z + 1$: $\#2 - x + 1 - y + 1 - z$		



Fig.S1 (a) Stick/ball view of the asymmetric unit of 2. The hydrogen atoms are omitted for clarity.(b) View of the 3D supramolecular structure of 2.



Fig.S2 View of the 3D supramolecular structure of 1.



Fig.S3 The IR spectra of compounds 1–3 and L.



Fig.S4 The thermogravimetric (TG) curves of compounds 1–3.



Fig.S5 The UV absorption curves of RhB and MO with 2.



Fig.S6 (a-e) The UV absorption curves of CR, MB, RhB, MO and MG with 1.



Fig.S7 The UV absorption curves of RhB and MO with 3; The inserted photographs show the change of solution colors with the increase of time.