

Electronic Supplementary Information

AgPd-MnO_x supported on carbon nanospheres: An efficient catalyst for dehydrogenation of formic acid

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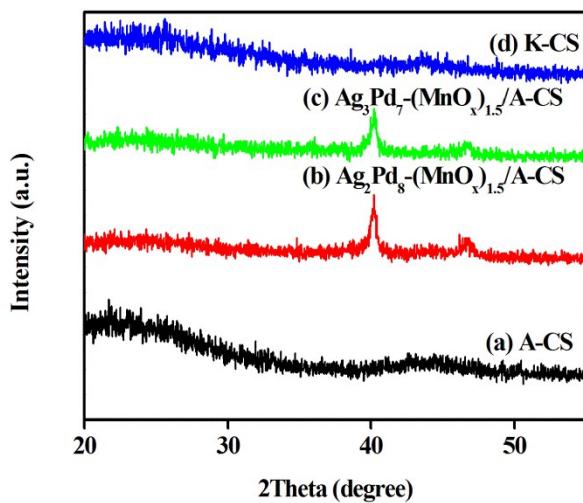


Fig. S1 X-ray diffraction patterns for as-prepared carbon spheres and AgPd-MnO_x/carbon spheres.
 (a) A-CS, (b) Ag₂Pd₈-(MnO_x)_{1.5}/A-CS, (c) Ag₃Pd₇-(MnO_x)_{1.5}/A-CS,(d) K-CS

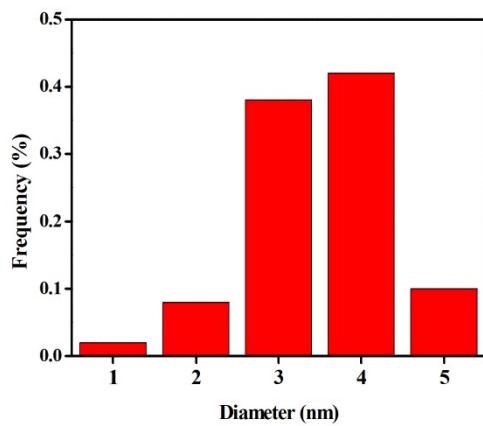


Fig. S2 The corresponding particle size distributions of Ag₁Pd₉-(MnO_x)_{1.5}/A-CS.

