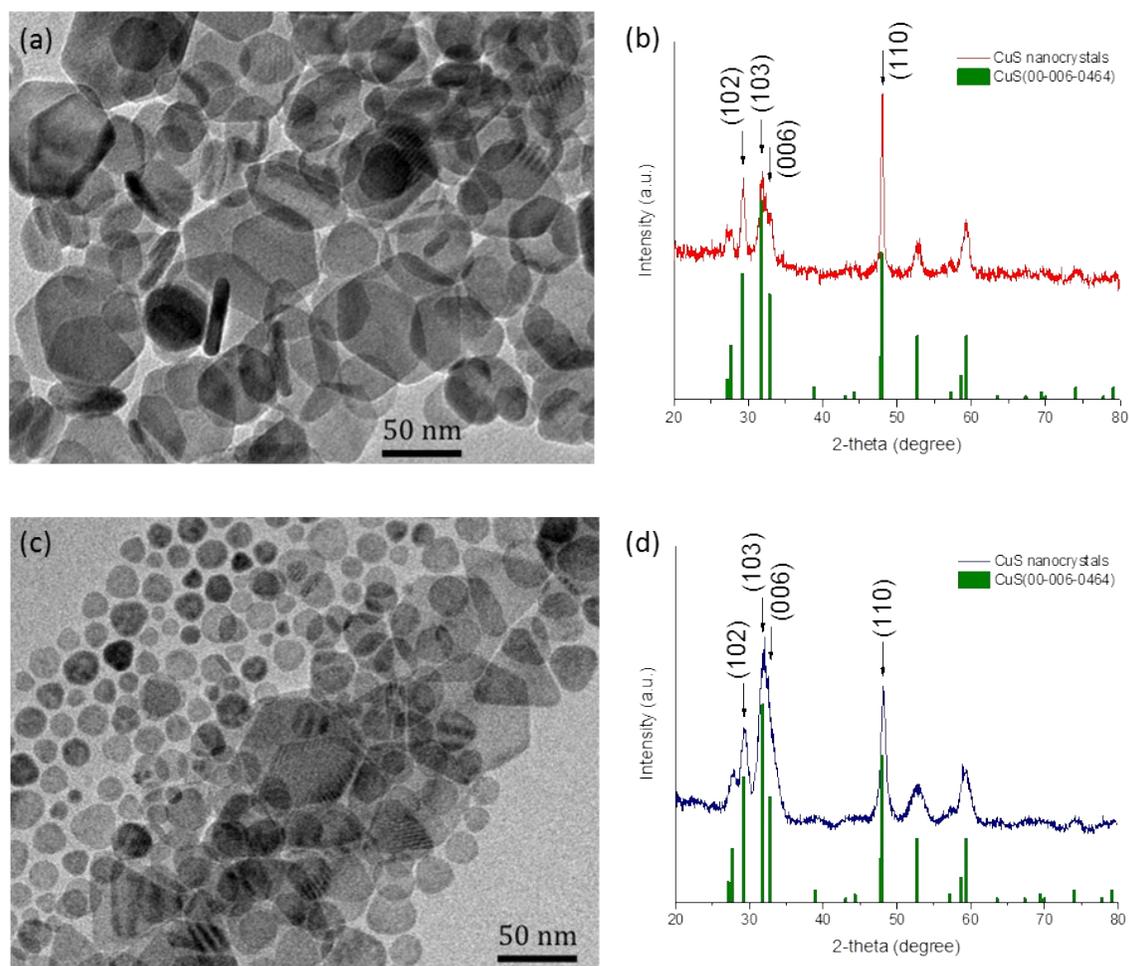


## Supporting Information

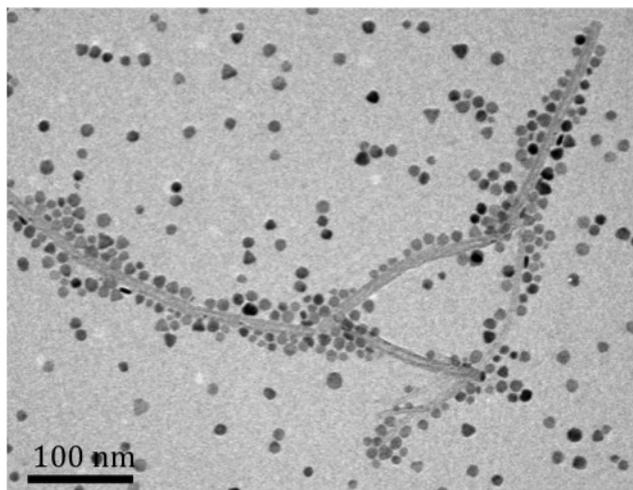
# Covellite CuS nanocrystals: realizing rapid microwave-assisted synthesis in air and unraveling the disappearance of their plasmon resonance after coupling with carbon nanotubes