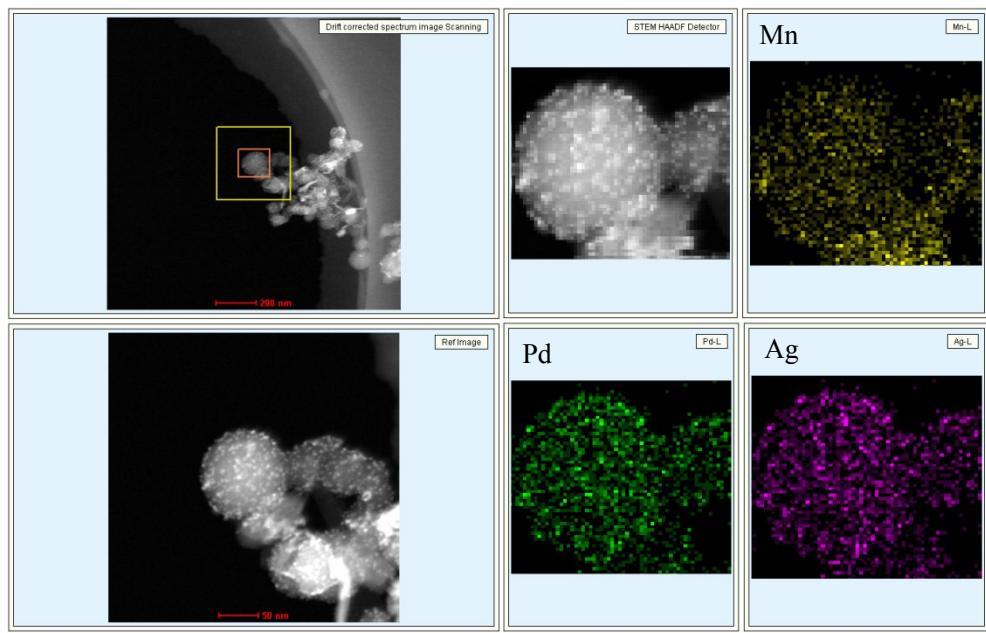


Fig. S3 The corresponding elemental mapping for Ag, Pd and Mn

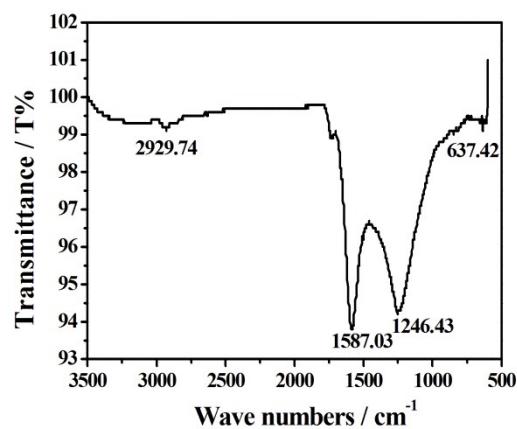


Fig. S4 The ATR FT-IR spectra of $\text{Ag}_1\text{Pd}_9\text{-}(\text{MnO}_x)_{1.5}/\text{A-CS}$

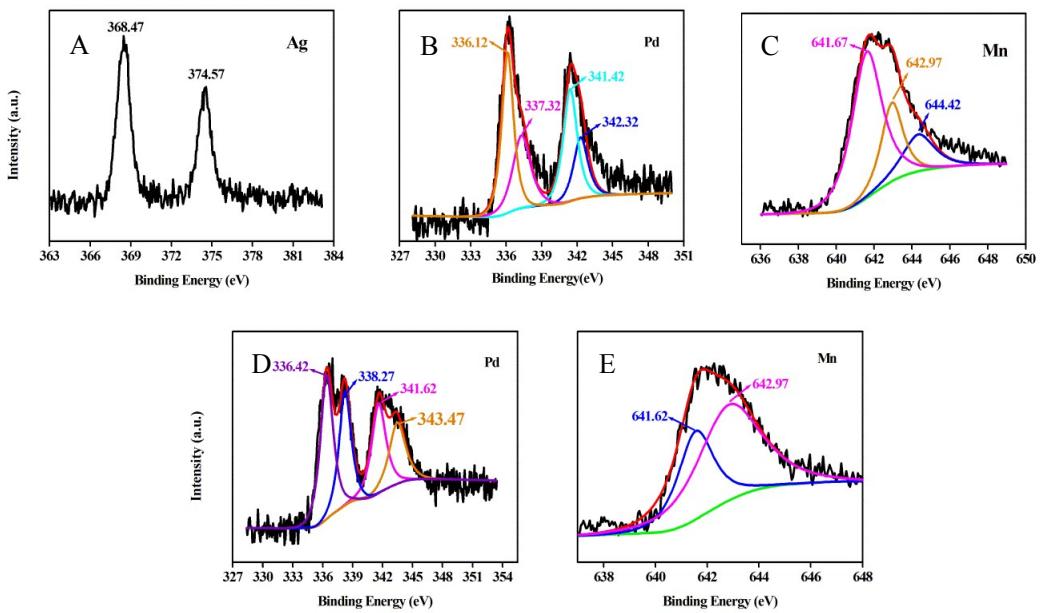


Fig. S5 XPS patterns of $\text{Ag}_7\text{Pd}_3\text{-}(\text{MnO}_x)_{1.5}/\text{A-CS}$ (A, B and C) and $\text{Pd}_{10}\text{-}(\text{MnO}_x)_{1.5}/\text{A-CS}$ (D and E)

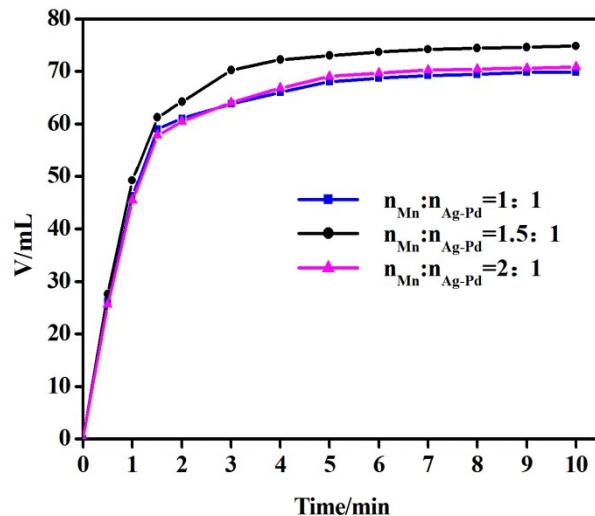


Fig. S6 Different molar ratios of Mn and Ag-Pd
(The reactions were performed at 50°C, $n_{\text{FA}}:n_{\text{PF}} = 2.5:7.5$)