Mee Rahn Kim,<sup>a</sup> Hassan A. Hafez,<sup>a</sup> Xin Chai,<sup>a</sup> Lucas V. Besteiro,<sup>b</sup> Long Tan,<sup>a</sup>  
Tsuneyuki Ozaki,<sup>\*,a</sup> Alexander O. Govorov,<sup>b</sup> Ricardo Izquierdo<sup>c</sup> and Dongling Ma<sup>\*,a</sup>

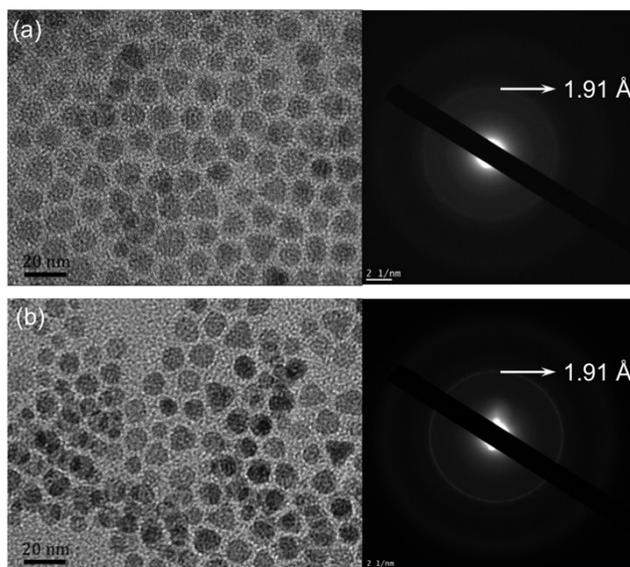
E-mail: ozaki@emt.inrs.ca, ma@emt.inrs.ca



**Fig. S1.** (a, b) TEM image and XRD pattern of CuS nanocrystals synthesized by the microwave-assisted method with the reaction time of 10 min; (c, d) TEM image and XRD pattern of CuS nanocrystals synthesized by the oil-bath method with the reaction time of 10 min. Much larger covellite CuS nanocrystals were obtained from a longer reaction time.

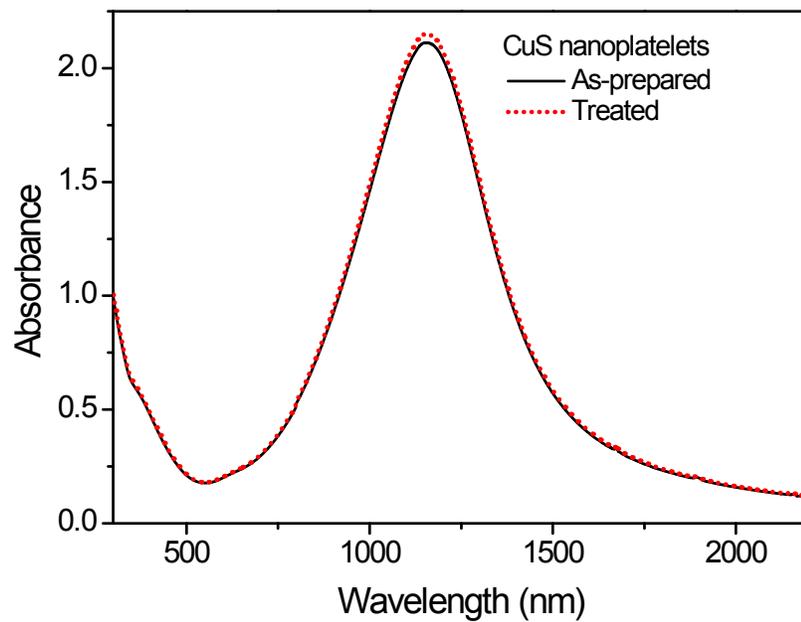


**Fig. S2.** TEM image showing the mixture of pristine MWCNTs and as-prepared CuS nanoplatelets after being sonicated in toluene for 30 min. The pristine MWCNTs did not undergo the carboxyl functionalization procedure.



**Fig. S3.** (Left) TEM images and (right) SAED patterns of (a) as-prepared CuS nanoplatelets, and (b) CuS nanoplatelets following the same procedure for the preparation of the CuS-MWCNT hybrid in the absence of MWCNTs. The nanoplatelet sizes in both

TEM images are similar and the SAED patterns in a and b prove both nanocrystals are covellite CuS.



**Fig. S4.** UV-Vis-NIR absorption spectra of as-prepared CuS nanoplatelets (solid, black) dispersed in toluene, and treated CuS nanoplatelets (dotted, red) dispersed in toluene for the comparison with those coupled with MWCNTs.