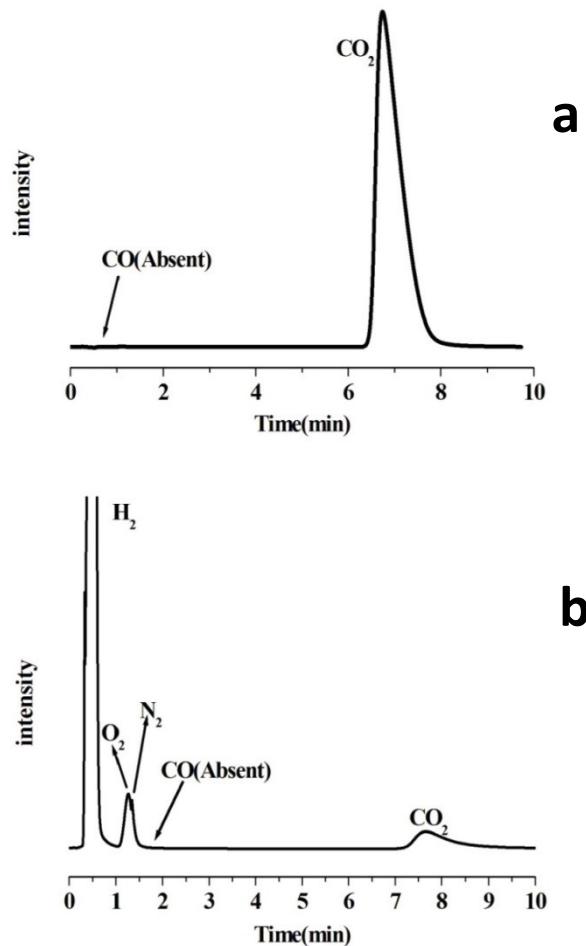


Fig. S7 GC spectrum using TCD for the evolved gas from FA/PF solution over Ag₁Pd₉-(MnO_x)_{1.5}/A-CS. The limit of detection for CO is 0.1 ppm.

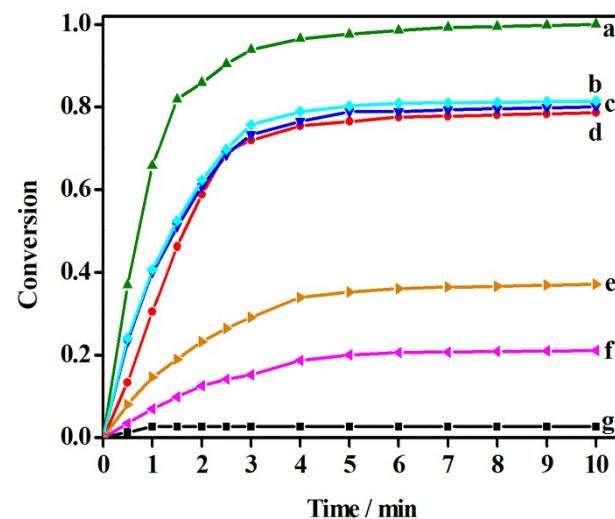


Fig. S8 Conversion of FA with different catalysts versus time at 50°C.

(a)Ag₁Pd₉-(MnO_x)_{1.5}/A-CS, (b)Ag₃Pd₇-(MnO_x)_{1.5}/A-CS, (c)Ag₂Pd₈-(MnO_x)_{1.5}/A-CS, (d)Pd₁₀-(MnO_x)_{1.5}/A-CS, (e)Ag₇Pd₃-(MnO_x)_{1.5}/A-CS, (f)Ag₈Pd₂-(MnO_x)_{1.5}/A-CS), (g) Ag₁₀-(MnO_x)_{1.5}/A-CS

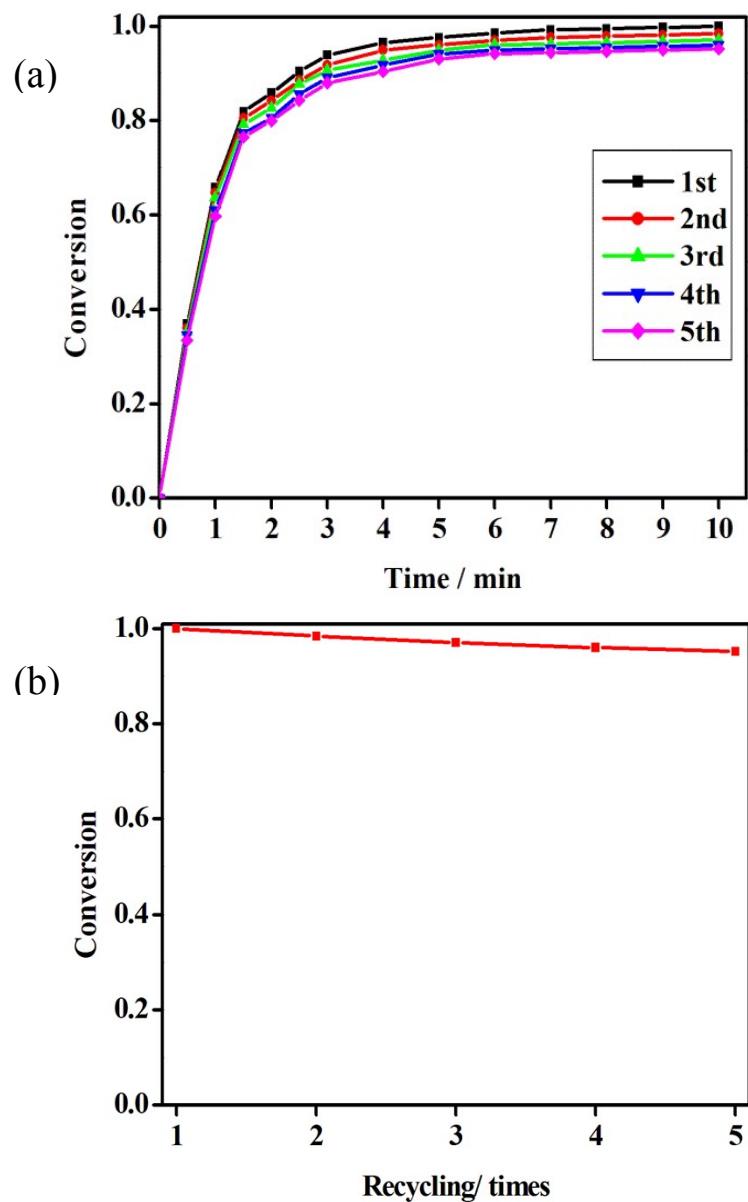


Fig. S9 Conversion of FA catalyzed by $\text{Ag}_1\text{Pd}_{9-(\text{MnO}_x)_{1.5}}/\text{A-CS}$ versus time (a) and recycling times (b) at 50°C .

Table S1 The dosages of reagents for the synthesis of different catalysts

Catalysts	CS/mg	AgNO ₃ /mL (1mg/mL)	H ₂ PdCl ₄ /mL (1mg/mL)	MnSO ₄ /mL (1mg/mL)
Ag ₁ Pd ₉ -(MnO _x) _{1.5} /A-CS	41	0.79	7.51	11.90
Ag ₂ Pd ₈ -(MnO _x) _{1.5} /A-CS	41	1.58	6.68	11.90
Ag ₃ Pd ₇ -(MnO _x) _{1.5} /A-CS	41	2.37	5.84	11.90
Ag ₇ Pd ₃ -(MnO _x) _{1.5} /A-CS	41	5.53	2.50	11.90
Ag ₈ Pd ₂ -(MnO _x) _{1.5} /A-CS	41	6.32	1.67	11.90
Ag ₁₀ -(MnO _x) _{1.5} /A-CS	41	7.9	0	11.90
Pd ₁₀ -(MnO _x) _{1.5} /A-CS	41	0	8.35	11.90
Ag ₁ Pd ₉ -(MnO _x) ₁ /A-CS	41	0.79	7.51	8.0
Ag ₁ Pd ₉ -(MnO _x) ₂ /A-CS	41	0.79	7.51	15.9

Table S2 The contents of Pd and Ag in AgPd-(MnO_x)_{1.5}/A-CS determined by ICP-AES

Catalysts	Ag (wt %)	Pd (wt %)	Initial mass ratio of Ag and Pd
Ag ₁ Pd ₉ -(MnO _x) _{1.5} /A-CS	0.97	8.92	1:9
Ag ₂ Pd ₈ -(MnO _x) _{1.5} /A-CS	1.96	7.94	2:8
Ag ₃ Pd ₇ -(MnO _x) _{1.5} /A-CS	2.92	6.90	3:7
Ag ₇ Pd ₃ -(MnO _x) _{1.5} /A-CS	6.91	2.91	7:3
Ag ₈ Pd ₂ -(MnO _x) _{1.5} /A-CS	7.93	1.95	8:2

Table S3 The N₂ adsorption-desorption isotherms of CS and Ag₁Pd₉-(MnO_x)_{1.5}/CS

Sample	S _{BET} (m ² g ⁻¹)	Pore volume (cm ³ g ⁻¹)	Pore diameter (nm)
A-CS	782.63	0.65	3.39
Ag ₁ Pd ₉ -(MnO _x) _{1.5} /A-CS	722.47	0.61	3.30
K-CS	2084.06	1.08	2.08
Ag ₁ Pd ₉ -(MnO _x) _{1.5} /K-CS	1519.24	0.80	2.12